

# Week 2-5

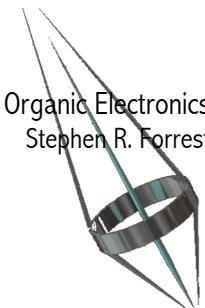
## Light emitters 5

### OLED Reliability

### Lasers

### Chapter 6.7-6.8

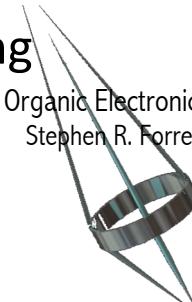
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# Reliability Testing Methodologies

- Need to set clear metrics for failure
  - Example: Operating time for initial luminance ( $L_0$ ) to decrease 10% from its initial value (called T90, or LT90)
  - Employ a population of equivalent devices and monitor their performance parameter (e.g. luminance) under normal operating conditions
  - If degradation slow, then an empirical degradation relationship is determined to extrapolate time to failure
    - Example: **Stretched exponential function:**
$$L(t) = L_0 \exp(-t/\tau)^\beta \quad \tau, \beta = \text{empirical constants}$$
  - If degradation too slow, need to accelerate via increased  $T$  or  $L_0$ .
    - Accelerated conditions must not introduce new failure modes
    - Need empirical relations to normalize lifetime to standard operating conditions (called **acceleration factors**)

$$LTx(L_0) = LTx(L_{0tst}) \cdot \left[ \frac{L_{0tst}}{L_0} \right]^n \quad n = \text{empirical acceleration factor}$$



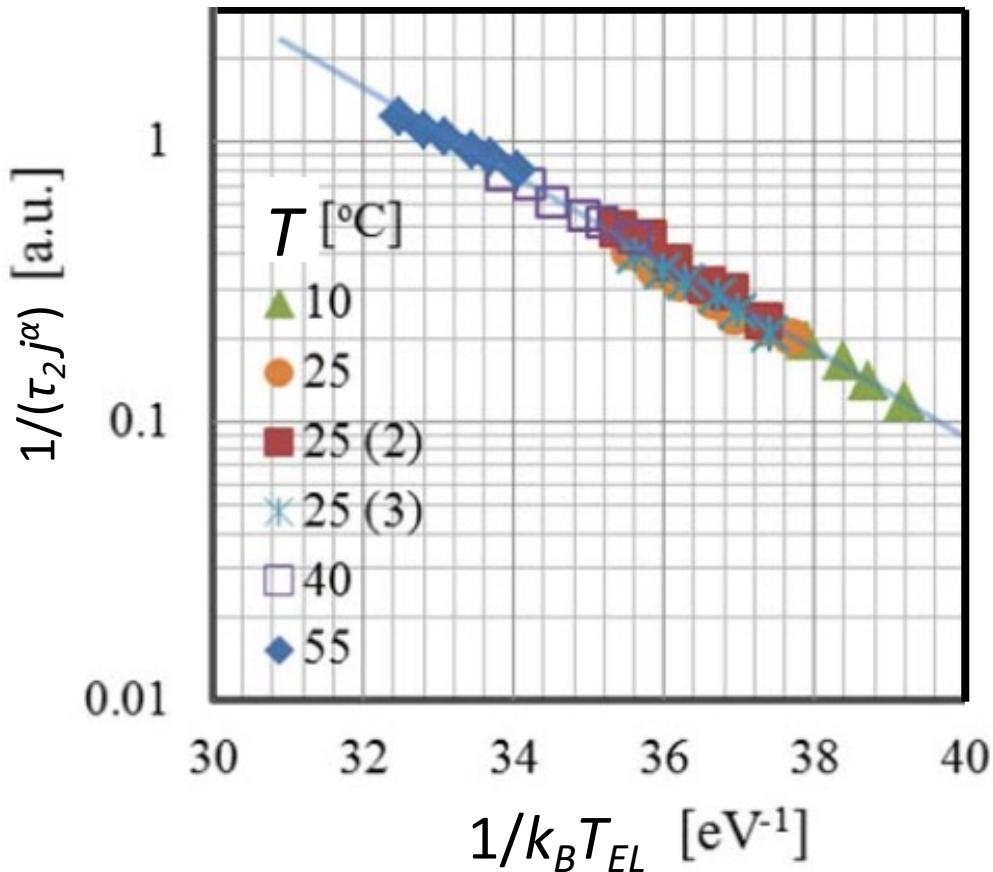
# Accelerated Degradation Methodologies

Sum of lifetimes alternative empirical relation):

$$L(t)/L_0 = \lambda \exp(-t/\tau_1) + (1-\lambda) \exp(-t/\tau_2)$$

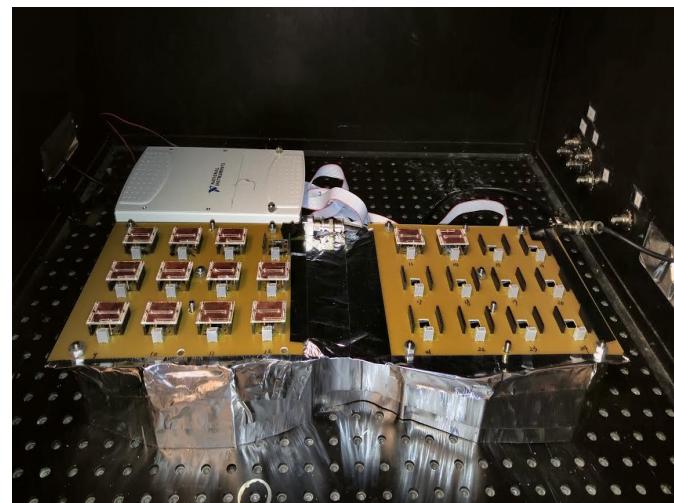
Burn-in                      Long term decay

Example data: Green PHOLED



$$\frac{1}{\tau_2} = K'' j^\alpha \exp(-\Delta E_{A0}/k_B T)$$

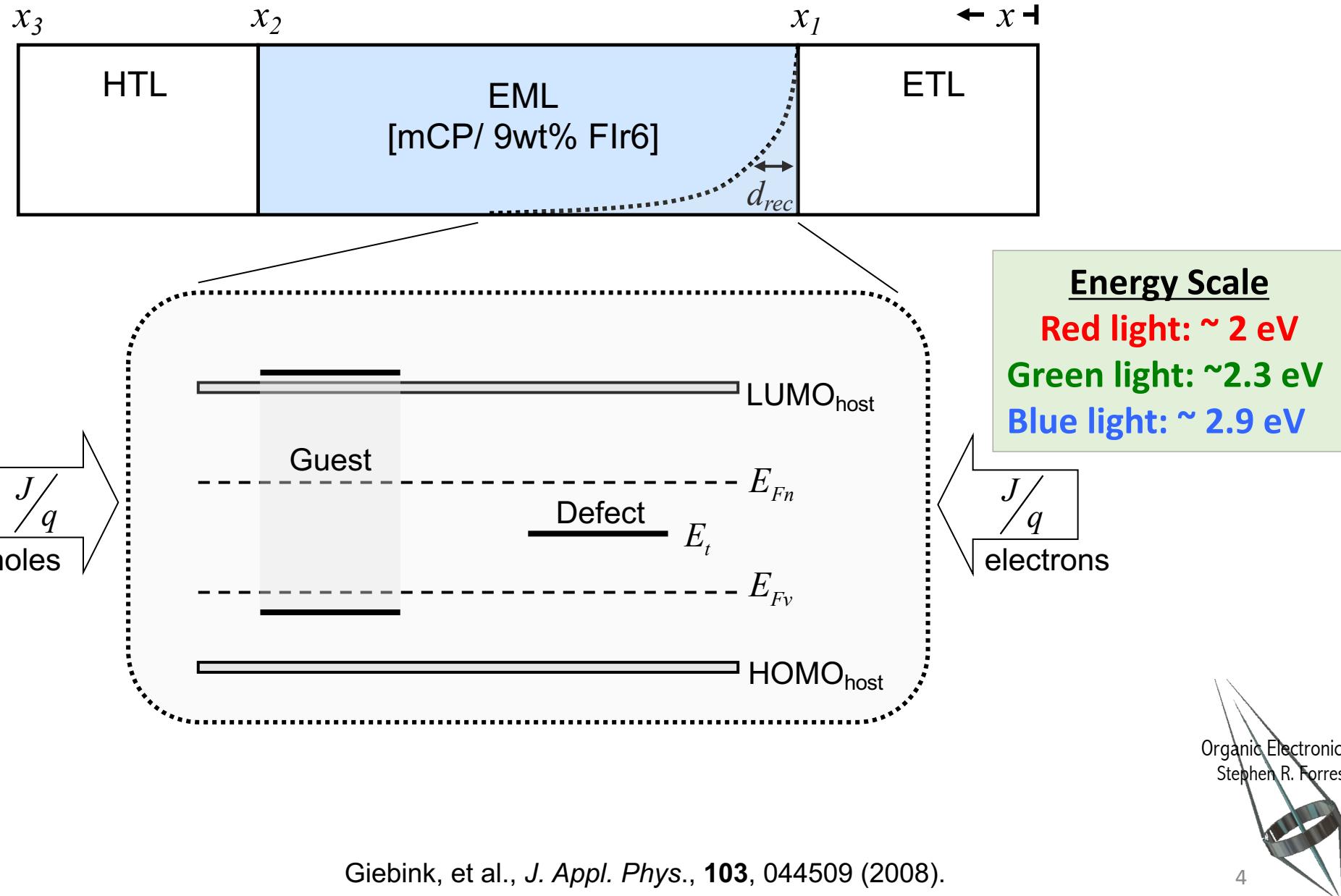
$\Delta E_{A0}$ =thermal activation of degradation  
 $\alpha$  = current acceleration factor



Measuring populations of identical devices

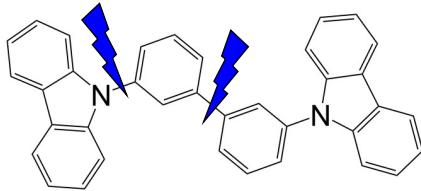


# Intrinsic Lifetime Limits of OLEDs



# Degradation Routes

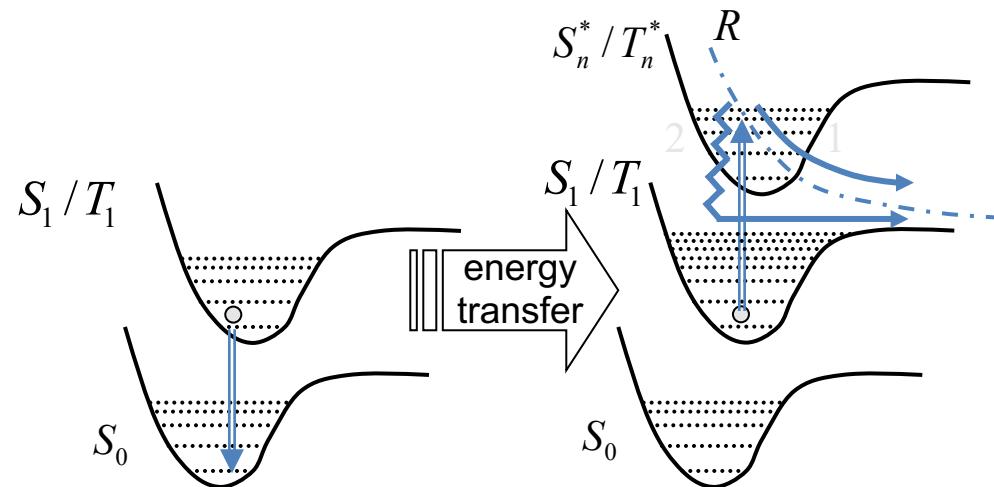
- Energetically Driven
  - Lifetime:  $R > G > B$
- Two particle interactions lead to luminance loss
  - Exciton on phosphor, polaron on host
  - Exciton-exciton also possible



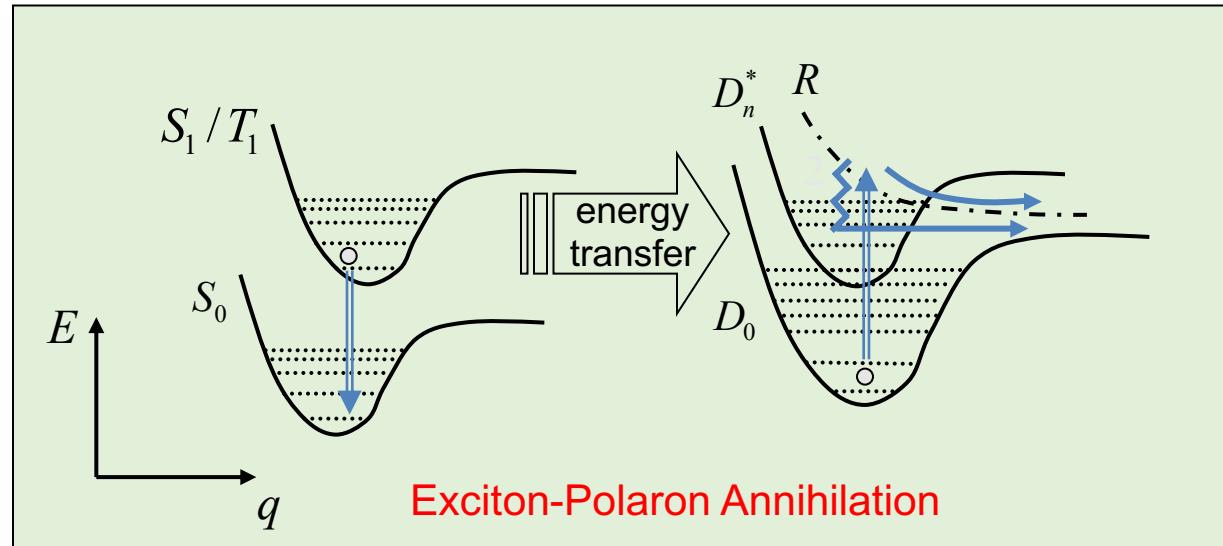
Bond	BE(eV)	Bond	BE(eV)
C-C	3.64	N-N	1.69
C-H	4.28	N-O	2.08
C-O	3.71	N-H	4.05
C-N	3.04	O-O	1.51
C-F	5.03	H-H	4.52

## Bond cleavage

Broken bonds?  $\rightarrow$  Defects!



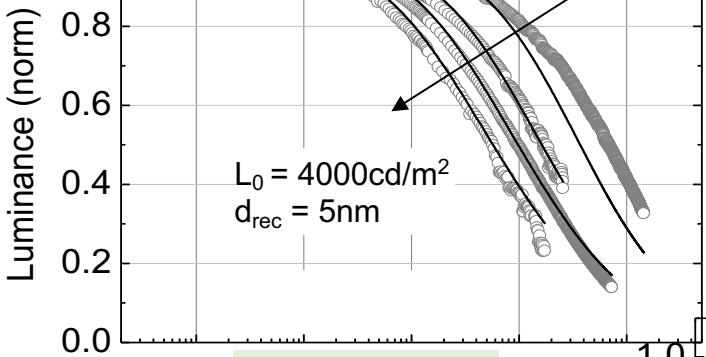
Exciton-Exciton Annihilation



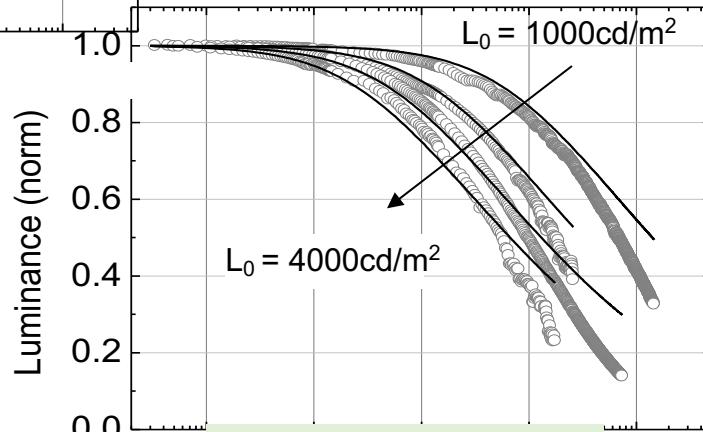
Exciton-Polaron Annihilation

Triplet energy (~2.8 eV) + polaron (~3.3 eV) = hot polaron ( $\geq 6$  eV)

# Luminance Decay vs Time



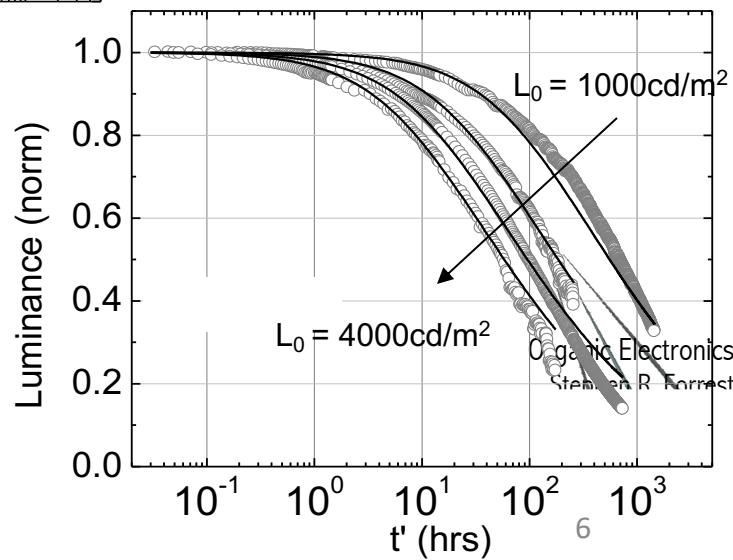
Exciton  
Localization



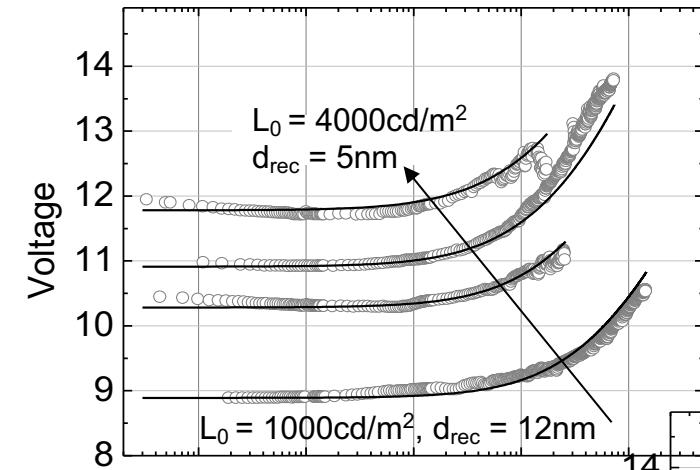
Exciton-Exciton  
Annihilation

Defect Generation Rates

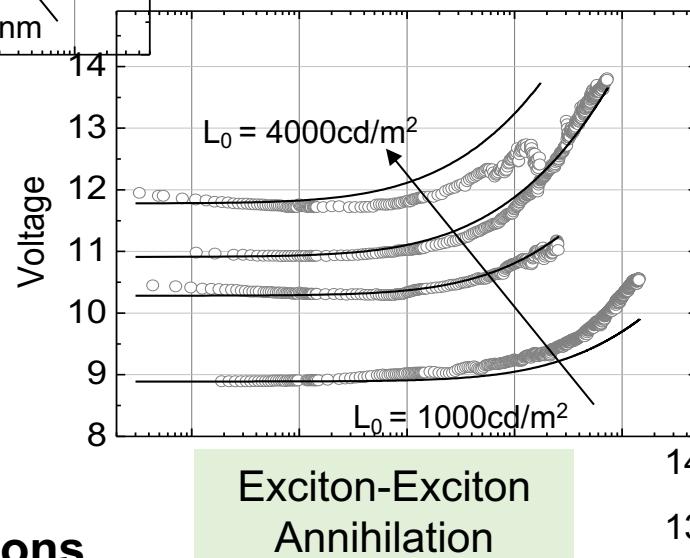
$$\frac{dQ(x, t')}{dt'} = \left\{ \begin{array}{ll} K_X n(x, t') & K_X p(x, t') \\ K_X N(x, t') & \\ K_X N^2(x, t') & \\ K_X N(x, t') n(x, t') & K_X N(x, t') p(x, t') \end{array} \right. \quad \begin{array}{l} P \\ E \\ E-E \\ E-P \end{array}$$



# Drive Voltage Drift with Aging



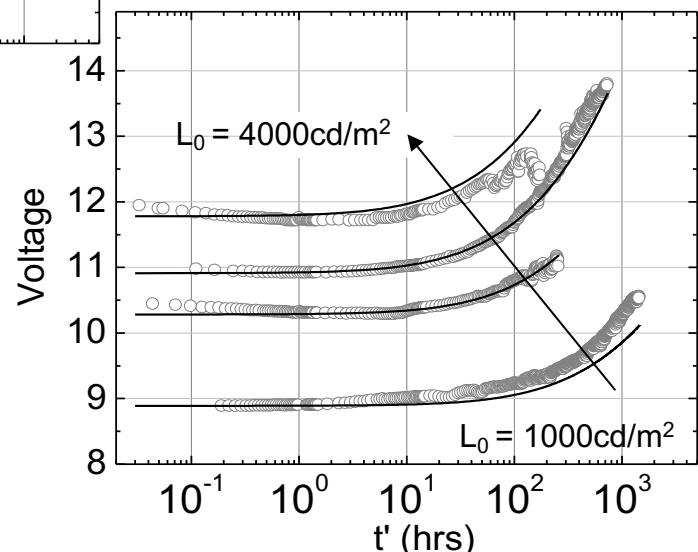
Exciton  
Localization



Exciton-Exciton  
Annihilation

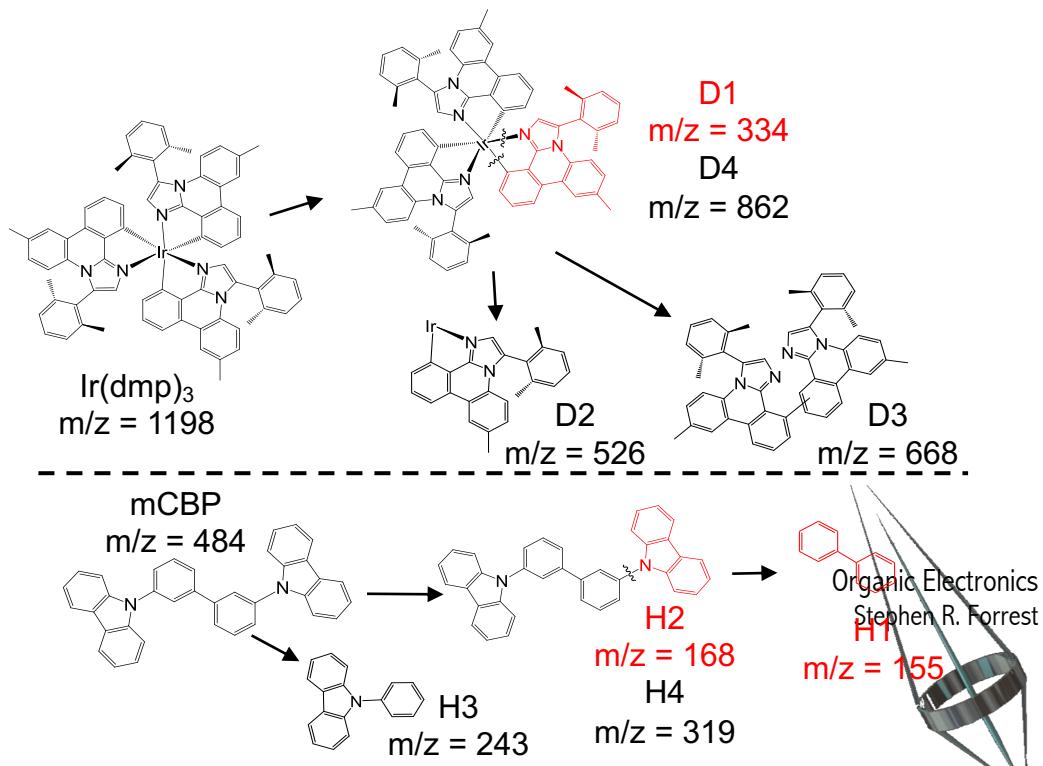
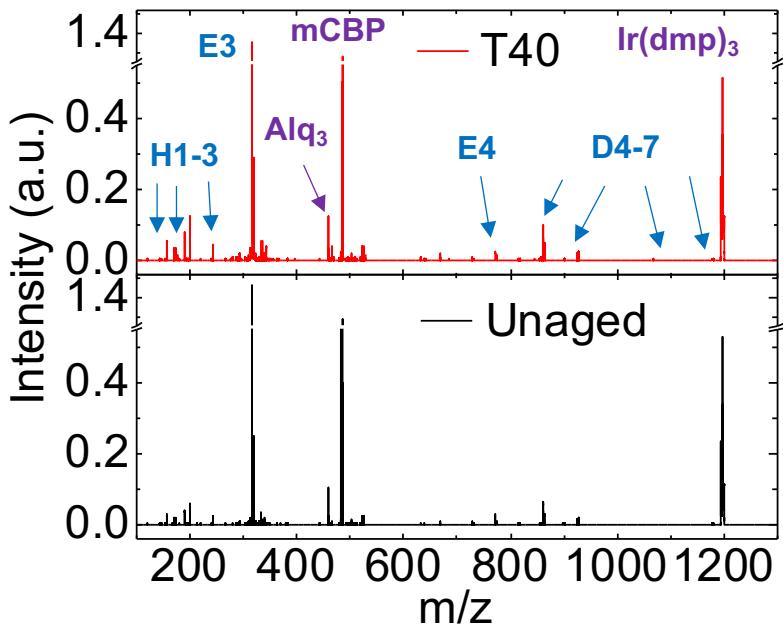
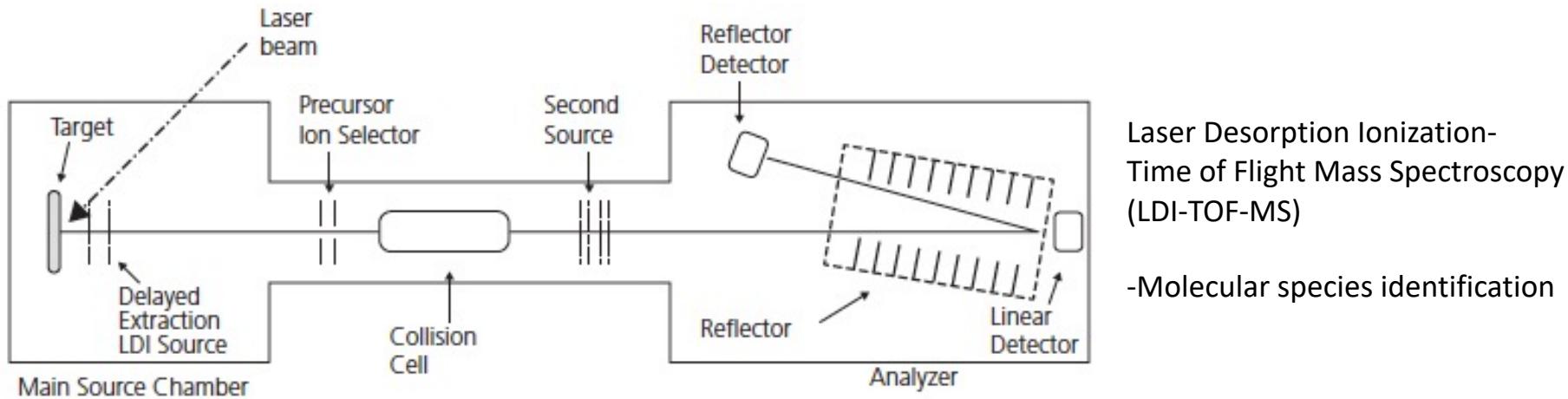
## Conclusions

- $\bullet Q \sim 10^{18} \text{ cm}^{-3} \rightarrow 50\% \text{ increase in quenching}$
- $\bullet \text{At } 1000 \text{ cd/m}^2, \text{ formation rate} = 10^{12} \text{ cm}^{-2} \text{s}^{-1}$ 
  - 1 in  $5 \times 10^8$  E-P encounters leads to defect
  - Increasing recombination zone width extends lifetime
  - Guest triplets/host polarons most active

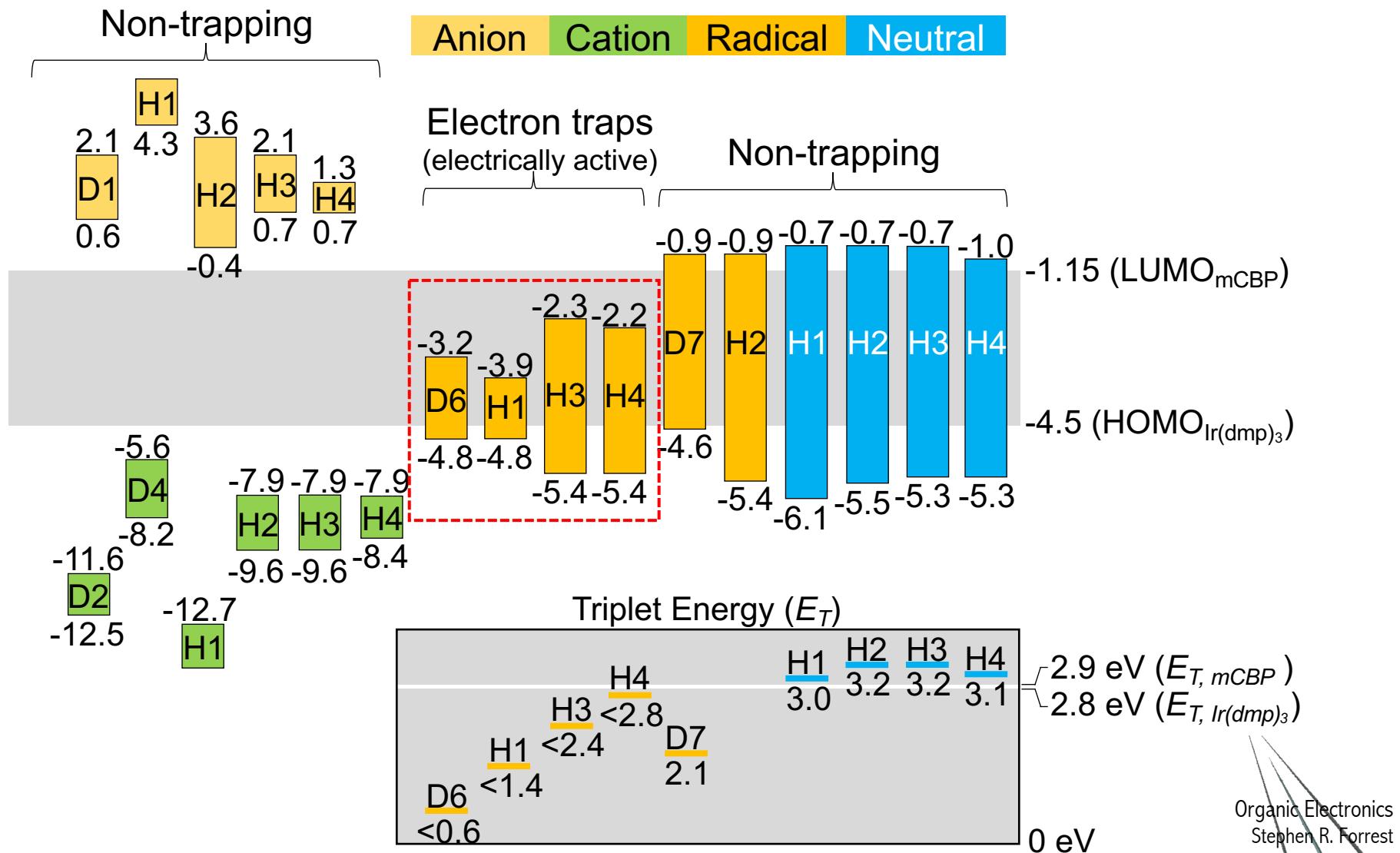


Exciton-Polaron  
Annihilation

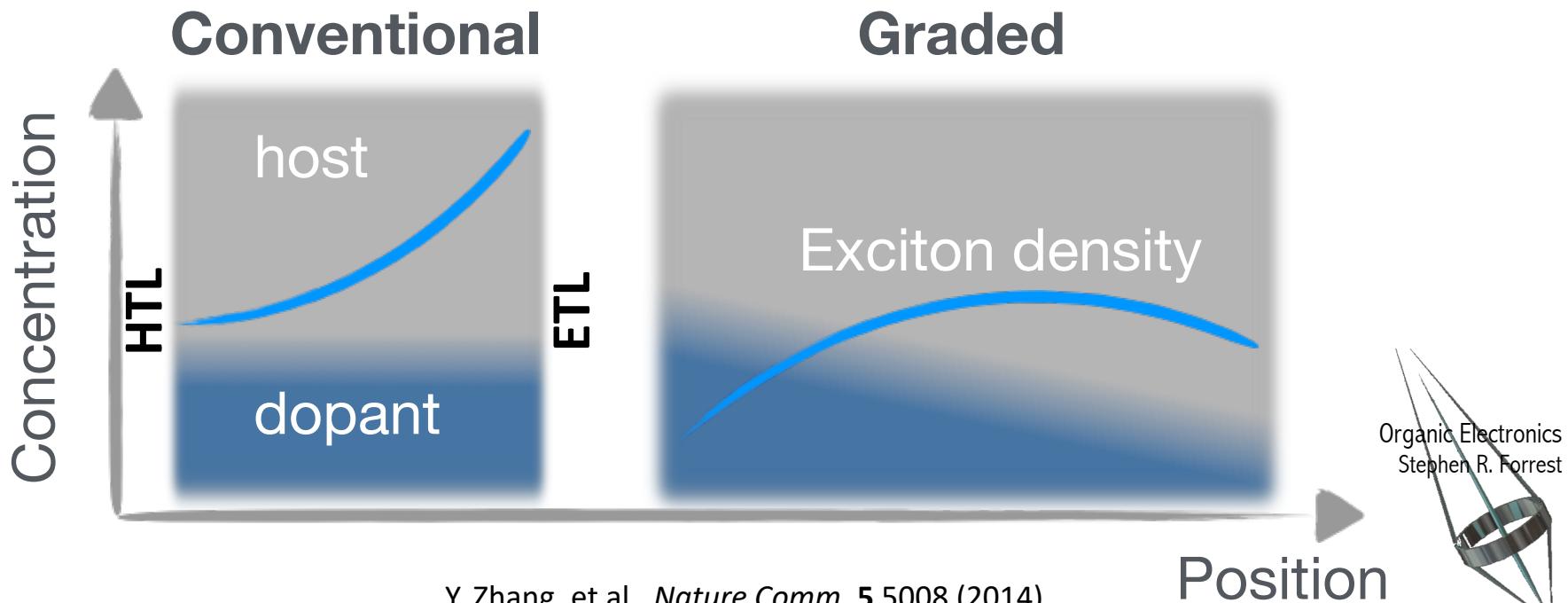
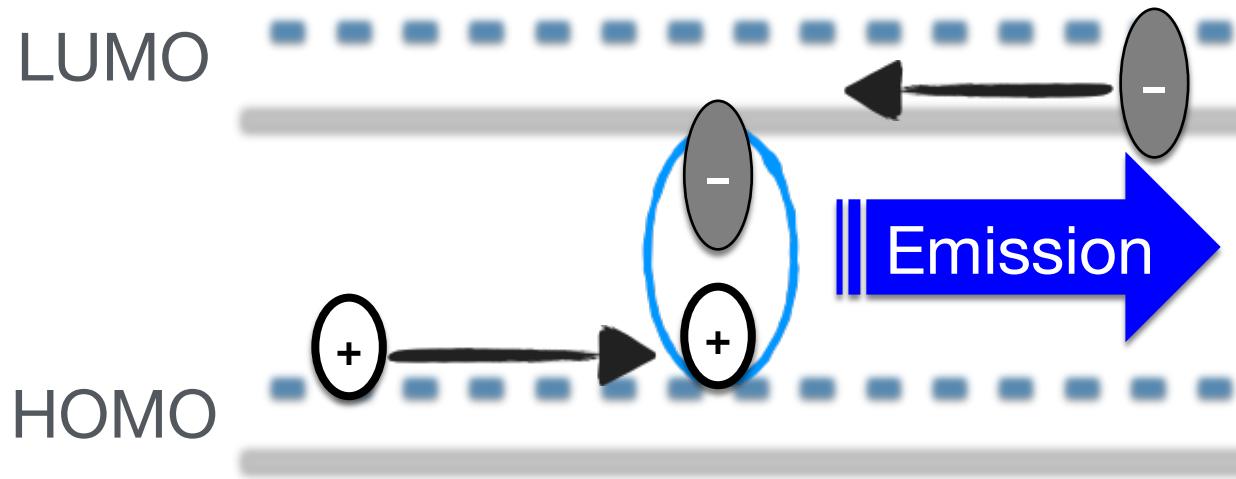
# Evidence for Defect Formation: Molecular Fragmentation



# Identification of Defect Energies

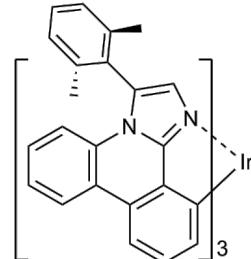
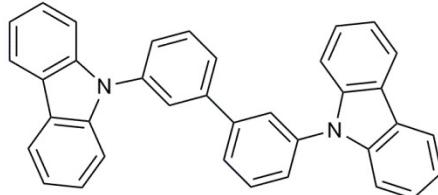


# Reducing Exciton Density to Increase Lifetime



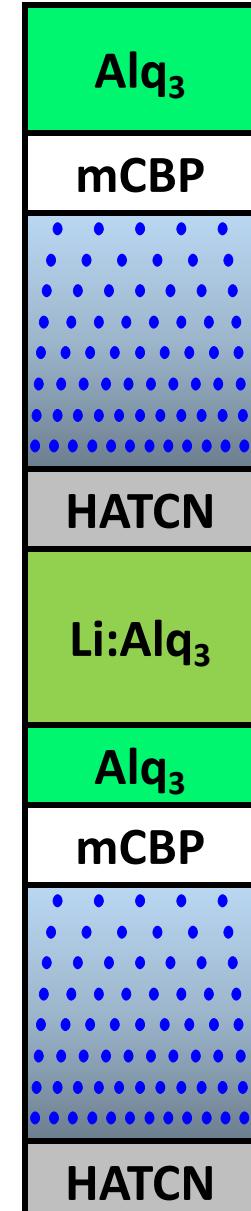
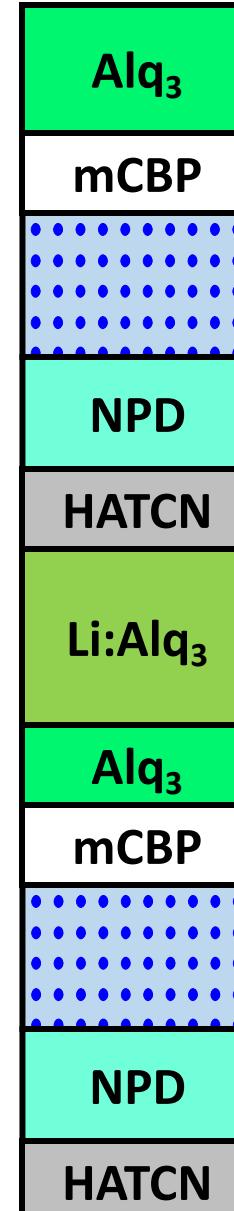
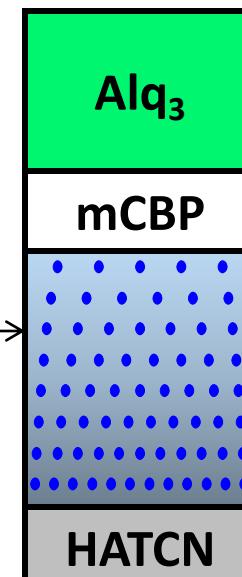
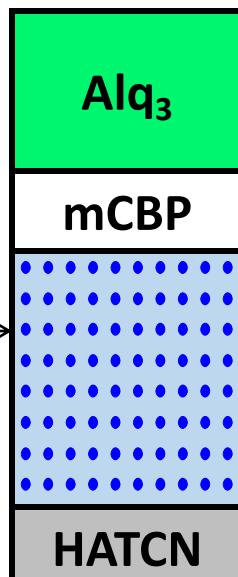
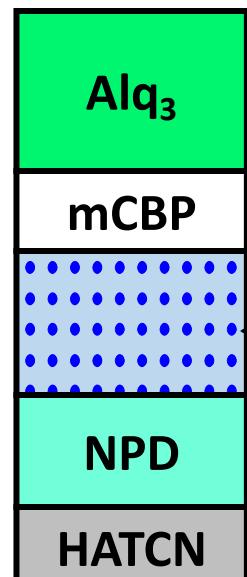
# Spreading the recombination zone: Dopant/Host Grading

### 3 Different test device structures

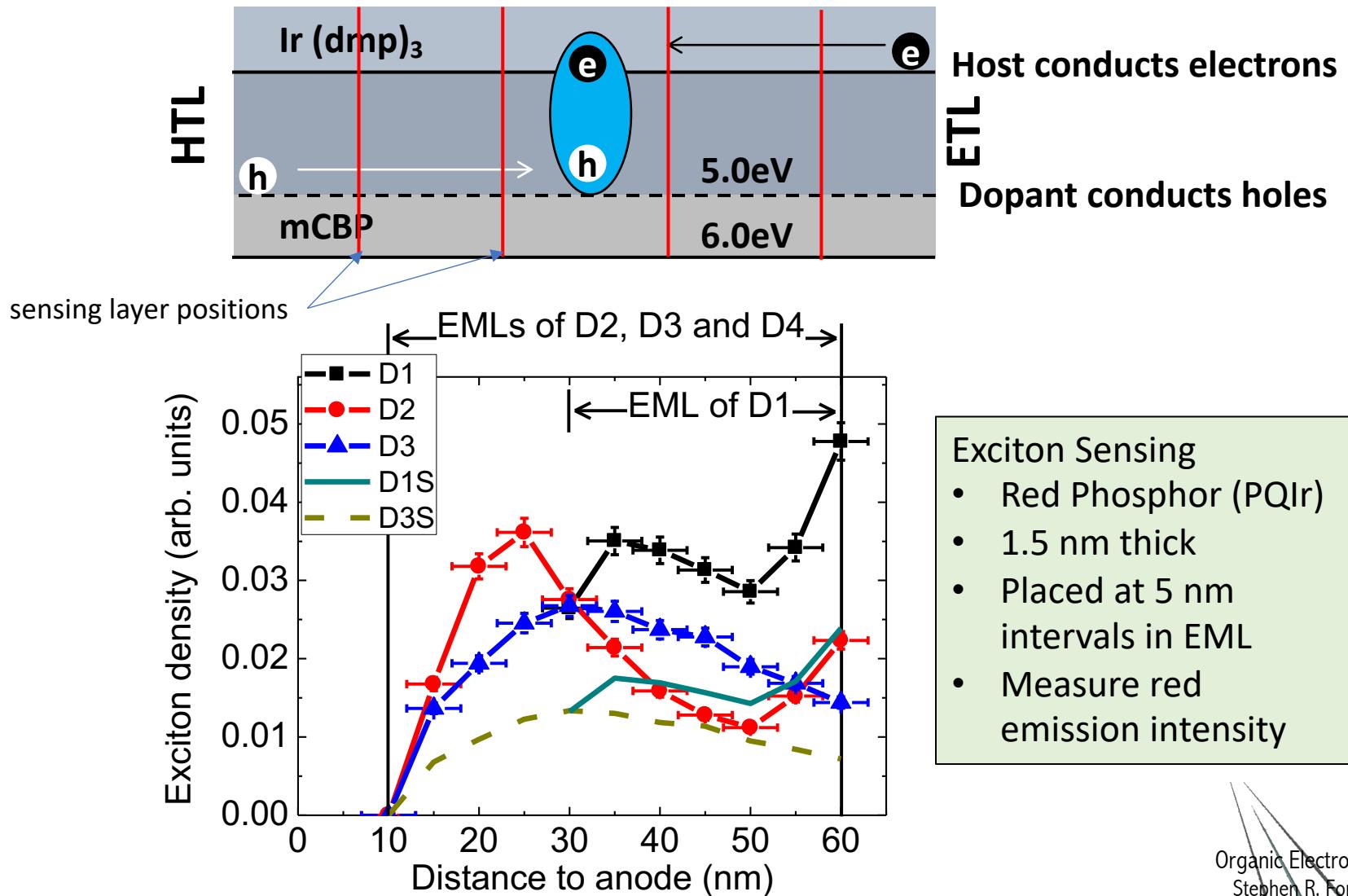


## 13 vol% uniform

**8 to 18% vol% graded**



# Excitons in the EML

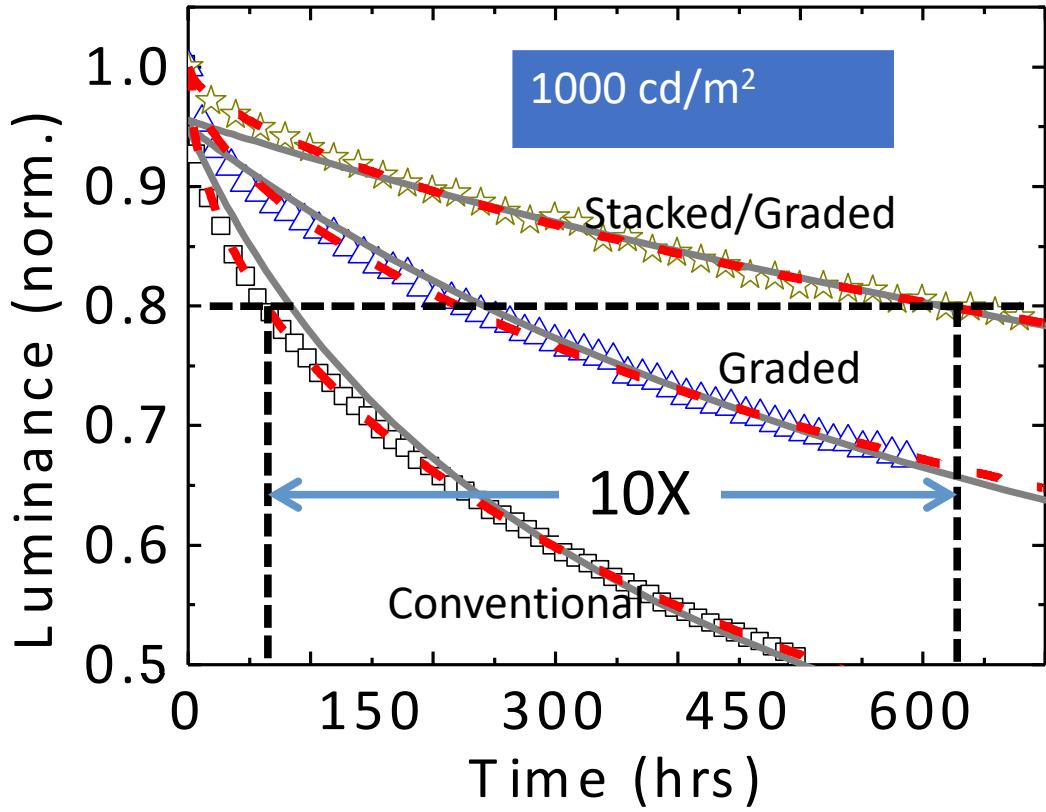


Host conducts electrons  
Dopant conducts holes

Exciton Sensing

- Red Phosphor (PQIr)
- 1.5 nm thick
- Placed at 5 nm intervals in EML
- Measure red emission intensity

# 10 X Lifetime Improvement Over Conventional



Stacking is essential!



Panel 15 cm x 15 cm 82% fill factor	2 Unit WSOLED
Luminance [cd/m <sup>2</sup> ]	3,000
Efficacy [lm/W]	48
CRI	86
Luminous Emittance [lm/m <sup>2</sup> ]	7,740
1931 CIE	(0.454, 0.426)
$LT_{70}$ [hrs]	13,000

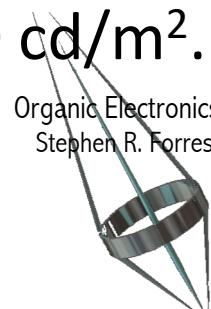
# Dopant Grading: Is it Good Enough?

using acceleration factors to predict lifetime

- Luminance to achieve sRGB color gamut for G is 10X that for B
- $\Rightarrow$  B sub-pixel  $L_0 = 100 \text{ cd/m}^2$  (c.f. G with  $L_0 > 1,000 \text{ cd/m}^2$ )
- $\Rightarrow$  B lifetime to T50 = 70,000 hr.
- Adopting Degradation acceleration factor:  $n = 1.55$  with

$$T50(100\text{cd/m}^2) = T50(1000\text{cd/m}^2) \times \left[ \frac{1000\text{cd/m}^2}{100\text{cd/m}^2} \right]^n$$

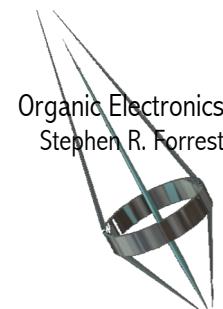
- $\Rightarrow$  B PHOLED lifetime to T50 =  $1.3 \times 10^5$  hr.
- Commercial G PHOLED lifetime =  $10^6$  hours at  $L_0 = 1000 \text{ cd/m}^2$ .



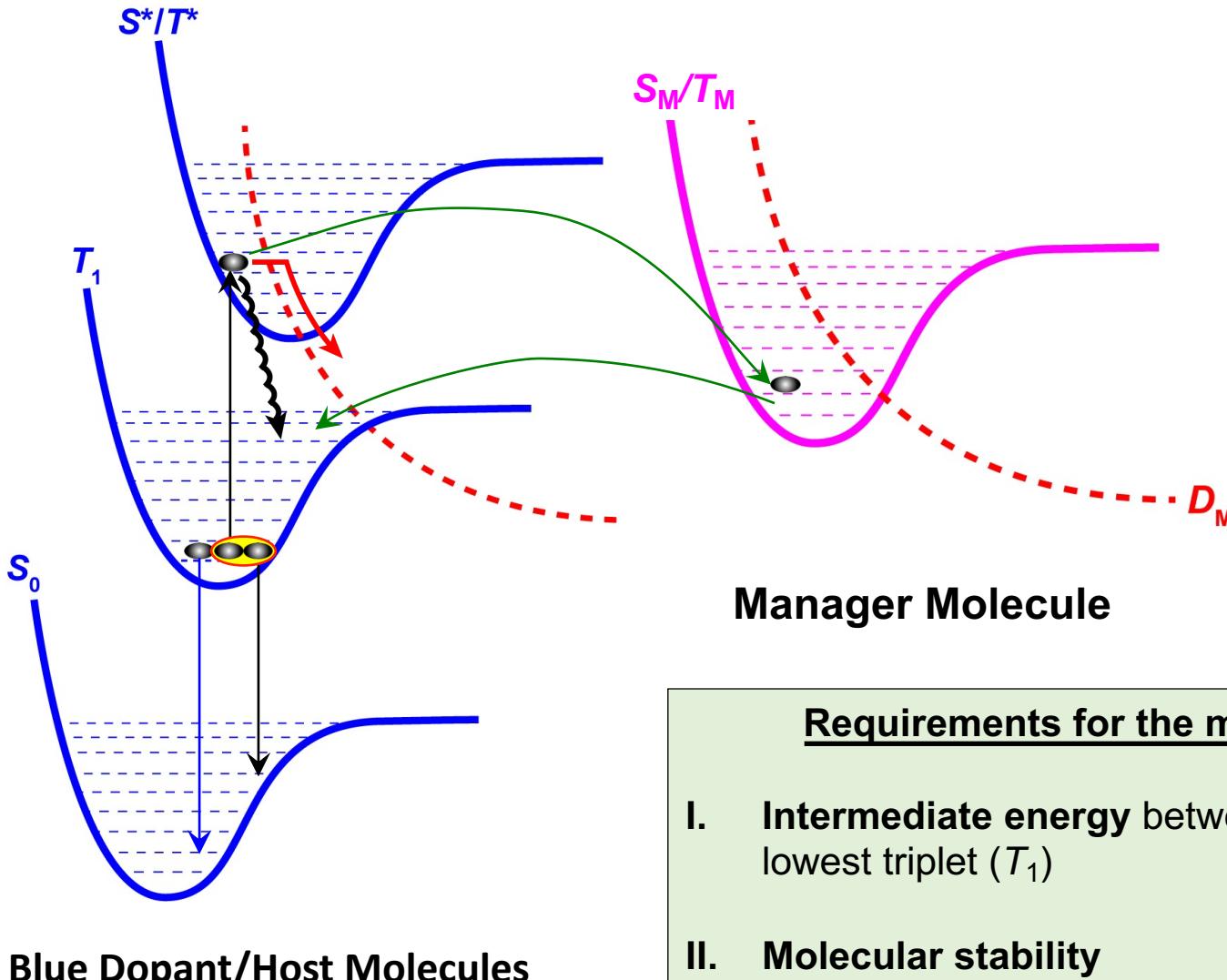
# Dopant Grading for Lighting: Is it OK?

- Current state of stacked WOLED:  $T_{70}=13,000$  hrs
- Mostly limited by blue lifetime
- Only light blue required
- Estimated increase in lifetime for stacked blue at lighting brightness:  $\sim 4X$
- Lifetime of blue lighting using grading: 50,000 hr

This is almost good enough



# Hot excited state management: Eliminating the highest energy excited states



Blue Dopant/Host Molecules

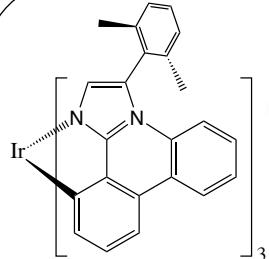
Manager Molecule

## Requirements for the manager molecule

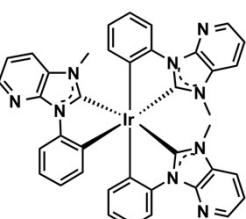
- I. **Intermediate energy** between hot state ( $T^*$ ) and lowest triplet ( $T_1$ )
- II. **Molecular stability**
- III. **Fast energy transfer** from dopant/host to manager

# Managed blue PHOLEDs

EML materials

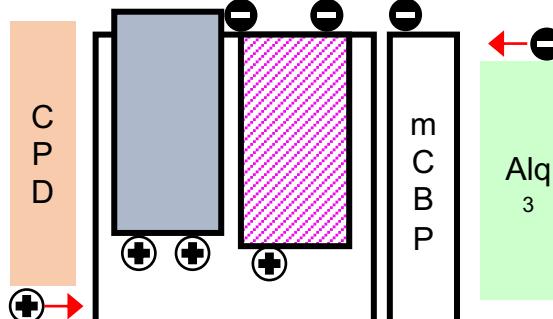


Blue dopant  
Ir(dmp)<sub>3</sub>  
 $E_T = 2.8 \text{ eV}$

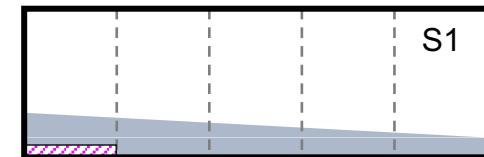


Manager  
mer-Ir(pmp)<sub>3</sub>  
 $E_T = 3.1 \text{ eV}$

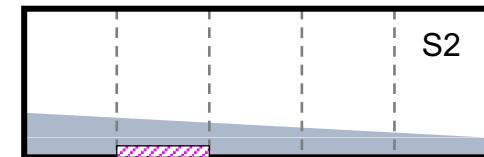
Energetics and charge transport



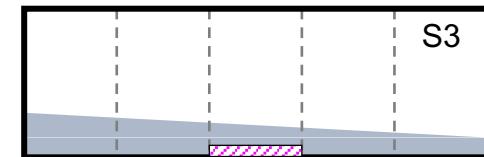
Managed EML (M1–M5)



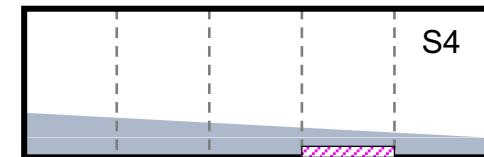
S1



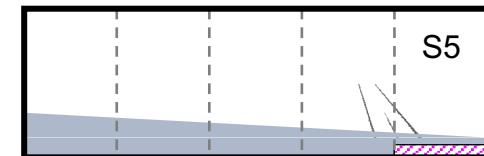
S2



S3

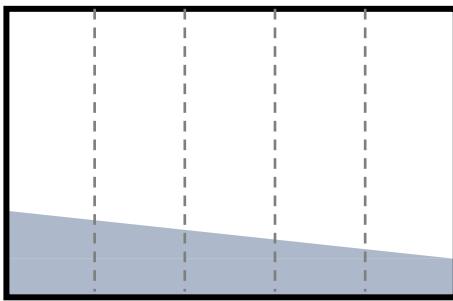


S4



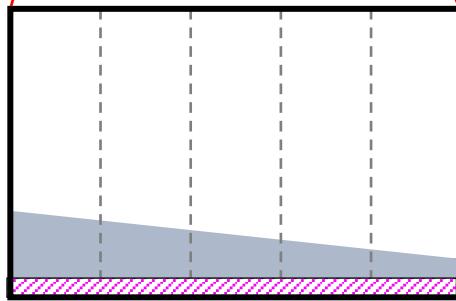
S5

Graded EML (GRAD)



■ : 18–8 vol%

Managed EML (S0)



■ : 15–5 vol%

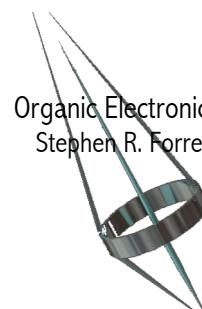
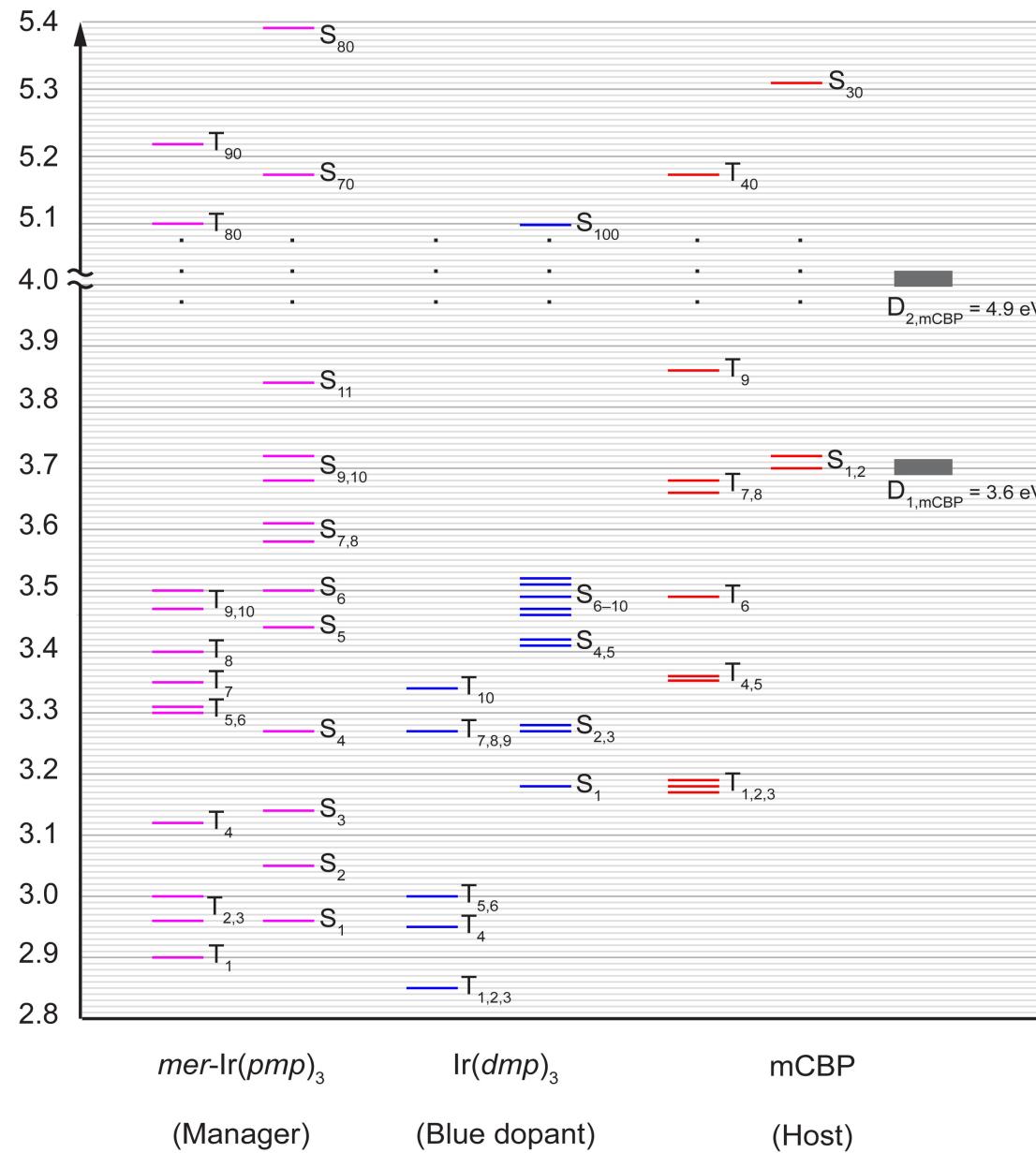
■ : 3 vol%

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Place manager at 10 nm  
sections of EML

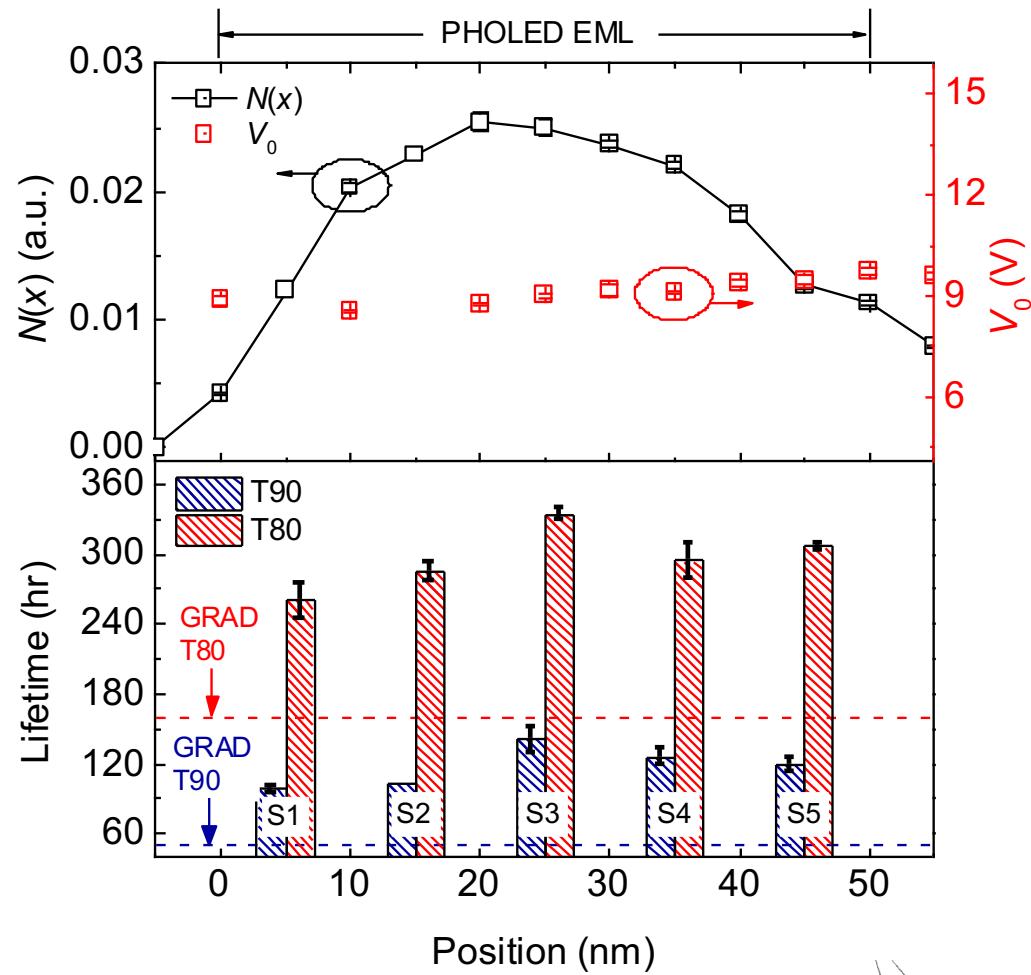
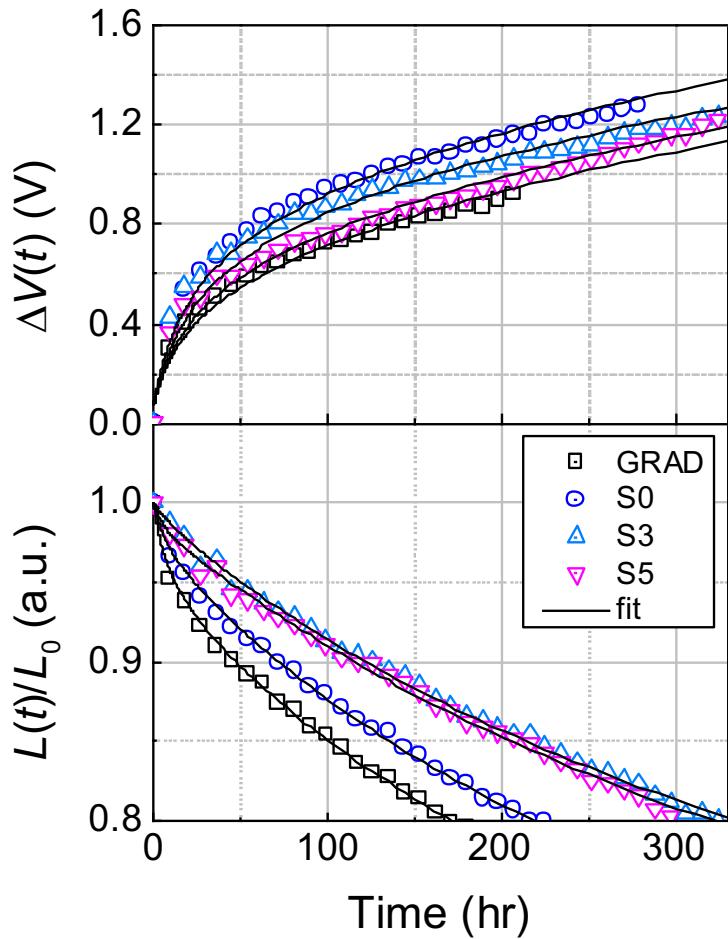
10 nm

# Plenty of Energy Levels to Access in the Management Process

Excited state energy (eV)

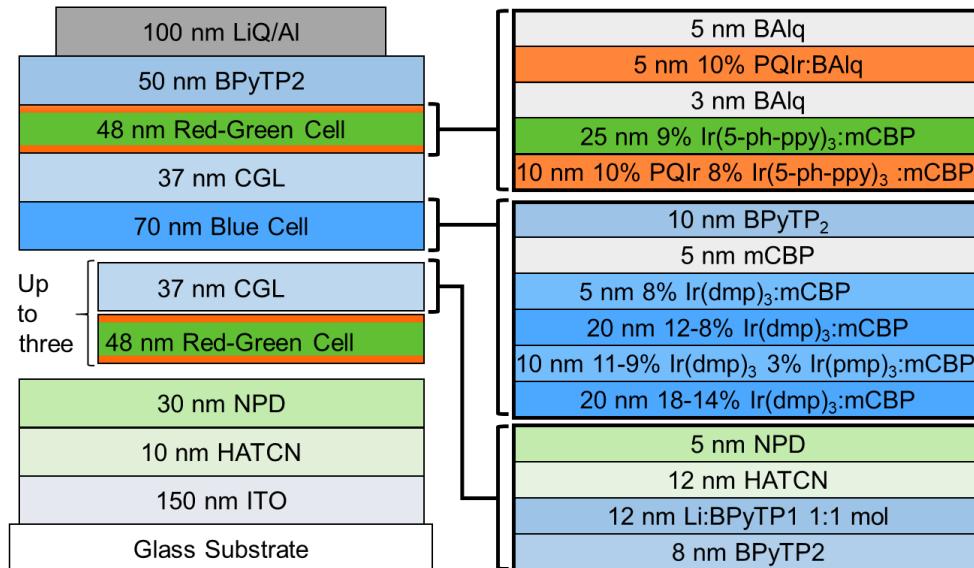


# Lifetime Improvements and TTA/TPA Model



- Greatest improvement in lifetime when manager at position of highest exciton density (S3)
- Fractional increase in lifetime decreases with time
  - Greater at T90 than T80  $\Rightarrow$  manager depletion

# Putting Grading Excited State Management to Work: Long lived all phosphor stacked WOLEDs

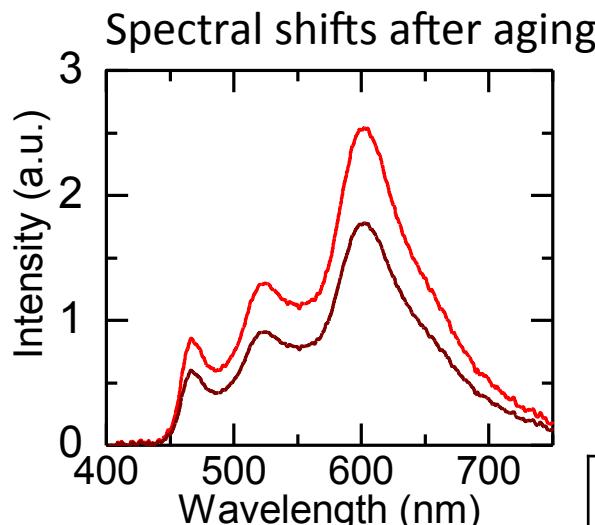
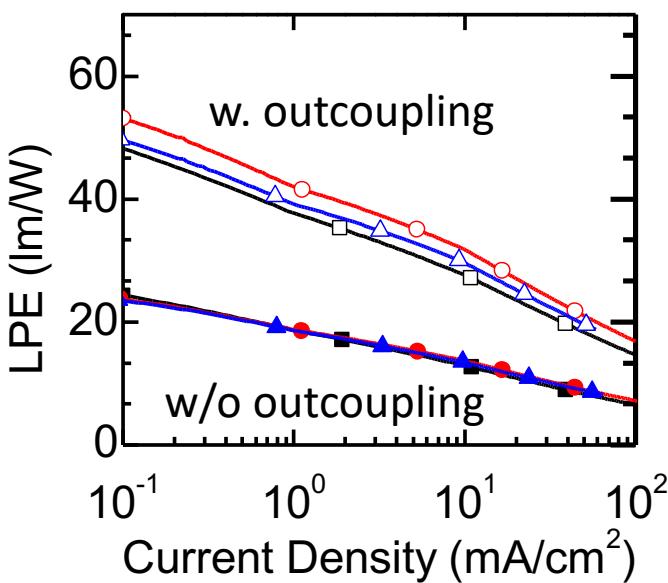


- Max Luminance > 200,000 nits
- 50 lm/W max
- CCT = 2780K
- CRI=89

Photo illustrating good color rendering of the SWOLEDs in this report. The luminaire comprises 36 pixels ( $2 \text{ mm}^2$ ) operated at 50-100k nits

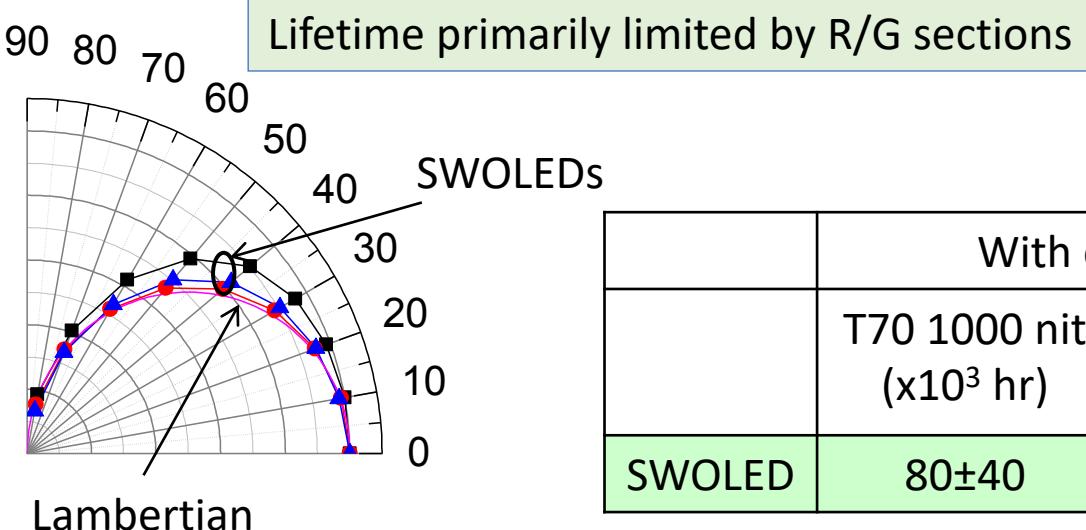
Electronics  
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# All Phosphor SWOLED Performance



T70	SWOLED
$\Delta\text{CCT}$	-360 K
$\Delta\text{CRI}$	-0.8
$\Delta\text{CIE}$	(0.03, 0)

SWOLED Architecture	Blue degradation @ WOLED T70:
Conv	T28
Grad-Managed	T48

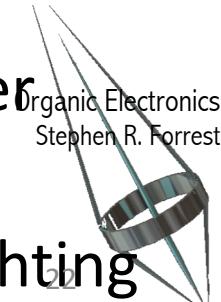


	With outcoupling		$\Delta V/V_0$ (T70) (%)
	T70 1000 nit ( $\times 10^3$ hr)	T70 3000 nit ( $\times 10^3$ hr)	
SWOLED	$80 \pm 40$	$14 \pm 5$	$\sim +10\%$

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# What we learned about OLEDs

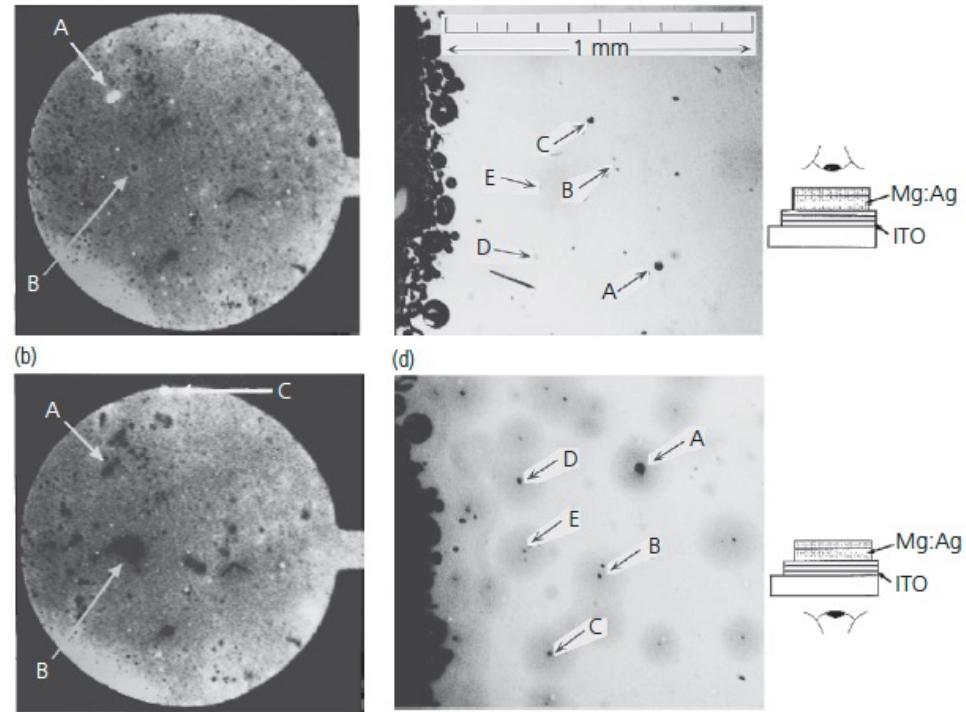
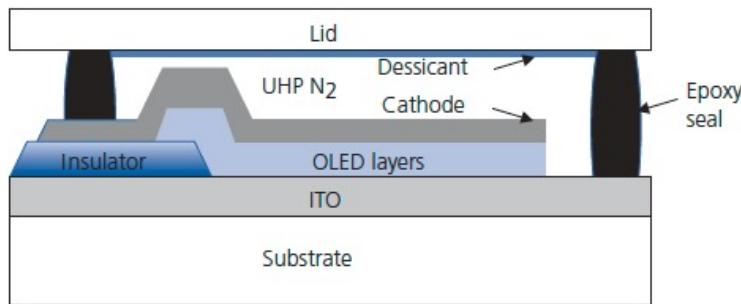
- Chromaticity and the perception of color is quantified based on eye response (photometric quantities)
- OLEDs reach highest efficiency when both singlets and triplets are harvested (heavy metal complexes and TADF molecules)
- Optimized OLEDs have many layers serving purposes ranging from charge conduction, contacting to electrodes, to light emission
- Outcoupling methods essential to view substrate and waveguide modes while limiting surface plasmons
- Degradation of OLEDs particularly severe for blue due to bimolecular annihilation
- Lighting requires broad spectral emission using multilayer devices or excimer emission
- OLEDs provide uniform, area lighting vs. specular LED lighting



# Packaging Matters

(see Ch. 5)

- Without packaging, there is rapid degradation of OLED luminance
- Dark spot defect formation of contacts more rapid in atmosphere than when packaged in inert (e.g. N<sub>2</sub>) gas.



- Dark spot formation on cathode as seen from the top and bottom sides
- Defect bright EL spots soon become dark
- Defects appear to be due to dust on substrate penetrating device active region

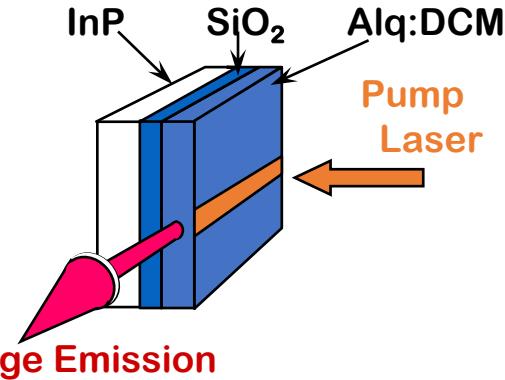
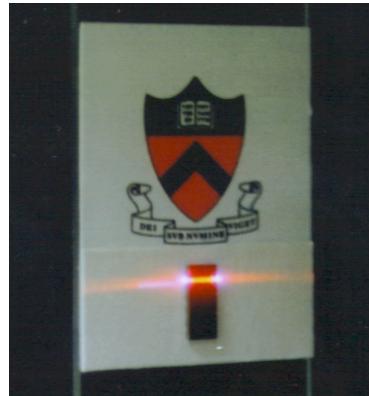
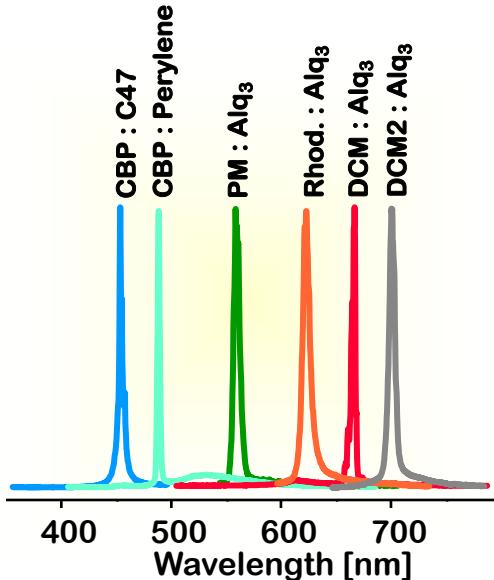
# Organic Lasers

- High intensity, monochromatic sources
- To date, only optical pumping has unambiguously shown lasing
- Large triplet and electrode losses, and low mobilities (hence low current) hamper electrical pumping

## How to identify a laser

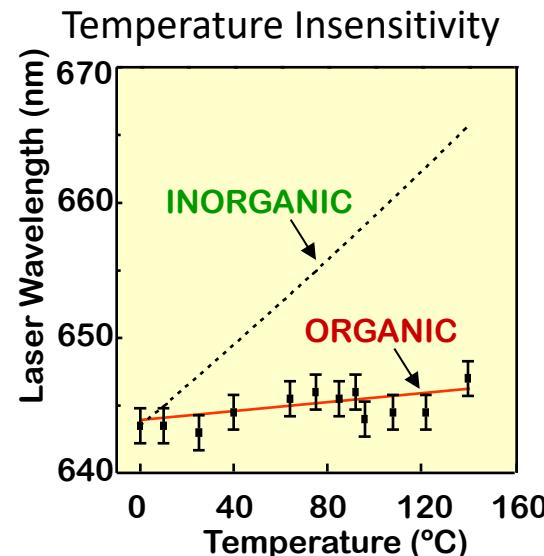
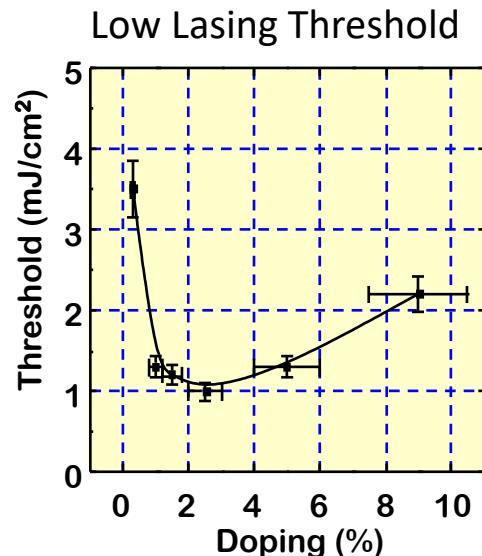
- A clear threshold between spontaneous and stimulated emission evident from an abrupt increase in output slope efficiency
- A significant narrowing of the spectral linewidth at threshold. Single mode lasers show linewidths  $\sim 1\text{\AA}$ , multimode lasers will have multiple emission lines coincident with the gain (PL) spectrum of the material, and whose separation is  $\Delta\lambda \sim 1/L$ , the cavity length
- A well-defined output beam
- Temporal and spatial coherence of the output beam

# Features of Organic Lasers



Kozlov, et al., *Nature* **389**, 362 (1997).

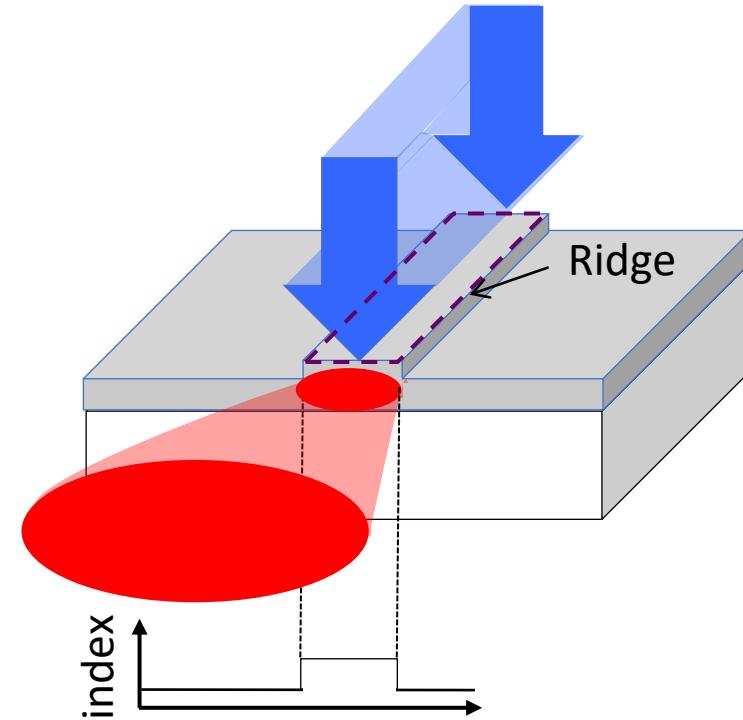
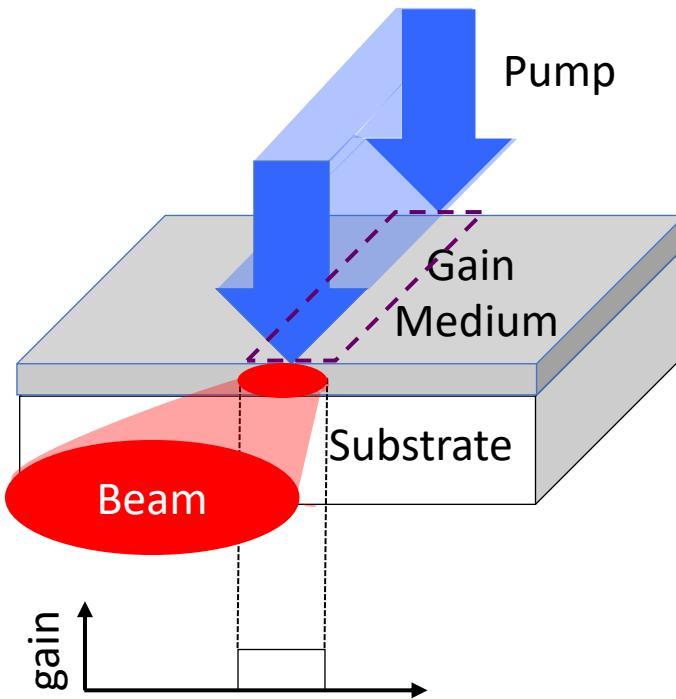
- Material Tunability
- Freedom from Epitaxial Limitations
- Natural Quantum Dots



Kozlov, et al., *Appl. Phys. Lett.* **71**, 2575 (1997).



# Gain vs. Index Guided Lasers



## Gain guiding

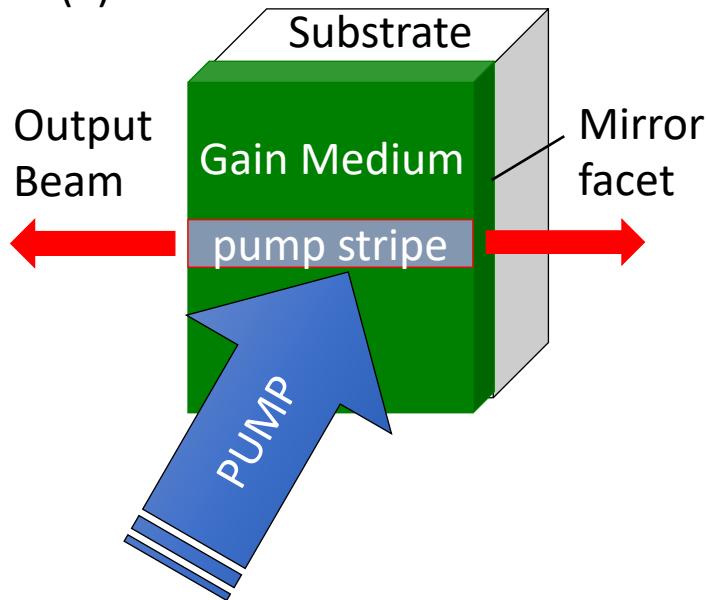
- Simple
- Can lead to high thresholds
- Can lead to modal instabilities

## Index guiding

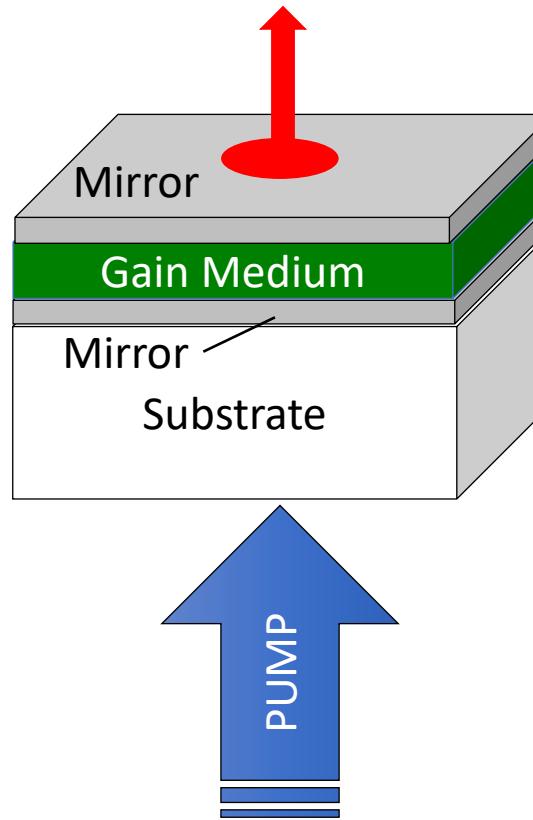
- More complex
- Can reduce thresholds
- Has modal instabilities only at very high power

# Optically pumped lasers

(a)



(b)



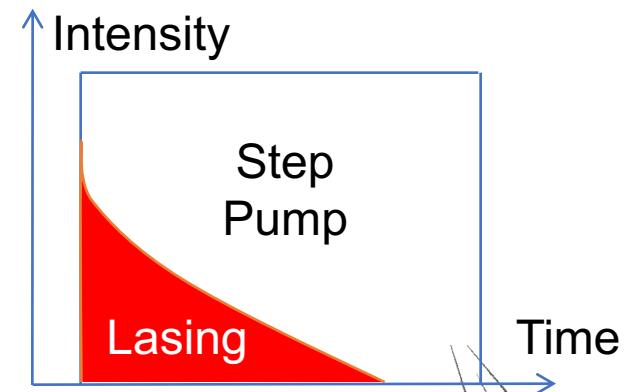
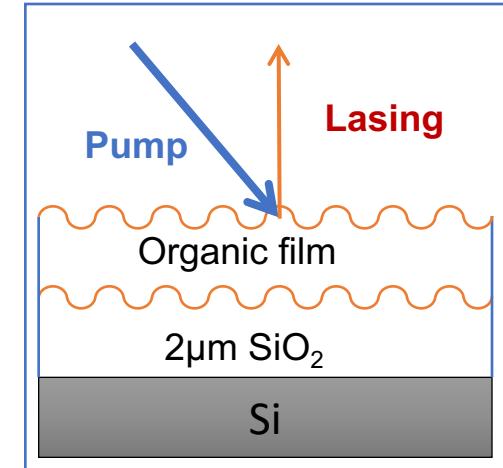
Longitudinal Configuration

Vertical Cavity Surface Emitting Laser  
(VCSEL)

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# Pulsed Organic Lasers

- Why does organic lasing last only <100ns?
  - Initially (<10ns)
    - Negligible T
    - Gain=Loss
  - Later (>100ns)
    - T builds up
    - Gain ↓ : S-T quenching
    - Loss ↑ : T absorption
  - Same source of loss prevents electrically pumped laser action

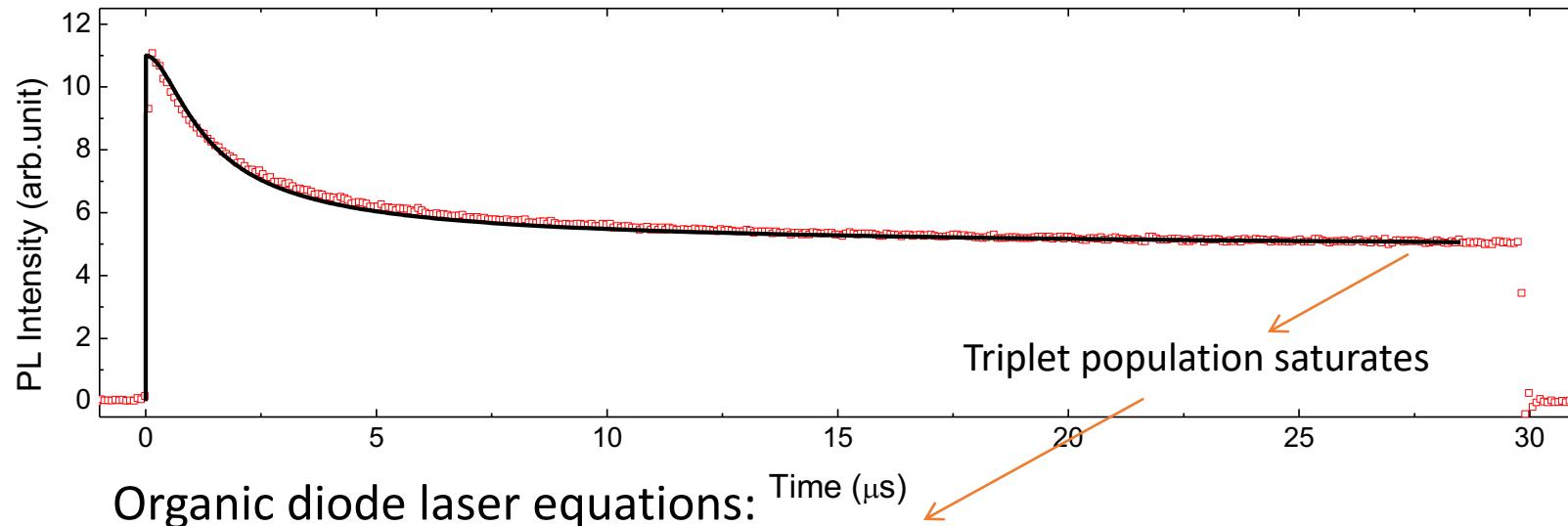


Organic Electronics  
Stephen R. Forrest

Giebink & Forrest, *Phys. Rev. B* **2009**, 79, 073302

Lehnhardt, et al. *Phys. Rev. B* **2010**, 81, 165206

# Triplet saturation and CW threshold



Organic diode laser equations:

Threshold:  $g_{net}(t) = \Gamma \sigma_{stim} S(t) - \alpha_{cav} - \Gamma \sigma_{TT} T_G(t) \geq 0$  (gain condition)

Pulsed threshold

$$I_{PS} = e_p d (k_S + k_{ISC}) \frac{\alpha_{CAV}}{\eta \Gamma \sigma_{stim}}$$

CW threshold

$$I_{CW} = e_p d (k_S + k_{ISC} + k_{ST} T_\infty) \frac{\alpha_{CAV} + \Gamma \sigma_{TT} T_\infty}{\eta \Gamma \sigma_{stim}}$$

S-T

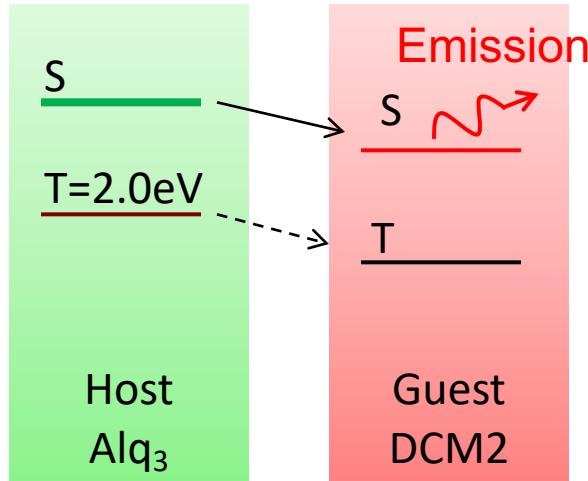
T abs

	$I_{PS}$ $(\text{kW/cm}^2)$	$T_\infty$ $(10^{18}\text{cm}^{-3})$	$I_{CW}$ $(\text{kW/cm}^2)$
DCM2:Alq <sub>3</sub>	0.93	5.0	32

Above damage  
threshold

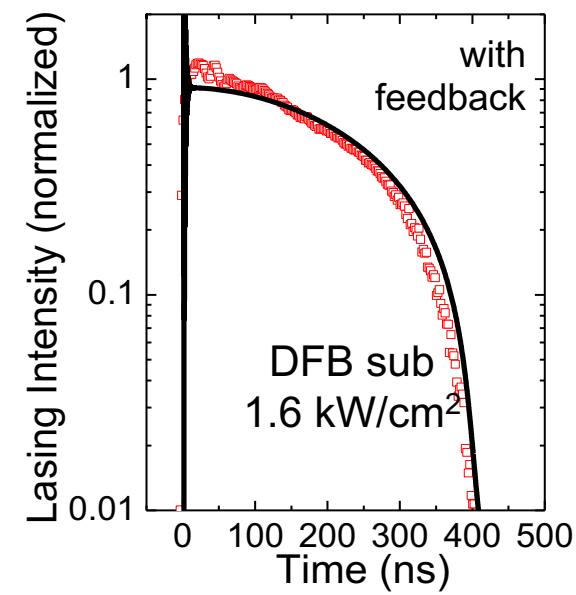
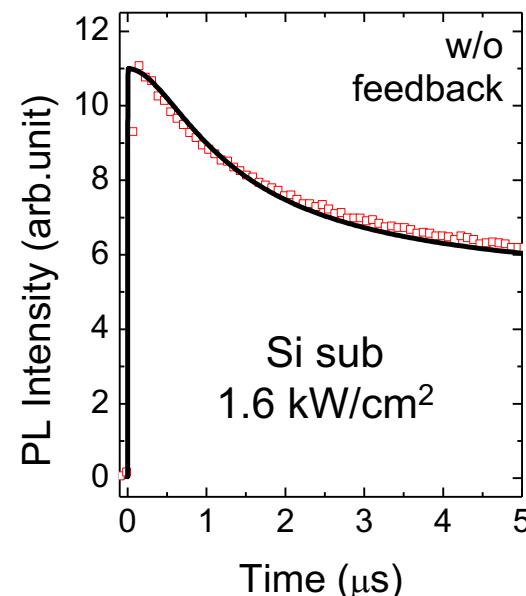
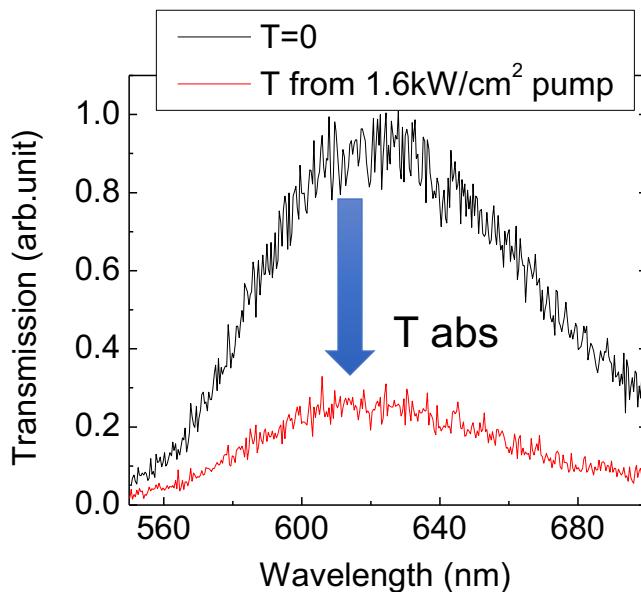
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# Example Laser -- DCM2:Alq<sub>3</sub>



- PL fit follows S-T quenching  
 $\square \text{S}^* + \text{T}^* \rightarrow \text{S}_0 + \text{T}^{**}$
- Lasing fit follows S-T and T absorption  
 $\square \text{P} + \text{T}^* \rightarrow \text{T}^{**}$
- Lasing condition

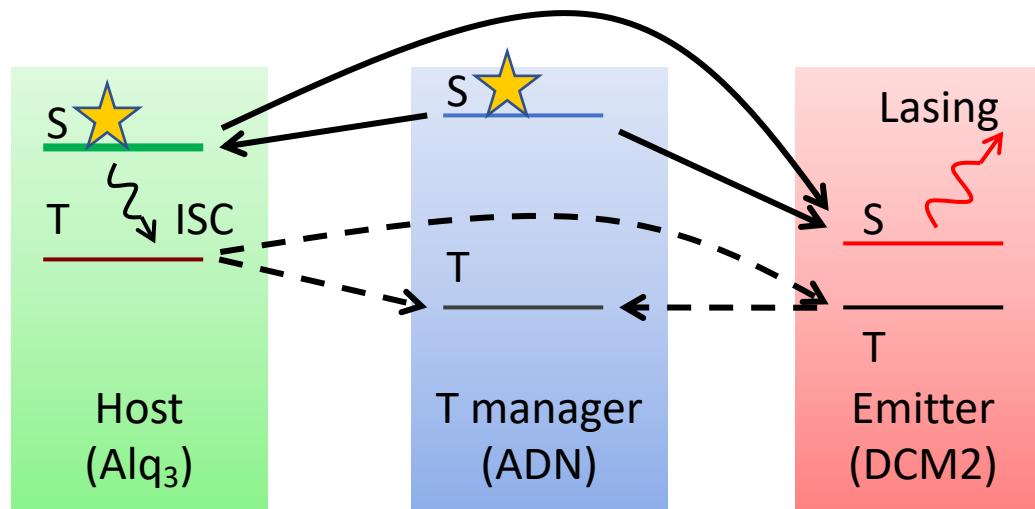
$$g_{net}(t) = \Gamma \sigma_{stim} S(t) - \alpha_{cav} - \Gamma \sigma_{TT} T_G(t) \geq 0$$



Lasing turns off once triplet loss exceeds gain

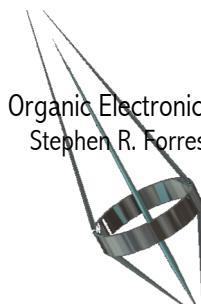
# The Triplet Management Concept

- Introduce a 3rd molecule into the gain region that quickly removes triplets  $\Rightarrow$  reduced absorption loss
- Same concept works in fluorescent OLEDs to reduce rolloff due to accumulation of triplets at high intensity



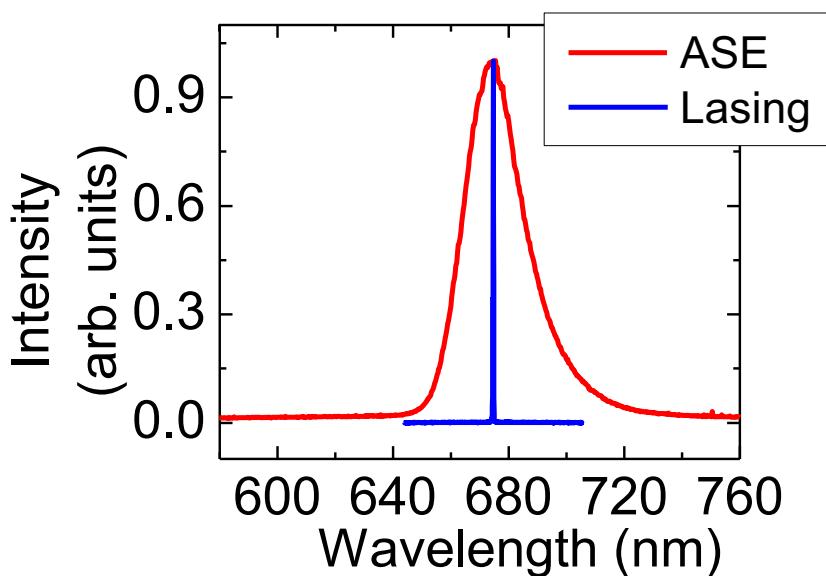
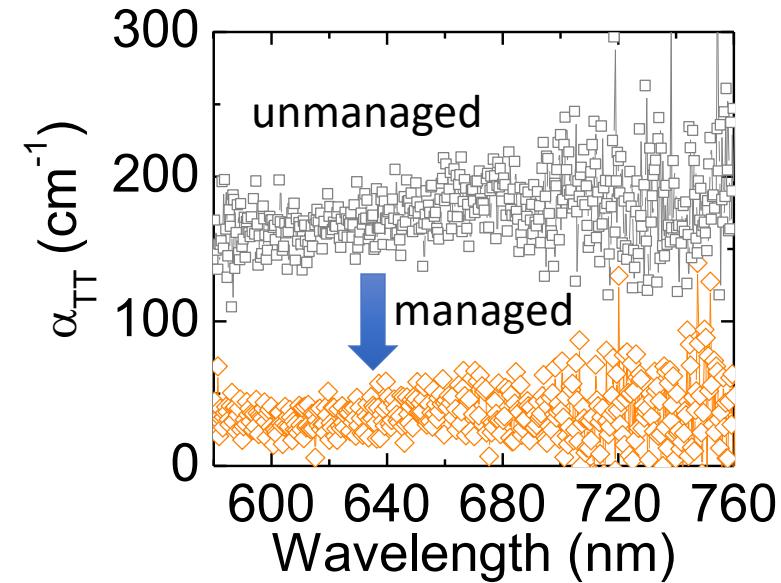
→ Förster transfer    → Dexter transfer

Example gain region composition

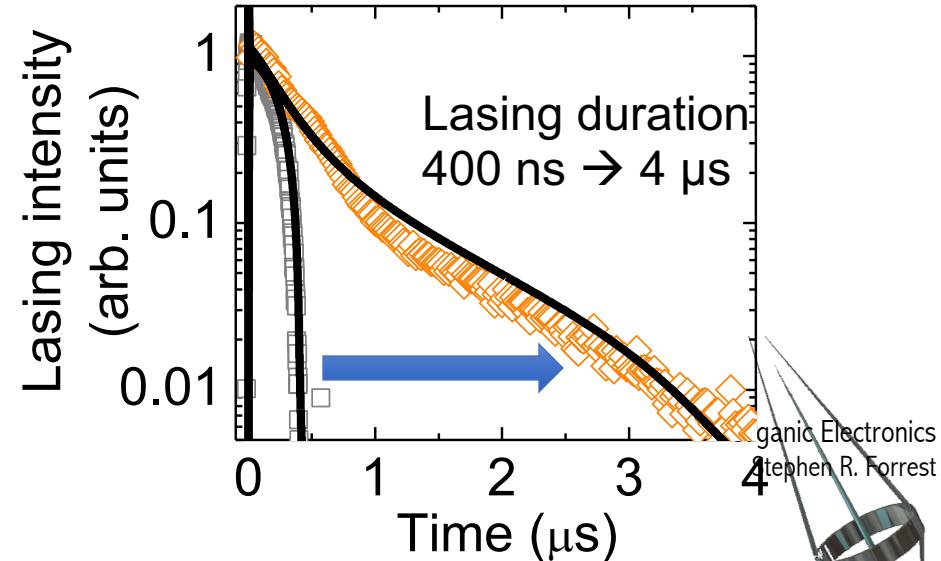
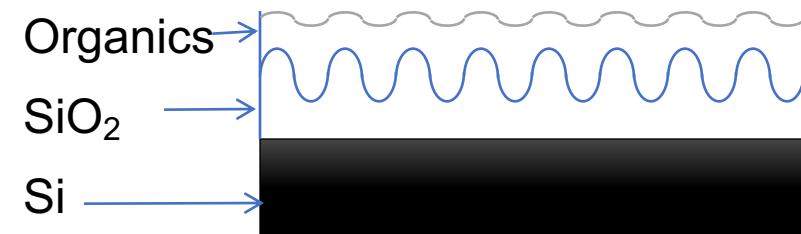


# T absorption and lasing with T management

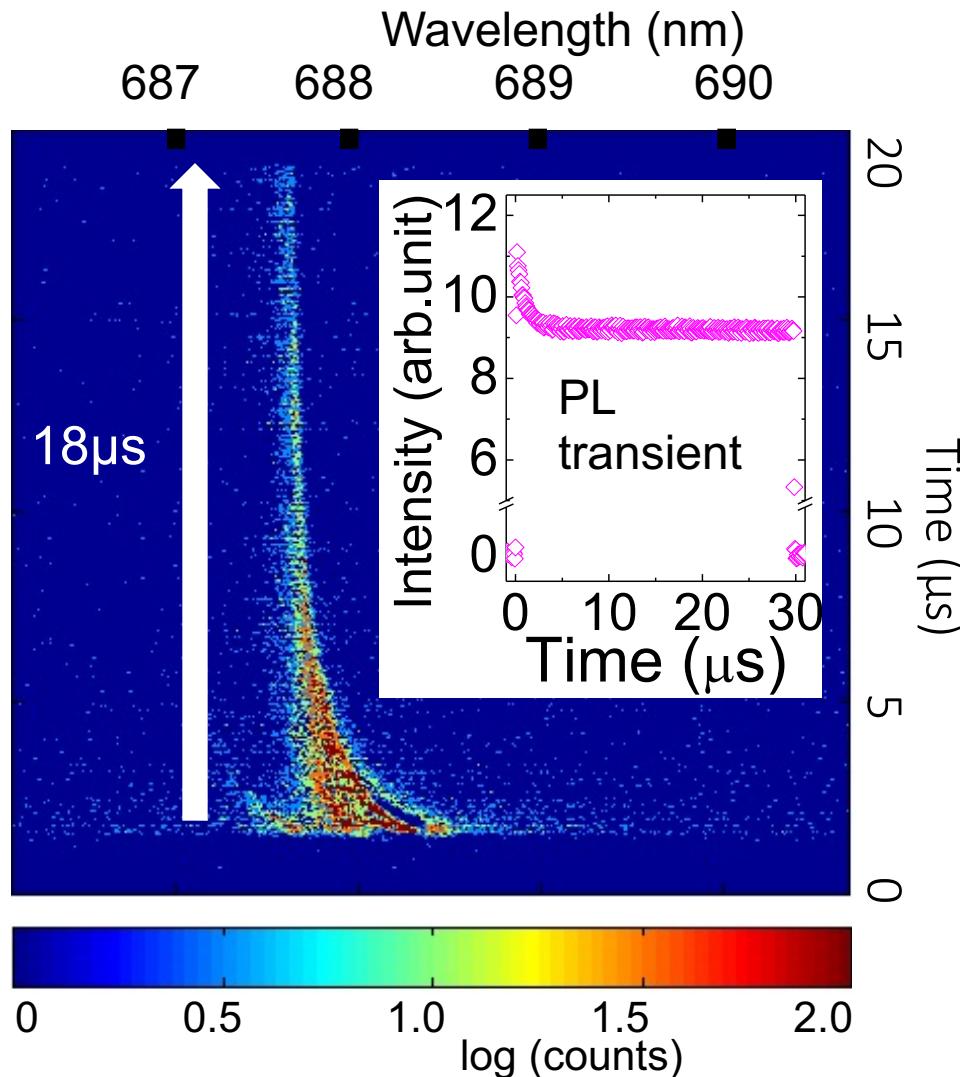
Alq<sub>3</sub>:DCM2



Organic laser with 2<sup>nd</sup> order distributed feedback structure



# Management Allows CW Lasing Above the CW threshold



- Condition
  - $2.4\text{kW/cm}^2$ , 10Hz/18 $\mu$ s
  - Integrate 1000 pulses
  - Degradation limited
- **Single pulse  $\rightarrow 100 \mu\text{s}$**
- Wavelength shift due to triplet induced index change
- $m\lambda=2n_{\text{eff}}\Lambda$
- CW lasing not limited by T

# What we have learned

- OLEDs are the leading application of organic electronics due to their features of:
  - Color versatility due to chemical modification
  - 100% internal efficiency in PHOLEDs and TADF molecules
  - Stability over long term operation (except in the blue)
  - Thin film, flexible form factors allowing for their use in mobile applications
  - Very attractive lighting colors and luminance characteristics
  - But...optical outcoupling losses and long lived PHOLEDs and TADF devices are remaining challenges to be solved.
- Organic lasers are excited primarily via optical pumping with features of:
  - Wavelength agility
  - Extraordinary temperature stability of its threshold and spectral properties
  - But...electrical pumped lasing remains a challenge due to large triplet and contact losses.

