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# <sup>13</sup>C of Carbon Dioxide (CO<sub>2</sub>) by GasBench-IRMS

## Methodology

Analysis of atmospheric  $CO_2$  is performed in 12 mL Labco Exetainer vials on a Thermo Scientific GasBench II coupled to a Thermo Finnigan Delta Plus XL isotope-ratio mass spectrometer <sup>[1]</sup>. An autosampler with a 2-hole sampling needle uses a helium carrier stream (2.0 mL/min) to push  $CO_2$  out of the vial and into a 100  $\mu$ L sample loop. Once the loop is adequately flushed, pulses of sample gas are injected into the IRMS for isotopic measurement. Prior to entering the IRMS, water is removed from the gas stream using a Nafion dryer, and sample  $CO_2$  is chromatographically separated from other gases (e.g.,  $N_2O$ ) using an Agilent PoraPLOT Q GC column (25 m x 0.32 mm ID x 10  $\mu$ m film thickness, 50 °C, 2 mL/min). Samples with  $CO_2$  concentrations above 5 % (i.e., 50,000 ppm) can be analyzed after dilution with helium in a second Exetainer vial.

Sample replicates are the responsibility of the client and must be paid for as individual samples. Replicates of the quality control and assurance reference materials are measured every ten samples.

# **Calibration and Reporting of Stable Isotope Ratios**

Calibration procedures for CO<sub>2</sub> are applied identically across reference and sample materials and are directly traceable to the primary isotopic reference material (i.e., VPDB).

First, a pure CO<sub>2</sub> reference gas is used to calculate provisional isotopic values of the sample peaks. Next, isotopic values are adjusted for changes in linearity and instrumental drift using secondary reference materials (i.e., UCDC1, and a dilution series from UCDC2). Finally, measurements are scale-normalized to the primary reference material using three pure CO<sub>2</sub> secondary reference materials (i.e., OZ-3, OZ-10, and OZ-40) that have been calibrated against certified standard reference materials (i.e., NBS 18, NBS 19, and LSVEC) available from NIST and the IAEA.

Final quality assessment is based on the accuracy and precision of unbiased quality control materials,  $\delta^{13}$ C-calibrated CO<sub>2</sub> gases (i.e., UCDC3 and UCDC4).

Quality assurance reference materials: OZ-3, OZ-10, OZ-40, UCDC1, UCDC2 Quality control reference materials: UCDC3, UCDC4

### **Measurement Uncertainty**

Sample materials are inherently variable in composition, and measurement error may vary between different sample types due to differences in composition. Mean measurement error  $(\sigma)$  and accuracy, as determined by replicate measurements of the quality control and assurance materials, must fall below expected measurement error ( $\pm$  0.10 % for  $\delta^{13}$ C). Accuracy and precision of the co-measured calibrated quality control and assurance materials are provided with data reports. Limit of quantification (LOQ), based on total peak area, is 3 V-s for  $\delta^{13}$ C.

#### **Revision Date**

September 6, 2019

## **Approved By**

Richard Doucett

Richard Dovett

# References

[1] Tu KP, Brooks PD, Dawson TE. 2001. Using septum-capped vials with continuous-flow isotope ratio mass spectrometric analysis of atmospheric CO<sub>2</sub> for Keeling plot applications. Rapid Communications in Mass Spectrometry 15: 952-956. doi.org/10.1002/rcm.320

# **Glossary**

°C degree Celsius

% percent

per mil; 1 ‰ is equivalent to 0.001 or 1 mUr
 stable isotope of carbon; mass number of 13

δ delta notation for isotopic composition; in per mil ("%") or mUr; 1 % equals 1 mUr

μL microliter

μm micron or micrometer
 σ standard deviation
 CO<sub>2</sub> carbon dioxide
 GC gas chromatograph

IAEA International Atomic Energy Agency

ID inner diameter

IRMS isotope-ratio mass spectrometry

L liter

LOQ limit of quantification; minimum signal for analyte to meet required signal-to-noise ratio

LSVEC lithium carbonate prepared in 1973 by H. Svec, Iowa State University

m meter
mg milligram
min minute
mL milliliter
mm millimeter

mUr milliurey; 1 mUr is equivalent to 0.001 or 1 ‰

N<sub>2</sub>O nitrous oxide

NBS National Bureau of Standards; renamed NIST in 1998

NBS 18 calcite prepared by I. Friedman, J. R. O'Neil, and G. Cebula, USGS NBS 19 carbonatite prepared by I. Friedman, J. R. O'Neil and G. Cebula, USGS

NIST
Oztech
OZ-3
OZ-10
OZ-40
National Institute of Standards and Technology
Oztech Trading Corporation, Safford, Arizona
pure CO<sub>2</sub> gas reference available from Oztech
pure CO<sub>2</sub> gas reference available from Oztech
pure CO<sub>2</sub> gas reference available from Oztech

ppm parts per million; equivalent to mg/L

QA quality assurance; overall laboratory measures to ensure measurement quality QC quality control; activities and procedures used to evaluate quality requirements

UCDC1 gas reference comprising 3 000 ppm CO<sub>2</sub> with balance helium gas reference comprising 50 000 ppm CO<sub>2</sub> with balance helium UCDC3 gas reference comprising 10 000 ppm CO<sub>2</sub> with balance helium UCDC4 gas reference comprising 400 ppm CO<sub>2</sub> with balance helium

VPDB Vienna PeeDee Belemnite; primary reference for measurements of carbon isotopes

V-s volt-second

#### **General Resources**

Berglund M, Wieser ME. 2011. Isotopic compositions of the elements 2009 (IUPAC Technical Report). *Pure Appl Chem.* 83(2):397–410. doi: 10.1351/PAC-REP-10-06-02.

Brand WA, Coplen TB, Vogl J, Rosner M, Prohaska T. 2014. Assessment of international reference materials for isotope-ratio analysis (IUPAC Technical Report). *Pure Appl Chem.* 86(3):425–467. doi: 10.1515/pac-2013-1023.

Commission on Isotopic Abundances and Atomic Weights. c2007–2008. CIAAW/IUPAC. http://www.ciaaw.org/index.htm.

Coplen TB, Hopple JA, Böhlke JK, Peiser HS, Rieder SE, Krouse HR, Rosman KJR, Ding T, Vocke RD Jr, Revesz KM, et al. 2002. Compilation of minimum and maximum isotope ratios of selected elements in naturally occurring terrestrial materials and reagents. *US Geological Survey Water-Resources Investigations Report 2001-4222*. doi: 10.3133/wri014222.

Coplen TB. 2011. Guidelines and recommended terms for expression of stable-isotope-ratio and gas-ratio measurement results. *Rapid Commun Mass Spectrom*. 25(17):2538–2560. doi: 10.1002/rcm.5129.

de Groot PA, editor. 2004. *Handbook of Stable Isotope Analytical Techniques*. Vol. 1. Amsterdam (NL): Elsevier.

de Groot PA, editor. 2008. *Handbook of Stable Isotope Analytical Techniques*. Vol. 2. Amsterdam (NL): Elsevier.

Dunn PJH, Carter JF, editors. 2018. *Good Practice Guide for Isotope Ratio Mass Spectrometry*. 2<sup>nd</sup> ed. Bristol (GB): FIRMS. Available from: http://www.forensic-isotopes.org/gpg.html.

Fry B. 2006. Stable Isotope Ecology. New York (NY): Springer. doi: 10.1007/0-387-33745-8.

Meier-Augenstein W. 2017. Stable Isotope Forensics: Methods and Forensic Applications of Stable Isotope Analysis. 2<sup>nd</sup> ed. Hoboken (NJ): Wiley. doi: 10.1002/9781119080190.

Mook W, de Vries JJ, editors. 2001. Environmental Isotopes in the Hydrological Cycle: Principles and Applications. *IHP-V Technical Documents in Hydrology No. 39*. Paris (FR): UNESCO/IAEA; 6 vol. Available from: http://www-naweb.iaea.org/napc/ih/IHS\_publication.html.

National Institute of Standards and Technology. Gaithersburg (MD): US Department of Commerce. https://www.nist.gov/.

Reference Products for Environment and Trade. c2009. Vienna (AT): International Atomic Energy Agency (IAEA). https://nucleus.iaea.org/rpst/index.htm.

Sharp Z. 2017. *Principles of Stable Isotope Geochemistry*. 2nd ed. Available from: https://digitalrepository.unm.edu/unm\_oer/1/.