



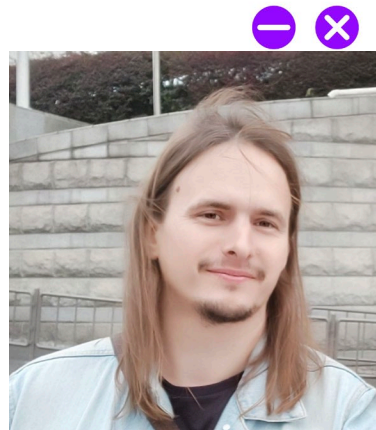
iTMO

Physical Nature of Light and Color

Image Processing

About me

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- PhD in Engineering
- 3 years of teaching experience
- More than 20 years of IT research and development
- Associate Professor and Leading Researcher at ITMO University
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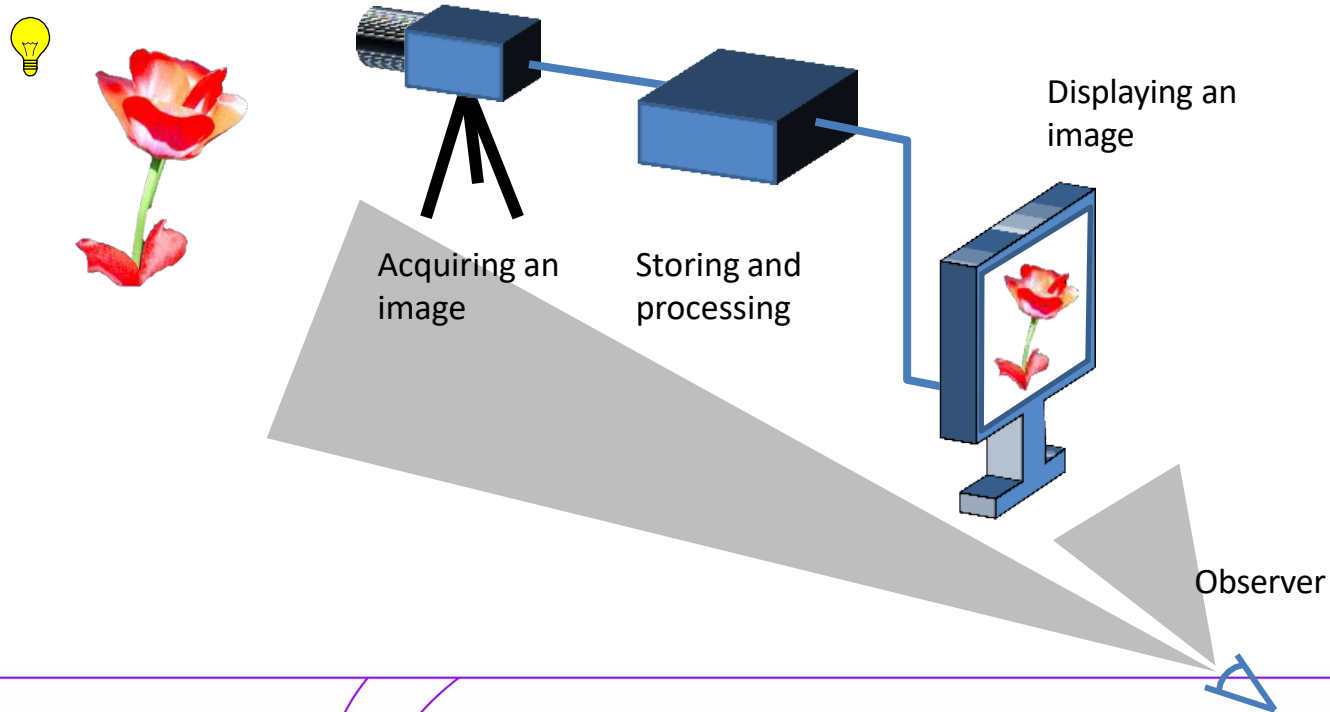


Outline

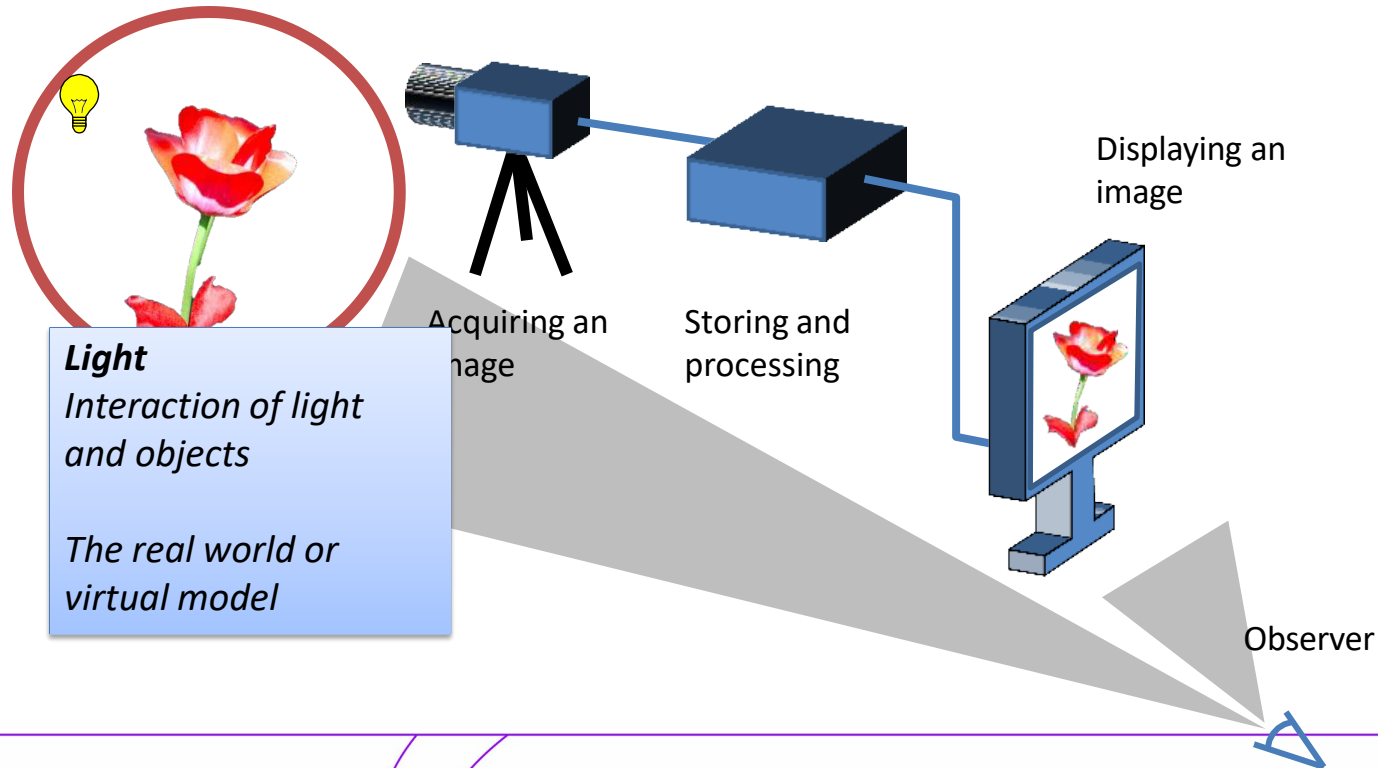
- Light and wave nature of light
- Radiometry: basic terms
- Photometry: basic terms



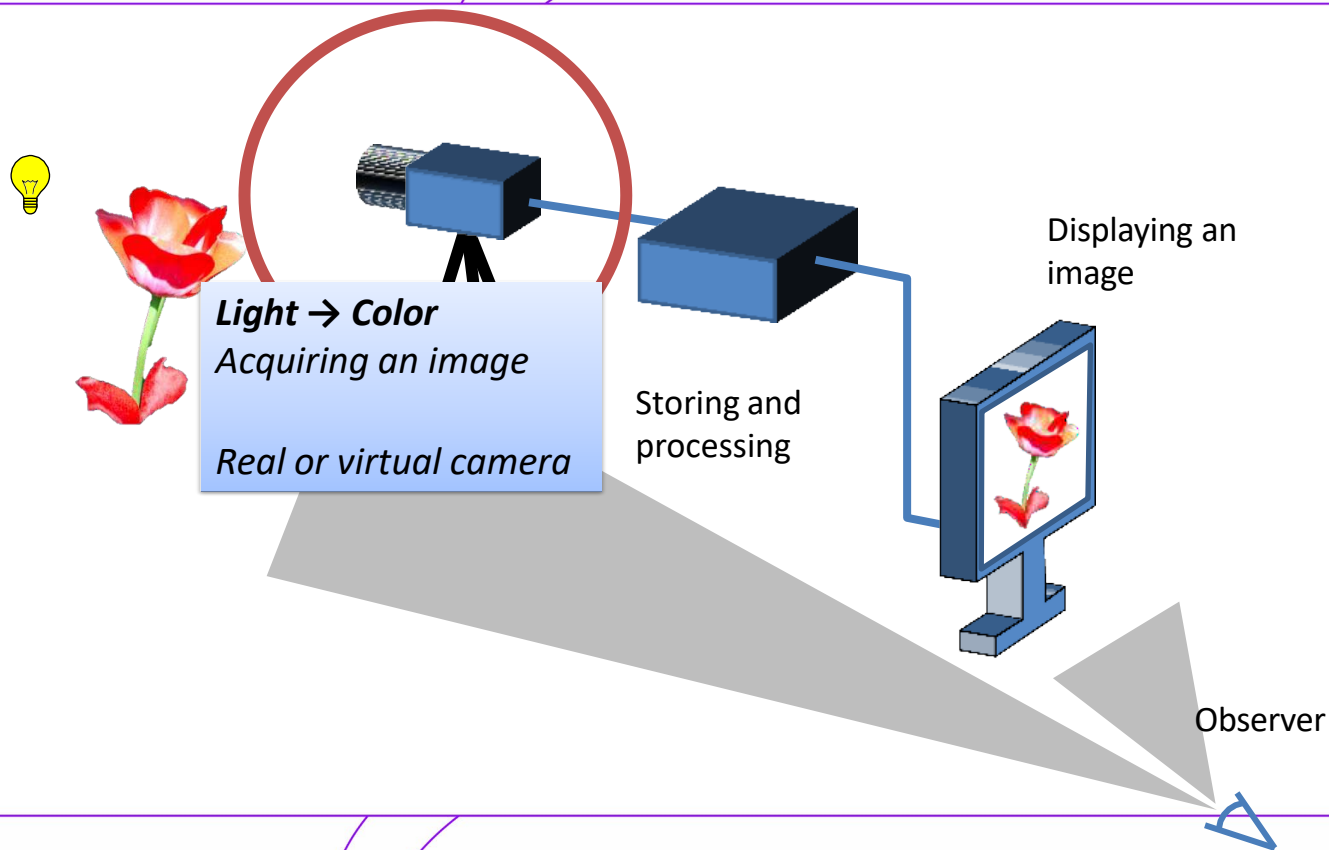
Light and color in a graphical system



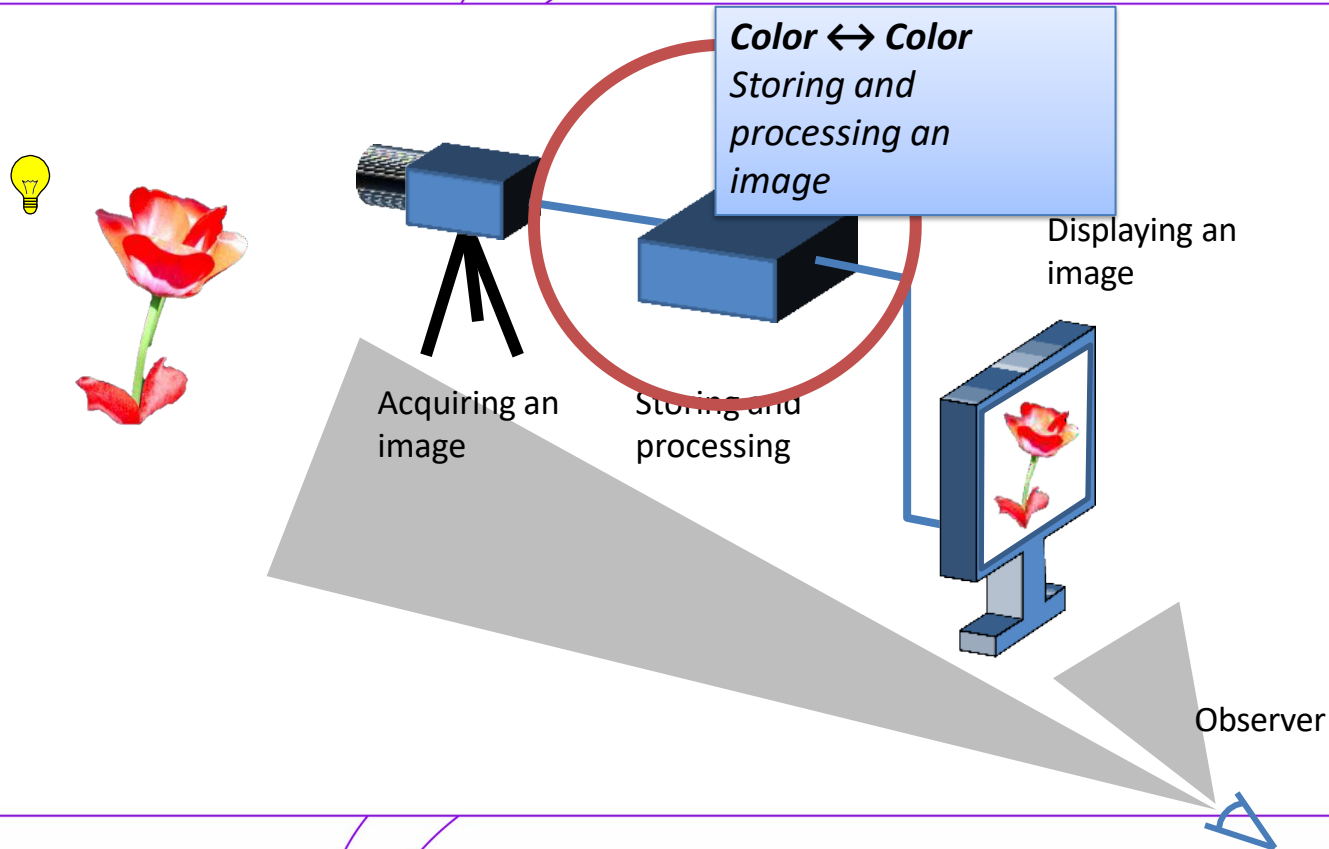
Light and color in a graphical system



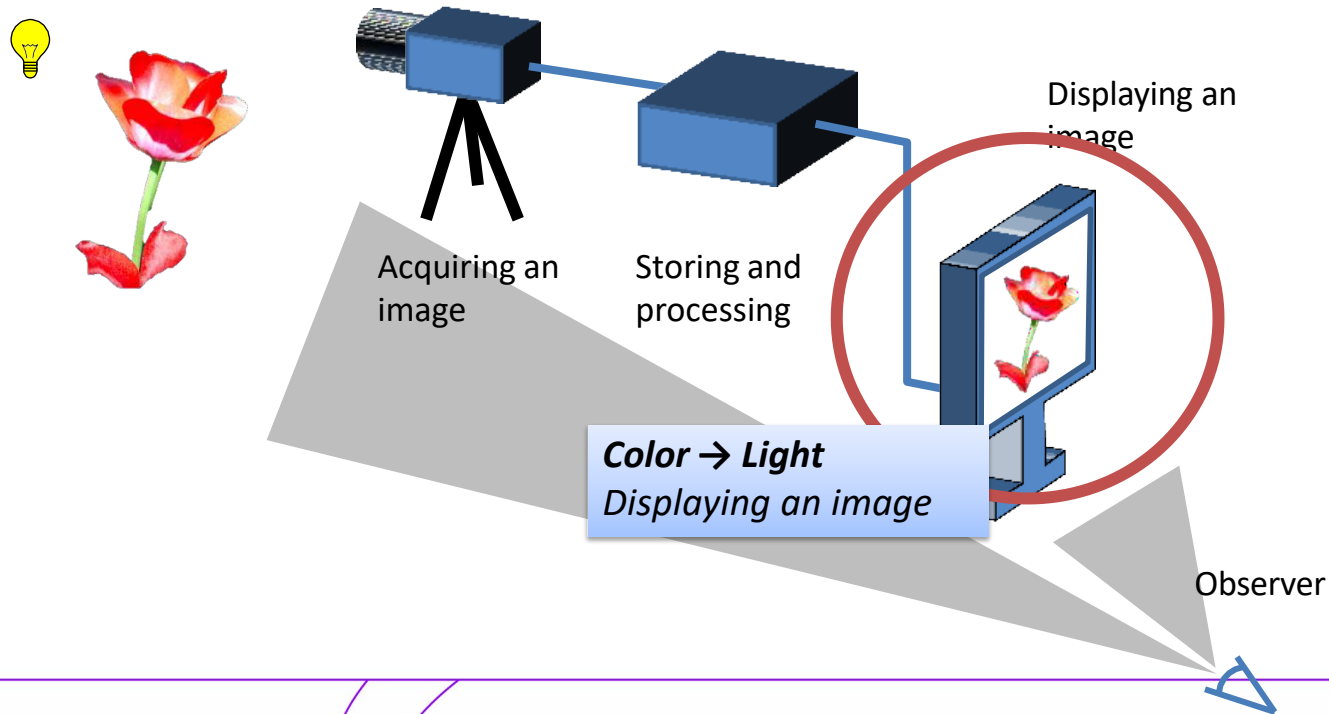
Light and color in a graphical system



Light and color in a graphical system



Light and color in a graphical system

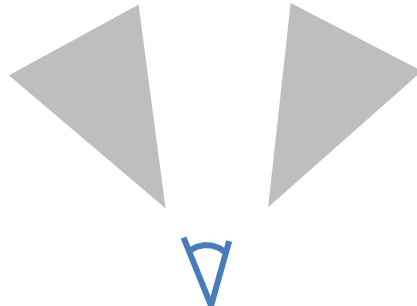


The «good» system

Expected



Received



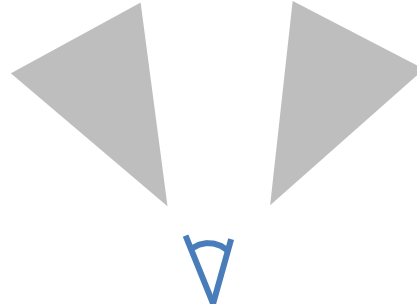
Observer

The «bad» system

Expected

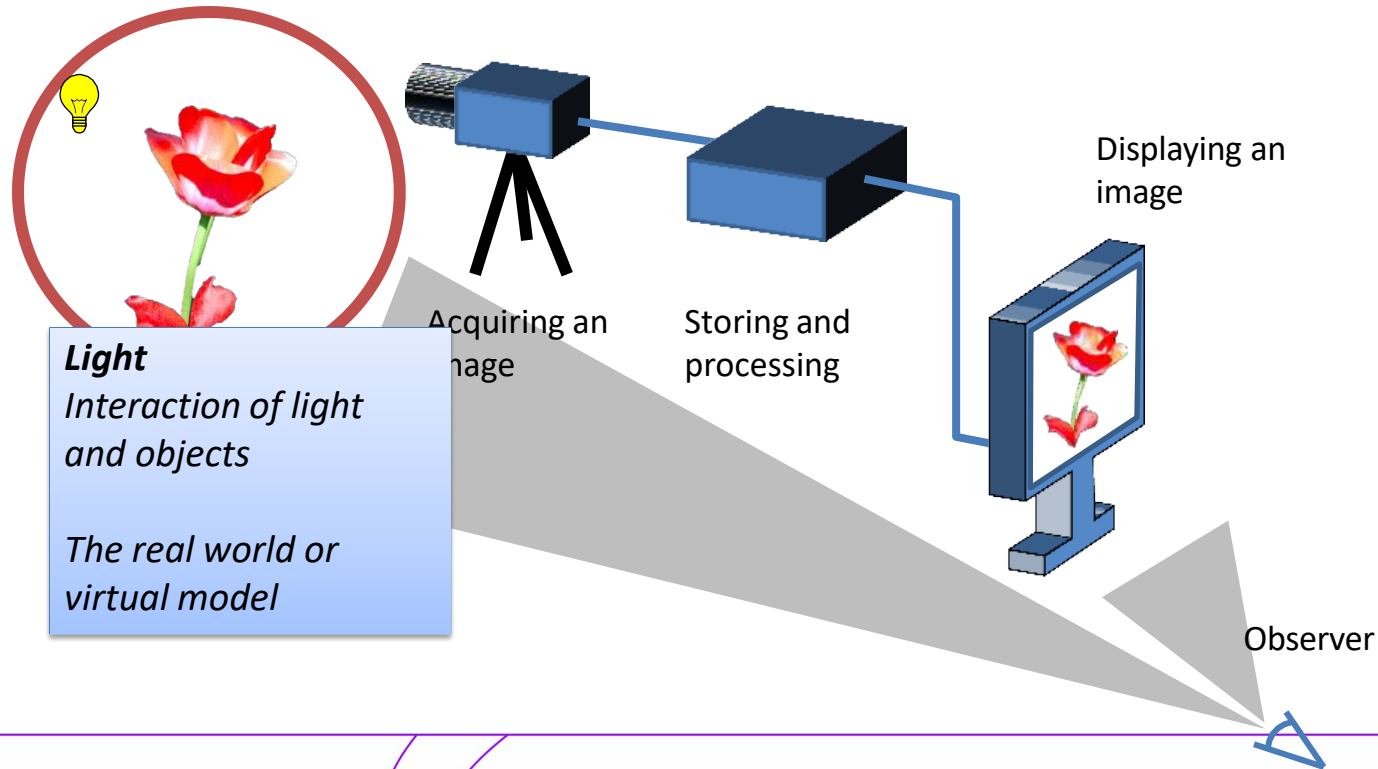


Received



Observer

Light and color in a graphical system



Light: dual nature



Electromagnetic wave

- wave optics

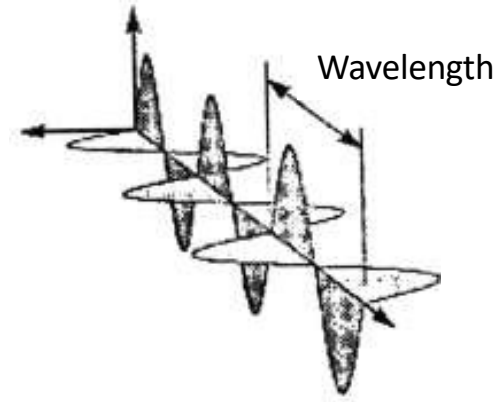
Stream of particles

- geometrical optics

The reasons for duality are explained in quantum optics

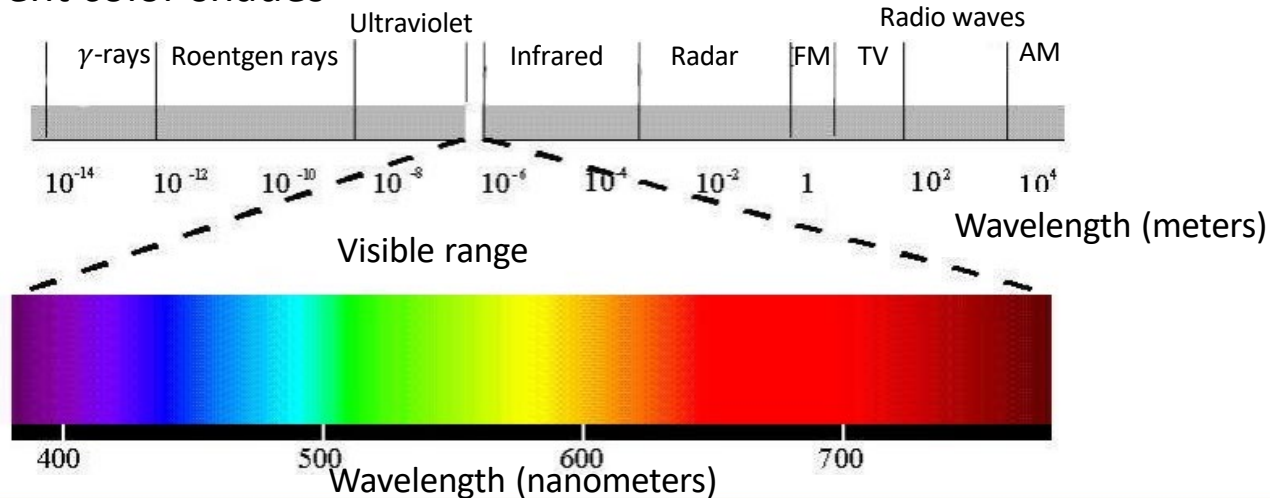
Light: electromagnetic waves or particles **iTMO**

- Dual nature of light
 - Electromagnetic waves
 - Stream of particles
- Light wave parameters: amplitude, wavelength, polarization
 - Amplitude \sim energy
 - Measured in Watts
(Watt = Joule/Second)
- Is emitted by discrete quanta called photons



Visible wavelengths range ~400-700nm

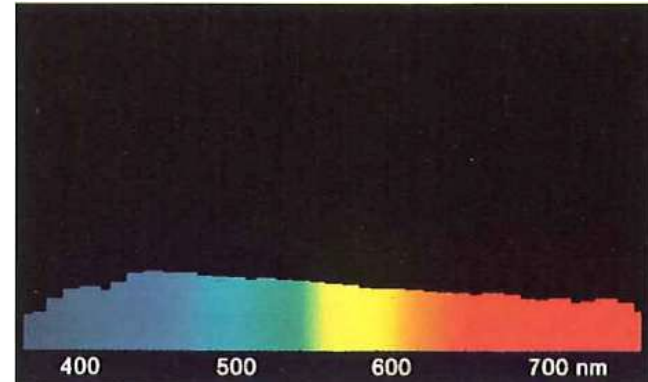
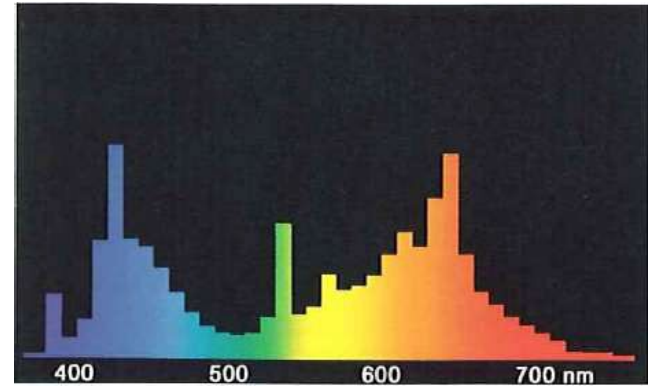
- Visible light is in range of 400-700 nanometers
 - 380-470 nm violet and blue color
 - 500-560 nm green color
 - 590-760 nm red color
- In smaller intervals of this range the color of radiation corresponds to different color shades



Light is a flow of waves

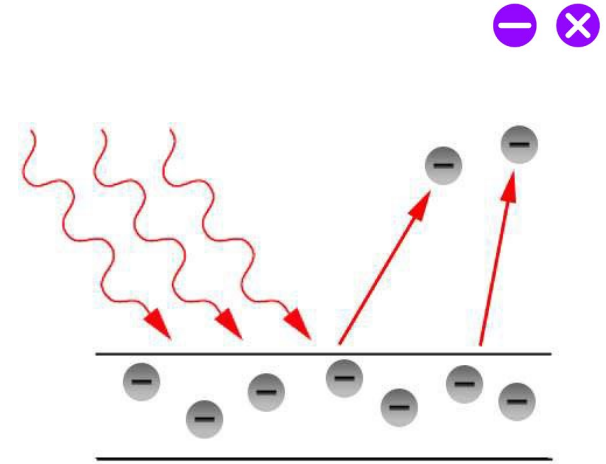
Light: flow of waves with different wavelengths, different amplitudes and different polarizations

- Can be considered as the sum of monochromatic emissions



Photoelectric effect

- Emission of electrons by an incident light
 - Albert Einstein's photoelectric effect experiment, 1905
 - A beam of light could eject electrons from metal
 - Light consists of photons with energy that depends on the frequency
 - If the frequency is over a certain limit, then it would have sufficient energy to eject an electron from metal
- Is one of the foundations of the photon theory (particle theory)

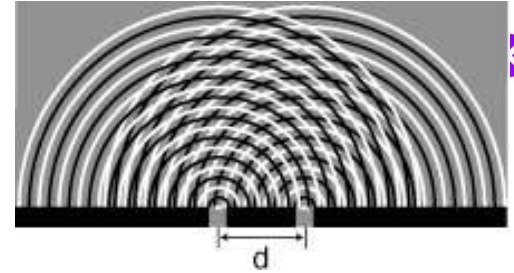




- The Fermat's principle
- The law of rectilinear light propagation
- The law of independent light ray propagation
- The law of light reflection
- The law of light refraction (Snell's law)
- The law of light ray reversibility

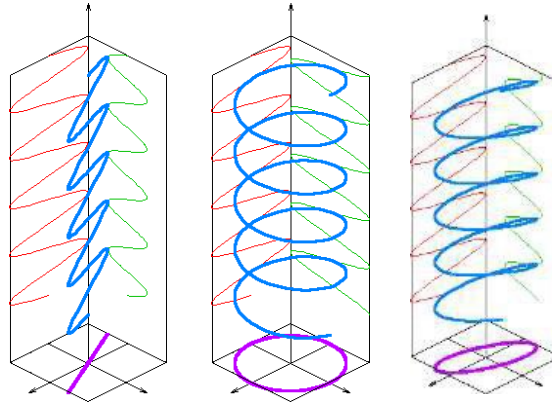
Wave nature of light: diffraction and interference

- The phenomenon of transformation of a wave propagating in space
 - Thomas Young experiments, 1901
 - If the light consisted of the small particles, the alternating light and dark bands would not have occurred
- Occurs at comparable wavelengths and sizes of medium inhomogeneity
- When the size of inhomogeneities is 3-4 orders of magnitude greater than the wavelength, diffraction can be neglected



Wave nature of light: polarization

- Light wave is a transverse wave



Polarization: example



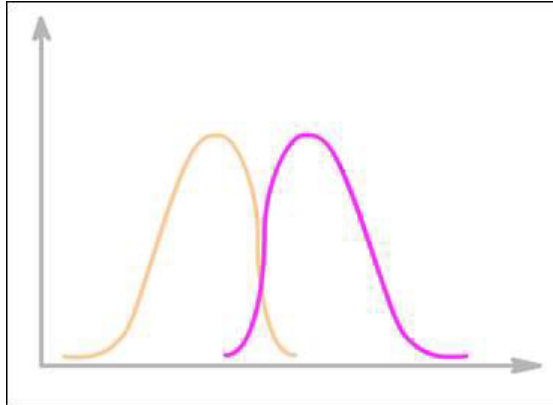
<https://commons.wikimedia.org/wiki/File:CircularPolarizer.jpg>

Fluorescence, phosphorescence

Non-thermal self luminescence of a substance.

Photoluminescence – self glow caused by an incident light (visible and UV)

- absorbs shorter wave and emits a longer one
- fluorescence (only when exposed to light)
- phosphorescence

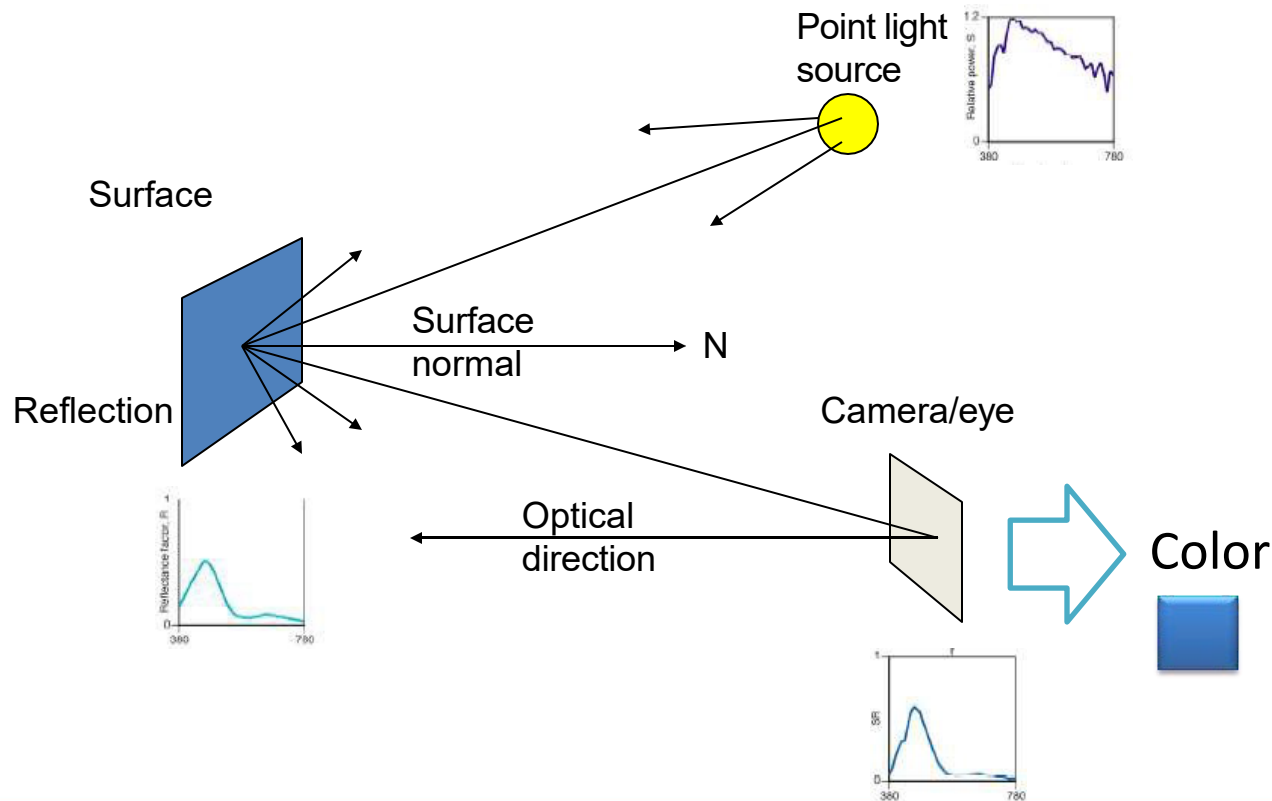


https://en.wikipedia.org/wiki/Image:Tonic_water_uv.jpg

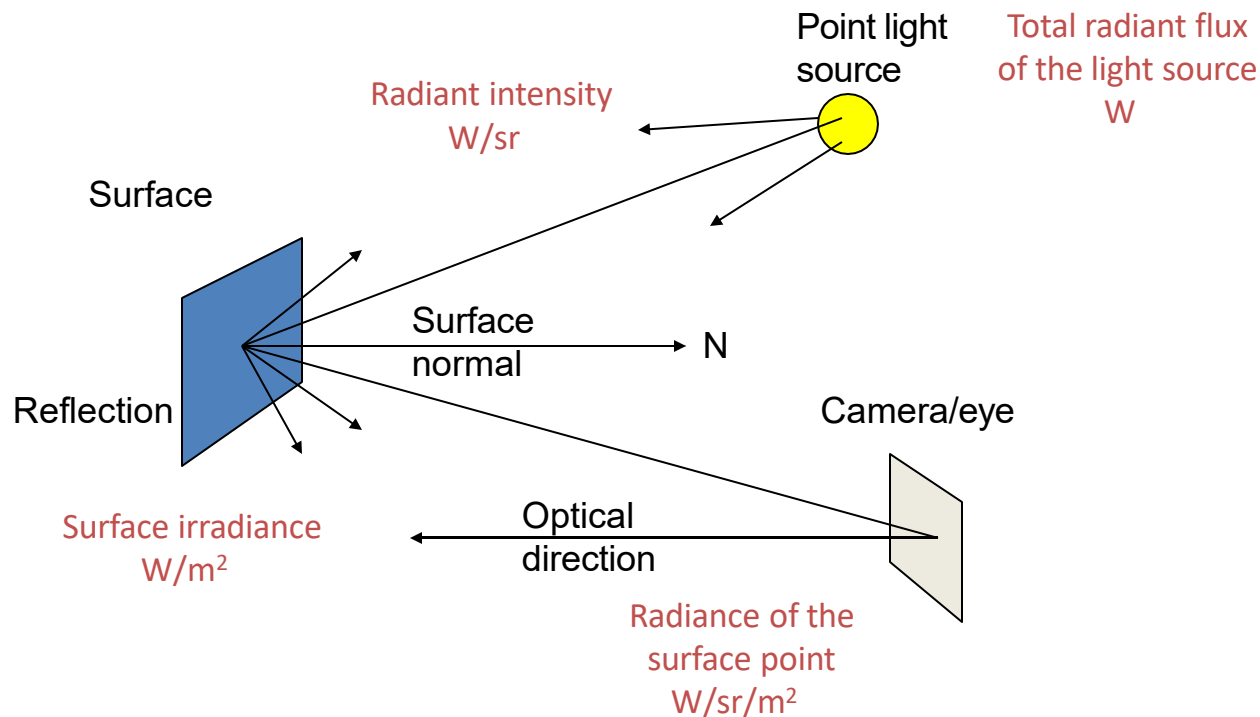


- Radiometry is the science of measuring electromagnetic radiation
 - Including visible light
- Radiometry does not consider the peculiarities of human perception
- Based on radiation as a flow of particles (geometric optics)
- However, it is possible to take some elements of wave optics into account

Light propagation



Radiometric units



Radiometry: base assumptions

- **Linearity**

The total effect of two input signals is always equal to the sum of the effects of each signal separately

- **Conservation of energy**

After transformation, an energy cannot produce more energy than it was initially

- **No polarization**

The only property of light is its wavelength distribution (frequency distribution)

- **No fluorescence and phosphorescence**

The behavior of light at one wavelength (frequency) is independent of the behavior at another

- **Stationary process**

The distribution of light energy is independent of time



Radiometry: limitations

Impossible to measure some physical effects :



- Diffraction
- Interference
- Polarization
- Fluorescence
- Phosphorescence

(However, the last three can be accounted)

Radiant energy



Notation: **Q**

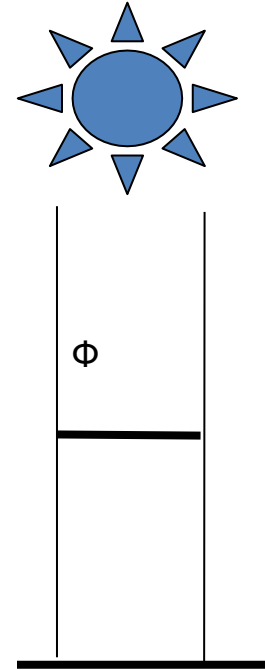
Units: **J** (Joule)

- Not too handy as have to account the whole energy emitted by body (e.g., when being heated)
- Have to describe the properties of the energy flow:
 - Velocity
 - Direction
 - Density
 - etc.

Radiant flux

Flux is the energy emitted per unit of time for a given surface

- Notation: Φ
- $\Phi = dQ / dt$
- Units: **W**
(Watt = Joule/Second)
- Stationary process



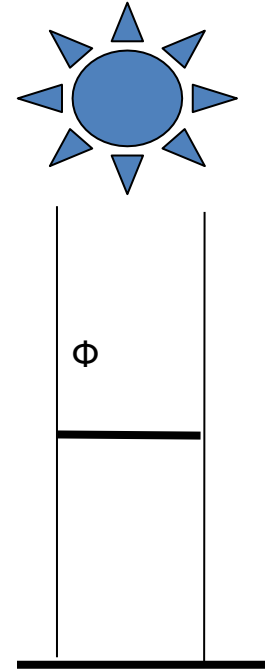
Radiant flux: measuring

- Place the light source
- Measure the change in the temperature of the plate at a given time interval

$$Q = cm\Delta T$$

$$\Phi = Q/\Delta t$$

c - specific heat capacity

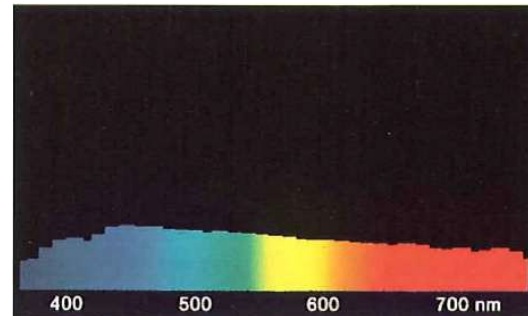
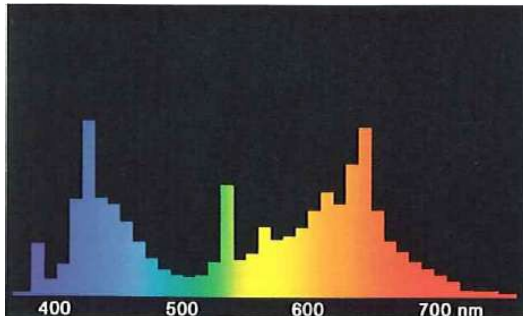


Radiant flux spectral density

- Φ defines the total characteristic over all wavelengths
- Φ_λ defined the radiant flux at a given wavelength

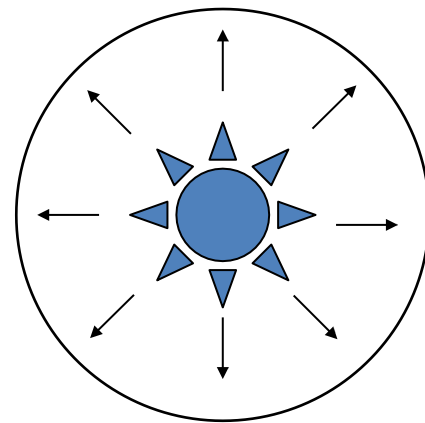


$$\Phi = \int \Phi_\lambda \, d\lambda$$



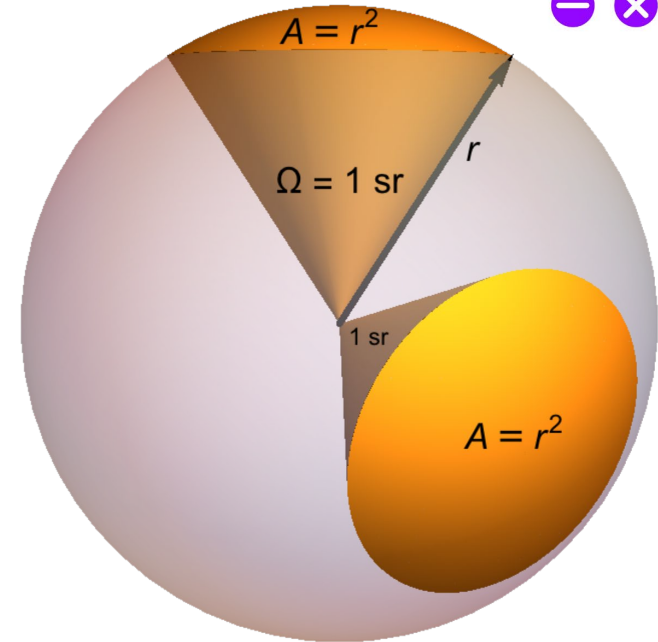
Total radiant flux

- It is often necessary to measure the total radiant flux of a light source
- The entire amount of energy emitted by the body over a period of time (exposure)
- Characterizes the light source, cannot be enlarged, only concentrated
- More detailed values are needed
 - Distribution by direction
 - Distribution by area



Solid angle

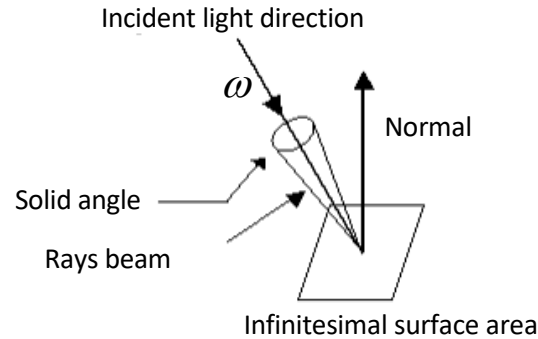
- The part of the space
 - Is the union of all rays emanating from a given point
 - And intersecting some surface
- Equal to the area of a segment of a unit sphere corresponding to the projection of the surface onto this sphere
- The whole sphere's solid angle is 4π
- For a sphere of radius R and projection area S , the solid angle is equal to S/R^2
- Units: *steradian (sr)*



https://commons.wikimedia.org/wiki/File:Solid_Angle,_1_Steradian.svg

Radiant intensity

- The density of the light radiant flux passing through the solid angle
- Notation: I
- Units: **W/sr**



$$I = \frac{d\Phi}{d\omega}$$

Connection between total radiant flux and radiant intensity



$$\Phi = \int I \, d\omega = I \int d\omega = 4\pi I$$

Irradiance and radiosity

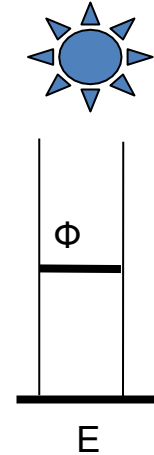
- Units are needed to describe the radiant flux incident on the surface or emitted from the surface
- The density of the light radiant flux passing through a given area
- Since we don't know the direction of the light propagation, two symmetrical terms are used
 - irradiance for incident light
 - radiant exitance (radiosity) for emitted light



Irradiance

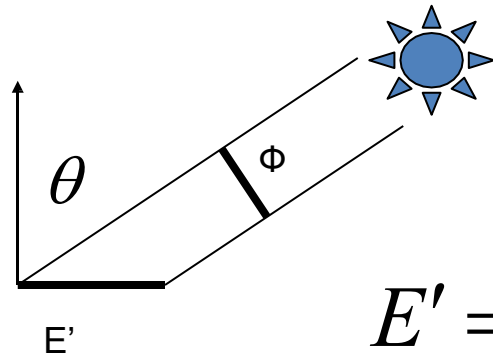
- Notation: **E**
- Units: **W/m²**
(Watts per square meter)

$$E = \frac{d\Phi}{dS}$$

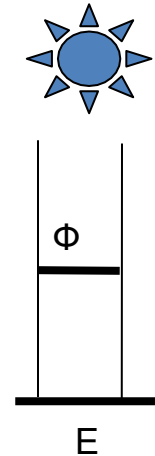


Irradiance and "cosines law"

- In many lighting models, *cosines* is used as a multiplier
- True for parallel beams



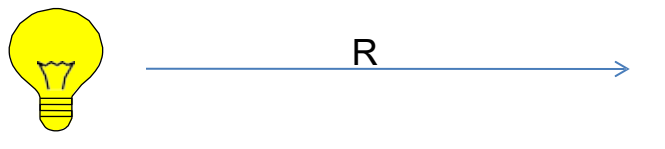
$$E' = E \cos \theta$$



Irradiance and radiance intensity: the inverse square law



Let a plane with an area A be illuminated by a point light source at a distance R

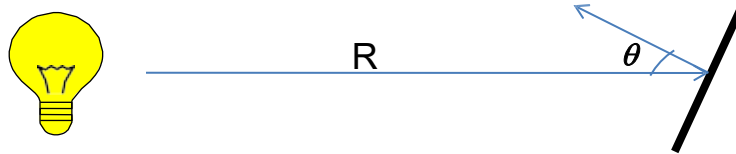


$$E = \frac{d\Phi}{dA} = \frac{I d\omega}{dA} = \frac{I dA}{R^2 dA} = \frac{I}{R^2}$$

$$d\omega = \frac{dA}{R^2}$$

The inverse square law and “cosines” law iTMO

- Let a plane with an area A be illuminated by a point light source at a distance R
- At an angle θ (*theta*)



$$E = \frac{d\Phi}{dA} = \frac{I d\omega}{dA} = \frac{I dA \cos \theta}{R^2 dA} = \frac{I \cos \theta}{R^2}$$

$$d\omega = \frac{dA \cos \theta}{R^2}$$

Radiant exitance

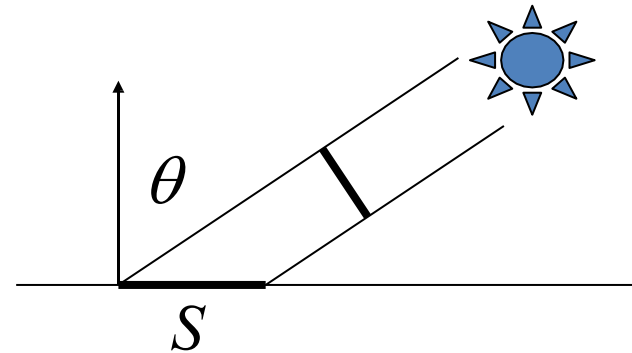
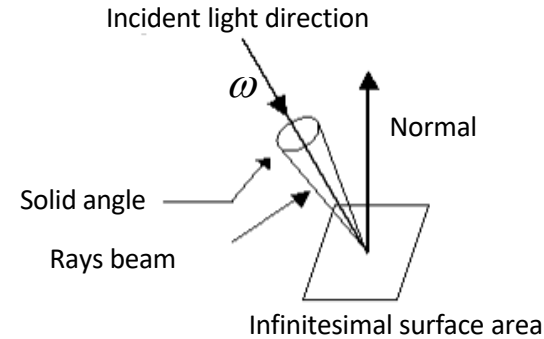
- Notation: **M**
- Units: **W/m²**
(Watts per square meter)
- Also called **radiosity**

$$M = \frac{d\Phi}{dS}$$

Radiance

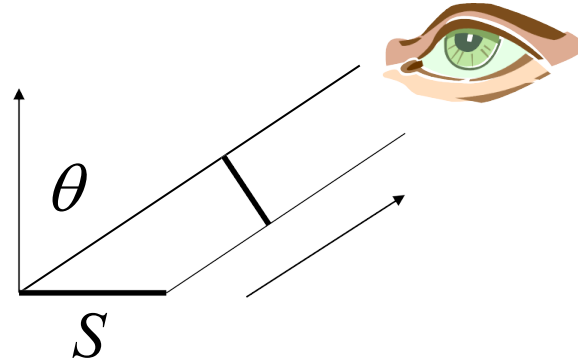
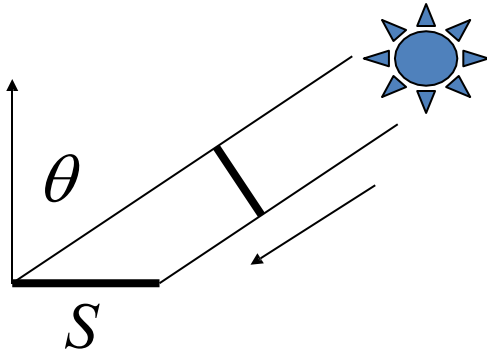
- The most important unit
- Arbitrary light source
- The density of the radiant flux incident at the unit area site, passing through a unit solid angle
- Notation: **L**
- Units: **W / (sr * m²)**
(Watts per steradian per square meter)

$$L = \frac{d^2\Phi}{d\omega dS \cos\theta}$$



Incident and emitted radiance

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Connection between radiometric units




Radiance

Radiant intensity

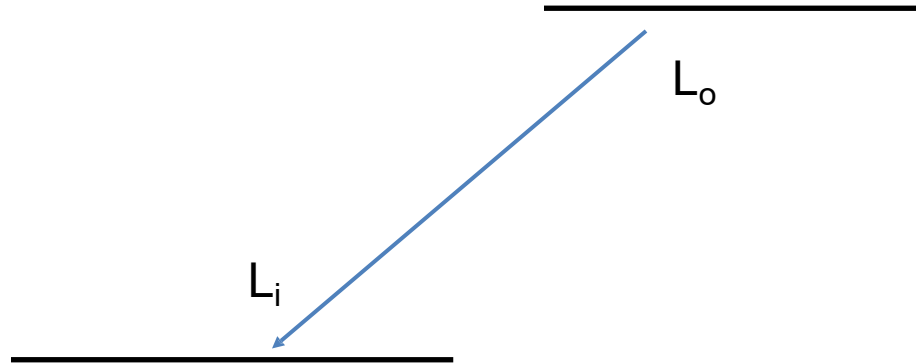
Irradiance

Radiosity


$$L = \frac{dI}{dS \cos \theta} = \frac{dE}{d\omega \cos \theta} = \frac{dM}{d\omega \cos \theta}$$

Radiance properties

- Losslessly transmitted in a vacuum



- The photo camera stores the radiance values
- The human eye reacts on radiance (brightness)

Lambertian light source

- Lambertian light source have constant radiance
 $L = \text{const}$
- So, it looks the same from all directions
- Properties:
 - $I = I_0 \cdot \cos(\Theta)$
 - $M = \pi \cdot L$
 - A luminous flat disk is indistinguishable from a hemisphere
 - Sun is a Lambertian light source



Sample radiometric values

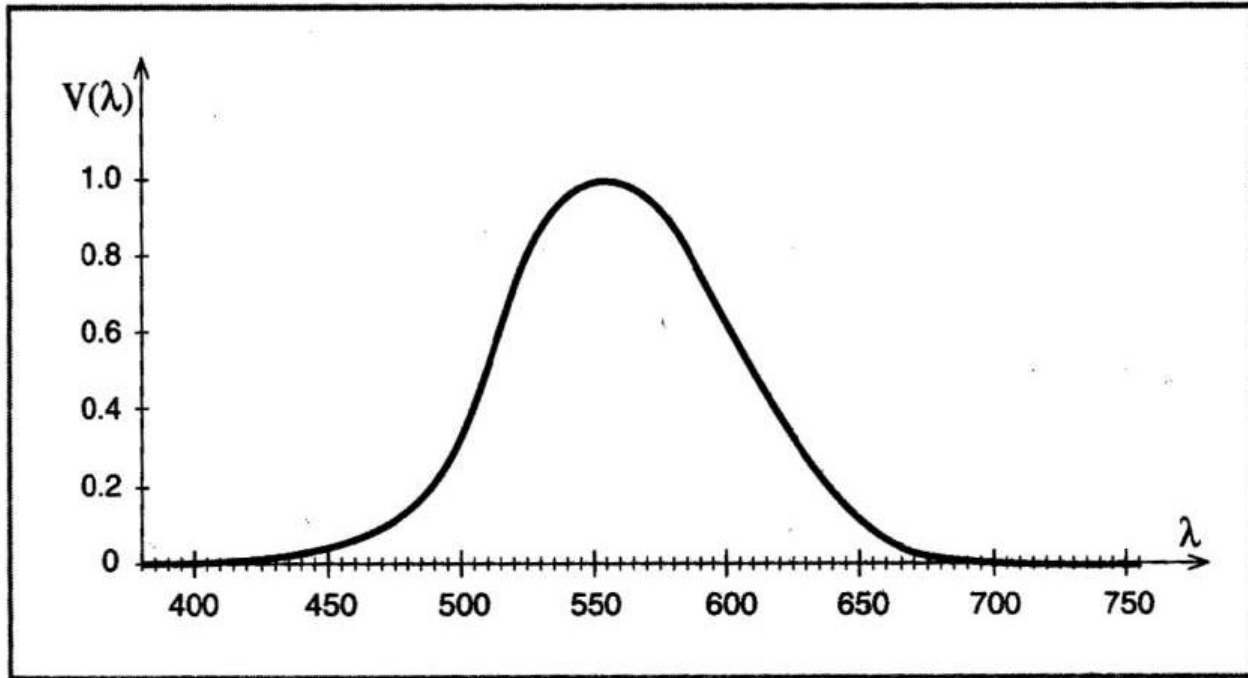
Name	Value
The total radiant flux of gas-filled tungsten incandescent lamp	82 W
The radiant flux of a typical helium-neon laser of medium power at 632.8nm	5 mW
The radiant flux of typical 40W daylight lamp	23.2 W
The beyond atmospheric irradiance at a mean earth orbit	1267 W/m ²
The direct irradiance of the Earth, clear sky, winter, South-East of the US, midday	852 W/m ²
The hemispherical irradiance of the Earth, clear sky, winter, South-East of the US, midday	686 W/m ²
The radiance at the Sun surface	2.3e7 W/m ² /sr
The visible radiance of the Sun from the Earth	1.4e7 W/m ² /sr



- Photometry can be considered of as a subset of radiometry in which all radiometric units have been modified to account for the sensitivity of the human eye



Spectral luminous efficiency function $V(\lambda)$ iTMO



Conversion between radiometric and photometric units



$$Q_v = 683 \int_{380}^{780} Q_\lambda V(\lambda) d\lambda$$

Q_v – one of units Φ , I , E , M , L

Luminous flux

- Luminous flux – lm (*lumen*)
 - Radiant flux (W, Watt) assessed by human eye visual perception
 - Radiant flux weighted by the spectral luminous efficiency function



Luminous intensity

- Luminous intensity - cd (*candela*, lm/sr)
 - Historically, It was measured with an “etalon candle”
- If light source luminous intensity is equal to 1 cd in all directions, then the total luminous flux is equal to 4π lm
- The coefficient 683 that we met before comes exactly from the definition of the candela, so that 1 candela is approximately equal to the luminous intensity of a candle



Illuminance

- Illuminance – lx (*lux*, lm/m^2)
– Irradiance - W/m^2



Luminance

- Luminance – nit ($\text{cd}/\text{m}^2/\text{sr}$)
– Radiance - $\text{W}/\text{m}^2/\text{sr}$



Radiometric and photometric units

Radiometric units	Units	Photometric units	Units
Radiant flux	W	Luminous flux	lm (lumen)
Radiance intensity	W/sr	Luminous intensity	cd (candela) = lm/sr
Irradiance, radiosity	W/m ²	Illuminance, luminosity	lx (lux) = lm/m ²
Radiance	W/m ² /sr	Luminance	nit = cd/m ² /sr



Additional reading



- DeCusatis, C., "Handbook of Applied Photometry." AIP Press (1997)
- McCluney, W. R., "Introduction to Radiometry and Photometry", Artech House (1994)

**THANK YOU
FOR YOUR TIME!**

it^{'s}**MO** *re than a*
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