# COMPARING MEASUREMENTS USING SCIENTIFIC NOTATION LEARNING GOAL

1. I can compare measurements using scientific notation.

#### **ASSIGNMENT**

Choose one thing that is very small and one thing that is very small, and compare the two sizes using scientific notation. Show all of the steps to receive credit!

Thing one: \_\_\_\_\_

Thing two: \_\_\_\_\_

#### **DISTANCE SCALE**

Mark approximately where the two things you chose fit on the scale. (**Note:** If your measurement is not a distance, use the scale on the next slide).

Atomic nucleus					Human					Galaxy	
$10^{-15}$	$10^{-12}$	$10^{-9}$	$10^{-6}$	$10^{-3}$	$10^{0}$	$10^{3}$	$10^{6}$	$10^{9}$	$10^{12}$	$10^{15}$	meters (m)

#### **NON-DISTANCE SCALE**

Use this scale if your measurements are not lengths.

Mark approximately where the two things you chose fit on the scale.

 $10^{-15}$   $10^{-12}$   $10^{-9}$   $10^{-6}$   $10^{-3}$   $10^{0}$   $10^{3}$   $10^{6}$   $10^{9}$   $10^{12}$   $10^{15}$ 

### Use the table to lookup how to write measurements (like km, cm, mm, etc.) using scientific notation.

The Prefixes Used with SI Units								
			Scientific					
Prefix	Symbol	Meaning	Notation					
exa-	E	1,000,000,000,000,000	1018					
peta-	P	1,000,000,000,000	1015					
tera-	T	1,000,000,000,000	1012					
giga-	G	1,000,000,000	10 <sup>9</sup>					
mega-	M	1,000,000	10 <sup>6</sup>					
kilo-	k	1,000	10 <sup>3</sup>					
hecto-	h	100	10 <sup>2</sup>					
deka-	da	10	10 <sup>1</sup>					
_	_	1	10°					
deci-	d	0.1	10-1					
centi-	С	0.01	10-2					
milli-	m	0.001	$10^{-3}$					
micro-	μ	0.000 001	$10^{-6}$					
nano-	n	0.000 000 001	10-9					
pico-	p	0.000 000 000 001	$10^{-12}$					
femto-	f	0.000 000 000 000 001	$10^{-15}$					

#### **COMPARE THE SIZES**

In the space below, calculate the ratio of the sizes, and write 1-3 sentences to interpret your ratio.

## PRACTICE WRITING MEASUREMENTS USING SCIENTIFIC NOTATION

Convert each measurements into meters using scientific notation:  $35,650 \text{ km} \text{ km = kilometers} \\
35,650 \cdot 10^{3} \text{ m}$   $124 \text{ mm} \text{ mm = millimeters} \\
124 \cdot 15^{3} \text{ m}$ 

 $51\,\mathrm{cm}$ 

 $0.051\,\mathrm{mm}$