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Background: Transistor Theory

Ahh yes, the transistor. We all owe you so much. The transistor is the reason why your cell phone fits inside your pocket and why your SpikerBox can measure spikes from neurons.

 Time 45 Minutes Difficulty Advanced

What will you learn?

In this lesson you will learn what a transistor is and how it works. It is a complicated, lovely mix of physics and chemistry, but with patience you can learn the principles behind the transistor. In the next lab (<http://www.backyardbrains.com/experiments/transistorDesign>) you will use this theory to build your very own junior SpikerBox on a solderless breadboard.

Prerequisite Labs

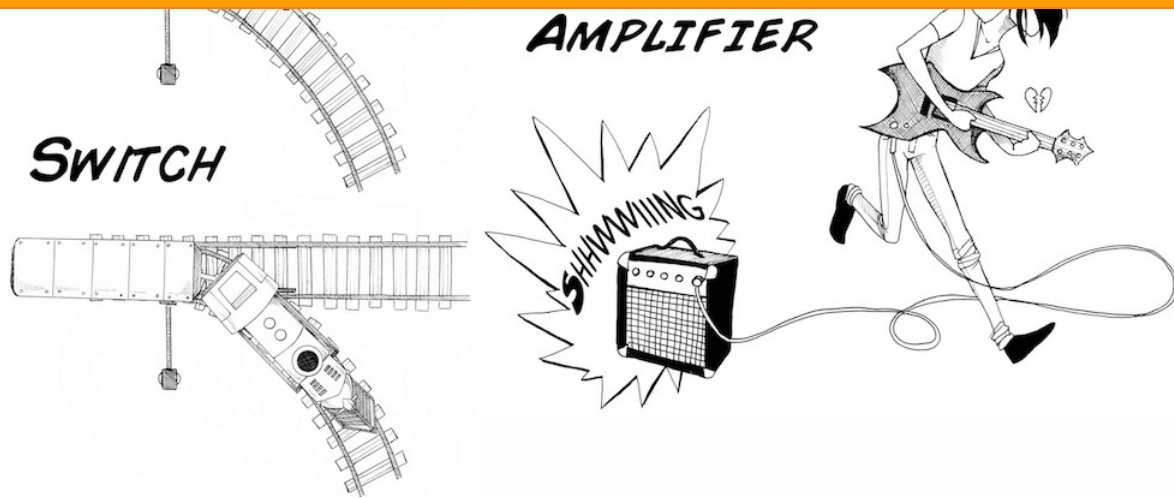
SpikerBox (spikerbox) - A familiarity with spikes will make this theory writeup more meaningful

Equipment

Your Brain (<http://en.wikipedia.org/wiki/Brain>)

Background

Modern computing and electronics are built upon the transistor, making it in our opinion the most important invention of the 20th century. Transistors are used as switches (devices telling signals where to go) or amplifiers (devices transforming "small" signals into bigger ones), and these two functions are what make your favorite mobile phones possible.

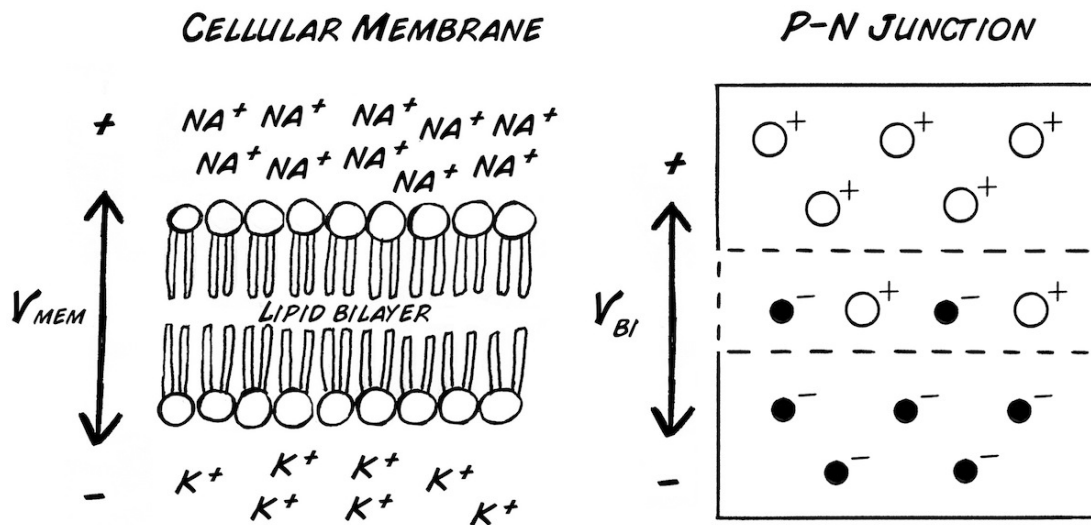


(img/Amptraincombo_smaller.jpeg)

Transistors can be mass produced at very low costs, and transistors are the reason that computers keep getting smaller yet more powerful every day. There are more than 60 million transistors built every year for every man, woman and child on earth. Transistors are the key to our modern world. So, if they are so wonderful, then how do they work?

p-n junctions and doping

In order to understand how transistors work, you first must understand the concept of the p-n junction. The p-n junction shares some similarities with the neural lipid bi-layer (cellular membrane) that we learned about in Experiment 3. Recall, the lipid bi-layer is the barrier between the inside and outside of the cell, and it is characterized by a buildup of charged ions on both sides of the barrier. The charged ions generate a difference in electric potential that ultimately allows action potentials. Similarly, a p-n junction is the border between two materials with different charges on them. Instead of ions, the charges in a p-n junction are controlled by the presence (-) or absence (+) of electrons.



(img/Figure7_PN_vs_CellMembrane.jpeg)

Electrons have a negative charge, and the motion of these charges through conductive material is the basis of electricity. In some materials (named semiconductors) we can manipulate how many electrons are present through a process called doping, which means introducing impurities into extremely pure semiconductors. Get out your periodic tables, because this process is only possible due to the chemical properties of certain elements.

1.sopn																4.sopn																						
lithium 3 Li		beryllium 4 Be														boron 5 B		carbon 6 C		nitrogen 7 N		oxygen 8 O		fluorine 9 F		neon 10 Ne												
sodium 11 Na		magnesium 12 Mg														aluminum 13 Al		silicon 14 Si		phosphorus 15 P		sulfur 16 S		chlorine 17 Cl		argon 18 Ar												
potassium 19 K		calcium 20 Ca		scandium 21 Sc	titanium 22 Ti	vanadium 23 V	chromium 24 Cr	manganese 25 Mn	iron 26 Fe	cobalt 27 Co	nickel 28 Ni	copper 29 Cu	zinc 30 Zn	gallium 31 Ga	germanium 32 Ge	arsenic 33 As	selenium 34 Se	bromine 35 Br	krypton 36 Kr																			
rubidium 37 Rb		strontium 38 Sr		yttrium 39 Y	zirconium 40 Zr	niobium 41 Nb	molybdenum 42 Mo	technetium 43 Tc	ruthenium 44 Ru	rhodium 45 Rh	palladium 46 Pd	silver 47 Ag	cadmium 48 Cd	indium 49 In	tin 50 Sn	antimony 51 Sb	tellurium 52 Te	iodine 53 I	xenon 54 Xe																			
cesium 55 Cs		barium 56 Ba														thallium 81 Tl		lead 82 Pb		bismuth 83 Bi		polonium 84 Po		astatine 85 At		radon 86 Rn												
francium 87 Fr		radium 88 Ra														copernicium 110 Ds		darmstadtium 111 Ds																				
lanthanum 57 La		cerium 58 Ce	praseodymium 59 Pr	neodymium 60 Nd	promethium 61 Pm	europium 62 Eu	gadolinium 63 Gd	terbium 64 Tb	dysprosium 65 Dy	holmium 66 Ho	erbium 67 Er	thulium 68 Tm	ytterbium 69 Yb	lutetium 70 Lu																								
actinium 89 Ac		thorium 90 Th	protactinium 91 Pa	uranium 92 U	neptunium 93 Np	plutonium 94 Pu	americium 95 Am	curium 96 Cm	berkelium 97 Bk	californium 98 Cf	esboium 99 Es	fermium 100 Fm	mendelevium 101 Md	nobelium 102 No	lawrencium 103 Lr																							

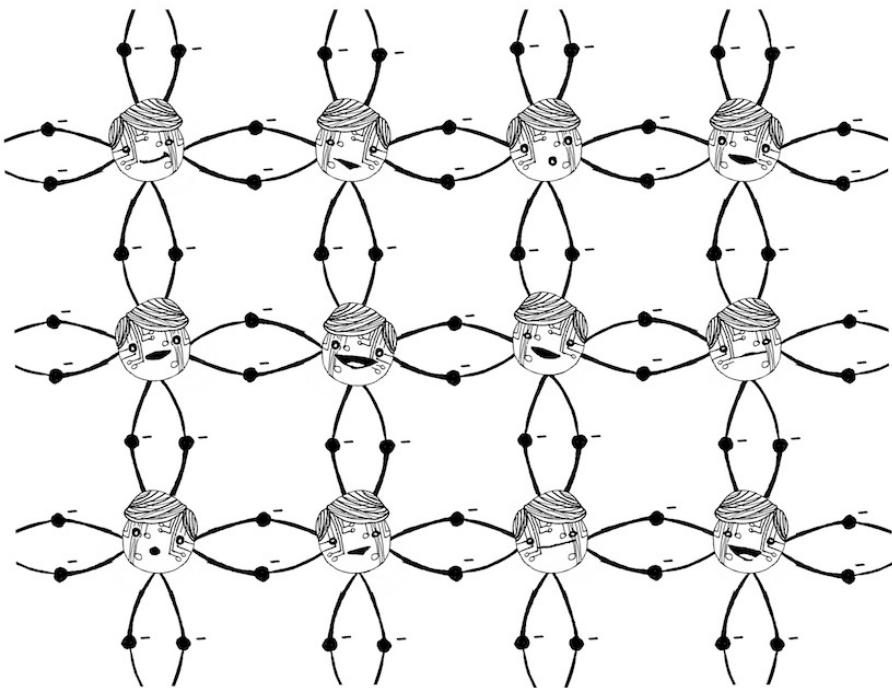
(img/Periodic_table.png)

Semiconductors come from what was known as Group IV of the periodic table, which includes carbon, silicon and germanium. Science fiction has often referenced these elements, because their properties make them such a key aspect of both biological and machine-based systems. Each Group IV element has four electrons in its outer energy level, but it can ultimately hold up to eight electrons. This is key, because these Group IV elements can then form four-way covalent bonds in a crystal lattice so that each atom's outer energy level is stable.

LONELY SILICON ATOM

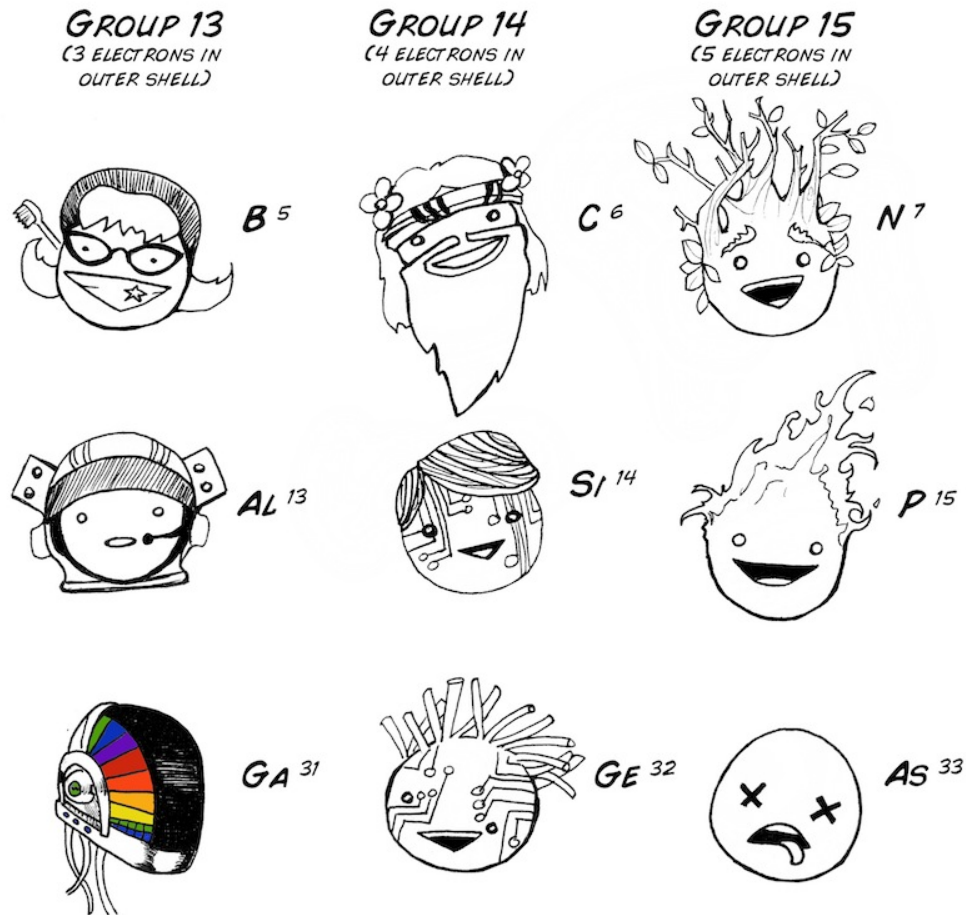


HAPPY SILICON MATRIX



(img/Figure3_Silicon.jpeg)

bonds with electrons.

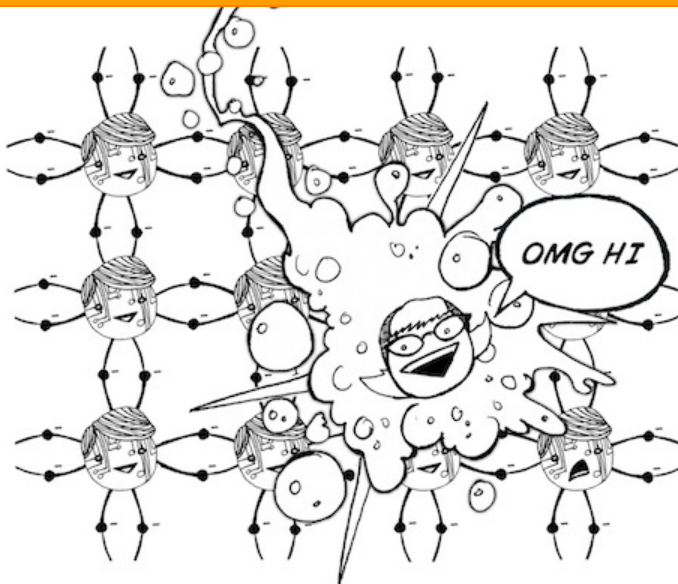


(img/Figure4_GroupIII_IV_V_atoms.jpeg)

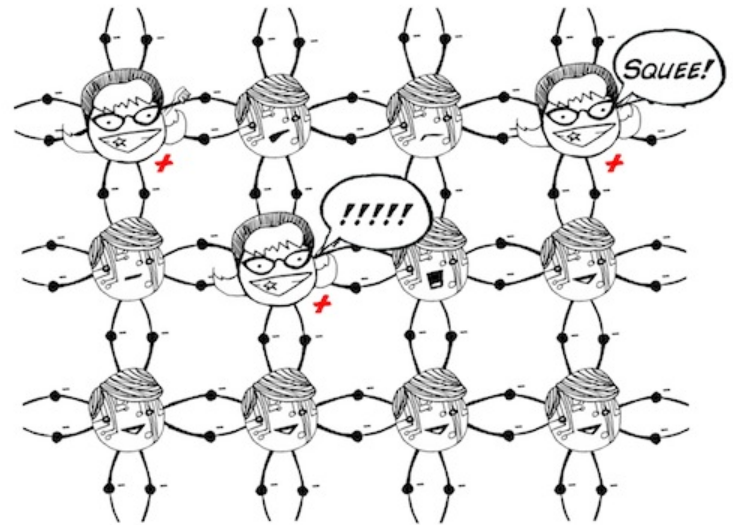
If you take a block of a pure Group IV element like silicon and zap it with some Group III atoms like boron, the boron will attempt to fit into the lattice. However, since the boron only has three electrons in the outer shell, one of the four silicon neighbors in the lattice will be short one electron in the covalent bond. Thus, the bond will have a net positive charge (absence of electron), which can attract and accept an electron from a neighboring bond. Group III atoms are known as acceptors.

Doping a semiconductor with acceptors will generate an excess of these absences-of-electrons (known as holes), which results in the appearance of a surplus of "positive" charges in the material, leading to this material being called "positively-doped" or "p-doped".

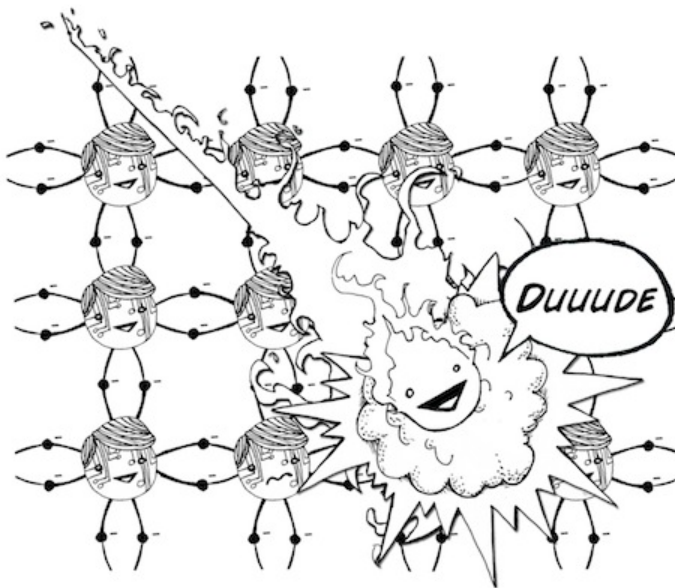
As you might conversely expect, when Group V elements like phosphorus that have five electrons are added to silicon, this forms bonds with an excess of electrons. Group V atoms are thus known as "donors". Doping a semiconductor with donors will generate a large concentration of negatively charged electrons, making the material "negatively doped" or "n-doped".



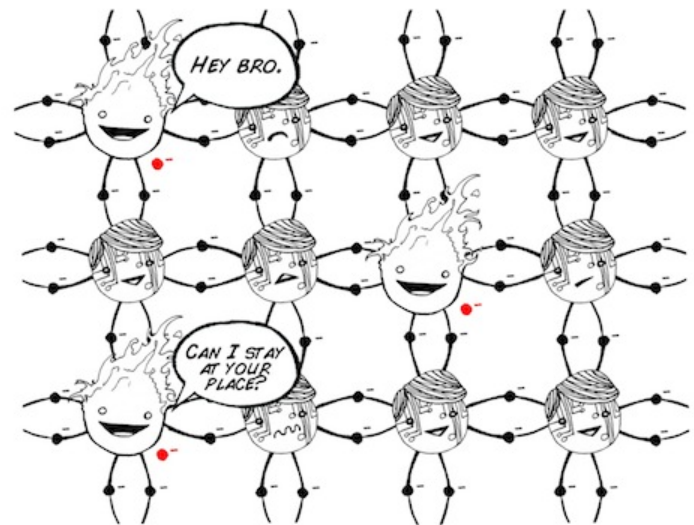
P-DOPED SILICON CRYSTAL LATTICE



EVERY PLACE BORON REPLACES SILICON THE MATRIX IS SHORT AN ELECTRON → FLOATING POSITIVE CHARGES → P-DOPED



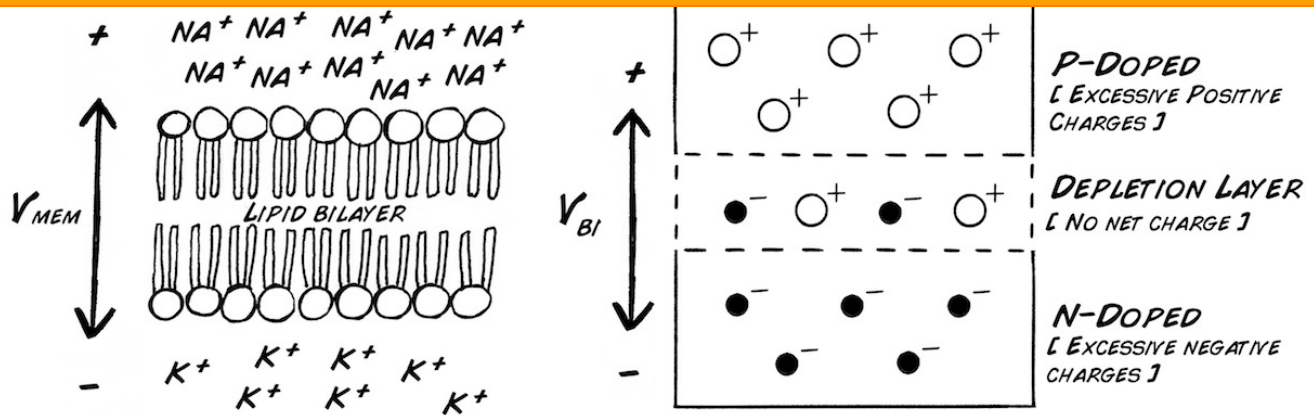
N-DOPED SILICON CRYSTAL LATTICE



EVERY PLACE PHOSPHORUS REPLACES SILICON THE MATRIX HAS AN EXTRA ELECTRON → FLOATING NEGATIVE CHARGES → N-DOPED

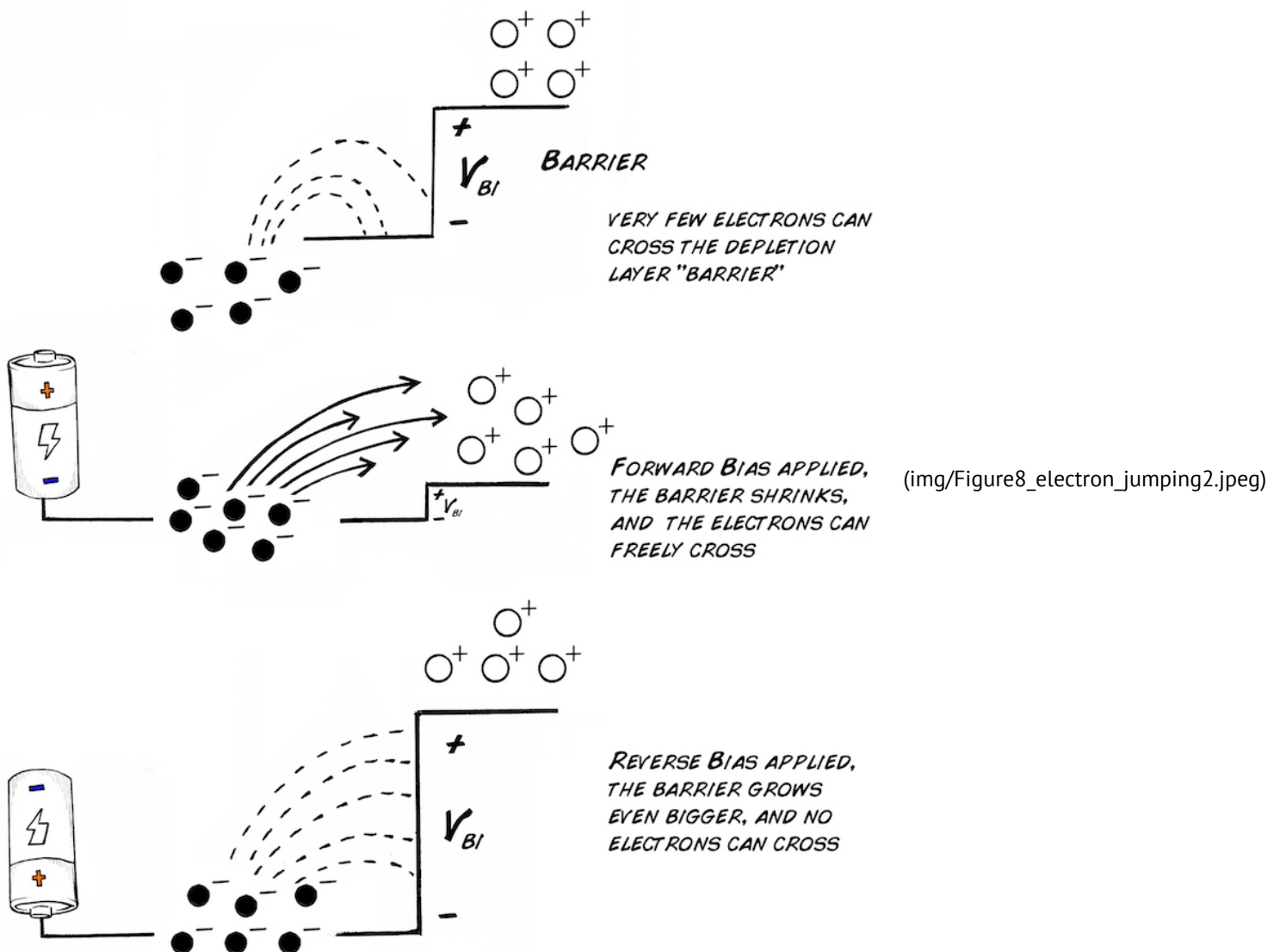
(img/Figure5_Doping2.jpeg)

Both p-doped and n-doped semiconductors are relatively electrically conductive on their own, but what happens when you put a block of p-doped semiconductor next to a block of n-doped semiconductor? The electrons in the n-doped material are attracted to the positively charged p-doped substance, and the excess electrons and positive charges meet in the middle at the junction between the two blocks. When the electrons and holes meet, they cancel each other out and form a layer that is depleted of charges, or a depletion layer. Like the neural bi-layer, the resultant electrical potential properties of the p-n junction allow many functions.



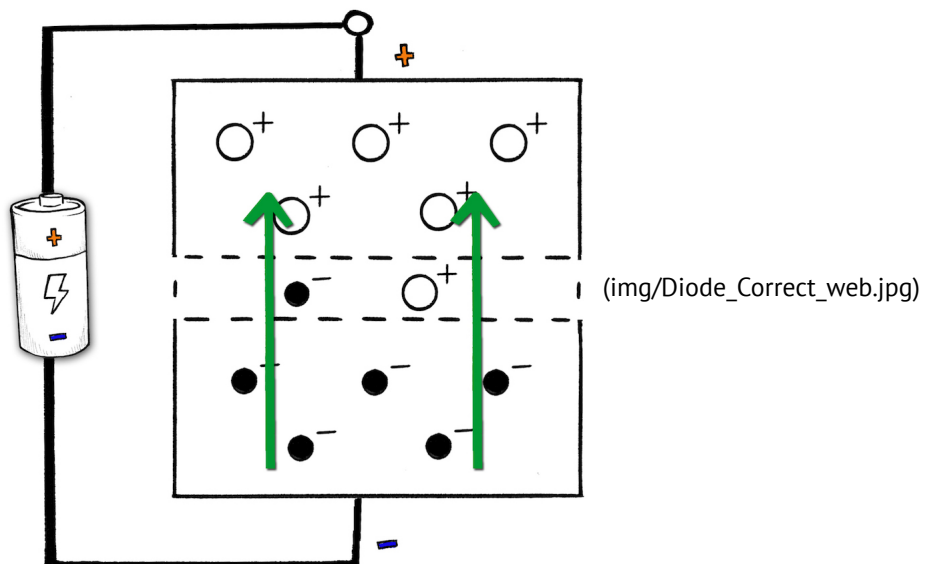
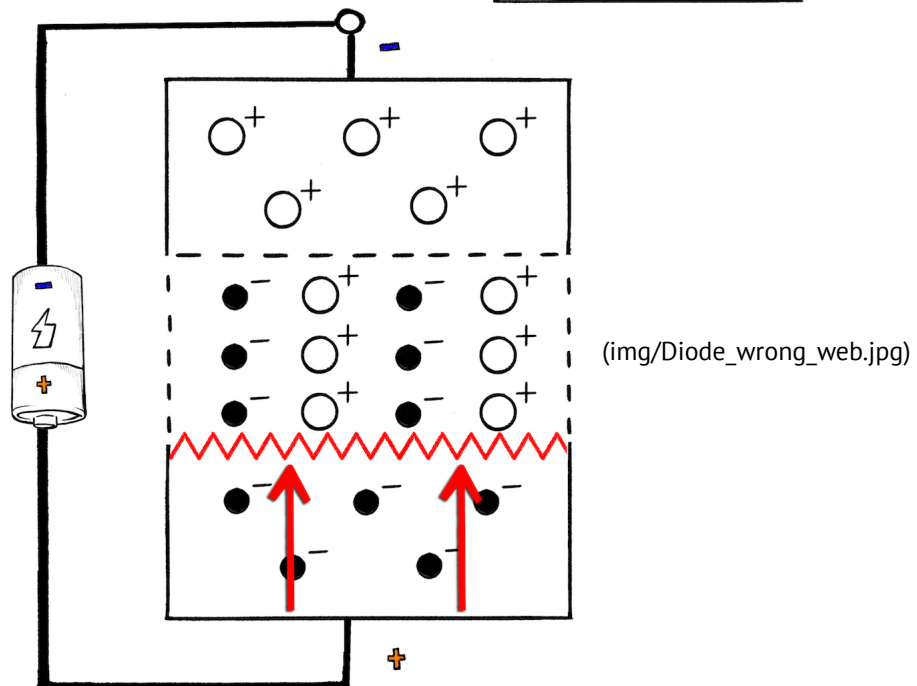
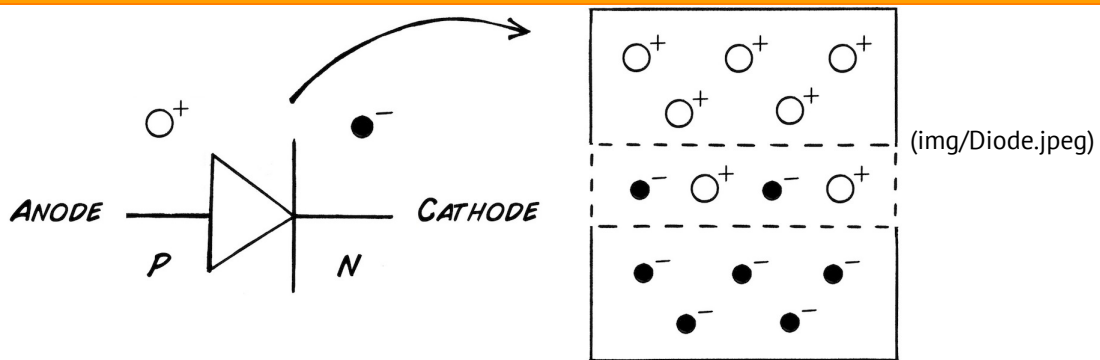
(img/Figure7_Partdeux_vs_CellMembrane.jpeg)

The depletion layer, due to the lack of free charges, is non-conductive without an applied external voltage. If the p-doped side of the p-n junction is connected to the positive voltage and the n-doped side with the negative voltage of a battery, this results in a lessening of the electric potential barrier and allows electrons to cross the p-n junction, which results in electric current flow. This process is called forward bias. If, on the other hand, the p-type semiconductor is connected with the negative voltage and the n-type with the positive voltage, the electrons and positive charges (holes) are pulled further away from the depletion region, which results in a larger electric potential barrier that behaves as an insulator. This is called reverse bias.

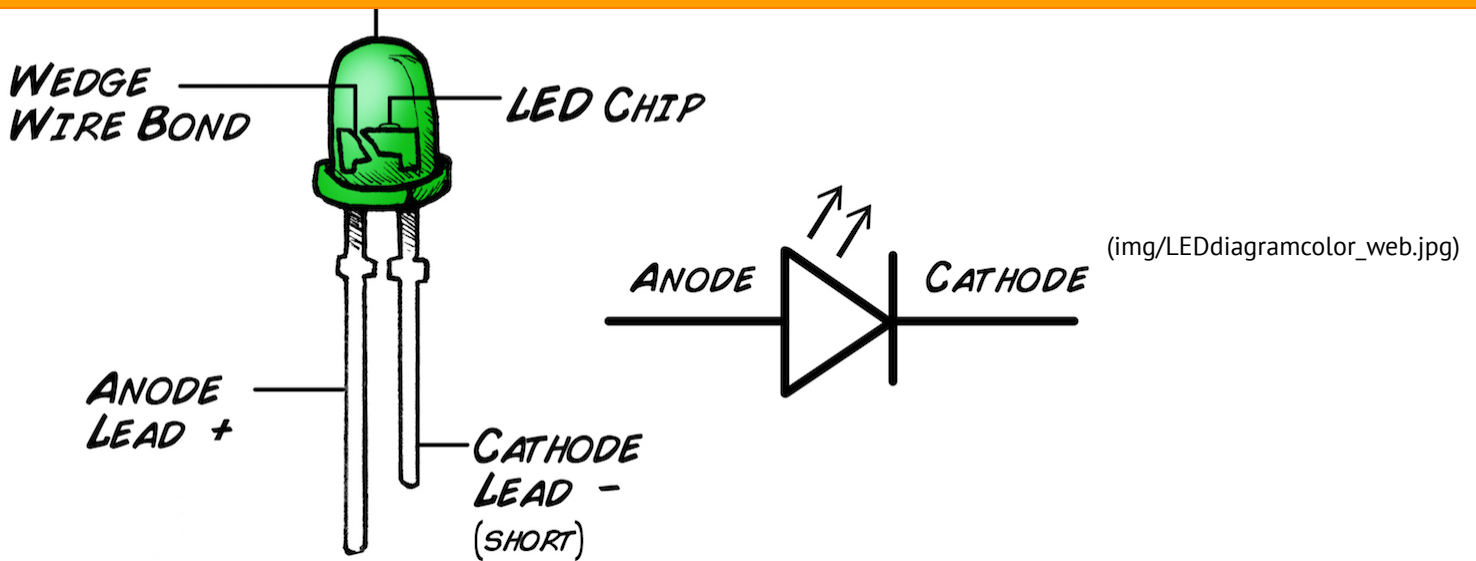


(img/Figure8_electron_jumping2.jpeg)

Thus, p-n junctions are therefore commonly used as diodes, which are devices that allow electricity to flow in one direction but not in the opposite direction. Importantly, diodes only allow one-way current through if a certain voltage is reached, or "forward voltage."

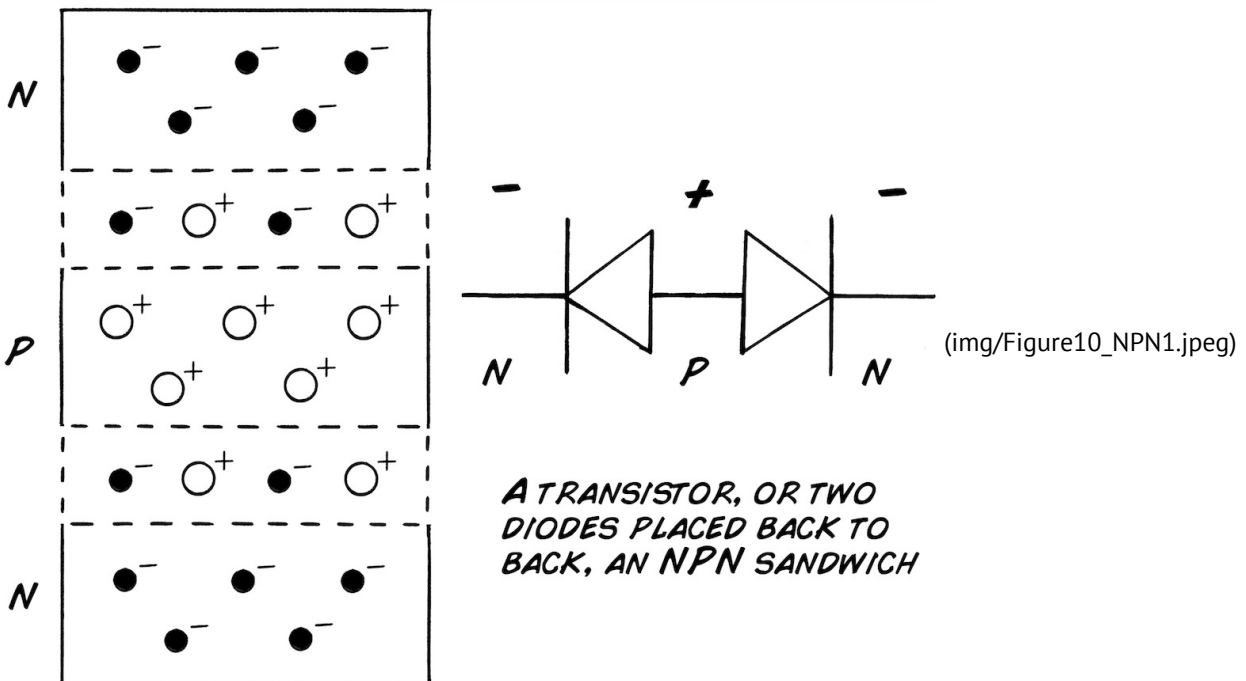


Some diodes release light with the current passes, hence the name "Light Emitting Diode" or "LED."

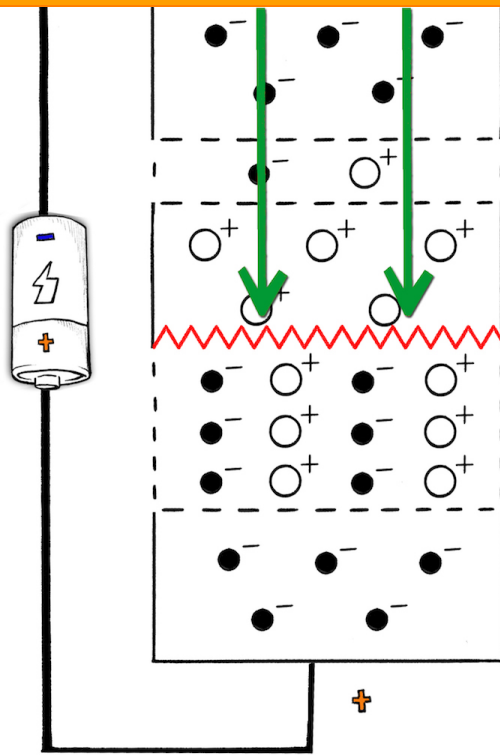


Transistor Theory

Now that we understand how p-n junctions and diodes work, what would happen if you made a "sandwich" with one block of p-doped material placed between two blocks of n-doped material?



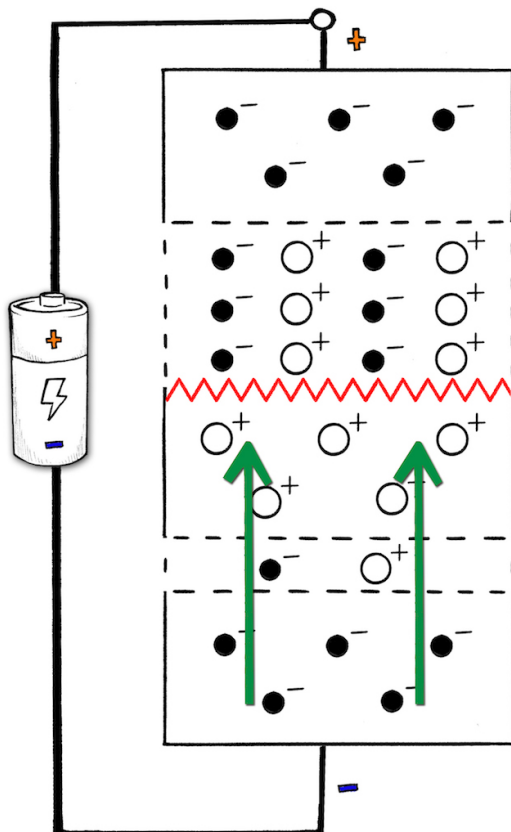
We now have a device with one "n-p" junction and one "p-n" junction that acts like two diodes* placed back to back. What would happen if you applied a large voltage across the whole sandwich?



BECAUSE THE APPLIED VOLTAGE MAKES ONE BARRIER LARGE AND THE OTHER SMALL, CURRENT STILL CANT PASS THROUGH THE TRANSISTOR

(img/fig9TransistorFlip_web.jpg)

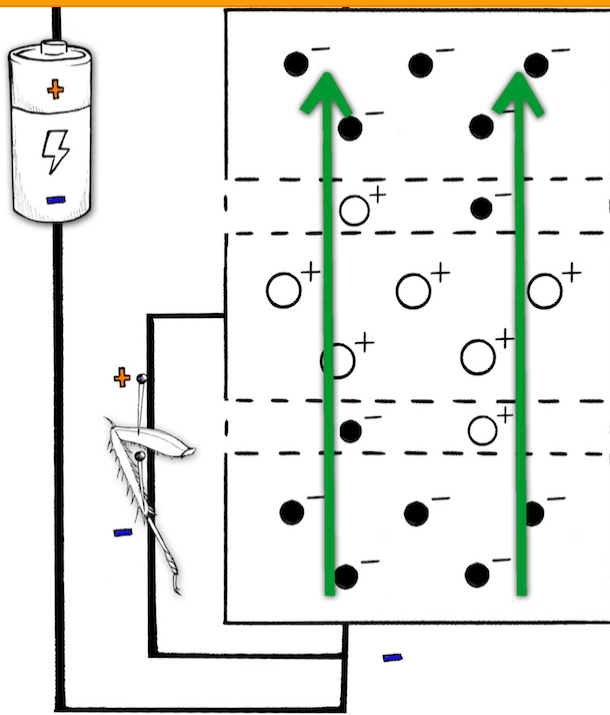
We didn't generate any current flow! What if we reversed the battery?



BECAUSE THE APPLIED VOLTAGE MAKES ONE BARRIER LARGE AND THE OTHER SMALL, CURRENT STILL CANT PASS THROUGH THE TRANSISTOR

(img/fig9_New_web.jpg)

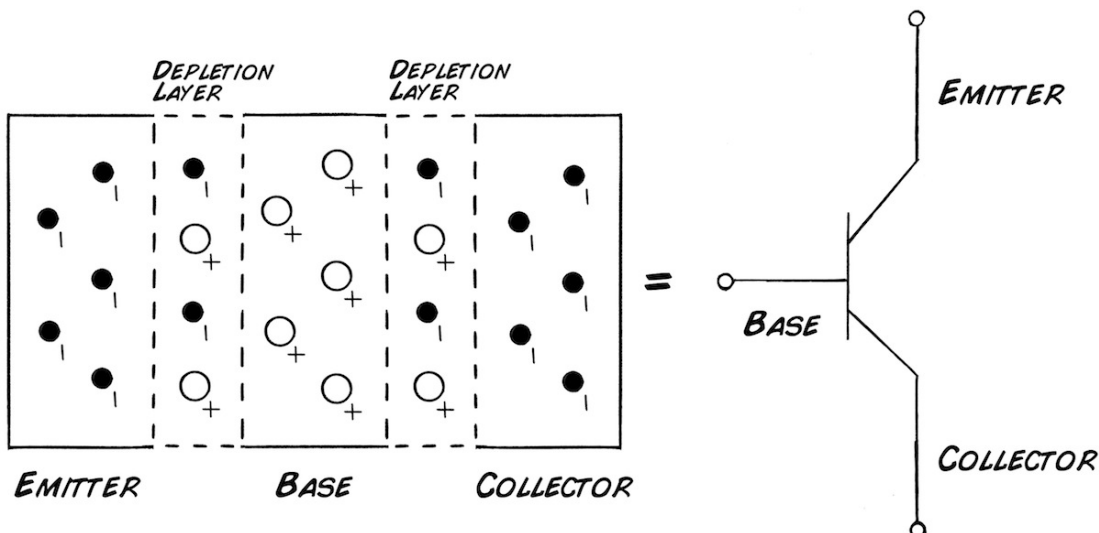
With the diodes back-to-back*, the applied voltage, no matter what direction it is in, will always reverse-bias one of the diodes and prevent current flow. But...hold on ... what if we add a smaller voltage to the terminal of the p-block? What then?



THE SMALLER ADDED VOLTAGE MAKES BOTH BARRIERS SMALL, (img/fig11NEW2withLeg_web.jpg) AND ALLOWS CURRENT TO FLOW THROUGH THE TRANSISTOR

The top n-p junction, which is reverse biased by the main battery voltage, prevents any current flow. But, **forward-biasing the lower p-n junction with the smaller voltage causes a huge number of electrons to shoot into the p-block.** This has the effect of lowering the current flow barriers in both depletion layers, and we get an exponential increase in electrons that can travel through the transistor. We've got current!

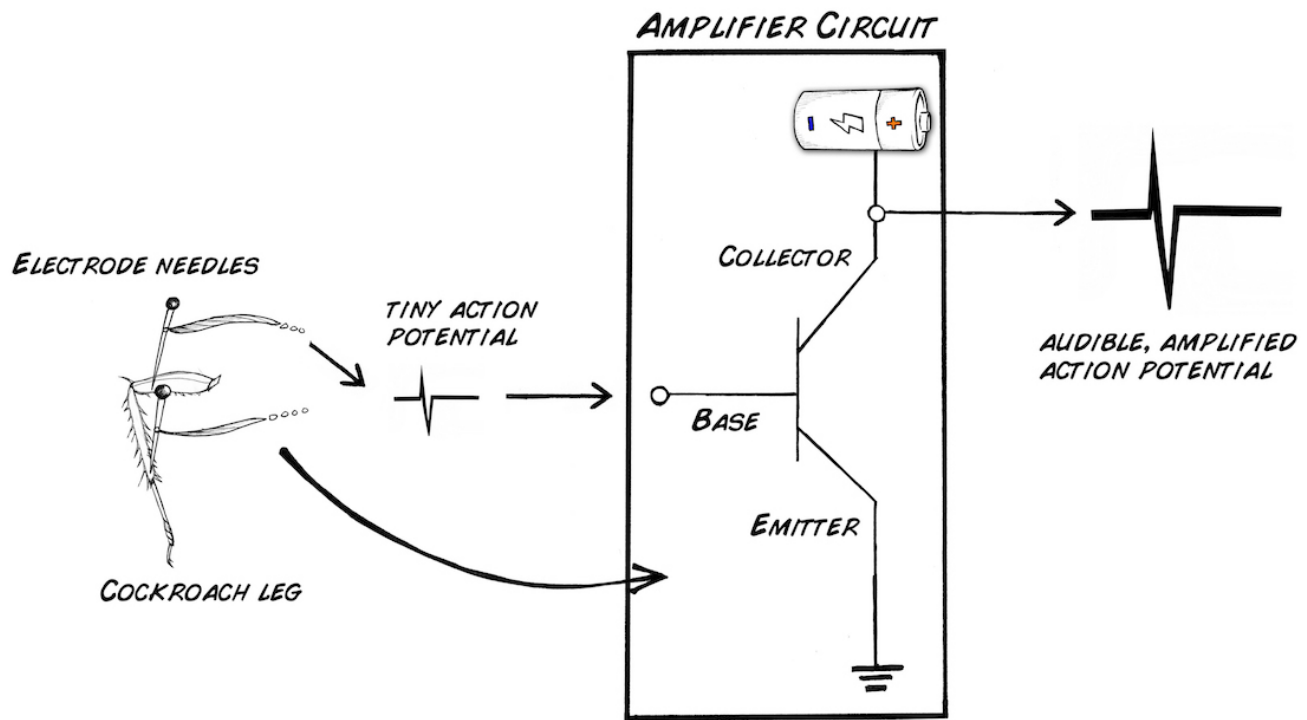
The "bipolar junction transistor" is the real-world component of this sandwich. There are two varieties, the "PNP" and the "NPN," but we will focus on the more common NPN configuration. In an NPN transistor, the three terminals are named the emitter (first N-block), the base (P-block), and the collector (second N-block).



(img/Figure12_NPN_to_BJT.jpeg)

Now we know how a transistor works, but why is its function important to our mission to learn about neural action potentials? A neural action potential has an extremely small voltage that needs to be amplified to be observed. If we set up our NPN transistor so that our little neural signal goes into the P-block (the base) and our large voltage (battery) going across the two n-blocks (collector and emitter), we have an amplifier! If we then monitor the current between the collector and emitter, we should see a signal that looks just like our action potential ...but a lot bigger!

enough that we can hear it through a simple speaker. And the world of neurons is ours to study.



(img/fig12ampNewLeg_web.jpg)

Now on to part II (<http://www.backyardbrains.com/experiments/transistorDesign>) and building your circuit...

*Note: Unfortunately, you can't simply buy two RadioShack diodes and place them back-to-back with solder to make a transistor. The effects described above happen at the crystal lattice level.

Discussion Questions

1. Why are the Group IV elements called "semi"-conductors? Do they conduct at all times? If not, what must be done to them to make them conduct?
2. What happens when Group III atoms are added into a block of Group IV material? What about when Group V atoms are added to Group IV instead? What do you think would happen if you just mixed Group III and Group V atoms? Do you think the resultant mix would be conductive?
3. In our temperature experiment (<http://www.backyardbrains.com/experiments/temperature>) we learned about the electro-chemical interactions that occur at the cellular membrane. How is a p-n junction similar to the lipid bilayer of a cell? How is it different?
4. How do diodes relate to transistors?

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