

Name: Madla, Celdrick Nheld S.

Course/Year/Section: BSIT 1A

1. Screen shot of your webpage

The Periodic Table

Alkali metals	Halogens																
Alkaline-earth metals	Noble gases																
Transition metals	Rare-earth elements (21, 39, 57-71) and lanthanoid elements (57-71 only)																
Other metals																	
Other nonmetals	Actinoid elements																
1 H	2 He																
3 Li	4 Be																
11 Na	12 Mg																
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
55 Cs	56 Ba	57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
87 Fr	88 Ra	89 Ac	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Me	116 Lv	117 Ts	118 Og

Type here to search

11:22 AM 3/29/2021

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55 Cs	56 Ba	57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
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58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu				
90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr				

Type here to search

11:22 AM 3/29/2021

2. Tags/ Codes used

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    <center>
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Ln 1, Col 1 100% Windows (CRLF) UTF-8
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Ln 1, Col 1 100% Windows (CRLF) UTF-8
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3. Image that you used as your reference in creating the activity.

The image shows a periodic table of elements with the following details:

- Periods:** 1, 2, 3, 4, 5, 6, 7
- Groups:** 1, 2, 13, 14, 15, 16, 17, 18
- Elements:** H, He, Li, Be, Na, Mg, K, Ca, Sc, Ti, V, Cr, Mn, Fe, Co, Ni, Cu, Zn, Ga, Ge, As, Se, Br, Kr, Rb, Sr, Y, Zr, Nb, Mo, Tc, Ru, Rh, Pd, Ag, Cd, In, Sn, Sb, Te, I, Xe, Cs, Ba, La, Hf, Ta, W, Re, Os, Ir, Pt, Au, Hg, Tl, Pb, Bi, Po, At, Rn, Fr, Ra, Ac, Rf, Db, Sg, Bh, Hs, Mt, Ds, Rg, Cn, Nh, Fl, Mc, Lv, Ts, Og
- Series:** Lanthanoid series (Ce-Lu), actinoid series (Th-Lr)

*Numbering system adopted by the International Union of Pure and Applied Chemistry (IUPAC). © Encyclopædia Britannica, Inc.