

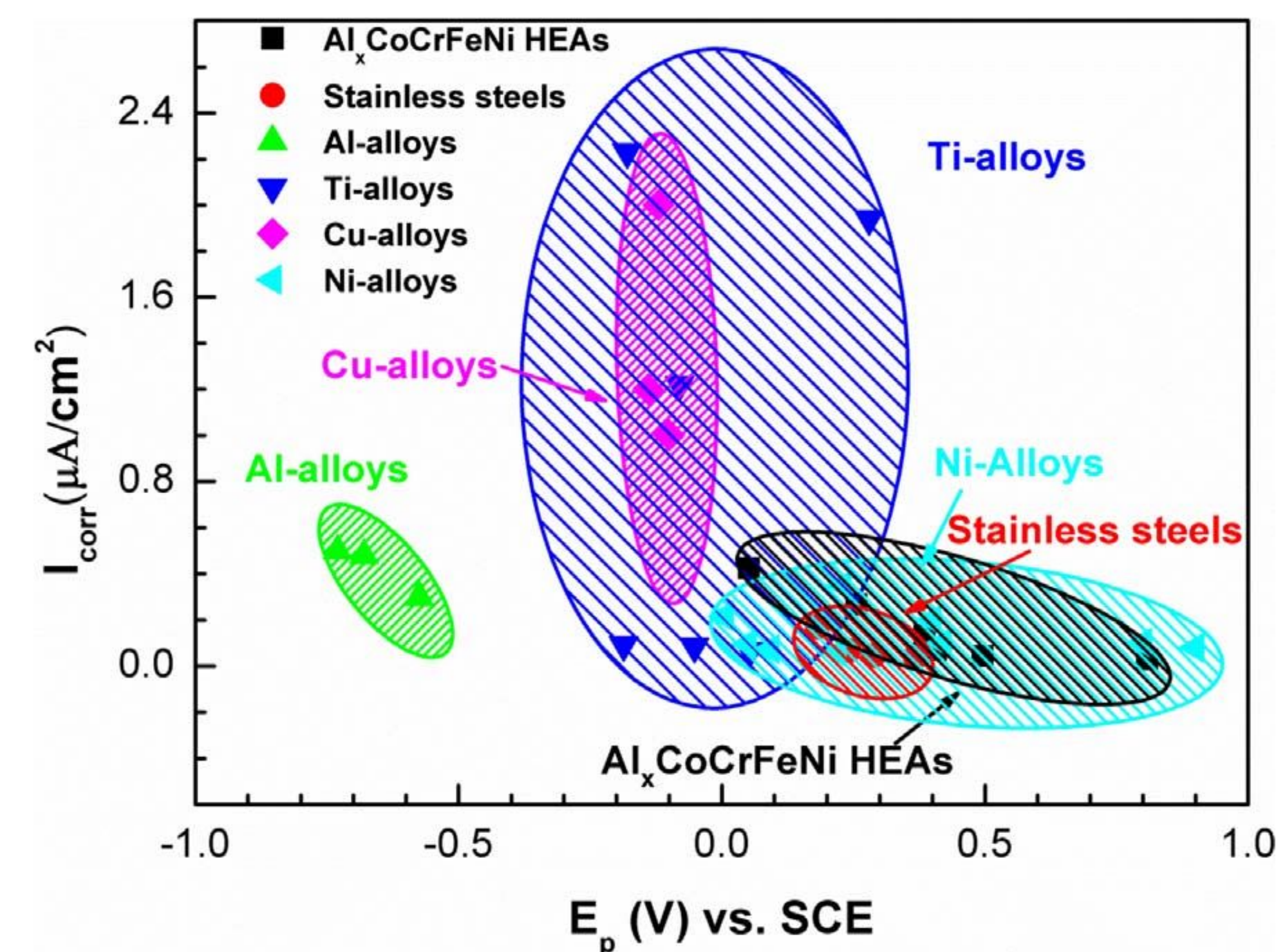
First-principles and Data-driven Discovery of High-entropy Alloys for Corrosion Protection



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Introduction

Corrosion performance by alloy types [1]



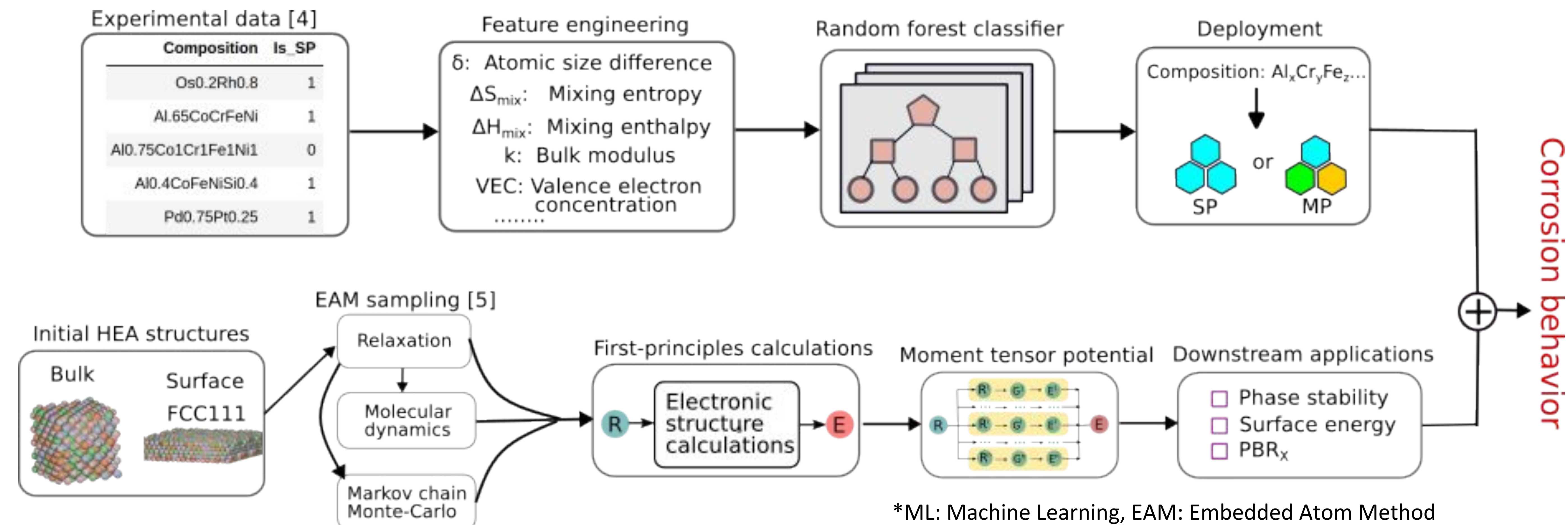
Background & Motivation

- Corrosion costs a lot.
- High-entropy alloys (HEA) are superior for corrosion protection.

Challenges

- HEAs live in high-dimensional search space.
- Unclear relation between compositions and corrosion behavior.

Workflow for ML-enabled discovery of corrosion-resistant HEA



→ Machine learning accelerated search of corrosion-resistant HEAs

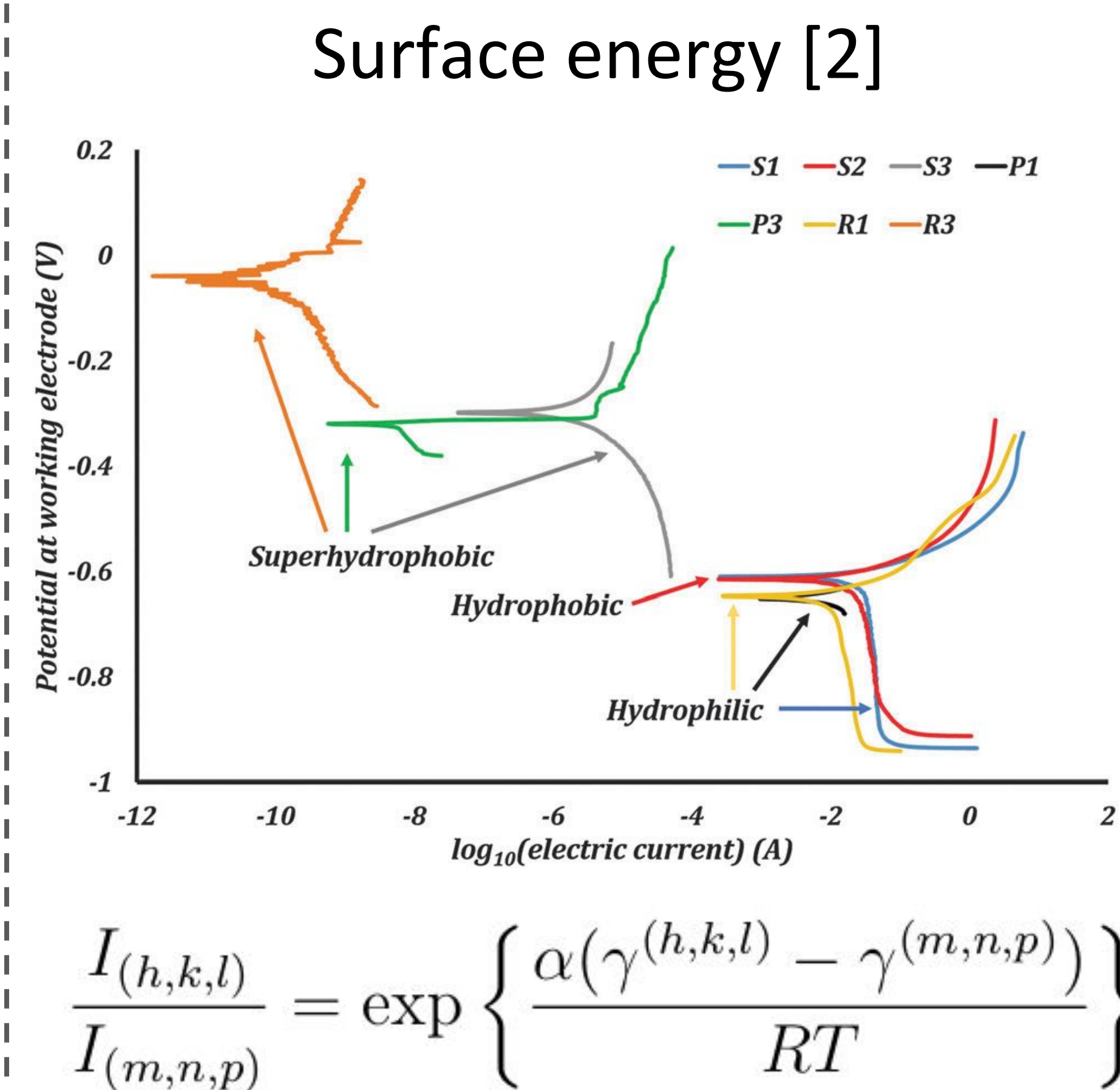
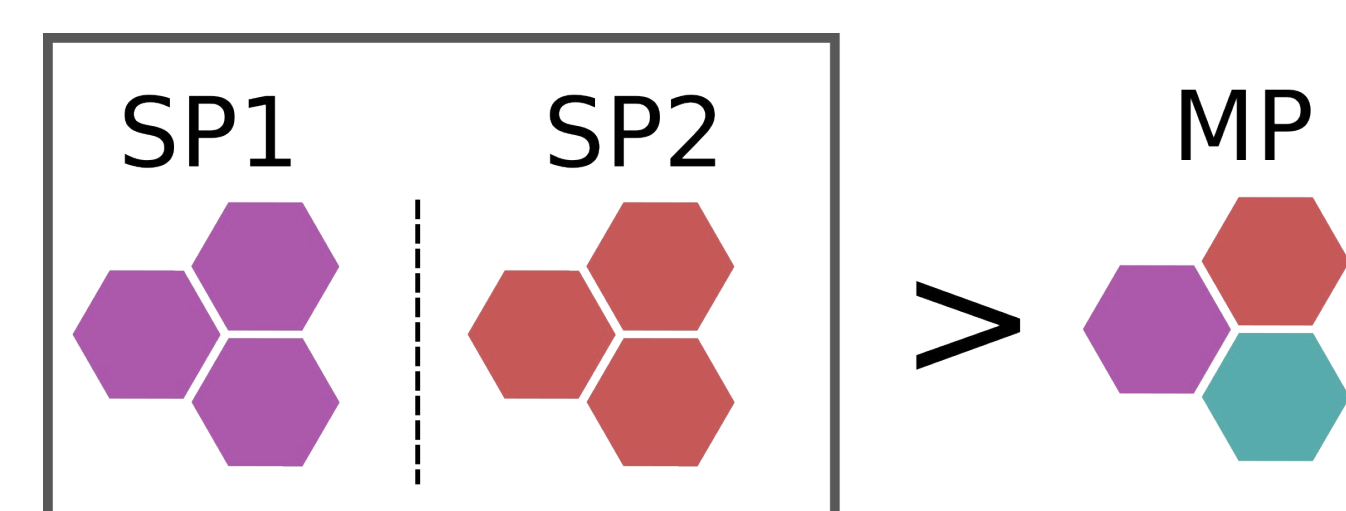
Three factors influencing corrosion

Synergistic effects of Al and Cr for $\text{Al}_x\text{Cr}_y\text{FeNiCo}$

Conclusions

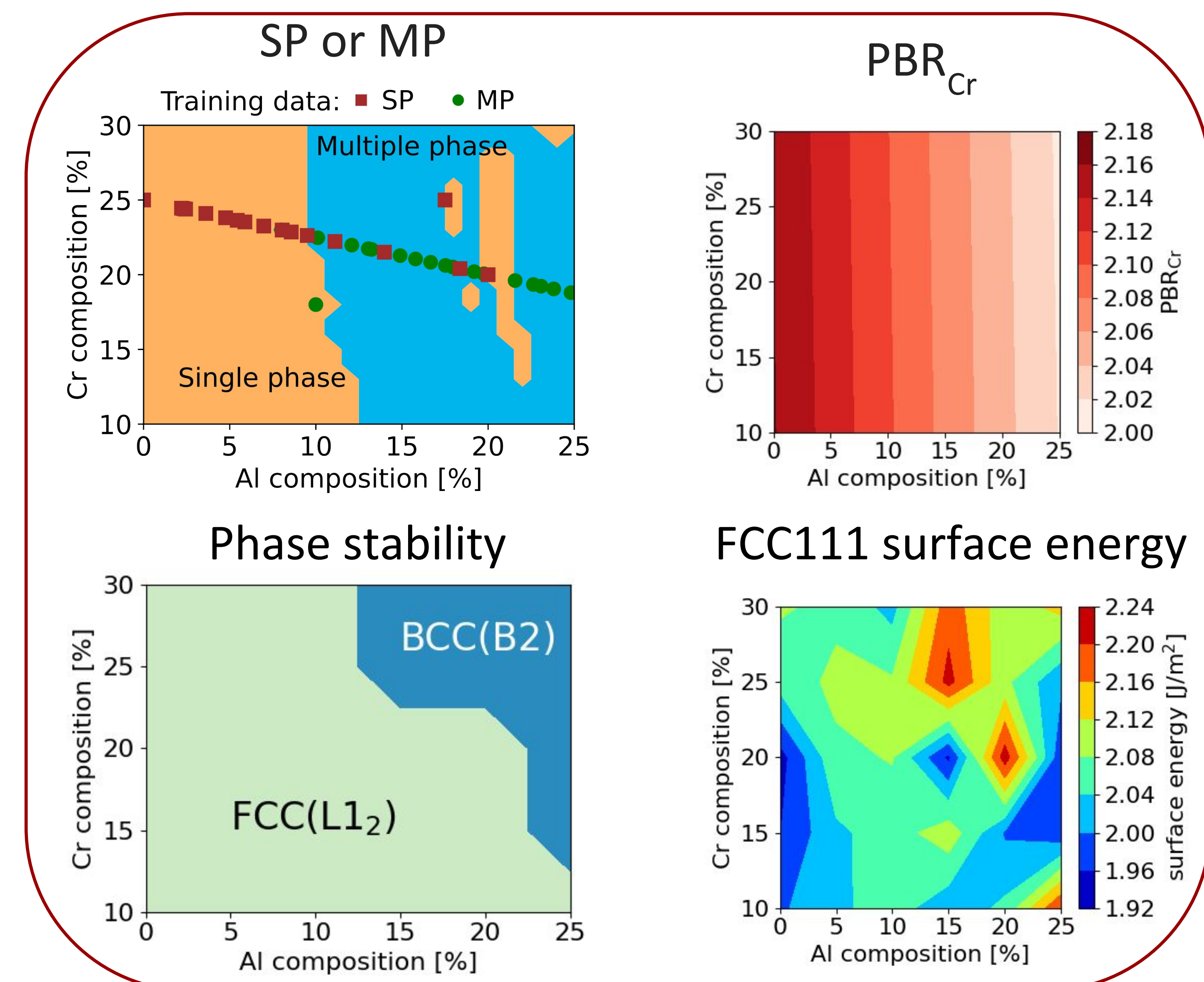
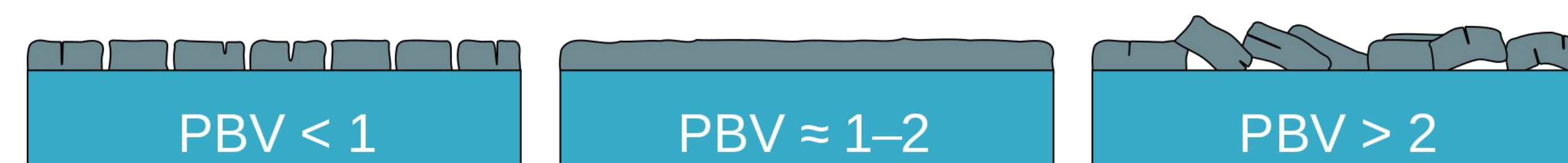
Phase type and stability

Single-phase (SP) is better than multiple-phase (MP)



Pilling-Bedworth Ratio (PBR) for oxidation of alloys [3]

$$PBR_{\text{alloy}} = \frac{\text{Volume of a mole of } B_xO_y}{\text{Volume of } x \text{ moles of } B \text{ in alloy}}$$



- A framework combining first-principles calculations and data-driven models to evaluate corrosion performance of HEAs is developed.
- Corrosion behavior is gauged in terms of SP formability, PBR and surface energy.
- SP formability is predicted by random forest classifier and the other properties are quantified by moment tensor potential.
- For $\text{Al}_x\text{Cr}_y\text{FeNiCo}$, the best Al and Cr composition ranges are around 0% and 15-18%.

References

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