ELEMENTS

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properties of chemical elements

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This package provides means for retrieving properties of chemical elements like atomic number, element symbol, element name, electron distribution or isotope number. Properties are defined for the elements up to the atomic number 112.

This package is a spin-off of the package bohr [Nie15] by the same author.

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1 Licence and Requirements

Permission is granted to copy, distribute and/or modify this software under the terms of the LATEX Project Public License (LPPL), version 1.3 or later (http://www.latex-project.org/lppl.txt). The software has the status "maintained."

ELEMENTS loads the packages etoolbox [Leh19] and translations [Nie17].

2 Element Names

 $\ensuremath{\mbox{\ensuremath}\ensuremath{\mbox{\ensuremath{\mbox{\ensuremath{\mbox{\ensuremath{\mbox{\ensuremath{\mbox{\ensuremath}\ensuremath{\mbox{\ensuremath}\ens$

Prints the element name of a given element as defined with \setatomname.

```
\setatomname[\langle alt. name \rangle] \{\langle atomic number \rangle\} \{\langle element name \rangle\}
```

Define or redefine the name of an element. If $\langle element\ name \rangle$ contains non-ascii symbols the optional argument $\langle alt.\ name \rangle$ must be given. In this case $\langle alt.\ name \rangle$ must be used in **ELEMENTS**' other macros where an element's name can be given as argument.

```
\DeclareAtomName[\langle alt. name \rangle] \{\langle atomic number \rangle\} \{\langle element name \rangle\}
```

This is the same as \setatomname but used before begin document or in packages/classes.

```
\saveelementname{\langle cs \rangle}{\langle atomic\ number \rangle | \langle element\ symbol \rangle | \langle element\ name \rangle} Saves the name of the given element as replacement text for the macro \langle cs \rangle.
```

```
1 \elementname{Cu} \par
2 \elementname{11} \par
3 \saveelementname\foo{28}
4 \ttfamily\meaning\foo

Copper
Sodium
macro:->Nickel
```

3 Element Symbols

```
\elementsymbol{\langle atomic number \rangle | \langle element name \rangle \rangle
```

Prints the element symbol of a given element as defined with \setatomsymbol.

```
\state{atomsymbol} {\langle atomic\ number \rangle} {\langle element\ symbol \rangle}
```

Define or redefine the symbol of an element.

```
\DeclareAtomSymbol{\langle atomic\ number\rangle}{\langle element\ symbol\rangle}
```

This is the same as \setatomsymbol but used before begin document or in packages/classes.

```
\saveelementsymbol{\langle cs \rangle}{\langle atomic\ number \rangle | \langle element\ symbol \rangle | \langle element\ name \rangle} Saves the symbol of the given element as replacement text for the macro \langle cs \rangle.
```

```
1 \elementsymbol{13} \par
2 \elementsymbol{Sulfur} \par
3 \saveelementsymbol\foo{83}
4 \ttfamily\meaning\foo
Al
S
macro:->Bi
```

4 Atomic Numbers

```
\atomicnumber{\langle element symbol \rangle | \langle element name \rangle }
```

Prints the atomic number of a given element.

```
\Z\{\langle element\ symbol\rangle \mid \langle element\ name\rangle\}
```

An alias of \atomicnumber but only defined at begin document and only if it isn't defined already.

 $\sin save a tomic number {\langle cs \rangle} {\langle atomic number \rangle | \langle element symbol \rangle | \langle element name \rangle}$

Saves the atomic number of the given element as replacement text for the macro $\langle cs \rangle$.

```
1 \atomicnumber{U} \par
2 \atomicnumber{Chlorine} \par
3 \saveatomicnumber\foo{Kr}
4 \ttfamily\meaning\foo

92
17
macro:->36
```

5 Electron Configuration

 $\ensuremath{\mbox{\ensuremath}\ensuremath{\mbox{\ensuremath{\mbox{\ensuremath}\ensuremath}\ensuremath}\ensuremath}\ensuremath}\engen}}}}}}}}} \end{subsite} \end{subsite} \end{subsite}\ensuremath{\mbox{\ensuremath}\$

Typesets the electron configuration of the given element.

```
\writeelconf{\langle electron distribution\rangle}
```

Typesets the electron distribution $\langle electron\ distribution \rangle$. The input is the same as described below for $\langle electron\ distribution \rangle$.

```
\stribution{\langle atomic\ number \rangle}{\langle electron\ distribution \rangle}
```

This set the electron distribution associated with the atom number $\langle atomic\ number \rangle$. $\langle electron\ distribution \rangle$ is a comma-separated list of the number of electrons placed on each shell from inner to outer shell. For example \setelectrondistribution{3}{2,0,1} would be an excited Lithium. The number of electrons with the same principal quantum number but a different angular quantum number are separated with a + ordered by the angular quantum number, *i. e.*, first *s*, then *p*, then *d*, and then *f* . Copper's distribution would be declared like this:

\setelectrondistribution{29}{2,2+6,2+6+10,1}.

A declaration with \setelectrondistribution{29}{2,8,18,1} would work but then \elconf {29} would give the wrong results.

\DeclareElectronDistribution

This is the same as \setelectrondistribution but used before begin document or in packages/classes.

```
\printangular momentum {\langle angular momentum \rangle}
```

Introduced in version 0.1e

This command is used be **\elconf** and **\writeelconf** for typesetting s, p, d, ... You can redefine it if you want to change how those functions are typeset (keeping in mind that it will be used in *math mode*). The initial definition is:

\newcommand*\printangularmomentum[1]{\mathrm{#1}}

There is currently *no way* to get the electron configuration in the shortened way (*e. g.*: $[Ar]d^{10}4s^{1}$).

6 Isotope Lists

```
\statement{setatomisotopes}{\langle atomic\ number \rangle}{\langle isotope\ list \rangle}
```

Defines or redefines the isotope list for a given element. (*isotope list*) should be a comma seperated list of integers. One of the integers may be preceded with a! to mark the main isotope for the given element: \setatomisotopes{6}{10,11,!12,13,14,15,16}

```
\DeclareAtomIsotopes{\langle atomic number \rangle}{\langle isotope list \rangle}
```

This is the same as \setatomisotopes but used before begin document or in packages/classes.

```
\saveelementisotopes{\langle cs \rangle}{\langle atomic\ number \rangle|\langle element\ symbol \rangle|\langle element\ name \rangle} Saves the isotope list of the given element as replacement text for the macro \langle cs \rangle.
```

\savemainelementisotope{ $\langle cs \rangle$ }{ $\langle atomic\ number \rangle$ | $\langle element\ symbol \rangle$ | $\langle element\ name \rangle$ } Saves the main isotope of the given element as replacement text for the macro $\langle cs \rangle$. If the isotope list of the element contains no main isotope $\langle cs \rangle$ will be equivalent to \@empty.

```
\mbox{\constrainelementisotope} \{ \langle atomic\ number \rangle \mid \langle element\ symbol \rangle \mid \langle element\ name \rangle \}
```

Introduced in version 0.1d

Print the main isotope of the given element. If the isotope list of the element contains no main isotope nothing will be printed.

7 Supported Languages

Currently the following languages are supported:

- English
- German
- French
- Spanish
- Portuguese

Introduced in version 0.2 Introduced in version 0.2d

Element Properties

On the following pages a table containing the properties known to **ELEMENTS** is printed. For those interested: the code used to get the table is as follows (using the packages Iscape [Caroo], longtable [Car19] and booktabs [Fea16]).

```
1 \setlength\LTleft\fill
2 \setlength\LTright\fill
3 \newcounter{element}
4 \setcounter{element}{1}
5 \begin{landscape}
6 \begin{longtable}{lllll}
      Number & Symbol & Name & Main Isotope & Electron Configuration \\
    \midrule
    \endhead
10
    \whileboolexpr{test{\ifnumless{\value{element}}{113}}}
11
12
        \theelement &
13
        \elementsymbol{\arabic{element}} &
14
        \elementname{\arabic{element}} &
        \mainelementisotope{\arabic{element}} &
        \elconf{\arabic{element}}
        \stepcounter{element} \\
18
      }
      {}
21 \end{longtable}
22 \end{landscape}
```

Electron Configuration	$1s^1$	$1s^2$	$1s^{2}2s^{1}$	$1s^22s^2$	$1s^2 2s^2 2p^1$	$1s^2 2s^2 2p^2$	$1s^2 2s^2 2p^3$	$1s^2 2s^2 2p^4$	$1s^2 2s^2 2p^5$	$1s^2 2s^2 2p^6$	$1s^2 2s^2 2p^6 3s^1$	$1s^2 2s^2 2p^6 3s^2$	$1s^2 2s^2 2p^6 3s^2 3p^1$	$1s^2 2s^2 2p^6 3s^2 3p^2$	$1s^2 2s^2 2p^6 3s^2 3p^3$	$1s^2 2s^2 2p^6 3s^2 3p^4$	$1s^2 2s^2 2p^6 3s^2 3p^5$	$1s^2 2s^2 2p^6 3s^2 3p^6$	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^1$	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^1 4s^2$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^2 4s^2$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^3 4s^2$	$1s^2 2s^2 2\bar{p}^6 3s^2 3\bar{p}^6 3d^5 4s^1$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^5 4s^2$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^6 4s^2$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^7 4s^2$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^8 4s^2$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^1$
Main Isotope	1	4	7	6	11	12	14	16	19	20	23	24	27	28	31	32	35	40	39	40	45	48	51	52	55	56	59	58	63
Name	Hydrogen	Helium	Lithium	Beryllium	Boron	Carbon	Nitrogen	Oxygen	Fluorine	Neon	Sodium	Magnesium	Aluminium	Silicon	Phosphorus	Sulfur	Chlorine	Argon	Potassium	Calcium	Scandium	Titanium	Vanadium	Chromium	Manganese	Iron	Cobalt	Nickel	Copper
Symbol	Н	He	Li	Be	В	С	Z	0	냄	Ne	Na	Mg	Al	Si	Ь	S	Cl	Ar	K	Ca	Sc	Ţi	>	Cr	Mn	Fe	Co	Ņ.	Cu
Number	1	2	3	4	2	9	7	8	6	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29

loqu		Main Isotope	Electron Configuration
Zn Zinc 64	4 0		$15^225^229^935^239^934^{10}45^2$
	Ž 4		$18^2 28^2 20^6 38^2 30^6 34^{10} 48^2 40^2$
Arsenic	ŕς		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^3$
Se Selenium 80	0		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^4$
3r Bromine 79	6.		$1s^22s^22p^63s^23p^63d^{10}4s^24p^5$
Kr Krypton 84	4		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6$
Rubidium 85	5		$1s^22s^22p^63s^23p^63d^{10}4s^24p^65s^1$
Sr Strontium 88	∞		$1s^22s^22p^63s^23p^63d^{10}4s^24p^65s^2$
Y Yttrium 89	6		$1s^22s^22p^63s^23p^63d^{10}4s^24p^64d^15s^2$
Zr Zirconium 90	0		$1s^22s^22p^63s^23p^63d^{10}4s^24p^64d^25s^2$
Nb Niobium 93	3		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^4 5s^1$
Mo Molybdenum 98	∞		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^5 5s^1$
Fc Technetium 99	6		$1s^22s^22p^63s^23p^63d^{10}4s^24p^64d^65s^1$
Ruthenium 102	02		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^7 5s^1$
Rhodium 103	03		$1s^22s^22p^63s^23p^63d^{10}4s^24p^64d^85s^1$
9d Palladium 106	90		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10}$
Ag Silver 107	07		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 5s^1$
Cd Cadmium 114	14		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 5s^2$
n Indium 115	15		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 5s^2 5p^1$
Sn Tin 120	20		$1\mathrm{s}^{2}2\mathrm{s}^{2}2\mathrm{p}^{6}3\mathrm{s}^{2}3\mathrm{p}^{6}3\mathrm{d}^{10}4\mathrm{s}^{2}4\mathrm{p}^{6}4\mathrm{d}^{10}5\mathrm{s}^{2}5\mathrm{p}^{2}$
Sb Antimony 121	21		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 5s^2 5p^3$
Fe Tellurium 130	30		$1s^22s^22p^63s^23p^63d^{10}4s^24p^64d^{10}5s^25p^4$
Iodine 127	27		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 5s^2 5p^5$
Xe Xenon 132	32		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 5s^2 5p^6$
Cs Caesium 133	33		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 5s^2 5p^6 6s^1$
3a Barium 138	38		$1s^22s^22p^63s^23p^63d^{10}4s^24p^64d^{10}5s^25p^66s^2$
Lanthanum	39		$1s^22s^22p^63s^23p^63d^{10}4s^24p^64d^{10}5s^25p^65d^16s^2$
Ce Cerium 140	40		$1s^22s^22p^63s^23p^63d^{10}4s^24p^64d^{10}4f^25s^25p^66s^2$

Electron Configuration	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^3 5s^2 5p^6 6s^2$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^4 5s^2 5p^6 6s^2$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^5 5s^2 5p^6 6s^2$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^6 5s^2 5p^6 6s^2$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^7 5s^2 5p^6 6s^2$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^7 5s^2 5p^6 5d^1 6s^2$	$1s^22s^22p^63s^23p^63d^{10}4s^24p^64d^{10}4f^95s^25p^66s^2$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{10} 5s^2 5p^6 6s^2$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{11} 5s^2 5p^6 6s^2$	$1s^22s^22p^63s^23p^63d^{10}4s^24p^64d^{10}4f^{12}5s^25p^66s^2$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{13} 5s^2 5p^6 6s^2$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 6s^2$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^1 6s^2$	$1s^22s^22p^63s^23p^63d^{10}4s^24p^64d^{10}4f^{14}5s^25p^65d^26s^2$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^3 6s^2$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^4 6s^2$	$1s^22s^22p^63s^23p^63d^{10}4s^24p^64d^{10}4f^{14}5s^25p^65d^56s^2$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^6 6s^2$	$1s^22s^22p^63s^23p^63d^{10}4s^24p^64d^{10}4f^{14}5s^25p^65d^76s^2$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^9 6s^1$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 6s^1$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 6s^2$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 6s^2 6p^1$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 6s^2 6p^2$	$1s^22s^22p^63s^23p^63d^{10}4s^24p^64d^{10}4f^{14}5s^25p^65d^{10}6s^26p^3$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 6s^2 6p^4$	$1s^22s^22p^63s^23p^63d^{10}4s^24p^64d^{10}4f^{14}5s^25p^65d^{10}6s^26p^5$	$1s^22s^22p^63s^23p^63d^{10}4s^24p^64d^{10}4f^{14}5s^25p^65d^{10}6s^26p^6$	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 6s^2 6p^6 7s^1$
Main Isotope	141	142	147	152	153	158	159	164	165	166	169	174	175	180	181	184	187	192	193	195	197	202	205	208	209	210		222	223
Name	Praseodymium	Neodymium	Promethium	Samarium	Europium	Gadolinium	Terbium	Dysprosium	Holmium	Erbium	Thulium	Ytterbium	Lutetium	Hafnium	Tantalium	Tungsten	Rhenium	Osmium	Iridium	Platinum	Gold	Mercury	Thallium	Lead	Bismuth	Polonium	Astatine	Radon	Francium
Symbol	Pr	pN	Pm	Sm	Eu	PS	Tb	Dy	Ho	Er	Tm	Yb	Lu	Hf	Ta	M	Re	Os	Ir	Pt	Au	Hg	I	Pb	Bi	Po	At	Rn	Fr
Number	59	09	61	62	63	64	65	99	29	89	69	70	71	72	73	74	75	9/	77	78	79	80	81	82	83	84	85	98	87

Number	Symbol	Name	Main Isotope	Electron Configuration
88	Ra	Radium	226	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 6s^2 6p^6 7s^2$
89	Ac	Actinium	227	$1s^2 2s^2 2p^6 3s^2 3p^5 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^5 5d^{10} 6s^2 6p^6 6d^1 7s^2$
90	Th	Thorium	232	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 6s^2 6p^6 6d^2 7s^2$
91	Pa	Protactinium	231	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 5f^2 6s^2 6p^6 6d^1 7s^2$
92	Ω	Uranium	238	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 5f^3 6s^2 6p^6 6d^1 7s^2$
93	dN	Neptunium		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 5f^4 6s^2 6p^6 6d^1 7s^2$
94	Pu	Plutonium	244	$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 5f^6 6s^2 6p^6 7s^2$
95	Am	Americium		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 5f^7 6s^2 6p^6 7s^2$
96	Cm	Curium		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 5f^7 6s^2 6p^6 6d^1 7s^2$
26	Bk	Berkelium		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 5f^9 6s^2 6p^6 7s^2$
98	Cf	Californium		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 5f^{10} 6s^2 6p^6 7s^2$
66	Es	Einsteinium		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 5f^{11} 6s^2 6p^6 7s^2$
100	Fm	Fermium		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 5f^{12} 6s^2 6p^6 7s^2$
101	Md	Mendelevium		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 5f^{13} 6s^2 6p^6 7s^2$
102	No	Nobelium		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 5f^{14} 6s^2 6p^6 7s^2$
103	Lr	Lawrencium		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 5f^{14} 6s^2 6p^6 6d^1 7s^2$
104	Rf	Rutherfordium		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 5f^{14} 6s^2 6p^6 6d^2 7s^2$
105	Db	Dubnium		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 5f^{14} 6s^2 6p^6 6d^3 7s^2$
106	Sg	Seaborgium		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 5f^{14} 6s^2 6p^6 6d^4 7s^2$
107	Bh	Bohrium		$1s^2 2s^2 2p^6 3s^2 3p^5 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^5 5d^{10} 5f^{14} 6s^2 6p^6 6d^5 7s^2$
108	Hs	Hassium		$1s^2 2s^2 2p^6 3s^2 3p^5 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^5 5d^{10} 5f^{14} 6s^2 6p^6 6d^6 7s^2$
109	Mt	Meitnerium		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 5f^{14} 6s^2 6p^6 6d^7 7s^2$
110	Ds	Darmstadtium		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 5f^{14} 6s^2 6p^6 6d^9 7s^1$
111	Rg	Roentgenium		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 5f^{14} 6s^2 6p^6 6d^{10} 7s^1$
112	Cn	Copernicium		$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 4f^{14} 5s^2 5p^6 5d^{10} 5f^{14} 6s^2 6p^6 6d^{10} 7s^2$

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