

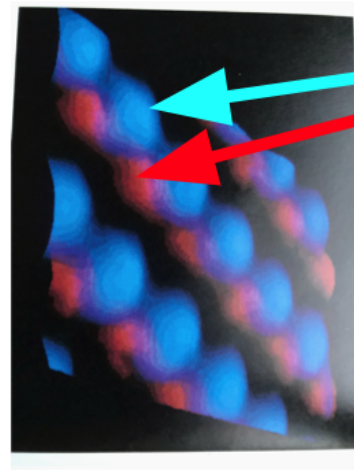
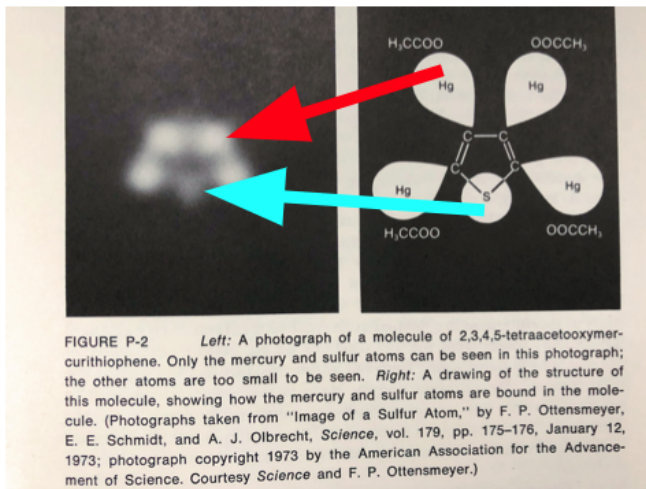
See Chandru Grading Comments from 11/4/23 below

1. Nice work!

✓ You have a 100% grade on this notebook

✓ Correct Any Errors identified above for an improved grade

TODAYS LAB - MEASURE THE DIAMETER OF AN ATOM



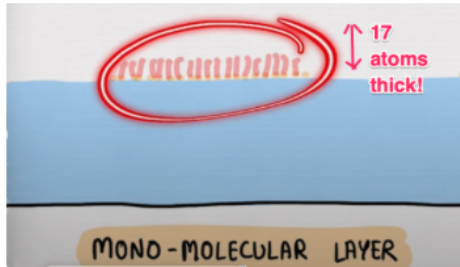
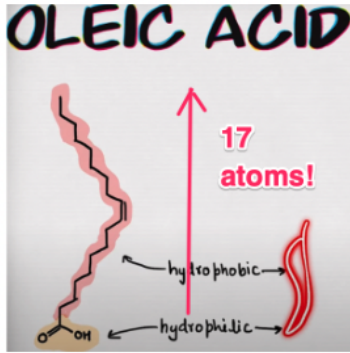
Gallium
Arsenide
Atoms!

OBSERVED COLORS OF STARS, NEBULAE AND OTHER OBJECTS!

n	Energy (J)	Energy (eV)	
6	-6.05×10^{-20}	-0.377	
5	-8.70×10^{-20}	-0.544	Violet (7.3×10^{14} Hz)
4	-1.36×10^{-19}	-0.850	Violet (6.9×10^{14} Hz)
3	-2.42×10^{-19}	-1.51	Blue-green (6.2×10^{14} Hz)
2	-5.44×10^{-19}	-3.40	Red (4.6×10^{14} Hz)
1	-2.18×10^{-18}	-13.6	



PROCESS TO MEASURE THE DIAMETER OF AN ATOM USING OLEIC ACID



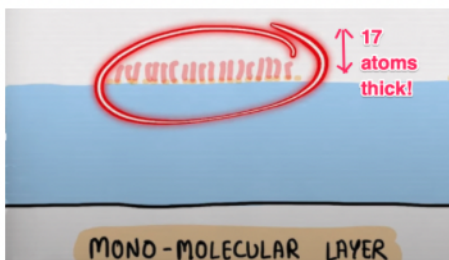
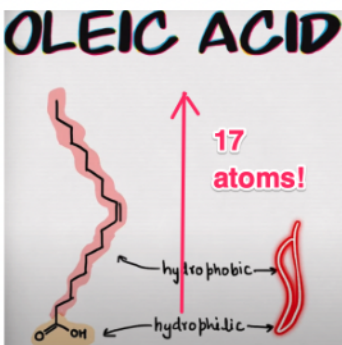
- Oleic acid is a simple molecule with hydrophilic and hydrophobic parts!
- Prepare a 1/400 solution of Oleic acid in alcohol
- Calculate the volume of Oleic acid in droplet of the this solution
- Create a mono-molecular layer of this solution floating on water
- Measure its diameter and calculate its thickness knowing the volume of the Oleic acid in the droplet
- The thickness is the diameter of an atom!!

1. Make a 1/400 Dilute solution of Oleic Acid

picture of your scetch and # of drops

amount_of_oleicacid_in_one_drop_of_soln: 4.6296e-05 ml

PROCESS TO MEASURE THE DIAMETER OF AN ATOM USING OLEIC ACID



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- The thickness is the diameter of an atom!!

diameter_mono_molecular_layer: 1.800e-01 m

oleic_acid_in_one_drop_of_solution: 4.630e-05 ml
oleic_acid_in_one_drop_of_solution: 4.630e-11 m³

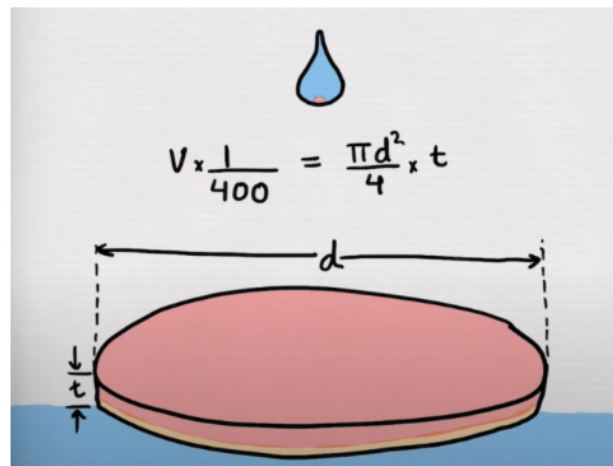
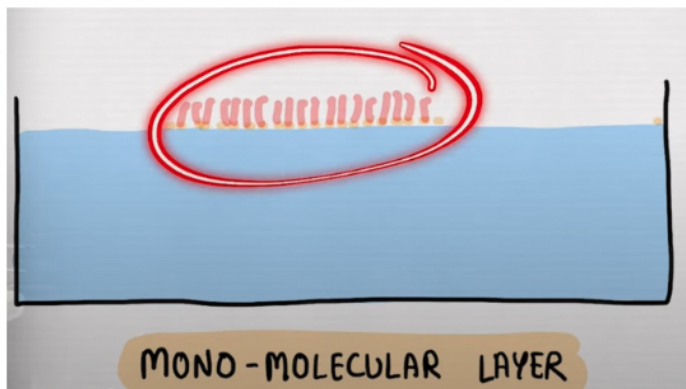
2. Calculate the area of the Mono Molecular Layer Disk

area_of_mono_molecular_disk: 2.545e-02 m²

3. Calculate the Thickness of the Mono Molecular Disk.

This is the Volume of the Oleic Acid in one droplet divided by the Area of the Mono Molecular Disk!

thickness_of_mono_molecular_layer: 1.819e-09 m
no_of_carbon_atoms_in_a_oleic_acid_molecule: 17



7. Calculate the Diameter of Carbon Atom !!

diameter_of_a_single_carbon_atom: 1.070e-10 m

[Hyperphysics Carbon Atom Diameter](#) is 0.22nm or

$$= 2.2 \times 10^{-10} \text{ m}$$

Error in experimental measurement of the diameter of a single carbon atom: 51.35%

REALLY ??? !!!