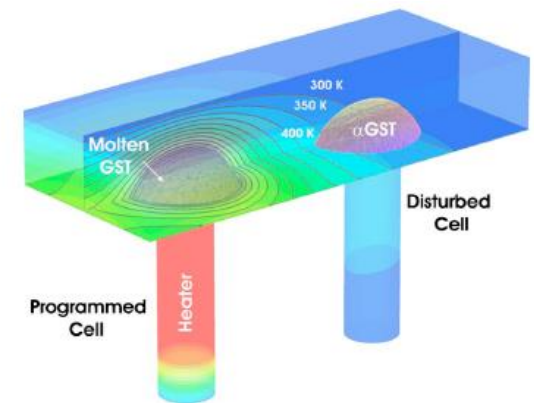
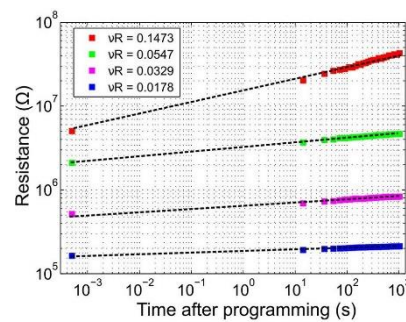
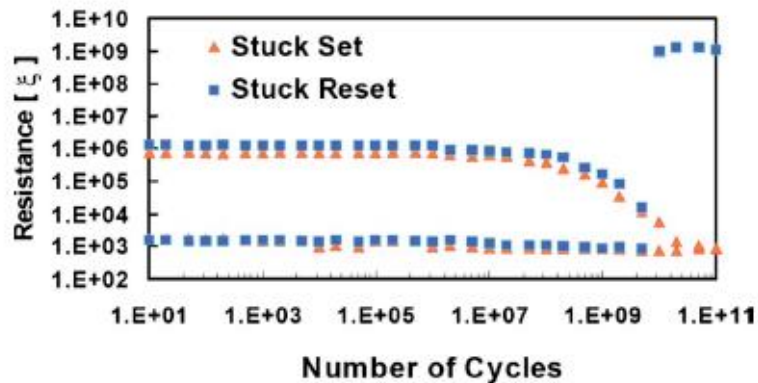
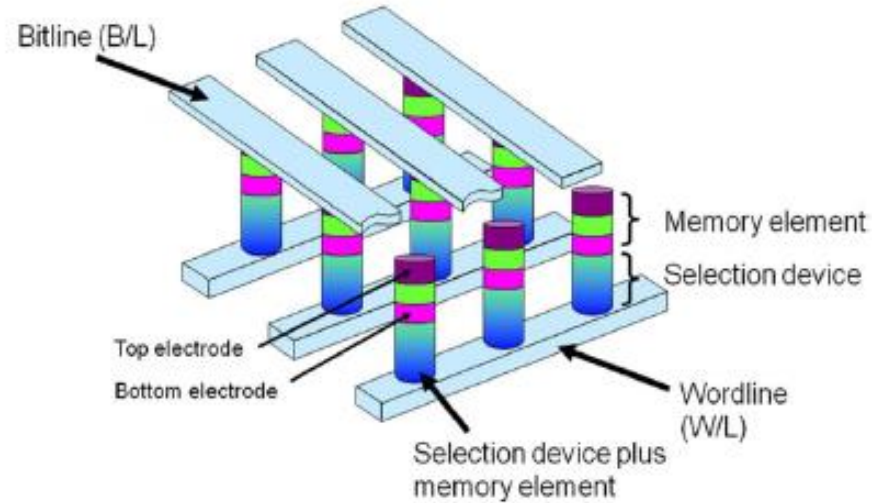
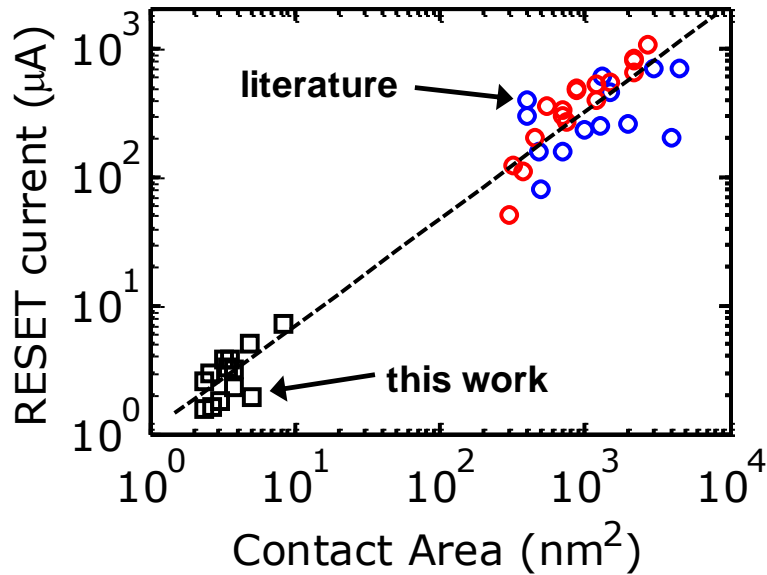


# **L5: Resistive RAM I**

# Recap

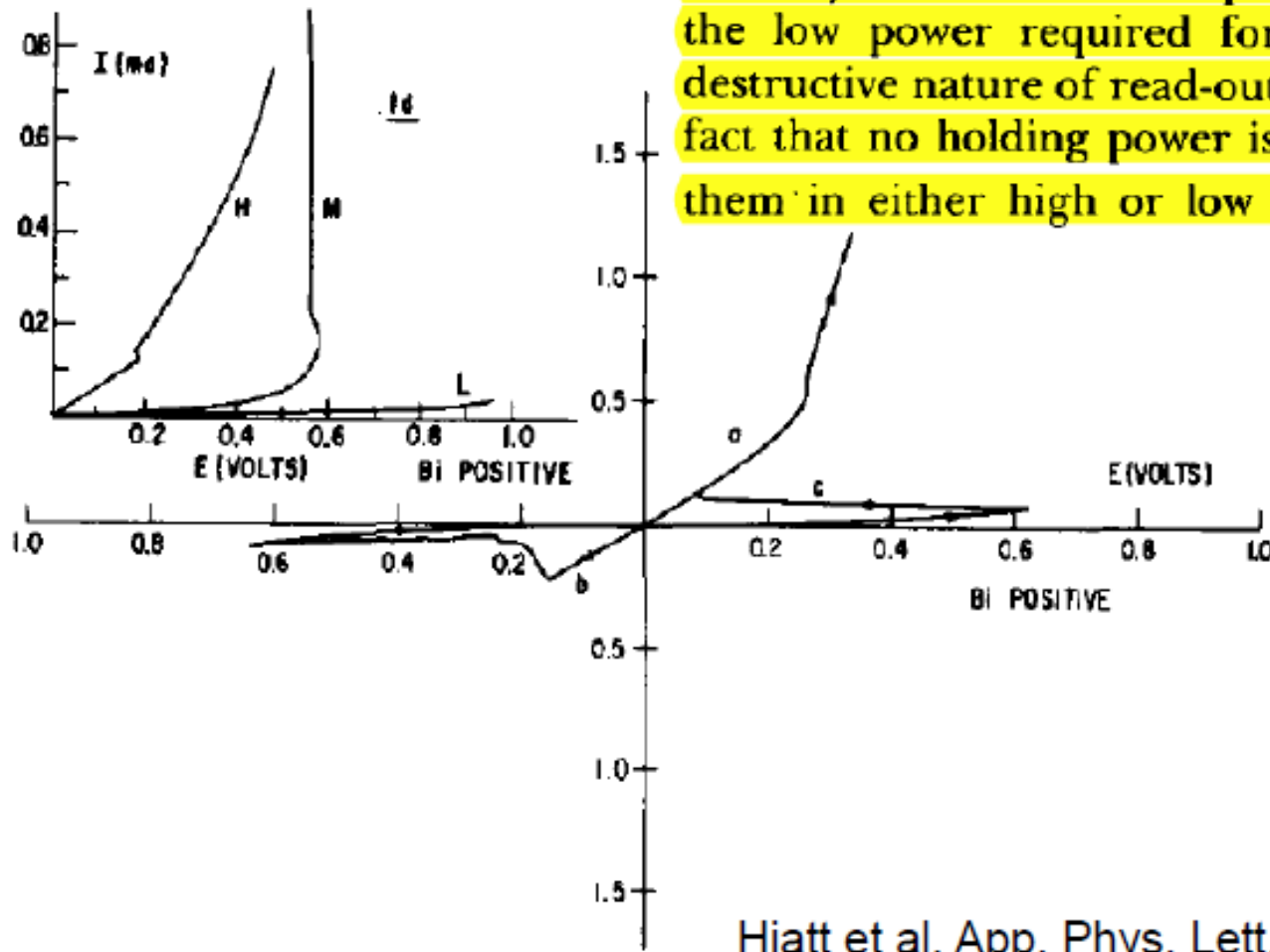


# Outline

- Introduction
  - History
  - pros and cons
  - device structure
- Unipolar vs. bipolar
- Switching mechanisms
- Cation switching
- Anion switching
- Forming, set and reset

# History of RRAM

The possible use of Nb-Nb<sub>2</sub>O<sub>5</sub>-metal diodes, particularly Nb-Nb<sub>2</sub>O<sub>5</sub>-Bi diodes, in a computer memory is obvious. Their potential advantages are the low power required for switching, the non-destructive nature of read-out of the device and the fact that no holding power is required to maintain them in either high or low conductivity state. At

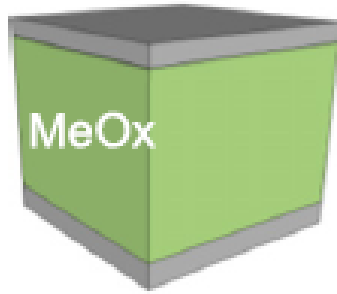


Hiatt et al. App. Phys. Lett. 6, 106, 1965

# Resistive Switching

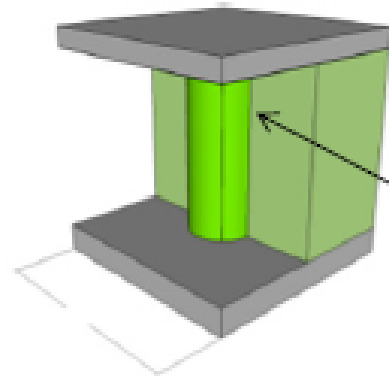
Top electrode (TE)

D. Ielmini et al., Semicon. Sci. & Tech., 2016

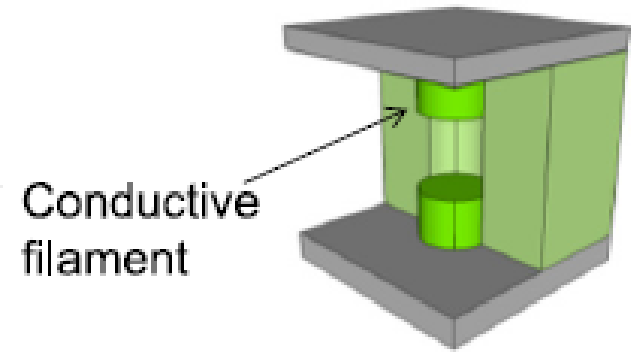


Bottom electrode (TE)

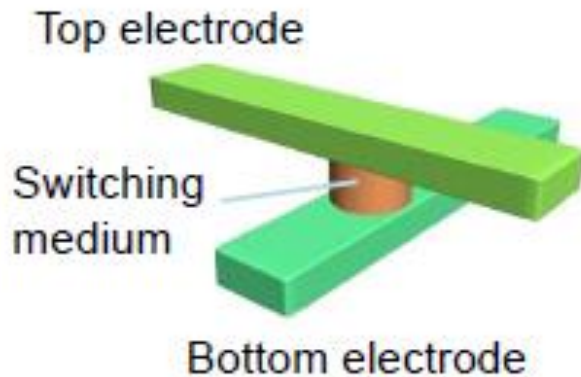
(a) Initial state



(b) Set state



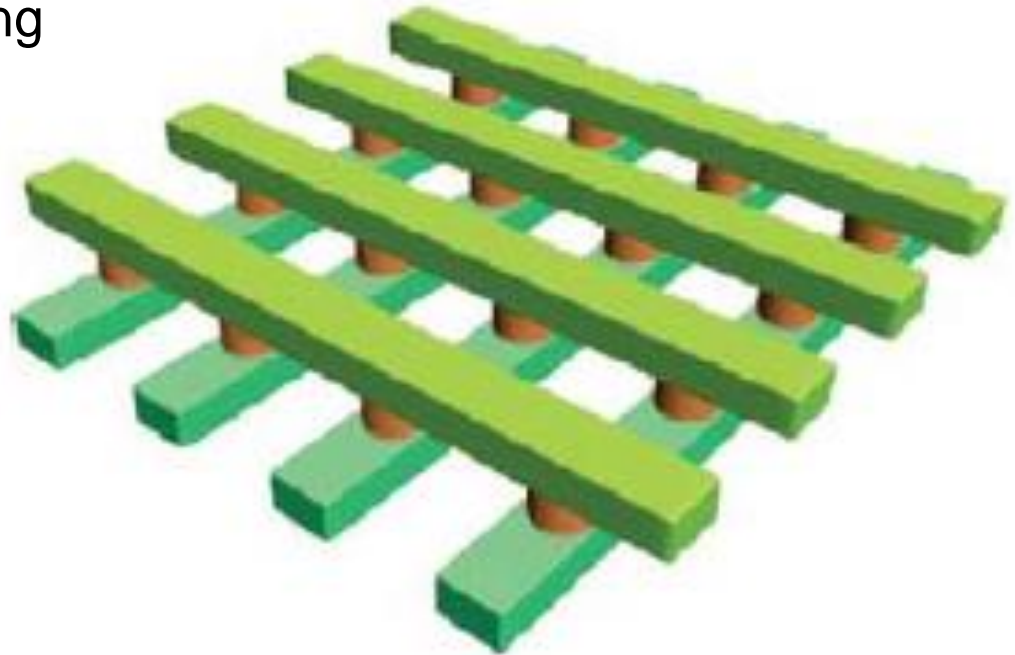
(c) Reset state



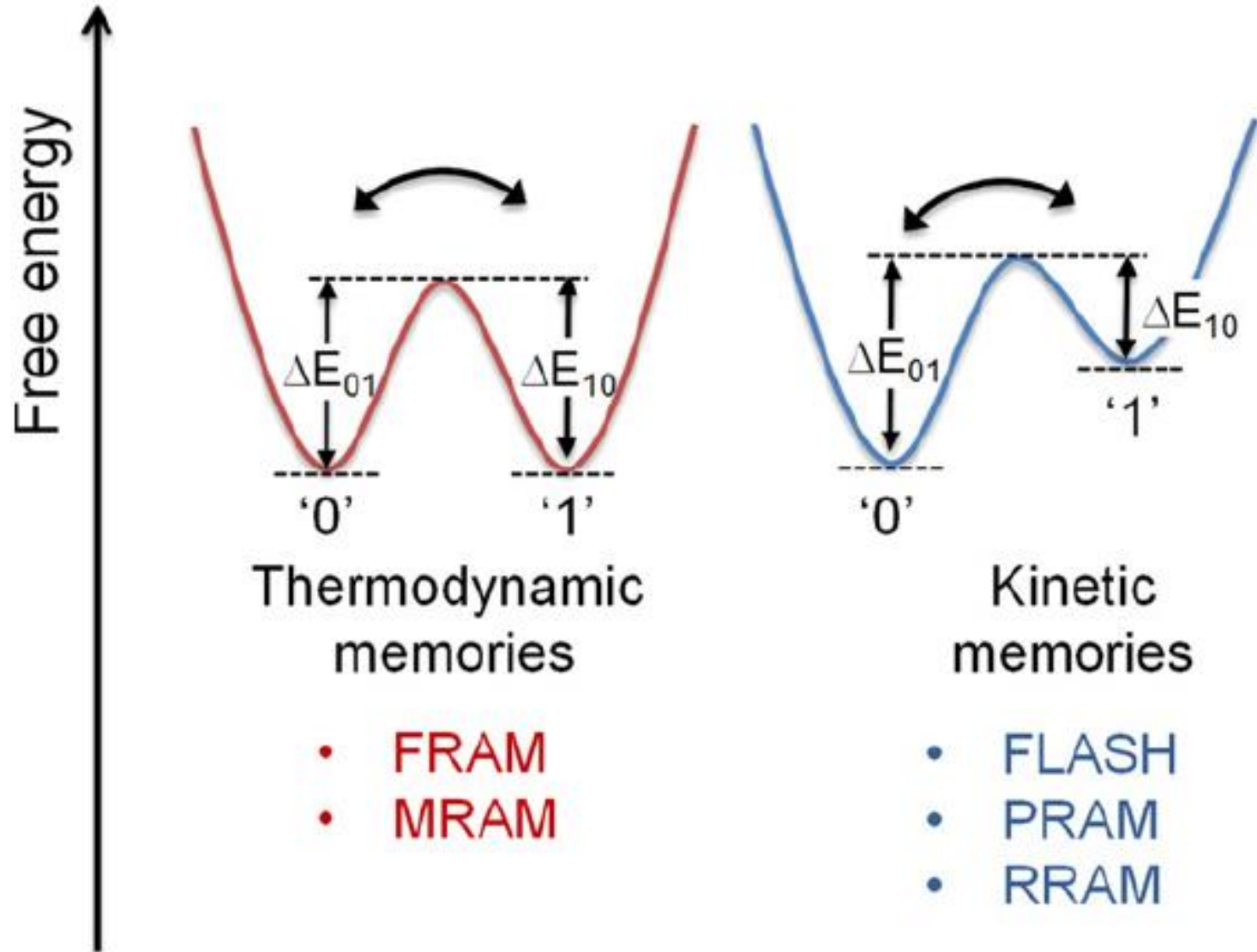
- E-field and Joule heating induced ionic movement
- Formation and dissolution of a conductive filament (CF)
- Compliance needed to prevent breakdown

# RRAM Pros and Cons

- Simple structure
  - two-terminal device
  - not limited by transistor scaling
- Ultra-high density
  - Cell size  $4F^2$
  - 3D stackable
  - terabit potential
  - Multi-level cell
- Large connectivity
- memory, logic and neuromorphic applications

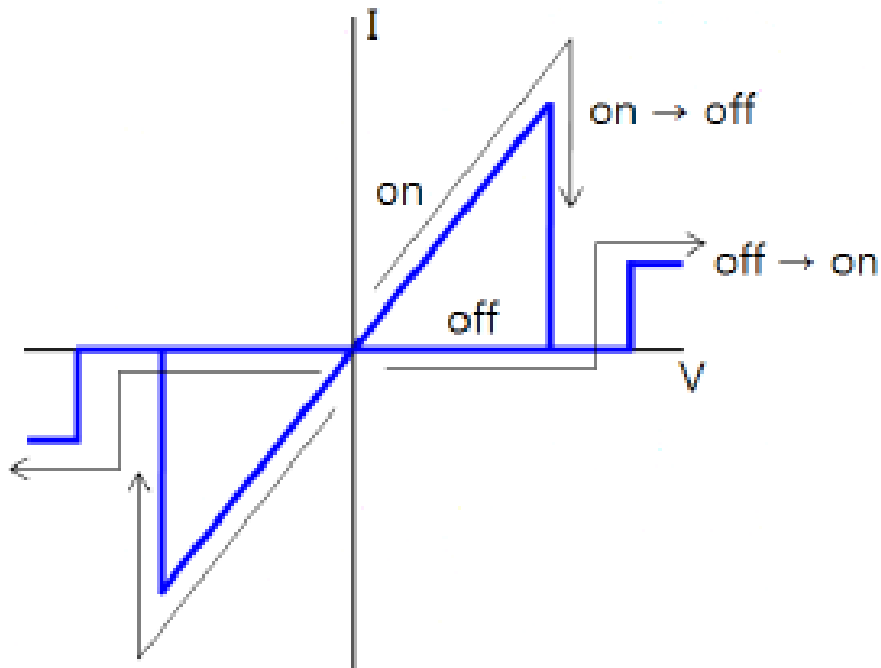


# RRAM Activation Energy

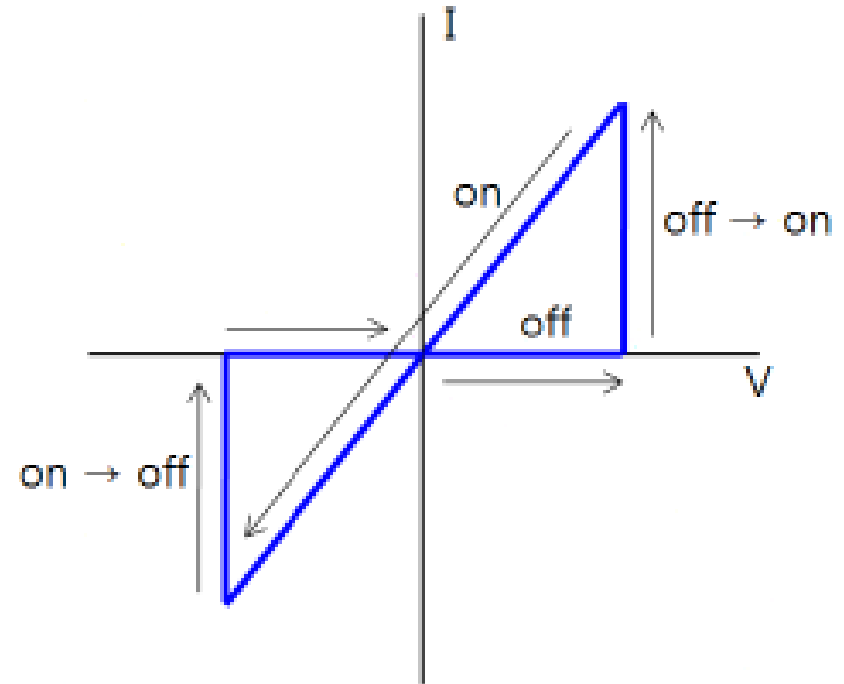


# Unipolar vs. Bipolar

## Unipolar



## Bipolar

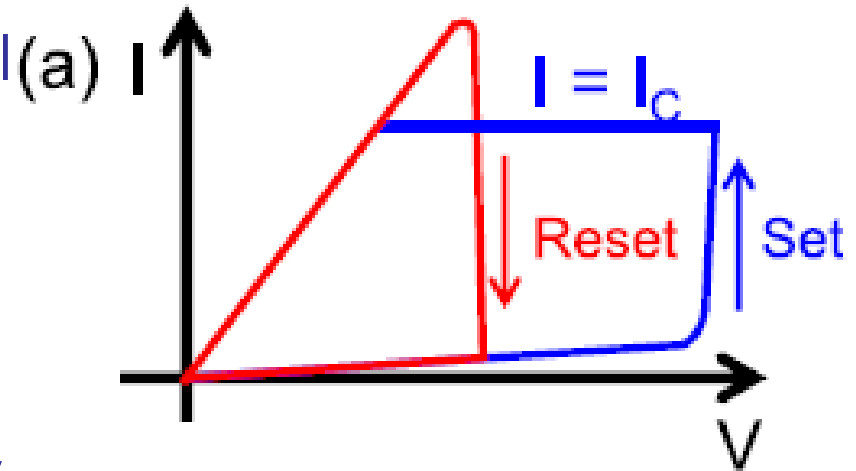


- Unipolar: both set and reset are achieved by applying voltages of the **same** polarity
- Bipolar: set and reset are achieved by applying voltages of **different** polarity



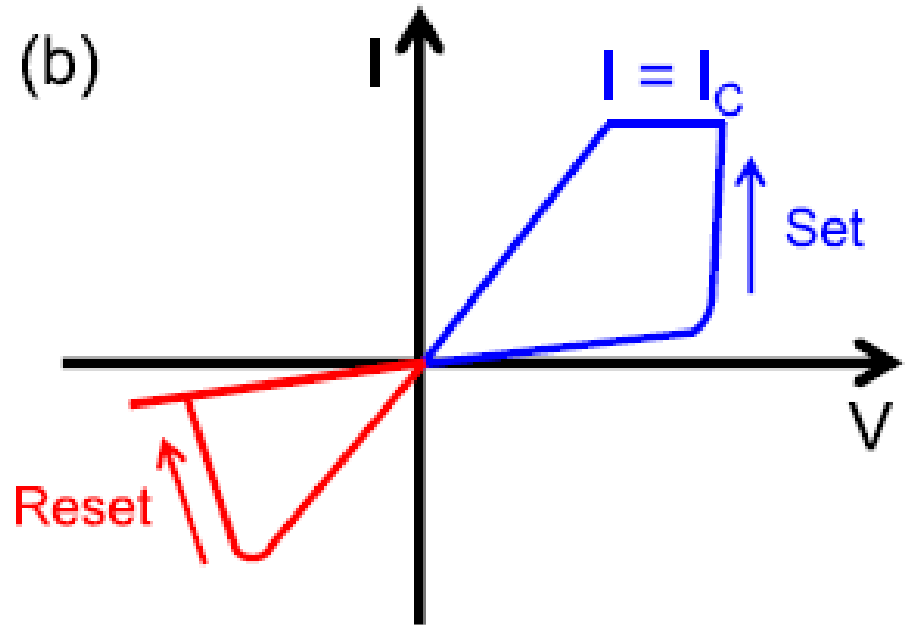
# Unipolar

- Thermally accelerated redox transitions
- Set: chemical reduction of metal oxides and CF re-connection
- Reset: oxidation of metallic filament and CF disconnection
- Preferred due to simple circuit and selection design in memory array
- Suffers from lower uniformity and low endurance



# Bipolar

- Depends on both temperature and E-field
- Reset: migration of ionized defects towards the –ve electrode → depletion region in the CF
- Set: injection of these defects back to the CF
- Better reliability and endurance
  - ionized defects are conserved during programming
- Current focus of most researches



# RRAM Switching Mechanisms

## Classification of the working principle

### Resistive Switching by Thermal / Chemical / Electronic Mechanisms

Phase  
Change  
Mechanism



PCM

Thermo-  
chemical  
Mechanism



TCM

Valency  
Change  
Mechanism



VCM

Electro-  
chemical  
Metallization



ECM

Electrostatic/  
Electronic  
Mechanism



EEM

### Material Impact

Chalcogenide Dominated

Electrode Dominated

### Switching Polarity

Unipolar

Bipolar

### Primary Mechanism

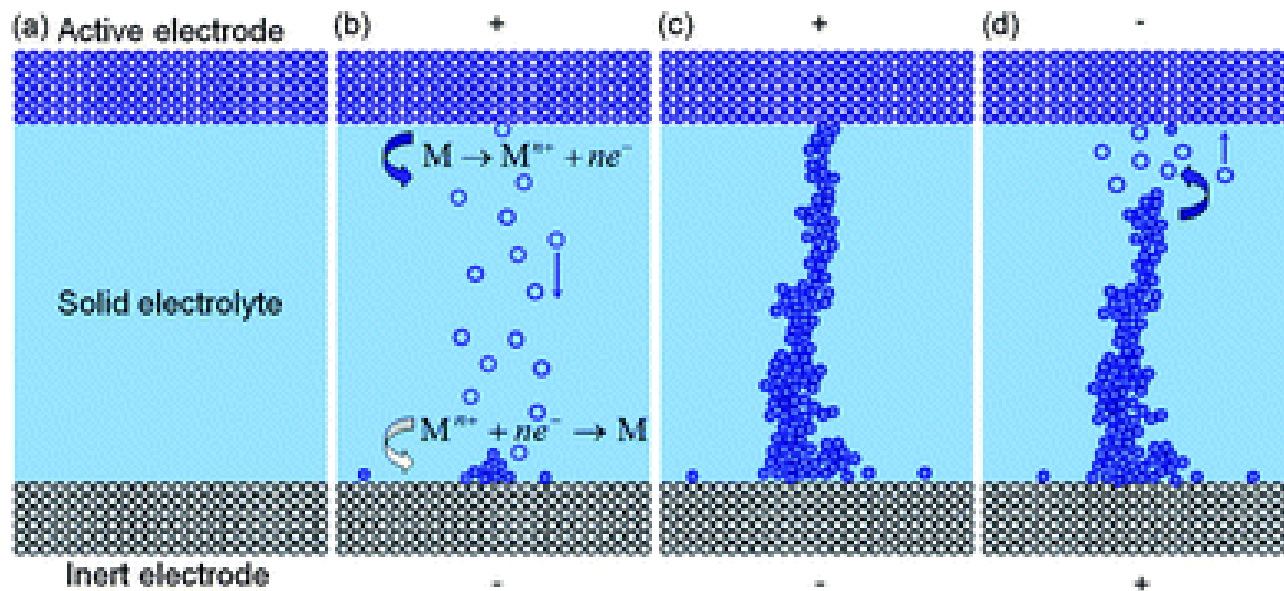
Thermal Effect

Redox-Related Chemical Effect

Electronic Effect

ITRS\_ERD  
workshop,  
April 2010

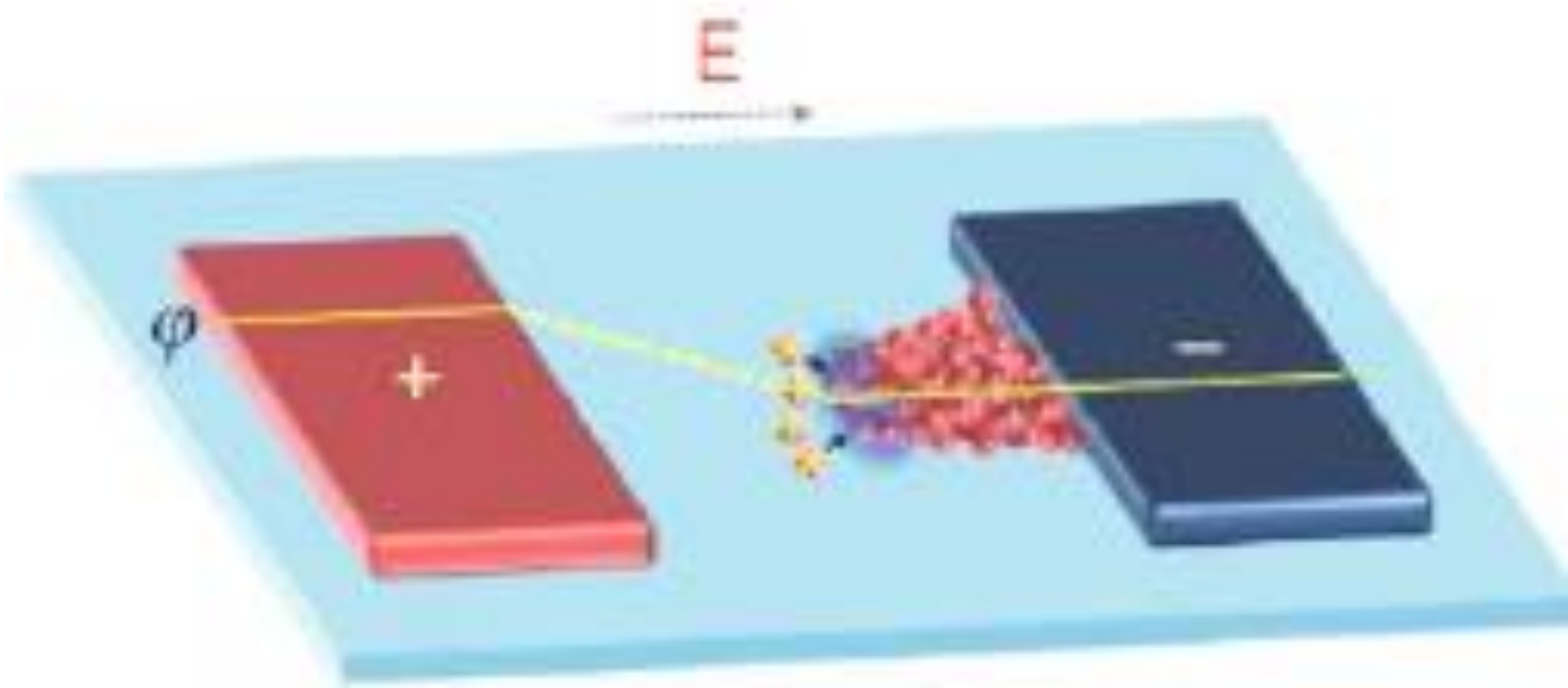
# Cation Switching



- Forming and dissolution of metallic filament inside a dielectric medium
- Mostly unipolar  $\rightarrow$  thermally accelerated redox reaction
- Key parameters:
  - ion mobility  $\mu$
  - redox reaction rate  $R$

# Cation Switching Case I

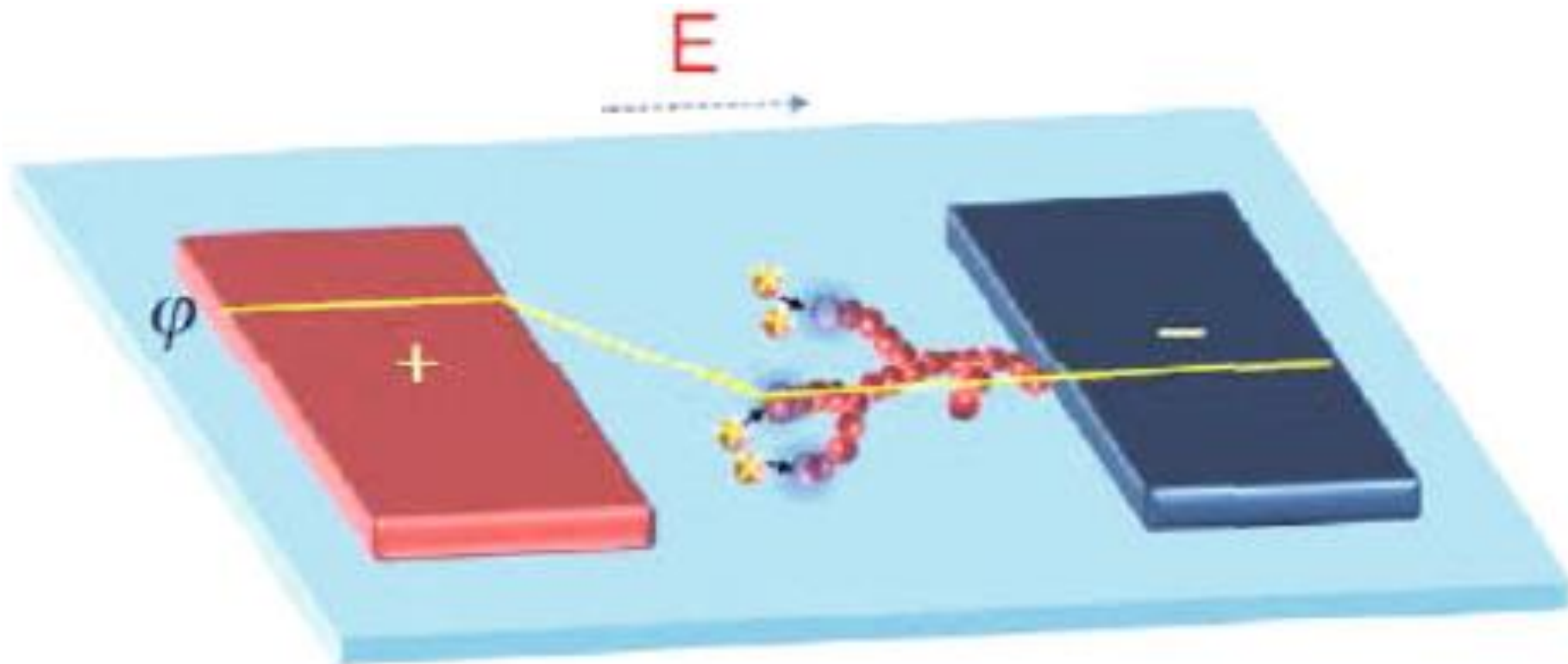
Yang et al., Nat. Comm., 2014



- High ion mobility  $\mu$  and high redox rate  $R$
- cations can reach anode (-ve) without reducing
- Filament **growth starts at anode**
- With large ion supplies (high  $R$ ), **inverted cone** shaped filament

# Cation Switching Case II

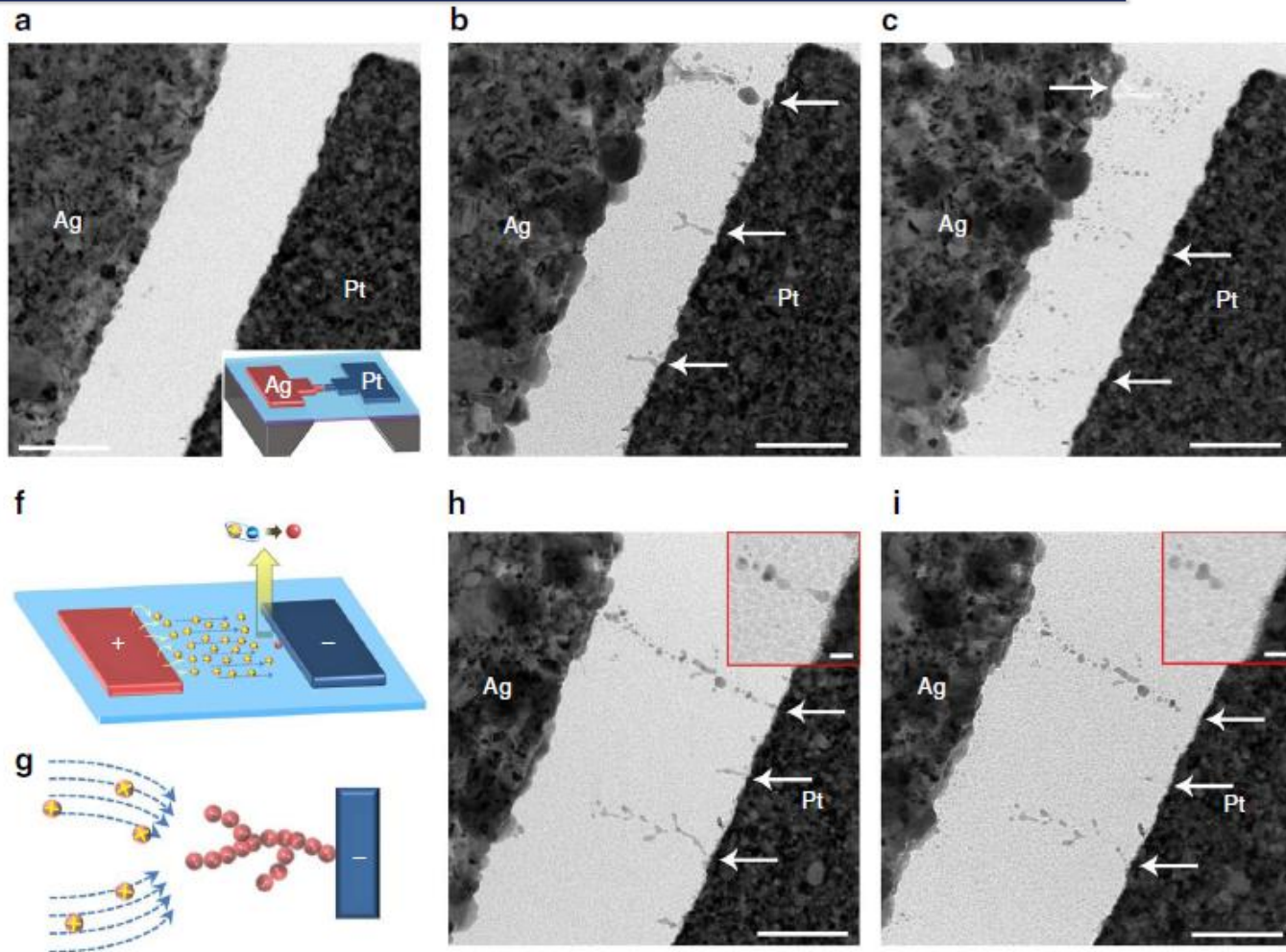
Yang et al., Nat. Comm., 2014



- High ion mobility  $\mu$  and low redox rate  $R$
- Ions are **reduced at anode** electrode
- With limited ion supply (low  $R$ ), reduction occurs at the edge  $\rightarrow$  **branched growth** towards cathode



# In-situ TEM of Filament Movements

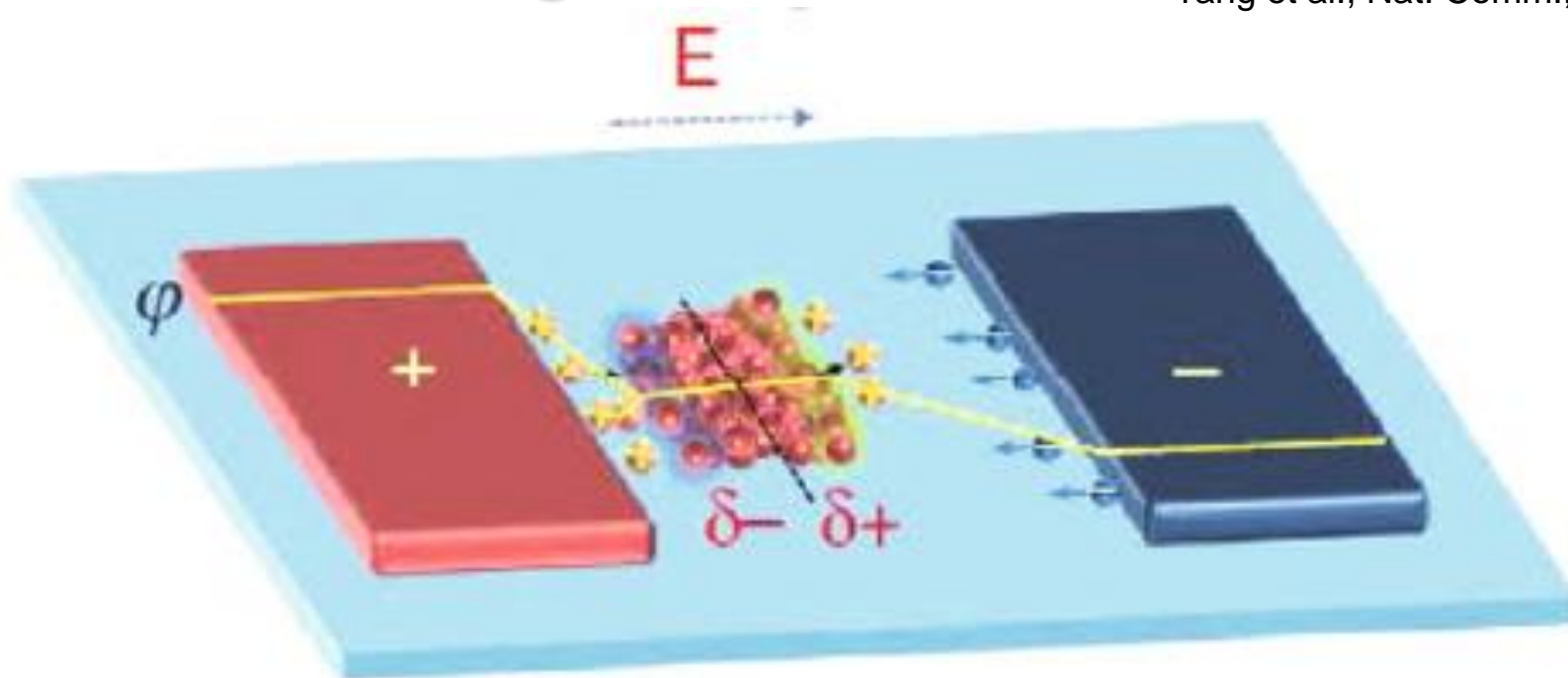


- High ion mobility  $\mu$  and low redox rate  $R$
- With limited ion supply (low  $R$ ), reduction occurs at the edge  
→ **branched growth** towards cathode

Yang et al., Nat. Comm., 2012

# Cation Switching Case III

Yang et al., Nat. Comm., 2014

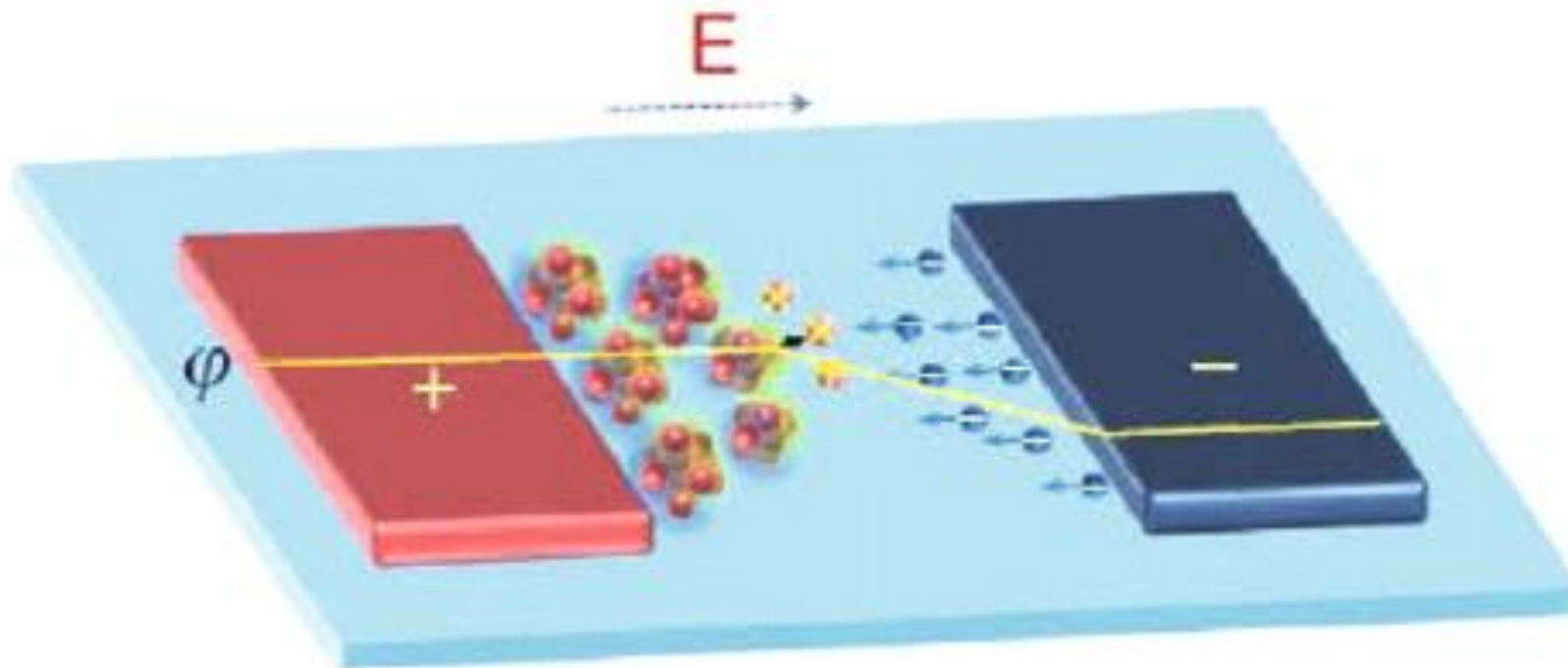


- Low ion mobility  $\mu$  and high redox rate  $R$
- Ions **nucleate inside dielectric**
- Incoming ions are **reduced at the site of nucleation** due to high redox rate
- **First** bridging the gap between nucleation and **cathode**
- **Then** filament growth towards **anode**



# Cation Switching Case IV

Yang et al., Nat. Comm., 2014



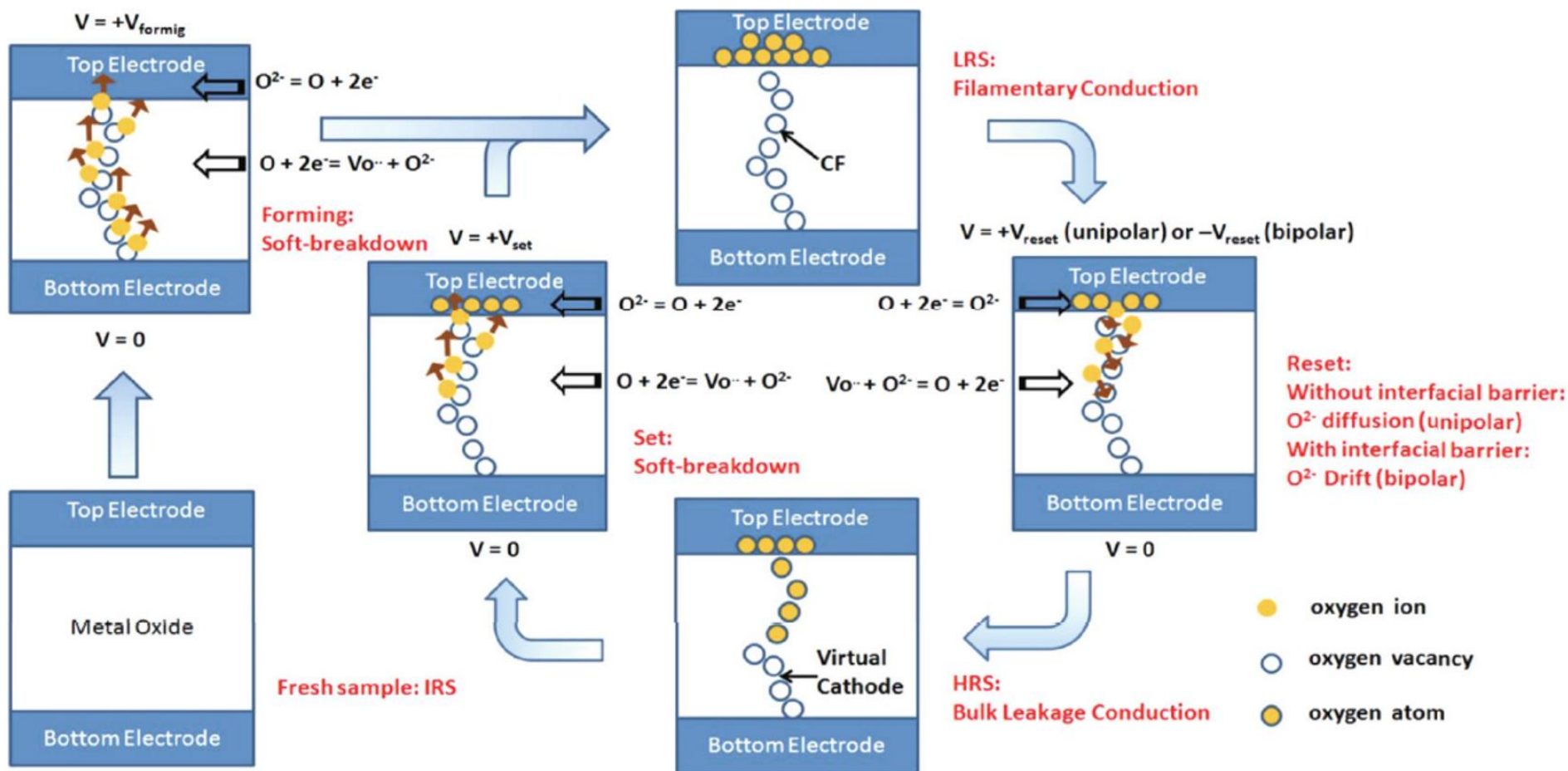
- Low ion mobility  $\mu$  and low redox rate  $R$
- Ions **nucleate inside the dielectric**
- Further growth through **cluster displacement** and repeated **splitting-merging process**

# Cation Summary

	High $\mu$	Low $\mu$
High R	Inverted Cone from anode	Nucleate from middle, grows towards cathode and then anode
Low R	Branched growth from anode	graduate cluster displacement and splitting-merging

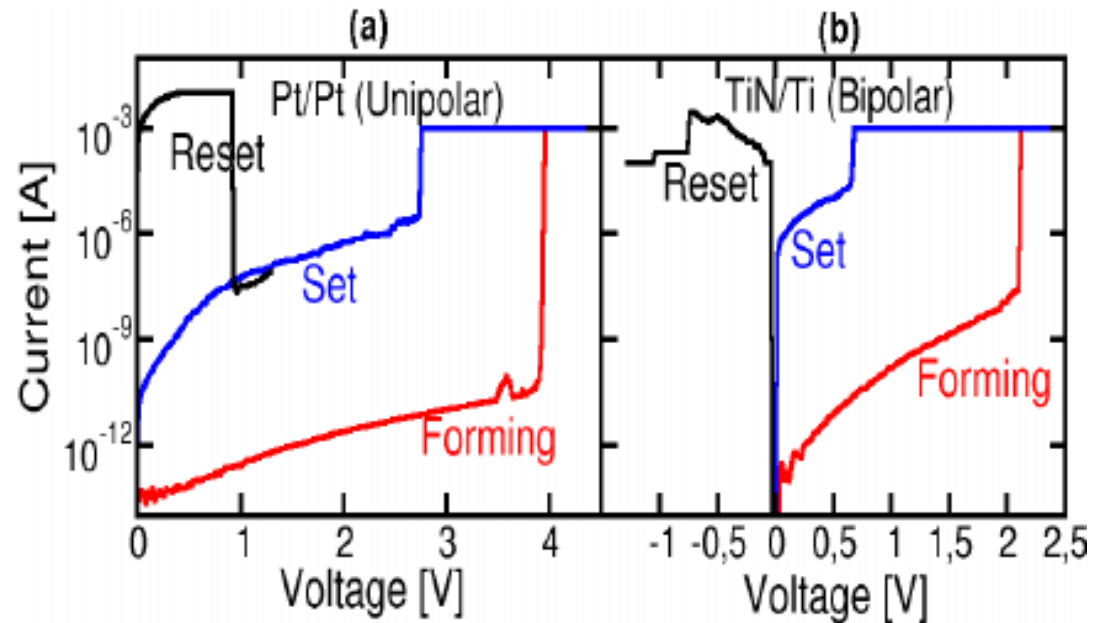
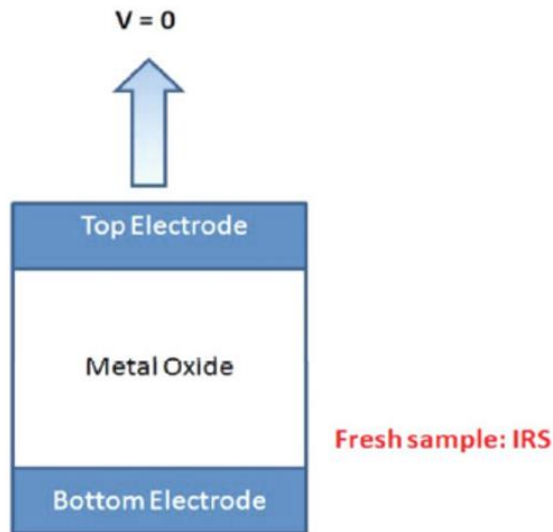
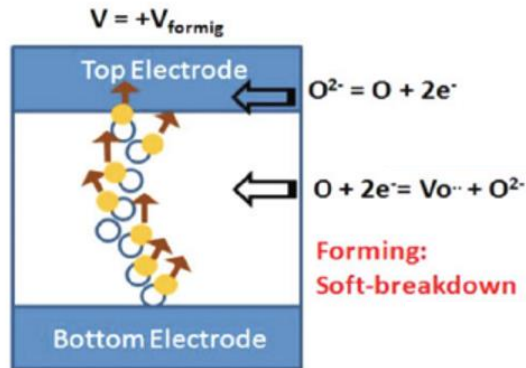
- High  $\mu$ : growth starts at anode side
- low  $\mu$ : growth starts inside dielectric
- high R: cone filament
- Low R: branched growth

# Anion Switching



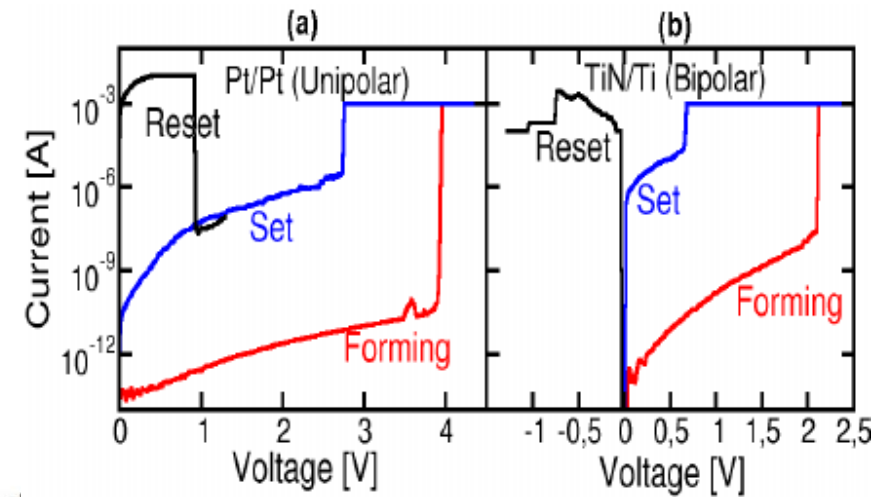
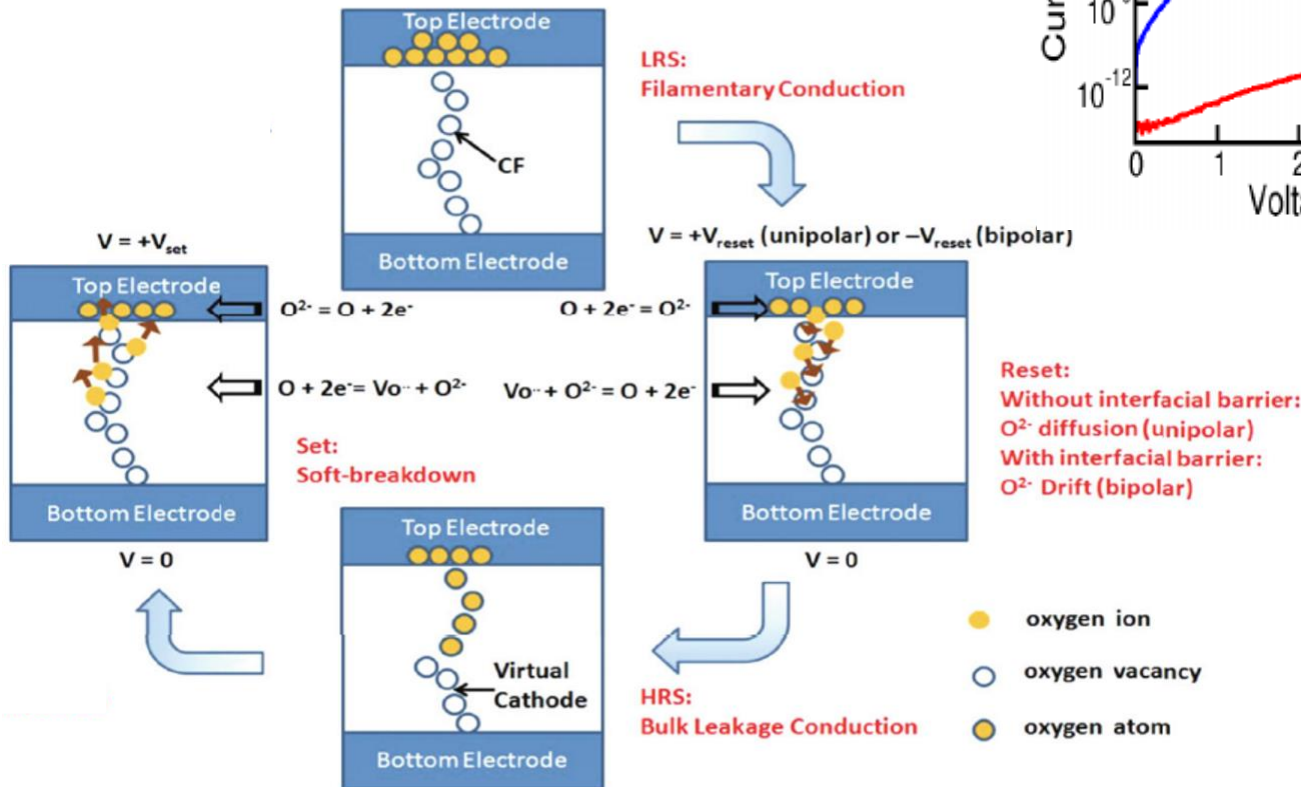
- Movement of oxygen vacancies ( $HfO_x$ ,  $TiO_x$ ,  $TaO_x$  ...)
- Mostly bipolar

# Forming



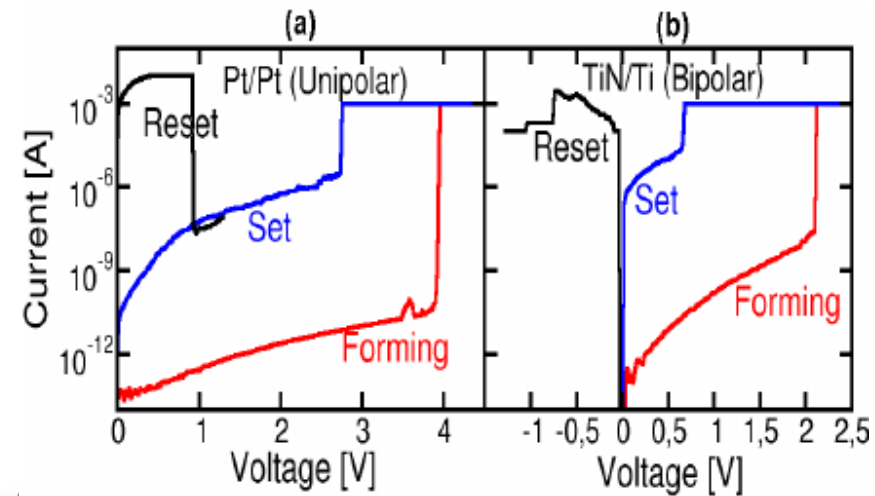
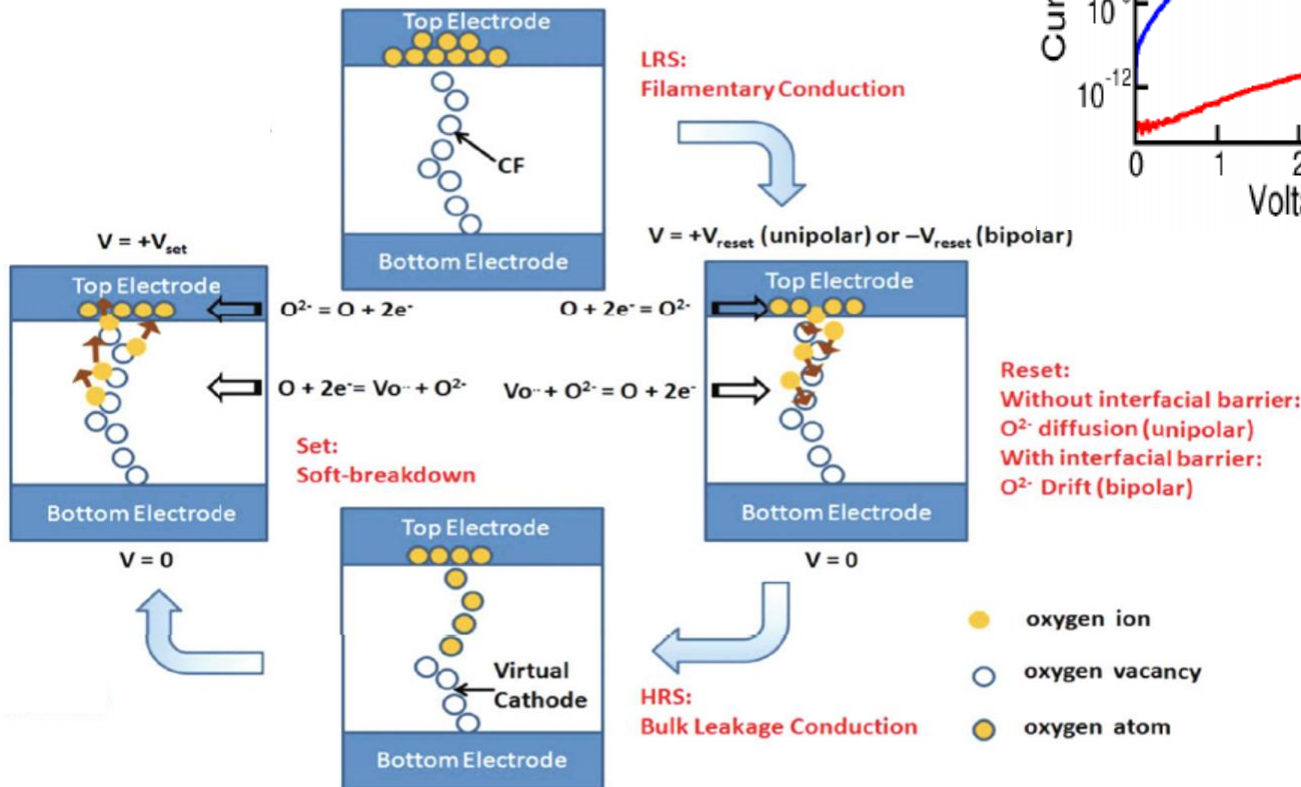
- First SET/soft breakdown process
- Requires higher voltage to form the filament pathway for the first time
- Current compliance is needed to prevent thermal runaway

# SET



- E-field induced soft breakdown
- Conductive filament made of oxygen vacancy
- Abrupt SET

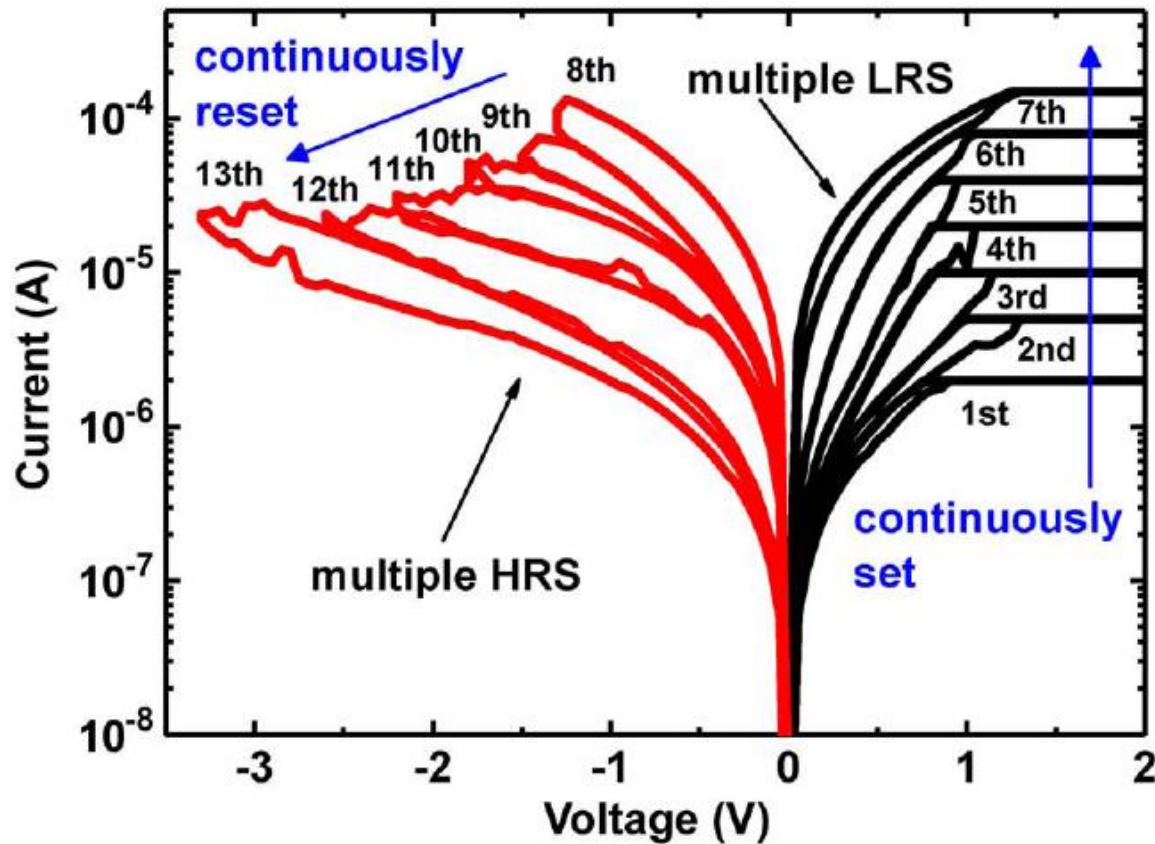
# RESET



- Gradual RESET
- Change of polarity for  $O_2$  ions to move in the other direction
- Breaking of conductive filament



# Gradual SET Process



- Gradual SET for (top) TiN/HfO<sub>x</sub>/AlO<sub>x</sub>/Pt (bottom)
- Possible reasons: multiple CF formed and/or stronger CF formed with larger diameter