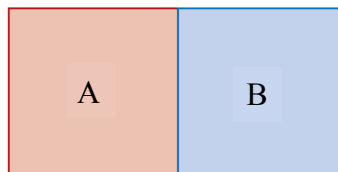


Material Science

Homework 3

Due Tuesday Nov 2, 2018

1. Compare interstitial and vacancy atomic mechanisms for diffusion, and cite two reasons why interstitial diffusion is normally more rapid than vacancy diffusion. (20%) (Problem 7.3)
2. A sheet of steel 1.8 mm thick has nitrogen atmospheres on both sides at 1200°C and is permitted to achieve a steady-state diffusion condition. The diffusion coefficient for nitrogen in steel at this temperature is $6 \times 10^{-11} \text{ m}^2/\text{s}$, and the diffusion flux is found to be $1.2 \times 10^{-7} \text{ kg/m}^2\text{-s}$. Also, it is known that the concentration of nitrogen in the steel at the high-pressure surface is 4 kg/m^3 . How far into the sheet from this high-pressure side will the concentration be 2.0 kg/m^3 ? Assume a linear concentration profile. (10%) (Problem 7.7)
3. A diffusion couple similar to as shown below is prepared using two hypothetical metals A and B. After a 30-h heat treatment at 1000 K (and subsequently cooling to room temperature) the concentration of A in B is 3.2 wt% at the 15.5 mm position within metal B. If another heat treatment is conducted on an identical diffusion couple, only at 800 K for 30 h, at what position will the composition be 3.2 wt% A? Assume that the preexponential and activation energy for the diffusion coefficient are $1.8 \times 10^{-5} \text{ m}^2/\text{s}$ and 152 kJ/mol, respectively. (15%) (Problem 7.29)



4. An γ iron-carbon alloy initially containing 0.20 wt% C is carburized at an elevated temperature and in an atmosphere wherein the surface carbon concentration is maintained at 1.0 wt%. If after 49.5 h the concentration of carbon is 0.35 wt% at a position 4.0 mm below the surface, determine the temperature at which the treatment was carried out. Use the diffusion data in Table. (20%) (Problem 7.31)

<i>Diffusing Species</i>	<i>Host Metal</i>	<i>$D_0(\text{m}^2/\text{s})$</i>	<i>$Q_d(\text{J/mol})$</i>
Interstitial Diffusion			
C^b	Fe (α or BCC) ^a	1.1×10^{-6}	87,400
C^c	Fe (γ or FCC) ^a	2.3×10^{-5}	148,000
N^b	Fe (α or BCC) ^a	5.0×10^{-7}	77,000
N^c	Fe (γ or FCC) ^a	9.1×10^{-5}	168,000
Self-Diffusion			
Fe^c	Fe (α or BCC) ^a	2.8×10^{-4}	251,000
Fe^c	Fe (γ or FCC) ^a	5.0×10^{-5}	284,000

5. A cylindrical specimen of a hypothetical metal alloy is stressed in compression. If its original and final diameters are 20.000 and 20.025 mm, respectively, and its final length is 74.96 mm, compute its original length if the deformation is totally elastic. The elastic and shear moduli for this alloy are 105 GPa and 39.7 GPa, respectively. (Problem 8.18)
6. A cylindrical rod 100 mm long and having a diameter of 10.0 mm is to be deformed using a tensile load of 27,500 N. It must not experience either plastic deformation or a diameter reduction of more than 7.5×10^{-3} mm. Of the materials listed as follows, which are possible candidates? Justify your choice(s). (Problem 8.23)

Material	Modulus of Elasticity (GPa)	Yield Strength (MPa)	Poisson's Ratio
<i>Aluminum alloy</i>	70	200	0.33
<i>Brass alloy</i>	101	300	0.34
<i>Steel alloy</i>	207	400	0.30
<i>Titanium alloy</i>	107	650	0.34

Tabulation of Error Function Values

<i>z</i>	<i>erf(z)</i>	<i>z</i>	<i>erf(z)</i>	<i>z</i>	<i>erf(z)</i>
0	0	0.55	0.5633	1.3	0.9340
0.025	0.0282	0.60	0.6039	1.4	0.9523
0.05	0.0564	0.65	0.6420	1.5	0.9661
0.10	0.1125	0.70	0.6778	1.6	0.9763
0.15	0.1680	0.75	0.7112	1.7	0.9838
0.20	0.2227	0.80	0.7421	1.8	0.9891
0.25	0.2763	0.85	0.7707	1.9	0.9928
0.30	0.3286	0.90	0.7970	2.0	0.9953
0.35	0.3794	0.95	0.8209	2.2	0.9981
0.40	0.4284	1.0	0.8427	2.4	0.9993
0.45	0.4755	1.1	0.8802	2.6	0.9998
0.50	0.5205	1.2	0.9103	2.8	0.9999