Introduction to Materials Science & Engineering

Course Objective...

Introduce fundamental concepts in Materials Science

You will learn about:

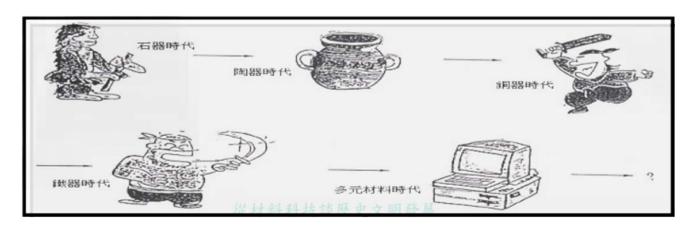
- material structure
- how structure dictates properties
- how processing can change structure

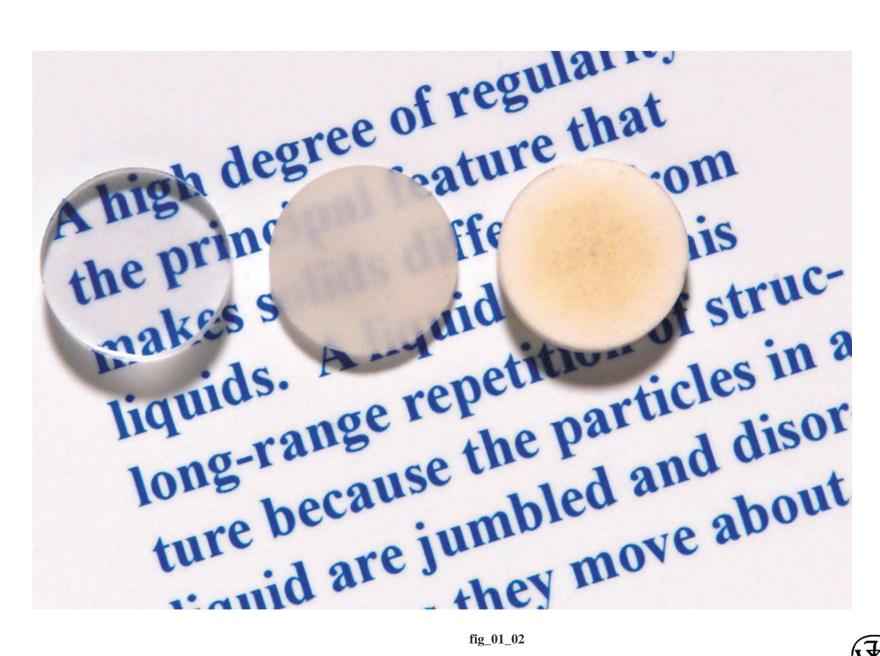
This course will help you to:

- use materials properly
- realize new design opportunities with materials

Chapter 1 - Introduction

- What is materials science?
- What is materials engineering?
- Why should we know about it?
 - → Materials drive our society
 - Stone Age → Bronze Age → Iron Age →
 Now → Future?





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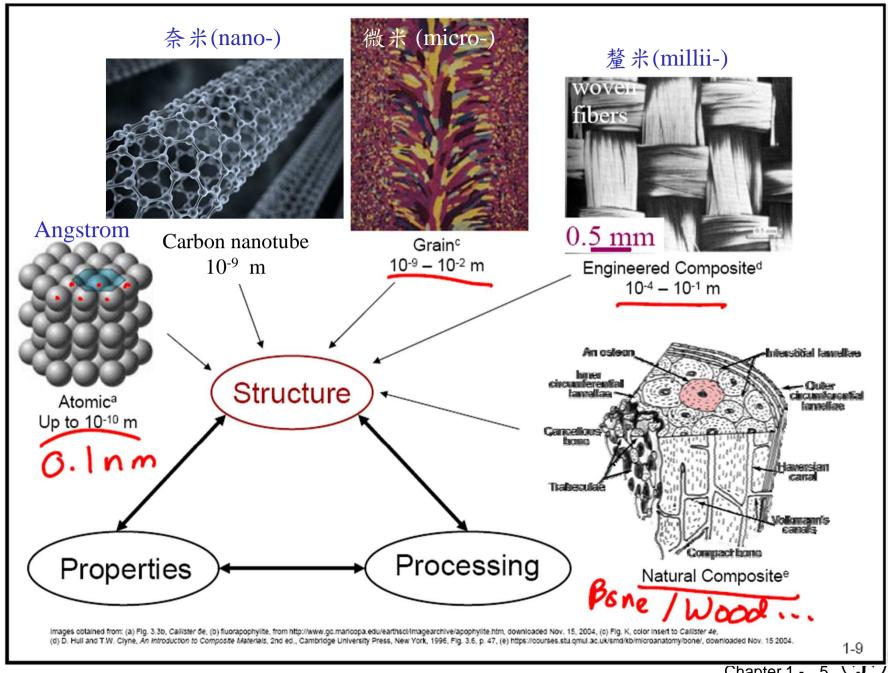


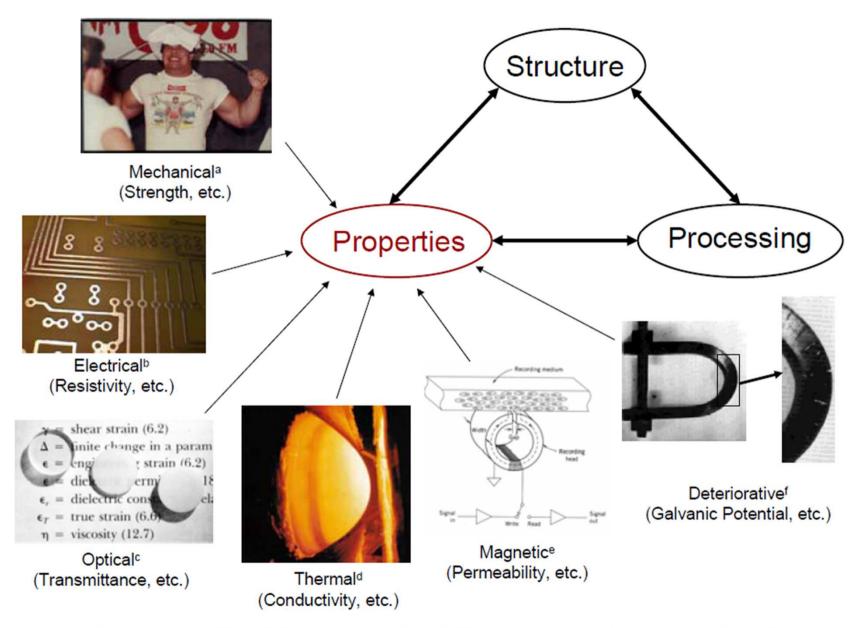
Glass (ceramic) can

Aluminum (metal) can

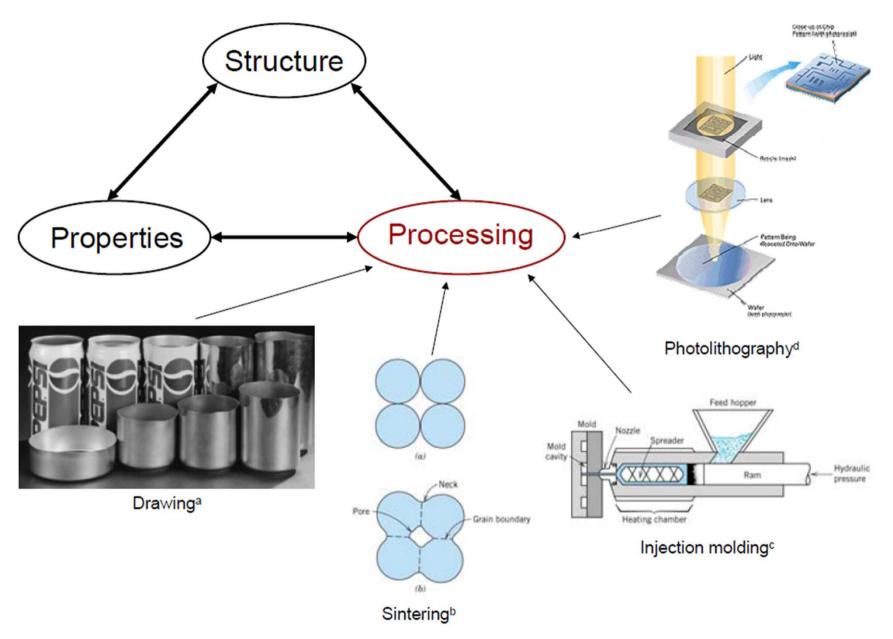
Plastic (polymer) can







Images obtained from: (a) http://www.mcshane-enterprises.com/ASL/Images/Mitcheli/barbendatvastrong.jpg, downloaded Aug. 24, 2004, (b) http://www.netadmintoois.com/cabletester/alltinned.jpg, downloaded Aug. 24, 2004, (c) Fig. 1.2, Callister 6e, (d) http://www.missliesandfirecontrol.com/our_products/spaceprograms/SHUTTLE/pic01-shuttle.html, downloaded Aug. 24, 2004, (e) Fig. 20.18, Callister 6e, (f) Fig. 17.0, Callister 6e.



(a) Adapted from opening picture in Ch. 11, Callister 6e, PEPSi is a registered trademark of PepsiCo inc., (b) Adapted from Fig 13.14, Callister 6e, (c) Fig. 15.23, Callister 6e, (d) www.just2good.co.uk/ opu\$llicon.htm downloaded Dec. 21, 2004.

Metals:

- Strong, ductile, stiff, dense
- high thermal & electrical conductivity
- opaque, reflective.
- Polymers/plastics: Covalent bonding → sharing of e's
 - Soft, ductile, low strength, low density
 - thermal & electrical insulators
 - Optically translucent or transparent.
- Ceramics: ionic bonding (refractory) compounds of metallic & non-metallic elements (oxides, carbides, nitrides, sulfides)
 - Brittle, glassy, elastic, stiff, strong, hard
 - non-conducting (insulators)
- Composites, semiconductors, biomaterials, smart materials, nanomaterials

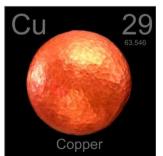
• Metals:

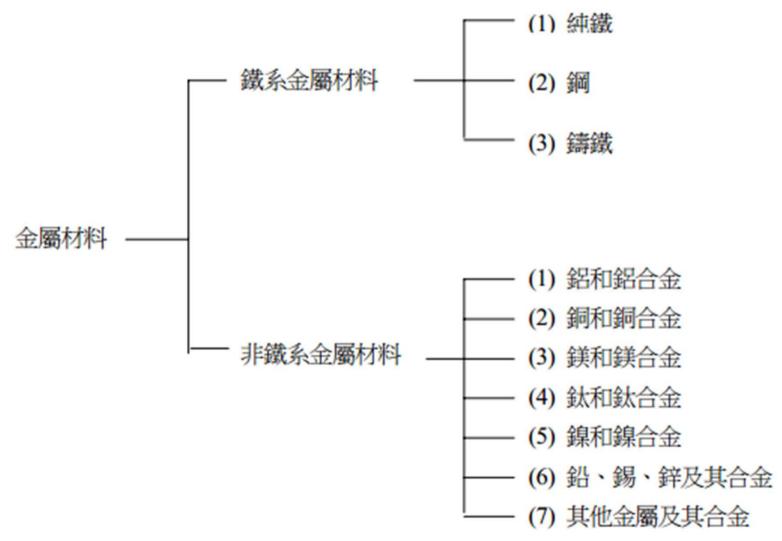
- Composed of one or more metallic elements (Fe, Al, Cu, Ti, Au and Ni....) and often also nonmetallic elements (C, O, N).
- Crystalline structure.
- Strong, ductile, stiff, dense
- High thermal & electrical conductivity
- Divided into two classes : Ferrous and nonferrous metals.













全鋁合金車型 EvaGT



航太交通工具



鋁合金輪圈



MOSSO 700C27速鋁合金自行車

MOSSO 7005航太鋁合金車架 前、後培林花鼓 微轉27速變速系統

3期0利率 31家	18期0利率 18家
6期0利率 31家	24期0利率 15家
10期0利率25家	30期0利率 1家
12期0利率10家	12期分期 15家

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鋁合金自行車

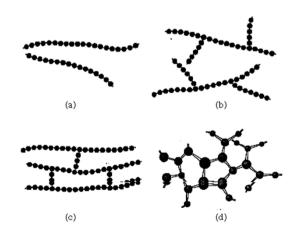


建議售價\$19999

網路價\$16800

• Polymers/plastics:

- Organic compounds chemically based on C, H and other nonmetallic elements (O, N and Si).
- Soft, ductile, low strength, low density.
- thermal & electrical insulators
- Optically translucent or transparent.





- **Ceramics**: ionic bonding (refractory)
 - Compounds of metallic & non-metallic elements (oxides, carbides, nitrides, sulfides) (ex: Al₂O₃, SiO₂).
 - Inorganic
 - Brittle, glassy, elastic, stiff, strong, hard
 - non-conducting (insulators)







Composites

Mixture of two or more materials.

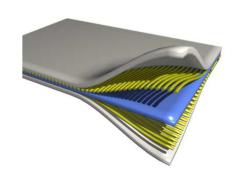
Polymer base composite, Metal base composite, Ceramic

base composite.

纖維強化高分子複合材料 (Fiber Reinforced Plastic, FRP)

Semiconductors

• Biomaterials, smart materials, nanomaterials







The Materials Selection Process

1. Pick Application

Determine required Properties

Properties: mechanical, electrical, thermal, magnetic, optical, deteriorative.

- 2. Properties → Identify candidate Material(s)
 Material: structure, composition.
- 3. Material → Identify required Processing

Processing: changes *structure* and overall *shape* ex: casting, sintering, vapor deposition, doping forming, joining, annealing.