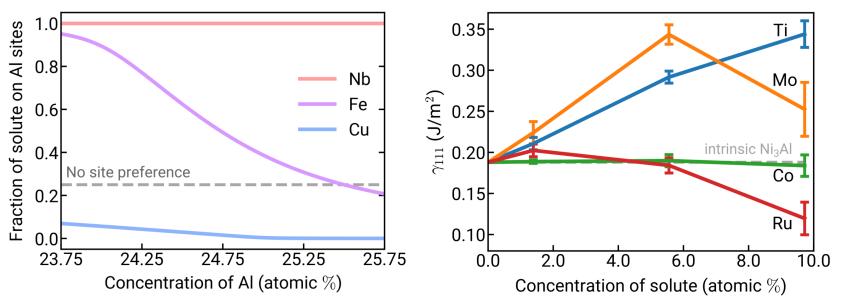
# Determining solute site preference and correlations to antiphase boundary energy in Ni-based superalloys

Enze Chen<sup>1</sup>, Tao Wang<sup>2</sup>, Mario E. Epler<sup>2</sup>, Timofey Frolov<sup>3</sup>, Mark Asta<sup>1</sup>

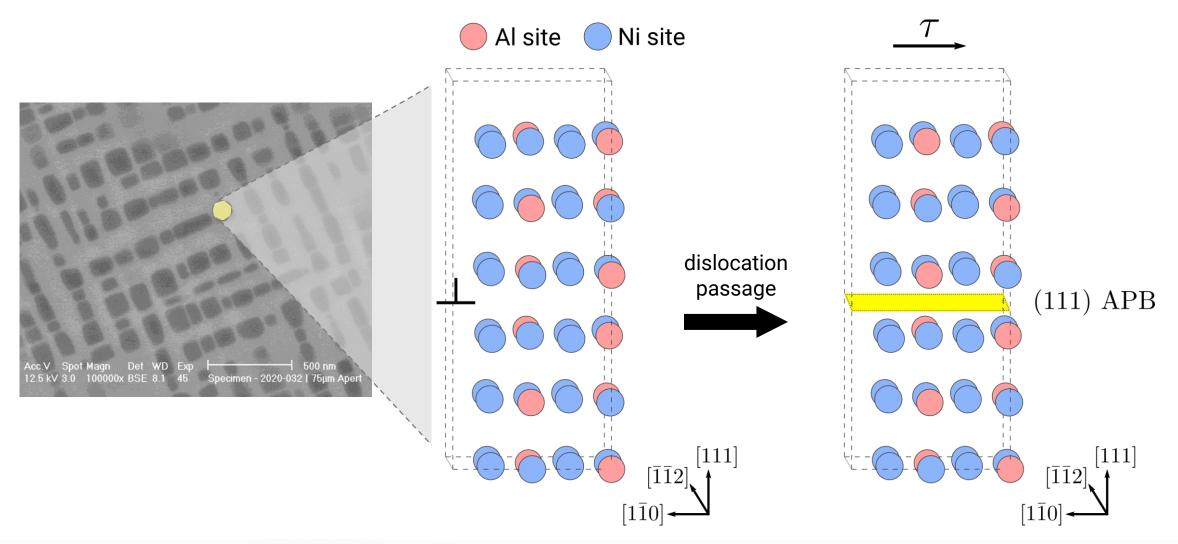
<sup>1</sup>University of California, Berkeley; <u>chenze@berkeley.edu</u> <sup>2</sup>Carpenter Technology Corporation; <sup>3</sup>Lawrence Livermore National Laboratory



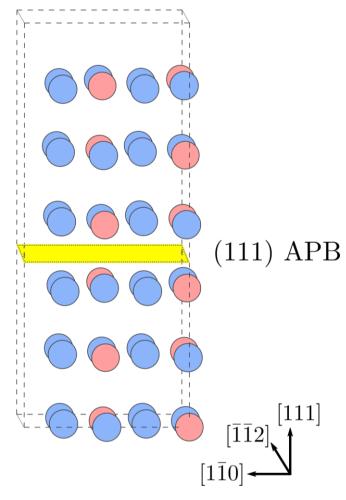
This work was performed under the auspices of the U.S. Department of Energy by Lawrence Livermore National Laboratory under Contract DE-AC52-07NA27344. LLNL-PRES-819127



## Superalloys and antiphase boundary (APB) energy



## Superalloys and antiphase boundary (APB) energy



$$\sigma_{y_{
m peak}} \propto rac{1}{2} \gamma_{
m apb} f^{1/2}/b$$

- ullet f is the volume fraction of the precipitates
- b is the magnitude of the Burgers vector
- ullet  $\gamma_{
  m apb}$  is the APB energy

Reed, R.C. The Superalloys, Cambridge, 2006

#### **Existing computational studies**

- At low concentrations for several ternary species, the APB energy varies approximately linearly and monotonically with solute concentration [1,2].
- The APB energy is related to the ordering energy, with a particularly strong correlation for 3d transition metal solutes at low concentration [2].

[1] Crudden, D.J. et al. Acta Materialia, 2014

[2] Gorbatov, O.I. et al. Phys. Rev. B, 2016

#### Our methodology

- We implement automated workflows that accelerate data generation and analysis of the APB energy.
- We further look for correlations of APB energy with other physical properties.
- 1. Determine the site preference of solute species in Ni<sub>3</sub>Al.
- Calculate the APB energy in Ni<sub>3</sub>Al-based alloys and analyze trends.
- 3. Identify correlations between physical properties and the APB energy.

#### Steps in our method

- 1. Determine the site preference of solute species in Ni<sub>3</sub>Al.
- 2. Calculate the APB energy in Ni<sub>3</sub>Al-based alloys and analyze trends.
- 3. Identify correlations between physical properties and the APB energy.

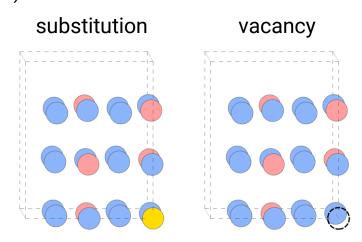
### Solute site preference – DSM formalism

• Dilute solution model (DSM) with first-order low-temperature expansion [1].

$$\Omega = E^0 - \sum_{i} \mu_i \sum_{p} \lambda(p) c_i^0(p) - k_B T \sum_{p} \lambda(p) \sum_{\epsilon} \exp\left(-\left[\delta E^{\epsilon}(p) - \sum_{i} \mu_i \delta c_i^{\epsilon}(p)\right] / k_B T\right)$$

$$\langle c_i(p) \rangle = c_i^0(p) + \sum_{\epsilon} \delta c_i^{\epsilon}(p) \exp\left(-\left[\delta E^{\epsilon}(p) - \sum_j \mu_j \delta c_j^{\epsilon}(p)\right] / k_B T\right)$$

• Implemented in PyDII [2].

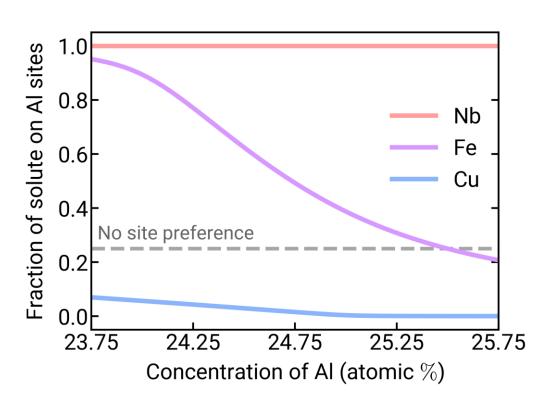


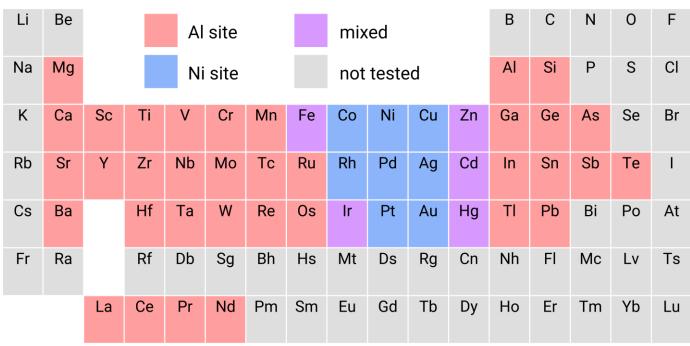
<sup>[1]</sup> Woodward, C. et al. Phys. Rev. B, 2001

<sup>[2]</sup> Ding, H. et al. Comp. Phys. Comm., 2015; GitHub

## Solute site preference - PyDII results

$$f_i(Al) = \frac{\langle c_i(Al) \rangle}{\langle c_i(Al) \rangle + 3\langle c_i(Ni) \rangle}$$





Sublattice preference in Ni<sub>3</sub>Al agrees with <u>Jiang, C. and Gleeson, B. Scripta Materialia, 2006</u>

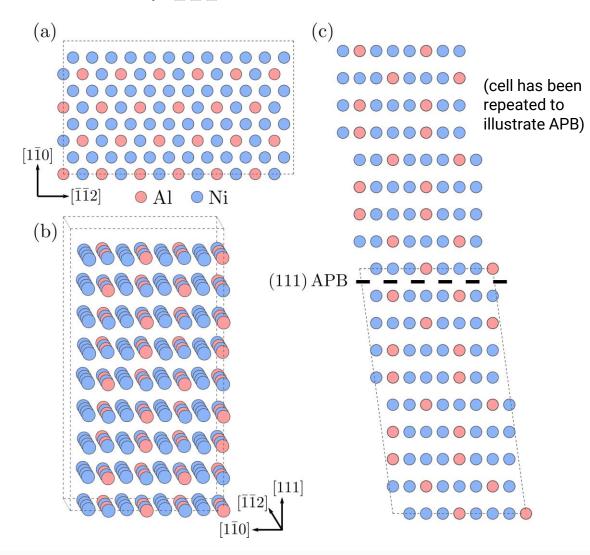
#### Steps in our method

- 1. Determine the site preference of solute species in Ni<sub>3</sub>Al.
- 2. Calculate the APB energy in Ni<sub>3</sub>Al-based alloys and analyze trends.
- 3. Identify correlations between physical properties and the APB energy.

## DFT calculations for $\gamma_{111}$

- 288-atom supercells with z-axis parallel to the [111] direction. Tilt creates an APB.
- At three concentrations for 30+ ternary elements, we randomly place solutes on the sublattice calculated by PyDII.

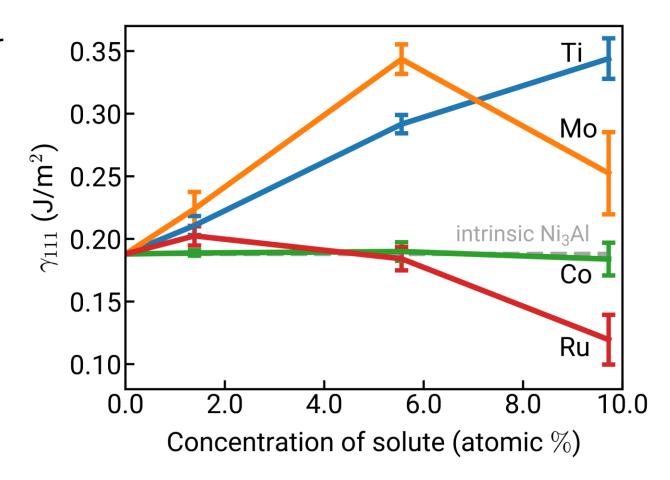
$$\gamma_{111} = \frac{E_{\rm apb} - E_{\rm bulk}}{A_{111}}$$





## Representative $\gamma_{111}$ vs. concentration profiles

- Most alloys with solutes that prefer the Al sublattice increase  $\gamma_{111}$  with increasing concentration.
- Elements with Ni or mixed sublattice preference tend to have little effect or decrease  $\gamma_{111}$ .
- Some are noticeably nonmonotonic, such as solutes in groups 6 and 7.





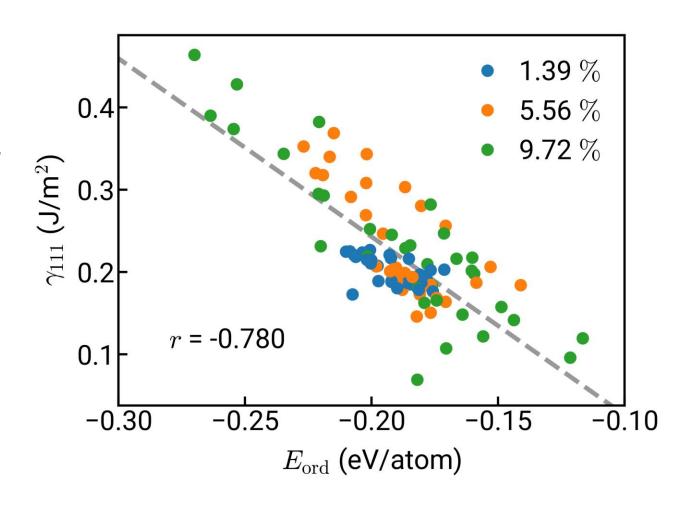
#### Steps in our method

- 1. Determine the site preference of solute species in Ni<sub>3</sub>Al.
- 2. Calculate the APB energy in Ni<sub>3</sub>Al-based alloys and analyze trends.
- 3. Identify correlations between physical properties and the APB energy.

## Correlation of $\gamma_{111}$ with ordering energy ( $E_{\rm ord}$ )

$$E_{\rm ord} = \frac{E_{\rm bulk} - E_{\rm random}}{N}$$

- We find a moderately strong linear correlation between  $\gamma_{111}$  and  $E_{\mathrm{ord}}$ .
- $E_{
  m ord}$  can also be calculated using CALPHAD methods [1].



[1] Miodownik, A. and Saunders, A. TMS, 1995

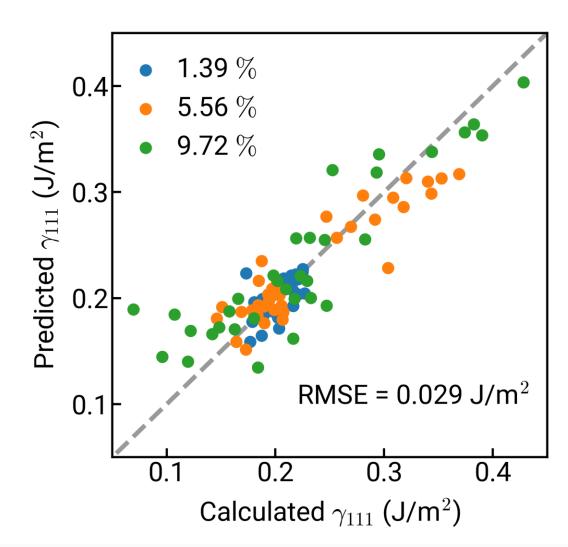
## Feature engineering for machine learning (ML)

- We use Magpie and Matminer libraries [1] to generate more physical descriptors that are composition-weighted averages of the elemental properties.
- We calculate the permutation importance of the features [2]:
  - Ordering energy
  - Mean Pettifor number
  - 3. Mean number of unfilled orbitals
  - 4. Mean space group number
  - 5. Mean column number
- After cross-validation (CV) we select only the top 7 features for model building.

<sup>[2]</sup> Breiman, L. Machine Learning, 2001

## Predicting $\gamma_{111}$ with an ML model

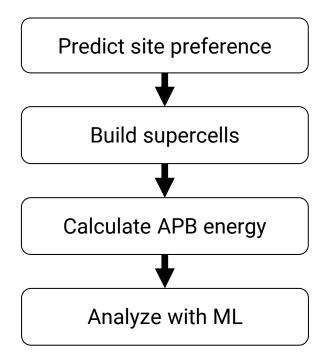
- We use the random forest ML model in Scikit-learn [1].
- We perform grouped 5-fold CV and calculate the average RMSE.

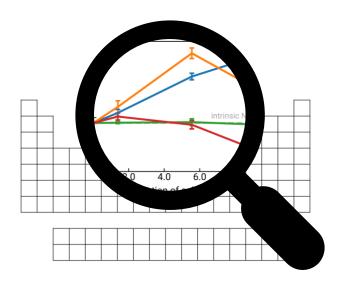


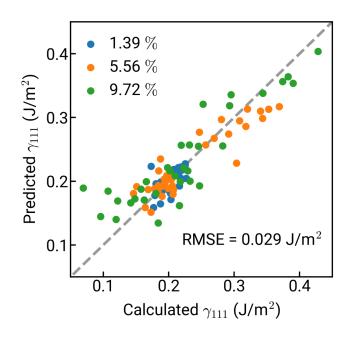
[1] Pedregosa, F. et al. JMLR, 2011



#### Conclusions







Created workflows to automate calculations

Analyzed trends in APB energy

Generated features for predictive ML models

## Acknowledgements

- Project funding: High Performance Computing for Materials (<u>HPC4Mtls</u>) Program of the DOE Vehicle Technologies Office, CRADA No. TC02309.
- Enze: NSF Graduate Research Fellowship Program under Grant No. DGE-1752814.



















