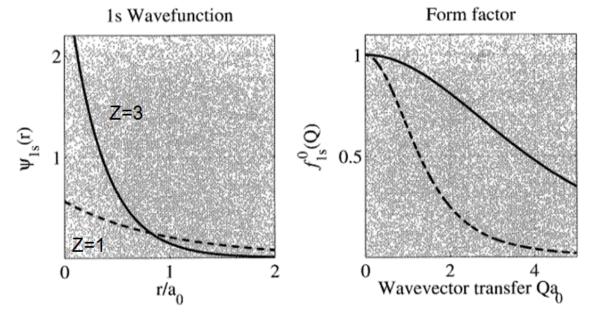


For bound electrons
$$\begin{array}{ll}
\sqrt{(\tilde{q})} & = \int g(\tilde{x}) e^{i \tilde{g} \cdot \tilde{x}} d\tilde{x} = \int \tilde{q} \to 0 \\
\Rightarrow \text{ atomic form factor}
\\
\text{Example} \quad |S| \text{ electron in H} \\
\forall_{1S} \text{ Cr}) & = \int \overline{\pi} a^{3} e^{-r/a} & a = \frac{a_{0}}{Z - Z_{S}} \\
Z_{S} \sim 0.3 \text{ (screening)}
\end{cases}$$

$$\begin{array}{ll}
f_{1S}(\tilde{q}) & = \frac{1}{\pi} a^{3} \int e^{-2r/a} e^{i \tilde{g} \cdot \tilde{x}} d\tilde{x}
\end{array}$$



Wavefn. localized => Form factor falls slowly with Q

Scattering from a molecule. U(3) constant (-ro f(z): atomic form factor molecule: Fmol (molecular form factor $= \sum_{i} f_{i}(\vec{\xi}) e^{i\vec{\xi}\cdot\vec{r}_{j}}$ CF4 (perfect tetrahedron) C: origin, C-F: a F: (0,0,a) $\left(\frac{2\sqrt{2}}{3}a, o, -\frac{a}{3}\right)$ $\left(-\frac{\sqrt{2}}{3}a, \frac{\sqrt{2}}{3}a, -\frac{\alpha}{3}\right)$ $\left(-\frac{\sqrt{2}}{3}a, -\frac{2}{3}a, -\frac{\alpha}{3}\right)$ (=) [3= iqa/3 + e iqa-**A.F.F.** molecular form factor

