

# Advancing predictive modeling through focused development of model systems to drive new modeling innovations

A funding proposal recently submitted to the NIH

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## SPECIFIC AIMS

While computational techniques are currently widely used in pharmaceutical drug discovery, current generation technologies (such as docking) are unsuitable for true molecular design. Specifically, these techniques fail to predict small molecule binding affinities to target and antitarget biomolecules with sufficient accuracy for the variety of applications currently of interest. Computational screening techniques can do better than random selection of compounds, but they lack the accuracy to guide molecular design or optimization. A new generation of physical techniques, alchemical free energy calculations, are poised to fill this void by providing a quantitative, predictive tool that can be used in multiple stages of the drug discovery pipeline, including lead optimization to improve affinity and selectivity or the retention of potency as other physical properties are optimized.

Recent success of alchemical methods in predicting accurate affinities has sparked considerable enthusiasm, but the domain of applicability of these techniques is currently highly limited; broad application and routine use will require further evaluation, refinement, and development. There is a vast gulf between targets within the domain of applicability and those which are outside it. Bridging this gulf to expand the domain will require focused study of carefully selected systems of intermediate complexity. Without such a bridge, these techniques may encounter the same problems faced by docking and related techniques: routine failure without clear insights into why, and years to decades spent making small methodological modifications without dramatic improvements in predictive power.

We propose to collect targeted experimental datasets, use them to conduct blind prediction challenges, and release curated benchmark sets in a manner designed to drive expansion of the domain of applicability and improvements to physical modeling techniques. The data we generate will provide a spectrum of difficulty between systems tractable with current methodologies to the pharmaceutically-relevant drug targets featured in the NIH-funded D3R effort, which fields blind challenges using protein-ligand datasets from pharma. Our systematic set of challenges aims to rapidly advance free energy techniques to the point of standard application in drug design. At the same time, the data we collect will play a long lasting role in the community, going through a life cycle of collection, curation, blind challenges, and then public dissemination to serve as benchmark sets and standard reference data, and to drive construction of new and improved models. While our work here focuses primarily on generating new targeted data for a series of blind SAMPL ("Statistical Assessment of Modeling of Proteins and Ligands") challenges and running those challenges, we also plan for subsequent data dissemination. Here, we will:

**Aim 1. Collect new physical property datasets to assess accuracy and spur improvements in force fields and modeling of protonation states and tautomers.**

We will develop new solution-phase datasets for druglike small molecules. These data can test critical aspects of small molecule modeling (including accounting for interactions and treatment of protonation/tautomeric state) and improve our ability to predict physical properties relevant to drug discovery in new regions of chemical space. We will initially focus on aqueous/nonpolar distribution coefficients and  $pK_a$  measurements, advancing to solubilities and membrane permeabilities, while using these data to drive improvements in the modeling of ligand interactions.

**Aim 2. Measure affinities of drug-like compounds in supramolecular hosts to challenge quantitative models of binding in systems not plagued by major receptor sampling issues.**

We will measure new host-guest binding free energies (using cucurbiturils and deep-cavity cavitands as hosts) to field binding challenges with varying complexity between physical property prediction and protein-ligand binding. Host guest systems are some of the simplest cases of molecular recognition, and thus these binding data will drive improvements in modeling of simple binding systems with techniques of relevance to drug discovery.

**Aim 3. Develop model protein-ligand systems that isolate specific modeling challenges of drug targets.**

We will identify suitable biological protein-ligand model systems that isolate individual modeling challenges (selected to push the limits of physical techniques) and develop these for blind challenges based on new protein-ligand affinity measurements. While the initial year will feature fragment binding to human serum albumin, subsequent challenge systems will be selected using a novel informatics platform to focus on timely modeling issues.

**Aim 4. Field community blind challenges to advance quantitative biomolecular design.**

The data collected in Aims 1–3 will drive annual SAMPL blind challenges, allowing the field to test the latest methods and force fields to assess progress, compare them against one another head-to-head, and perform sensitivity analysis to learn how much different factors (protonation state, tautomer selection, solvent model, force field, sampling method, etc.) affect predictive power. Results will then feed back into improved treatment of these factors for subsequent challenges, driving regular cycles of application, learning, and advancement.

Overall, the data generated here and the cycles of tests in SAMPL challenges will guide new innovations in physical methods for predicting binding and physical properties, providing a foundation for the next several generations of computational methods for pharmaceutical drug discovery.

## SIGNIFICANCE

**Physical modeling is poised to transform drug discovery and chemical biology by enabling true molecular design.** While modeling is used extensively in drug discovery, its main role at present is to aid with idea generation or filter large libraries of compounds for screening. Instead, we imagine using computational techniques extensively to guide the design process. Consider a medicinal chemist in the not-too-distant future who has just finished synthesizing several new derivatives of an existing inhibitor, and has obtained binding affinity or potency data against the desired biomolecular target. Before leaving work, she generates ideas for perhaps 100 new compounds which could be synthesized next, setting her computer to work overnight. By morning, the idea compounds have been prioritized based on reliable predictions of their affinity for the desired target, selectivity against antitargets, solubility, and membrane permeability. The chemist looks through the predicted properties for the top few compounds, selecting some for synthesis. If synthesizing and testing each compound takes several days, this workflow compresses roughly a year's work into a few days.

While this workflow is not yet a reality, significant strides have been made toward accurate binding affinities [97–104], solubilities [105–107], selectivity and drug resistance [108], and membrane permeability [109, 110]. A considerable amount of science and engineering still remains to make this vision a reality. Given recent progress, the question now seems more one of *when* rather than *whether*.

Widespread availability of inexpensive graphics processing units (GPUs) provides a 100-fold increase in price-to-performance ratio over CPUs, while advances in automation [111] and sampling protocols have helped simulation-based techniques reach the point where they now begin to be genuinely useful in guiding drug discovery for a limited *domain of applicability* [100–104, 112, 113]. Specifically, in some situations, free energy calculations appear to be capable of achieving RMS errors of 1–2 kcal/mol with current force fields, even in prospective applications, sufficient to drastically reduce the number of molecules that must be synthesized and assayed [114]. As a consequence, pharmaceutical companies are beginning to use these methods in active discovery projects.

**Despite progress, current physical modeling methodologies suffer from severe limitations hindering their widespread use in molecular design.** For example, even “small” protein conformational changes not gracefully handled by current methodologies can yield errors up to 5 kcal/mol in calculated binding free energies [115], force field limitations still pose major challenges [116], and the inability to treat important chemical effects like protonation and tautomer equilibria drastically limits the domain of applicability. **For many pharmaceutically relevant systems, the most important sources of error—and modeling challenges—are not yet clear.**

Progress on addressing these challenges has been frustratingly slow, hindered by a lack of high-quality data and community focus. **Neither retrospective tests nor prospective application in discovery projects provides the necessary impetus and data to rapidly overcome remaining barriers to widespread utility.** Large-scale retrospective tests can assess *retrospective* performance, but they do not provide accurate guidance on utility for prospective design, nor do they effectively identify the most important sources of error. Retrospective tests can also easily result in over-fitting, where researchers apply a variety of protocols until apparently significant results are obtained by chance [117]. In retrospective tests, performance may also not be indicative of expected performance in applications because even well-meaning researchers can take advantage of prior knowledge. For example, if the binding mode of a ligand is already known crystallographically, a researcher may use that binding mode in retrospective tests, whereas prospective or design work would require first selecting among candidate binding modes, introducing substantial uncertainty unaccounted for in the retrospective statistics [97, 118, 119]. This also means that in retrospective tests, researchers almost invariably try far fewer methods than in prospective tests, resulting in much less new insight. Prospective tests, in contrast, force researchers to anticipate a multitude of potential situations rather than only those observed in a known benchmark dataset. Prospective application in actual discovery projects, while important, also does not provide the necessary impetus, partly because often, the predicted compounds are in fact never tested [102] or the experimental data necessary to assess the quality of the predictions is absent—for example, because binding affinities are not measured or no crystallography is available.

**To accelerate progress in quantitative predictive physical modeling, we need a series of community blind prediction challenges focused on pushing the limits of predictive techniques, providing a bridge between challenging but tractable problems and pharmacologically relevant but currently intractable problems.** These challenges should be designed to have the necessary high quality experimental data, but also be prospective, predictive tests. While the Drug Design Data Resource (D3R [120], discussed further below) provides an existing community blind challenge on protein-ligand binding, it focuses on using pre-existing pharmaceutical datasets, rather than on measuring new data carefully selected maximize community learning [120]. D3R serves well to assess where we are now—but we need a carefully-designed effort focused on improving modeling.

**Physical modeling accuracy advances most rapidly when progress toward a complex goal can be decomposed into resolving a series of tractable problems, as revealed by carefully collected and curated data.** To make rapid progress, our field needs an effort which focuses on specific *component* problems of the overall problem of interest, collects and curates data that highlights these problems, and drives progress via prospective challenges. This process allows the entire community to learn from both methodological success and failure. The model we propose here has been proven to drive dramatic improvements in modeling, as evidenced by our **Statistical Assessment of Modeling of Proteins and Ligands (SAMPL)** series of challenges. SAMPL, born out of frustration with the lack of venues for comparing predictive accuracy on a level playing field, was initiated by Anthony Nicholls of OpenEye software in 2007/2008 [121], and has run challenges approximately every two years since then [122–129]. Governance transitioned to an unfunded academic collaboration during SAMPL3 in 2012; this collaboration ran subsequent challenges as SAMPL4 (2014) and SAMPL5 (2016). The PI of this proposal (Mobley) was a primary organizer of SAMPL4/5 (2014–2016). Much as unit testing is an indispensable tool for discovering where bugs in a program are hiding when complex integration tests fail, exercises like SAMPL are valuable in pinpointing and correcting modeling errors when the overall performance fails to live up to expectations for complex pharmacological targets. To accomplish this, SAMPL has historically focused on both simple challenges that attempt to isolate likely sources of modeling errors, such as physical properties of small molecules (hydration free energies, aqueous tautomer ratios, partition or distribution coefficients between aqueous and nonpolar phases) as well as small molecule binding to targets of reduced complexity (such as host-guest binding, and binding of fragments to trypsin and HIV integrase). **SAMPL has already been a tremendous community resource, resulting in nearly 100 publications which are typically cited 5–50 times or more each [1–96].**

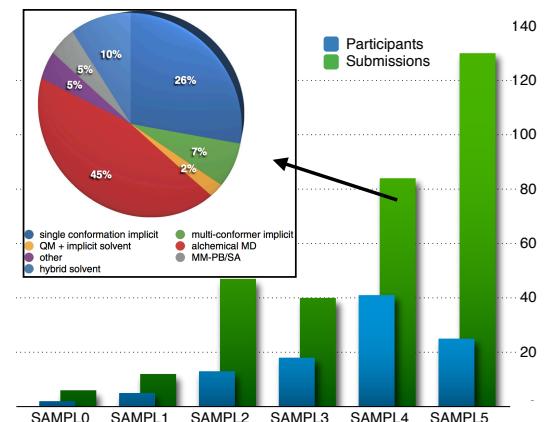
Here, we **design a new series of SAMPL challenges specifically to guide the improvement of models**. Until now, this has been impossible, because SAMPL has been entirely unfunded; its very existence has required “donation” of data and time, giving us no ability to gather datasets tailored to our purpose. Our proposed new challenges bridge the gap between calculations of simple physical properties that isolate forcefield inaccuracies from sampling challenges, like hydration—which can already be calculated fairly accurately [126]—and the D3R Grand Challenges on protein-ligand binding, which are a major source of consternation for the community so far [120, 130–132]. Unless this gap is bridged, there is the very real possibility that modeling may simply continue to fall far short of expectations in pharmaceutical challenges like D3R for reasons which are unclear. Here, we design challenges to highlight major reasons for failure and drive progress towards resolving them.

**Our major goal is to rapidly advance predictive modeling to where it can guide biomolecular design, and extension of the SAMPL challenges will do exactly that.** This work will play a vital role in enhancing the work being done on *existing* data by D3R, helping prepare methods for application to the more challenging systems emerging from pharma in D3R’s challenges.

## INNOVATION

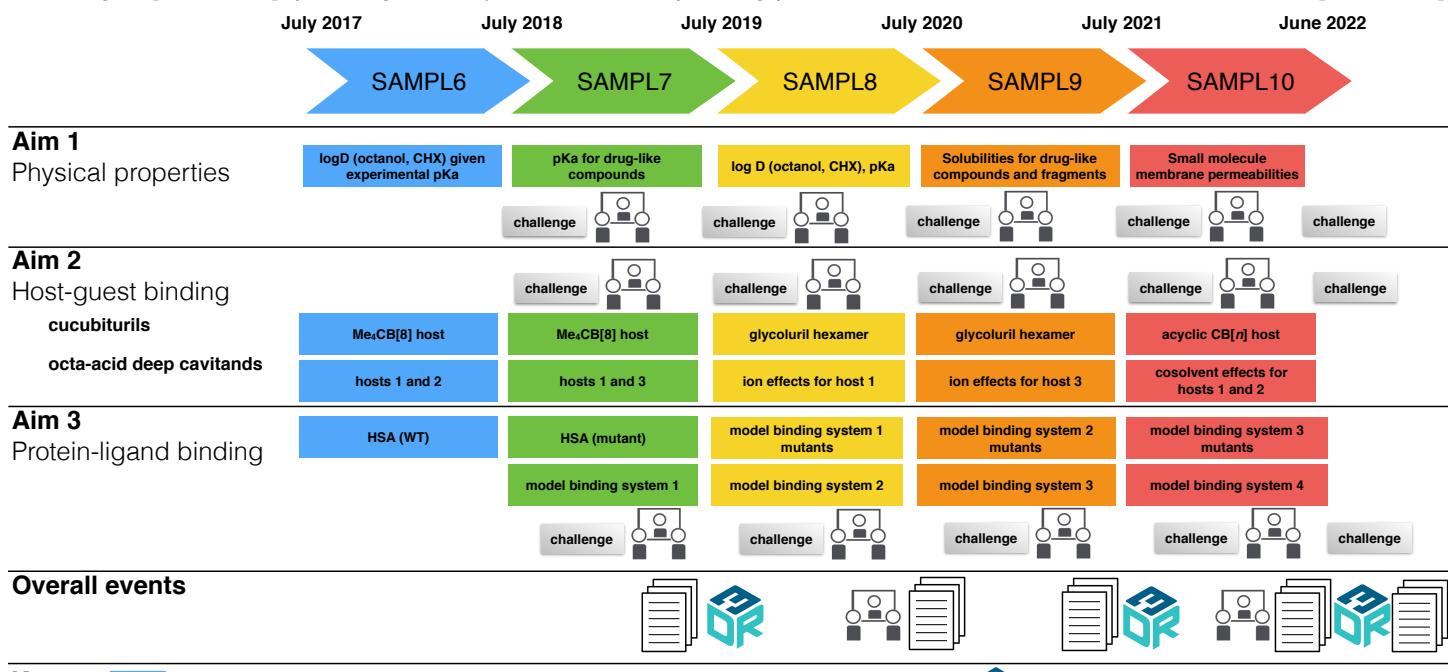
Blind predictive challenges—and SAMPL in particular—have already led to important new science on method development, evaluation, robustness, and force field improvements. However, they have not yet produced dramatic improvements in predictive molecular design, largely because the challenges posed have not been selected and staged for optimal impact. Indeed, SAMPL challenges to date have had to be opportunistic, making best use of measurements that contributing experimentalists were carrying out to advance other scientific goals. A central goal of this proposal, then, is to maximize the impact of future SAMPL exercises by crafting a series of challenges specifically designed to test and advance computational methods, and gathering the requisite experimental data.

Several historical examples serve to highlight how SAMPL can foster innovation (though far more examples are available in our SAMPL bibliography below; see also Figure 3). The first several SAMPL challenges on hydration free energies had rather hit-and-miss performance, highlighting pitfalls of existing methods and force fields which led to marked improvements in PB models [122, 133, 134], recognition of some limitations of fixed-charge force fields [135, 136], repair of some of these force field deficiencies via additional polarization or introduction of off-site charges [135–137], and helped motivate alternate implicit or hybrid solvent models [138–140]. Shifts in protonation



**Figure 1. SAMPL historical participation [126].** Historical participation in SAMPL host-guest + solvation/distribution challenges has climbed rapidly, and we expect this trend to continue. The number of participating groups is shown in blue, and the number of submissions in green. The inset shows the diversity of methods employed for the SAMPL4 hydration challenge, which is typical for SAMPL.

state and tautomer proved particularly important in the recent SAMPL5 log D challenge [128, 141]. This challenge provided a tractable opportunity to isolate and explore these specific physical effects, which are so important in protein-ligand binding, while avoiding the full complexity of pharmaceutical binding studies. Host-guest binding studies have also been particularly important [142], highlighting the importance of salt effects [129, 142, 143] and in some cases revealing more severe force field limitations than observed in hydration and distribution challenges [144, 145], pointing the way forward for improving predictive models of molecular interactions [142, 146].



**Figure 2. Timeline for our activities.** Activities covered by this grant include data collection and SAMPL challenges on our three major components (physical properties, host-guest binding, and protein-ligand binding), with each challenge cycle color-coded separately. Data collection within each Aim is shown by a colored bar indicating what is measured and curated. Data collection/curation is followed by a submission window for that challenge component, then all results and analysis are returned to participants and posted on the SAMPL website; this also will nucleate more detailed long-term discussion on the relevant Slack channel. At this point, we will also release the data to the public as a high quality benchmark. Each component will then wrap up with a virtual meeting focused on lessons learned and areas which need further exploration; these will be recorded and posted on our website to assist in rapid dissemination of new insights. Virtual meetings precede the submission window for the next SAMPL challenge, giving the opportunity to incorporate lessons learned for the next challenge. Submission windows and virtual meetings are staggered across categories so that participants can be involved in all three major areas without multiple simultaneous deadlines. In-person meetings are co-hosted with D3R and will occur every two years, supplemented by effort-wide virtual meetings in between. Special issues of JCAMPD will have deadlines shortly after the virtual meeting on the protein-ligand challenge for that year, and a 4-5 month timeline (based on historical experience) from the submission deadline until the special issue appears (with the first papers appearing online substantially sooner). Rapid dissemination of insights is critical for rapid progress, so we highly encourage the use of preprints and informal reports to supplement the special issue.

This work is also innovative because of the uniqueness of SAMPL. While there are other predictive challenges in the area of biomolecular modeling, such as D3R [120], the  $pK_a$  cooperative [147], CAPRI [148] and CASP [149], **no other blind challenge focuses specifically on data tailored and collected to drive quantitative protein-ligand modeling**. The SAMPL expansion we propose here is unique in its *specific design* to drive improvements in modeling accuracy rather than simply serving an evaluative role. SAMPL benefits the whole modeling community—for example, protein-ligand docking software has improved as a direct result of SAMPL hydration challenges [150], and commercial software vendors have introduced new features or scientific improvements based on participation in SAMPL challenges [141, 151]. In effect, **SAMPL serves as an engine to spur innovation by soliciting novel approaches to complex problems from the community and evaluating their success at predictions**.

This work also focuses on innovative experimental methods. Specifically, in Aim 3, we develop a new informatics platform to facilitate the rapid identification and study of particularly informative protein-ligand systems that are both experimentally tractable for high-throughput biophysical measurements and focus on specific challenges of interest. We employ a fully automated wetlab to screen potential model systems for expression, carry out high-accuracy biophysical measurements, and perform automated error analysis to carefully assess experimental uncertainty. **This work is at the forefront of innovation in high-throughput, automated biophysical experiments to produce high quality data with well-characterized uncertainties**. Not only will the data be of prime importance, but the techniques themselves will help future experiments.

In Aim 4, in addition to running SAMPL challenges, we will also perform reference calculations to test the accuracy of current state-of-the-art techniques. Both the Mobley and Chodera labs are experts in development of free energy methods for application to physical properties (e.g., [126, 152, 153] and binding (e.g., [115, 116, 154]), and **these reference calculations will drive innovation** as well, serving several key roles: (1) Benchmarking the latest method developments against current “best practices” methods (by doing calculations via both approaches); (2) Facilitating learning, allowing others to compare against our results to determine how a change in method or force field impacts results; (3) Focusing the field on key issues by doing sensitivity analysis to whether conditions such as ionic strength, protonation state, tautomer choice, etc., impact computed values.

Careful analysis of challenge predictions and results to identify why models fail and what specific problems need further attention is a critical and powerful aspect of SAMPL that spurs further innovation. Both organizers and participants play key roles in this; organizers identify global patterns and provide a venue for participants to explore these issues, while participants probe failure modes of individual methodologies in greater detail. When methods differ in performance, it is critical to understand whether the differences are statistically significant and important, and to provide an accurate accounting of the uncertainty in performance measures. Thus, careful and innovative analysis of challenge outcomes is particularly important in SAMPL [126, 128, 129], in some cases driving experimentation with new performance metrics [126].

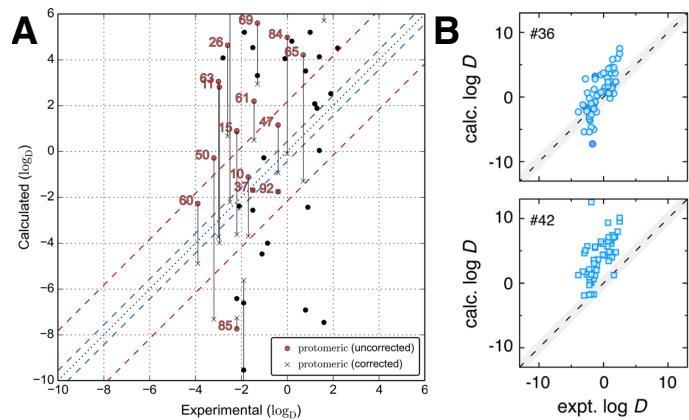
## APPROACH

Our approach to systematically advancing modeling for biomolecular design involves collecting carefully targeted experimental datasets for challenges focusing on physical property prediction, host-guest binding, and protein-ligand binding. These datasets, spanning a spectrum of complexity, help isolate individual limitations in quantitative physical modeling to encourage and evaluate multiple solutions from the community. Aims 1–3 focus on tailoring and generating this experimental data, while Aim 4 focuses on fielding annual SAMPL challenges. Each annual challenge includes one or more components from each of the Aims. Aims 1–3 bring together multiple laboratories and both theorists and experimentalists: graduate students from the Mobley and Chodera laboratories are paired with well-equipped experimental groups in industry to collect physical property data (Aim 1); Gibb and Isaacs, leading experimentalists in supramolecular chemistry, work with theorist Mobley to perform host-guest affinity measurements (Aim 2); and the Chodera lab applies new automated approaches to identify suitable protein-ligand systems and measure binding (Aim 3). The annual SAMPL challenges organized by the Mobley and Chodera labs (Aim 4) will leverage the data of Aims 1–3 (Figure 2) and best practices reference calculations to drive progress.

### **Aim 1: Collect new physical property datasets to assess accuracy and spur improvements in force fields and modeling of protonation states and tautomers.**

Simple physical properties such as solvation, partitioning, and protonation equilibria can be calculated quite precisely (but not necessarily accurately) with physical methods, allowing quantitative comparison between calculations and experiment and revealing and isolating deficiencies in our models. These properties allow us to directly probe force field accuracy and chemical effects like protonation and tautomer handling in the absence of slow conformational changes and other effects which complicate assessment in protein-ligand systems.

**Rationale:** We will generate new solution-phase physical property measurements for drug-like molecules to motivate improvements in force fields and handling of protonation states and tautomers. This builds on our work on water-cyclohexane distribution coefficients for SAMPL5 (in partnership with Genentech), which revealed major issues with handling of protonation states and tautomers [128] as well as serious forcefield limitations [137] (Figure 3). Distribution coefficients give the equilibrium ratio of concentrations of a solute between aqueous and nonpolar phases, and thus relate to transfer free energies from aqueous to more protein- or membrane-like environments. Thus, they capture many of the characteristics of transfer of drugs from water into binding sites but absent challenges with receptor conformational sampling and specific ligand-receptor interactions. In SAMPL5,



**Figure 3. Lessons learned from SAMPL5 log D predictions.** Predictions of log D values for SAMPL5 provided a number of key lessons. (A) Methods which treated multiple protonation and tautomeric states in their predictions performed dramatically better than those which did not; here, red dots move to x symbols when these effects are treated, improving accuracy in every case [155]. (B) Re-parameterization of a force field to more accurately reproduce pure solvent dielectric constants resulted in dramatically better predictions (top) than the original force field (bottom) [137].

distribution coefficients were challenging enough that many methods performed poorly, with even the best methods having accuracies less than would be expected based on hydration free energies in water [128], yet failures were informative and the major sources of error were issues which will also plague prediction of ligand-receptor interactions. In some respects, distribution coefficients posed the ideal SAMPL challenge, hitting the sweet spot in terms of difficulty—difficult enough that clear failures were frequent, with ample room for improvement, but not so difficult that the reasons for failure were unclear in general. Still, many models consistently disagreed with experiment for some compounds [128, 137, 141, 156], revealing the impact that targeted follow-up experiments (such as those we will conduct here) could have on improving models.

**This new experimental data is critical to maximize impact on the modeling community.** While the community has already benefitted considerably from SAMPL, as indicated by the nearly 100 publications on SAMPL (see attached list), the citation count, and the growing participation (Figure 1), progress will be greatly accelerated by the proposed initiative. A funded, coordinated effort allows the targeted collection of datasets designed to focus on the most important problems. With multiple investigators and collaborators, we are poised to respond and adapt to new challenges and opportunities which emerge in a manner not previously possible. Additionally, we will be able to continue experimental work until the necessary data is collected rather than terminating it at a specific time point dictated by industry internships—allowing us to do things like ensure the full dynamic range of log D values is covered, unlike in SAMPL5 [128, 156]. This will improve our ability to learn from the data—for example, the lack of dynamic range for SAMPL5 meant that, when calculated values often spanned a larger dynamic range than the experimental values, it was unclear if this was an artifact of the data set itself, experimental limitations, or force field problems [128, 137, 141, 156]. With funding to extend the experimental set via follow-up experiments, we will be able to resolve similar issues, providing further impetus to improve models.

Below, we summarize plans for data collection for the SAMPL6-10 challenges of Aim 4 (see also Figure 2). These specific data sets were selected based on a desire to help the field resolve the problems encountered in SAMPL5 then progress to accurate estimation of new properties. Data sets will typically consist of at least 96 compounds for good statistics, though when possible, a much larger amount of data will be collected.

**SAMPL6: Cyclohexane/water and octanol/water distribution coefficients.** Building on the success of distribution coefficient measurements in SAMPL5 and their surprising ability to motivate rapid advances in physical modeling methodologies [128], we will measure cyclohexane-water distribution coefficients at pH 7.4 for a new batch of commercially-available drug- and fragment-like molecules. Given the routine nature of octanol-water distribution coefficient measurements and indications that their prediction may be computationally tractable [153, 157] despite the heterogeneous structure of the wet octanol phase [158], we will also measure octanol-water distribution coefficients for the same compounds. Because  $pK_a$  prediction was difficult but critical for SAMPL5, we will focus SAMPL6 on forcefields and tautomers by measuring  $pK_a$  values, revisiting  $pK_a$  prediction in SAMPL7 and 8.

**SAMPL7: pKa measurements for drug-like molecules.** While much less complex than protein-ligand affinities, distribution coefficient measurements still conflate the challenging issues of protonation state and tautomer prediction, as well as transfer into different environments which may contain small but important quantities of co-solvents. Thus, we will separate these issues and improve our handling of them one at a time. For SAMPL7, then, we will measure  $pK_a$  values for an extensive set of drug-like molecules in water which will serve as the focus of the challenge, paving the way for SAMPL8.

**SAMPL8: pKa measurements and distribution coefficients.** In the next challenge, we will jointly explore  $pK_a$  and transfer issues, measuring distribution coefficients and  $pK_a$  values for the same set of compounds, with participants predicting (a) log D; (b)  $pK_a$ ; and (c) log P. Unlike SAMPL6,  $pK_a$  values will not be provided.

**SAMPL9-10: Solubility prediction and membrane permeability.** With solubility predictions now becoming tractable [105–107] (with Schrödinger also working on amorphous solubility prediction), solubility measurements will be a valuable test for SAMPL9, combining the solvation aspects of SAMPL1-8 with a new solid phase component. New computational techniques are targeting membrane permeability [109, 110], and this is experimentally accessible (see support letters from Pfizer and Merck), leading to our interest in permeability for SAMPL10.

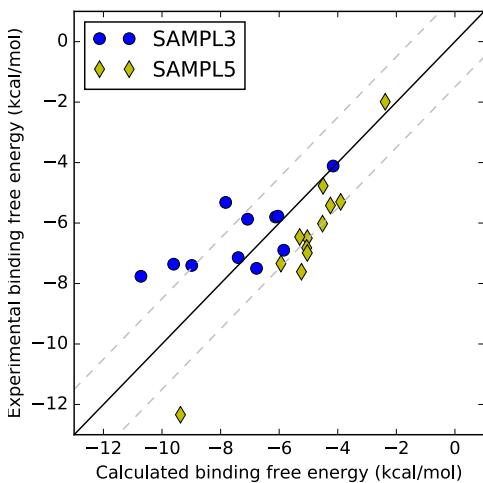
drug	features
memantine	adamantane; 1:1
saxagliptin	adamantane; 1:1
premarin	steroid
pancuronium	steroid
varenicline	1:1 vs 1:2
valsartan	$pK_a$ 4.37
omeprazole	$pK_a$ 4.77
ranolazine	$pK_a$ 7.17; epitopes
pradaxa	$pK_a$ 3.87; epitopes
nilotinib	epitopes; $pK_a$ 6.3
sensipar	epitopes; folding
vyvance	diamine; epitopes; folding
minocycline	tetracyclin; amino aniline

Table 1. Selected drugs whose binding to CB[n] hosts will be assayed for SAMPL6, 8, and 10 challenges (SA 2.1). These drugs bind to the cucurbituril-based host systems considered here, some at high affinity, so measuring their affinities provides a way to test methods for predicting binding interactions absent complexities present in protein-ligand systems.

**Experimental plan:** Experimental data will be collected in collaboration with our pharma partners (see support letters), roughly following the model used for SAMPL5, where Chodera lab student Bas Rustenburg went to Genentech to conduct cyclohexane-water log D measurements by adapting a Genentech high-throughput mass spectrometry workflow [156]. To collect this data, the Mobley and Chodera labs will send graduate students on visits or internships to industry collaborators to collect targeted datasets. Working with industry collaborators (see Letters of Collaboration from Genentech, Pfizer, and Merck) gives us substantial access to equipment and high-throughput measurement workflows—such as the Sirius T3 from Sirius Analytical (which can measure partition/distribution coefficients,  $pK_a$ s, and solubilities for molecules with titratable groups)—automation equipment, and compound libraries—for the purposes of rapidly collecting targeted datasets. Our previous experience demonstrates this model will work [156], and our partners see the value of this data and SAMPL to the modeling community.

Overall, Aim 1 extends prior SAMPL challenges via data focused on quantitative prediction of physical properties of tremendous relevance to accurately predicting biomolecular interactions, paving the way to applications in more complex systems addressed in Aims 2 and 3.

## Aim 2: Measure affinities of drug-like compounds in supramolecular hosts to challenge quantitative models of binding in systems not plagued by major receptor sampling issues.



**Figure 4. The best host-guest binding predictions of SAMPL3 [159] and SAMPL5 [144].** Binding free energy predictions have shown clear improvements from SAMPL3 to SAMPL5 as the major challenges become understood and are treated better by models, though a systematic offset remains in the best SAMPL5 predictions (yellow). Dashed lines denote errors of  $\pm 1.5$  kcal/mol.

importance of adequately sampling water rearrangements [127, 129, 142, 160].

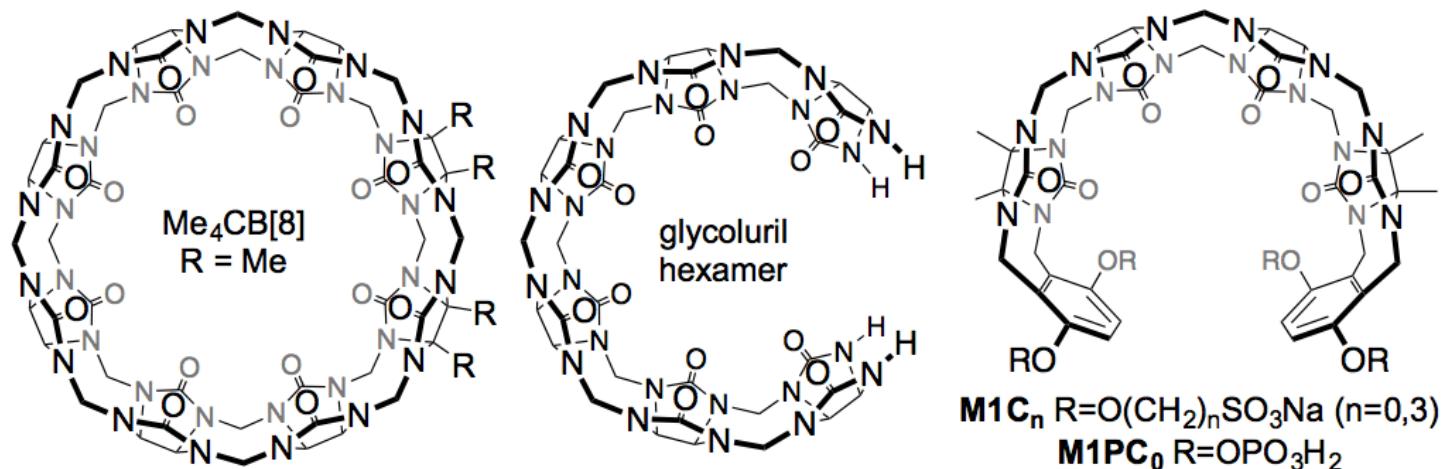
This new attention has resulted in clear improvements as participants begin to treat the relevant effects more accurately (Figure 4). Host-guest binding proves remarkably difficult to model accurately [161], in part due to force field limitations (resulting in new force field work [146]).

Here, we design a series of SAMPL challenges focused on two classes of host-guest systems—cucurbiturils and analogs (SA 2.1) and Gibb's deep-cavity cavitands (GDCCs, SA 2.2)—both of which build on prior SAMPL challenges. These two sets of systems exhibit different challenges as recently reviewed [142], with the hosts of 2.1 bringing relatively modest co-solvent and ion effects but some receptor sampling problems for the acyclic hosts, and the GDCCs of 2.2 bringing profound ion and co-solvent effects as well as water sampling challenges. Methods which perform well on one class may not perform well on the other [142], since the distinct sets of challenges highlight different limitations. This diversity drives more innovation than would a focus on a single host class.

### Subaim 2.1: Cucurbituril-based receptors as model binding systems

**Cucurbituril derivatives for host-guest binding.** Building on previous success with cucurbit[ $n$ ]uril (abbreviated CB[ $n$ ]) experiments for SAMPL challenges [162–164], we will conduct a series of new experiments on these receptors for five new challenges, with experimental work conducted by co-investigator Isaacs, an expert on these systems who provided data for previous SAMPL challenges. CB[ $n$ ] receptors are particularly well suited to our

Aim 1 focuses on the behavior of small molecules and its environment-dependence, in the absence of receptors and the associated potential for slow sampling, strong specific interactions, and other challenges such as salt effects. Binding in host-guest systems retains many of the same challenges seen in Aim 1 and introduces strong specific interactions and other challenges like salt effects [142], while still avoiding many of the issues with slow sampling (of protein conformational changes, ions, and ligand binding modes) seen in protein-ligand interactions. That is, binding in host-guest systems introduces a wider variety of challenges relevant to biomolecular interactions, but without the full array of challenges seen in protein-ligand interactions, as reviewed recently [142]. Thus, new data for SAMPL challenges on host-guest binding is critical to provide challenges of intermediate complexity between those of physical properties and those in biomolecular binding. We believe that host-guest binding challenges provide a vital step towards accurately modeling biomolecular interactions, focusing the field on issues not commonly encountered in physical property challenges (such as the importance of accurately modeling ionic conditions) that are highly relevant for protein-ligand interactions. Already, over the past several SAMPL challenges, host-guest systems have provided key tests for modeling of binding interactions, resulting in new attention paid to how co-solvents and ions modulate binding (resulting in errors of up to 5 kcal/mol when these effects are neglected) and the



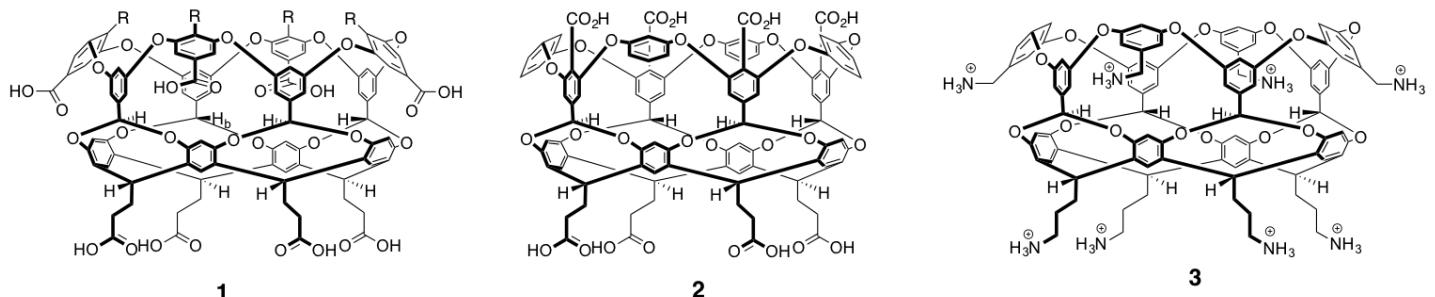
**Figure 5.** SAMPL6-10 host-guest challenges will feature cucurbituril hosts and analogs, including **Me<sub>4</sub>CB[8]**, glycoluril hexamer, and acyclic CB[n]-type receptors (SA 2.1). These receptors bind a variety of drug-like molecules, some with high affinity.

goals because they exhibit: (1) high binding affinities toward suitable guests in water comparable to protein-ligand affinities (routinely  $\mu$ M to nM; occasionally pM to fM) [165–171], (2) high selectivities between structurally related guests which translate into large  $\Delta\Delta G$  values [172], (3) low molecular weights (1–2 kDa) permitting high levels of theory to be used, and (4) highly restricted conformational degrees of freedom, reducing conformational sampling challenges often seen in protein-ligand binding. For SAMPL6-10, we will resynthesize a series of CB[n]-type receptors of increasing complexity, measure  $K_a$  values, and determine host-guest stoichiometry and geometry toward pharmaceutically relevant guests (selected drugs) in order to stringently test methods for predicting binding. Figure 5 shows the chemical structures of three hosts—Me<sub>4</sub>CB[8] [173], glycoluril hexamer [174], and acyclic CB[n]-type receptors [175–180] which span a range in terms of level of preorganization and formal charge.

**SAMPL6-10 cucurbituril challenges.** For SAMPL6, we will measure  $K_a$  and  $\Delta H$  values, stoichiometry, and geometry for the interaction of Me<sub>4</sub>CB[8] (a soluble CB[8] derivative) with 15 guests (selected top drugs, Table 1) by either direct or competition isothermal titration calorimetry (ITC), UV/Vis or fluorescence indicator displacement assay, or NMR competition experiments, as previously [164–166, 181]. Our selection of Me<sub>4</sub>CB[8] binding to top drugs allows us to modulate the computational complexity by: 1) changing host flexibility (e.g. Me<sub>4</sub>CB[8] can exhibit ellipsoidal deformation) [173], 2) allowing the possibility of binary or ternary (e.g. 1:1 and/or 1:2 host:guest) complexes [182–184], 3) using drugs with several potential binding epitopes or modes to induce sampling issues. Host:guest stoichiometry and geometry (e.g., which binding epitope is complexed) will be addressed by ITC  $n$  values, Job plots monitored by UV/Vis or NMR [185], and by <sup>1</sup>H NMR complexation induced changes in chemical shifts [186]. All studies will be conducted in phosphate buffered saline (pH 7.4 with physiological salt) which introduces its own complexities due to salt competition for binding [142, 187]. SAMPL7 will revisit the same host, but use 15 different guests be selected from commercial sources on the basis of reference calculations (on a larger set of guests) to ensure that they cover substantial dynamic range and/or exhibit affinities that depend substantially on the force field or water model, thus effectively testing our force fields and methods. For SAMPL8, we will focus on binding of the same 15 drugs (Table 1), but to glycoluril hexamer. This host introduces the complication of increased conformational dynamics, and influences the number and energy of solvating (and unusually coordinated) water molecules implicated in the high binding constants for CB[n]-guest complexes [171, 188]. The selected drugs include several with  $pK_a$  values in the 3.8 to 7.4 range; given that CB[n]-type receptors (like biomolecular receptors) can induce  $pK_a$  shifts in their guests of up to 4  $pK_a$  units [189–191], this will test how well models can predict these effects. Additionally, it will couple nicely with the focus on  $pK_a$  values in Aim 1. SAMPL9 will revisit glycouril hexamer with the same 15 guests from SAMPL7. SAMPL10 will shift to acyclic CB[n]-type receptors (e.g. M1C<sub>3</sub>, M1C<sub>0</sub>, and M1PC<sub>0</sub>) that contain anionic solubilizing groups attached via different linker lengths. As in SAMPL3 [159], these acyclic CB[n]-type receptors introduce conformational complexity, and water interactions play a key role. Moreover, the presence of 4 anionic groups near the cavity will likely impact the balance between ion-dipole interactions and solvation of the free host.

## Subaim 2.2. Gibb deep cavity cavitands for host-guest studies

**History of GDCC SAMPL challenges.** During SAMPL4 [192] and SAMPL5 [193] we focused on two specific GDCC hosts: the octa-acid 1 (R = H) and another octa-acid variant with four methyl groups at the portal of the binding pocket (1, R = Me). These studies used isothermal titration calorimetry (ITC) to measure the thermodynamics of (1) host 1 (R = H) binding a range of 9 carboxylate guests, and (2) the binding of 6 carboxylate



**Figure 6. Gibb deep cavity cavitands for the SAMPL6-10 datasets (SA 2.2).** These hosts bind a variety of carboxylate and trimethylammonium guests in a strongly salt-dependent manner, providing a stringent test of our ability to model salt-dependent binding.

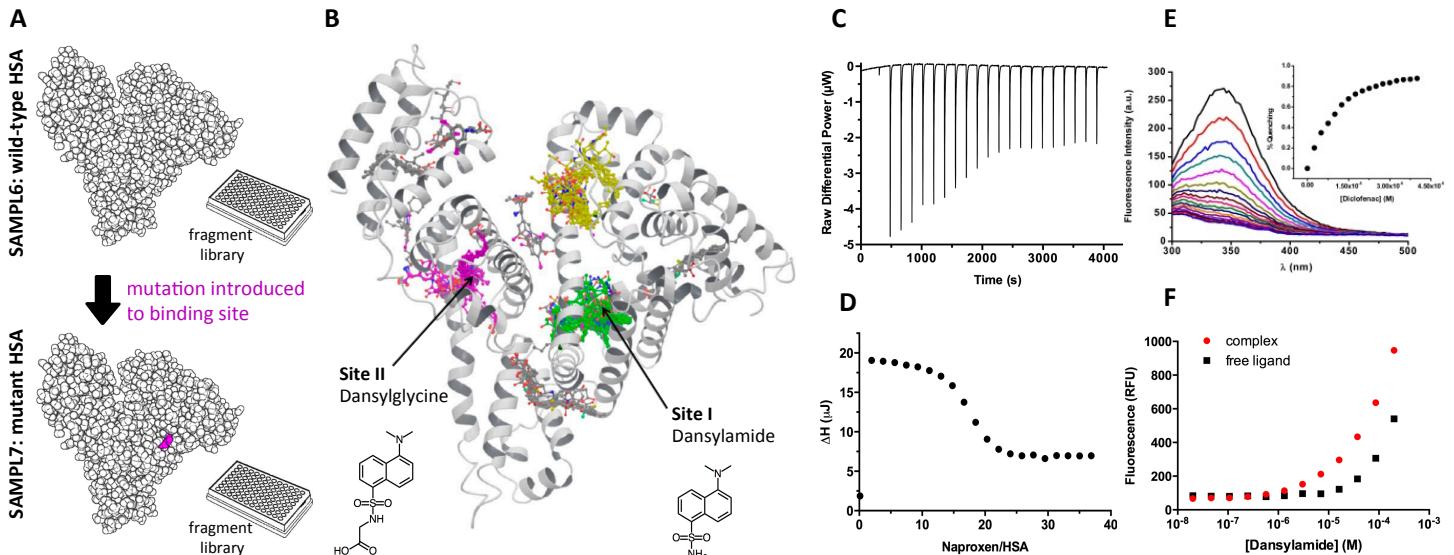
and trimethylammonium guests to both hosts (**1**, R = H and Me; Figure 6). In both cases <sup>1</sup>H-NMR titration was also used to confirm ITC-derived free energies of binding. As noted above, co-solvent effects and water rearrangements posed particular challenges for predicting binding in these hosts. SAMPL5 emphasized how differences in the shape of the hydrophobic pocket of the host can have a profound effect on affinity for some guests [129].

**Novel deep cavity hosts probe the effects of binding site charge constellations.** For future GDCC datasets, we will expand on the range of hosts by including **2** and **3** in our ITC studies (Figure 6). Like cavitand **1**, host **2** is an octa-acid derivative. However, the four benzoate groups are relocated from the extreme exterior in the case of **1**, to the rim of the binding pocket in **2**. We surmise that this will have a direct effect on the binding of charged guests as well as an indirect effect on guest complexation via changes to the solvation of the empty host. Octa-trimethylammonium cavitand (“positand” **3**) has the same overall architecture as host **1**, but inverts the charges on the water solubilizing exterior coat. While it is not yet clear if this switch in groups relatively remote from the pocket will directly affect guest complexation, results from related systems suggest it can (unpublished).

**SAMPL6-10 deep cavity cavitand datasets.** Data for SAMPL6 will focus on how well the effect of host carboxylate substituent location can be predicted, and will involve hosts **1** and **2** with a set of five previously uninvestigated guests. Guests will be selected from commercial sources on the basis of reference calculations in a similar manner to SAMPL7 in Subaim 2.1, specifically picking guests which have broad dynamic range and, here, have marked differences in affinities between hosts. SAMPL7 will provide a second iteration of this experiment to test algorithmic improvements in predictive modeling following SAMPL6 by comparing hosts **1** and **3** with a different set of guests. We anticipate that because of the relative remoteness of the charged groups in these two hosts, the effects of switching charges will be subtler than the differences between **1** and **2**. SAMPL8 will consider the effect of common biologically-relevant counterions/salts on guest binding, comparing the effects of NaCl and NaI on the complexation of five guests to **1**. We have previously shown that iodide has a weak affinity for the binding pocket of **1**, while sodium ions have an affinity for the outer carboxylates [194], requiring modeling to capture the differential affinities of these ions in addition to guest affinities to successfully model the observed affinities. SAMPL9 will follow up on this by examining the effects of these same two salts on the complexation of five guests to **3**, again giving the modeling community time to incorporate algorithmic improvements following SAMPL8. While we have not yet quantified salt affinities to host **3**, we expect the iodide to have affinity for both the pocket and the positively charged solubilizing groups. For SAMPL10 we will consider the effects of co-solvents on the binding of five guests to **1** and **2** to probe the effect of co-solvent competition for the binding site, as well as effects co-solvents may have in weakening the hydrophobic effect. While the number of guests considered in each challenge is relatively small, the total number of binding affinities measured is significant across the full family of hosts, meaning that the full data set will be of considerable value as a benchmark set [142].

### Aim 3: Develop model protein-ligand systems that isolate specific modeling challenges of drug targets.

We seek to drive advances in quantitative modeling of protein-ligand interactions. While D3R [120] benchmarks accuracy for targets of pharmaceutical interest, it does not provide a clear route to improving poor performance because the large number complexities exhibited by these targets make it difficult to identify clear points of failure [120, 130–132]. For example, while kinases are targets of great interest to drug discovery, blind challenges involving kinase targets conflate issues of slow protein conformational dynamics [195], protonation state effects of both protein [196] and ligand [197, 198], charged ligands, and the modeling of complex divalent salt environments and phosphorylation state effects along with the standard challenges of conformational sampling and forcefield accuracy. Thus D3R exercises serve the community well to understand current accuracy, but **blind challenges on complex pharmaceutical targets have limited ability to rapidly advance quantitative predictive modeling.**



**Figure 7. The SAMPL6/7 protein-ligand challenge focuses on soluble drug fragment binding to human serum albumin (HSA) (Aim 3).** (A) SAMPL6 will study binding of a library of 96 small soluble druglike fragments to recombinant HSA, with an engineered HSA mutant used for SAMPL7. (B) HSA has at least eight known binding sites, with two major well-characterized sites (green, Sudlow's Site I; purple, Site II) that bind a variety of drugs (figure from [199]). Two fluorescent probes—dansylamide and dansylglycine—bind with  $\sim\mu\text{M}$  affinity and high selectivity to Site I and Site II, respectively; both exhibit binding-enhanced fluorescence at 480 nm, and can be used to site-specifically probe ligand affinities by competition. (C, D) Binding affinities of soluble molecules can be measured by isothermal titration calorimetry (ITC); here, we show the (C) differential power and (D) integrated injection heats for the ITC titration of HSA by naproxen sodium collected using the Chodera lab automation pipeline; (E) HSA tryptophan fluorescence quenching can also be used to measure ligand binding affinity; here, HSA titration by diclofenac is shown, with the inset plot showing percent quenching at 346 nm [200, 201]. (F) Direct fluorescence binding assay of Dansylamide (fluorescent ligand) and HSA collected on the Chodera lab automation system. The binding curve can be constructed based on binding-induced fluorescence emission at 480 nm.

**We take the alternative approach of identifying and developing specific protein-ligand systems which isolate individual accuracy-limiting effects in a series of prospective challenges.** By developing *model binding systems*—real protein-ligand systems that may be of pharmacological interest, but comprised of single-domain proteins binding to a simple ligand series free of complex phenomena—we can study systems of complexity intermediate between completely artificial systems (such as the T4 lysozyme L99A system developed by Shoichet and Matthews [118, 142, 202]) and complex pharmaceutical targets where multiple modeling challenges make it difficult to learn from failure (Figure 8A). This process focuses the field on identifying and evaluating multiple solutions to selected accuracy-limiting effects (such as how to deal with ligand and protein protonation-state issues [203], slow protein conformational dynamics, etc.) while avoiding other complicating factors.

While model systems have had ample success in driving progress in individual research laboratories, community participation in blind challenges amplifies their power. For example, SAMPL3 featured the binding of small, rigid charged molecules to bovine trypsin [204], and rapidly focused the field on the deficiencies of current alchemical free energy methodologies in treating the binding of charged ligands. Within two years, multiple laboratories had developed practical solutions to effectively handle charged ligand binding [205–207].

**SAMPL6-10 model protein-ligand challenges.** We will introduce a new model protein-ligand system each year (revisiting the prior year's system if this becomes too difficult), with multiple challenges on each system (Figure 2) to allow iterative improvement and assessment. Our SAMPL6 data will focus on binding of small soluble drug fragments to one particular protein (below). However, maximizing gains in this area requires adapting subsequent challenges based on deficiencies identified by previous D3R/SAMPL challenges. Therefore, subsequent model systems will be rapidly identified and developed using our new informatics platform (below).

**SAMPL6: Assessing predictive modeling of binding to multiple weak sites via measuring fragment binding to human serum albumin (HSA).** HSA, the most abundant blood plasma protein, has a remarkable ability to bind a great variety of small molecule drugs in multiple binding sites (Figure 7B) [208]. As a result, HSA not only helps isolate the challenge of multiple weak ligands binding to a stable rigid protein, but it is also a pharmacologically relevant because of how it drastically modulates drug pharmacokinetics [199]. HSA has at least *eight* known binding sites, with numerous crystal structures available for drugs binding to two predominant sites (Site I and II) [199]. Small soluble molecules resembling drug fragments are highly likely to bind to HSA ( $\geq 90\%$  of such fragments, as detected by SPR [209]), providing an experimentally-tractable diverse set of ligands spanning several orders of magnitude in affinity [209]. As current advanced methodologies such as alchemical free energy calculations currently assume a single well-defined binding site with high affinity [210], this dataset will allow the

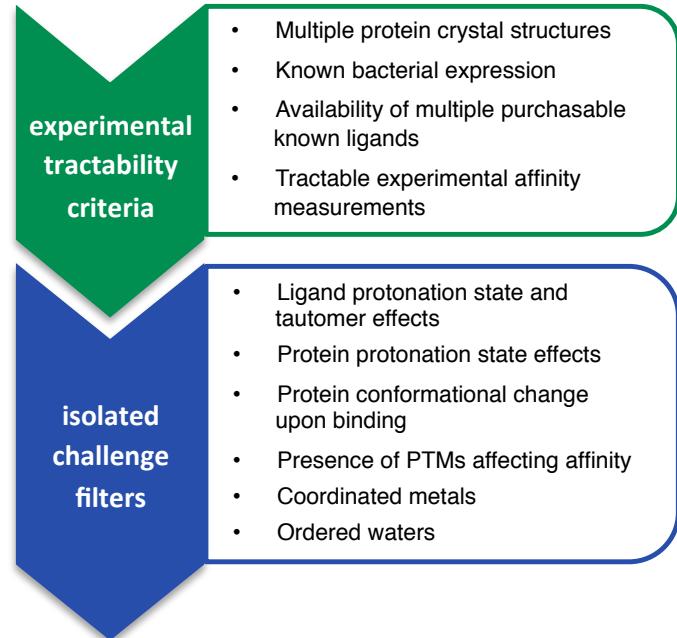
isolation of the effect of weak multiple binding from the majority of other confounding factors in protein-ligand binding. As HSA is relatively rigid, and computational methods already show some promise in computing binding affinities to HSA [199, 211, 212], this is an optimal model system for SAMPL6.

Recombinant HSA will be expressed in *E. coli* and purified via refolding from inclusion bodies [215], then defatted at low pH [216] to ensure the resulting protein is free of the glycosylation and bound fatty acids found in plasma-isolated HSA [216]. Recombinant expression will also allow a mutant form of HSA (engineered via single-primer mutagenesis) to be fielded for SAMPL7 (Figure 7). We will obtain a diverse library of 96 soluble drug-fragment-like molecules in pre-plated format as dry compound, and use our automated isothermal titration calorimetry (ITC) pipeline (Figure 7C,D) to characterize overall binding affinities to HSA. The same ligands pre-plated in DMSO format will be used to conduct a separate set of fluorescence titration assays (monitoring tryptophan fluorescence quenching, Figure 7D) and competition assays with site-specific fluorescent probes (Figure 7B) to resolve site-specific affinities to Sites I and II. We will field several levels of challenges, including challenges focused on affinities to Sites I and II, as well as challenges focused on predicting overall affinity and stoichiometry.

**SAMPL7-10: Rapid development of new tailored model systems using a novel informatics platform.** We are developing a novel informatics platform aimed at identifying protein targets that can be rapidly developed into experimentally- and computationally-tractable model systems focusing on individual challenges (Figure 8). This tool filters all known protein targets with structural data available in the PDB, first selecting for experimental tractability, then annotating experimentally tractable targets to determine which targets possess (or are likely to be free of) specific challenges for physical modeling. This will allow us to select systems which introduce only specific modeling challenges.

The Chodera lab has developed an automated wetlab to facilitate the development of such model protein-ligand systems using bacterial expression (see Equipment and Facilities). Potential targets matching desired challenge criteria will be screened for bacterial expression using high-throughput cloning, transformation, and expression testing, with purity and yield assessed by capillary electrophoresis on a Caliper GXII. Targets will be screened for stability in various buffers using ThermoFluor thermal shift assays [217]. Ligands identified via TargetExplorer as spanning a wide dynamic range of binding affinities will be purchased as dry powder stocks and prepared for assay by highly accurate gravimetric solution preparation techniques using a Quantos automated balance. Our lab has access to a wide variety of biophysical techniques for quantitative measurement of protein-ligand binding affinities, including fluorescence (if fluorescent probe ligands are available), absorption (e.g. Soret band shifts), automated isothermal titration calorimetry (provided ligands are sufficiently soluble), surface plasmon resonance, microscale thermophoresis (MST), luminescence, and alphascreen; all except MST are fully automated.

We take a twofold approach to developing challenge datasets: First, we will purchase and assay small molecules similar to known ligands, presuming that these molecules are likely to have measurable affinities. Second, using single-primer quick-change mutagenesis, we will introduce site-directed mutants to modulate the binding affinities of known ligands. This can be performed and screened for expression in 96-well format. Thus, datasets will consist of a matrix of protein mutants and ligands, providing opportunity to deeply explore the effects of interest.



**Figure 8. Mining model protein-ligand systems to focus on individual modeling challenges via a structural and chemical informatics platform (Aim 3).** We are developing a structural and chemical informatics system called TargetExplorer [<https://github.com/choderalab/targetexplorer>] that applies successive filters to all potential protein-ligand systems for which structural data is available. Suitable model systems should meet all experimental tractability criteria (green box) and possess only a few challenging properties, ideally only one (blue box). Tractability of experimental affinity measurements includes properties like known ligands with potentially fluorescent scaffolds (for fluorescence competition assays), highly soluble ligands (for ITC), or ligands above a minimal mass (for SPR or MST). Additional filters annotate experimentally tractable systems with their potential computational challenges, including charged ligands or potential ligand protonation state or tautomer effects [213] (deduced from predicted aqueous protonation/tautomer energies); potential protein protonation state effects (deduced from MCCE2 calculations [214]); protein conformational changes (deduced from variation in protein conformation or the presence of unresolved loops in protein-ligand crystal structures); the presence of post-translational modifications that may affect affinity (deduced from Uniprot annotations); coordinated metals (identified in crystal structures); and ordered waters (present in multiple crystal structures).

**Aim 4: Field community blind challenges to advance quantitative biomolecular design.** The value of the targeted datasets generated in Aims 1–3 will be amplified enormously by the strategic release of this data through iterative, coordinated SAMPL blind challenges (Figure 2). These blind challenges are designed to test the state of the art, provoke new methodological and force field innovations, allow comparative evaluation of methods, and drive downstream improvements. The new, progressive, targeted nature of the data generated means that SAMPL challenges will now build on one another, and for success in later challenges, participants must build on lessons learned from prior challenges. SAMPL challenges and subsequent data release activities will therefore facilitate rapid cycles of application, learning, and improvement. Each iteration will likely yield its own incremental benefits (e.g., as in Figure 3) for molecular design, in addition to contributing to progress towards our larger goals.

**SAMPL blind challenges.** Challenges will have yearly submission deadlines and involve roughly the same size data sets as prior challenges. The full timeline for SAMPL challenges (Figure 2) will be made available on the website [<https://drugdesigndata.org/about/samp1>] at the outset, allowing participants to plan their work and select what challenges to be involved in. As experimental data for each component becomes available and is curated, input files and challenge details will be made available at least six months prior to the challenge deadline; data not yet available at that time will be held for a subsequent challenge (with the exception of three months for year 1 due to startup timescales). As in prior SAMPLs, submissions will be handled by a web upload service on the SAMPL website (which will be migrated to separate hosting if the D3R effort is not renewed) which validates submissions to ensure that they meet format standards we specify along with the challenge details. As in SAMPL4 and SAMPL5, analysis will also be conducted by our automated Python framework, and results returned automatically online. All participant submissions and methodology descriptions will (as before) be made available publicly on the website, along with participant information (except for participants who specifically request to remain anonymous prior to submission). Aggregate statistics and historical performance will also be made available on our website, along with a record linking publications to historical submissions.

**Our goal is not just to run blind challenges, but to advance modeling by helping participants identify both modeling failures and potential solutions.** To achieve this, we provide guidance to participants as to what known modeling issues we expect may be relevant when providing details on each SAMPL component. For a host-guest system, for example, we might highlight known buffer/salt effects, protonation state challenges, and point out previous work on sampling challenges, with pointers to the relevant experimental work and to modeling work from past SAMPL challenges and elsewhere [142]. This helps participants design their approach. **Additionally, we will run reference calculations using current best practices.** This serves several purposes: It provides a test of the current methods and force fields we select; it helps facilitate learning—we announce what calculations we plan to perform, make input files available in a wide variety of formats [128, 129, 218], and others can repeat our calculations with a different method but same system and force field to compare methods, or swap force field but keep the method and system fixed to compare force fields, etc.; and it allows us to conduct sensitivity analysis, as by varying the conditions of our simulations (protonation state, tautomer, etc., [128]) we can see how much this impacts calculated values and thereby how important it is, even if participants don't do these tests. Reference calculations have, for example, helped us highlight the importance of a small amount of water in cyclohexane for accurately calculating log D values, show how an incorrect tautomer could affect calculated values by many log units [128], and discover that small forcefield modifications could significantly improve results on hydration free energies [126]. To further aid follow-up studies, we will make the input files, results, and simulation workflows used for our reference calculations—along with the data—available via GitHub and Docker Hub.

Physical methods are only valuable if they can reliably outperform alternate methods, so **a new focus of SAMPL6–10 will be selecting quality null models and running them to provide a point of comparison for participants**, going far beyond previous SAMPL nulls [127, 128, 137]).

Following submission and analysis of each SAMPL challenge, challenge results will be released and discussed, with SAMPL workshops allowing more formal presentations on and discussion of results in years 1, 3, and 5. Workshops will run every two years at the request of past participants, and will be co-run and co-hosted with D3R Grand Challenge workshops (see support letter). During the off years, SAMPL challenges will still run, but discussion of and dissemination of results will be via asynchronous means (as discussed below) and a “virtual workshop” consisting of talks and interaction over Google Hangouts or YouTube Live. While coordination with D3R will mostly be at the level of workshops, we will also ensure that SAMPL challenge submission deadlines are offset from D3R deadlines to allow maximum community participation in both efforts. If the D3R effort is not renewed beyond its current funding cycle, we will run SAMPL workshops independently, controlling costs via the model we use for our Workshops on Free Energy Methods in Drug Design—specifically, most participants will pay their own way to the workshop, and we will seek pharmaceutical and software industry sponsorships to defray costs.

**Dissemination of results and data.** Rapid dissemination of results is critical so that new insights can be used in subsequent challenges. We will continue to publish special issues of the *Journal of Computer Aided Molecular Design* (JCAMD) collecting publications related to each year's SAMPL challenges (see Letter of Support). To ensure immediate availability of reports, we will strongly encourage prepublication sharing of results and analysis, including both slides and posters from SAMPL meetings (via F1000 Research) and paper preprints (via bioRxiv). We also want to ensure that participants learn from one another by rapid exchange of ideas outside of formal workshops and meetings. While this has happened in the past—for example, when participants using similar methods work together after the SAMPL meeting to identify the origin of these discrepancies [129, 142, 160, 219, 220]—we hope to accelerate this kind of collaboration. To facilitate more open communication between the community, we will use collaboration software—such as Slack, which facilitates scientific communication for the NASA/JPL Mars Rover teams and NSF antarctic scientific research teams—to build a community discussion platform, facilitating a process of learning from one another more rapidly than normal publication channels.

**Each dataset will have a life cycle of collection, curation, blind challenges, and public dissemination.** In the past, the unfunded nature of SAMPL has forced us to primarily emphasize the *blind challenges* and pre-challenge *curation* aspects, with isolated forays into collection [156, 204]. This work now considers the full life cycle, with Aims 1–3 dealing primarily with collection and pre-challenge curation. Post-challenge, datasets will receive additional curation, then be released as standard test or benchmark sets that allow retrospective evaluation of methodologies on high-quality data [142]. The FreeSolv dataset, for example, includes a large number of calculated and experimental hydration free energies from SAMPL0–4, and provides a standard benchmark dataset for hydration free energy calculations [221]. Post-challenge curation will receive new attention here; in the past, lack of resources has always prevented follow-up experimental work, even when the data clearly indicated it was warranted (such as the puzzling issues with dynamic range for log D values in SAMPL5 [128, 156]). The requested budget will allow follow up experiments motivated by computation when warranted. Dissemination is the final stage in the data life cycle (see Resource Sharing Plan); we will make the data (including primary data, processed data, and our analysis of challenge submissions) available freely and publicly with permanent, citeable DOIs; ensure relevant data is deposited in standard community repositories (e.g. BindingDB [222]); and guarantee data longevity via backup hosting on library archival facilities (such as the UC's DASH (<https://dash.cdlib.org/>)).

We will also push for containerization of tools and methods in conjunction with other efforts such as AutoDesks's Molecular Design Toolkit and the NSF Molecular Sciences (MolSci) initiative. Our vision is that long-term, instead of participants submitting a set of predictions, they would also submit the entire workflow they applied via Docker containers allowing reproducibility and repurposing, ensuring dissemination workflows, not just results.

## COLLABORATION MANAGEMENT PLAN

We have a strong previous history of successful collaboration, with Mobley and Chodera having co-authored roughly a dozen publications and organized several workshops and other initiatives. Mobley, Isaacs, and Gibb have also worked together to coordinate past SAMPL challenges, and Mobley and Gibb a previous NSF workshop. PI Mobley will oversee the project, with teams for the other aims (Aim 1: Mobley & Chodera; Aim 2.1: Isaacs; Aim 2.2: Gibb; Aim 3: Chodera; Aim 4: Mobley & Chodera) involving the other co-investigators as needed. Meetings will consist of a monthly Google Hangout and a yearly in-person planning meeting. Chodera and Mobley will communicate more frequently due to the interlinked nature of their work. Publications are expected to be largely dictated by the overall Timeline, with an experimental publication associated with each challenge component being prepared for distribution to participants along with their results. Conflict resolution is expected to be straightforward, but if any serious difficulties arise, Michael Gilson (UCSD) will arbitrate given our close connections with D3R.

## OUTLOOK

Physical methods have been slow to achieve their promise in binding prediction, in part because truly significant innovations are so hard to recognize due to a lack of standard tests and benchmarks, and in part because of an “applications first” approach which seems to plague our community where we rush to apply our methods to problems of pharmaceutical relevance without ensuring they can tackle simpler, better-understood problems first. Here, we propose an innovative extension of the successful series of SAMPL blind challenges, generating novel experimental data to drive improvement of the methods in our field and help them become pharmaceutically relevant – beginning with relatively simple physical property prediction and progressing to challenging problems in biomolecular recognition via a series of carefully designed intermediate steps. SAMPL already has a strong track record of success, and funding will ensure dramatically increased impact and continued success. The proposed series of carefully tailored challenges will focus our community on a variety of problems which we can realistically resolve in the near term, resulting in dramatic improvements in computational molecular design.

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**BIOGRAPHICAL SKETCH**

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NAME: David L. Mobley

ERA COMMONS USER NAME (credential, e.g., agency login): dlmobley

POSITION TITLE: Associate Professor, Departments of Pharmaceutical Sciences and Chemistry

EDUCATION/TRAINING (*Begin with baccalaureate or other initial professional education, such as nursing, include postdoctoral training and residency training if applicable. Add/delete rows as necessary.*)

INSTITUTION AND LOCATION	DEGREE <i>(if applicable)</i>	Completion Date MM/YYYY	FIELD OF STUDY
University of California, Davis	B.S.	06/2000	Physics
University of California, Davis	M.S.	06/2002	Physics
University of California, Davis	Ph.D.	06/2004	Physics
University of California, San Francisco	Postdoctoral	2004-2008	Comp. chem./modeling

**A. PERSONAL STATEMENT**

My research focuses specifically on using molecular simulations to understand and predict molecular interactions, solvation, solubility, and other properties relating to molecular design. A major goal is enabling molecular design for applications in drug discovery, bioengineering and other areas. I also have major interests – and an extensive background – in using blind prediction challenges to drive progress in my own work and in the field. The present project focuses on developing model systems of varying complexity to span between physical modeling exercises which are tractable with today's methods and pharmaceutical applications which are, in some cases, far too complex; generating experimental data; and running blind challenges to use this data to advance predictive modeling.

My group is particularly well suited to this task, given our extensive experience with predictive modeling and blind challenges and long history with the SAMPL challenge itself. In my postdoctoral work with Ken Dill, I participated in informal blind predictive challenges with experimentalists in the Shoichet lab, where I predicted binding free energies and the Shoichet lab separated measured the same. This led me to early see the value of blind tests for method development, as they involve much more careful planning and analysis of protocols than do retrospective tests, and a great deal more is typically learned. This early involvement in blind challenges led to my participation in an OpenEye CUP blind prediction challenge on hydration free energies which we now refer to as SAMPL0, and subsequently I participated in every SAMPL challenge through SAMPL3 (where I also oversaw the hydration free energy component of the challenger), when SAMPL transitioned out of being run by OpenEye. At that point, I took over running SAMPL, running SAMPL4 largely within my group with help on host-guest systems from Michael Gilson (see support letter). Subsequently, I worked with John Chodera to coordinate experimental work at Genentech for log D data for SAMPL5, and with Mike Gilson to bring in host-guest measurements by Bruce Gibb and Lyle Isaacs. So my role in SAMPL0-3 was as a participant and partial organizer, and in SAMPL4-5 I've been the primary organizer with help from others.

In addition to my work on SAMPL, I have extensive experience applying free energy calculations and physical methods both to predicting binding and to prediction of physical properties, driving my group's role in performing reference calculations (Aim 4) for the challenges proposed in this work. My group consists of a postdoctoral researcher, eight graduate students, and several undergraduates, and we are actively involved in methodology and application of free energy calculations to biomolecular binding, drug discovery, and prediction

of solvation and solubility. We also have a major force field effort underway, currently in its early stages, which will have overlap with the science covered in this proposal.

The four publications below highlight our specific expertise for this project and focus on our role in, and the community's lessons learned from, the past two SAMPL challenges, of which I was an organizer:

- a. Rustenburg, A.S, Dancer, J., Lin, B., Feng, J.A., Ortwine, D.F., **Mobley, D.L.**, and Chodera, J.D. Measuring experimental cyclohexane-water distribution coefficients for the SAMPL5 challenge. *Journal of Computer Aided Molecular Design*, in press.
- b. **D. L. Mobley**, S. Liu, N. M. Lim, K. L. Wymer, A. L. Perryman, S. Forli, N. Deng, J. Su, K. Branson, and A. J. Olson, "Blind prediction of HIV integrase binding from the SAMPL4 challenge", *J. Comput. Aided Mol. Design* 28(4):327-345 (2014). PMCID PMC4331050.
- c. H. S. Muddana, A. T. Fenley, **D. L. Mobley**, and M. K. Gilson, "The SAMPL4 host-guest blind prediction challenge: an overview", *J. Comput. Aided Mol. Design* 28(4):305-317 (2014). PMCID PMC4053502
- d. **D. L. Mobley**, K. L. Wymer, and N. M. Lim, "Blind prediction of solvation free energies from the SAMPL4 challenge", *J. Comput. Aided Mol. Design* 28:135-150 (2014). PMCID PMC4006301

## B. POSITIONS AND HONORS

### **Positions and Employment**

06/2004 – 12/2007	Postdoctoral researcher, Dept. of Pharm. Chem., University of California, San Francisco
12/2007 – 06/2008	Chief Science Officer, Simprota Corporation, San Francisco, CA
06/2008 – 08/2008	Postdoctoral researcher, Dept. of Pharm. Chem., University of California, San Francisco
08/2008 – 06/2012	Assistant Professor, Department of Chemistry, University of New Orleans, LA
07/2012 – present	Adjunct Professor, Department of Chemistry, University of New Orleans, LA
07/2012 – 06/2014	Assistant Professor, Department of Pharmaceutical Sciences, Univ. of California, Irvine
12/2012 – 06/2014	Assistant Professor (joint), Department of Chemistry, Univ. of California, Irvine
07/2014 – present	Associate Professor, Department of Pharmaceutical Sciences, Univ. of California, Irvine
07/2014 – present	Associate Professor (joint), Department of Chemistry, Univ. of California, Irvine
06/2013 – present	Scientific Advisory Board, Schrödinger Software, New York, NY

### **Honors and Awards**

2014	National Science Foundation CAREER Award
2009	Hewlett Packard Outstanding Junior Faculty Award, American Chemical Society
2003	Graduate Fellowship, Kavli Institute for Theoretical Physics, UC Santa Barbara
2001-2003	National Science Foundation NEAT-IGERT Fellowship, UC Davis
2000-2001	UC Davis Physics Department Award
2000	Saxon-Patten Prize, UC Davis
2000	Departmental Citation in Physics, Physics Department, UC Davis
2000	Howard Hughes Medical Institute Science Teaching Internship

### **Other Experience and Professional Memberships**

American Chemical Society  
Biophysical Society

## C. CONTRIBUTIONS TO SCIENCE

To date, I have published more than 50 articles in peer-reviewed journals, which have collectively received over 3800 citations in the literature. My current h-index is 30, and my i10-index is 43.

**1. Predictive absolute binding free energy calculations:** Alchemical binding free energy calculations initially showed considerable promise for helping to guide early stage pharmaceutical drug discovery by predicting protein-ligand interactions in advance of the synthesis of new ligands. However, by the time I began my postdoctoral work early enthusiasm had waned and applications of these techniques were relatively rare, largely because of challenges relating to automation (addressed more below) and technical and conceptual

issues relating to the calculation of absolute binding free energies. However, publications in 1997 and 2003 removed the major challenges hampering application of absolute techniques and paved the way for new success with free energy calculations. My work in this area was the first to apply absolute binding free energy calculations prospectively, to make (experimentally verified) blind predictions in a series of three different binding sites. We also demonstrated the thermodynamic importance of key aspects of binding such as ligand orientational sampling and protein conformational changes. This work made absolute binding calculations practical for the first time, and helped spark a resurgence of interest in these techniques. My role in this work was to design, oversee, and in some cases conduct the research (the work began when I was a postdoc, and I actually conducted it, but then as I started my own group I oversaw the work of students on these projects).

- a. D. L. Mobley, A. P. Graves, J. D. Chodera, A. C. McReynolds, B. K. Shoichet and K. A. Dill, "Predicting absolute ligand binding free energies to a simple model site," *J. Mol. Biol.* **371**(4): 1118-1134 (2007).
- b. S. E. Boyce, D. L. Mobley, G. Rocklin, A. P. Graves, K. A. Dill, B. K. Shoichet. "Predicting ligand binding affinity with alchemical free energy methods in a polar model binding site", *J. Mol. Biol.* **394**: 747-763 (2009).
- c. G. J. Rocklin, S. E. Boyce, M. Fisher, I. Fish, D. L. Mobley, B. K. Shoichet\*, and K. A. Dill\*. "Blind prediction of charged ligand binding affinities in a model binding site", *J. Mol. Biol.* **425**(22):4569-4583 (2013).

**2. Automation of relative free energy calculations.** Relative free energy techniques were more established than absolute calculations, but applications were small in scale and limited due to the difficulty of setting up and conducting these calculations. We developed an automated approach for planning and setting up relative free energy calculations, and also improved automation of analysis. This allowed these calculations to be applied on a large scale, including in drug discovery applications, for the first time. I planned and oversaw the work on automated planning, with collaborators inside and outside my group, and then collaborated with colleagues in industry who implemented the approach in their software for industrial applications.

- a. S. Liu, Y. Wu, T. Lin, R. Abel, J. Redmann, C. M. Summa, V. R. Jaber, N. M. Lim, and D. L. Mobley, "Lead Optimization Mapper: Automating free energy calculations for lead optimization", *J. Comput. Aided Mol. Design* **27**(9):755-770 (2013).
- b. L. Wang, Y. Wu, Y. Deng, B. Kim, L. Pierce, G. Krilov, D. Lupyan, S. Robinson, M. K. Dahlgren, J. Greenwood, D. L. Romero, C. Masse, J. L. Knight, T. Steinbrecher, T. Beuming, W. Damm, E. Harder, W. Sherman, M. Brewer, R. Wester, M. Murcko, L. Frye, R. Farid, T. Lin, D. L. Mobley, W. L. Jorgensen, B. J. Berne, R. A. Friesner, and R. Abel. "Accurate and reliable prediction of relative ligand binding potency in prospective drug discovery by way of a modern free-energy calculation protocol and force field", *J. Am. Chem. Soc.* **137**(7):2695-2703 (2015).
- c. P. V. Klimovich, M. R. Shirts and D. L. Mobley, "Guidelines for the analysis of free energy calculations", *J. Comput. Aided Mol. Design.* **29**(5):397-411 (2015).

**3. Testing force fields using solvation free energies.** The accuracy of computed binding free energies depends crucially on the accuracy of the underlying force fields. We realized this and began using solvation free energies as a way to test and identify systematic errors in our force fields, with benefits for binding free energy calculations. We also used our work comparing to experiment in this area to develop a database of calculated and experimental hydration free energies that is widely used in a variety of contexts to help test and improve models. I planned, oversaw, and in some cases conducted this work, which took place in my group (after the initial study, which I did as a postdoc) with help from outside collaborators.

- a. D. L. Mobley, C. I. Bayly, M. D. Cooper, M. R. Shirts, and K. A. Dill. "Small molecule hydration free energies in explicit solvent: An extensive test of fixed-charge force fields", *J. Chem. Theory Comput.* **5**: 350-358 (2009).
- b. P. V. Klimovich and D. L. Mobley, "Predicting hydration free energies using all-atom molecular dynamics simulations and multiple starting conformations", *J. Computer-Aided Molecular Design* **24**: 307-316 (2010).
- c. D. L. Mobley, S. Liu, D. Cerutti, W. C. Swope, and J. Rice, "Alchemical prediction of hydration free energies for SAMPL", special issue, *J. Comput. Aided Mol. Design* **26**(5):551-562 (2012).
- d. C. J. Fennell, K. L. Wymer, and D. L. Mobley, "A fixed-charge model for alcohol polarization in the condensed phase, and its role in small molecule hydration", *J. Phys. Chem. B* **118**(24):6438-6446 (2014). Invited article.

- e. D. L. Mobley and J. P. Guthrie, "FreeSolv: A database of experimental and calculated hydration free energies, with input files," *J. Comput. Aided Mol. Design* **28**:711-720 (2014).

**4. Participating in and organizing blind SAMPL challenges:** As noted above, SAMPL blind predictive challenges have played a key role in our work since our first entrance into blind challenges in #1 above. In 2014, I actually transitioned to running the SAMPL challenge, soliciting predictions of a variety of properties from the computational community all over the world. This challenge provides, and will continue to provide, a valuable opportunity for testing and comparing different methods. In addition to the work noted under A, the following important publications have resulted from our work in SAMPL:

- a. C. C. Bannan, K. H. Burley, M. Chiu, M. R. Shirts, M. K. Gilson, and D. L. Mobley, "Blind prediction of cyclohexane-water distribution coefficients from the SAMPL5 challenge", *J. Comput. Aided Mol. Design*, in press.
  - b. T. S. Peat, O. Dolezal, J. Newman, D. L. Mobley, J. J. Deadman, "Interrogating HIV integrase for compounds that bind – a SAMPL4 challenge", *J. Comput. Aided Mol. Design* **28**:347-362 (2014).
  - c. **D. L. Mobley**, S. Liu, D. Cerutti, W. C. Swope, and J. Rice, "Alchemical prediction of hydration free energies for SAMPL", special issue, *J. Comput. Aided Mol. Design* **26**(5):551-562 (2012).
  - d. P. V. Klimovich and **D. L. Mobley**, "Predicting hydration free energies using all-atom molecular dynamics simulations and multiple starting conformations", *J. Computer-Aided Molecular Design* **24**: 307-316 (2010).

Full list of published work available at MyNCBI:

<https://www.ncbi.nlm.nih.gov/myncbi/browse/collection/43376427/?sort=date&direction=ascending>

#### **D. RESEARCH SUPPORT**

### **Ongoing Research Support**

1R01GM108889-01 (NIH)      Mobley (PI)      9/1/2014-8/31/2019

Alchemical free energy methods for efficient drug lead optimization

This study focuses on automating planning of protein simulation-based relative free energy calculations for drug lead discovery, on improving methods for sampling ligand and protein motions in molecular simulations, and on applying these techniques to several drug discovery projects.

Role: PI

CHE-1352608 (NSF)      Mobley (PI)      07/15/2014-07/14/2019

CAREER: Computing accurate free energies for solubility, solvation and transfer

This study focuses on developing and applying molecular-simulation-based techniques for prediction of solvation free energies, transfer free energies and transfer properties, and solubilities.

Role: PI

### ***Completed Research Support***

1R15GM096257-01A1 (NIH) Mobley (PI) 3/1/2012-2/28/2015

(At the University of New Orleans)

## Improving alchemical methods for predicting protein-ligand binding

The goal of this study is to test alchemical binding free energy techniques on three different binding sites, incorporating sampling enhancements relating to binding mode and sidechain sampling as needed.

Role: PI

1450865 (NSF) Mobley (PI) 02/15/2015-01/31/2016

## Accelerating our Understanding of Supramolecular Chemistry in Aqueous Solutions: A Workshop Proposal

This provided funds for a workshop in Arlington, VA in June, 2015 bringing together physical and supramolecular chemists to plan how to move the field forward and facilitate knowledge transfer between the physical and supramolecular chemistry communities.

Role: co-PI

**BIOGRAPHICAL SKETCH**NAME: **John D. Chodera**ERA COMMONS USER NAME: **JCHODERA**POSITION TITLE: **Assistant Member**, Computational Biology Program**EDUCATION/TRAINING**

INSTITUTION AND LOCATION	DEGREE	Completion Date	FIELD OF STUDY
California Institute of Technology	BS	06/1999	Biology
University of California, San Francisco	PhD	12/2006	Biophysics
Stanford University	Postdoc	2007-2008	Chemistry
QB3 Fellow, University of California, Berkeley	Postdoc	2008-2012	Quantitative Biosciences

**A. PERSONAL STATEMENT**

My research focuses on the use of rigorous statistical mechanics and physical modeling to develop predictive, quantitative computational models to enable rapid rational engineering of small molecule ligands for use as tool compounds for computational biology or potential therapeutics. This project focuses on developing model systems of complexity intermediate between trivial physical modeling exercises and complex pharmaceutical drug targets that focus on specific challenges to predictive physical modeling; generating datasets for these model systems with particular attention to experimental error and concordance between experimental and modeled systems; running blind challenges that focus the community in ways that significantly advance predictive modeling; and curating and disseminating this data and the results of these competitions.

My laboratory is particularly well suited to developing model protein:ligand systems of intermediate complexity. We have significant expertise in the structural and chemical bioinformatics necessary to mine public databases for potential model systems; we have developed an automated cloning and bacterial expression pipeline to screen for bacterial expression; we develop and employ high-quality automated biophysical assays and Bayesian experimental analysis techniques to measure binding affinities; and we both extensively utilize alchemical free energy calculations and co-organize the primary meetings for this field (Alchemical Free Energy Calculations in Drug Discovery) so are intimately aware of which model systems will provide maximal value to the community for blind challenges and high-quality datasets. In addition, we have experience in coordinating the measurement of physical property datasets by sending students to work with pharma collaborators, analyzing and curating this data, and releasing his data for blind challenges.

At the Sloan Kettering Institute, my laboratory consists of twelve theorists and experimentalists that combine theory, advanced simulation algorithms, high performance computing, and automated biophysical measurements to develop quantitative models for predicting and understanding how small molecules (such as drugs) modulate cellular pathways, how mutations lead to drug resistance, and how this resistance can be circumvented or suppressed. My laboratory has extensive experience in the use of alchemical free energy calculations for the computation of protein-small molecule binding affinities, and is actively engaged in efforts to scale our methodologies to aid in the design of high-affinity ligands that bind selectively to desired members of protein families. My laboratory makes heavy use of large-scale computational resources, including the Folding@Home distributed computing platform, national supercomputing resources, and high-performance GPU computing resources at MSKCC. I am also actively involved in developing new high-throughput protocols for high-quality, high dynamic range binding affinity and physical property measurements; laboratory automation techniques; experimental design guided by Bayesian inference and information theoretic principles; and the use of Bayesian inference and bootstrap simulation for accurate assessment of measurement error.

The four publications highlighting our specific expertise for this proposal demonstrate (a) our grasp of the challenges facing predictive modeling using modern free energy methods in drug discovery, (b) our early recognition of the value in iterative blinded challenges to drive improvements in quantitative predictive modeling, (c) our ability to collect useful blind challenge experimental datasets in collaboration with pharma, and (d) our focus on the thorough characterization of all sources of experimental error in affinity measurement.

- a. **Chodera, J.D.**, Mobley, D.L., Shirts, M.R., Dixon, R.W., Branson, K.M., and Pande, V.S. Free energy methods in drug discovery and design: Progress and challenges. *Current Opinion in Structural Biology*, 21:150-160, 2011. PMCID: PMC3085996
- b. Mobley, D.L., Graves, A.P., **Chodera, J.D.**, McReynolds, A.C., Shoichet, B.K., and Dill, K.A. Predicting absolute ligand binding free energies to a simple model site. *Journal of Molecular Biology*, 371:1118-1134, 2007. PMCID: PMC2104542
- c. Rustenburg, A.S., Dancer, J., Lin, B., Feng, J.A., Ortwine, D.F., Mobley, D.L., and **Chodera, J.D.**. Measuring experimental cyclohexane-water distribution coefficients for the SAMPL5 challenge. *Journal of Computer Aided Molecular Design*, in press.
- d. Hanson, S.M., Ekins, S., and **Chodera, J.D.**. Modeling error in experimental assays using the bootstrap principle: understanding discrepancies between assays using different dispensing technologies. *Journal of Computer Aided Molecular Design* 29:1073, 2015. PMCID: PMC4696763.

## B. Positions and Honors

### **POSITIONS AND EMPLOYMENT (current positions in bold)**

2005 IBM Almaden Research summer internship, Blue Gene project, under William C. Swope  
2007-2008 Postdoctoral Fellow, Department of Chemistry, Stanford University  
2008-2012 QB3 Distinguished Postdoctoral Fellow, University of California, Berkeley, Berkeley, CA  
2012-present **Assistant Member** and Laboratory Head, Computational Biology Program,  
Sloan Kettering Institute for Cancer Research, MSKCC (primary appointment)  
2013-present **Assistant Professor**, Program in Physiology, Biophysics, and Systems Biology,  
Weill Cornell Graduate School of Medical Sciences  
2013-present **Faculty Member**, Tri-Institutional PhD Program in Chemical Biology  
2013-present **Faculty Member**, Tri-Institutional PhD Program in Computational Biology and Medicine  
2015-present **Faculty Member**, Gerstner Sloan Kettering Graduate School of Medical Sciences, MSKCC

### **HONORS AND AWARDS**

2000-2005 Howard Hughes Medical Institute Predoctoral Fellowship  
2005 Frank M. Goyan Award for outstanding work in Physical Chemistry, UCSF  
2005-2006 IBM Predoctoral Fellowship  
2008-2012 QB3-Berkeley Distinguished Postdoctoral Fellowship  
2013-2106 Louis V. Gerstner Young Investigator Award

### **OTHER EXPERIENCE AND PROFESSIONAL MEMBERSHIPS**

2000-present Member, American Chemical Society  
2014-present Scientific Advisory Board, Schrödinger

## C. CONTRIBUTIONS TO SCIENCE

To date, I have published more than 50 articles in peer-reviewed journals, which have collectively received over 4950 citations in the literature. My current h-index is 30, and my i10-index is 46.

**1. Accurate alchemical free energy calculations of ligand binding affinities.** With the aim of enabling true computer-guided design of small molecules as potential therapeutics and chemical probes, I have spent the better part of a decade developing alchemical free energy methodologies into a quantitative, predictive tool for accurate computation of small molecule binding affinities to biomolecular targets. Work I have led or contributed to has benchmarked and improved the accuracies of free energy calculations, fixed deficiencies in methodologies, helped establish best practices, developed new efficient simulation algorithms, and exploited high-performance graphics computing hardware (GPUs) to greatly advance our progress toward this goal. We have made effective use of model systems and blind tests as a means of identifying systematic improvements in methodologies. Key papers demonstrate the capability of GPU-based free energy calculations to discover and compute affinities to new binding sites, review challenges facing the deployment of these techniques in drug discovery, address the problem of multiple kinetically-trapped conformational states contributing to binding, and demonstrate the power of cycles of experiment and computation to drive improvements.

- a. Wang, K., **Chodera, J.D.**, Yang, Y., and Shirts, M.R. Identifying ligand binding sites and poses using GPU-accelerated Hamiltonian replica exchange molecular dynamics. *Journal of Computer-Aided Molecular Design* 27:989, 2013. PMCID: PMC4154199
- b. **Chodera, J.D.**, Mobley, D.L., Shirts, M.R., Dixon, R.W., Branson, K.M., and Pande, V.S. Free energy methods in drug discovery and design: Progress and challenges. *Current Opinion in Structural Biology*, 21:150-160, 2011. PMCID: PMC3085996
- c. Mobley, D.L., **Chodera, J.D.**, and Dill, K.A. Confine-and-release method: Obtaining correct binding free energies in the presence of protein conformational change. *Journal of Chemical Theory and Computation*, 3:1231-1235, 2007. PMCID: PMC2562444
- d. Mobley, D.L., Graves, A.P., **Chodera, J.D.**, McReynolds, A.C., Shoichet, B.K., and Dill, K.A. Predicting absolute ligand binding free energies to a simple model site. *Journal of Molecular Biology*, 371:1118-1134, 2007. PMCID: PMC2104542

**2. Quantitative experimental biophysics.** I have been involved in the development of new techniques for the analysis of a variety of biophysical measurements. In the field of single-molecule force spectroscopy, I developed new techniques for the analysis of both nonequilibrium and equilibrium experiments. Working with force spectroscopists at UC Berkeley, I developed data analysis techniques crucial to demonstrating that nascent polypeptide chains translated by the ribosome have their folding properties modulated by electrostatic interactions with the ribosome (*a*), mechanical characterization of the molten globule state of a protein (*b*), and limitations of constant-force-feedback experiments (*c*). I have also worked extensively with biophysical techniques for the measurement of protein-ligand binding affinities, focusing on highly accurate quantitative measurements using techniques such as isothermal titration calorimetry (*d*).

- a. Kaiser, C., Goldman, D.H., **Chodera, J.D.**, Tinoco, I. Jr., and Bustamante, C. (2011) The ribosome modulates nascent protein folding. *Science* 334:1723. PMCID: PMC4172366
- b. Elms, P.J., **Chodera, J.D.**, Bustamante, C., Marqsee, S. (2012) The molten globule state is unusually deformable under mechanical force. *Proceedings of the National Academy of Sciences USA* 109:3796. PMCID: PMC3309780.
- c. Elms, P.J., **Chodera, J.D.**, Bustamante, C.J., Marqsee, S. (2012) Limitations of constant-force-feedback experiments. *Biophysical Journal*, 103, 1490, 2012. PMCID: PMC3471466
- d. Tellinghuisen, J., and **Chodera, J.D.** Systematic errors in isothermal titration calorimetry: Concentrations and baselines. *Analytical Biochemistry* 414:297, 2011.

**3. Biomolecular conformational dynamics and structural biology.** Biological macromolecules are not static entities, but populate a variety of kinetically metastable conformational states critical to binding and function. The long lifetimes of these metastable states present a challenge for molecular simulation, which are generally limited in length to a few microseconds. Together with collaborators at Stanford, the IBM Almaden Research Center, and the Freie Universität Berlin, I developed an approach to use *Markov state models* (MSMs) to build stochastic models of the long-time dynamics of biomolecules from many short atomistically-detailed molecular simulations. This technique allows for the characterization of thermally accessible metastable

conformational states, along with their associated interconversion kinetics and equilibrium free energies, and is now utilized by many laboratories around the world.

- a. Chodera, J.D., Signhal, N., Pande, V.S., Dill, K.A., and Swope, W.C. (2007). Automatic discovery of metastable states for the construction of Markov models of macromolecular conformational dynamics. *Journal of Chemical Physics* 126, 155101. PMID: 174616665
- b. Pitera, J.W. and Chodera, J.D. On the use of experimental observations to bias simulated observables. *Journal of Chemical Theory and Computation* 8:3445, 2012.
- c. Noé, F., Doose, S., Daidone, I., Löllmann, M., Sauer, M., Chodera, J.D., and Smith, J.C. (2011). Dynamical fingerprints: A theoretical framework for understanding biomolecular processes by combination of simulation and kinetic experiments. *Proceedings of the National Academy of Sciences USA* 108:4822, 2011. PMCID: PMC3064371
- d. Prinz, J.H., Wu, H., Sarich, M., Keller, B., Fischbach, M., Held, M., Chodera, J.D., Schütte, C., and Noé, F. (2011). Markov models of molecular kinetics: Generation and validation. *Journal of Chemical Physics* 134:174105. PMID: 21548671

**4. Advances in molecular simulation algorithms and methodologies.** Throughout my career, I have been active in the development of new algorithms to increase the efficiency of molecular simulations, establish best practices, benchmark and improve molecular mechanics forcefields, and exploit novel computing paradigms. Key advances include recognizing replica exchange simulations can be considered a form of Gibbs sampling (*a*), new estimators for combining simulation data from a variety of temperatures (*b*), the development of a new GPU-accelerated molecular simulation framework (*c*), and a simple solution to the longstanding problem of detecting when a simulation has sufficiently equilibrated (*d*).

- a. Chodera, J.D., and Shirts, M.R. Replica exchange and expanded ensemble simulations as Gibbs sampling: Simple improvements for enhanced mixing. *Journal of Chemical Physics* 135:194110, 2011. PMID: 22112069
- b. Prinz, J.H., Chodera, J.D., Pande, V.S., Swope, W.C., Smith, J.C., Noé, F. (2011) Optimal use of data in parallel tempering simulations for the construction of discrete-state Markov models of biomolecular dynamics. *Journal of Chemical Physics* 134, 244108. PMCID: PMC3139503
- c. Eastman, P., Friedrichs, M., Chodera, J.D., Radmer, R., Bruns, C., Ku, J., Beauchamp, K., Lane, T.J., Wang, L.P., Shukla, D., Tye, T., Houston, M., Stinch, T., Klein, C., Shirts, M.R., and Pande, V.S. OpenMM 4: A reusable, extensible, hardware independent library for high performance molecular simulation. *Journal of Chemical Theory and Computation* 9:461, 2012. PMCID: PMC3539733
- d. Chodera, J.D. A simple method for automated equilibration detection in molecular simulations. *Journal of Chemical Theory and Computation*, in press. Submitted to PMC.

**5. Nonequilibrium statistical mechanics.** The discovery of the Jarzynski equality (JE) in 1997 and the Crooks fluctuation theorem (CFT) in 1999 touched off a revolution in the field of statistical mechanics, providing for the first time exact relationships between the behavior of systems driven out of equilibrium and their equilibrium counterparts. I have been heavily involved in efforts to produce robust, reliable, and useful statistical estimators from these theorems, enabling the analysis of both nonequilibrium molecular simulations and real nonequilibrium biophysical experimental data to produce optimal estimates of physical properties like free energies and equilibrium expectations, along with good estimates of error (*a* and *b*). Together with Gavin Crooks and David Min, I developed a new efficient simulation methodology that exploits nonequilibrium driving--nonequilibrium candidate Monte Carlo (NCMC)---which can increase the acceptance probability of Monte Carlo in complex systems moves by orders of magnitude (*c*). More recently, we have shown how nonequilibrium theorems and estimators can yield new insight into the errors made in simulating physical systems by discretizing dynamical equations of motion for computer simulation (*d*).

- a. Minh, D.D.L and Chodera, J.D. Optimal estimators and asymptotic variances for nonequilibrium path-ensemble averages. *Journal of Chemical Physics* 131, 134110, 2009. PMCID: PMC2771048
- b. Minh, D.D.L. and Chodera, J.D. Estimating equilibrium ensemble averages using multiple time slices from driven nonequilibrium processes: Theory and application to free energies, moments, and

- thermodynamic length in single-molecule pulling experiments. *Journal of Chemical Physics* 134, 024111, 2011. PMID 21241084.
- c. Nilmeier, J.P., Crooks, G.E., Minh, D.D.L., and **Chodera, J.D.** Nonequilibrium candidate Monte Carlo is an efficient tool for equilibrium simulation. *Proceedings of the National Academy of Sciences USA* 108, E1009, 2011. PMCID: PMC3215031
  - d. Sivak, D.A., **Chodera, J.D.**, and Crooks, G.E. Using nonequilibrium fluctuation theorems to understand and correct errors in equilibrium and nonequilibrium simulations of discrete Langevin dynamics. *Physical Review X*, 011007, 2013.

Complete list of published work available at MyNCBI Collections:

<http://www.ncbi.nlm.nih.gov/sites/myncbi/john.chodera.1/bibliography/43349161/public>

## D. RESEARCH SUPPORT

### Ongoing Research Support

**I8-A8-058** (PI: Luo) 1/1/2015 – 12/31/2016

Starr Cancer Consortium

Designing Sinefungin Scaffolds as Specific Protein Methyltransferase Inhibitors

Our long-term goal is to develop PMT inhibitors for epigenetic cancer therapy, with the current objective to establish a drug-discovery pipeline with sinefungin analogues.

Role: Co-Investigator

**SK2015-0252** (PI: Chodera) 7/1/2015 – 10/31/2016

AstraZeneca

Evaluating the potential for Markov state models of conformational dynamics

Our goal is to evaluate the potential for Markov state models of conformational dynamics to describe the mechanism of slow off-rate inhibition in the human kinases CK2 and SYK.

Role: Investigator

**SK2016-0731** (PI: Chodera) 8/29/2016 – 8/28/2018

Merck KG

Benchmarking absolute alchemical free energy calculations with YANK

The objective of this project is to perform a large-scale benchmark of alchemical free energy calculations using the GPU-accelerated open source code YANK.

Role: Investigator

### Completed Research Support

None

**BIOGRAPHICAL SKETCH**

Provide the following information for the Senior/key personnel and other significant contributors.  
Follow this format for each person. **DO NOT EXCEED FIVE PAGES.**

NAME: Isaacs, Lyle

ERA COMMONS USER NAME (credential, e.g., agency login): ISAACS

POSITION TITLE: Professor of Chemistry

**EDUCATION/TRAINING** (*Begin with baccalaureate or other initial professional education, such as nursing, include postdoctoral training and residency training if applicable. Add/delete rows as necessary.*)

INSTITUTION AND LOCATION	DEGREE <i>(if applicable)</i>	Completion Date MM/YYYY	FIELD OF STUDY
University of Chicago	B.S.	06/1991	Chemistry
University of California, Los Angeles	M.S.	06/1992	Organic Chemistry
Swiss Federal Institute of Technology, Zurich	Ph.D.	08/1995	Organic Chemistry
Harvard University	Postdoc	06/1998	Supramolecular Chem.

**A. Personal Statement**

For his independent work at the University of Maryland, Lyle Isaacs is recognized as a world leader in the area of cucurbit[n]uril (CB[n]) molecular containers. The focus of this work has been manifold: 1) developing a thorough mechanistic understanding of the CB[n] forming reaction, 2) using this mechanistic understanding to create new members of the CB[n] family of molecular containers with exciting new structures, and 3) investigating their unique molecular recognition properties. For example, we have discovered the ultra tight affinity binding of CB[n] toward its best guests and used CB[n]-type receptors to regulate the enzymatic activity of bovine carbonic anhydrase, to perform homotropic allosteric binding, and to trigger the decloaking of cytotoxic nanoparticles in the cell. In more recent work, Isaacs and collaborators (Briken and Eikermann) have explored the use of acyclic CB[n] type molecular containers as: 1) a solubilizing excipient for insoluble drugs, and 2) as an *in vivo* reversal agent for neuromuscular block induced by rocuronium, vecuronium, and cis-atracurium. Acyclic CB[n] compounds (a.k.a. Calabadiols) do not display toxicity in both *in vitro* and *in vivo* assays which sets the stage for their further development as reversal agents for diverse biomedical applications. Over the past five years, the Isaacs group has provided blind challenge data for SAMPL3, SAMPL4, and SAMPL5 by synthesizing cyclic and acyclic CB[n]-type receptors and measuring their binding properties toward appropriate guest molecules. We have found the SAMPL challenges to be scientifically stimulating, and our participation has led our work in new directions. Accordingly, Dr. Isaacs has both the scientific expertise and a high level of enthusiasm to serve as a co-investigator on this project.

**B. Positions and Honors****Positions and Employment**

10/95 – 6/98	Harvard University, NIH postdoctoral fellow (with George M. Whitesides)
6/98 – 8/04	University of Maryland, Assistant Professor, Department of Chemistry and Biochemistry
8/04 – 8/08	University of Maryland, Associate Professor, Department of Chemistry and Biochemistry
8/08 – present	University of Maryland, Professor, Department of Chemistry and Biochemistry
7/10 – 6/13	University of Maryland, Director, Chemistry Graduate Program

8/12 – present Director, UMD Chemistry Graduate Assistance in Areas of National Needs (GAANN) Program

### **Other Experience and Professional Memberships**

1991 – present	Member, American Chemical Society
2001 – present	Member, American Academy of Arts and Sciences
2005 – present	Member, Maryland Nanocenter
2007	Organizer of the NSF Workshop on Cucurbit[n]uril Molecular Containers
2009 – present	Member, International Advisory Board of the International Conference on Cucurbiturils
2009 – present	Member of the Editorial Board of the Journal of Systems Chemistry
2010 – 2016	Member, International Advisory Board of the International Symposium on Macrocyclic and Supramolecular Chemistry
2013	Organizer of the International Symposium on Macrocyclic and Supramolecular Chemistry
2015	Local Co-Organizer, American Chemical Society, National Organic Symposium
2015	Co-organizer of the “Molecular Containers” symposium at Pacificchem 2015

### **Honors & Awards**

- U.S. Department of Defense Graduate Fellow (1991)  
Silver Medallion Dissertation Award (ETH Zürich, 1996)  
National Institutes of Health Postdoctoral Fellow (1996 – 1998)  
National Science Foundation Career Award (2000 – 2004, declined)  
Cottrell Scholar, Research Corporation (2001 – 2006)  
Junior Faculty Award, College of Life Sciences, University of Maryland (2001)  
Outstanding Invention of 2010 “*Molecular Container to Enhance Solubility of Drugs*” University of Maryland, Office of Technology Commercialization.  
Elected Fellow, American Association for the Advancement of Science (2013)

### **C. Contribution to Science**

Dr. Isaacs has made internationally recognized and highly cited contributions to science at all stages of his career development. Dr. Isaacs has published a total of 134 articles in peer reviewed journals. These publications have received over 8000 citations in the literature. Dr. Isaacs has an h-index of 47.

1. *Cucurbit[n]uril molecular container chemistry and recognition properties.* The Isaacs group is a world-leader in the area of cucurbit[n]uril molecular containers. Specifically, we are renowned for: 1) our investigation of the mechanism of CB[n] formation that allowed the creation of numerous new CB[n]-type receptors (e.g. chiral CB[n], acyclic CB[n], double cavity CB[n], chromophoric CB[n], monofunctionalized CB[n]), 2) our discovery of the remarkable affinity of CB[n] toward their guests in water (e.g.  $K_a$  up to  $10^{17} \text{ M}^{-1}$ ), 3) the use of the recognition properties of CB[n] in diverse applications (e.g. homotropic allosteric, chiral recognition, non-natural folding, chemical sensing, supramolecular polymers, drug delivery), and 4) our landmark review article that launched >100 groups to join the CB[n] field.

Lagona, J.; Mukhopadhyay, P.; Chakrabarti, S.; Isaacs, L. The Cucurbit[n]uril Family. *Angew. Chem. Int. Ed.* **2005**, *44*, 4844-4870. (review article; cited 1158 times)

Rekharsky, M. V.; Mori, T.; Yang, C.; Ko, Y. H.; Selvapalam, N.; Kim, H.; Sobransingh, D.; Kaifer, A. E.; Liu, S.; Isaacs, L.; Chen, W.; Gilson, M. K.; Kim, K.; Inoue, Y. A synthetic host-guest system achieves avidin-biotin affinity by overcoming enthalpy-entropy compensation. *Proc. Natl. Acad. Sci.* **2007**, *104*, 20737-20742. (cited 248 times)

Huang, W.-H.; Zavalij, P. Y.; Isaacs, L. Cucurbit[n]uril Formation Proceeds by Step-Growth Cyclo-Oligomerization. *J. Am. Chem. Soc.* **2008**, *130*, 8446-8454. (cited 53 times)

Cao, L.; Hettiarachichi, G.; Briken, V.; Isaacs, L. Cucurbit[7]uril Containers for Targeted Delivery of Oxaliplatin to Cancer Cells. *Angew. Chem. Int. Ed.* **2013**, *52*, 12033-12037. (cited 20 times)

**2. Biomedical Applications of Acyclic CB[n] Containers.** Dr. Isaacs research group used our mechanistic knowledge of CB[n] formation to create acyclic CB[n] containers that function well in two important biomedical application areas. First, a major problem facing the pharmaceutical industry is that many of their most promising drugs and drug candidates are so insoluble that they cannot be formulated on their own. Isaacs and Briken created acyclic CB[n] compounds that act as containers that solubilize and thereby formulate these drugs for in vitro and in vivo testing. Second, a major issue facing anesthesiologists is that residual neuromuscular block at the end of surgery results in patients having difficulty breathing which increases healthcare costs and mortality. Isaacs and Eikermann demonstrated that certain acyclic CB[n] compounds are able to sequester the neuromuscular blocking agent rocuronium from the bloodstream of rats and thereby allow them to regain their neuromuscular function rapidly. Acyclic CB[n] are under active development toward real world application.

Ma, D.; Hettiarachchi, G.; Nguyen, D.; Zhang, B.; Wittenberg, J. B.; Zavalij, P. Y.; Briken, V.; Isaacs, L. Acyclic Cucurbit[n]uril Molecular Containers Enhance the Solubility and Bioactivity of Poorly Soluble Pharmaceuticals. *Nat. Chem.* **2012**, 4, 503-510. (cited 82 times)

Ma, D.; Zhang, B.; Hoffmann, U.; Grosse Sundrup, M.; Eikermann, M.; Isaacs, L. Acyclic Cucurbit[n]uril Type Molecular Containers Bind Neuromuscular Blocking Agents in Vitro and Reverse Neuromuscular Block In Vivo. *Angew. Chem. Int. Ed.* **2012**, 51, 11358-11362. (cited 31 times)

Hoffmann, U.; Grosse-Sundrup, M.; Eikermann-Haerter, K.; Ayata, C.; Zhang, B.; Ma, D.; Isaacs, L.; Eikermann, M. Calabadion, a new broad spectrum agent to reverse the effects of benzylisoquinoline and steroid neuromuscular blocking agents. *Anesthesiology* **2013**, 119, 317-325. (cited 3 times)

Zhang, B.; Isaacs, L. Acyclic Cucurbit[n]uril-Type Molecular Containers: Influence of Aromatic Walls on their Function as Solubilizing Excipients for Insoluble Drugs. *J. Med. Chem.* **2014**, 57, 9554-9563. (cited 1 time)

**3. The conceptualization and demonstration of multi-component self sorting systems.** Researchers in the field of supramolecular chemistry take inspiration from Nature to create non-natural receptors that perform useful recognition, transport, or catalytic function. However, a widespread misconception was that supramolecular chemists synthetic systems were not as selective as natural systems and would be incapable of functioning in complex multicomponent mixtures. The Isaacs group shifted the viewpoint of the field in a series of papers that showed that mixtures of supramolecular receptors performed their functions faithfully even withing complex mixtures. The work has been highly cited and has launched numerous research groups worldwide to start research programs on self-sorting systems.

Wu, A.; Isaacs, L. Self-Sorting: The Exception or the Rule? *J. Am. Chem. Soc.* **2003**, 125, 4831-4835. (cited 196 times)

Mukhopadhyay, P.; Wu, A.; Isaacs, L. Social Self-Sorting in Aqueous Solution. *J. Org. Chem.* **2004**, 69, 6157-6164. (cited 114 times)

Liu, S.-M.; Ruspic, C.; Mukhopadhyay, P.; Chakrabarti, S.; Zavalij, P.; Isaacs, L. The CB[n] Family: Prime Components for Self-Sorting Systems. *J. Am. Chem. Soc.* **2005**, 127, 15959-15967. (cited 358 times)

Mukhopadhyay, P.; Zavalij, P. Y.; Isaacs, L. High Fidelity Kinetic Self-Sorting in Multi-Component Systems Based on Guests with Multiple Binding Epitopes. *J. Am. Chem. Soc.* **2006**, 128, 14093-14102. (cited 111 times)

**4. Development of the tether directed remote functionalization approach to regioselectively multiply functionalize buckminsterfullerene ( $C_{60}$ ).** My PhD dissertation with Prof. Francois Diederich focused on the functionalization of  $C_{60}$  as a means to tune the properties of this important class of compounds for specific applications. In the early 1990s, a major challenges to the field was that  $C_{60}$  features 30 reactive C=C double bonds and therefore numerous regioisomers can result when two or more groups are added to the carbon sphere. My research introduced the concept of tether directed remote functionalization to create bis, tris,

tetrakis, pentakis, and hexakis adducts of C<sub>60</sub>. The work stimulated significant follow up research in numerous labs worldwide.

Isaacs, L.; Haldimann, R. F.; Diederich, F. Tether-Directed Remote Functionalization of Buckminsterfullerene: Regiospecific Hexaadduct Formation. *Angew. Chem. Int. Ed. Engl.* **1994**, 33, 2339-2342. (cited 140 times)

Isaacs, L.; Seiler, P.; Diederich, F. Solubilized Derivatives of C<sub>195</sub> and C<sub>260</sub>: The First Members of a New Class of Carbon Allotropes C<sub>n(60 + 5)</sub>. *Angew. Chem. Int. Ed. Engl.* **1995**, 34, 1466-1469. (cited 76 times)

#### Complete List of Published Work in MyBibliography:

<http://www.ncbi.nlm.nih.gov/sites/myncbi/lyle.isaacs.1/bibliography/45414278/public/?sort=date&direction=asc>  
[ending](#)

#### D. Research Support

##### Ongoing Research Support

CHE-1404911 National Science Foundation “Synthesis and Application of Cucurbituril Type Receptors” This project explores the preparation of cucurbituril-type receptors and exploitation of their molecular recognition properties including ultratight host-guest binding, peptide and protein recognition, and the use of immobilized cucurbituril type receptors in bioanalytical assays. Role: PI. Dr. Isaacs is responsible for the conception of the projects, supervision of their execution, and their dissemination in the scientific literature.	Isaacs (PI)	09/01/14 – 08/31/18
1R01 CA168365-01A1 National Institutes of Health “Acyclic Cucurbit[n]uril Molecular Containers for Drug Solubilization and Delivery” Role: Isaacs (PI), Briken (co-PI). Dr. Isaacs is responsible for overseeing the design of new acyclic CB[n]-type receptors and investigation of their solubilization ability toward insoluble drugs as well as the administrative reporting and dissemination of the results in the literature.	Isaacs (PI)	04/01/2013 – 03/31/2018
P200A150033 Department of Education “UMD Chemistry GAANN” This grant provides funding to the Department of Chemistry and Biochemistry to support five students from the Chemistry graduate program as GAANN fellows each year. Role: PI, Jeffery Davis and Herman Sintim (Co-PIs).	Isaacs (PI)	09/01/2015 – 08/31/2018

**BIOGRAPHICAL SKETCH**

Provide the following information for the Senior/key personnel and other significant contributors.  
Follow this format for each person. **DO NOT EXCEED FIVE PAGES.**

NAME: Bruce C Gibb

ERA COMMONS USER NAME (credential, e.g., agency login): bkgibb

POSITION TITLE: Professor of Chemistry

**EDUCATION/TRAINING**

INSTITUTION AND LOCATION	DEGREE <i>(if applicable)</i>	Completion Date MM/YYYY	FIELD OF STUDY
Robert Gordon's University, Aberdeen, UK	B.Sc.	06/1987	Physical Sciences
Robert Gordon's University, Aberdeen, UK	Ph.D.	06/1992	Organic Chemistry
University of British Columbia, Vancouver	Postdoc	10/1994	Supramolecular Chem.
New York University, New York	Postdoc	07/1996	Bioorganic Chemistry

**A. Personal Statement**

The current application builds on our five-year collaboration with Profs. David Mobley (UC Irvine) and Michael Gilson (UC San Diego), which in turn stems from my 20 years of independent research on aqueous supramolecular chemistry. Our work with Mobley and Gilson has centered round the Statistical Assessment of the Modeling of Proteins and Ligands (SAMPL) exercise (OpenEye Scientific).<sup>1,2</sup> In this collaboration our group synthesizes host and guest molecules and then utilizes Isothermal Titration Calorimetry (ITC) to determine the thermodynamics of host-guest complexation. Our data is sealed until the group of computational chemists marshaled by Mobley and Gilson has, *a priori*, calculated the thermodynamics of the different binding events. Whilst this exercise has proven to be invaluable for computational chemists to test the predictive powers of their different approaches, the scrutiny our results receives also ensures exceedingly high standards of data collection within my own group; a fact that I consider essential to student training as research scientists.

The SAMPL exercise dove-tails with our ongoing research into a family of hosts we use<sup>3</sup> to probe the complex relationship between the Hydrophobic effect and solute shape,<sup>4</sup> and how salts influence aqueous solutions, i.e., the Hofmeister effect. Additionally, as is described below, we use our understanding of these phenomena to study molecular compartmentalization within yocto liter spaces, and how this can be used to bring about novel purification protocols and unusual chemical reactions.

1. Sullivan M. R., Sokkalingam P., Nguyen T., Donahue J. P., Gibb B. C. Binding of carboxylate and trimethylammonium salts to octa-acid and TEMOA deep-cavity cavitands. *J. Comput. Aided Mol. Des.*, 2016, SAMPL5 Special Issue. doi: 10.1007/s10822-016-9925-0. PubMed PMID: 27432339.
2. Gibb C. L. D., Gibb B. C. *Binding of cyclic carboxylates to octa-acid deep-cavity cavitand*. *J. Comput. Aided Mol. Des.*, 2014, 28(4), 319-25. doi: 10.1007/s10822-013-9690-2. PubMed PMID: 24218290; PubMed Central PMCID: PMC4018434.
3. Jordan J. H., Gibb B. C. *Molecular containers assembled through the hydrophobic effect*. *Chem Soc Rev*, 2015, 44(2), 547-85. doi: 10.1039/c4cs00191e. PubMed PMID: 25088697.
4. Hillyer M. B., Gibb B. C. *Molecular Shape and the Hydrophobic Effect*. *Annu. Rev. Phys. Chem.*, 2016, 67, 307-29. doi: 10.1146/annurev-physchem-040215-112316. PubMed PMID: 27215816.

## B. Positions and Honors

### Positions:

- 1996-2002 Assistant Professor, Department of Chemistry, Univ. of New Orleans (UNO)  
2002-2005 Associate Professor, Department of Chemistry, UNO, LA  
2005-2008 Professor of Chemistry, Department of Chemistry, UNO, LA  
2008-2011 University Research Professor of Chemistry, Department of Chemistry, UNO, LA  
2012- Professor of Chemistry, Tulane University, LA

### Other experiences and professional memberships:

- 1996- Member, American Chemical Society, American Association for the Advancement of Science, Royal Society of Chemistry  
1998-2005 Co-organizer, placement program for (30+) students from the IUT system in France  
2003- Co-chair of the 7th International Conference on Calixarenes, Vancouver, B.C., Canada, 08/03  
2004- Ad hoc Reviewer, NIH General Medicine ZRG1 BCMB-M (10) B study section  
2004- Invited participant, NSF Young Investigator Supramolecular Chemistry Workshop  
2006- Ad hoc Reviewer, NIH, Synthetic and Biological Chemistry: A (SBC-A) Study Section  
2007- Invited participant, NSF Workshop on Complexity and Emergent Phenomena in Chemistry  
2007- Member for the NIH, Synthetic and Biological Chemistry: A (SBC-A) Study Section (2007-2011)  
2009-2012 Associate Chair, Department of Chemistry, UNO, LA  
2009- Columnist, Nature Chemistry  
2010- Organizer, Supramolecular Chemistry Symposium, Joint SE/SW Regional ACS meeting, New Orleans  
2011- Member, International Advisory Committee, International Conference on Calixarenes  
2013- Co-Editor-in-Chief, *Supramolecular Chemistry* (Taylor and Francis Group)  
2013- Organizer and co-founder of the C. David Gutsche Award, a biennial \$5000 award for senior researchers who have made a significant impact in the field of calixarene chemistry  
2014- Chair: Mardi Gras Symposium, Tulane University; a one-day symposium with 7 international speakers and 105 attendees (2014)  
2015- Visiting Professor, Wuhan University of Science and Technology as a Chair Professor of the Chutian Scholars Program (2015-2017)  
2015- Member, International Scientific Committee for the International Symposium on Macrocyclic and Supramolecular Chemistry (ISMSC) series  
2015- PI, NSF Workshop; "Accelerating our Understanding of Supramolecular Chemistry in Aqueous Solutions"  
2015- Consultant, Statistical Assessment of the Modeling of Proteins and Ligands (SAMPL), OpenEye Scientific

### Honors and Awards:

- 1996 UNO Scholar Award  
1998 Research Corporation, Research Innovation Award  
1999 UNO Scholar Award  
2001 UNO Science Research Development Award  
2008-2011 UNO University Research Professor

## C. Contributions to Science

**The Hofmeister Effect.** In large part because of improved molecular dynamics simulations and new surface-specific spectroscopic techniques, the last decade or so has seen a revival in the age-old problem of understanding the Hofmeister effect, i.e., the ubiquitous phenomenon whereby, irrespective of the physicochemical technique, ranges of salts are seen to influence the properties of water or aqueous solutions in predictable ways. As outlined in this application, our recent research with deep-cavity cavitands has affected science's understanding of the Hofmeister effect by unequivocally demonstrating that even under unfavorable conditions, anions have a measurable and significant affinity for hydrophobic surfaces.<sup>1-4</sup> Furthermore, we have shown that anion binding influences the thermodynamics of (hydrophobic) guest binding to hosts in a manner that parallels the Hofmeister series.

From a personal perspective, understanding the Hofmeister effect is a large component of aqueous supramolecular chemistry. Yet the two groups of scientists who respectively study the Hofmeister effect and supramolecular chemistry in water have essentially remained separate. It is for this reason that our work has not only involved fundamental research at the boundary between these two fields, but has also involved efforts to bring these two fields together. This application – involving collaboration between a supramolecular chemist (the PI), a leader in molecular simulations and statistical thermodynamic theory of aqueous solutions (Ashbaugh), and an expert in hydration-shell vibration spectroscopic methods, densitometry, and theory (Ben-Amotz) – is one such local example of bringing these fields together. On a larger scale, the PI recently led a NSF supported workshop entitled, “*Accelerating our Understanding of Supramolecular Chemistry in Aqueous Solutions*”. Held last May in Washington DC, this meeting brought together sixteen water scientists (physical chemists, computationalists, instrumentalists, physicists and biophysicists) and sixteen supramolecular chemists (physical-(in)organic chemists) to discuss each side’s approach to solving mutual/related problems in the field. One metric of the success of this meeting was that an exit survey suggested twenty-four new collaborations between pairs of attendees in the two fields. Further such endeavors are ongoing.

- 1) Sokkalingam P., Shrberg J., Rick S. W., Gibb B. C. Binding Hydrated Anions with Hydrophobic Pockets. *J. Am. Chem. Soc.*, **2016**, 138(1), 48-51.
- 2) Gibb C. L. D., Oertling E. E., Velaga S., Gibb B. C. *Thermodynamic Profiles of Salt Effects on a Host-Guest System: New Insight into the Hofmeister Effect*. *Journal of Physical Chemistry B*, **2015**, 119(17), 5624-38
- 3) Carnegie R., Gibb C. L. D., Gibb B. C. *Anion Complexation and The Hofmeister Effect*. *Angew. Chem. Int. Ed.*, **2014**, 53(43), 11498-500
- 4) Gibb C. L. D., Gibb B. C. *Anion binding to hydrophobic concavity is central to the salting-in effects of Hofmeister chaotropes*. *J. Am. Chem. Soc.*, **2011**, 133(19), 7344-7

**Assemblies driven by the Hydrophobic Effect.** With the exception of cyclodextrin chemistry, which had mostly peeled away from supramolecular chemistry and moved decidedly into a more applied realm, when I began my independent career, supramolecular chemistry was primarily studied in organic solvents. State-of-the-art in regards to self-assembled hosts/supramolecular containers were the (hydrogen bonded) Rebek “jelly doughnut” and (metal coordinated)  $L_4M_6$  cages from the Fujita lab. My goal was to work in water, and I wanted to avoid these two common strategies for assembly – hydrogen-bonding and metal coordination – and use the Hydrophobic effect to drive the formation of structurally specific, mono-dispersed supramolecular containers large enough to encapsulate sizable guests. We first designed and synthesized more straightforward hosts soluble in organic solvents, before devising the first host that relied solely on the Hydrophobic effect to drive assembly.<sup>1</sup> This deep-cavity cavitand, as well as related hosts we have synthesized, define a dry, yocto-liter sized environment by dimerizing around guest molecules. Their ease of synthesis, combined with their encapsulation properties, is reflected in the fact that at least six laboratories around the world have independently worked with these types of hosts.<sup>1</sup> Since that time, others have entered this new field with their own self-assembling systems; most notably Mitsuhiro Shionoya (U. Tokyo) and Michito Yoshizawa (Tokyo Institute of Technology). Our research has now expanded to include the formation of tetrameric and hexameric assemblies that represent the largest such structures devised to date.<sup>2,3</sup> *En masse*, these supramolecular containers have provided valuable information about the structural requirements for controlled assembly using the Hydrophobic effect, and in conjunction with the concomitantly developing cucurbiturils, are shedding light on how the Hydrophobic effect can be orchestrated to control supramolecular properties.<sup>4</sup>

- 1) Jordan, J. H., Gibb, B. C., Molecular containers assembled through the hydrophobic effect, *Chem. Soc. Rev.*, **2015**, 44 (2), 547-85
- 2) Gan H., Benjamin C. J., Gibb B. C. *Non-monotonic Assembly of a Deep-Cavity Cavitand*. *J. Am. Chem. Soc.*, **2011**, 133, 4770-3
- 3) Gan H., Gibb B. C. *Guest-Mediated Switching of the Assembly State of a Water-Soluble Deep-Cavity Cavitand* *Chemical Communications*, **2012**, 49, 1395-7.
- 4) Hillyer M. B., Gibb B. C. *Molecular Shape and the Hydrophobic Effect*. *Annu. Rev. Phys. Chem.*, **2016**, 67, 307-29.

**Controlling Photochemistry and Photophysics.** In collaboration with V. Ramamurthy (U. Miami) we have advanced the ability to control photochemical and photophysical processes in aqueous solutions.<sup>1-4</sup> Our self-assembling containers have opened up the range of processes that can be studied in water, and brought a new level of control in regards to not only cage-effects, but also to the regioselectivity and stereoselectivity of these processes. In short, we have demonstrated that the hydrophobic effect can be orchestrated to bring about selective complexation and assembly so as to force photochemical and photophysical processes down

pathways that do not exist in free solution (or even in structurally defined media such as zeolites). Furthermore, our work demonstrated for the first time cross-talk between supramolecular complexes that mimic chemical signaling between enzymes.<sup>3</sup>

- 1) Kaanumalle L. S., Gibb C. L. D., Gibb B. C., Ramamurthy V. *Controlling Photochemistry with Distinct Hydrophobic Nano-Environments*. *J. Am. Chem. Soc.*, **2004**, 126, 14366-7
- 2) Kaanumalle L. S., Gibb C. L. D., Gibb B. C., Ramamurthy V. *A Hydrophobic Nano-Capsule Controls the Photophysics of Aromatic Molecules by Suppressing their Favored Solution Pathways*. *J. Am. Chem. Soc.*, **2005**, 127, 3674-5.
- 3) Natarajan A., Kaanumalle L. S., Jockusch S., Gibb C. L. D., Gibb B. C., Turro N. J., et al. *Controlling Photoreactions with Restricted Spaces and Weak Intermolecular Forces: Remarkable Product Selectivity during Oxidation of Olefins by Singlet Oxygen*. *J. Am. Chem. Soc.*, **2007**, 129, 4132-3. See also commentary: Greer A. *Molecular Cross-Talk*. *Nature*, **2007**, 447, 273-4.
- 4) Gibb, C. L. D.; Sundaresan, A. K.; Ramamurthy, V.; Gibb, B. C., Temptation of the Excited-State chemistry of  $\alpha$ -(n-Alkyl) Dibenzyl Ketones: How guest packing with a nano-scale supramolecular capsule influences photochemistry. *J. Am. Chem. Soc.* **2008**, 130, 4069-4080

**Molecular Protections and Separations.** The supramolecular systems devised in the group have many unusual and unique properties resulting from their ability to engender compartmentalization. One area that we have helped pioneer, and an area that is very much ongoing in our lab, is to use these nano-containers as a means to bring about molecular separations. Although these processes can be made more complex by the binding of two different guests within a container,<sup>1</sup> we have reported on a number of different approaches. One example is the physical separation/purification of hydrocarbon gases using aqueous solutions.<sup>2</sup> Thus, when a mixture of, for example, propane and butane is added to the head space above a solution of one of our water-soluble hosts, it is observed that only butane is pulled into solution (to form the corresponding 2:2 host-guest complex), leaving behind an enriched/pure sample of propane in the gas-phase. Such purifications are beyond the state-of-the-art of semi-permeable membrane technology. We have also demonstrated how molecular containers can be used to bring about chemical separations of normally difficult-to-separate mixtures.<sup>3</sup> One example of such kinetic resolutions is to add a mixture of two guests to a hydrolytic solution containing a host. The guest that binds preferentially is protected within the nano-container, whilst the less complementary guest is exposed to the reactive solution and hydrolyzed. This type of kinetic resolution via molecular protection represents a new strategy for separating complex mixtures.

- 1) Sullivan M. R., Gibb B. C. *Differentiation of small alkane and alkyl halide constitutional isomers via encapsulation*. Special issue on aqueous supramolecular chemistry, *Org. Biomol. Chem.*, **2015**, 13(6), 1869-77
- 2) Gibb C. L. D., Gibb B. C. *Templated Assembly of Water-Soluble Nano-Capsules: Inter-Phase Sequestration, Storage, and Separation of Hydrocarbon Gases*. *J. Am. Chem. Soc.*, **2006**, 128(51), 16498-9
- 3) Liu, S.; Gan, H.; Hermann, A. T.; Rick, S. W.; Gibb, B. C., Kinetic Resolution of Constitutional Isomers Controlled by Selective Protection inside a Supramolecular Nanocapsule. *Nature Chemistry* **2010**, 2 (10), 847-852

**Editorship.** As co-editor in chief of *Supramolecular Chemistry* (along with Philip Gale, U. of Southampton) I am fortunate to be in a position to help shape the future of the field. Although relatively specialized, the journal provides a valuable forum for countless authors from both developed nations and newly industrialized countries such as China and India. Moreover, the journal works closely with a number of international conferences including: the International Symposium on Macrocyclic and Supramolecular Chemistry, the International Conference on Molecular Sensors and Molecular Logic Gates, the International Conference on Calixarenes, and the International Conference on Cucurbiturils. Through these collaborations the journal helps support these meetings and produces special issues dedicated to each conference. Relatedly, the journal also publishes thematic special editions on a range of subjects, as well as those dedicated to honoring leaders in the field. In addition, we also publish invited “big picture” essays from world leaders on their opinions of the state-of-the-art and future of the field. In combination, these different activities mean that *Supramolecular Chemistry* is not only a key forum for everything supramolecular, but that it also helps glue, shape and direct the field.

Complete list of Published Work: <http://www.gibbgrou.org/blog/wp-content/uploads/2013/08/Gibb-CV.pdf>

#### D. Research Support

NIH R01 GM098141      Gibb (PI)      06/01/12-02/28/17  
*Using Deep-Cavity Cavitands to Study Supramolecular Chemistry in Water*  
The goal of this study is to contribute to our understanding of the Hofmeister Effect by probing how anions influence binding events driven by the Hydrophobic effect.  
Role: PI

NSF CHE 1450865      Gibb (PI)      02/15/15-01/31/17  
*Accelerating our Understanding of Supramolecular Chemistry in Aqueous Solutions: A Workshop Proposal*  
The goal of this study was to organize a workshop to bring together two separate groups of scientists: water and aqueous solutions scientists, and supramolecular chemists.  
Role: PI

NSF CHE 1507344      Gibb (PI)      08/15/15-08/14/18  
*Nano-Containers for Novel Separations and Reactions*  
The goal of this study is to investigate self-assembled dimeric nano-capsules as yocto-liter reaction flasks and as containers for the separation of complex mixtures of small (< C20) molecules.  
Role: PI

NSF CBET 1403167      Ashbaugh (PI)      07/01/14-06/30/17  
*Capacious Deep-Cavity Cavitand/Hydrophobic Guest Assemblies with Tunable Interior Volumes*  
The goal of this grant is to understand how hydrophobe shape influences the Hydrophobic effect and how this in turns influences the formation of self-assembled supramolecular containers.  
Role: co-PI

NSF 143280      Khonsari (PI)      08/01/14-07/31/17  
*The Smart MATerial Design, Analysis, and Processing (SMATDAP) consortium: Building next-generation polymers and the tools to accelerate cost-effective commercial production*  
The goals of this award are to develop new, real-time and multi-channel analysis of polymerization processes, particularly in regard to the development of supramolecular polymers.  
Role: co-PI

**Completed Research Support**  
NIH 1S10OD020117-01      Jayawickramarajah (PI)      03/09/15-03/08/16  
*Acquisition of a Surface Plasmon Resonance Biosensor: Enhancing Biomedical (Bio)Molecular Recognition Projects*  
This was an equipment award for the purchase of a new Surface Plasmon Resonance.  
Role: co-PI

Louisiana Board of Regents (LEQSF(2015-16)-ENH-TR-39      Grayson (PI)      03/09/15-03/08/16  
*Nano-materials Separation and Characterization Equipment*  
This was an award for the purchase of new GPC equipment.  
Role: co-PI

NSF CHE 1244024      Gibb (PI)      05/31/12-07/31/15  
*Studies of Deep Cavity Cavitands*  
The goals of this award were to: devise multi-(supra)molecule systems possessing unusual relationships between the components, the environment and the net overall state of the system, and study encapsulation approaches to the separation of constitutional isomers.  
Role: PI

NSF CHE 1228232      Mague (PI)      09/01/12-08/31/15  
*MRI: Acquisition of a Single-Crystal, Micro-Source Diffractometer*  
This was an award for the purchase of new diffractometer.  
Role: co-PI