# SAMPL6 pKa reference calculations using Epik and Jaguar

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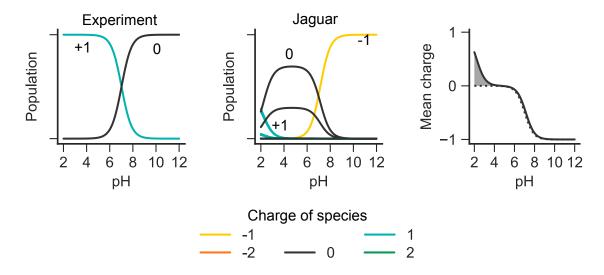
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# **NOTES: Objectives for this manuscript**

track down type II artifacts, and dont use type I labels, use epik scan, sequential, et cetera

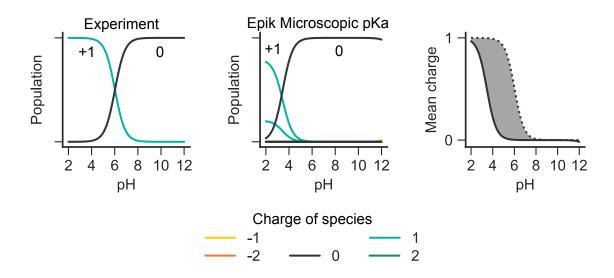
- Provide a baseline for the expectation of performance of empirical models (Epik) and DFT models (Jaguar) on kinase inhibitor-like molecules and molecular fragments; draw comparisons to earlier benchmarks for these tools
- · Describe basic concepts, along with strengths and weaknesses of each approach (Epik and Jaguar)
- Lessons learned for participating in the challenge
  - How should we predict macroscopic pKas from microscopic pKas? Explain possibilities and justify choice
    - \* Sequential titration (Epik)
    - \* <n protons>/charge vs pH inflection points: Matches electrochemical titration, but not UVmetric
    - \* Is there a way to predict what UV-metric would observe from populations vs pH?
- For microscopic pKa prediction: Epik may enumerate a different set of microstates than other tools (we proposed these as newly labeled states)
- Epik does not return state populations below a certain threshold, so we could not consider states proposed by SAMPL6 [Check this]
- For worst predicted compounds, provide an explanation as to why the methods may perform poorly (descriptors?) [ may not have enough data points to draw conclusion]

# Figures and Tables

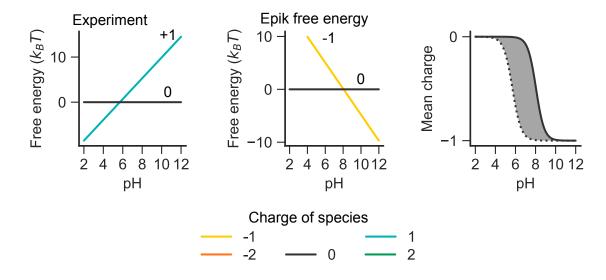


**Figure 1. Comparing UV experiments for SM03 to microscopic pKa Jaguar predictions**. This example shows how the UV experimental populations (left) for macrostates molecule SM03 as a function of the pH can be compared to Jaguar predictions. Jaguar was used to predict microscopic pKas (denoted as Type I predictions in SAMPL6), which can be used to derive relative microstate populations (middle). In the UV experiment, charge reflects the number of protonations, since absolute charge information is unknown. The charge combined with populations produces a macroscopic titration curve (right), with the prediction as a solid, and the experiment as a dashed line. The area between the curves (gray) is calculated after correcting for an integer offset in charges.

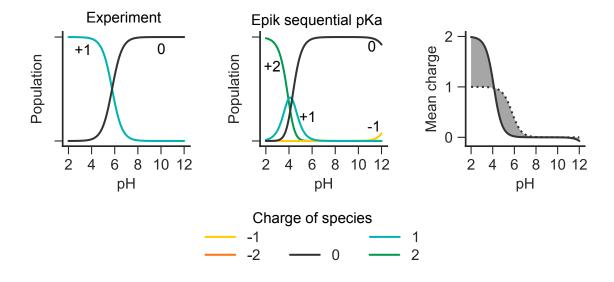
ASR: Add number of area to plot



**Figure 2. Comparing UV experiments for SM04 to Epik microscopic pKa predictions**. This example shows how the UV experimental populations (left) for macrostates molecule SM04 as a function of the pH can be used to compare to microscopic pKa Epik predictions. Epik was used to predict microscopic pKas (denoted as Type I predictions in SAMPL6), which can be used to derive relative microstate populations (middle). In the UV experiment, charge reflects the number of protonations, since absolute charge information is unknown. The charge combined with populations produces a macroscopic titration curve (right), with the prediction as a solid, and the experiment as a dashed line. The area between the curves (gray) is calculated after correcting for an integer offset in charges.



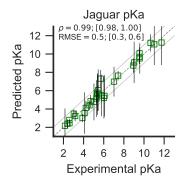
**Figure 3. Comparing UV experiments for SM20 to Epik microscopic free energy predictions**. This example shows how the UV experimental free energy (left) for macrostates molecule SM20 as a function of the pH can be compared to microscopic free energy Epik predictions. Epik was used to predict microscopic free energy (denoted as Type II predictions in SAMPL6) (middle). In the UV experiment, charge reflects the number of protonations, since absolute charge information is unknown. The charge combined with free energies produces a macroscopic titration curve (right), with the prediction as a solid, and the experiment as a dashed line. The area between the curves (gray) is calculated after correcting for an integer offset in charges.



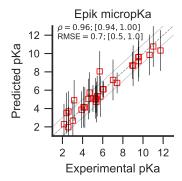
**Figure 4. Comparing UV experiments for SM13 to Epik macroscopic pKa predictions**. This example shows how the UV experimental populations (left) for macrostates molecule SM13 as a function of the pH can be compared to sequential pKa Epik predictions. Epik was used to predict macroscopic pKa values (denoted as Type III predictions in SAMPL6), which can be translated into relative macrostate populations (middle). In the UV experiment, charge reflects the number of protonations, since absolute charge information is unknown. The charge combined with populations produces a macroscopic titration curve (right), with the prediction as a solid, and the experiment as a dashed line. The area between the curves (gray) is calculated after correcting for an integer offset in charges.

**Table 1.** Overall performance of each method using either the closest pKa matching, Hungarian pKa matching, and the area between the experimental and predicted macroscopic charge titration curve

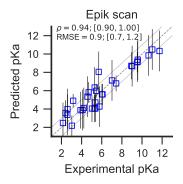
	C	losest		Hungarian			Titration curve
	RMSE	MAE	$R^2$	RMSE	MAE	$R^2$	Δ area
Epik sequential							
Epik micropka							
Jaguar micropKa							



**Figure 5. Comparison of microscopic Jaguar pKa values to experimental macroscopic pKa values.** Microscopic pKa values for Jaguar were matched of the experimental dataset using algorithm 1. An alternative alignment of pKa values using the Hungarian algorithm (**??**) is available as a supplementary figure.



**Figure 6. Comparison of microscopic Epik pKa values to experimental macroscopic pKa values.** Microscopic pKa values for Epik were matched of the experimental dataset using algorithm 1. An alternative alignment of pKa values using the Hungarian algorithm (??) is available as a supplementary figure.



**Figure 7. Comparison of macroscopic Epik pKa values to experimental macroscopic pKa values.** Macroscopic pKa values for Jaguar were matched of the experimental dataset using algorithm 1. An alternative alignment of pKa values using the Hungarian algorithm (??) is available as a supplementary figure.

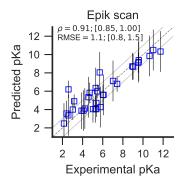


Figure 8. Epik macroscopic pKa values compared to macroscopic experimental pKa values, sequentially matching pKa values Comparison of the Epik sequential scan with entire experimental data set is shown, allowing only sequential pKa values to match experiment.

Add columns for the other two algorithms

**Table 2. Macroscopic pKas computed via Epik's sequential scan procedure.** Experimental and computed pKa values matched using algorithm 1.

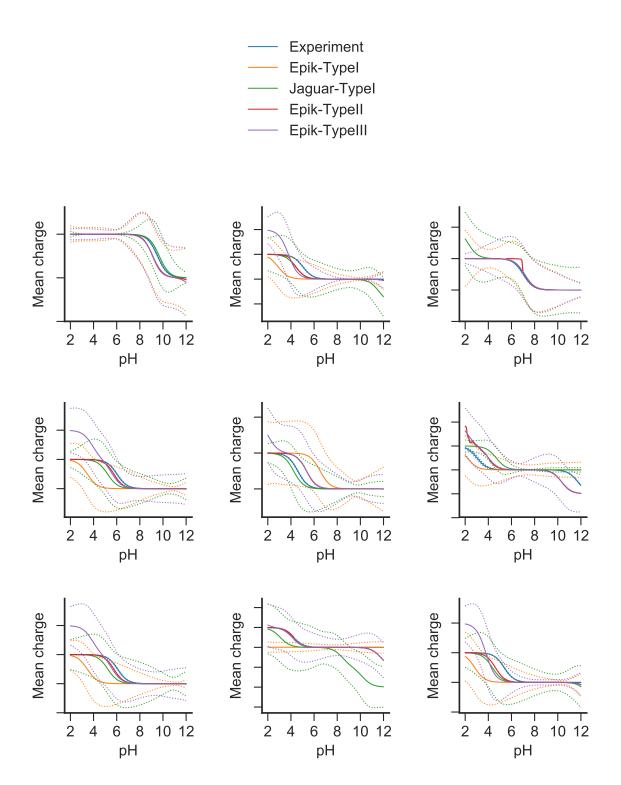
Molecule	Experiment $\pm$ SEM	Epik closest $\pm$ SEM
SM01	9.53±0.01	9 <u>±</u> 1
SM02	5.03±0.01	4±2
SM03	7.02±0.01	7±2
SM04	6.02±0.01	6 <u>±</u> 1
SM05	4.59±0.01	5 <u>+</u> 2
SM06	3.03±0.04	2±1
SM06	11.74 <u>±</u> 0.01	10 <u>±</u> 2
SM07	6.08 <u>±</u> 0.01	6 <u>±</u> 1
SM08	4.22 <u>±</u> 0.01	4 <u>±</u> 2
SM09	5.37±0.01	4 <u>±</u> 2
SM10	9.02 <u>±</u> 0.01	9 <u>±</u> 2
SM11	3.89 <u>±</u> 0.01	4 <u>±</u> 2
SM12	5.28 <u>±</u> 0.01	4 <u>±</u> 2
SM13	5.77 <u>±</u> 0.01	4 <u>+</u> 2
SM14	2.58 <u>+</u> 0.01	3 <u>±</u> 2
SM14	5.30±0.01	6 <u>±</u> 2
SM15	8.94 <u>±</u> 0.01	9 <u>±</u> 1
SM15	4.70±0.01	6 <u>±</u> 1
SM16	5.37±0.01	$4.7\pm0.9$
SM16	10.65±0.01	10 <u>±</u> 2
SM17	3.16±0.01	$4.9\pm0.7$
SM18	9.58±0.03	9 <u>+</u> 2
SM18	2.15±0.02	2.5 <u>±</u> 2
SM18	11.02±0.04	11 <u>±</u> 2
SM19	9.56±0.02	9 <u>+</u> 2
SM20	5.70±0.03	8 <u>±</u> 2
SM21	4.10±0.01	4 <u>±</u> 2
SM22	7.43 <u>±</u> 0.01	6.8 <u>±</u> 0.9
SM22	2.40±0.02	4 <u>±</u> 1
SM23	5.45 <u>±</u> 0.01	6.0±0.8
SM24	2.60 <u>+</u> 0.01	4 <u>±</u> 2

Add table for the Hungarian algorithm to SI.

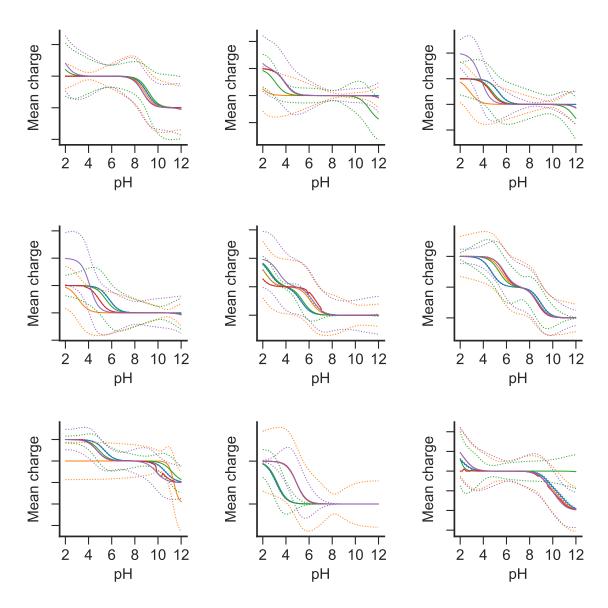
Table 3. Microscopic pKa per molecule for each method as matched by algorithm 1, compared to experiment.

	Experimental pKa	Epik pKa	Epik pKa uncertainty	Epik error	Jaguar pKa	Jaguar error
SM01 pKa1						
SM01 pKa2						
SM02 pKa 1						
SM03 pKa 1						

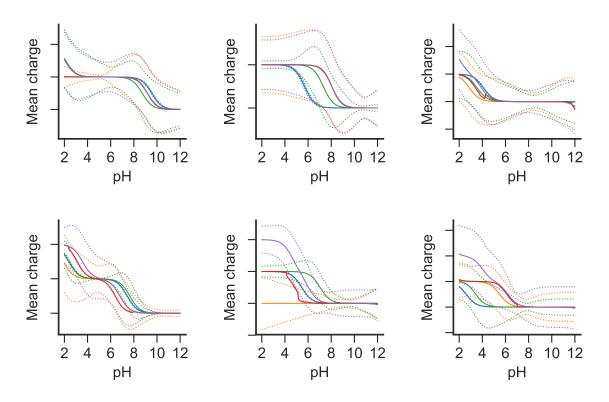
ASR: Find a good place to put the legend, and regenerate figures with lines less thick, or more alpha, add molecules to these. Plot experiment as dashed with each curve.



**Figure 9. Bootstrap titration curves for each method compared to experiment for molecules SM01-SM09.** The titration curve for each method is shown as a solid line, and 96 % confidence intervals from bootstrap have been shown as dotted lines. A numerical comparison to experiment is presented in table 4.



**Figure 10. Bootstrap titration curves for each method compared to experiment for molecules SM10-SM18.** The titration curve for each method is shown as a solid line, and 96 % confidence intervals from bootstrap have been shown as dotted lines. A numerical comparison to experiment is presented in table 4.



**Figure 11. Bootstrap titration curves for each method compared to experiment for molecules SM19-SM24.** The titration curve for each method is shown as a solid line, and 96 % confidence intervals from bootstrap have been shown as dotted lines. A numerical comparison to experiment is presented in table 4.

ASR:Need to add new data, and bootstrap error bars.

Table 4. Difference in area between derived macroscopic charge titration curves ( $\Delta$  area ) for each method. The macroscopic charge curve was constructed for each of the methods, and after alignment with the experimental curve by an integer offset, the area between the curves is reported.

Molecule	Epik microscopic pKa	Epik microscopic population	Epik scan	Jaguar microscopic pKa
SM01	0.43	0.44	0.46	0.15
SM02	2.12	0.59	1.99	1.21
SM03	0.10	0.29	0.09	0.57
SM04	2.55	0.26	2.20	0.81
SM05	1.93	0.74	1.08	0.51
SM06	2.60	2.52	2.62	2.09
SM07	2.61	0.32	2.25	0.84
SM08	1.27	0.57	0.63	5.82
SM09	2.53	0.78	2.32	1.09
SM10	0.17	0.34	0.62	0.28
SM11	1.84	0.06	0.14	1.68
SM12	2.48	0.71	2.26	0.88
SM13	2.64	0.88	2.80	0.30
SM14	1.07	1.74	1.75	0.25
SM15	1.07	1.32	1.31	0.85
SM16	4.36	1.10	1.44	0.97
SM17	1.71	1.71	1.71	0.07
SM18	0.74	1.02	1.00	NaN
SM19	0.38	0.39	0.73	1.22
SM20	2.34	2.34	2.34	1.60
SM21	1.26	0.70	0.55	1.68
SM22	0.94	1.62	1.64	0.23
SM23	0.38	0.68	3.17	NaN
SM24	0.61	3.46	5.62	NaN
Average	1.59	1.03	1.70	1.10



**Figure 12. Problematic lewis structure generated by Epik.** Epik generated protonation states with invalid Lewis structure for SM03 and SM18, with a pentavalent nitrogen present in the thiazole/thiadiazole (the protonation pattern displayed on the right). One of these was also present in the set provided as part of SAMPL6 microstates, namely SM03\_micro021. A valid Lewis structure could not be figured out for this protonation state.

See Images folder for titration curves and free energy images for each category. Will be added to supplementary info later.

**Abstract** The goal of the SAMPL6 pKa Challenge was to evaluate the performance of small molecule pKa prediction methods from challenge participants in a blinded fashion on a set of small, drug-like molecules resembling kinase inhibitor fragments. To provide a useful point of comparison for blind participant predictions that may use experimental methods still under development, we performed reference benchmark calculations using a popular empirical model (Epik) and quantum chemical approach (Jaguar) from Schrödinger. Epik predicts microstate populations and pKas using a Hammett-Taft type model, while Jaguar is a fast DFT quantum chemical method. In this work, we discuss how these reference calculations were performed, provide a broad assessment of the performance of the method, and highlight challenges and considerations in predicting pKas to benchmark against experiment macroscopic pKa measurements.

#### Introduction

Titratable sites are ubiquitous in druglike small molecules.

JDC: It would be very useful if we could cite some sort of reference discussing how important protonation states are in small molecule drugs or druglike molecules, but I'm failing to find useful references right now. Alternatively we could pull pKas from DrugBank <a href="https://www.drugbank.ca/releases/latest">https://www.drugbank.ca/releases/latest</a> or use Epik to make our own plots, like the kinase inhibitor pKa plot we had previously generated.

Large-scale computational surveys suggest that 60% of all protein-ligand complexes undergo a change in ionization state upon binding [1], either due to protonation state changes of the small molecule or the protein (where roughly a third of all protein residues are ionizable [2]). More generally, protonation state effects—in which the dominant protonation, charge, or tautomer state shifts upon binding, or a mixture of protonation states are significantly populated in complex or solution—has the potential to cause large modeling errors if these effects are neglected. In the SAMPL5 distribution coefficient (logD) challenge, for example, protonation state effects were determined to be a major contributor to loss in accuracy for the otherwise mundane task of predicting a transfer free energy between aqueous and cyclohexane phases [31].

To isolate the question of how well pKa effects could be modeled—and therefore how accurately the community could address these effects—the SAMPL6 challenge featured a blind pKa prediction component, as an intermediate step to logD predictions in which we provide participants with pKas and later have them predict both pKa and logD [?]. The SAMPL6 pKa challenge consisted of predicting macroscopic pKas measured by UV-metric titration for a set of small moelcules that resembled kinase inhibitors and their fragments [?]. As participants in the SAMPL6 pKa challenge were expected to utilize a wide variety of methods still under development, we endeavored to provide a useful baseline reference set of predictions using well-established, widely-deployed, commercially available methods. We selected both an empirical method (Epik [4]) and quantum chemical method (Jaguar [5]) from the Schrödinger Suite of computational chemistry software, version 20XX-X.

Reference calculations—which were not fully blinded—were performed in a manner that attempted to mimic standard use, using recommended settings for each program, without significantly modifying the input parameters. As the computation of UV-metric macroscopic pKas from the microscopic pKas predicted by the tools is not necessarily completely straightforward, we considered several alternative possibilities, which we discuss in more detail. We provide an analysis and broad assessment of the performance of the two methods, and highlight challenges and considerations in predicting pKas to benchmark against experiment. All analysis tools used to perform this study are available via GitHub at https://github.com/choderalab/SAMPL6-Reference-pKa-Calculations

#### Methods

## 74 Epik

75 Summary of how Epik works

Could be filled in by John Shelley?

# 77 Jaguar

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Summary of how Jaguar computes pKas

Could be filled in by Art Bochevarov?

#### 80 Microstate predictions

- 81 Epik: Fast empirical pKa predictions
  - were performed between pH 2-12
  - Reported states with a minimum population of  $e^{-10/RT}$ 
    - States proposed by SAMPL6 but not predicted by Epik were not considered in analysis.

# 85 Jaguar: Ab initio quantum chemical pKa predictions

- Ran using default settings (5 conformations) for each of the pre-enumerated specified microstate pairs.
- Input structures were first minimized using MMFF.
  - Additional minimization was performed for structures for which scf/geopt did not converge

# 89 Predicting macroscopic pKas

- 90 Epik and Jaguar predict microscopic pKas or microstate energies, which must be translated into macroscopic
- 91 pKas for comparison to experiment. Several reasonable choices are possible for translating these microscopic
- 92 properties into macroscopic pKas.

# 93 Sequential titration of dominant species

- Sequential titration avoids the need to compute the energies of all protonation states
- Epik can automatically perform a sequential scan (which can also be used in Jaguar with some automation)
  - For this experiment, we started from the predicted highest-occupancy microstate at pH 7 and went in both directions, returning pKas between 2-12

#### 99 Virtual electrochemical titration

 Average charge vs pH to match electrochemical titrations, look at the mean signed deviation (MSD) to see whether the charge curve behavior matches

### 102 Predicting UV-metric titration

- · Actual experiments were UV-metric, where only UV-active transitions are observable
- There is potentially a route to predicting UV-active transitions via identifying of microstates that would have different UV spectra, perhaps using SMARTS matches or simple QM absorption spectra calculations?

#### 107 Results

- How much conformation-dependence is there in Jaguar-derived pKas (or state energies)?
- Does sequential scan or mean molecular charge provide better agreement with experimental macroscopic pKas? Would it be worthwhile to develop predictive UV-metric models?
- Comparison of observed accuracies to previously reported/expected accuracies; expected accuracy on kinase inhibitors derived from this study
- How does Epik compare to Jaguar in terms of accuracy and computational cost?
- Discussion of outliers

#### 15 Discussion

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## 116 Overall performance

Some low probability structures produced by Epik included questionable protonation states of a heterycyclic moiety, present in SM03 and SM18 ( Figure 12).

Overall performance of Epik and Jaguar based on various metrics (figures/tables)

Matching of experimental and calculated pKas

The experiments do not provide any microscopic information on what atom, or microstate a pKa belongs to. Therefore, it is necessary to perform a matching of pKa values between experiment and predicted pKa values. There are several ways one could go about this, each strategy can prioritize a different aspect to match. We consider three algorithms:

125 Closest pKa matching

126 Hungarian pKa matching

This algorithm, also known as linear sum assignment. It finds the combination of pKa that minimizes the overall cost by picking rows and columns in a matrix, also considering the cost of not mapping certain pKa values in the case of different numbers of predictions and experimental values.

130 Sequential pKa matching

For macroscopic only, because it doesn't make sense to sequentially align microscopic pKas to a set of macroscopic pKas algorithm 2

# Macroscopic titration curves

· Compare macroscopic titration results

135 Population curves from pKa

Description of the population curve generation from pKa

$$g_i(pH) = \beta \left( n_i * pH - \sum_j pKa_j \right)$$
 (1)

$$\pi_i(pH) = \frac{e^{-g_i(pH)}}{\sum_i e^{-g_i(pH)}}$$
 (2)

Virtual electrochemical titration

• Description of the calculation of the mean charge curve

$$\langle q_{\text{total}}\rangle(\text{pH}) = \sum_{i} q_{i} \times \pi_{i}(\text{pH})$$
 (3)

#### 41 Microscopic pKa values

· Compare microscopic pKa prediction results

## 143 Descriptor analysis

- which moieties are harder to predict?
- Do certain descriptors correlate with variance/absolute errors?
- Do number of rotatable bonds affect epik/jaguar results (conformations missing from prediction)?

#### Comparison with other methods

This subsection depends on how much information we already have in the overview paper

- Compared to other Hammet-Taft methods, does Epik stand out? [based on early overview results?]
- Compared to other QM methods, does |aguar stand out? [based on early overview results?]

## 151 Suggestions for future challenges

- Future challenges could benefit from NMR experiments for microstates
- Potentiometric titrations could capture states that may have been left out
- Probabilistic models (i.e. bayesian hierarchical models) could be constructed for the analysis of experiments.

#### 156 Conclusions

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- Epik and Jaguar can be useful as baselines
- We can use various ways of judging each method.
- A conclusion about the accuracy
- Some take away points about difficulties about comparing microscopic to macroscopic
  - How a future challenge could make this easier

# Code and data availability

 Input files and analysis scripts are available at https://github.com/choderalab/SAMPL6-reference-pkacalculations

#### **Author Contributions**

(Follow the CRediT Taxonomy)

#### 167 Acknowledgments

ASR, MI, AR, PBG, and JDC acknowledge support from the Sloan Kettering Institute. JDC acknowledges support from NIH grant P30 CA008748.

#### 170 Disclosures

171 JDC is a member of the Scientific Advisory Board for Schrödinger, LLC.

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# Supplementary Information SM01

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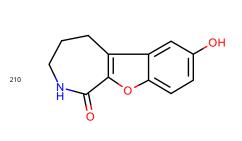
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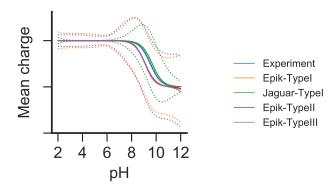
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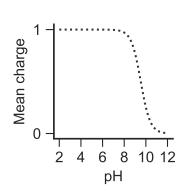
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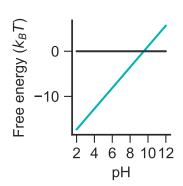
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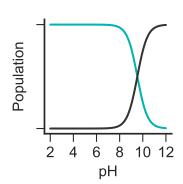


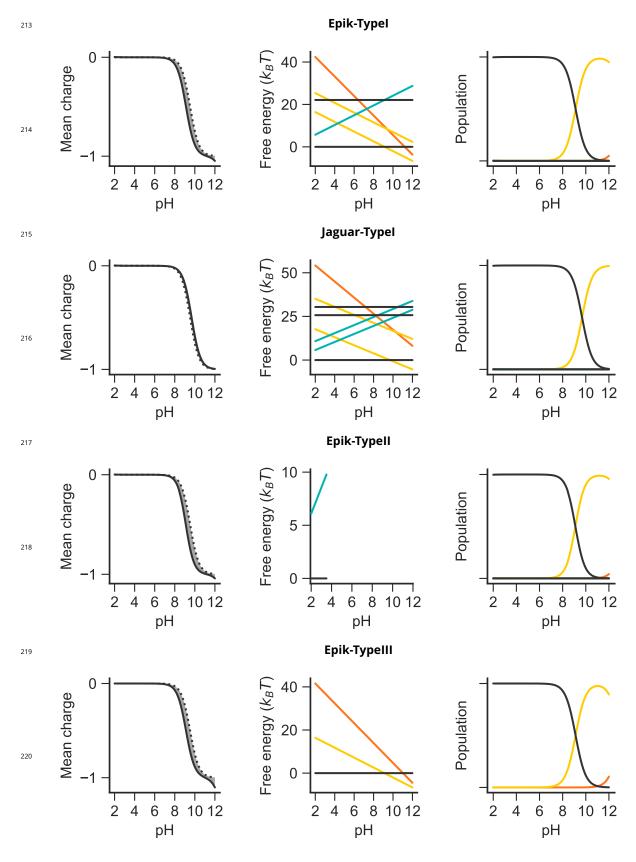






**Experiment** 





• Titration curves and free energy plots for each compound, by each method

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• The mean charge/deviation curves for each compound and each method

```
Result: Mapping of each experimental pKa i to predicted pKa j C is constructed, where every row i is an experiment, and every column j a prediction; C_{ij} = \text{cost}(\text{pKa}_{\text{exp},i}, \text{pKa}_{\text{pred},j}); while C.\text{size} > 0 do
\begin{array}{c} k, l = \text{arg min}(C_{ij});\\ \text{assign experimental value } k \text{ to prediction } l;\\ \text{remove row } k, \text{ column } l \text{ from } C;\\ \text{end} \end{array}
```

remaining values are unmatched;

**Algorithm 1:** This algorithm matches experiment with prediction based on how close each value is, one pKa value at a time. Unless the matrix *C* is square, some values will be unmatched. Those leftover pKas are returned at the end. It uses a cost function, such as root mean square deviation, to assess how close two values are.

```
Result: Mapping of each experimental pKa i to predicted pKa j
I = sorted experimental pKas;
J = sorted predicted pKas;
length = max(I.size, J.size);
Append placeholders such that I.size = length;
Append placeholders such that J.size = length;
min = \infty;
solution = J rolled 0 times;
for n in 0..length do
   S = J rolled n times;
   total = 0.0:
   for m in 0..length do
       if I_m is placeholder then unmatched experiment
          if J_{m} <= 7.0 then
           I_m = 0.0
          else
           I_m = 14.0
          end
       else if J_m is placeholder then unmatched prediction
          if I_m \leq 7.0 then
           J_m = 0.0
          else
           J_m = 14.0
          end
       end
       total = total + cost(I_m, J_m);
   if total < min then solution is better
       min = total;
       solution = S;
   end
end
```

**Algorithm 2:** Sequential pKa mapping. It uses a cost function to measure cost, and it rolls (shifts by one, and reintroduces last element as first). Any unmatched pKas are represented by matching with a placeholder value. To calculate the cost, the placeholder is replaced by either 0, or 14, depending on whether the unmatched value is above or below 7.0.

- .mae and .sdf files with results
- scripts and or jupyter notebooks for analysis
- csv version of tables