2.3 Modeling With first-Order equations

We have already done some modeling, mainly modeling population growth:

The exponential growth model dt = RP

The logistic po growde model $\frac{dP}{dt} = RP(1-\frac{P}{N})$.

In some exercises, we modeled ladioactive decay and another problem on ambient temperature.

Now we look at "mixing" ploblens. We do the modeling via an example.

Example 1 (Textbook page 52).

At time t=0, a tank contains Q. 16 (pound) of salt dissolved in 100 gal of water.

Assume that water containing 1 b of salt/gal is entering the tank at a rate of a gal/min, and that the mixed solution leaves the tank at the same rate.

Set up an IVP that describes the rate of change of the solt of this the tank.

Solution: Let Q(t) denote the amount of salt inside the trank at time t.

Instially there was Q, Ub of solt inside the tank.
ie Q(0)=Q0.

Now as selt water powers inside the tank, and as
the mixed solution leaves the tank, the amount of salt
is changing inside the tank.

We claim: $\frac{dQ}{dt} = (\text{rate at which salt is flowing in})$ - (rate at which salt is flowing out)

Now, rate in = $\left(\frac{1}{4}\frac{1b}{gal}\right)$. $\left(r\frac{gal}{min}\right) = \frac{r}{4}\frac{b}{min}$ $\frac{1}{4}$ Lbs of solt on flowing in per minute (rate in).

We will write a similar expression for the rate out.

The main difference is that the rate of the per gallon
is changing, it because Q(t) = amount of solt
inside tank is changing (is unknown) => amount
of Us per gallon is Q(t) (100 being the volume of tank)

$$= \frac{\partial Q}{\partial t} = \frac{1}{4} - \frac{100}{100} \text{ or } \frac{\partial Q}{\partial t} + \frac{1}{100} Q = \frac{1}{4}$$

this is a linear OOE where $p(t) = \frac{r}{100}$ and $g(t) = \frac{r}{4}$.

The pollutions take the form: $Q(t) = 25 + k \cdot e^{\frac{-r}{100}}t$.

but Q(0)=Q0 => Q0 = 25 + k => R=Q0-25.

Remark: Since the liquid pouring in has a concentration of I los of goldens salt per golden, then we expect after some time for the concentration inside the tank to reach that same amount: 0.25 lbs/golden Since the tank has low goldens =) eventually (i.e after some time), we expect the amount of Aaat to be: I × 100 = 25 lbs. Dow the solution Solvishy this property?

Take lin
$$Q(t) = \lim_{t \to \infty} 25 + (R_5 - 25) e^{-\frac{t}{100}t}$$

= 25, as expected.

Example 2 (Exercise 4 or page 60).

A tank with Capacity 500 gal contains originally 200 gal of water with 100 lb of soll in the solution.

Water containing 1 to of solt per galler is entiring at the valing of 3 gal/min

The mixture is flowing out at the rate of 2 gallous per minute.

because the liquid is flowing in at the 12th of

3 gallows per minute, and flowing out at

the late of 2 gallows per minute => Volume

of liquid inside the tout is not constant but

lather increasing by 1 gallow per minute.

3 the Volume of liquid inside tank after to

minute is: 200 + t.

= rate out =
$$\left(\frac{R(t)}{200+t}\right) \times (2) = \frac{20}{200+t}$$
 Ubs/gol Min

$$\frac{dQ}{dt} = 3 - \frac{2}{250+t}Q$$
; $Q(0) = 150$

or
$$\frac{dQ}{dt} + \frac{2}{200+t}Q = 3$$
; $Q(0) = 100$

$$\mu(t) = e$$

$$= \frac{2 + \lambda t}{200 + t}$$

$$= (200 + t)^{2}$$

$$=) Q(t) = \frac{1}{(2\omega_{\tau}t)^2} \int (2\omega_{\tau}t)^2 . 3 dt$$

$$= \frac{1}{(200+t)^{2}} \left[(200+t)^{3} + C \right] = (250+t) + \frac{C}{(250+t)^{2}}$$

but
$$Q(0) = 100 = 100 = 200 + \frac{C}{(200)^2} = C = (250)^{\frac{1}{2}}(100)$$
.

Final the amount of solt inside the tank when the solution begins to overflow.

Since the whene is increasing by I yallow per minute, then

after 300 minutes the tank well start overflowing.

$$Q(300) = (200 + 300) - \frac{4 \times 10^{6}}{(500)^{2}} = \frac{(500)^{3} - 4 \times 10^{6}}{(500)^{2}} = \frac{121 \times 10^{6}}{25 \times 10^{4}}$$
$$= \frac{121}{25} \times 10^{2}$$

Now, the concentration of salt = inside the tank at the time when it overflows = $\frac{Q(3 \text{ orb})}{Volume} = \frac{121}{25} \times 10^2$

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$$= \left| \frac{121}{125} \frac{\text{Lbs}}{\text{gal}} \right|$$

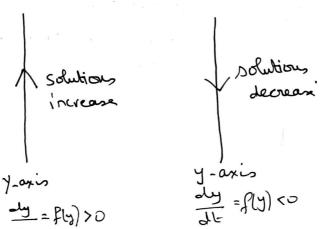
A 1st order ODE dy = P(y) to called automous.

We have seen earlier that solutions to autonomous ode's are horizontal translates of each other.

Furthermore, we define equilibrium solutions to be those solutions y, where f(y)=0. Those are horizontel solutions.

Since f(y) depends only on y, then the sign of dy depends also on y only: where fly) >0, dy >0 and y ? and you and you where fly)=0, dy = 0 and y is a constant (y = quilibrium solution).

This can be summarized on a vertical line /a y-axis, called the phase line:



y-axis = P(y) (0 constant solution

m =fy)=0

This can be better understood through examples. Ex1: dy = y-1 (+19) 20 if y=1 -> y=1 is equilibrium Phase line Ex2 dy = (y-1)(2-y) = +(y=0 if y=1 or y=2 1(y) co if y c1 or y>2. J: 2- SINK STABLE - y =1 - SOURCE | UNSTABLE | Phase line Ex3. = (y-1)2(2-y) = f(y) >0 if y <2 > 8(y) (> if y > 2 y=2 - (SINK) (STABLE) J=1 -> (NODE) | SEMI_STABLE) Phase line