

This is a Very Important Title!

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This abstract is abstract.

If you want to learn more about using L^AT_EX, you should check UiO's official tutorials: <https://www.mn.uio.no/ifi/tjenester/it/hjelp/latex/>

If you are familiar with L^AT_EX and you want to learn more about the REVTeX4-1 document class, check: http://www.physics.csbsju.edu/370/papers/Journal_Style_Manuals/auguide4-1.pdf

I. INTRODUCTION

II. THEORY

III. METHOD

IV. RESULTS

V. DISCUSSION

VI. CONCLUSION

ACKNOWLEDGMENTS

I would like to thank myself for writing this beautiful document.

REFERENCES

- Reference 1
- Reference 2

We will now show that

$$FWHM = 2\sqrt{2\ln 2}\sigma$$

The areas

$$(FWHM) \cdot (P(x_1)) = \frac{P(\mu)}{2} \cdot (x_2 - x_1)$$

And since

$$\begin{aligned} FWHM &= x_2 - x_1 \wedge FWHM = 2(\mu - x_1) \\ &\Downarrow \\ P(x_1) &= \frac{P(\mu)}{2} \\ e^{\frac{-(x_1-\mu)^2}{2\sigma^2}} &= \frac{1}{2} \\ -\frac{(x_1-\mu)^2}{2\sigma^2} &= -\ln 2 \end{aligned}$$

But we know that

$$-FWHM = 2(x_1 - \mu)$$

This then means

$$\begin{aligned} \frac{\left(\frac{1}{2}(-FWHM)^2\right)}{2\sigma^2} &= \ln 2 \\ FWHM &= 2\sqrt{2\ln 2}\sigma \end{aligned} \quad (1)$$

We are now going to derive the average energy of a molecule in an ideal gas. We will have use of knowing this integral.

$$\int_0^{\infty} x^{\frac{3}{2}-x} dx = \frac{3\sqrt{\pi}}{4}$$

Appendix A: Name of appendix

This will be the body of the appendix.

Appendix B: This is another appendix

Tada.

Note that this document is written in the two-column format. If you want to display a large equation, a large

figure, or whatever, in one-column format, you can do this like so:

This text and this equation are both in one-column format.

[1]

$$\frac{-\hbar^2}{2m}\nabla^2\Psi + V\Psi = i\hbar\frac{\partial}{\partial t}\Psi \quad (\text{B1})$$

Note that the equation numbering (this: B1) follows the appendix as this text is technically inside Appendix B. If you want a detailed listing of (almost) every available math command, check: <https://en.wikibooks.org/wiki/LaTeX/Mathematics>.

And now we're back to two-column format. It's really easy to switch between the two. It's recommended to keep the two-column format, because it is easier to read, it's not very cluttered, etc. Pro Tip: You should also get used to working with REVTeX because it is really helpful in FYS2150.

One last thing, this is a code listing:

```
This will be displayed with a cool programming font!
```

You can add extra arguments using optional parameters:

```
This will be displayed with a cool programming font!
```

You can also list code from a file using `lstinputlisting`. If you're interested, check https://en.wikibooks.org/wiki/LaTeX/Source_Code_Listings.

This is a basic table:

Table I. This is a nice table

Hey	Hey	Hey
Hello	Hello	Hello
Bye	Bye	Bye

You can a detailed description of tables here: <https://en.wikibooks.org/wiki/LaTeX/Tables>.

I'm not going to delve into Tikz in any level detail, but here's a quick picture:

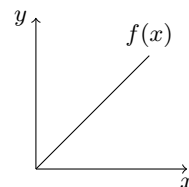


Figure 1. This is great caption

If you want to know more, check: <https://en.wikibooks.org/wiki/LaTeX/PGF/TikZ>.

[1] This equation is actually from quantum mechanics. “It’s called Schrödinger’s Time-Dependent Wave Equation”, named after the awesome Austrian physicist Erwin Rudolf Josef Alexander Schrödinger. Yep, the “Schrödinger’s cat” guy. Pretty cool dude actually, check his wiki page: https://en.wikipedia.org/wiki/Erwin_Schrodinger. He was

physics’ no. 1 Ladies’ man if there ever was one. Anyway, you will learn more about this equation in FYS2140. You can also find it printed on a glass wall in the UiO Physics Building (it really is that important).