

Flatulence Diffusion in a Confined Space

Virtual Earth: Simulating the Environment 01

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Motivation

- Imagine nine strangers packed into a $2\text{ m} \times 2\text{ m}$ elevator.
- A silent puff makes its debut—who's the first to notice the stench?
- Is there a scientific way to predict how quickly a puff spreads?



Research Goals

- In a 2D horizontal plane, develop a simulation of a flatulence "puff" in a confined space.
- Record when and where bystanders (detectors) first hit the smell threshold.

Physical Model Overview

- **Diffusion:** random molecular motion smooths concentration gradients.
- **Advection:** initial burst velocity + decaying "wind" transports the gas.
- **Initial Puff:** represented by a Gaussian concentration spot at the source.

Governing Equations

We solve the 2D unsteady advection-diffusion equation:

$$\frac{\partial C}{\partial t} + \mathbf{u} \cdot \nabla C = D \nabla^2 C$$

with zero-flux boundaries:


$$\left. \frac{\partial C}{\partial n} \right|_{\partial \Omega} = 0$$

Discretization & Numerical Scheme

- **Advection** ($\mathbf{u} \cdot \nabla C$): Explicit upwind scheme, CFL-limited.
- **Diffusion** ($D\nabla^2 C$): Crank-Nicolson via alternating-direction implicit (ADI) split:
 - Step 1: Implicit solve in x (fix y)
 - Step 2: Implicit solve in y (fix x)
 - **Advantage vs. Explicit Euler:** Unconditionally stable, allowing a much larger Δt for a faster simulation.

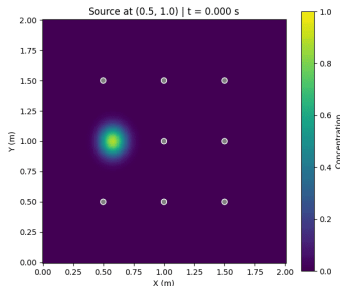
Model Parameters

- Domain: $2\text{ m} \times 2\text{ m}$, grid 200×200
- Time: total 10 s,
 $\Delta t = 0.02\text{ s}$
- Diffusion coeff.: $D = \underline{1.6 \times 10^{-5}\text{ m}^2/\text{s}}$ $0.01\text{ m}^2/\text{s}$
- Initial speed: 1.0 m/s ,
decays linearly over 2.0 s

Components of a human fart - 	
Nitrogen	20-90%
Hydrogen	0-50% (flammable)
Carbon Dioxide	10-30%
Oxygen	0-10%
Methane	0-10% (flammable)

Initial Conditions & Detector Setup

- **Source locations:** 9 preset "launchpads,"
- **Detectors:** 8 fixed points, start grey (safe), turn red when $C > C_{\text{thresh}}$.
- Detection threshold:
 $C_{\text{thresh}} = 0.03$



Simulation Results

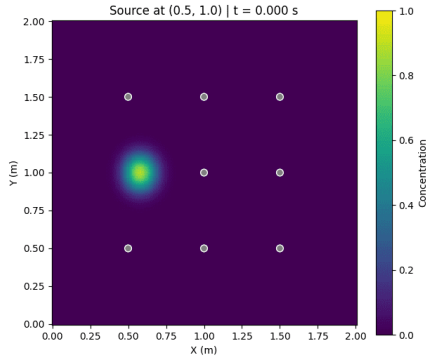


Figure 1: Source 4 (angle 0°)

Simulation Results

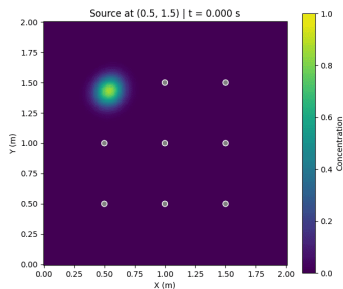


Figure 2: Source 1 (angle 300°)

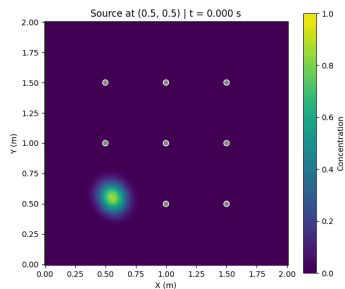


Figure 3: Source 7 (angle 42°)

Discussion, Conclusions & Future Work

- **Key insights:**
 - Burst speed & direction drastically alter who smells it first.
 - Crank-Nicolson keeps diffusion stable with larger Δt .
- **Limitations:**
 - 2D, no turbulence, constant D , linear u .
 - Real elevators have complex airflow patterns.
- **Next steps:**
 - Incorporate turbulence models.
 - Extend to full 3D to capture vertical mixing.