HW3

Mihel

# atoms 
$$275U = \frac{6.022 \times 10^{23}}{235} = 2.56 \times 10^{21} \Rightarrow 2.55 \times 10^{22}$$
# atoms  $237U = \frac{6.022 \times 10^{23}}{238} = 2.58 \times 10^{21} \Rightarrow 2.53 \times 10^{22}$ 
# atoms  $0 = \frac{6.022 \times 10^{23}}{11} = 7.50 \times 10^{22}$ 

Now must calculate enrichment factor

$$y = \frac{N_{235}}{N_{233}} = \frac{2.56}{2.53} = 1.012$$

$$Z = \{ 0, N : = (8.9 \times 10^{-27}, 2.56 \times 10^{23}), (8.9 \times 10^{-24}, 2.53 \times 10^{23}) + 2(3.75 \times 10^{-24}, 7.53 \times 10^{23}) \}$$

a) Isotropic source, assume emitted spherically with radius R

((1) = 34772dr = 47713 = P1

Osphragle Poszmuthelangle Invert COF: 1= RAS Sample PDF assume uniform neutron transport  $\rho(\theta, \phi) dd\phi = \frac{\sin \theta d\theta}{2} \cdot \frac{\partial \theta}{\partial \theta} = \frac{\sin \theta d\theta}{2} \cdot \frac{\partial \theta}{\partial \theta} = \frac{\cos (1-2)}{2\pi}$  Set  $u = \cos \theta = (1-2)\rho_2 \cdot \cos \theta = \frac{\cos (1-2)\rho_2}{2\pi}$ 

$$\begin{array}{ll}
\mathcal{L}(\phi) = \int_{0}^{10} z = \frac{1}{27} = p, & \phi = \sqrt{\pi}p, \\
x = r \sin\theta \cos\phi = Rp, & (1-\mu^{2})^{\frac{1}{2}} & \cos(2\pi p, ) & 2 = r \cos\theta = Rp, & \mu \\
y = r \sin\theta \sin\phi = Rp, & (1-\mu^{2})^{\frac{1}{2}} \sin(2\pi p, ) & = \pi \\
\end{array}$$

26 \(\frac{2+s}{5}\) = \(\

2c See Jupyter notebook

3a See text file

36 Dimensions: cylinder w/ radius=.41 cm, length:400cm
cylinder w/ radius=.42 cm, length=400cm
cylinder w/ radius=.48 cm, length=400cm
box w/ limensions of 1.26cm for Ww/A
Isotropic composition: 275 U, 238 U
Grichment: 5% 275 U, 950, 238 U

2

9

5

C

0

0

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Density
10.41 g/cm³ for Uranium Oxide
6.59 s/cm³ for Zirconium
0.7 g/cm² for water

Cross sections: UB = 0.73 cm from EWDF 7, fission v, prompt Hall = 0.71 cm GNDF 7 = total cross section thermal 2r = 0.58 cm GNDF 7 = Fission Q lib may

3 c) Kint = 1.3862, super-entical

Average neutron flux in fuel, cladding, and moderator

Fuel = 6.9354 × 10<sup>-2</sup> 1/cm<sup>2</sup>

Cladding = 6.9504 × 10<sup>-2</sup> 1/cm<sup>2</sup>

Moderator: 6.9488 × 10<sup>-2</sup> 1/cm<sup>2</sup>

Absorption: 2.74724 × 10 = 1/cm²

Absorption: 2.74724 × 10 = 1/cm²

Fission ate: \$ 3.80242×10 = 1/cm²

3c 1) See Jupytor Notebook 3c 5-6) See text file and Jupytor notebook

Proper notebook

Collegention. Worked 152anith John Floria. Worked 2b-3c with Major Freeman and Copt Chapman, who Walked us through a lot of Problem 4 is from Ashvin because I could not figure it out on moun.