

Homework 8
Due Wednesday 10/5 at the beginning of class

1. (5 pts) Dry air is bubbled through 25.0 liters of water at a rate of 15.0 liters (STP)/min. The air leaving the liquid is saturated with water at 25°C and 1.5 atm. How long will it take for all of the water to vaporize?
2. (5 pts) Ethyl acetate has a vapor pressure of 118.3 mm Hg at 29.5°C and a normal boiling point of 77.0°C. Estimate the vapor pressure at 45°C using:
 - a. The Antoine equation and the constants from Table B.4 of the textbook
 - b. The Clausius-Clapeyron equation and the two given data points
 - c. Linear interpolation between the two given points.
 - d. Assuming the answer from part a) to be the actual vapor pressure, calculate the percentage error associated with the estimates from parts b) and c).
3. (10 pts) A hot-air dryer is used to reduce the moisture content of 1500 kg/min of wet wood pulp from 0.75 w/w water to 0.002 w/w water. Air is drawn from the atmosphere at 28°C, 760 mm Hg, and 50% relative humidity ($\frac{P_{H_2O}}{P_{H_2O}^*}$). The air is heated and then passes through the dryer. The air leaves the dryer at 80°C and 10 mm Hg (gauge). A sample of the air is drawn into a chamber and slowly cooled while keeping the pressure constant. A mist is observed to form on the mirror in the chamber at 40.0°C.
 - a) What is the rate (kg/min) that water is removed from the wood?
 - b) What is the volumetric flow rate of air entering the system to achieve this?