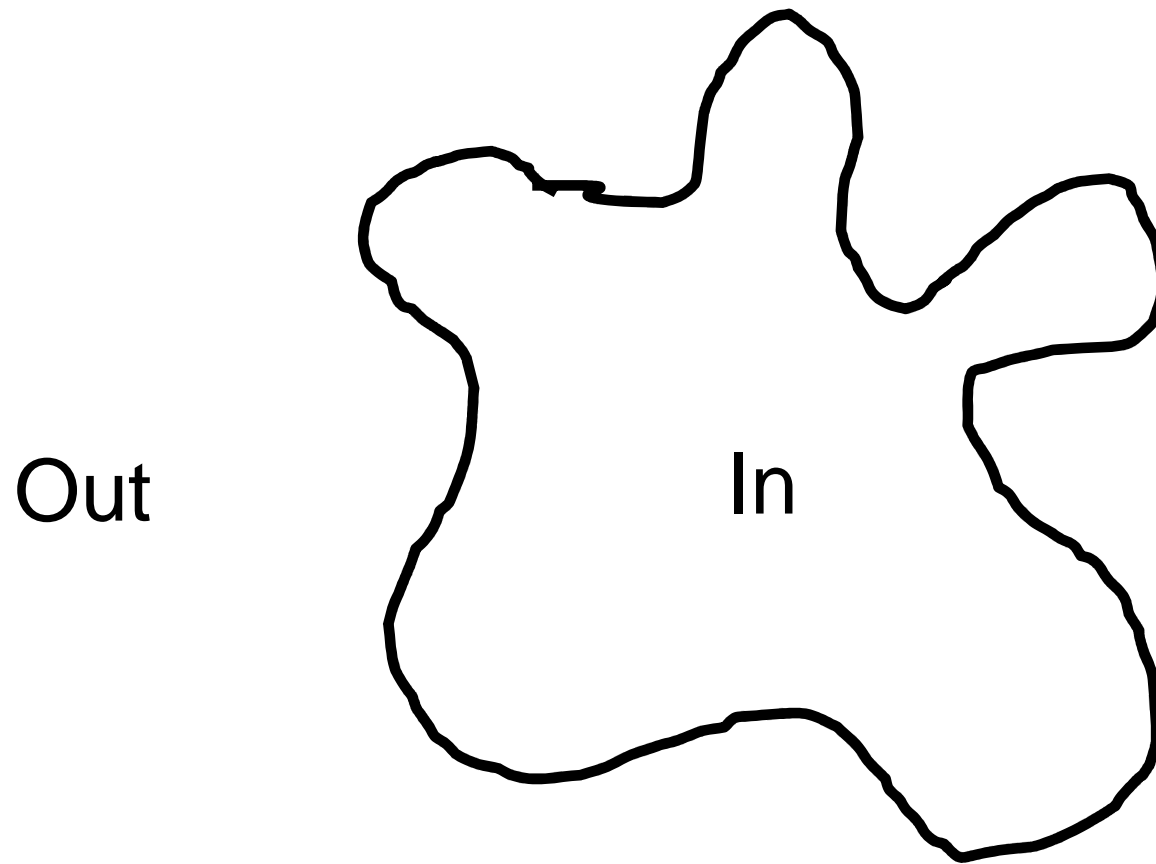


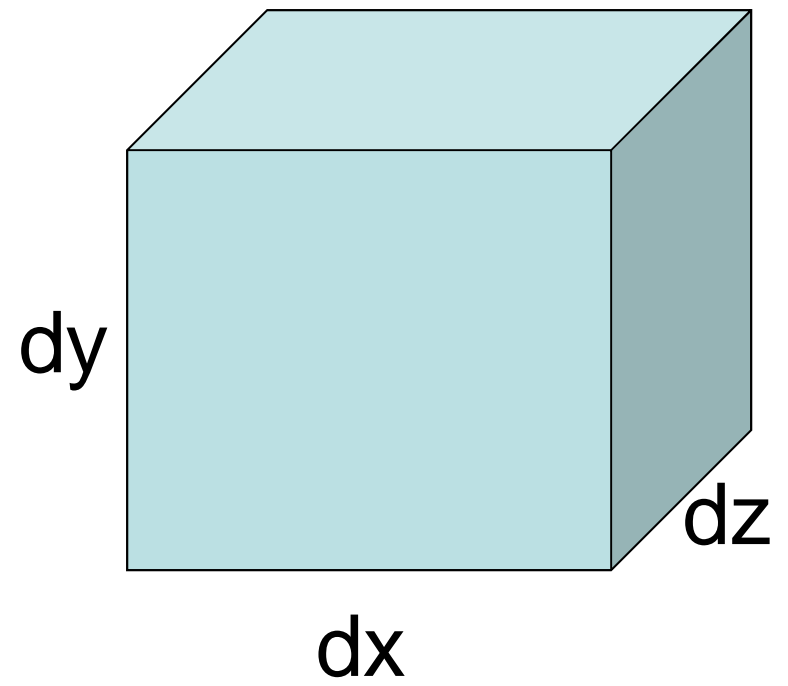
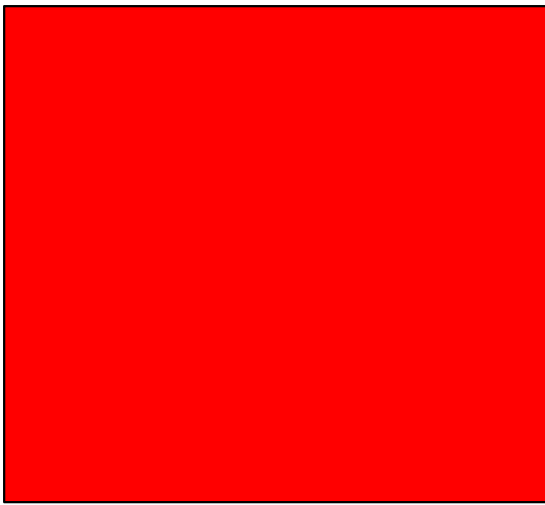
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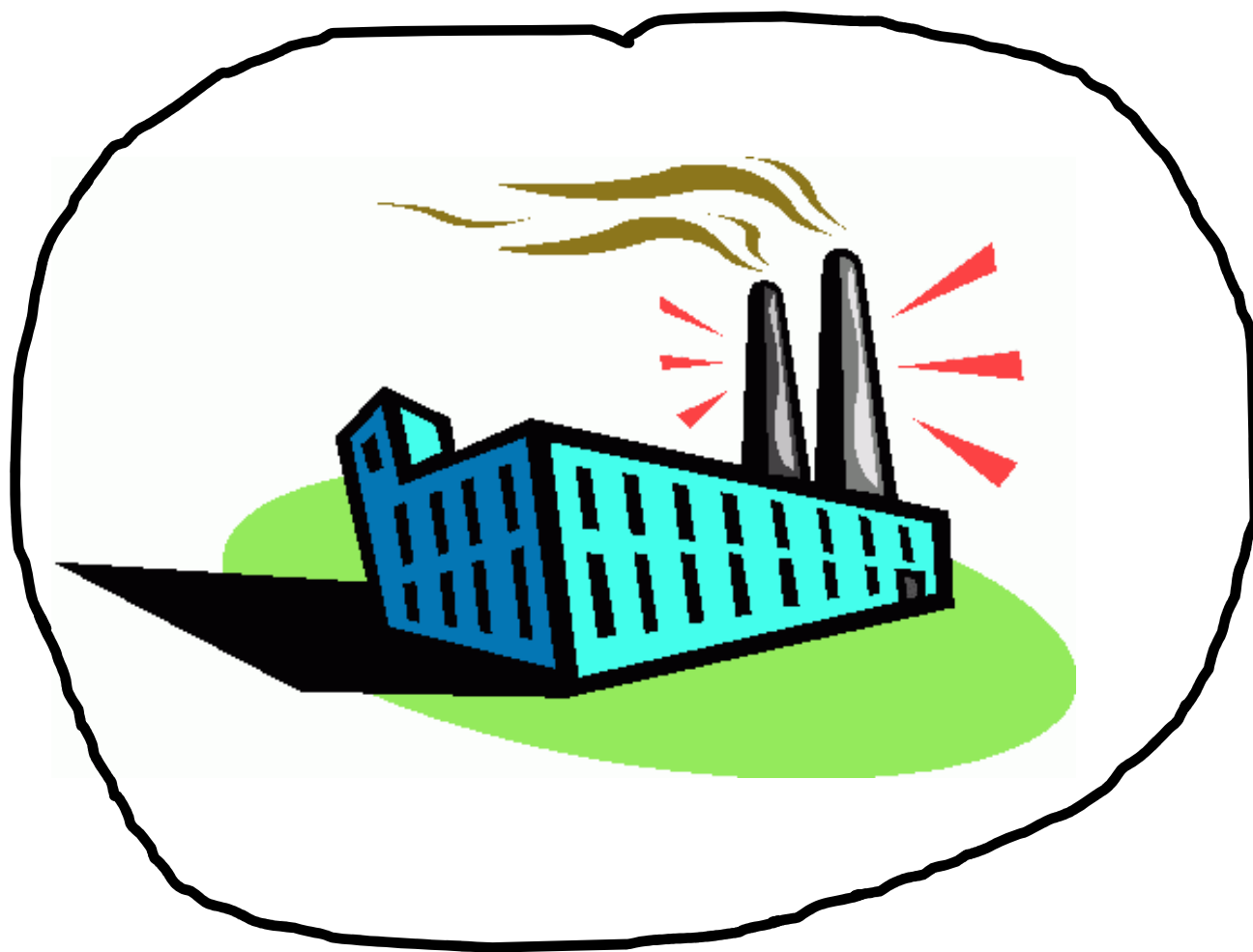
Biological Thermodynamics 1

Mass Balance Fundamentals

System Definition







Material Balances

- We can think of material balances in the same way as we think of balancing our checking accounts.

Generalized Continuity Equation

Accumulation = in – out + generation - consumption

$$\frac{\partial \varphi}{\partial t} + \nabla \cdot \vec{f} = \sigma \qquad \frac{dq}{dt} + \oint \vec{f} \cdot d\vec{S} = \Sigma$$

Definitions

- Input: •Mass and energy entering the system
- Output: •Mass and energy exiting the system
- Generation: •Mass and energy produced by the system
- Consumption: •Mass and energy being converted in the system
- Accumulation: •Mass and energy building up in the system

Generation and Consumption

- Chemical compounds can be generated through chemical reactions (products)
- Chemical compounds can be consumed through chemical reactions (reactants)
- Elements (C,N,P,S, etc) are neither generated nor consumed
- Total mass is neither generated nor consumed

Example: Population

- Chicago has 4 million people. Per yr. 250,000 move to the city, 200,000 leave the city, 50,000 are born and 100,000 die. What are the terms?
 - Input: 250,000 (move in)
 - Output: 200,000 (move out)
 - Generation: 50,000 (births)
 - Consumption: 100,000 (deaths)

Groundwater

- A well is measured at a level of 10" (each inch corresponds to 100 gallons). Today it rains and increases the level to 12". The well provides water for a farm household that used 200 gallons today. What are all of the terms? What is the final level?
 - Input: $(12'' - 10'')(100 \text{ gal/in}) = 200 \text{ gal}$
 - Output: 200 gal
 - Accumulation = Input – Output = 200 gal – 200 gal = 0 gal, the level remains 10"

Special Terms

- Steady State

$$\text{Accumulation} = 0$$

$$dX/dt = 0$$

- Transient

$$\text{Accumulation} \neq 0$$

$$dX/dt \neq 0$$

- Non-reactive

$$\text{Generation} = \text{Consumption} = 0$$

Types of Processes

- Batch: No mass crosses the system boundary
- Continuous: Mass inputs and outputs exist
- Semibatch: Processes not batch or continuous, usually a stream entering (input) or a stream leaving (output) but not both, contributing to the value of the accumulation term.

Solving Mass Balance Problems

- When solving mass balance problems the proper choice of a basis for the calculations can make life easier!
- Ask yourself:
 - What do I have to start with?
 - What answer is called for?
 - What is the most convenient **basis** to use?

What type of basis?

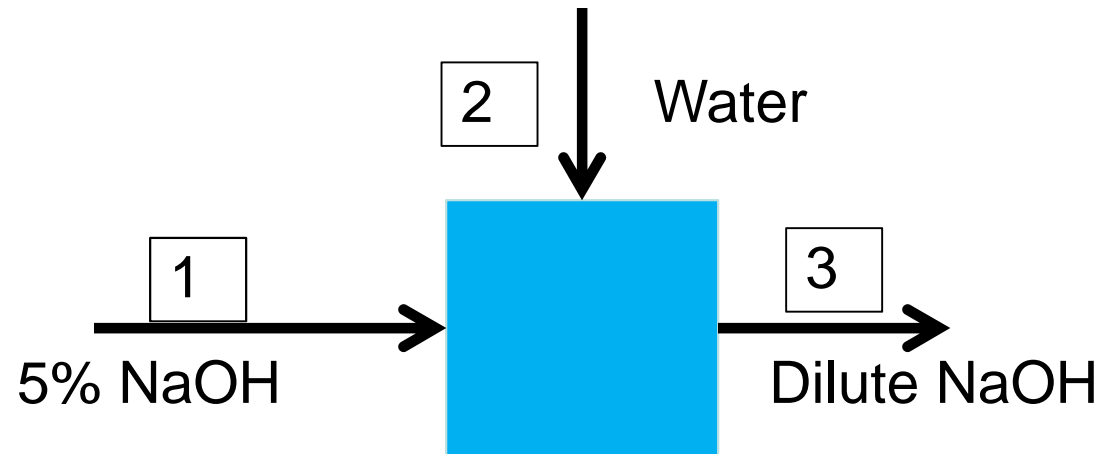
- If compositions are given as mass fractions then choose a total mass basis.
- If mole fractions are known, choose total number of moles.
- If some amount is already given:
 - a) use it as the basis
 - b) use 100 (percentage) as the basis and then scale the answer.

Mass Balance Problems: Develop a Procedure

- Write down all of the information with units in an organized form, with a picture if applicable.
- Write down what you want to find and the units.
- Pick a basis if necessary.
- Do the calculations, for multiple calculations use a table format
- Check to see if the result is dimensionally correct and does it make sense, is it in the right range.

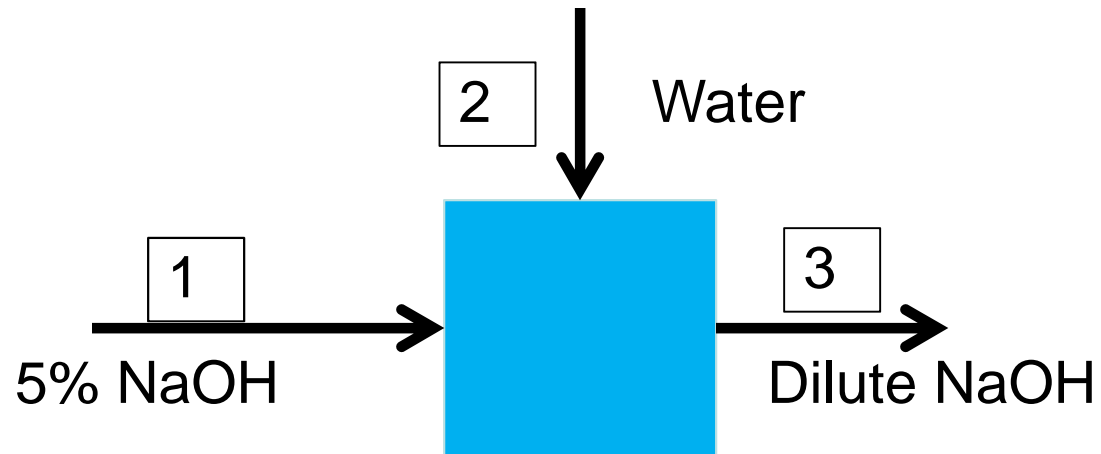
Sodium Hydroxide Example

- A 5.00% wt solution of sodium hydroxide is being diluted by a 10.0 gpm stream of pure water.
- What is the weight percent of sodium hydroxide in the stream exiting the system at 2.00×10^2 lbm/min?
- Assume steady-state conditions.



| Composition | 1 | 2 | 3 |
|-------------|------|---|---|
| NaOH | 0.05 | 0 | ? |
| Water | 0.95 | 1 | ? |
| | | | |

$$\Sigma(x\%) = 100\%$$



| Composition | 1 | 2 | 3 |
|-------------|------|---|---|
| NaOH | 0.05 | 0 | ? |
| Water | 0.95 | 1 | ? |
| | | | |

| Mass | 1 | 2 | 3 |
|-------|---|----------|---------|
| NaOH | ? | 0 | ? |
| Water | ? | 83.4 lbm | ? |
| Total | ? | 83.4 lbm | 200 lbm |

$$\rho_{\text{water}} = 8.34 \text{ lbm/gal}$$

| Mass | 1 | 2 | 3 |
|--------------|----------------|-----------------|----------------|
| NaOH | $X_{1n} * m_1$ | 0 | $X_{3n} * m_3$ |
| Water | $X_{1w} * m_1$ | 83.4 lbm | $X_{3w} * m_3$ |
| Total | m1 | 83.4 lbm | 200 lbm |

$$\text{Acc} = \text{In} - \text{Out} + \text{Gen} - \text{Con}$$

$$\text{Acc} = 0 \text{ (Steady State)}$$

Total Mass Balance

$$0 = (m_1 + m_2) - m_3 = m_1 + 83.4 - 200$$

$$m_1 = 116 \text{ lbm}$$

NaOH Mass Balance

$$0 = (x_{1n} * m_1 + x_{2n} * m_2) - x_{3n} * m_3$$

$$0 = (0.05 * 116 + 0 * 83.4) - x_{3n} * (200)$$

$$X_{3n} = 0.0290 = 2.90\%$$

Summary

- Systems are defined by their boundary
 - A boundary is a closed loop
- Accumulation = in – out
+ generation - consumption
- Solving Material Balance Problems
 - Organize your information
 - Draw a diagram
 - Assemble your equations
 - Solve!