

## Rubric for VLE calculations

- 1) For a given  $T, P$ , use EOS to calculate  $\underline{V}$  (use a computer)

in  
matlab - roots

for vapor

- 2) Choose the low density (biggest)  $\underline{V}$

bigger  $\underline{V}$  is, the more ideal gas-like it is.

max( )

- 3) Use the value of  $\underline{V}$  and sub into your equation for fugacity.

$$\ln \left( \frac{f^V(T, P)}{P} \right) = \frac{1}{RT} \int_{\underline{V}=\infty}^{\underline{V} = z^V RT/P} \left( \frac{RT}{\underline{V}} - P \right) d\underline{V} - \ln z^V + (z^V - 1)$$

for liquid

$$\ln\left(\frac{f^L(T,P)}{P}\right) = \frac{1}{RT} \int_{V=\infty}^{\underline{V} = Z^L RT/P} \left( \frac{RT}{\underline{V}} - P \right) d\underline{V} - \ln Z^L + (Z^L - 1)$$

Vapor, no problem  $\underline{V}$  goes from  $\infty$  to  
Some  $\underline{V}$  and no phase changes

For liquid,  $\rightarrow$  goes through vapor phase, condenses  
so there is a phase change

Same procedure for  $f^V$ , or  $f^L$  but how  
do we handle the phase change?

$$RT \ln\left(\frac{f^L}{P}\right) = \underline{G}(T,P) - \underline{G}^{IG}(T,P)$$

$$= \underbrace{\int_{P=0}^{P_{\text{vap}}} \left( \underline{V} - \frac{RT}{P} \right) dP}_{\text{(1) Vapor}} + \underbrace{RT \Delta \ln \left( \frac{f}{P} \right)}_{\text{(2) transition}} + \underbrace{\int_{P_{\text{vap}}}^P \left( \underline{V} - \frac{RT}{P} \right) dP}_{\text{(3) liquid.}}$$

② what is  $RT \Delta \ln \left( \frac{f}{P} \right)$ ?

fugacity between phases @ equilibrium.

Same in both!      So

$$\Delta \ln \left( \frac{f}{P} \right) = 0$$

$$\textcircled{1} \int_0^{P_{\text{vap}}} \left( \underline{V} - \frac{RT}{P} \right) dP = RT \ln \left( \frac{f}{P} \right)_{\text{sat}}$$

$$\textcircled{3} \int_{P_{\text{vap}}}^P \left( \underline{V} - \frac{RT}{P} \right) dP = \int_{P_{\text{vap}}}^P \underline{V} dP - RT \int_{P_{\text{vap}}}^P \frac{1}{P} dP$$

$$= \int_{P_{\text{vap}}}^P \underline{V} dP - RT \ln\left(\frac{P}{P_{\text{vap}}}\right)$$

So, now  
 (1) + ~~(2)~~ + (3)

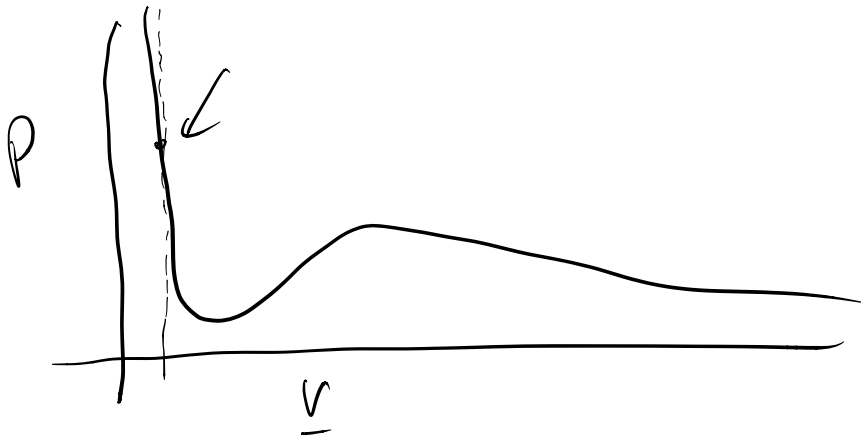
$$f^L(T, P) = \underbrace{P^{\text{vap}}(T) \left(\frac{f}{P}\right)_{\text{sat}}}_{f_{\text{sat}}} \underbrace{\exp\left[\frac{1}{RT} \int_{P_{\text{vap}}}^P \underline{V} dP\right]}_{\text{Poynting correction factor}}$$

Most of the  
 time.

$$f^L(T, P) \approx P_{\text{vap}}(T)$$

$$f^L(T, P) = f_{\text{sat}}^L(T) = f^V(T, P_{\text{vap}})$$

$$= P_{\text{vap}} \left( \frac{f}{P} \right)_{\text{sat.}}$$



if you assume liquid is incompressible,

then the fugacity is

$$f^L(T, P) = P_{\text{vap}}(T) \left( \frac{f}{P} \right)^{\text{sat}, T} \exp \left[ \frac{V (P - P_{\text{vap}})}{RT} \right]$$

# A1 Gore R<sub>y</sub>thm

