



Alginate: Properties and biomedical applications

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ABSTRACT

Alginate is a biomaterial that has found numerous applications in biomedical science and engineering due to its favorable properties, including biocompatibility and ease of gelation. Alginate hydrogels have been particularly attractive in wound healing, drug delivery, and tissue engineering applications to date, as these gels retain structural similarity to the extracellular matrices in tissues and can be manipulated to play several critical roles. This review will provide a comprehensive overview of general properties of alginate and its hydrogels, their biomedical applications, and suggest new perspectives for future studies with these polymers.

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1. Introduction

Biomaterials have traditionally been designed to be inert and not to interact with biological systems in the host. Materials derived from natural sources (e.g., wood) have a long history as biomaterials, and have often been used to replace tissues lost to disease or trauma (e.g., prosthetics). Since the early twentieth century, however, these materials began to be replaced by synthetic polymers, ceramics, and metal alloys, due to their better performance and more reproducible properties, as compared to naturally derived materials [1,2]. The more recent evolution in this field has now led to the definition of a biomaterial as a material intended to interface with biological systems to evaluate, treat, augment or replace any tissue, organ or function of the body [3], and boundaries for the use of biomaterials are still expanding. The design of new biomaterials is now focused on mimicking many functions of the extracellular matrices of body tissues, as these can regulate host responses in a well-defined manner, and naturally-derived materials have recently been regaining much attention owing to their inherent biocompatibility.

Alginate is a naturally occurring anionic polymer typically obtained from brown seaweed, and has been extensively investigated and used for many biomedical applications, due to its biocompatibility, low toxicity, relatively low cost, and mild gelation by addition of divalent cations such as Ca^{2+} [4]. Alginate hydrogels can be prepared by various cross-linking methods, and their structural similarity to extracellular matrices of living tissues allows wide applications in wound healing, delivery of bioactive agents such as small chemical drugs and proteins, and cell transplantation. Alginate wound dressings maintain a physiologically moist microenvironment, minimize bacterial infection at the wound site, and facilitate wound healing. Drug molecules, from small chemical drugs to macromolecular proteins, can be released from alginate gels in a controlled manner, depending on the cross-linker types and cross-linking methods. In addition, alginate gels can be orally administered or injected into the body in a minimally invasive manner, which allows extensive applications in the pharmaceutical arena. Alginate gels are also promising for cell transplantation in tissue engineering. Tissue engineering aims to provide man-made tissue and organ replacements to patients who suffer the loss or failure of an organ or tissue [5]. In this approach, hydrogels are used to deliver cells to the desired site, provide a space for new tissue formation, and control the structure and

function of the engineered tissue [6]. In this review, general properties of alginate and its current and potential applications in biomedical science and engineering will be discussed.

2. Alginate: general properties

Commercially available alginate is typically extracted from brown algae (*Phaeophyceae*), including *Laminaria hyperborea*, *Laminaria digitata*, *Laminaria japonica*, *Ascomphyllum nodosum*, and *Macrocystis pyrifera* [7] by treatment with aqueous alkali solutions, typically with NaOH [8]. The extract is filtered, and either sodium or calcium chloride is added to the filtrate in order to precipitate alginate. This alginate salt can be transformed into alginic acid by treatment with dilute HCl. After further purification and conversion, water-soluble sodium alginate powder is produced [9]. On a dry weight basis, the alginate contents are 22–30% for *A. nodosum* and 25–44% for *L. digitata* [10].

Bacterial biosynthesis may provide alginate with more defined chemical structures and physical properties than can be obtained from seaweed-derived alginate. Bacterial alginate can be produced from *Azotobacter* and *Pseudomonas*. The pathway of alginate biosynthesis is generally divided into (i) synthesis of precursor substrate, (ii) polymerization and cytoplasmic membrane transfer, (iii) periplasmic transfer and modification, and (iv) export through the outer membrane [11]. Recent progress in regulation of alginate biosynthesis in bacteria, and the relative ease of bacteria modification may enable production of alginate with tailor-made features and wide applications in biomedical applications.

2.1. Structure and characterization

Until Fischer and Dörfel identified the L-gulonate residue [12], D-mannuronate was regarded as the major component of alginate. Fractional precipitation with manganese and calcium salts demonstrated later that alginates are actually block copolymers, and that the ratio of guluronate to mannuronate varies depending on the natural source [13]. Alginate is now known to be a whole family of linear copolymers containing blocks of (1,4)-linked β -D-mannuronate (M) and α -L-gulonate (G) residues. The blocks are composed of consecutive G residues (GGGGG), consecutive M residues (MMMMMM), and alternating M and G residues (GMGMGM) (Fig. 1). Alginates extracted from different sources differ in M and G contents as well as

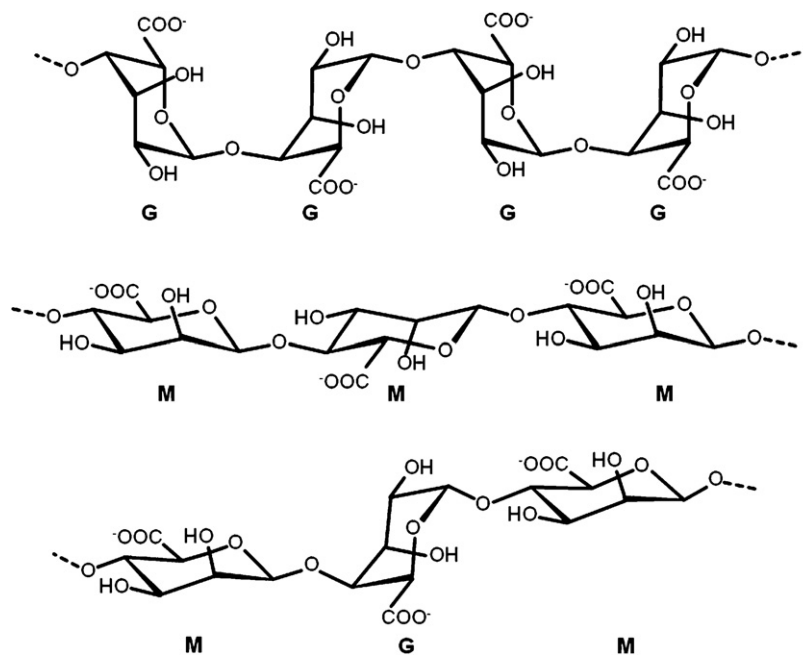


Fig. 1. Chemical structures of G-block, M-block, and alternating block in alginate.

the length of each block, and more than 200 different alginates are currently being manufactured [14]. The G-block content of *L. hyperborean* stems is 60%, and for other commercially available alginates is in the range of 14.0–31.0% [10].

Only the G-blocks of alginate are believed to participate in intermolecular cross-linking with divalent cations (e.g., Ca^{2+}) to form hydrogels. The composition (i.e., M/G ratio), sequence, G-block length, and molecular weight are thus critical factors affecting the physical properties of alginate and its resultant hydrogels [15]; the mechanism of gelation will be addressed in more detail in a later section. The mechanical properties of alginate gels typically are enhanced by increasing the length of G-block and molecular weight. It is important to note that different alginate sources provide polymers with a range of chemical structures (e.g., bacterial alginate produced from *Azotobacter* has a high concentration of G-blocks and its gels have a relatively high stiffness [16]). The physical properties significantly control the stability of the gels, the rate of drug release from gels, and the phenotype and function of cells encapsulated in alginate gels.

2.2. Molecular weight and solubility

The molecular weights of commercially available sodium alginates range between 32,000 and 400,000 g/mol. The parameters of the Mark–Houwink relationship ($[\eta] = KM_v^a$) for sodium alginate in 0.1 M NaCl solution at 25 °C are $K = 2 \times 10^{-3}$ and $a = 0.97$, where $[\eta]$ is intrinsic viscosity (mL/g) and M_v is the viscosity-average molecular weight (g/mol) [17]. The viscosity of alginate solutions increase as pH decreases, and reach a maximum around pH 3–3.5, as carboxylate groups in the alginate backbone become protonated and form hydrogen bonds.

Increasing the molecular weight of alginate can improve the physical properties of resultant gels. However, an alginate solution formed from high molecular weight polymer becomes greatly viscous, which is often undesirable in processing [18]. For example, proteins or cells mixed with an alginate solution of high viscosity risk damage from the high shear forces generated during mixing and injection into the body [19]. Manipulation of the molecular weight and its distribution can independently control the pre-gel solution viscosity and post-gelling stiffness. The elastic modulus of gels can be increased significantly, while the viscosity of the solution minimally raises, by using a combination of high and low molecular weight alginate polymers [20].

2.3. Biocompatibility

Although the biocompatibility of alginate has been extensively evaluated *in vitro* as well as *in vivo*, there is still debate regarding the impact of the alginate composition. Much of this confusion, though, likely relates to varying levels of purity in the alginate studied in various reports. For example, it has been reported that high M content alginates were immunogenic and approximately 10 times more potent in inducing cytokine production compared with high G alginates [21], but others found little or no immunoresponse around alginate implants [22]. The immunogenic response at the injection or implantation sites might be attributed to impurities remaining in the alginate. Since alginate is obtained from natural sources, various impurities such as heavy metals, endotoxins, proteins, and polyphenolic compounds could potentially be present. Importantly, alginate purified by a multi-step extraction procedure to a very high purity did not induce any significant foreign body reaction when implanted

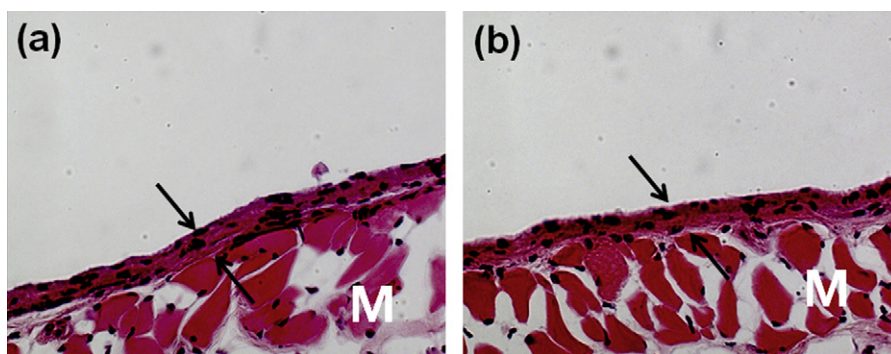


Fig. 2. Photomicrographs of tissue sections three weeks post-injection with (a) PBS and (b) alginate hydrogels. Original pictures were taken at 100 \times magnification. Photomicrographs have labels for the muscle layer (M). Arrows indicate the newly formed granulation tissue adjacent to the site of injection: this is a connective tissue typically formed following biomaterial implantation [24]. Copyright 2009, Springer, New York, USA.

into animals [23]. Similarly, no significant inflammatory response was observed when gels formed from commercially available, highly purified alginate have been subcutaneously injected into mice (Fig. 2) [24].

2.4. Derivatives

2.4.1. Amphiphilic alginate

Various alginate derivatives used in a range of biomedical applications have been reported to date. Amphiphilic alginate derivatives have been synthesized by introducing hydrophobic moieties (e.g., alkyl chains, hydrophobic polymers) to the alginate backbone. These derivatives can form self-assembled structures such as particles and gels in aqueous media, and have potential in many drug delivery applications.

Amphiphilic derivatives of sodium alginate have been prepared by conjugation of long alkyl chains (i.e., dodecyl, octadecyl) to the alginate backbone via ester bond formation. Aqueous solutions of these alginate derivatives exhibited the typical rheological properties of physically cross-linked, gel-like networks in the semidilute regime, which could be useful for cartilage repair and regeneration [25]. Microparticles were prepared from these derivatives by dispersion in a sodium chloride solution, as this technique can allow encapsulation of proteins and their subsequent release by the addition of either surfactants that disrupt intermolecular hydrophobic junctions or esterases that hydrolyze the ester bond between alkyl chains and the alginate backbone [26]. Dodecylamine was also conjugated to the alginate backbone via amide linkage formation using 2-chloro-1-methylpyridinium iodide as a coupling reagent. Hydrogels prepared from this alginate derivative exhibited long-term stability in aqueous media, compared to those prepared from alginate derivatives with dodecyl ester, which are labile to hydrolysis [27]. Water soluble, amphiphilic alginate derivatives grafted with cholesteryl groups were also synthesized using *N,N'*-dicyclohexylcarbodiimide as a coupling agent and 4-(*N,N'*-dimethylamino)pyridine as a catalyst at room temperature. The derivatives formed self-aggregates with a mean diameter of 136 nm in an aqueous sodium chloride solution [28]. The associative behavior of alginate grafted

with poly(ϵ -caprolactone) (PCL) in an aqueous sodium chloride solution was strongly dependent on the length of PCL chains, unlike the behavior of derivatives in water [29]. Sodium alginate has also been hydrophobically modified with poly(butyl methacrylate), leading to prolonged release of model drugs as compared with unmodified alginate gels [30].

2.4.2. Cell-interactive alginate

Alginate derivatives containing cell-adhesive peptides have been gaining significant attraction recently. These derivatives are typically prepared by chemically introducing peptides as side-chains, using carbodiimide chemistry to couple via the carboxylic groups of the sugar residues. Since alginate inherently lacks mammalian cell-adhesivity, appropriate ligands are crucial to promote and regulate cellular interactions, especially for cell culture and tissue engineering applications. Peptides including the sequence arginine–glycine–aspartic acid (RGD) have been extensively used as model adhesion ligands, due to the wide-spread presence of integrin receptors (e.g., $\alpha_v\beta_3$, $\alpha_5\beta_1$) for this ligand on various cell types [31,32]. RGD containing peptides can be chemically coupled to the alginate backbone using water-soluble carbodiimide chemistry (Fig. 3) [33]. A minimum concentration of RGD peptides in alginate gels is needed for the adhesion and growth of cells, and this minimum is likely cell type specific. For example, minimal concentrations for substantial adhesion of MC3T3-E1 and C2C12 cells to alginate gels *in vitro* have been reported as 12.5 $\mu\text{g}/\text{mg}$ [34,100] and 10.0 $\mu\text{g}/\text{mg}$ [98] alginate, respectively. The affinity of the RGD peptide will also play an important role, and cyclic RGD peptides have been demonstrated to be more potent and are needed at lower concentrations than linear RGD peptides [105]. Various peptides containing the DGEA (Asp–Gly–Glu–Ala) [34] and YIGSR (Tyr–Ile–Gly–Ser–Arg) [35] sequences derived from other extracellular matrix proteins have also been exploited to modify alginate gels and enhance the adhesive interactions with various cell types. For example, alginate has been modified with YIGSR peptides using water-soluble carbodiimide chemistry to promote neural cell adhesion [35]. However, the majority of work to date has been with RGD–alginate. Cell-interactive alginates have

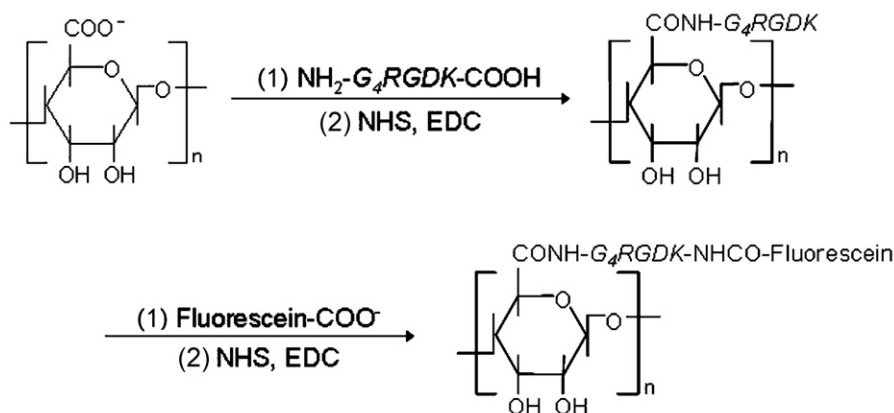


Fig. 3. Synthetic scheme of RGD-modified alginate using water-soluble carbodiimide chemistry (NHS, *N*-hydroxysulfosuccinimide; EDC, 1-ethyl-3-(dimethylaminopropyl) carbodiimide). For further analysis, RGD peptide can be labeled with fluorescein [33], as demonstrated in this scheme. Copyright 2008, Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim, Germany.

been widely utilized as cell culture substrates both in 2-D and 3-D culture, and as scaffolds in tissue engineering applications, as described in detail in a later section.

3. Hydrogel formation and its properties

3.1. Methods of gelling

Alginate is typically used in the form of a hydrogel in biomedicine, including wound healing, drug delivery and tissue engineering applications. Hydrogels are three-dimensionally cross-linked networks composed of hydrophilic polymers with high water content. Hydrogels are often biocompatible, as they are structurally similar to the macromolecular-based components in the body, and can often be delivered into the body via minimally invasive administration [36]. Chemical and/or physical cross-linking of hydrophilic polymers are typical approaches to form hydrogels, and their physicochemical properties are highly dependent on the cross-linking type and cross-linking density, in addition to the molecular weight and chemical composition of the polymers [37,38]. Here, we summarize various approaches to cross-link alginate chains to prepare gels, and how these approaches alter the hydrogel properties relevant to biomedical applications.

3.1.1. Ionic cross-linking

The most common method to prepare hydrogels from an aqueous alginate solution is to combine the solution with ionic cross-linking agents, such as divalent cations (i.e., Ca^{2+}). The divalent cations are believed to bind solely to guluronate blocks of the alginate chains, as the structure of the guluronate blocks allows a high degree of coordination of the divalent ions. The guluronate blocks of one polymer then form junctions with the guluronate blocks of adjacent polymer chains in what is termed the egg-box model of cross-linking, resulting in a gel structure (Fig. 4) [39]. Calcium chloride (CaCl_2) is one of the most frequently used agents to ionically cross-link alginate. However, it

typically leads to rapid and poorly controlled gelation due to its high solubility in aqueous solutions. One approach to slow and control gelation is to utilize a buffer containing phosphate (e.g., sodium hexametaphosphate), as phosphate groups in the buffer compete with carboxylate groups of alginate in the reaction with calcium ions, and retard gelation. Calcium sulfate (CaSO_4) and calcium carbonate (CaCO_3), due to their lower solubilities, can also slow the gelation rate and widen the working time for alginate gels. For example, an alginate solution can be mixed with CaCO_3 , which is not soluble in water at neutral pH. Glucono- δ -lactone is then added to the alginate/ CaCO_3 mixture in order to dissociate Ca^{2+} from the CaCO_3 by lowering the pH. The released Ca^{2+} subsequently initiates the gelation of the alginate solution in a more gradual manner [40].

The gelation rate is a critical factor in controlling gel uniformity and strength when using divalent cations, and slower gelation produces more uniform structures and greater mechanical integrity [41]. The gelation temperature also influences gelation rate, and the resultant mechanical properties of the gels. At lower temperatures, the reactivity of ionic cross-linkers (e.g., Ca^{2+}) is reduced, and cross-linking becomes slower. The resulting cross-linked network structure has greater order, leading to enhanced mechanical properties [42]. In addition, the mechanical properties of ionically cross-linked alginate gels can vary significantly depending on the chemical structure of alginate. For example, gels prepared from alginate with a high content of G residues exhibit higher stiffness than those with a low amount of G residues [43].

One critical drawback of ionically cross-linked alginate gels is the limited long-term stability in physiological conditions, because these gels can be dissolved due to release of divalent ions into the surrounding media due to exchange reactions with monovalent cations. In addition, the calcium ions released from the gel may promote hemostasis, while the gel serves as a matrix for aggregation of platelets and erythrocytes [44]. These features may be beneficial or negative, depending on the situation, but a desire to avoid these

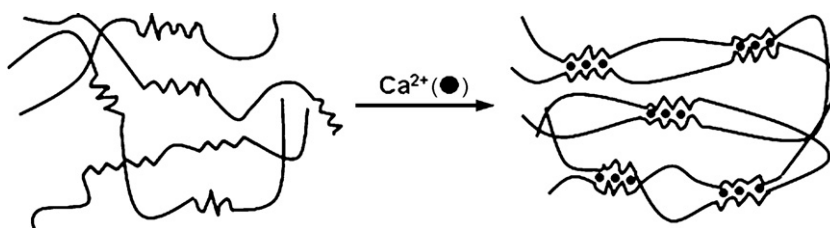


Fig. 4. Alginate hydrogels prepared by ionic cross-linking (egg-box model) [38]. Only guluronate blocks participate in the formation of a corrugated egg-box-like structure with interstices in which calcium ions are placed. Copyright 2007, Elsevier Science Ltd., Oxford, UK.

biological reactions, along with more general limitations of ionically cross-linked gels has led to interest in covalently cross-linked alginate hydrogels.

3.1.2. Covalent cross-linking

Covalent cross-linking has been widely investigated in an effort to improve the physical properties of gels for many applications, including tissue engineering. The stress applied to an ionically cross-linked alginate gel relaxes as the cross-links dissociate and reform elsewhere, and water is lost from the gel, leading to plastic deformation. While water migration also occurs in covalently cross-linked gels, leading to stress relaxation, the inability to dissociate and reform bonds leads to significant elastic deformation [45]. However, covalent cross-linking reagents may be toxic, and the unreacted chemicals may need to be removed thoroughly from gels.

Covalent cross-linking of alginate with poly(ethylene glycol)-diamines of various molecular weights was first investigated in order to prepare gels with a wide range of the mechanical properties. While the elastic modulus initially increased gradually with an increase in the cross-linking density or weight fraction of poly(ethylene glycol) (PEG) in the gel, it subsequently decreased when the molecular weight between cross-links (M_c) became less than the molecular weight of the softer PEG [46]. It was subsequently demonstrated that the mechanical properties and swelling of alginate hydrogels can be tightly regulated by using different kinds of cross-linking molecules, and by controlling the cross-linking densities. The chemistry of the cross-linking molecules also significantly influences hydrogel swelling, as would be expected. The introduction of hydrophilic cross-linking molecules as a second macromolecule (e.g. PEG) can compensate for the loss of hydrophilic character of the hydrogel resulting from the cross-linking reaction [47].

The use of multi-functional cross-linking molecules to form hydrogels provides a wider range and tighter control over degradation rates and mechanical stiffness than bi-functional cross-linking molecules. For example, the physical properties and degradation behavior of poly(aldehyde guluronate) (PAG) gels prepared with either poly(acrylamide-co-hydrazide) (PAH) as a multi-functional cross-linker or adipic acid dihydrazide (AAD) as a bi-functional cross-linker were monitored *in vitro*. PAG/PAH gels showed higher mechanical stiffness before degradation and degraded more slowly than PAG/AAD gels. The enhanced mechanical stiffness and prolonged degradation

behavior could be attributed to the multiple attachment points of PAH in the gel even at the same concentration of overall functional groups [48].

Photo cross-linking is an exciting approach to *in situ* gelation that exploits covalent cross-linking. Photo cross-linking can be carried out in mild reaction conditions, even in direct contact with drugs and cells, with the appropriate chemical initiators. Alginate, modified with methacrylate and cross-linked by exposure to a laser (argon ion laser, 514 nm) for 30 s in the presence of eosin and triethanol amine, forms clear and flexible hydrogels. The gels were useful for sealing corneal perforation *in vivo*, indicating a potential clinical use in sutureless surgery [49]. Photo cross-linking reactions typically involve the use of a light sensitizer or the release of acid, which may be harmful to the body. An alternative photo cross-linking approach uses polyallylamine partially modified with α -phenoxycinnamyl diene acetylchloride, which dimerizes on light exposure at about 330 nm, and releases no toxic byproducts during the cross-linking reaction [50]. The mechanical properties of the gels formed from this photosensitive polyallylamine and alginate were significantly improved by light irradiation, and the gels were freely permeable to cytochrome c and myoglobin [51].

3.1.3. Thermal gelation

Thermo-sensitive hydrogels have been widely investigated to date in many drug delivery applications, due to their adjustable swelling properties in response to temperature changes, leading to on-demand modulation of drug release from the gels [52]. Poly(*N*-isopropylacrylamide) (PNIPAAm) hydrogels are the most extensively exploited thermo-sensitive gels, and these undergo a reversible phase transition near body temperature in aqueous media (lower critical solution temperature near 32 °C). The transition temperature can be altered by copolymerization with hydrophilic monomers such as acrylic acid and acrylamide [53]. Despite the potential importance of thermo-sensitive hydrogels in biomedical applications, few systems using alginate have been reported yet, as alginate is not inherently thermo-sensitive. However, semi-interpenetrating polymer network (semi-IPN) structures were prepared via *in situ* copolymerization of *N*-isopropylacrylamide (NIPAAm) with poly(ethylene glycol)-*co*-poly(ϵ -caprolactone) (PEG-*co*-PCL) macromer in the presence of sodium alginate by UV irradiation (Fig. 5). The swelling ratio of the gels increased with the concentration of sodium alginate at a constant temperature, and

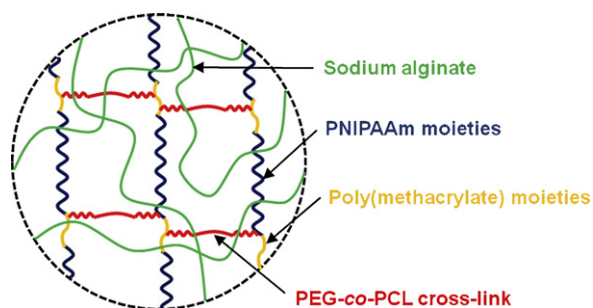


Fig. 5. Schematic description of thermo-sensitive semi-IPN alginate hydrogels [54].

decreased with an increase in temperature. The use of sodium alginate in semi-IPN structures also improved the mechanical strength and the cumulative release of BSA from the gels, indicating potential in drug delivery applications [54]. Graft copolymerization of NIPAAm onto the alginate backbone after reaction with ceric ions also provided a useful means to prepare temperature-responsive alginate gels, with sensitivity near body temperature (Lee and Mooney, unpublished data) (Fig. 6).

3.1.4. Cell cross-linking

While a number of chemical and physical methods have been reported to form alginate gels, the ability of cells to contribute to gel formation has been largely ignored. When alginate is modified with cell adhesion ligands, the ability of cells to bind multiple polymer chains can lead to long-distance, reversible network formation even in the absence of chemical cross-linking agents. Cells added to an RGD-modified alginate solution form a uniform dispersion within the solution, and this system subsequently generates the cross-linked network structure via specific receptor-ligand interactions without using any additional cross-linking molecules (Fig. 7) [55]. In contrast, cells added to non-modified alginate solutions aggregate and form a non-uniform structure, due to the dominance of cell-cell interactions in that system. This gelation behavior is shear reversible and can be repeated multiple times. Once the gel structure is broken down by applying shear forces, cross-linked structures are recovered within a few minute. This

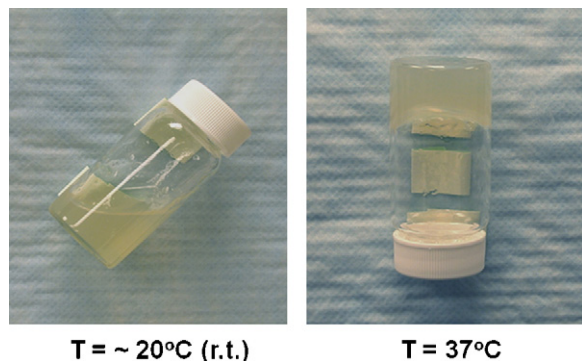


Fig. 6. Thermal gelation of an aqueous alginate-g-NIPAAm solution at 37 °C.

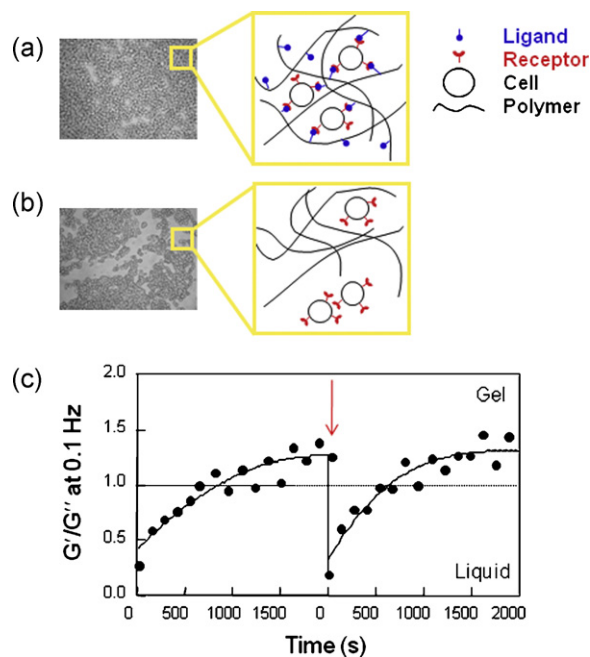


Fig. 7. Photomicrographs and schematics of cell-cross-linked network structures when cells are mixed with (a) RGD-modified alginate and (b) non-modified alginate. (c) Shear-reversible gelation of cell-cross-linked RGD-alginate hydrogels. The arrow indicates disruption of the gel structure with intentional shear force. The cross-over point where G' (storage shear modulus) = G'' (loss shear modulus) provides a measure of the gelation threshold (dotted line) [55]. Copyright 2003, Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim, Germany.

behavior is governed by the weak and reversible ligand-receptor interactions in the system. This system might be ideal for cell delivery in tissue engineering because a gel can flow like a liquid during injection into the body, but solidify once it is placed in the body. Further, it was reported that cells can provide additional mechanical integrity to RGD-alginate gels that are ionically cross-linked with calcium ions, again via binding interactions between cells and the adhesion ligands coupled to the alginate chains [56].

3.2. Biodegradation of alginate and its hydrogels

Alginate is inherently non-degradable in mammals, as they lack the enzyme (i.e., alginase) which can cleave the polymer chains, but ionically cross-linked alginate gels can be dissolved by release of the divalent ions cross-linking the gel into the surrounding media due to exchange reactions with monovalent cations such as sodium ions. Even if the gel dissolves, though, the average molecular weights of many commercially available alginates are higher than the renal clearance threshold of the kidneys, and likely will not be completely removed from the body [57]. An attractive approach to make alginate degradable in physiological conditions includes partial oxidation of alginate chains. Slightly oxidized alginate can degrade in aqueous media, and these materials have demonstrated potential as a delivery vehicle of drugs and cells for various applications. Alginate is typically oxidized with sodium

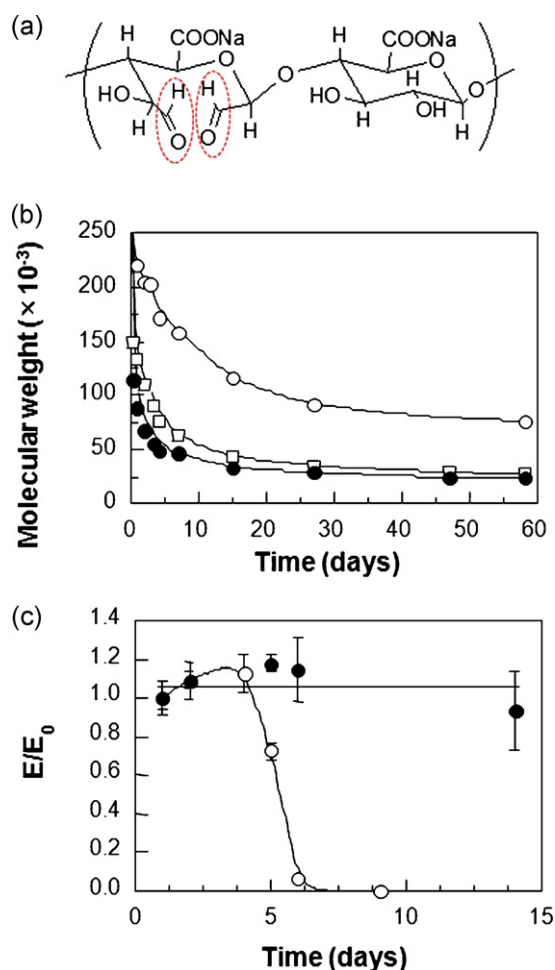


Fig. 8. (a) Chemical structure of partially oxidized alginate (aldehyde groups generated after oxidation reaction highlighted in red) and (b) its degradation behavior at various pHs (○, pH 4.5; □, pH 7.4; ●, pH 9.2). (c) Changes in the compressive modulus of hydrogels prepared from alginate (●) and partially oxidized alginate (○) over time (pH 7.2, 37 °C) [58]. Copyright 2001, John Wiley & Sons, New York, USA.

periodate (Fig. 8a). The periodate oxidation cleaves the carbon-carbon bond of the *cis*-diol group in the uronate residue and alters the chair conformation to an open-chain adduct, which enables degradation of the alginate backbone. A slight reduction in the molecular weight during oxidation is expected. However, partial oxidation of alginate does not significantly interfere with its gel forming capability in the presence of divalent cations. The resulting degradation rate of the gels is strongly dependent on the degree of oxidation, as well as on the pH and temperature of the media (Fig. 8b and c) [58].

One can create gels solely from G-blocks isolated from alginate; partial oxidation of the G-blocks enables degradable gel formation. For example, polyguluronate (PG) was isolated from alginate at pH 2.85, and then oxidized with sodium periodate to prepare PAG. PAG can be covalently cross-linked with AAD to form gels (Fig. 9), in place of ionic cross-linking. The reaction between aldehydes and hydrazides is very fast and the resulting hydrazone bonds

will be labile to hydrolysis, which leads to gels that degrade in aqueous media. The higher the AAD concentration used to form gels, the slower the degradation rate. The degradation rate and mechanical properties have been regarded as critical factors in new tissue formation in tissue engineering, but are typically coupled, making it difficult to decouple their effects. Surprisingly, PAG gels with a high content of dangling single-end AAD molecules showed retarded degradation behavior, irrespective of the low cross-linking density, because this large number of single-end AAD molecules allows re-cross-linking of PAG chains following hydrolysis of the initial hydrazone bond [59]. This finding clearly indicates that one can make soft gels that degrade slowly over time, unlike conventional gels.

In addition, the degradation rate and mechanical properties of alginate gels can be decoupled by adjusting the molecular weight distribution of alginate. Binary alginate gels have been formed from partially oxidized alginates with low and high molecular weights by either ionic or covalent cross-linking. The molecular weight (MW) of alginate was varied via γ -irradiation; the length of G-blocks was nearly unchanged by this treatment. Increasing the fraction of low MW alginate to 0.50 maintained the mechanical stiffness of the gels compared with the high MW alginate gels, but led to faster degradation, irrespective of the cross-linking method [60]. Alternatively, gels prepared from two types of alginate with a size mismatch in G-block length exhibited a more rapid ion exchange, and resultant gel dissolution [61]. These various approaches may be useful alone or in combination for manipulating the physical properties of various hydrogels in the development of drug delivery and cell transplantation vehicles.

4. Biomedical applications

4.1. Pharmaceutical applications

The conventional role of alginate in pharmaceuticals includes serving as thickening, gel forming, and stabilizing agents, as alginate can play a significant role in controlled-release drug products. Oral dosage forms are currently the most frequent use of alginate in pharmaceutical applications, but the use of alginate hydrogels as depots for tissue localized drug delivery is growing. Here, we briefly describe recent progress in controlled drug delivery using alginate and/or its derivatives.

4.1.1. Delivery of small chemical drugs

Alginate gels have been investigated for the delivery of a variety of low molecular weight drugs, and are likely most useful when a primary or secondary bond between the drug and the alginate can be exploited to regulate the kinetics of drug release. Alginate gels are typically nanoporous (pore size ~ 5 nm) [62], leading to rapid diffusion of small molecules through the gel. For example, the release of flurbiprofen from ionically cross-linked, partially oxidized alginate gels is almost complete in 1.5 h. However, incorporation into beads formed from partially oxidized alginate in the presence of both calcium ions and adipic acid dihydrazide (combination of ionic and covalent cross-linking) led to a prolonged release due to the

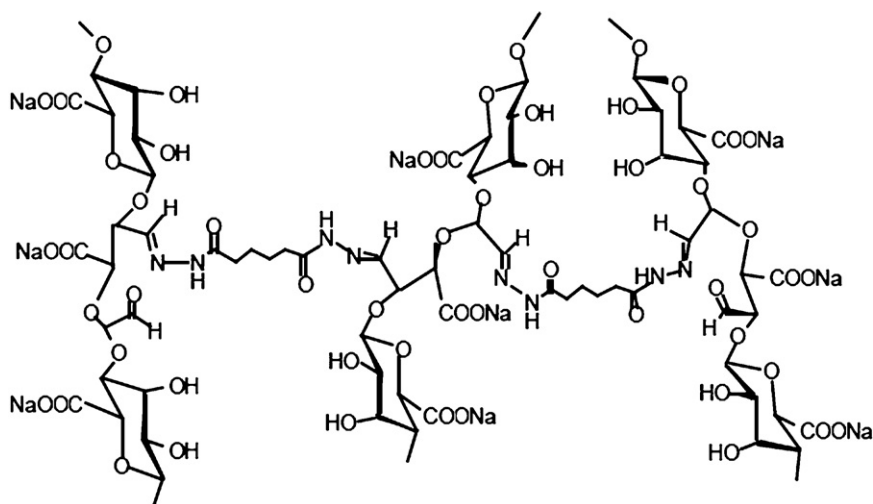


Fig. 9. Chemical structure of poly(aldehyde guluronate) gels cross-linked with adipic acid dihydrazide [59]. Copyright 2000, American Chemical Society, Washington DC, USA.

increased number of cross-links and resultant reduced swelling [63]. The controlled and localized delivery of anti-neoplastic agents has also been achieved using partially oxidized alginate gels. Multiple drugs can be loaded into alginate-based gels for simultaneous or sequential delivery, as the chemical structure of the drug and mode of incorporation will dramatically alter the release kinetics. For example, methotrexate (non-interactive with alginate) was rapidly released by diffusion, while doxorubicin, covalently attached to the alginate, was released via chemical hydrolysis of the cross-linker. Mitoxantrone, ionically complexed to alginate, was only released after the dissociation of the gel [64].

Amphiphilic gel beads have also prepared to modulate the release of hydrophobic drugs. Alginate grafted with poly(ϵ -caprolactone) (PCL) has been cross-linked with calcium ions for controlled delivery of theophylline, a model drug with poor water-solubility. The length of the hydrophobic PCL chains controlled the swelling behavior of the gel beads, and the PCL slowed the release of theophylline. The drug release was complete within 2 h for alginate-g-PCL/ Ca^{2+} beads, but in 1 h for alginate/ Ca^{2+} beads [65]. The sustained release of theophylline was also achieved from carbon nanotube (CNT)-incorporated alginate microspheres. The addition of CNT enhanced the mechanical stability of gels, without affecting the structure and morphology of the microspheres and no significant cytotoxicity was observed, indicating potential application as a delivery carrier to the intestine and colon [66].

Alginate has also been widely exploited in many drug delivery applications in combination with chitosan, as the combination forms ionic complexes. Chitosan is a derivative of chitin, the second most abundant natural polymer in the world, and has a repeating structure of (1,4) linked β -D-glucosamine, with an apparent pK of 6.5. Traditionally, commercial products are composed of 80% β -D-glucosamine and 20% *N*-acetyl- β -D-glucosamine (Fig. 10) [67]. Chitosan is a cationic polymer and has been

widely used in the areas of food, cosmetics, biomedical and pharmaceutical applications, [68], due to its biocompatibility and other favorable properties. Multiparticulate systems of alginate and chitosan containing triamcinolone were prepared by a complex coacervation/ionotropic gelation method for colonic drug delivery. A higher swelling degree and faster drug release were observed from the particulate systems in a simulated enteric environment (pH 7.5), as compared to a simulated gastric environment (pH 1.2) [69]. Magnetic alginate-chitosan beads loaded with albendazole (ABZ) were also prepared for passive targeting to the gastrointestinal tract using physical capture mechanisms (e.g., magnetic field, pH). The beads showed unique pH-dependent swelling behaviors and a continuous release of ABZ [70]. Chitosan-treated alginate microparticles containing all-trans retinoic acid (ATRA) have also been shown to enhance dermal localization and achieved the sustained release of ATRA into the skin [71]. Metronidazole was also entrapped into chitosan-treated alginate beads by an ionotropic gelation method, and the beads were effective in eradication of *Helicobacter pylori* when orally administered into mice [72]. Alginate gels have also been utilized to form a matrix in which depots releasing small drugs can be incorporated; Amoxicillin-loaded chitosan/poly(γ -glutamic acid) nanoparticles have been incorporated into alginate/ Ca^{2+} hydrogels for effec-

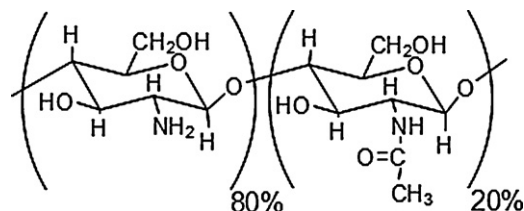


Fig. 10. Chemical structure of chitosan.

tive treatment of *H. pylori* infection. The alginate gel outer layer protected the amoxicillin-loaded nanoparticles in the gastric environment, and facilitated amoxicillin interactions specifically with intercellular spaces, which is the infection site of *H. pylori* [73].

4.1.2. Protein delivery

The protein drug market is rapidly growing, and various protein drugs are now available owing to the development of recombinant DNA technology. Alginate is an excellent candidate for delivery of protein drugs, since proteins can be incorporated into alginate-based formulations under relatively mild conditions that minimize their denaturation, and the gels can protect them from degradation until their release. A variety of strategies have been investigated to control the rate of protein release from alginate gels.

In general, the release rate of proteins from alginate gels is rapid, due to the inherent porosity and hydrophilic nature of the gels. However, heparin binding growth factors such as vascular endothelial growth factor (VEGF) or basic fibroblast growth factor (bFGF) exhibit similar, reversible binding to alginate hydrogels, enabling a sustained and localized release [74,75]. The release in this scenario can be readily manipulated by altering the degradation rate of the gels (e.g., use of partially oxidized alginate), in order to make protein release at least partially dependent on the degradation reaction [75]. Many attempts have been made to further control the release of angiogenic molecules from alginate gels, particularly for factors that are not heparin binding.

Ionically cross-linked alginate microspheres efficiently encapsulated high pI proteins such as lysozyme and chymotrypsin; these proteins appear to physically cross-link the sodium alginate, allowing for more sustained release [76]. Amino group-terminated poly((2-dimethylamino) ethyl methacrylate) has also been reacted with oxidized alginate without using a catalyst, and gel beads were prepared by dropping the aqueous solution of the alginate derivative into an aqueous CaCl_2 solution to form particles for oral delivery of proteins [77]. Alginate was also used as a building block in the synthesis of a tetra-functional acetal-linked polymer network for stimuli-responsive gels with adjustable pore sizes. The gels protected acid-labile proteins such as insulin from denaturation in the gastric environment (pH 1.2), while releasing the loaded protein at near zero-order kinetics in neutral pH [78].

The low encapsulation efficiency and fast release from alginate gels exhibited by many proteins can also be addressed with various cross-linking or encapsulation techniques, and/or by enhancing protein-hydrogel interactions [15]. For example, insulin-loaded alginate microspheres were prepared by blending alginate with anionic polymers (e.g., cellulose acetate phthalate, polyphosphate, dextran sulfate), followed by chitosan-coating in order to protect insulin at gastric pH and obtain its sustained release at intestinal pH [79]. Alginate microspheres have also been coated with *Bombyx mori* silk fibroin using layer-by-layer deposition techniques, which provided mechanically stable shells as well as a diffusion barrier to the encapsulated proteins [80]. A combination of microspheres that serve as a depot for proteins and alginate hydrogels also enables

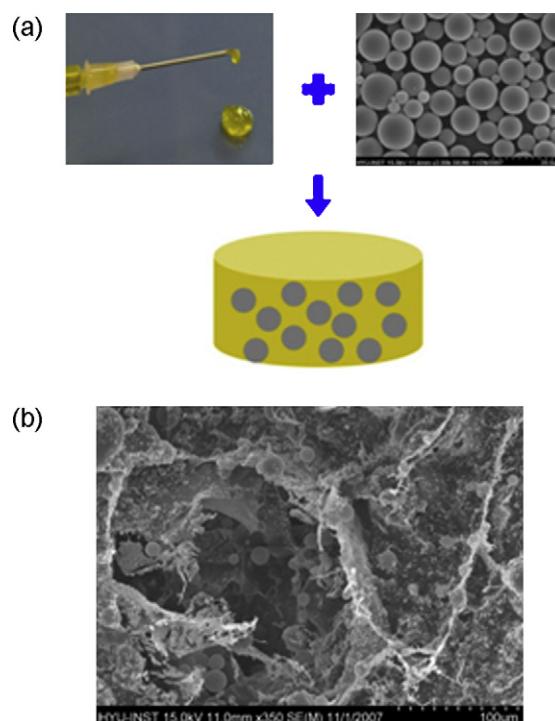


Fig. 11. (a) A microspheres/hydrogel combination system can be prepared by ionic cross-linking of a suspension of poly(D,L-lactide-co-glycolide) (PLGA) microspheres containing protein drugs in an alginate solution. (b) Scanning electron microscopic image clearly shows even dispersion of PLGA microspheres in an alginate hydrogel [81]. Copyright 2009, Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim, Germany.

sustained protein release. Hydrogels loaded with microspheres were prepared by encapsulation of a suspension of poly(D,L-lactide-co-glycolide) (PLGA) microspheres in alginate prior to ionic cross-linking. A homogeneous dispersion of PLGA microspheres was observed by SEM (Fig. 11), and the release of bovine serum albumin (BSA), a model protein, from this combination delivery system was primarily controlled by the mixing ratio between PLGA microspheres and alginate hydrogels, independent of total BSA content and the size of PLGA microspheres used [81]. The release behavior of TAT-HSP27 (heat shock protein 27 fused with transcriptional activator) was also regulated by varying the mixing ratio between microspheres and gels (Fig. 12) [82]. Alginate gels releasing proteins are being widely explored in tissue engineering and regeneration, as described in the following sections on blood vessel, bone, and muscle regeneration.

4.2. Wound dressings

The treatment of acute and chronic wounds is a pressing need in many facets of medicine, and alginate-based wound dressings offer many advantageous features. Traditional wound dressings (e.g., gauze) have provided mainly a barrier function – keeping the wound dry by allowing evaporation of wound exudates while preventing entry of pathogen into the wound [83]. In contrast, modern dressings (e.g., alginate dressings) provide a moist wound

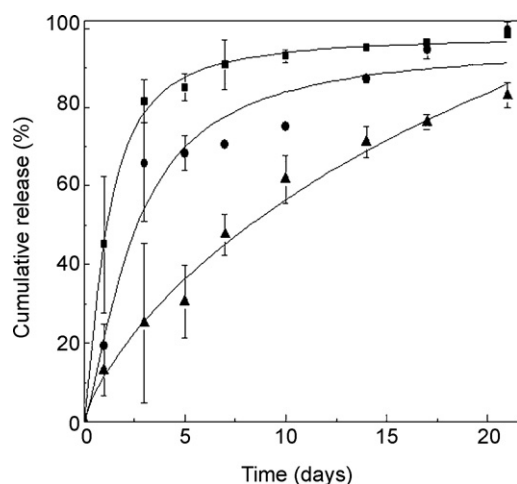


Fig. 12. *In vitro* TAT-HSP27 release from microsphere/hydrogel combination delivery systems prepared at different mixing ratios (■, 0; ●, 1.0; ▲, 1.5; PLGA/alginate = w/w) [82]. Copyright 2009, Elsevier Science Ltd., Oxford, UK.

environment and facilitate wound healing [84]. Alginate dressings are typically produced by ionic cross-linking of an alginate solution with calcium ions to form a gel, followed by processing to form freeze-dried porous sheets (i.e., foam), and fibrous non-woven dressings. Alginate dressings in the dry form absorb wound fluid to re-gel, and the gels then can supply water to a dry wound, maintaining a physiologically moist microenvironment and minimizing bacterial infection at the wound site. These functions can also promote granulation tissue formation, rapid epithelialization, and healing. Various alginate dressings including Algicell™ (Derma Sciences) AlgiSite M™ (Smith & Nephew), Comfeel Plus™ (Coloplast), Kaltostat™ (ConvaTec), Sorbsan™ (UDL Laboratories), and Tegagen™ (3M Healthcare) are commercially available.

A variety of more functional and bioactive alginate-based wound dressings have also been studied to date. The sustained release of dibutyryl cyclic adenosine monophosphate, a regulator of human keratinocyte proliferation, from partially oxidized alginate gels accelerated wound healing, leading to complete re-epithelialization of full-thickness wounds within 10 days in a rat model [85]. Alginate gels releasing stromal cell-derived factor-1 were

also effective in accelerating wound closure rates and reducing scar formation in pigs with acute surgical wounds [86]. Incorporation of silver into alginate dressings increased antimicrobial activity and improved the binding affinity for elastase, matrix metalloproteinases-2 (MMP-2), and proinflammatory cytokines (e.g., TNF- α , IL-8). The addition of silver into alginate dressings also enhanced the antioxidant capacity [87]. Alginate fibers cross-linked with zinc ions have also been proposed for wound dressings, as zinc ions may generate immunomodulatory and anti-microbial effects, as well as enhanced keratinocyte migration and increased levels of endogenous growth factors [88]. Blends of alginate, chitin/chitosan, and fucoidan gels have been reported to provide a moist healing environment in rats, with an ease of application and removal [89].

4.3. Cell culture

Alginate gels are increasingly being utilized as a model system for mammalian cell culture in biomedical studies. These gels can be readily adapted to serve as either 2-D or more physiologically relevant 3-D culture systems. The lack of mammalian cell receptors for alginate, combined with the low protein adsorption to alginate gels allows these materials to serve in many ways as an ideal blank slate, upon which highly specific and quantitative modes for cell adhesion can be incorporated (e.g., coupling of synthetic peptides specific for cellular adhesion receptors). Further, basic findings uncovered with *in vitro* studies can be readily translated *in vivo*, due to the biocompatibility and easy introduction of alginate into the body.

RGD-modified alginate gels have been most frequently used as *in vitro* cell culture substrates to date. The presence of RGD peptides in alginate gels allows one to control the phenotype of interacting myoblasts [90], chondrocytes [91,92], osteoblasts [93], ovarian follicle [94], as well as bone marrow stromal cells (BMSCs) [95–97]. For example, the adhesion and proliferation of myoblasts cultured on alginate gels were dramatically enhanced by chemical conjugation of RGD peptides to the alginate backbone, compared with non-modified alginate gels (Fig. 13) [98]. Further, the number of cells adherent to the gels, as well as the growth rate, were strongly dependent on the bulk RGD density in the gels. The length of the spacer arm

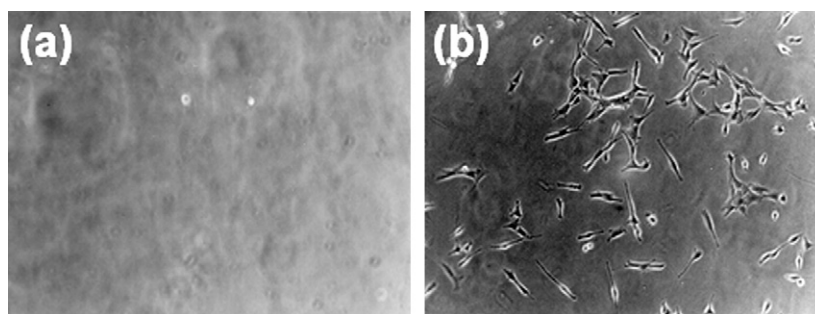


Fig. 13. Optical microscopic images of C2C12 myoblasts adhered to the surface of (a) non-peptide-modified alginate gels and (b) RGD-modified alginate gels. No cells attach and spread on the unmodified gels, while large numbers of well-spread cells are found on the RGD-modified alginate. Images were taken after 24 h culture at 100 \times magnification [98]. Copyright 1999, Elsevier Science Ltd., Oxford, UK.

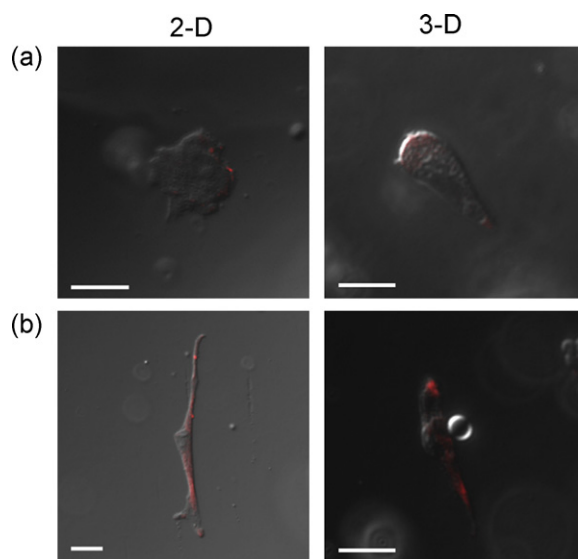


Fig. 14. Confocal microscopic images of primary human fibroblasts cultured on alginate gels (2-D) modified with either (a) RGDSP or (b) G_{12} RGDSP, and cells encapsulated within the same two types of gels (3-D). Images were taken after cells were treated with anti-vinculin antibody, followed by visualization with rhodamine-conjugated donkey anti-mouse IgG (scale bar, 20 μ m) [99]. Cells cultured with G_{12} RGDSP-alginate gels, both in 2-D and 3-D, clearly display focal contact formation, which is a sign of strong cell adhesion and regulates cell spreading and migration, as demonstrated by positive immunostaining for vinculin. Copyright 2010, Elsevier Science Ltd., Oxford, UK.

between the RGD peptide and the alginate chain is a key parameter in regulation of cellular responses. The adhesion and growth of primary human fibroblasts cultured on alginate gels modified with a peptide with the sequence of (glycine) $_n$ -arginine-glycine-aspartic acid-serine-proline (G_n RGDSP) was dramatically influenced by the spacer arm length, irrespective of the same total concentration of the peptides in the gels (Fig. 14). At least four glycine units as a spacer arm allowed proper binding to the cellular receptors, but using more than 12 glycine units led to no further improvement in cell adhesion and growth [99]. The number of RGD peptides per alginate chain, and the spacing between clusters of RGD peptides, even independently of the overall density of RGD ligands, dramatically impact the response of cells to RGD-modified alginate gels [100,101], likely due to the ability of these variables to affect the clustering of integrin receptors [102]. While the presence of the RGD ligands typically enhances cell adhesion and differentiation, chondrogenic gene expression and matrix accumulation of BMSCs encapsulated in RGD-alginate gels (3-D) was inhibited with an increase of the RGD density *in vitro* [103]. Interestingly, alginate gels have recently been formed in a microfluidic device through light-triggered release of caged calcium using DM-nitrophenTM compounds, and used as a 3-D cell culture substrate. Preosteoblasts (MC3T3-E1) and human umbilical vein endothelial cells were co-cultured in the microfluidic device using photo-patterning of alginate hydrogels, and this system may provide a useful means for integrating 3-D culture microenvironments into microfluidic systems [104].

The affinity of the cell adhesion peptides for cellular receptors is also important to the cell response, as demonstrated with studies using high affinity cyclic RGD-containing peptides. Alginate gels presenting cyclic RGD peptides (glycine $_4$ -cystine-arginine-glycine-aspartate-serine-proline-cystine; G_4 CRGDSPC) enhanced osteogenic differentiation of stem cells (primary human bone marrow stromal cells and mouse bone marrow stromal D1 cell line) better than gels modified with linear RGD peptides [105]. Cyclic RGD peptides are more resistant to proteolysis [106], and have higher binding affinity and selectivity than linear RGD peptides [107,108]. Synthesis of alginate derivatives presenting appropriate cyclic RGD peptides capable of promoting stem cell differentiation may enhance tissue regeneration by reducing the need for exogenous soluble factors.

Recent studies utilizing alginate gels as 3-D cell culture substrates have revealed key insights regarding stem cell and cancer biology. The fate of mesenchymal stem cells was demonstrated to be controlled by the elastic modulus of the RGD-alginate gels in which they were encapsulated, as differentiation down fat and bone pathways was promoted at different values of gel stiffness. Strikingly, and in contrast to 2-D culture systems used in previous mechanotransduction studies, the control over stem cell fate was related to the number of adhesive bonds formed between the gel and the cells, as well as alterations in the receptors cells utilized to adhere to the RGD peptides in 3-D versus 2-D culture. The cells actively reorganized on the nanoscale the adhesion ligands presented from the gels [109]. Alginate gels have also been used to examine how a 3-D culture microenvironment influences cancer cell signaling and tumor vascularization. Integrin engagement within a 3-D tumor microenvironment (i.e., encapsulation in RGD-alginate gels) dramatically altered how cancer cells signal to recruit blood vessels, and this finding may lead to the development of new anti-angiogenic cancer therapies [110].

A crucial limitation of most 3-D cell culture systems is the difficulty in analyzing and quantifying cell-matrix interactions, particularly in a non-invasive, real time manner. However the development of several FRET techniques has recently enabled a previously unprecedented ability to quantitatively probe the relation between cell adhesion and decision-making. In one FRET technique, cell membranes are pre-stained with fluorescent molecules (i.e., acceptor), and a different fluorophore (i.e., donor) can be coupled to the cell adhesion peptides conjugated to the polymer chains (Fig. 3) [33]. This FRET technique allows one to quantify cell receptor-ligand binding, and a similar FRET technique provides information on cell-mediated rearrangements, at the nanometer size scale, of the adhesion ligands attached to gels [111,112]. The relationship between cell behavior and the number of receptor-ligand bonds was investigated by encapsulating either preosteoblasts (MC3T3-E1) or myoblasts (C2C12) in alginate gels presenting RGD peptides using a FRET technique [113]. The adhesive interactions can be directly visualized, as the green emission of fluorescein in the cell membrane was greatly decreased and the red emission of rhodamine at the interface between cells and gels was

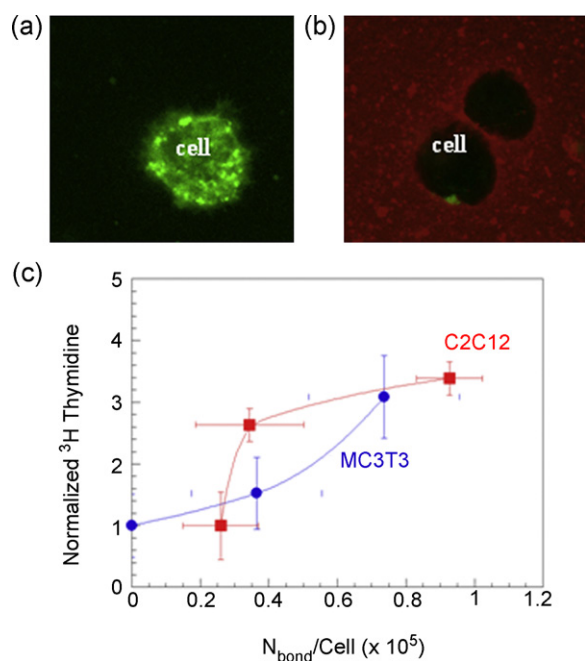


Fig. 15. Directly visualizing and quantifying cell-gel adhesions, and their relation to cell phenotype. (a) A strong green emission of fluorescein in the cell membrane within unmodified gels was observed. (b) The color intensity of fluorescein in the cell membrane was greatly decreased and the color intensity of rhodamine at the interface between cells and gels was increased when cells were encapsulated in alginate gels presenting rhodamine- $\text{G}_4\text{RGDASSKY}$, due to fluorescence resonance energy transfer (FRET). (c) The relationship between the amount of [^3H]thymidine uptake by cells (indication of cell multiplication) and $N_{\text{bond}}/\text{cell}$ for two cell types: muscle cells (C2C12) and bone cells (MC3T3-E1). The number of cell receptor–ligand bonds (N_{bond}) was determined using the FRET measurements [113]. Copyright 2006, National Academy of Sciences, Washington, DC, USA.

increased when the cells were encapsulated in rhodamine- $\text{G}_4\text{RGDASSKY}$ -alginate gels, due to FRET (Fig. 15). The proliferation and differentiation of both cell types were significantly dependent on the number of receptor–ligand bonds calculated using the FRET signal. This type of analysis may allow one to predict cell behavior, particularly in 3-D culture, and to design proper 3-D cell culture substrates for many applications.

4.4. Tissue regeneration with protein and cell delivery

Alginate gels have been widely explored over the past several decades as a vehicle to deliver proteins or cell populations that can direct the regeneration or engineering of various tissues and organs in the body. The various applications of alginate gels have exploited the wide range of gelling approaches, physical properties, cell adhesion, and degradation behavior of this family of materials. There are limits to the size of a regenerative agent that can be released from alginate hydrogels with diffusion, due to the pore size of ~ 5 nm. Most proteins readily diffuse out from alginate gels, even in the absence of gel degradation, although degradation can speed release [75]. Molecules too large to have significant diffusion-based release can still be delivered if the gel degrades. For example, condensed plasmid

DNA (size ~ 100 nm) [177] can be released from degrading alginate gels [171], and antibody might be released from alginate gels by the same mechanism. Cells must migrate out of alginate hydrogels, and/or be released as the gel degrades. There have been a number of qualitative studies of cell migration in various nanoporous alginate gels, in which migration occurred but was not quantitatively analyzed [178]. The number of cells migrating outward as a function of the porosity and RGD presentation in macroporous alginate gels has been quantified [124,150], as has the migration speed of cells both within a macroporous RGD-alginate gel and in the surrounding ECM gel [179].

4.4.1. Blood vessels

Networks of blood vessels are critical for transport of oxygen and nutrients to all tissues, removal of metabolic waste products, and trafficking of stem and progenitor cells, which are critical for organ growth in the embryo and wound repair in the adult [114]. New blood vessel formation is crucial for tissue engineering, as cells more than a few hundred microns from blood vessels will suffer from hypoxia and limited supply of nutrients. In addition, new blood vessel formation (neovascularization) is also a promising alternative to treat patients suffering from restricted or obstructed blood flow caused by coronary and peripheral arterial diseases. Neovascularization can be achieved by transplantation of various cell types into the body, delivery of angiogenic molecules such as recombinant proteins or genes, or a combination of both. Spatiotemporal control over the delivery of therapeutic molecules has been especially attractive for neovascularization [115], and alginate gels have been widely exploited as a delivery vehicle of various angiogenic molecules.

The most widely examined application of alginate gels to promote blood vessel formation has exploited their ability to provide a sustained and localized release of heparin binding growth factors such as VEGF [73,116,117]. Injection of alginate gels directly into ischemic muscle tissue has been demonstrated to lead to long-term (>14 days) release of VEGF in the ischemic tissue, and formation of VEGF gradients in the surrounding tissue capable of guiding new capillary formation into the ischemic tissue and relieving tissue ischemia [74]. The differential binding of various growth factors to alginate has also been exploited to provide sequential delivery of factors involved in early and late stages of angiogenesis, in order to promote maturation of the new networks of vessels, and increased functionality. Sequential delivery of VEGF followed by platelet-derived growth factor-BB (PDGF-BB) using alginate gels resulted in enhanced blood vessel formation, maturation and function when injected into ischemic hindlimbs of mice [118] and sites of myocardial infarction [119]. In general, VEGF plays an important role in initiating angiogenesis and forming new capillaries, while PDGF promotes the maturation of the resulting capillaries into larger, more functional vessels. Two strategies have been used to achieve sequential release of these factors from alginate gels. In the first, the PDGF was pre-encapsulated into poly(lactide-co-glycolide) (PLG) microspheres that were then encapsulated into gels with free VEGF [118]. In the second approach, the greater heparin binding of PDGF was exploited to slow down its

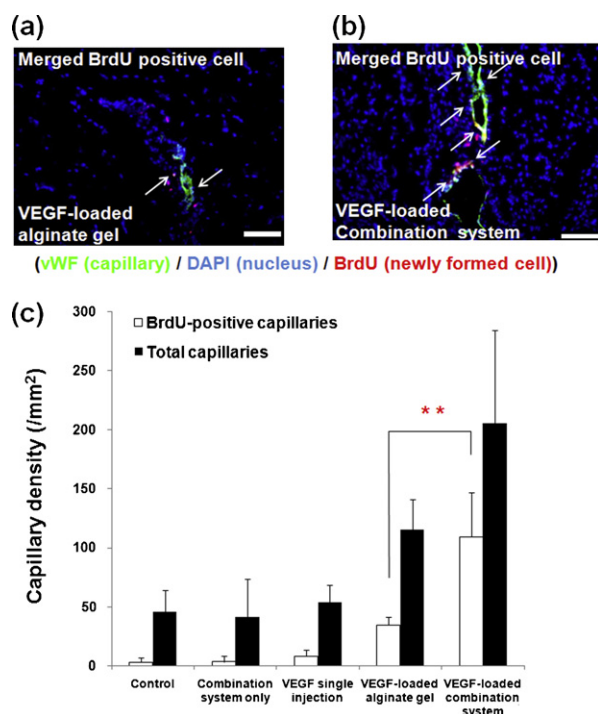


Fig. 16. Tissues from mouse ischemic hindlimbs stained with anti-von Willebrand factor (vWF) (indicator of endothelial cells lining blood vessels) 4 weeks post-treatment with either (a) VEGF-loaded alginate hydrogel or (b) VEGF-loaded microsphere/hydrogel combination system. Blue, green and red colors represent DAPI (cell nuclei), vWF, and BrdU (indicating cells actively multiplying), respectively. Arrows indicate newly formed capillaries (scale bar, 20 μ m). (c) Quantification of the capillary density in tissue [121]. Copyright 2010, Springer, New York, USA.

release, relative to VEGF, by simply encapsulating both factors in free form into the gels [119]. In both situations, VEGF was released faster than PDGF. Similarly, a sequential release of IGF-1 followed by HGF from alginate-sulfate gels preserved scar thickness, attenuated infarct expansion, and reduced scar fibrosis after 4 weeks, as well as enhanced mature blood vessel formation at the infarct site [120]. The release of VEGF can also be slowed by first encapsulating it into PLG microspheres, and including these microspheres in alginate gels. This approach to VEGF delivery enhanced the number of newly formed blood vessels in ischemic hindlimbs of mice (Fig. 16) [121].

The transplantation of cells is an attractive approach to promote new blood vessel formation, particularly when host cells capable of responding to delivered growth factors are lacking or dysfunctional. However, transplantation of endothelial cells or endothelial progenitor cells has not been effective in clinical trials, likely due to massive death of transplanted cells, insufficient integration of the transplanted cells with the host vascular network, and poor recruitment of host smooth muscle cells to promote mature blood vessel formation. Delivery of VEGF from alginate-PLG vehicles in concert with endothelial cell transplantation has been demonstrated to significantly increase the number of blood vessels formed by transplanted endothelial cells [122]. Further, transplantation of endothelial cells in combination with dual delivery of VEGF and monocyte

chemotactic protein-1 (MCP-1) using alginate microparticles enhanced functional vessel formation and increased the number of smooth muscle cell-invested, mature vessels in mice [123]. Recently, alginate delivery vehicles capable of actively promoting the outward migration of transplanted endothelial progenitor cells and their dispersion throughout ischemic tissues have been developed [124]. Endothelial progenitor cells were transplanted on vehicles formed from RGD-alginate to allow for cell adhesion and migration, and the gels also released VEGF to promote cell movement. Mice that would otherwise suffer from complete auto-amputation of an ischemic hindlimb demonstrated normal limb structure and function with this new delivery approach [124].

4.4.2. Bone

Despite recent progress, treatment of bone injuries is still often limited due to poor healing, and alginate gels have found potential in bone regeneration by delivery of osteoinductive factors, bone-forming cells, or combination of both. Alginate gels have advantages for bone and cartilage regeneration, as compared to other materials, due to their ability to be introduced into the body in a minimally invasive manner, their ability to fill irregularly shaped defects, and the ease of chemical modification with adhesion ligands (e.g., RGD) and controlled release of tissue induction factors (e.g., BMP, TGF- β). However, alginate gels do not have sufficient mechanical properties to allow load bearing in the initial stages of regeneration without fixation. They are also not inherently degradable in physiological conditions, as reviewed earlier, highlighting the need to control their degradation in order that residual gels do not interfere with regeneration. Alginate gels have proven useful in animal models for the delivery of growth factors that can effectively drive bone regeneration (e.g., bone morphogenetic proteins). The use of RGD-alginate gels allows complete regeneration of critical-sized femoral defects in rodents with a low dose of BMP [125]. Alginate gels that deliver DNA encoding bone morphogenetic proteins (BMPs) have also demonstrated significant bone tissue can be regenerated [126,127]. The delivery of multiple factors, either in combination or sequence, is also being explored, in a similar manner as described for angiogenesis. Sequential delivery of BMP-2 and BMP-7 using alginate gels enhanced osteogenic differentiation of bone marrow derived stem cells *in vitro* [128], and co-delivery of BMP-2 and VEGF releasing from alginate gels enhanced the repair and regeneration of critical sized bone defects [129].

When host cells capable of responding to morphogens are lacking, or one desires to accelerate tissue formation, alginate-based devices may also be used to transplant cell populations that directly participate in bone formation. Transplantation of primary rat calvarial osteoblasts into mice using RGD-alginate gels enhanced *in vivo* bone formation [34], as compared to control alginate gels. In addition, co-transplantation of primary chondrocytes and osteoblasts into mice using RGD-alginate gels enabled the formation of growth-plate-like structures, which may be potentially used to replace dysfunctional epiphyses (Fig. 17) [130]. Degradable and injectable alginate-derived gels, composed of PAG and AAD, have also been mixed

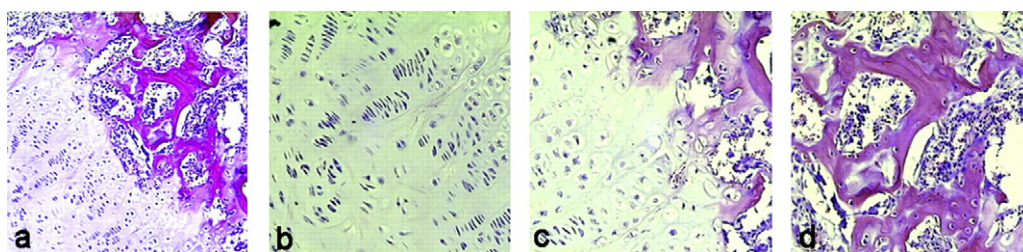


Fig. 17. (a) Growth-plate-like structures formed by co-transplantation of chondrocytes and osteoblasts, similar to that observed in developing long bones (100 \times). Magnified images of the (b) cartilage, (c) transition, and (d) bone and marrow space regions (200 \times magnification) [130]. Copyright 2002, National Academy of Sciences, Washington, DC, USA.

with rat primary calvarial osteoblasts, and subcutaneously injected into the backs of mice. Mineralized bone tissues were observed after 9 weeks [131].

The transplantation of stem cells using alginate hydrogels has been widely explored in bone tissue engineering. The thickness of calcium cross-linked alginate gels was demonstrated to alter the behavior of rat bone marrow cells; however, different geometries did not influence cell differentiation [132]. Bone marrow stromal cells, after being induced down the osteoblast pathway *in vitro* and mixed with calcium cross-linked alginate gels, repaired horizontal alveolar bone defects in dogs [133]. Alginate/chitosan gels entrapping mesenchymal stem cells and bone morphogenetic protein-2 also showed potential for trabecular bone formation in mice [134].

Alginate has also been combined with inorganic materials to enhance bone tissue formation. Alginate/hydroxyapatite (HAP) composite scaffolds with interconnected porous structures were prepared by a phase separation method, which enhanced the adhesion of osteosarcoma cells [135]. Cell-encapsulating alginate gel beads were introduced into calcium phosphate cement, and demonstrated potential for bone tissue engineering under moderate stress-bearing conditions [136]. In addition, alginate gels containing collagen type I and β -tricalcium phosphate enhanced adhesion and proliferation of human bone marrow stromal cells that do not readily attach or proliferate on pure alginate gels [137].

4.4.3. Cartilage

Repair of damaged or degraded cartilage is still one of the major challenges facing the orthopedics field, but tissue engineering approaches have recently shown potential in cartilage regeneration. Alginate gels have proved to be useful for transplanting chondrogenic cells to restore damaged cartilage in animal models. Early studies utilized a suspension of chondrocytes in an alginate solution mixed with calcium sulfate, and injected into molds of facial implants in order to produce pre-shaped cartilage. These constructs formed cartilage with three-dimensional shape retention after 30 weeks of subcutaneous implantation into mice and sheep, and the contents of proteoglycan and collagen as well as the elastic modulus of the engineered cartilage reached about 80% of those found in native cartilage [138,139]. Shape-memory alginate gels were subsequently developed to engineer cartilage with desired

shape and size *in vivo* following minimally-invasive delivery. In brief, macroporous alginate gels with predefined geometries were compressed into a significantly smaller form (dry state) and introduced into mice through a small catheter. The gels were then rehydrated *in situ* with a suspension of primary bovine articular chondrocytes, and recovered their original shape and size within 1 h, which allowed cartilage formation in mice with the desired geometry [140].

The use of stem cells in cartilage regeneration is very attractive, due to the invasive and destructive processes required to obtain primary chondrocytes from tissues. Encapsulation in alginate can regulate differentiation of stem cells, and in particular chondrogenesis may be enhanced. It has been demonstrated that chondrogenic lineage of adult stem cells could be regulated via the introduction of soluble factors and biophysical cues in 3D cell culture systems [180]. In addition, it has been hypothesized that chondrogenesis of stem cells relates to the morphology of the encapsulated cells (i.e., rounded cell shape) [181], and alginate gels promote a rounded morphology that may promote the cellular differentiation process [182].

Human mesenchymal stem cells (MSCs) encapsulated in alginate gel beads have been cultured in serum-free medium with the addition of transforming growth factor (TGF)- β 1, dexamethasone, and ascorbate 2-phosphate for more than one week, and demonstrated to form cartilage in large osteochondral defects [141]. Rabbit bone marrow stromal cells cultured in alginate gels have also been injected into osteochondral defects in rabbit knees without the use of a periosteal patch, which significantly enhanced the cellular proliferation and chondrogenic differentiation of BMSCs. This resulted in histologically and mechanically improved tissues in osteochondral defects [142]. The chondrogenic potential of human adipose derived stem cells (hASCs) suggests these cells as a possible cell source for cartilage regeneration, and chondrogenic differentiation of hASCs seeded in alginate gels was greatly improved in the presence of TGF- β 1 [143]. Pre-differentiated hASCs induced by transduction with an adenovirus carrying a TGF- β 2 plasmid maintained the chondrogenic phenotype *in vivo* and led to new cartilage formation when the cells were encapsulated into alginate gel beads and subcutaneously transplanted into mice [144].

Coupling of RGD-containing peptides to alginate can greatly enhance adhesive interactions with chondrocytes,

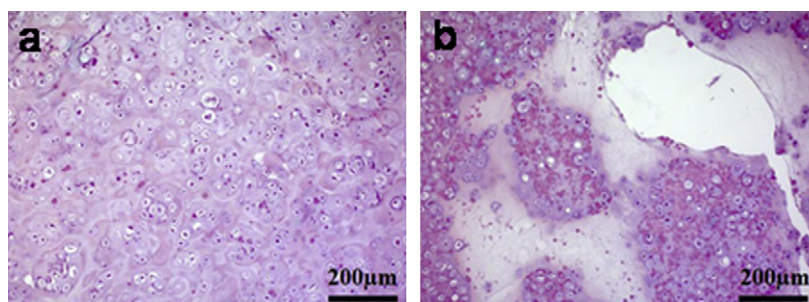


Fig. 18. Images of H&E-stained tissues after 6 weeks following transplantation of primary chondrocytes into mice using either (a) cell cross-linked RGD-alginate gels or (b) media only [145]. Chondrocytes in lacunae (typically found in natural cartilage) were clearly observed for cell cross-linked RGD-alginate gels. Copyright 2009, Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim, Germany.

allowing control of cell phenotype. Shear-reversibly cross-linked gels, which can recover their gel structure after shear-induced breakdown, were formed using an RGD-alginate solution and primary rabbit chondrocytes. These gels were injectable into mice in a minimally invasive manner, and were effective in engineering cartilage *in vivo* (Fig. 18) [145].

4.4.4. Muscle, nerve, pancreas, and liver

Alginate gels are also being actively investigated for their ability to mediate the regeneration and engineering of a variety of other tissues and organs, including skeletal muscle, nerve, pancreas, and liver. Current strategies for skeletal muscle regeneration include cell transplantation, growth factor delivery, or a combination of both approaches [146,147], and alginate gels have found potential in these strategies. A combined delivery of VEGF and insulin-like growth factor-1 (IGF-1) from alginate gels was used to modulate both angiogenesis and myogenesis. The localized and sustained delivery of both growth factors led to significant muscle regeneration and functional muscle formation, due to satellite cell activation and proliferation, and cellular protection from apoptosis by the released factors [148]. Long-term survival and outward migration of primary myoblasts into damaged muscle tissue *in vivo* from RGD-alginate gels were dramatically enhanced by the sustained delivery of hepatocyte growth factor (HGF) and fibroblast growth factor 2 (FGF 2) from the gels [149]. This led to extensive repopulation of host muscle tissues and increased the regeneration of muscle fibers at the wound site [150].

Alginate gels have also been investigated for the repair of the central and peripheral nerve systems. Alginate-based highly anisotropic capillary gels, introduced into acute cervical spinal cord lesions in adult rats, were integrated into the spinal cord parenchyma without major inflammatory responses and directed axonal regrowth [151]. Alginate gels, covalently cross-linked with ethylenediamine, were useful to restore a 50-mm gap in cat sciatic nerves [152], and promoted the outgrowth of regenerating axons and astrocyte reactions at the stump of transected spinal cords in young rats [153]. Alginate gels were also used as glue for repair of peripheral nerve gaps that could not be sutured [154]. Alginate gels may be useful for cell-based neural

therapies, as mouse-derived neural stem cells cultured in calcium alginate beads maintained their capacity for multilineage differentiation into neurons and glial cells [155]. Alginate gels modified with a peptide containing the YIGSR (Tyr-Ile-Gly-Ser-Arg) sequence promoted adhesion of NB2a neuroblastoma cells and neurite outgrowth from the cells, depending on the peptide density in the gels [35].

Tissue engineering is a potential approach to provide hepatic tissues for replacement of a failing liver, and alginate gels encapsulating hepatocytes may offer a suitable platform for developing a bioartificial liver as they are easily manipulated and can be cryopreserved [156,157]. The hydrophilic nature of alginate gels processed to exhibit an interconnected porous structure allows efficient seeding of hepatocytes into the gels, while maintaining high hepatocellular functions [158,159]. Primary rat hepatocytes maintained viability in alginate gels and appeared to synthesize fibronectin, which was deposited on the spheroids and promoted their functional expression [160]. Hepatocyte engraftment was improved when hepatocytes were transplanted into the liver lobe of Lewis rats using VEGF-releasing porous alginate gels [161].

One of the first applications of alginate gels in tissue engineering involved the transplantation of encapsulated pancreatic islet allografts and xenografts in an effort to cure Type I diabetes. In this approach, the gel was used to provide protection from the host immune system, in order to avoid the use of immunosuppressive drugs that would otherwise be required to prevent graft rejection. This approach has been successfully used to treat animal models of Type I diabetes without the use of immunosuppressive drugs [162–164]. These alginate beads encapsulating islets are generally coated with poly(amino acids), such as poly-L-lysine, to decrease the outer pore size, while keeping a liquid core structure [165]. The transplant volume of microencapsulated islet cells can be reduced by choosing an appropriate alginate composition and purity [166]. However, the mechanical and chemical instability of alginate beads is believed to be a limiting factor for long term survival of the transplanted islets *in vivo*, prompting continued investigations into the use of different poly(amino acids) as a coating material and the use of various microfabrication methods [167].

5. Conclusions and future perspectives

Alginate has demonstrated great utility and potential as a biomaterial for many biomedical applications, particularly in the areas of wound healing, drug delivery, *in vitro* cell culture, and tissue engineering. The most attractive features of alginate for these applications include biocompatibility, mild gelation conditions, and simple modifications to prepare alginate derivatives with new properties. Alginate has a track record of safe clinical uses as a wound healing dressing material and pharmaceutical component, and has been safely implanted in a variety of applications, including islet transplantation for treatment of type 1 diabetes [168] and chondrocyte transplantation for treatment of urinary incontinence and vesicoureteral reflux [169,170]. A chemically modified alginate has also been widely used as a carrier to promote periodontal regeneration [176]. Like other hydrogels, however, alginate gels have very limited mechanical stiffness, and more generally physical properties. A continuing challenge is matching the physical properties of alginate gels to the need in a particular application. Consideration of the range of different available cross-linking strategies, using molecules with various chemical structures, molecular weights, and cross-linking functionality will often yield gels suitable for each application. Although not a focus here, many of the concepts reviewed in this article are directly relevant to cell encapsulation strategies. However, covalent cross-linking reactions can cause toxicity to the cells to be encapsulated, and an appropriate choice of cell-compatible chemical reagents (e.g., initiator), and thorough removal of unreacted reagents and by-products will likely be needed in those applications [49].

As one looks to the future, the alginate-based materials used in medicine are likely to evolve considerably. While alginate gels are already used clinically in wound healing applications, they play a fairly passive role. Future dressings will likely play a much more active role. One or more bioactive agents that facilitate wound healing can be incorporated into alginate dressings, as these gels have demonstrated utility in maintaining local concentrations of biological factors, such as proteins, for extended time periods. In wound healing, and more generally drug delivery applications, precise control over the delivery of single vs. multiple drugs, or sustained vs. sequential release in response to external environmental changes is highly desirable.

Dynamical control over delivery can potentially improve the safety and effectiveness of drugs, and provide new therapies. On-demand drug release from alginate gels in response to external cues such as mechanical signals [172] and magnetic fields [173] can be used to design active depots of many drugs, including therapeutic cells. The introduction of appropriate cell-interactive features to alginate will also be crucial in many tissue engineering applications. The type of adhesion ligands and their spatial organization in gels are key variables, as they can regulate cell phenotype and the resultant function of regenerated tissues. While RGD peptides have been extensively exploited to date as a cell adhesion ligand, multiple ligands and/or a combination of ligands and soluble factors may

be required to properly produce replacement tissues and organs.

Furthering our current understating of fundamental properties of alginate, and developing new types of cell and tissue-interactive alginate gels may enable future advances in biomedical science and engineering. The design and creation of new alginate polymers with better or distinct properties can potentially be achieved using genetic engineering techniques to control bacterial synthesis. Various polypeptides and proteins with improved structural properties and novel functions have already been prepared by this approach and explored for biomedical applications [174,175]. The ability to engineer new classes of alginates with precisely controlled chemical and physical characteristics, unlike the limited repertoire available from natural sources, designed for a specific application could revolutionize the use of these materials.

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