

$$q_1 = \text{HEAT CONVECTION} = h_1 A (T_{\infty} - T_{i1}) = \frac{T_{\infty} - T_{i1}}{1/h_1 A}$$

$$q_2 = \text{HEAT CONDUCTION} = \frac{k A (T_{i1} - T_{i2})}{L} = \frac{T_{i1} - T_{i2}}{L/kA}$$

$$q_3 = \text{HEAT CONVECTION} = h_2 A (T_{i2} - T_{out}) = \frac{T_{i2} - T_{out}}{1/h_2 A}$$

Steady state $q_1 = q_2 = q_3 = q$

$$T_{\infty} - T_{i1} = q \left(\frac{1}{h_1 A} \right) \quad (1)$$

$$T_{i1} - T_{i2} = q / (L/kA) \quad (2)$$

$$T_{i2} - T_{out} = q \left[\frac{1}{h_2 A} \right] \quad (3) \quad (2)$$

Sum Eqs (1) to (3)

$$\cancel{T_{oo} - T_{i1}} + \cancel{T_{i1} - T_{i2}} + \cancel{T_{i2} - T_{out}} = q \left[\frac{1}{h_1 A} + \frac{L}{K A} + \frac{1}{h_2 A} \right]$$

$$T_{oo} - T_{out} = q \left[\frac{1}{h_1 A} + \frac{L}{K A} + \frac{1}{h_2 A} \right]$$

$$q = \frac{T_{oo} - T_{out}}{\left[\frac{1}{h_1 A} + \frac{L}{K A} + \frac{1}{h_2 A} \right]}$$

$\underbrace{\hspace{10em}}_{\Sigma \text{ Resistances}}$

$$\boxed{\frac{q}{A} = \text{Flux} = \frac{T_{oo} - T_{out}}{\frac{1}{h_1} + \frac{L}{k} + \frac{1}{h_2}}}$$

↳ COMPARE WITH EQUATION
IN SLIDE 7

MAIN DIFFERENCES BETWEEN HT & MT

THERMAL
CONDUCTIVITY

DIFFUSION

K

D_{AB}

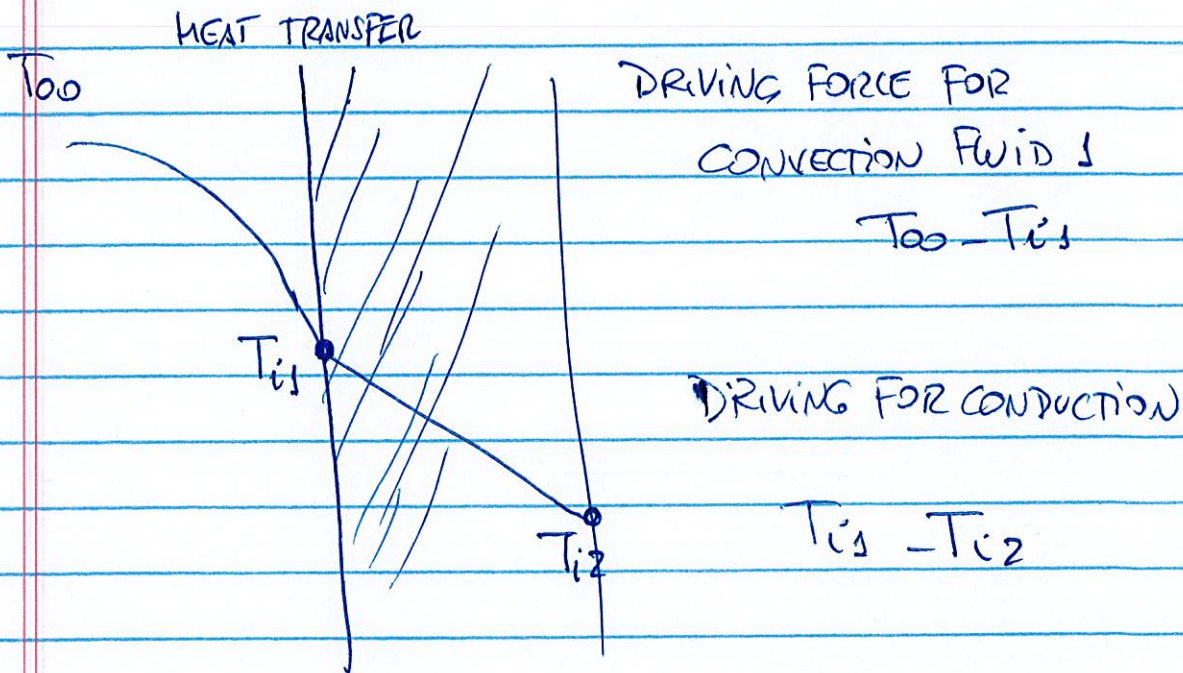
matrix

(3)

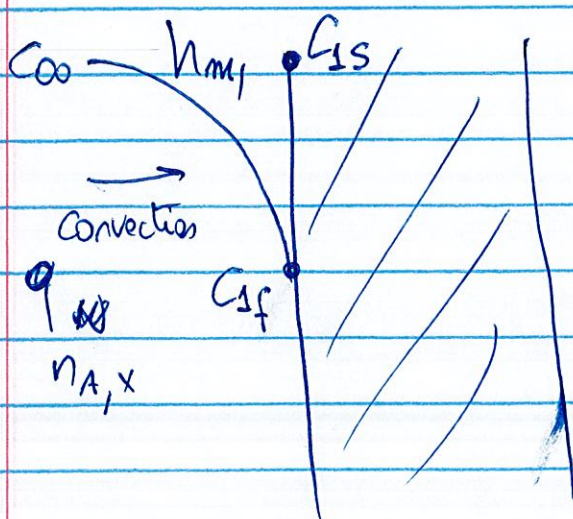
↑ corresponds to diffusing species A

$h_1 \& h_2 \rightarrow h_{m1} \& h_{m2}$

Big difference is that we have K^*



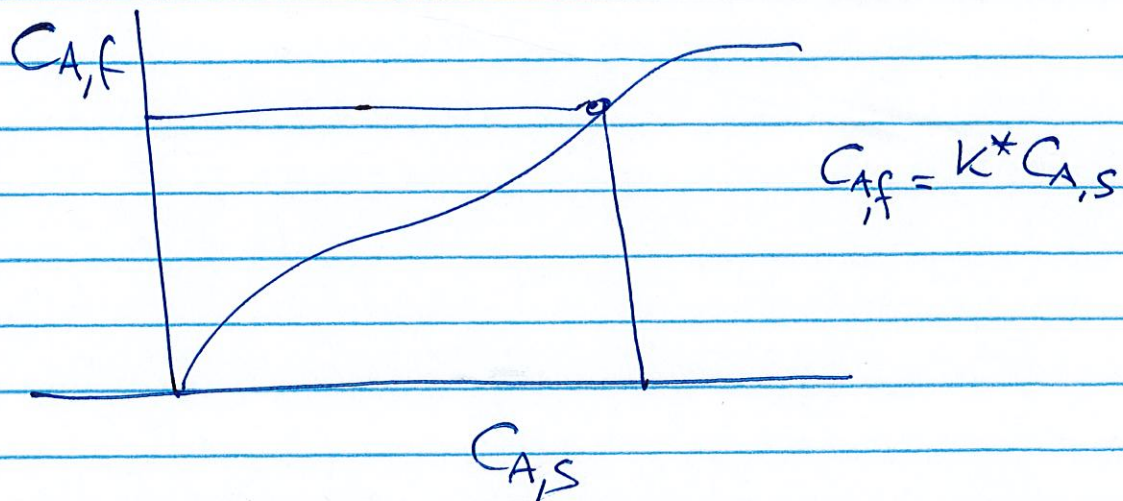
MASS TRANSFER



$$n_{A,x} = h_{m1} [C_{\infty 1} - C_s]$$

↑
f: indicates
that
we are
measuring concentration
of A in fluid phase.

In the solid phase the concentration of (4)
A will be different than the
concentration of A in the fluid phase



— Flux of A $\left[\frac{\text{moles}}{\text{m}^2 \cdot \text{s}} \right]$ in the left fluid.

$$n_{A,x} = h m_1 [C_1 - C_2]$$

↙ concentration of A in the fluid phase.

— Flux of A $\left[\frac{\text{moles}}{\text{m}^2 \cdot \text{s}} \right]$ in the membrane.

$$n_{A,x} = D_{AB} \frac{C_{2i} - C_{1i}}{L}$$

↙ Concentrations in the solid membrane

= Flux of A $\left[\text{moles/m}^2 \cdot \text{s} \right]$ convection on the right

$$n_{A,x} = h m_2 (C_3 - C_4)$$

(5)

$$C_1 - \cancel{C_2} = \frac{n_{A,x}}{h_{m1}} = n_{A,x} \left[\frac{1}{h_{m1}} \right]$$

$$+ \quad \underbrace{K^* C_2}_{C_{2i}} - \underbrace{K^* C_3}_{C_{3i}} = \frac{n_{A,x}}{\frac{D_{AB}}{L}} = n_{A,x} \left[\frac{1}{\frac{D_{AB}}{L}} \right] \Rightarrow \cancel{C_2} - \cancel{C_3} = n_{A,x} \left[\frac{1}{\frac{D_{AB} K^*}{L}} \right]$$

$$+ \quad \cancel{C_3} - C_4 = \frac{n_{A,x}}{h_{m2}} = n_{A,x} \left[\frac{1}{h_{m2}} \right]$$

$$C_1 - C_4 = n_{A,x} \left[\frac{1}{h_{m1}} + \frac{1}{\frac{D_{AB} K^*}{L}} + \frac{1}{h_{m2}} \right]$$

$$n_{A,x} = \frac{C_1 - C_4}{\left[\frac{1}{h_{m1}} + \frac{1}{\frac{D_{AB} K^*}{L}} + \frac{1}{h_{m2}} \right]}$$

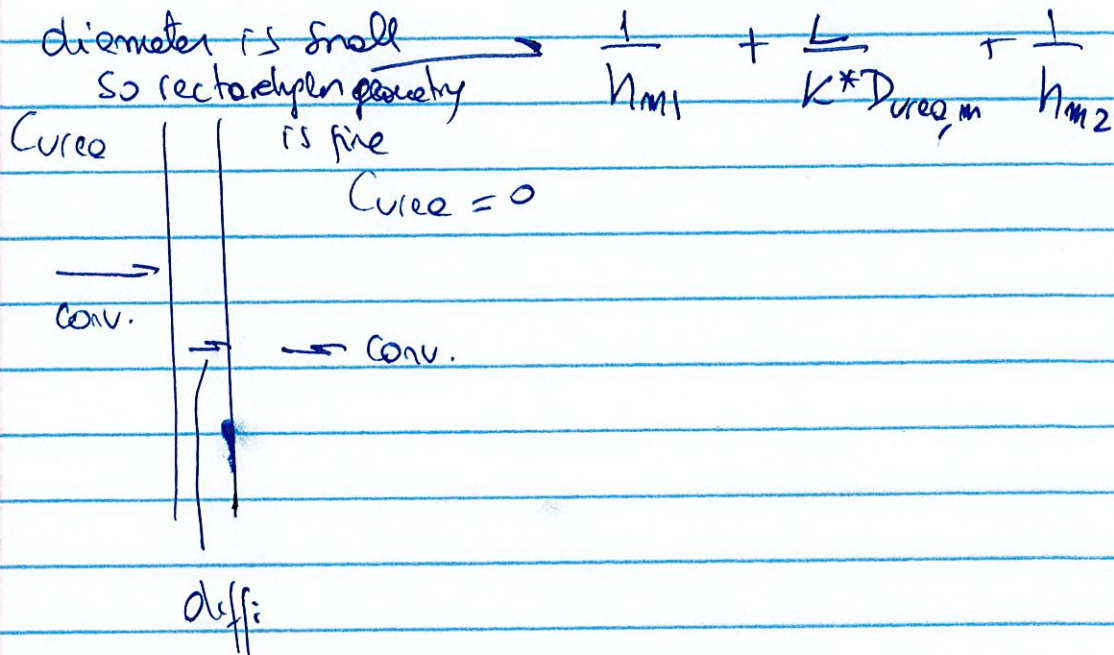
Σ Resistance

$$\boxed{\Delta L \equiv L}$$

For consistency with the
power point

Calculate the Flux of urea in the dialysis membrane (6)

$$n_{\text{urea}} = \frac{(0.02 \text{ g/100 cm}^3) \cdot A}{\text{convert units to g/l or g/m}^3}$$



$$\frac{d^2 C_A(x)}{dx^2} - m^2 C_A(x)$$

characteristic equation

$$r^2 - m^2 = 0 \Rightarrow r_{1,2} = \pm \sqrt{m^2} = \pm m$$

$$C_A(x) = A e^{r_1 x} + B e^{-r_1 x}$$

$$C_A(x) = A e^{mx} + B e^{-mx}$$

We need to get Boundary conditions
to find out A & B

(7)