

PHYS 486: Lecture # 23

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Solving the radial equation

Let

$$R = \frac{u}{r} \quad (1)$$

Then

$$\frac{dR}{dr} = \left(r \frac{du}{dr} - u \right) \frac{1}{r^2} \quad (2)$$

Then, our radial equation becomes

$$-\frac{\hbar^2}{2m} \frac{d^2u}{dr^2} + \left[V + \frac{\hbar^2}{2m} \frac{l(l+1)}{r^2} \right] u = E_n \quad (3)$$

As a toy system, we examine the infinite well in 3-D in spherical coordinates.

$$V(r) = \begin{cases} 0 & r \leq a \\ \infty & r > a \end{cases} \quad (4)$$

Inside the well,

$$\frac{d^2u}{dr^2} = \left[\frac{l(l+1)}{r^2} - k^2 \right] u \quad (5)$$

For $l = 0$, then

$$u(r) = A \sin(kr) + B \cos(kr) \quad (6)$$

Since the WF = u/r , then $B = 0$. with BC: $\sin(ka) = 0$, then $ka = N\pi$. Thus,

$$E = \frac{N^2 \pi^2 \hbar^2}{2ma^2} \quad (7)$$

For $l > 0$, generally:

$$u(r) = Ar j_l(kr) + Br u_l(kr) \quad (8)$$

$$j_l(x) = (-x)^l \left(\frac{1}{x} \frac{d}{dx} \right)^l \frac{\sin(x)}{x} \quad (9)$$

Then,

$$R(r) = Aj_l(kr) \wedge j_l(ka) = 0 \quad (10)$$

The wave function is then

$$\psi = AR(r)Y(\phi, \theta) \quad (11)$$

The Hydrogen Atom

Given a Hydrogen Atom with an immobile proton and an electron that is free to move around. Then

$$V(r) = -\frac{e^2}{4\pi\epsilon_0} \frac{1}{r} \quad (12)$$

Thus, the radial equation is

$$-\frac{\hbar^2}{2m} \frac{d^2u}{dr^2} + \left[-\frac{e^2}{4\pi\epsilon_0} \frac{1}{r} + \frac{\hbar^2}{2m_e} \frac{l(l+1)}{r^2} \right] u = E_u \quad (13)$$