

Estimating the carbon content of the deep mantle with Icelandic melt inclusions

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Abstract

Earth's carbon budget is central to our understanding of the long-term co-evolution of life and the planet. Direct observations of surface reservoirs allow for the detailed quantification of their carbon content. However, the carbon content of Earth's deep interior remains poorly constrained. Here we study olivine-hosted melt inclusions from two Icelandic eruptions, with those from the Miðfell eruption allowing us to investigate the carbon content of the deep mantle. Comparison with the previously studied Borgarhraun eruption highlights the presence of deep, plume-sourced mantle material within the Miðfell source region. Miðfell contains trace element-depleted melt inclusions undersaturated in CO₂, which have high CO₂/Ba ($= 396 \pm 48$) and CO₂/Nb ($= 1832 \pm 316$), though some inclusions preserve even greater relative carbon enrichment. These observations allow us to reconstruct the CO₂ content of the bulk Miðfell source as being > 690 ppm. By identifying that Miðfell is a mixture of depleted and deep mantle components, we can estimate a CO₂ content for the deep mantle component of 1350 ± 350 ppm; a concentration that is over ten times higher than depleted MORB mantle estimates. Assuming that the deep mantle component

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identified in Miðfell is representative of a global reservoir, then with our new CO_2 estimate and by considering a range of representative mantle fractions for this reservoir, we calculate that it contains up to 14 times more carbon than that of the atmosphere, oceans, and crust combined. Our result of elevated CO_2/Ba and CO_2/Nb ratios, and carbon enrichment support geochemical bulk Earth carbon models that call for the presence of carbon-rich deep mantle domains to balance Earth’s relatively carbon-poor upper mantle and surface environment.

Keywords: Deep mantle carbon; Iceland, melt inclusions

1. Introduction

Geological processes have modulated the Earth’s atmospheric carbon content for billions of years (Hayes and Waldbauer, 2006). At the heart of this cycle is a partitioning of carbon between planetary reservoirs, one that has placed the overwhelming majority of carbon in Earth’s crust, mantle, and core (e.g., Dasgupta and Hirschmann, 2010). The carbon flux from these solid-Earth reservoirs to the atmosphere is efficiently returned via silicate weathering (e.g., Walker et al., 1981), maintaining a habitable climate, and closing the loop on a cycle that has helped maintain liquid water at Earth’s surface over almost its entire history (Mojzsis et al., 2001).

To understand how Earth has come to operate such a stable and long-lasting chemical cycle, it is key to know how carbon is distributed among its reservoirs (Hirschmann, 2016). Whilst for the atmosphere, oceans, and to some extent the crust, their carbon content can be measured directly, quantifying the carbon content of mantle reservoirs is more challenging. Two basic problems frustrate accurate estimation of mantle carbon content: (i) the low solubility of carbon in basaltic melts (Shishkina et al., 2010), which means that information on high-carbon mantle domains is preferentially lost as their melts begin to degas at high pressure in the crust or shallow mantle; and (ii) the partial view that

20 volcanism provides of the mantle, with the vast majority of magmas tapping
21 the depleted upper mantle. Only rarely do volcanics display evidence for the
22 involvement of deep-sourced material in their petrogenesis: such occurrences are
23 usually linked to the presence of mantle plumes. These compounding problems
24 cause considerable uncertainty in previously published mantle carbon estimates.

25 In this paper we present new observations that help constrain the amount
26 of carbon in the deep mantle reservoir. In Section 2, we review the geochemical
27 constraints available on mantle carbon, and identify the observational gaps in
28 our present reservoir inventories. We next present our methods (Section 3)
29 and new data (Section 4) from two Icelandic eruptions, demonstrating that
30 their geochemical characteristics are suited to quantify the amount of carbon in
31 the deep mantle. We show that one of these eruptions, Miðfell, contains melt
32 inclusions that are undegassed and relatively enriched in carbon compared to
33 lithophile trace elements of similar mineral-melt compatibility (Section 5). In
34 Section 6, we use these results to place a new constraint on the deep mantle
35 carbon content, and in Section 7 discuss the implications of this constraint for
36 the origin and distribution of carbon in the Earth.

37 2. Measuring mantle carbon

38 An enormous amount of work, especially over the last two decades, has fo-
39 cused on measuring the carbon content of mantle-derived volcanics. Despite this
40 effort, surprisingly few observations provide tight constraints on upper mantle
41 carbon content, and fewer still on deep mantle carbon. This section aims to
42 contextualise the present study with this body of work, and identify the fea-
43 tures that, in subsequent sections, will mark our new observations as distinct
44 from pre-existing data sets.

45 *2.1. The upper mantle*

46 Mid-ocean ridge basalts (MORBs) sample the most accessible mantle reser-
47 voir, the depleted MORB mantle (DMM). Although a shallow reservoir in the
48 context of the mantle, the DMM can be a component of deep-sourced mantle
49 plume magmas, as is the case on Iceland (Stracke, 2012).

50 A key technique for estimating carbon in DMM is to find an incompati-
51 ble lithophile trace element (ITE) to reference MORB carbon concentrations
52 against, thereby providing a control for fractionation during mantle melting
53 and crustal differentiation. Estimates of source mantle ITE concentration (e.g.,
54 Workman and Hart, 2005) then enable calculation of a source carbon content.
55 Ba and Nb are commonly chosen as reference elements because they have sim-
56 ilar compatibility to carbon during peridotite melting (Rosenthal et al., 2015).
57 However, as Ba and Nb are not volatile, accurate estimation of source carbon
58 is dependent on either finding melts that are CO_2 undersaturated, or recon-
59 structing pre-degassed CO_2 . A degassing correction has been applied to some
60 highly vesicular glass samples, which re-coupled CO_2 to ITE concentrations,
61 giving DMM carbon estimates of 393 ± 82 ppm CO_2 using $\text{CO}_2/\text{Nb} = 534$,
62 and 427 ± 45 ppm CO_2 using $\text{CO}_2/\text{Ba} = 106$ (Table 1; 14°N Mid-Atlantic
63 Ridge; Cartigny et al., 2008). We note that whilst carbon exists in the mantle
64 in oxidised and reduced forms (Dasgupta and Hirschmann, 2010), for ease of
65 comparison between data sets, we report total carbon as CO_2 .

66 Only rare suites, such as some pillow glasses erupted at relatively high pres-
67 sure and with intrinsically low carbon content, may show undersaturated volatile
68 concentrations. Michael and Graham (2015) used such samples to estimate a
69 DMM CO_2 of 59 ± 39 ppm (Table 1; $\text{CO}_2/\text{Ba} = 105$; Global MORB). A diffi-
70 culty with this approach is in having confidence that measured melts are truly
71 undegassed. Saal et al. (2002) emphasised how linearly correlated carbon-trace

72 element concentrations can validate a data set as having undersaturated melt
 73 populations. However, Matthews et al. (2017) presented a forward model to
 74 suggest that such correlations can readily arise from the mixing of variably de-
 75 gassed melts at low pressure. They concluded that to have more confidence in
 76 identifying undegassed melts, data will ideally show some melts with high ITE
 77 content that have clearly degassed to their saturation concentration (i.e., no
 78 ITE-carbon correlation), and others with lower ITE and carbon content where
 79 an ITE-carbon correlation persists; such suites evidence incomplete melt mix-
 80 ing. Data sets exhibiting such structure offer the best chance for empirically
 81 separating degassed and undegassed melt populations.

82 Applying the insights from Matthews et al. (2017) can make certain data
 83 sets difficult to empirically validate for carbon undersaturation. For example,
 84 observations of quenched matrix glasses, for which the melt has been largely ho-
 85 mogenised prior to eruption (i.e., many seafloor basalts; Shorttle, 2015; Michael
 86 and Graham, 2015), do not provide data sets that can be internally validated:
 87 giving only one CO₂ and ITE content per eruption. One solution is to use
 88 olivine-hosted melt inclusions, which, when trapped at high pressure, may cap-
 89 ture melts with both undegassed carbon and chemical diversity that has not
 90 been homogenised by mixing (MacLennan, 2008b). Melt inclusion studies have
 91 thus been key in characterising upper mantle carbon: their results suggest that
 92 the DMM contains carbon heterogeneity, as both CO₂/ITE ratios and esti-
 93 mated ITE concentrations vary (Le Voyer et al., 2017), with carbon concen-
 94 trations ranging from 22–427 ppm CO₂ (Table 1; Siqueiros, Saal et al., 2002;
 95 Mid-Atlantic Ridge, Cartigny et al., 2008, Le Voyer et al., 2017).

96 *2.2. The deep mantle*

97 We use the term ‘deep mantle’ to refer loosely to the region of Earth’s con-
 98 vecting mantle that is not represented by the composition of typical MORB.

99 The geometry of this reservoir is poorly constrained, and its upper horizon
 100 could fall anywhere from the mantle transition zone, to the large low shear
 101 velocity provinces above the core-mantle boundary (e.g., Hofmann, 1997). Up-
 102 welling plumes can bring this deep mantle material into the upper mantle and
 103 melt it at shallow levels. Whatever its locus, the deep mantle appears to com-
 104 prise primitive material that has remained isolated for billions of years (e.g.,
 105 Mukhopadhyay, 2012; Peters et al., 2018), as well as crustal material intro-
 106 duced via recycling processes (e.g., Nestola et al., 2018). Both components
 107 could be significant carbon reservoirs: primitive material potentially containing
 108 solar nebular carbon, which dissolved into Earth’s early magma ocean along
 109 with noble gases (e.g., Williams and Mukhopadhyay, 2018); whereas, recycled
 110 material could introduce organic or inorganic carbon from the surface (e.g.,
 111 Nestola et al., 2018).

112 Ocean island basalts (OIBs) are prime candidates in the search for deep
 113 mantle carbon. Geochemical observations have consistently shown OIB to have
 114 more evidence of both recycled and primitive mantle components than MORB
 115 (e.g., Hofmann, 1997), and in many cases geophysical observations support their
 116 lower mantle origins (e.g., Montelli et al., 2006). Beneath Iceland specifically,
 117 the hot, low velocity, plume conduit has been seismically imaged through the
 118 deflected transition zone (Jenkins et al., 2016), and down into the lower mantle
 119 (Yuan and Romanowicz, 2017).

120 The nature of magma generation and eruption at ocean island settings typ-
 121 ically results in OIB degassing (Gonnermann and Mukhopadhyay, 2007): low
 122 eruption or melt inclusion entrapment pressures combine with high initial dis-
 123 solved CO_2 content, derived from low degree melting and/or carbon-rich source
 124 material. This degassing can, however, be an advantage if rather than measuring
 125 the melt, the degassed CO_2 flux is measured. Anderson and Poland (2017) mea-

126 sured CO₂ degassing at Hawaii, along with a melt production rate to estimate
 127 the Hawaiian source mantle carbon content to be $962 \pm 296/-227$ ppm CO₂.
 128 This estimate is significantly higher than that obtained from MORB (Table 1),
 129 a discrepancy suggesting the presence of high carbon regions in Earth’s deep
 130 mantle. However, bulk degassing cannot be used to uniquely assign carbon to
 131 the recycled (e.g., Sobolev et al., 2005) or primitive (e.g., Tieloff et al., 2000)
 132 components inferred for the Hawaiian source. Our approach minimises this
 133 ambiguity by focusing on melt inclusions from single eruptions.

134 *2.3. Bulk silicate Earth carbon estimates*

135 In principle, the size of the bulk silicate Earth (BSE) carbon reservoir could
 136 be reconstructed using carbon estimates of all mantle reservoirs, including the
 137 deep mantle. However, as emphasised above, a key piece of information is poorly
 138 known: the size of this deep mantle reservoir (Hofmann, 1997). A common
 139 approach to this problem is to assume, or infer from geochemical mass balance,
 140 the size of the deep mantle reservoir and then use carbon-gas or carbon-trace
 141 element ratios to extrapolate to bulk Earth.

142 A recent review paper from Halliday (2013) presented several models for es-
 143 timating bulk Earth carbon content. The ‘basalt’ model used the water content
 144 of MORB and OIB to infer a bulk water content for the entire mantle, and
 145 then took representative H/C ratios to calculate a bulk mantle carbon content
 146 (163 ppm CO₂). The ‘layered mantle’ model used Ar isotope budgets across
 147 all Earth’s reservoirs, along with volatile ratios, to derive a bulk Earth carbon
 148 budget of 2462 ppm CO₂. This estimate is very similar to the value from Marty
 149 (2012), who also used an ⁴⁰Ar budget to extrapolate observed C/N and C/⁴He
 150 ratios to the whole Earth (2831 ppm CO₂).

151 Carbon-ITE ratios in MORB, enriched-MORB, and OIB have also been used
 152 to reconstruct BSE carbon. Hirschmann (2018) presented a recent implementa-

tion of this approach, taking $\text{CO}_2/\text{Ba} = 100 \pm 20$ for the convecting mantle and a representative Ba concentration to estimate 514 ± 147 ppm CO_2 in BSE. This new estimate assumes that oceanic basalts, regardless of enrichment, preserve similar CO_2/Ba ratios, and that analysed oceanic basalts sample all significant geochemical reservoirs within the Earth. Our new results will show, in contrast, that there is evidence for both significant CO_2/Ba variability, and that there exist mantle reservoirs with far higher CO_2/Ba than have previously been identified in MORB or OIB.

Whilst the above BSE carbon estimates vary significantly, they have a common implication: assuming DMM carbon concentrations of 20–100 ppm CO_2 (e.g., Dasgupta and Hirschmann, 2010), a significantly more carbon-rich reservoir must exist in the deep Earth to balance high BSE carbon estimates. For example, Hirschmann (2016) calculated that this reservoir needs $\text{CO}_2/\text{Nb} = 5000 \pm 2000$ and $\text{CO}_2/\text{Ba} = 850 \pm 350$ to match the BSE carbon content calculated by Marty (2012).

2.4. Carbon in the Icelandic mantle

Previous studies of Icelandic geothermal fluids, glasses and vesicles have concluded that the Icelandic mantle source has little to no carbon enrichment with respect to DMM. While these studies used bulk analyses to assess average source carbon, our study uses olivine-hosted melt inclusions to access signals of source heterogeneity that are preserved in melts prior to mixing.

Only one melt inclusion suite from Iceland has previously been interrogated for mantle carbon signatures; the Borgarhraun eruption in the Northern Rift Zone (NRZ, Fig. 1a). Olivine-hosted melt inclusions from Borgarhraun record compositional heterogeneity (MacLennan et al., 2003), show evidence of concurrent crystallisation and melt mixing (MacLennan, 2008a), and preserve a CO_2 -ITE correlation (Hauri et al., 2018), which has been used as evidence for

an undegassed CO_2 signature ($\text{CO}_2/\text{Ba} = 48.3 \pm 2.7$, $\text{CO}_2/\text{Nb} = 391 \pm 16$; Table 1). The mantle carbon content inferred from these observations is similar to that of DMM, with CO_2/Nb comparable to undegassed MORB suites (Table 1), but with CO_2/Ba half the inferred global average (Hirschmann, 2018).

Borgarhraun melts are not strongly enriched in lithophile elements, being depleted relative to other Icelandic melts, and its He isotope signature ($R/R_a = 7.9$; Fig. 1b) lies within the MORB range (Füri et al., 2010). Hauri et al. (2018) suggested that the Borgarhraun mantle contains a source common to the Atlantic depleted mantle: hydrated and carbonated material originating from subduction-related modification. These features strongly suggest that there is no deep mantle component in the Borgarhraun source, which fits the regional systematics of ‘MORB-like’ He-Ne in NRZ eruptions in contrast to more ‘solar-like’ noble gas compositions of Western Rift Zone (WRZ) eruptions (e.g., Füri et al., 2010). Such observations align with Pb isotope constraints indicating distinct mantle source components across Iceland (Shorttle et al., 2013), and merit investigation of whether there are associated differences in mantle carbon abundances between the two rift zones. Our new observations from eruptions in the WRZ and central Iceland (Fig. 1a) directly address this question.

3. Samples and methods

3.1. Geological context

We present new data from two Icelandic eruptions, Miðfell (also known as Dagmálafell; $64^\circ 10.456' \text{ N}$, $021^\circ 02.859' \text{ W}$) and Kistufell ($64^\circ 47.442' \text{ N}$, $017^\circ 10.456' \text{ W}$; Fig. 1a). Miðfell is a ~ 300 m high mountain in Iceland’s WRZ, lying on the eastern edge of Þingvallavatn and within the Hengill volcanic system. It is a northeast-southwest striking ridge, which is thought to have erupted beneath ice during the last glacial period (Gurenko and Chaussidon, 1995, and

206 references therein). The mountain’s lower flanks comprise vesicular, olivine-
 207 phyric, pillow basalts with glassy rinds, whereas the upper portions are com-
 208 posed of hyaloclastite. Kistufell is a table mountain located in central Iceland,
 209 at the northern edge of Vatnajökull. Kistufell may have erupted towards the
 210 end of the last glacial period as the ice-sheet retreated (Breddam, 2002). Glassy
 211 pillows are exposed at the northern flanks of the mountain.

212 *3.2. Geochemical context*

213 Previous Miðfell melt inclusion studies have measured significant trace ele-
 214 ment variability, including high-degree melts with diluted ITE concentrations
 215 (Gurenko and Chaussidon, 1995). If carbon remained coupled to these ITE’s,
 216 then it is likely that some Miðfell melt inclusions have avoided degassing, provid-
 217 ing an opportunity to recover mantle carbon content. Both Miðfell and Kistufell
 218 have noble gas isotope data that support the presence of primordial material
 219 in their source regions: $R/R_a \sim 17$ (Breddam et al., 2000; Füre et al., 2010)
 220 and, where heavy noble gases have been measured in Miðfell, primordial Xe
 221 and Ne isotopic ratios (Mukhopadhyay, 2012). Miðfell olivines and matrix glass
 222 also show noble gas evidence for other components, including recycled air (e.g.,
 223 Harrison et al., 1999). Combined, these observations make Miðfell and Kistufell
 224 good targets for identifying deep mantle carbon.

225 *3.3. Analytical methods*

226 Olivine phenocrysts 0.5–4 mm in size were picked from hand-crushed pillow
 227 glasses. Care was taken to pick unaltered olivine macrocrysts containing glassy
 228 melt inclusions without post-entrapment crystals. Olivines were individually
 229 mounted and polished to expose melt inclusions before being re-mounted and
 230 polished for analysis. Melt inclusions were analysed for trace elements, CO_2 ,
 231 and H_2O by secondary ion mass spectroscopy at the Edinburgh ion microprobe

232 facility. Major elements were measured by electron probe microanalysis. A
 233 subset of the Miðfell melt inclusion suite was processed at Woods Hole Oceano-
 234 graphic Institution to reconstruct their initial CO₂ content from vapour bubble
 235 and inclusion glass CO₂ concentrations. Melt inclusion and vapour bubble vol-
 236 umes were determined by X-ray tomography. Vapour bubble CO₂ content was
 237 determined using confocal Raman spectroscopy. For full method details and
 238 representative melt inclusion micrographs, see Supplementary Material.

239 4. Results

240 4.1. Major elements

241 Miðfell olivine phenocryst compositions range from Fo_{90.9} to Fo_{85.2} (Fig. 2),
 242 the matrix glass and olivine-hosted melt inclusion compositions are basaltic
 243 (SiO₂ ~48 wt%), giving a picritic whole rock composition (Gurenko and Chaus-
 244 sidon, 1995). Kistufell olivines range from Fo_{89.7} to Fo_{87.9} with basaltic melt
 245 inclusions (SiO₂ ~48 wt%).

246 Melt inclusion major element compositions have been corrected for post-
 247 entrapment crystallisation by iteratively adding an equilibrium olivine composi-
 248 tion, assuming an Fe-Mg K_D = 0.34 (Matzen et al., 2011), until the melt inclu-
 249 sion reached equilibrium with its host (Danyushevsky et al., 2000). For Miðfell,
 250 this correction takes into account the ferric iron content of each melt inclusion
 251 (see Supplementary Material), whereas for Kistufell we assumed a constant fer-
 252 ric to total iron ratio of 0.07 based on matrix glass measurements (Breddam,
 253 2002). The majority of melt inclusions required < 5% olivine addition to bring
 254 them back into Fe-Mg equilibrium with their olivine host.

255 4.2. Trace elements

256 Miðfell melt inclusions record substantially more variability than those from
 257 Kistufell (Fig. 2 & 3). The method of MacLennan et al. (2003) was used to

258 establish that the Miðfell analyses for all major, trace, and volatile elements
 259 (apart from H₂O, Yb and Lu) have a signal-to-noise ratio > 1 , and $> 99\%$
 260 confidence that natural variability can be resolved from analytical noise. These
 261 inclusions preserve trace element variability comparable with that found by the
 262 combination of all previously published undegassed melt inclusion and MORB
 263 glass suites (e.g., Saal et al., 2002). Kistufell melt inclusions recover a signal-to-
 264 noise ratio > 1 for the majority of major, trace, and volatile elements. However,
 265 the Kistufell melt inclusion suite has much lower melt heterogeneity than Miðfell,
 266 as indicated by their contrasting signal-to-noise ratios (e.g., Miðfell $\sigma_t/\sigma_r|_{\text{Ba}} =$
 267 86.4 ; Kistufell $\sigma_t/\sigma_r|_{\text{Ba}} = 3.06$).

268 Multi-element plots (Fig. 3) illustrate Miðfell and Kistufell trace element
 269 variability compared to that of the Borgarhraun eruption (Hauri et al., 2018).
 270 As is typical for many mantle-derived basalts, the most incompatible trace ele-
 271 ments show the largest concentration range. In Miðfell, Ba content ranges from
 272 $0.37\text{--}115$ ppm ($29.8\text{--}3.5\%$ relative error; matrix glass 6.3 ppm), and Nb content
 273 ranges from $0.04\text{--}22.9$ ppm ($17.6\text{--}9.1\%$ relative error; matrix glass 0.78 ppm).
 274 Vapour bubbles only occur in Miðfell melt inclusions with high trace element
 275 concentrations (> 12.4 ppm Ba and > 2.47 ppm Nb). Kistufell melt inclusions
 276 rarely contain vapour bubbles, so no inclusions with vapour bubbles were mea-
 277 sured for this data set. Trace element patterns in all three eruptions show that
 278 some of the melt inclusions preserve positive Sr and negative Zr anomalies, con-
 279 sistent with previous Miðfell melt inclusion analyses (Gurenko and Chaussidon,
 280 1995).

281 4.3. Volatile elements

282 Kistufell melt inclusions have a water content that matches that of the matrix
 283 glass surrounding host olivines, but the sample population shows some variabil-
 284 ity ($0.10\text{--}0.19$ wt%). Water content in the Miðfell melt inclusions and matrix

285 glass is almost constant at ~ 0.06 wt%, which is low compared to other Icelandic
 286 melt inclusion suites (e.g., Laki, ~ 0.65 wt% H_2O , Hartley et al., 2015). Melt
 287 inclusion CO_2 content varies from 20–1120 ppm (Fig. 4a). At low trace element
 288 concentrations ($\text{Ba} < 3.0$ ppm, $\text{Nb} < 0.2$ ppm) the maximum melt inclusion
 289 CO_2 content is controlled by CO_2/ITE ratios. At higher trace element concen-
 290 trations, maximum CO_2 content reaches a plateau at 1120 ppm. Melt inclusions
 291 containing vapour bubbles record a glass CO_2 content of 720–1170 ppm, con-
 292 sistent with the vapour bubble-absent inclusions, and a total (reconstructed)
 293 CO_2 content of 1340–4550 ppm (Fig. 4a). Matrix glass has a CO_2 content of
 294 < 200 ppm.

295 4.4. CO_2/ITE ratios

296 Kistufell melt inclusions record maximum $\text{CO}_2/\text{Nb} = 391 \pm 70$ and CO_2/Ba
 297 $= 71.9 \pm 13.9$, which are associated with the lowest ITE concentration melts.
 298 Miðfell melt inclusions preserve maximum $\text{CO}_2/\text{Nb} = 5737 \pm 987$ and CO_2/Ba
 299 $= 566 \pm 68$, also associated with low trace element concentrations, i.e., lower
 300 than the matrix glass (Fig. 5), and hosted in the most forsterite-rich olivines
 301 (Fig. 2b). Melt inclusions with CO_2 content reconstructed from vapour bub-
 302 ble concentrations also have high CO_2/ITE ratios, with maximum $\text{CO}_2/\text{Nb} =$
 303 1186 ± 127 and $\text{CO}_2/\text{Ba} = 236 \pm 25$. Comparison to previously analysed melt
 304 inclusion and MORB glass suites shows that the Miðfell melt inclusions pre-
 305 serve some of the highest CO_2/ITE ratios ever measured in natural basaltic
 306 glass (e.g., Le Voyer et al., 2017). Importantly, Miðfell inclusions exhibit these
 307 higher CO_2/ITE ratios at the same ITE concentrations as in previously stud-
 308 ied suites, indicating that their relative carbon enrichment is not a feature of
 309 anomalous trace element depletion or enrichment (Fig. 5).

310 5. Does Miðfell preserve a mantle carbon signature?

311 The previous section demonstrated that whilst Miðfell melt inclusions pre-
312 serve significant trace element variability, Kistufell contains a more homoge-
313 neous melt inclusion population. On this basis, the Kistufell melt inclusions do
314 not enable empirical validation of whether they have degassed carbon (Fig. 4b;
315 Matthews et al., 2017). However, Kistufell melt inclusions have ITE concentra-
316 tions that in the Miðfell melt inclusion population are associated with carbon
317 loss (Fig. 4). For Miðfell, this carbon loss is shown by systematically decreasing
318 CO_2/ITE ratios with increasing ITE concentration (Fig. 5c,d). Kistufell inclu-
319 sions have likely degassed, therefore we focus our investigation of deep mantle
320 carbon onto the Miðfell inclusions, where we can be sure of recovering unde-
321 gassed systematics.

322 Trace element and CO_2 data from Miðfell melt inclusions suggest that some
323 process has enriched these basaltic glasses in carbon, relative to ITE's, compared
324 with other inclusion and glass suites (e.g., Michael and Graham, 2015). To use
325 this observation to constrain the carbon content of the Miðfell mantle source
326 first requires an assessment of the influence of crustal processes, which could
327 have affected the carbon and ITE content of Miðfell melts.

328 5.1. Degassing and olivine decrepitation

329 The strongest signal in the Miðfell CO_2 data, seen most clearly at trace el-
330 ement concentrations greater than that of the matrix glass, is one of degassing
331 (Fig. 4a). At these high ITE concentrations, melt inclusions with variable trace
332 element concentrations have constant CO_2 content, indicating the partial loss
333 of initial CO_2 , and therefore loss of information on carbon from deeper in the
334 system. If degassing occurred in some of the Miðfell melts prior to entrap-
335 ment, then solubility models suggest entrapment pressures of ~ 1.8 kbar (taking
336 1170 ppm CO_2 ; Shishkina et al., 2010).

337 However, the presence of vapour bubbles in some of the trace element en-
 338 riched melt inclusions suggests that some Miðfell melts were initially trapped
 339 with higher dissolved CO₂ concentrations and underwent exsolution post-entrapment.
 340 Reconstructed CO₂ concentrations from these inclusions are consistent with sat-
 341 uration pressures of up to ~4.5 kbar (~13 km depth). If all melts started with
 342 the same CO₂/ITE ratio, then those CO₂-saturated melt inclusions lacking
 343 vapour bubbles must have lost exsolved CO₂ during ascent to eruption, perhaps
 344 by decrepitation (MacLennan, 2017).

345 ITE-rich Miðfell melts appear to have been degassed to their saturation
 346 concentration, but at lower ITE concentrations melts show an ITE-CO₂ cor-
 347 relation. This observation suggests that ITE-depleted melt inclusions preserve
 348 an undegassed CO₂ signal (Matthews et al., 2017); an inference supported by
 349 bubble-reconstructed melt inclusions that have CO₂/ITE ratios in broad agree-
 350 ment with those of depleted inclusions (Fig. 4 & 5). Therefore, relative carbon
 351 enrichment in Miðfell was likely not restricted to the most ITE-depleted melts,
 352 but rather a characteristic of all ITE concentrations (Fig. 4a & 5). With this
 353 result, the key question is what source or process led to high CO₂/ITE ratios
 354 in Miðfell inclusions?

355 5.2. Crustal melt modification

356 Gurenko and Sobolev (2006) analysed olivine-hosted melt inclusions and gab-
 357 broic xenoliths from Miðfell to infer that the trace element chemistry of Miðfell
 358 melt inclusions had been modified by interaction with lower crustal gabbro. To
 359 attribute a carbon enrichment signature to the mantle, we must identify melt
 360 inclusions that have avoided CO₂/ITE ratio modification by crustal interaction.

361 We have modelled the mixing of a depleted Miðfell melt composition, which
 362 has a smooth trace element pattern, with crustal components. We discount
 363 a role for significant plagioclase addition to Miðfell melts (Fig. S5). However,

364 mixing with a 10% fractional melt derived from Miðfell gabbro best matches the
 365 shape of positive Sr, negative Zr, and Ba > Nb anomalies observed within some
 366 of the Miðfell melt inclusion population (Fig. 3 & S6), providing good evidence
 367 for interaction between primitive Miðfell melts and gabbro in the crust.

368 Extending our analysis to assess the impact of crustal interaction on CO₂/ITE
 369 ratios; an important observation is that carbonate phases have not been seen
 370 in Miðfell gabbroic xenoliths (Gurenko and Sobolev, 2006), though carbonate
 371 breakdown on decompression during eruption may remove visible evidence of
 372 gabbroic carbon (e.g., Canil, 1990). Therefore, the addition of carbon to Miðfell
 373 melts cannot be ruled out from textural observations, but geochemical obser-
 374 vations can be used to avoid melt inclusions that have strongly interacted with
 375 gabbroic material, i.e., those having large trace element anomalies.

376 To avoid the effects of any potential CO₂-ITE modification, in the subsequent
 377 analysis we have only used melt inclusions with smooth trace element patterns.
 378 We believe that these melt inclusions have had minimal interaction with crustal
 379 gabbro during ascent through the Miðfell magmatic system.

380 Melt inclusions with smooth trace element patterns record maximum CO₂/Nb
 381 = 1832 ± 316 and CO₂/Ba = 396 ± 48 (Fig. 5c,d), again showing higher ra-
 382 tios than observed in MORB suites, and suggesting that the relative carbon
 383 enrichment in Miðfell melts likely originates in its source mantle.

384 6. Estimating the carbon content of the Miðfell mantle source

385 The key question we address in this section is whether the observed high
 386 CO₂/Ba and CO₂/Nb ratios in Miðfell inclusions require an anomalously carbon-
 387 rich mantle source. First we identify which ITE's have stayed coupled to carbon
 388 during the melting process, and hence which CO₂/ITE ratios are unfractionated
 389 from their source values. We then review the mantle components contributing

390 to Miðfell and calculate in what relative proportions they are represented in its
 391 chemistry. Finally, we use constraints on ITE concentrations in these sources to
 392 convert the observed CO_2 and ITE systematics of Miðfell inclusions into source
 393 CO_2 concentrations.

394 *6.1. Carbon-ITE coupling in Miðfell*

395 The silicate-melt partition coefficient for carbon, as estimated by carbonated-
 396 lherzolite melting experiments, places it between Ba and Nb in terms of com-
 397 patibility (Rosenthal et al., 2015). Therefore, we would expect CO_2/Ba and
 398 CO_2/Nb in Miðfell to be similar to that of its mantle source, provided it repre-
 399 sents an aggregation of melts from across the melting region (e.g., Fig. 6 solid
 400 line). If instead, Miðfell was produced from high-degree melts from the shallow
 401 part of the melting region, a region already extensively depleted by prior melt-
 402 ing, then even highly incompatible elements may have been fractionated from
 403 each other, so CO_2/Ba and CO_2/Nb may not be faithful to the source value
 404 (e.g., Fig. 6 dashed and dashed-dotted lines).

405 We have two tests for whether CO_2/ITE ratios have been fractionated during
 406 melting. The first uses the fact that carbon partitioning between silicate and
 407 melt lies between that of Ba and Nb (Rosenthal et al., 2015). Therefore, if
 408 Miðfell represents preferential sampling of high-degree shallow melts, we might
 409 expect to observe anomalously high CO_2/Ba . However, this observation would
 410 be associated with correspondingly low CO_2/Nb . Instead, Miðfell inclusions
 411 show correlated high CO_2/Nb and CO_2/Ba values (Fig. 6), indicating that no
 412 such fractionation has occurred.

413 Secondly, fractionation between trace elements during fractional melting will
 414 be manifest in the elements' relative variability (their standard deviation divided
 415 by their mean concentration). One prediction of fractional melting models is
 416 that trace element variability should increase with decreasing partition coeffi-

417 cient (red line, Fig. 7; Schiano et al., 1993). Figure 7 shows that in Miðfell
 418 inclusions, the most incompatible elements measured (Ba, K, Nb, La) have con-
 419 stant relative variability, which is consistent with a residual porosity during
 420 melting having damped the variability generated by low degree melting. As
 421 these low degree melts will have contained almost all the Ba, K, Nb, and La
 422 that was in the source, these elements will not have been fractionated from
 423 each other during the melting event that produced Miðfell. As carbon’s parti-
 424 tion coefficient lies within the range of Ba to La, we can infer from Fig. 7 that
 425 carbon will not have been fractionated from any elements across this range of
 426 partition coefficients. Therefore, the uncommonly high CO_2/Ba and CO_2/Nb
 427 ratios observed in undegassed Miðfell melts (Fig. 5) reflect the composition of
 428 their mantle sources.

429 6.2. Components in the Miðfell mantle source

430 *Miðfell’s depleted mantle component:* the Borgarhraun eruption provides
 431 an on-Iceland sample of the local depleted mantle component in the plume
 432 (Thirlwall et al., 2004; Stracke et al., 2003), one that has been suggested to be
 433 ubiquitous in the Atlantic mantle (Hauri et al., 2018). Similarities in lithophile
 434 elements (Fig. 3), including Pb isotopes (Kokfelt et al., 2006; Halldórsson et al.,
 435 2016a), between Miðfell and Borgarhraun, suggest that DMM-like Borgarhraun
 436 source material is also present as a component within the Miðfell mantle source.

437 *Is there a pyroxenitic component in Miðfell?:* The variability in trace element
 438 enrichment and major element chemistry of primitive melts from across Iceland
 439 has been interpreted to reflect recycled pyroxenitic components in the Icelandic
 440 mantle (Sobolev et al., 2008; Peate et al., 2010; Shorttle and MacLennan, 2011).
 441 In the case of Miðfell though, trace element ratios (e.g., $\text{Nb}/\text{Zr} \sim 0.05$), major
 442 elements ($\text{FeO} \sim 9.4 \text{ wt\%}$), and Cl isotopes (Halldórsson et al., 2016a) suggest
 443 a minimal contribution from pyroxenite material to this eruption.

444 This is not to say that Miðfell has had no recycled contribution to its com-
 445 position. It contains a typical DMM component, which has itself been argued
 446 to contain recycled signatures, both as a regional contamination of the upper
 447 mantle (Hauri et al., 2018) and as a global phenomenon (Andersen et al., 2015).
 448 However, as Borgarhraun does not show significant carbon enrichment with re-
 449 spect to MORB and significantly less relative carbon enrichment than Miðfell,
 450 there is no evidence that the Icelandic mantle has been enriched in carbon from
 451 pyroxenite addition.

452 *Evidence for a deep mantle component:* the Miðfell source incorporates ma-
 453 terial distinct from that of the MORB source, having a more solar- or carbona-
 454 ceous chondrite-like composition that is often ascribed to an ancient primordial
 455 reservoir. This signal is most distinct in the heavy noble gas isotopes (e.g.,
 456 Harrison et al., 1999; Trieloff and Kunz, 2005; Mukhopadhyay, 2012; Caracausi
 457 et al., 2016).

458 However, the heavy noble gases also indicate that the Miðfell source has had
 459 atmosphere recycled into it (e.g., Harrison et al., 1999). This conclusion is in
 460 general consistent with constraints from other isotopic systems, e.g., N and Os,
 461 which suggest that ancient (> 1.5 Ga) recycled crust is present within the deep
 462 mantle component of the Icelandic mantle plume as crustal isotopic signatures
 463 are coupled with high- $^3\text{He}/^4\text{He}$ values (Brandon et al., 2007; Halldórsson et al.,
 464 2016b). Therefore, a deep mantle component could contain carbon of both
 465 primordial and recycled origins.

466 The above observations suggest that the Miðfell source mantle comprises a
 467 mixture of a depleted component (DM), akin to Borgarhraun source mantle, and
 468 a deep mantle component (Deep). Both the Deep and DM components within
 469 the Miðfell source can have plausible estimates made of their ITE content, which
 470 will be important for estimating source CO_2 concentrations. In subsequent mod-

471 elling we adopt the BSE composition given by McDonough and Sun (1995) for
 472 the Deep component, though we note that its ITE composition could be more
 473 enriched due to recycled material. As there are no constraints on the nature
 474 of this enrichment, we do not propagate this uncertainty, although we empha-
 475 sise that by assuming a BSE source rather than BSE + recycled source, our
 476 source carbon estimates will be a lower bound. We assign the DM component
 477 the Borgarhraun source mantle ITE composition (Hauri et al., 2018). As the
 478 CO₂ of DM has been estimated, only Deep CO₂ and the DM:Deep proportions
 479 are unknowns in estimating source CO₂. Neither choice of source composition
 480 is critical to the results we subsequently obtain: similar source carbon concen-
 481 trations are inferred if we take, for example, ITE abundances from Palme and
 482 O'Neill (2014) for the Deep estimate and Workman and Hart (2005) for the
 483 DM.

484 *6.3. Proportions of depleted and deep components in the Miðfell mantle source*

485 If Ba, K, Nb, La, and carbon have not been fractionated from each other
 486 during melting, then their proportions in Miðfell reflect the degree of melting,
 487 F ; melt mixing; and melt transport: partial melting will have enriched these
 488 ITE's in the melt compared to their source concentrations. Using estimates
 489 of Ba, K, Nb, and La content of both DM and Deep components constituting
 490 the Miðfell source, we can calculate how much the melting process has enriched
 491 Miðfell compared to its mantle source, and thereby calculate the source carbon
 492 content. We define the enrichment factor for an element, i , as

$$E_i = \frac{C_i^l}{C_i^0}, \quad (1)$$

493 where C_i^l is the concentration of the element in the liquid (i.e., observed Miðfell)
 494 and C_i^0 is the concentration of the element in the source (DM and Deep taken
 495 from Hauri et al., 2018 and McDonough and Sun, 1995, respectively). For Ba,

496 K, Nb, and La we can calculate E_i directly, assuming a DM:Deep ratio. For
 497 carbon, we take the mean enrichment factor estimated from the lithophile trace
 498 elements, and use it in (1) with the observed carbon content of Miðfell inclusions
 499 to estimate C_C^0 . The main question is what mass fraction of the Miðfell source
 500 is the Deep component? The enrichment factor contains information on this
 501 fraction: given that Ba, K, Nb, and La have not been fractionated during
 502 melting, and if we have identified the correct Deep and DM source compositions,
 503 then $E_{\text{Ba}} = E_{\text{K}} = E_{\text{Nb}} = E_{\text{La}}$. Therefore, by sweeping through Deep component
 504 fractions from 0 to 1 and identifying the minimum amount of variability in E_i ,
 505 we can identify the optimal source mixture.

506 In Fig. 8a we show the mean enrichment factor, \bar{E} , calculated from Ba,
 507 K, Nb, and La content of the Miðfell matrix glass (i.e., the composition of
 508 the aggregated melt) as a function of Deep fraction in the source. Figure 8
 509 shows that \bar{E} lies between 1 and 3 in these models. These small enrichment
 510 factors likely reflect both the high peak mantle melt extent under Iceland ($\sim 30\%$;
 511 MacLennan et al., 2001) and the incomplete mixing of mantle melts leading to
 512 a bias towards relatively shallow fractional melts from the full melting column
 513 in the mean Miðfell composition.

514 When the source is mostly DM (i.e., low X_{Deep}) the C_i^0 values are low, and
 515 the E_i 's are correspondingly high. The variability in the calculated enrichment
 516 factors is minimised at $X_{\text{Deep}} = 0.47$ (Fig. 8b), i.e., a nearly 50:50 mixture of
 517 DM and Deep components in the Miðfell source, which corresponds to $\bar{E} = 1.56$.

518 6.4. Translating enrichment factors to source CO_2

519 The final step in using the calculated enrichment factors to estimate bulk
 520 Miðfell source carbon is to choose a carbon content for the Miðfell magma. For
 521 this calculation we take the CO_2 content of the melt inclusions most chemically
 522 similar to the matrix glass, which have $\text{CO}_2 = 1079$ ppm: these inclusions likely

523 trap the mixed magma prior to its extensive low pressure degassing, albeit they
 524 do not have the highest CO₂/ITE ratios we observe and therefore may have
 525 already lost some CO₂. Most importantly, this choice is consistent with our
 526 use of the matrix glass trace element composition to calculate the enrichment
 527 factors. Our calculation is also independent of chosen DM CO₂ value, as it is
 528 determined by erupted CO₂ content and \bar{E} values.

529 Taking this value of CO₂, Fig. 8c reports Miðfell source carbon content
 530 calculated using enrichment factors over a range of source Deep fractions (blue
 531 line). For $X_{\text{Deep}} = 0.47$, the bulk Miðfell source contains 690 ppm CO₂ (Fig. 8c
 532 green bar).

533 Mass balance between the DM and Deep components allows us to convert
 534 the bulk source CO₂ (blue) into Deep CO₂ (orange). Here our calculation does
 535 depend on knowing the DM CO₂ content, which has been previously constrained
 536 as 105 ± 57 ppm (black; Hauri et al., 2018). For $X_{\text{Deep}} = 0.47$, Deep has a CO₂
 537 content of 1350 ± 350 ppm (90% confidence interval considering only propagated
 538 analytical and \bar{E} uncertainties).

539 These calculations are robust for a number of different assumptions: (i) as
 540 Fig. 8c shows, the full range of source DM:Deep proportions predict a Deep
 541 CO₂ content > 1000 ppm; (ii) if the modelling is repeated using more depleted
 542 melt inclusions, which are less likely to have undergone degassing and have peak
 543 CO₂ of 300 ppm, then Deep compositions of > 1250 ppm CO₂ are inferred; and
 544 (iii) adaptations to the model to explore the consequences of mixing fractional
 545 melts from two sources produce the same requirement of > 1000 ppm CO₂ in
 546 the Deep source.

547 7. Summary & Discussion

548 We have shown that melt inclusions from the Icelandic Miðfell eruption
549 record some of the highest CO_2/ITE ratios reported in basalts, and that these
550 inclusions reflect the composition of their mantle source. In comparison with
551 Borgarhraun, Miðfell’s noble gas and lithophile element composition argues for
552 its source comprising a deep mantle component in addition to the depleted
553 mantle common to Icelandic magmas. By combining existing work with our new
554 carbon and trace element observations, we have been able to assign a carbon
555 content to this deep component. We took the depleted mantle component to
556 be that of the well-studied Borgarhraun eruption, which lacks primitive noble
557 gas isotope signatures and has a source $\text{CO}_2 = 105 \pm 57$ ppm (using Ba, Nb
558 estimates; Hauri et al., 2018). We estimate the deep mantle component to have
559 a CO_2 concentration of at least 1000 ppm (preferred value 1350 ± 350 ppm),
560 which increases when the proportion of deep mantle component assumed to be
561 present in the source is decreased.

562 7.1. Recycled carbon?

563 Xenon isotopes indicate that a significant proportion of Xe in Miðfell is de-
564 rived from recycled air ($\sim 90\%$; Mukhopadhyay, 2012). If carbon was coupled
565 to Xe during recycling, then some amount of carbon in the Miðfell source could
566 also have come from recycling. However, it would be surprising if this compo-
567 nent was the origin of the high CO_2/ITE ratios we observe. Miðfell lavas are
568 not unique in containing recycled air: MORB have been interpreted to have a
569 similar abundance of recycled air in their DMM source (Mukhopadhyay, 2012;
570 Parai and Mukhopadhyay, 2015). Moreover, DMM, in contrast to the Icelandic
571 mantle, is also known from Pb isotopes, U isotopes, and CO_2/Ba ratios to
572 have been pervasively contaminated by surface material (Andersen et al., 2015;
573 Hirschmann, 2018). Despite this, MORB have lower CO_2/ITE and inferred

574 source CO₂ concentrations than Miðfell (e.g., Le Voyer et al., 2017). Therefore
575 recycling, at least as seen by MORB, does not appear to significantly enrich
576 carbon with respect to ITE’s.

577 7.2. Ancient mantle carbon?

578 If Miðfell’s carbon enrichment is a signature of entrained deep mantle ma-
579 terial, to what extent is it ancient mantle carbon? This question is difficult to
580 answer and will require more eruptions to be identified with primitive noble gas
581 signatures and carbon undersaturated melt inclusions. However, the two erup-
582 tions we now have on Iceland with CO₂/ITE ratios that can be linked to their
583 mantle sources, differ almost exclusively in their geochemistry by the presence
584 of primitive noble gas signatures in Miðfell, which are absent in Borgarhraun
585 — their lithophile radiogenic isotope compositions are otherwise very similar
586 (Fig. 1b,c). This signature suggests an association between the primitive noble
587 gases, which have been attributed to solar nebula ingassing during Earth’s ear-
588 liest history (Mukhopadhyay, 2012; Williams and Mukhopadhyay, 2018), and
589 carbon. If true, it would imply a significant fraction of Earth’s carbon came
590 directly from the solar nebula rather than later accretion.

591 However, the coupling of high-³He/⁴He with recycled material signatures
592 (Brandon et al., 2007; Halldórsson et al., 2016b) could also suggest that deep
593 mantle carbon is of a recycled origin, making it difficult to differentiate between
594 primordial and recycled carbon from deep mantle material. Especially as there is
595 no certainty that the nature of deep recycled material is the same as the upper
596 mantle recycled material assumed to be present in the Borgarhraun source,
597 which evidences no enrichment in carbon (Hauri et al., 2018).

598 7.3. Size of the PM carbon reservoir

599 If we assume that the Deep carbon estimate derived from Miðfell is repre-
600 sentative of the global deep mantle reservoir, then we can calculate a revised

mantle carbon budget. Estimates for the deep mantle fraction of the total mantle range from the seismically-defined 660 km transition zone, down to the seismically-defined D'' layer above the core-mantle boundary. Figure 9 presents carbon content estimates of four potential deep mantle reservoir fractions: (i) 5% representing the D'' layer (blue; Tolstikhin and Hofmann, 2005); (ii) 20% for the abyssal layer of the lower mantle, defined by U/K mass balance (green; Arevalo et al., 2009); (iii) 42% for the mass of primitive mantle calculated by ^{40}Ar mass balance (orange; 90% confidence envelope in grey; this study and Alègre et al., 1996); and (iv) 75% for the lower mantle as defined by the 660 km seismic discontinuity (red; Hofmann, 1997).

The carbon content of the atmosphere, crust, and oceans combined (ACO) is $\sim 3.1 \times 10^{23}$ g CO_2 (Hirschmann, 2018). Carbon stored in the deep mantle, depending on mantle fraction, is therefore up to 14 times greater than the ACO carbon reservoir, while the DMM is approximately equal to the ACO (Fig. 9). Our calculations do not include the potential carbon contribution from the lithospheric mantle, which could be host to one AOC of carbon (Sleep, 2009; Kelemen and Manning, 2015). These carbon reservoir estimates for the deep Earth are necessarily speculative, but comparable to the range of previous bulk mantle estimates (e.g., Halliday, 2013). Our carbon estimates are also a lower bound in two important respects: (i) if the deep mantle component in Miðfell is less than the high value we used, then the implied carbon content in it is higher; (ii) if the deep component contains recycled material that is enriched in Ba and Nb, then our use of a BSE composition will have led to an underestimate of its carbon content.

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637 financial interests.

Setting Locality Reference	Ocean Island		Mid Ocean Ridge			14°N MAR Cartigny et al. (2008)	Global MORB Michael & Graham (2015)
	Midfell This study	Bulk	Borgarhraun Hauri et al. (2018)	Anderson & Poland (2017)	Hawaii Le Voyer et al. (2017)		
CO ₂ /Nb	1832	> 1520	391		758	534	607
CO ₂ /Ba	396	> 150	48.3		86	106	105
Mantle CO ₂ (Nb)	887	-	129		-	393	90
Mantle CO ₂ (Ba)	1580	-	80		-	427	59
Mantle CO ₂ (other)	> 690	> 1000	-		263	-	-
Nb ppm	0.484	0.658	0.330		0.347	0.100	0.149
Ba ppm	3.99	6.60	1.66		3.06	0.736	0.562
						4.04	

Table 1: Estimates of mantle carbon reservoirs. This study presents carbon estimates for Midfell source mantle and a deep mantle reservoir, calculated from olivine-hosted melt inclusion compositions and a DM-Deep mixing model. CO₂/ITE ratios for each data set are as published in their original publications. Source trace element and CO₂ estimates are the values presented by Hauri et al. (2018), who used the method of Shimizu et al. (2016) to calculate source trace element concentrations.

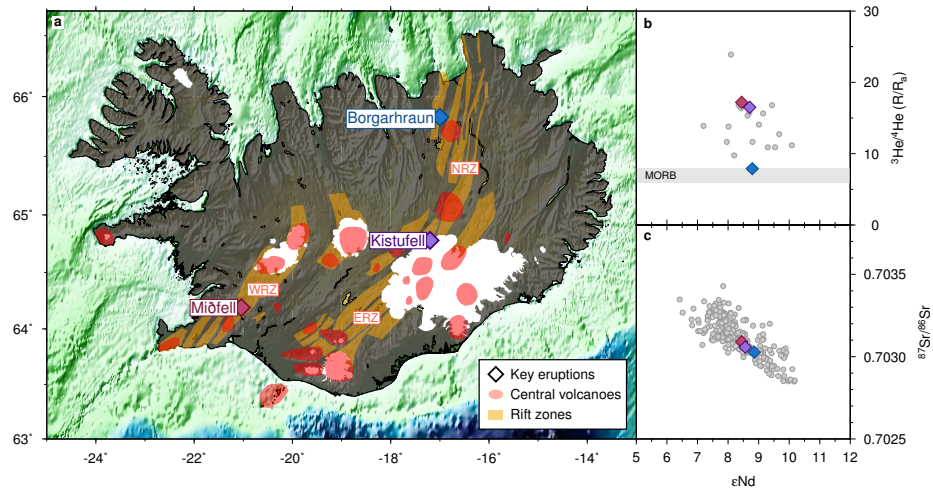


Figure 1: Icelandic eruptions discussed in this study: Miðfell (red), Kistufell (purple), and Borgarhraun (blue). Map of Iceland in (a) shows eruption locations, central volcanoes (red areas), and main rift zones (orange): WRZ, western rift zone; ERZ, eastern rift zone; and NRZ, northern rift zone. In (b) He isotope compositions relative to atmospheric $^3\text{He}/^4\text{He}$ (R/R_a) and in (c) Sr isotope compositions against ϵNd for key eruptions (diamonds) and young Icelandic eruptions (< 120 kyrs) from rift zones (grey). Kistufell and Miðfell show elevated $^3\text{He}/^4\text{He}$ with respect to Borgarhraun. Data compiled in Shorttle and MacLennan (2011).

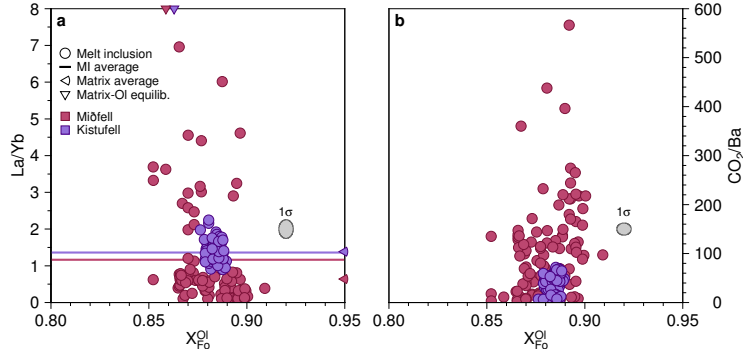


Figure 2: (a) La/Yb ratio of Miðfell (red) and Kistufell (purple) melt inclusions against host olivine forsterite content. Average melt inclusion La/Yb ratios are shown by horizontal lines, while matrix glass averages are indicated by arrowheads at $X_{\text{Fo}}^{\text{Ol}} = 0.95$. The olivine forsterite compositions in equilibrium with the matrix glasses are shown by inverted triangles at La/Yb = 8. Melt inclusion variability is preserved with decreasing forsterite content in the Miðfell suite, while the Kistufell melt inclusions have much lower La/Yb variability and are clustered within a smaller forsterite content range. (b) CO_2/Ba against forsterite, indicating that the highest CO_2/Ba ratios are preserved in the most forsteritic olivines. One sigma error ellipse plotted in grey.

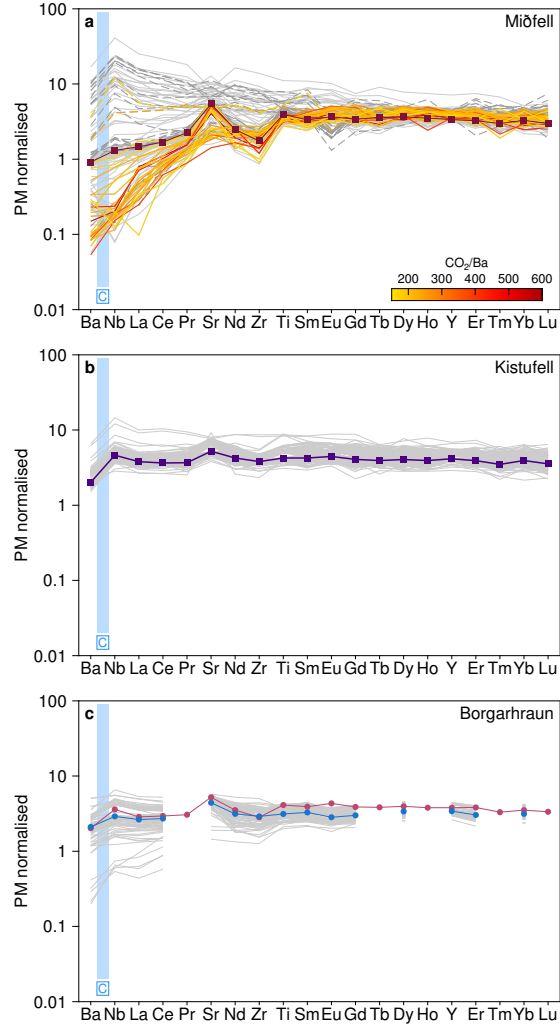


Figure 3: Trace element spider diagrams showing compositional variability in (a) Miðfell, (b) Kistufell, and (c) Borgarhraun (Hauri et al., 2018) melt inclusion suites, normalised to primitive mantle (PM; BSE composition, McDonough and Sun, 1995). Inclusions with $\text{CO}_2/\text{Ba} > 150$ are coloured, the rest are grey (Borgarhraun melt inclusions have an average $\text{CO}_2/\text{Ba} \sim 48$). Matrix glass compositions are shown as dark squares. The likely position of carbon, given its measured compatibility (Rosenthal et al., 2015), is shown as a blue vertical line. In (c) the light red line and circles are the Miðfell average melt inclusion composition, and in light blue is the Borgarhraun melt inclusion average. Miðfell and Borgarhraun matrix glasses are very similar, but Miðfell shows greater trace element variability within the melt inclusion population.

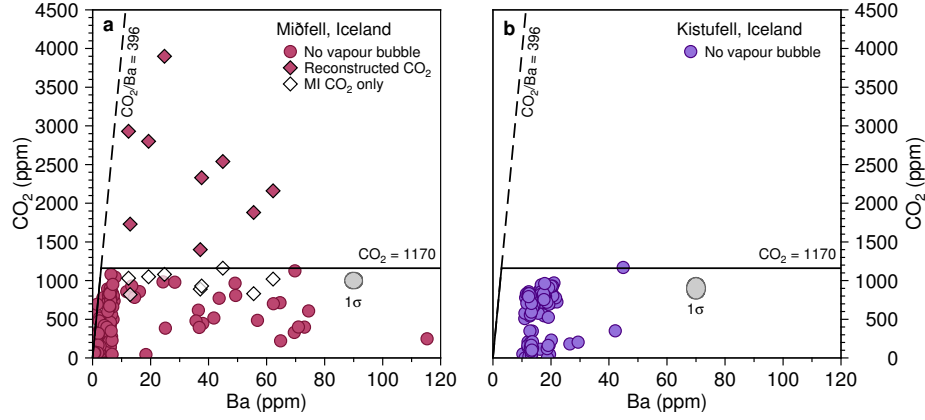


Figure 4: Variation in CO₂ content as a function of Ba concentration for (a) Miðfell and (b) Kistufell melt inclusions (diamonds, with vapour bubbles; circles, without vapour bubbles). The maximum CO₂ recovered from melt inclusion glass (circles and unfilled diamonds) is 1170 ppm, as shown by the horizontal line. Combined glass and vapour bubble CO₂ measurements plot between this line and CO₂/Ba = 396 (the maximum ratio preserved by a smooth trace element pattern melt inclusion), as shown by the dashed line. Miðfell contains melt inclusions with low enough Ba concentrations that they have likely avoided degassing, while Kistufell melt inclusions are too enriched and have all likely lost CO₂. One sigma error ellipse plotted in grey.

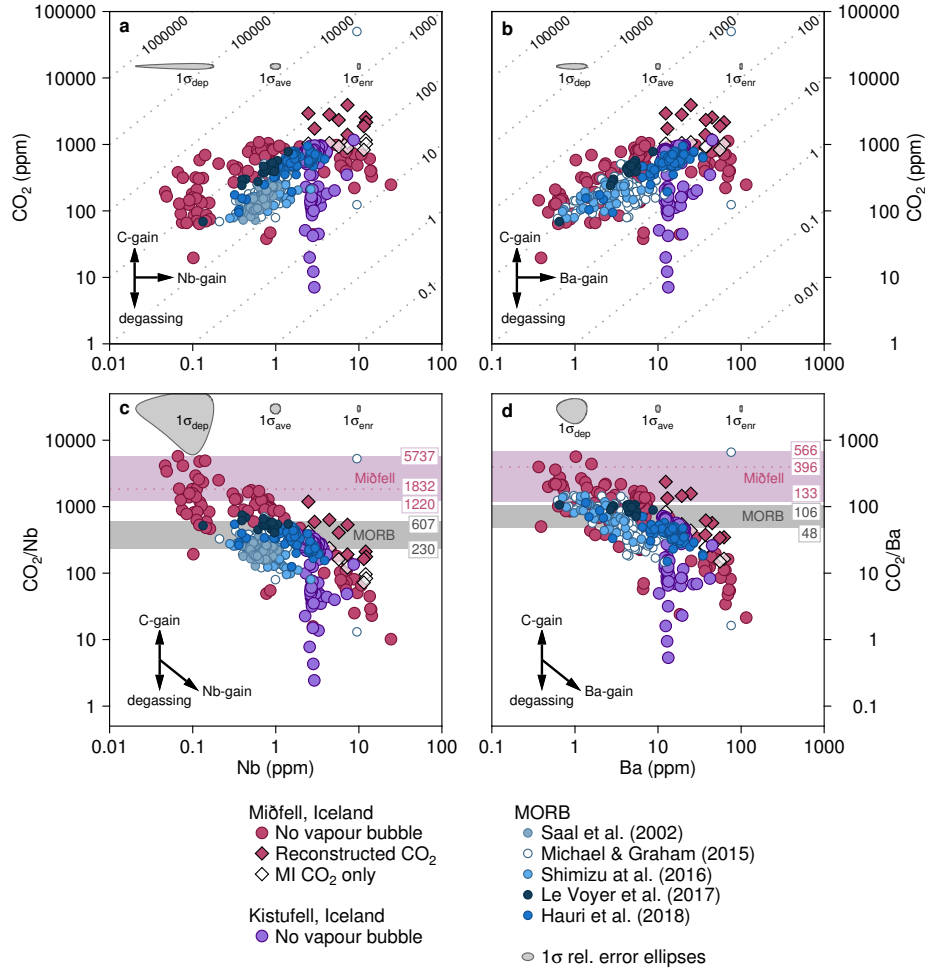


Figure 5: CO₂ content against (a) Nb abundance and (b) Ba abundance, and CO₂/ITE against ITE for (c) Nb and (d) Ba, for melt inclusion and glass suites. Miðfell melt inclusions (red) were measured at Edinburgh (circles) and Woods Hole Oceanic Institution (diamonds). Filled diamonds show reconstructed CO₂ content, while empty diamonds are glass only CO₂. Previously analysed suites are from Siquieros (light blue; Saal et al., 2002), Borgarhraun (royal blue; Hauri et al., 2018), and Equatorial Atlantic (dark blue; Le Voyer et al., 2017) melt inclusions, and MORB glasses (white, Michael and Graham, 2015; sky blue, D-MORB, Shimizu et al., 2016). Dashed lines show constant CO₂/ITE ratios in (a) & (b). Light red regions in (c) & (d) indicate the maximum CO₂/ITE ratios recorded (from high to low) in: (i) the whole population, (ii) melt inclusions with a smooth trace element pattern, and (iii) melt inclusions with matrix glass ITE concentration. Miðfell contains melt inclusions that are enriched in carbon with respect to MORB suites and Kistufell for a range of ITE concentrations. Grey bands show MORB source estimates. One sigma error ellipses are dependent on trace element concentration.

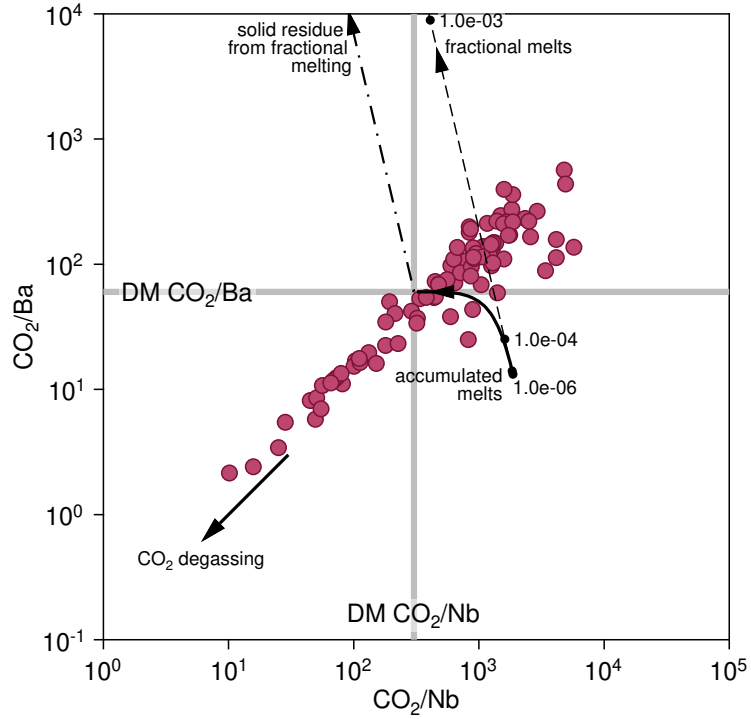


Figure 6: The relationship between CO_2/Nb and CO_2/Ba in Miðfell melt inclusions. The CO_2/ITE ratios of the depleted mantle component (DM), as inferred from the Borgarhraun eruption, are shown as grey bars. These ratios represent the starting source composition for modelling fractional melting. The composition of instantaneous melts of this source follow the dashed line, accumulated melts the solid line, and the residual solid the dashed-dotted line. Points along the instantaneous melts curve are labelled by melt fraction. The highest CO_2/ITE melts cannot be produced by fractional melting of a DM source. Partition coefficients for melting were taken from Rosenthal et al. (2015): $D_C = 0.00055$, $D_{\text{Nb}} = 0.0034$, and $D_{\text{Ba}} = 0.00012$.

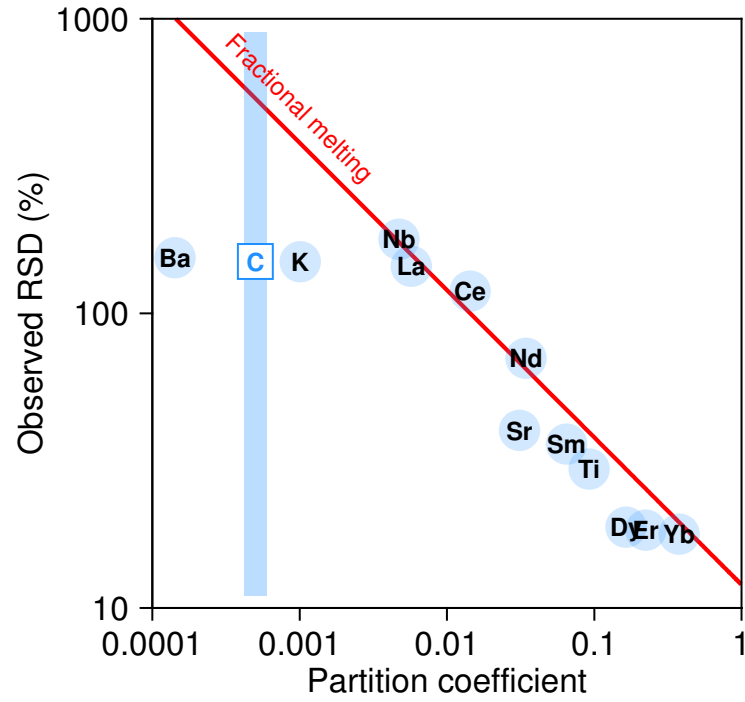


Figure 7: Trace element variability in Miðfell melt inclusions shown by observed relative standard deviation (RSD) of trace elements with varying incompatibility on mantle melting. The results of a pure fractional melting model and the partition coefficients for garnet peridotite melting are shown as a red solid line. The expected behaviour of carbon can be inferred from the blue vertical line, based on its measured compatibility (Rosenthal et al., 2015). Ba, K, Nb, La show similar RSD, suggesting that they remained coupled during melting.

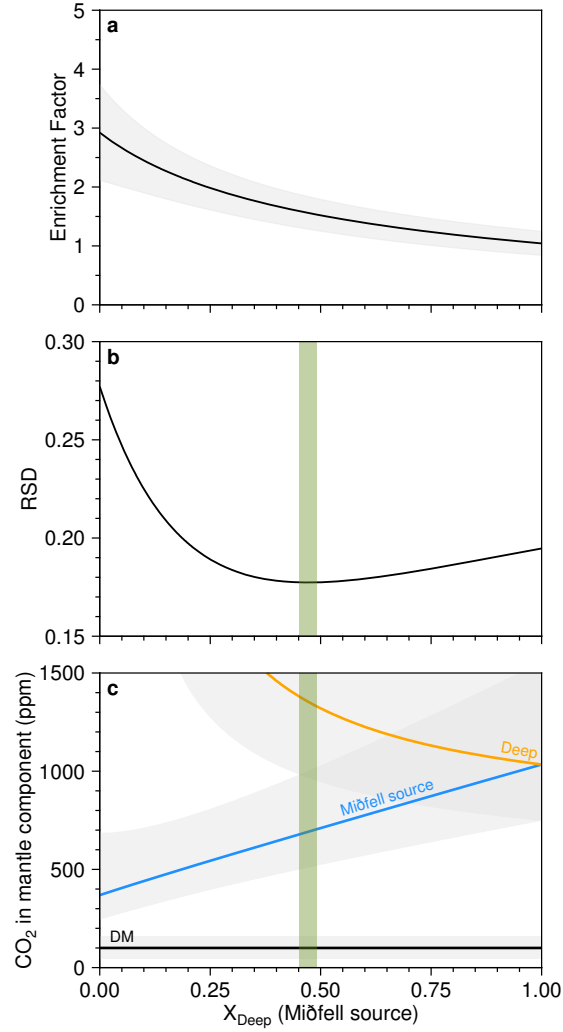


Figure 8: Estimating the fraction of deep mantle component (X_{Deep}) in the Miðfell source (a & b), and the CO_2 concentration of the bulk Miðfell source and its depleted (DM) and deep (Deep) components (c). The best-fit mean enrichment factor (\bar{E}) calculated from Ba, K, Nb and La in Miðfell relative to a mixed source (a; 2σ error as grey field) is defined by the X_{Deep} composition where relative standard deviation of the enrichment factor for the four elements is minimised (b; green band). In (c), the Miðfell source CO_2 content was calculated using \bar{E} and measured CO_2 ($= 1079$ ppm). The CO_2 concentration of the Deep component (orange) was calculated by balancing the Miðfell source content with the DM component (105 ± 57 ppm; black; Ba and Nb estimate average from Hauri et al., 2018). 90% confidence envelopes in grey.

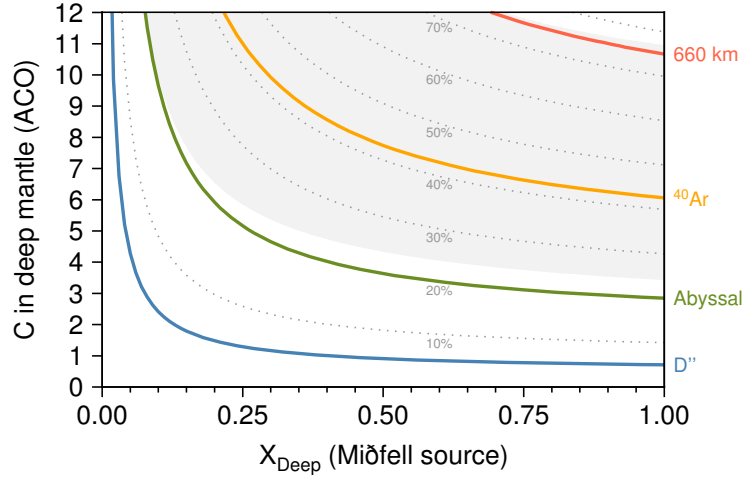


Figure 9: Estimates of deep mantle carbon reservoir mass ($ACO = \text{Atmosphere} + \text{Crust} + \text{Oceans} = 3.1 \times 10^{23} \text{ g CO}_2$), using the Deep concentration curve in Fig. 8c for a range of mantle fractions (grey dotted lines). Four mantle fractions have been highlighted: (i) 5% representing the seismically-defined D'' layer (blue; Tolstikhin and Hofmann, 2005); (ii) 20% for the abyssal layer of the lower mantle, defined by U/K mass balance (green; Arevalo et al., 2009); (iii) 42% for the mass of primitive mantle calculated by ^{40}Ar mass balance (orange; 90% confidence envelope in grey; this study and Allègre et al., 1996); and (iv) 75% for the lower mantle as defined by the 660 km seismic discontinuity (red; Hofmann, 1997).

640 References

- 641 Allègre, C.J., Hofmann, A., O’Nions, K., 1996. The argon constraints on mantle
642 structure. *Geophysical Research Letters* 23, 3555–3557.
- 643 Andersen, M.B., Elliott, T., Freymuth, H., Sims, K.W.W., Niu, Y., Kelley,
644 K.A., 2015. The terrestrial uranium isotope cycle. *Nature* 517, 356–359.
- 645 Anderson, K.R., Poland, M.P., 2017. Abundant carbon in the mantle beneath
646 Hawai’i. *Nature Geoscience* 10, 704–708.
- 647 Arevalo, R., McDonough, W.F., Luong, M., 2009. The K/U ratio of the silicate
648 Earth: Insights into mantle composition, structure and thermal evolution.
649 *Earth and Planetary Science Letters* 278, 361–369. doi:10.1016/j.epsl.
650 2008.12.023.
- 651 Brandon, A.D., Graham, D.W., Waight, T., Gautason, B., 2007. ^{186}Os and
652 ^{187}Os enrichments and high $^3\text{He}/^4\text{He}$ sources in the Earth’s mantle: Evidence
653 from Icelandic picrites. *Geochimica et Cosmochimica Acta* 71, 4570–4591.
- 654 Breddam, K., 2002. Kistufell: Primitive melt from the Iceland mantle plume.
655 *Journal of Petrology* 43, 345–373.
- 656 Breddam, K., Kurz, M.D., Storey, M., 2000. Mapping out the conduit of the
657 Iceland mantle plume with helium isotopes. *Earth and Planetary Science*
658 *Letters* 176, 45–55.
- 659 Canil, D., 1990. Experimental study bearing on the absence of carbonate in
660 mantle-derived xenoliths. *Geology* 18, 1011–1013.
- 661 Caracausi, A., Avice, G., Burnard, P.G., Fűri, E., Marty, B., 2016. Chondritic
662 xenon in the Earth’s mantle. *Nature* 533, 82–85. doi:10.1038/nature17434.

663 Cartigny, P., Pineau, F., Aubaud, C., Javoy, M., 2008. Towards a consistent
 664 mantle carbon flux estimate: Insights from volatile systematics ($\text{H}_2\text{O}/\text{Ce}$,
 665 δD , CO_2/Nb) in the North Atlantic mantle (14° N and 34° N). *Earth and*
 666 *Planetary Science Letters* 265, 672–685.

667 Danyushevsky, L.V., Della-Pasqua, F.N., Sokolov, S., 2000. Re-equilibration of
 668 melt inclusions trapped by magnesian olivine phenocrysts from subduction-
 669 related magmas: petrological implications. *Contributions to Mineralogy and*
 670 *Petrology* 138, 68–83. doi:10.1007/PL00007664.

671 Dasgupta, R., Hirschmann, M.M., 2010. The deep carbon cycle and melting in
 672 Earth’s interior. *Earth and Planetary Science Letters* 298, 1–13.

673 Füri, E., Hilton, D.R., Halldórsson, S.A., Barry, P.H., Hahm, D., Fischer, T.P.,
 674 Grönvold, K., 2010. Apparent decoupling of the He and Ne isotope systematics
 675 of the Icelandic mantle: The role of He depletion, melt mixing, degassing
 676 fractionation and air interaction. *Geochimica et Cosmochimica Acta* 74, 3307–
 677 3332.

678 Gonnermann, H.M., Mukhopadhyay, S., 2007. Non-equilibrium degassing and
 679 a primordial source for helium in ocean-island volcanism. *Nature* 449, 1037–
 680 1040. doi:10.1038/nature06240.

681 Gurenko, A.A., Chaussidon, M., 1995. Enriched and primitive melts included
 682 in olivine from Icelandic tholeiites: origin by continuous melting of a single
 683 mantle column. *Geochimica et Cosmochimica Acta* 59, 2905–2917.

684 Gurenko, A.A., Sobolev, A.V., 2006. Crust-primitive magma interaction be-
 685 neath neovolcanic rift zone of Iceland recorded in gabbro xenoliths from Mid-
 686 fell, SW Iceland. *Contributions to Mineralogy and Petrology* 151, 495–520.

687 Halldórsson, S.A., Barnes, J.D., Stefánsson, A., Hilton, D.R., Hauri, E.H., Mar-
688 shall, E.W., 2016a. Subducted lithosphere controls halogen enrichments in
689 the Iceland mantle plume source. *Geology* 44, 679–682.

690 Halldórsson, S.A., Hilton, D.R., Barry, P.H., Füre, E., Grönvold, K., 2016b.
691 Recycling of crustal material by the Iceland mantle plume: New evidence from
692 nitrogen elemental and isotope systematics of subglacial basalts. *Geochimica*
693 *et Cosmochimica Acta* 176, 206–226. doi:10.1016/j.gca.2015.12.021.

694 Halliday, A.N., 2013. The origins of volatiles in the terrestrial planets. *Geochim-*
695 *ica et Cosmochimica Acta* 105, 146–171.

696 Harrison, D., Burnard, P., Turner, G., 1999. Noble gas behaviour and composi-
697 tion in the mantle: constraints from the Iceland Plume. *Earth and Planetary*
698 *Science Letters* 171, 199–207.

699 Hartley, M.E., Neave, D.A., MacLennan, J., Edmonds, M., Thordarson, T.,
700 2015. Diffusive over-hydration of olivine-hosted melt inclusions. *Earth and*
701 *Planetary Science Letters* 425, 168–178.

702 Hauri, E.H., MacLennan, J., McKenzie, D., Grönvold, K., Oskarsson, N.,
703 Shimizu, N., 2018. CO₂ content beneath northern Iceland and the variability
704 of mantle carbon. *Geology* 46, 55–58.

705 Hayes, J.M., Waldbauer, J.R., 2006. The carbon cycle and associated redox
706 processes through time. *Philosophical Transactions of the Royal Society B*
707 361, 931–950.

708 Hirschmann, M.M., 2016. Constraints on the early delivery and fractionation
709 of Earth’s major volatiles from C/H, C/N, and C/S ratios. *American Miner-*
710 *alogist* 101, 540–553.

711 Hirschmann, M.M., 2018. Comparative deep Earth volatile cycles: The case
 712 for C recycling from exosphere/mantle fractionation of major (H₂O, C, N)
 713 volatiles and from H₂O/Ce, CO₂/Ba, and CO₂/Nb exosphere ratios. *Earth*
 714 *and Planetary Science Letters* 502, 262–273. doi:10.1016/j.epsl.2018.08.
 715 023.

716 Hofmann, A.W., 1997. Mantle geochemistry: the message from oceanic volcan-
 717 ism. *Nature* 385, 219–229.

718 Jenkins, J., Cottaar, S., White, R.S., Deuss, A., 2016. Depressed mantle dis-
 719 continuities beneath Iceland: Evidence of a garnet controlled 660 km discon-
 720 tinuity? *Earth and Planetary Science Letters* 433, 159–168.

721 Kelemen, P.B., Manning, C.E., 2015. Reevaluating carbon fluxes in subduc-
 722 tion zones, what goes down, mostly comes up. *Proceedings of the National*
 723 *Academy of Sciences* 112, E3997–E4006. doi:10.1073/pnas.1507889112.

724 Kokfelt, T., Hoernle, K., Hauff, F., Fiebig, J., Werner, R., Garbe-Schönberg, D.,
 725 2006. Combined trace element and Pb-Nd-Sr-O isotope evidence for recycled
 726 oceanic crust (upper and lower) in the Iceland mantle plume. *Journal of*
 727 *Petrology* 47, 1705–1749.

728 Le Voyer, M., Kelley, K., Cottrell, E., Hauri, E., 2017. Heterogeneity in mantle
 729 carbon content from CO₂-undersaturated basalts. *Nature Communications*
 730 8, 14062.

731 MacLennan, J., 2008a. Concurrent mixing and cooling of melts under Iceland.
 732 *Journal of Petrology* 49, 1931–1953. doi:10.1093/petrology/egn052.

733 MacLennan, J., 2008b. Lead isotope variability in olivine-hosted melt inclusions
 734 from Iceland. *Geochimica et Cosmochimica Acta* 72, 4159–4176.

735 MacLennan, J., 2017. Bubble formation and decrepitation control the CO₂ con-
736 tent of olivine-hosted melt inclusions. *Geochemistry, Geophysics, Geosystems*
737 18, 597–616. doi:10.1002/2016GC006633.

738 MacLennan, J., McKenzie, D., Grönvold, K., 2001. Plume-driven upwelling
739 under central Iceland. *Earth and Planetary Science Letters* 194, 67–82.

740 MacLennan, J., McKenzie, D., Hilton, F., Grönvold, K., Shimizu, N., 2003.
741 Geochemical variability in a single flow from northern Iceland. *Journal of*
742 *Geophysical Research* 108. doi:10.1029/2000JB000142.

743 Marty, B., 2012. The origins and concentrations of water, carbon, nitrogen and
744 noble gases on Earth. *Earth and Planetary Science Letters* 313–314, 56–66.

745 Matthews, S., Shorttle, O., Rudge, J.F., MacLennan, J., 2017. Constraining
746 mantle carbon: CO₂-trace element systematics in basalts and the roles of
747 magma mixing and degassing. *Earth and Planetary Science Letters* 480, 1–
748 14. doi:10.1016/j.epsl.2017.09.047.

749 Matzen, A.K., Baker, M.B., Beckett, J.R., Stolper, E.M., 2011. Fe-Mg parti-
750 tioning between olivine and high-magnesian melts and the nature of Hawai-
751 ian parental liquids. *Journal of Petrology* 52, 1243–1263. doi:10.1093/
752 *petrology/egq089*.

753 McDonough, W.F., Sun, S., 1995. The composition of the Earth. *Chemical*
754 *Geology* 120, 223–253.

755 Michael, P.J., Graham, D.W., 2015. The behavior and concentration of CO₂
756 in the suboceanic mantle: Inferences from undegassed ocean ridge and ocean
757 island basalts. *Lithos* 236–237, 338–351.

758 Mojzsis, S.J., Harrison, T.M., Pidgeon, R.T., 2001. Oxygen-isotope evidence

759 from ancient zircons for liquid water at the Earth's surface 4,300 Myr ago.
760 Nature 409, 178–181.

761 Montelli, R., Nolet, G., Dahlen, F.A., Masters, G., 2006. A catalogue of deep
762 mantle plumes: New results from finite-frequency tomography. *Geochemistry,*
763 *Geophysics, Geosystems* 7. doi:10.1029/2006GC001248.

764 Mukhopadhyay, S., 2012. Early differentiation and volatile accretion recorded
765 in deep-mantle neon and xenon. *Nature* 486, 101–104.

766 Nestola, F., Korolev, N., Koylova, M., Rotiroti, N., Pearson, D.G., Pamato,
767 M.G., Alvaro, M., Peruzzo, L., Gurney, J.J., Moore, A.E., Davidson, J.,
768 2018. CaSiO₃ perovskite in diamond indicates the recycling of oceanic crust
769 into the lower mantle. *Nature* 555, 237–241. doi:10.1038/nature25972.

770 Palme, H., O'Neill, H., 2014. Cosmochemical Estimates of Mantle Composition,
771 in: *Treatise on Geochemistry*. 2 ed.. Elsevier. volume 3, pp. 1–39. doi:10.
772 1016/B978-0-08-095975-7.00201-1.

773 Parai, R., Mukhopadhyay, S., 2015. The evolution of MORB and plume mantle
774 volatile budgets: Constraints from fission Xe isotopes in Southwest Indian
775 Ridge basalts. *Geochemistry, Geophysics, Geosystems* 16, 719–735. doi:10.
776 1002/2014GC005566.

777 Peate, D.W., Breddam, K., Baker, J.A., Kurz, M.D., Barker, A.K., Prestvik,
778 T., Grassineau, N., Skovgaard, A.C., 2010. Compositional characteristics
779 and spatial distribution of enriched Icelandic mantle components. *Journal of*
780 *Petrology* 51, 1447–1475. doi:10.1093/petrology/egq025.

781 Peters, B.J., Carlson, R.W., Day, J.M.D., Horan, M.F., 2018. Hadean silicate
782 differentiation preserved by anomalous ¹⁴²Nd/¹⁴⁴Nd ratios in the Réunion
783 hotspot source. *Nature* 555. doi:10.1038/nature25754.

784 Rosenthal, A., Hauri, E., Hirschmann, M., 2015. Experimental determination
 785 of C, F, and H partitioning between mantle minerals and carbonated basalt,
 786 CO₂/Ba and CO₂/Nb systematics of partial melting, and the CO₂ contents
 787 of basaltic source regions. *Earth and Planetary Science Letters* 412, 77–87.

788 Saal, A.E., Hauri, E., Langmuir, C.H., Perfit, M., 2002. Vapour undersaturation
 789 in primitive mid-ocean-ridge basalt and the volatile content of Earth’s upper
 790 mantle. *Nature* 419, 451–455.

791 Schiano, P., Allègre, C.J., Dupré, B., Lewin, E., Joron, J.L., 1993. Variability
 792 of trace elements in basaltic suites. *Earth and Planetary Science Letters* 119,
 793 37–51.

794 Shimizu, K., Saal, A.E., Myers, C.E., Nagle, A.N., Hauri, E.H., Forsyth, D.W.,
 795 Kamenetsky, V.S., Niu, Y., 2016. Two-component mantle melting-mixing
 796 model for the generation of mid-ocean ridge basalts: Implications for the
 797 volatile content of the Pacific upper mantle. *Geochimica et Cosmochimica*
 798 *Acta* 176, 44–80.

799 Shishkina, T., Botcharnikov, R., Holtz, F., Almeev, R., Portnyagin, M., 2010.
 800 Solubility of H₂O- and CO₂-bearing fluids in tholeiitic basalts at pressures up
 801 to 500 MPa. *Chemical Geology* 277, 115–125.

802 Shorttle, O., 2015. Geochemical variability in MORB controlled by concurrent
 803 mixing and crystallisation. *Earth and Planetary Science Letters* 424, 1–14.

804 Shorttle, O., MacLennan, J., 2011. Compositional trends of Icelandic basalts:
 805 Implications for short-lengthscale lithological heterogeneity in mantle plumes.
 806 *Geochemistry, Geophysics, Geosystems* 12. doi:10.1029/2011GC003748.

807 Shorttle, O., MacLennan, J., Piotrowski, A.M., 2013. Geochemical provincialism
 808 in the Iceland plume. *Geochimica et Cosmochimica Acta* 122, 363–397.

- 809 Sleep, N.H., 2009. Stagnant lid convection and carbonate metasomatism of
 810 the deep continental lithosphere. *Geochemistry, Geophysics, Geosystems* 10.
 811 doi:10.1029/2009GC002702.
- 812 Sobolev, A.V., Hofmann, A.W., Brüggmann, G., Batanova, V.G., Kuzmin, D.V.,
 813 2008. A quantitative link between recycling and osmium isotopes. *Science*
 814 321, 321.
- 815 Sobolev, A.V., Hofmann, A.W., Sobolev, S.V., Nikogosian, I.K., 2005. An
 816 olivine-free mantle source of Hawaiian shield basalts. *Nature* 434, 412–417.
- 817 Stracke, A., 2012. Earth’s heterogeneous mantle: a product of convection driven
 818 interaction between crust and mantle. *Chemical Geology* 330–331, 274–299.
- 819 Stracke, A., Zindler, A., Salters, V.J.M., McKenzie, D., Blichert-Toft, J., Al-
 820 barède, F., Grönvold, K., 2003. Theistareykir revisited. *Geochemistry, Geo-*
 821 *physics, Geosystems* 4. doi:10.1029/2001GC000201.
- 822 Thirlwall, M.F., Gee, M.A.M., Taylor, R.N., Murton, B.J., 2004. Mantle com-
 823 ponents in Iceland and adjacent ridges investigated using double-spike Pb
 824 isotope ratios. *Geochimica et Cosmochimica Acta* 68, 361–386.
- 825 Tolstikhin, I., Hofmann, A.W., 2005. Early crust on top of the Earth’s core.
 826 *Physics of the Earth and Planetary Interiors* 148, 109–130. doi:10.1016/j.
 827 *pepi*.2004.05.011.
- 828 Trieloff, M., Kunz, J., 2005. Isotope systematics of noble gases in the Earth’s
 829 mantle: possible sources of primordial isotopes and implications for mantle
 830 structure. *Physics of the Earth and Planetary Interiors* 148, 13–38.
- 831 Trieloff, M., Kunz, J., Clague, D.A., Harrison, D., Allège, C.J., 2000. The
 832 nature of pristine noble gases in mantle plumes. *Science* 288, 1036–1038.

- 833 Walker, J.C.G., Hayes, P.B., Kasting, J.F., 1981. A negative feedback mecha-
834 nism for the long-term stabilization of Earth's surface temperature. *Journal*
835 *of Geophysical Research* 86, 9776–9782.
- 836 Williams, C.D., Mukhopadhyay, S., 2018. Capture of nebular gases during
837 Earth's accretion is preserved in deep-mantle neon. *Nature* doi:10.1038/
838 s41586-018-0771-1.
- 839 Workman, R.K., Hart, S.R., 2005. Major and trace element composition of the
840 depleted MORB mantle (DMM). *Earth and Planetary Science Letters* 231,
841 53–72. doi:10.1016/j.epsl.2004.12.005.
- 842 Yuan, K., Romanowicz, B., 2017. Seismic evidence for partial melting at the
843 root of major hot spot plumes. *Science* 357, 393–397.