

Radiation as a Mechanism for Heat Transfer

Cody Petrie

Southern Utah University

NO QUIZ TODAY

I was told that that my lesson can't have any effect on your grades

Radiation

What do you think of when you hear the word radiation?

Radiation

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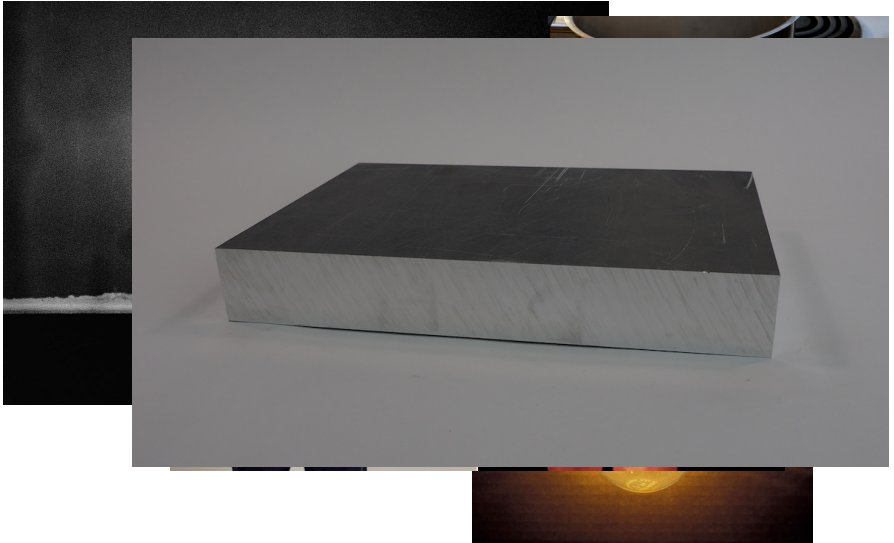
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Review

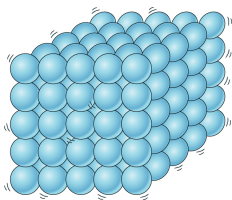
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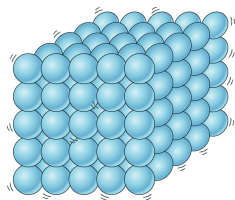
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- What is **internal energy**?
 - The energy of all of the internal pieces (atoms and molecules) of an object, as seen at rest.



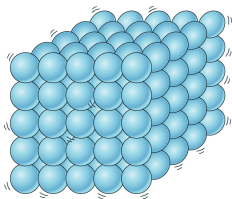
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- What about **work** and **heat**?



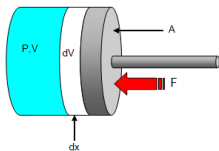
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 - Work is energy being transferred by some force.



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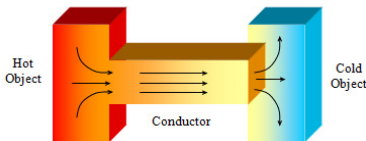
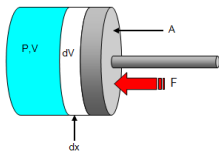
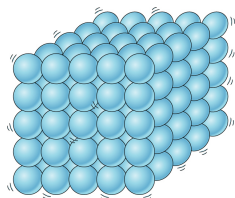
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- Heat is the flow of energy from one thing to another, usually because of a temperature difference.



First Law of Thermodynamics

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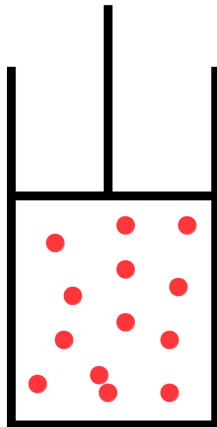
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$$\Delta E_{\text{int}} = Q_{\text{added to}} + W_{\text{done on}}$$

Check for Understanding

I have a gas contained in a metal cylinder with a piston. I want to temporarily raise the energy of the gas contained in a cylinder. How should I do it and how much will it change the energy by?

- A. Push the piston down, $\Delta E_{\text{int}} = \int_{V_i}^{V_f} P dV$
- B. Push the piston down, $\Delta E_{\text{int}} = - \int_{V_i}^{V_f} P dV$
- C. Pull the piston up, $\Delta E_{\text{int}} = \int_{V_i}^{V_f} P dV$
- D. Pull the piston up, $\Delta E_{\text{int}} = - \int_{V_i}^{V_f} P dV$
- E. Quit and give up



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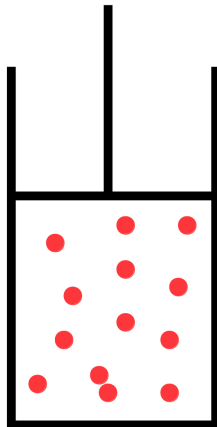
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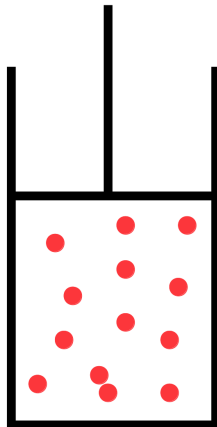
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Check for Understanding - Again

Same situation, we want to add energy, but since we used a metal cylinder the piston rusted over. What should we do now to add energy?

- A. Keep pushing on the piston
- B. Wait for a really long time for something to happen
- C. Give up, but give the container a good hard kick to make us feel better

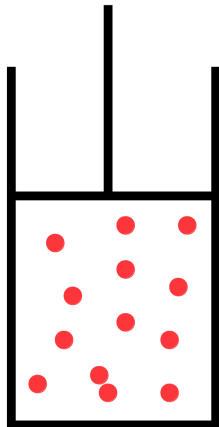


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And this leads us into our next topic: heat transfer



Picture References

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