

Problem I/ A) False; $Sh \leftrightarrow Nu$

B) True; $Sc = \frac{\nu}{D_{AB}}$

C) False; on L-side of equilibrium line

D) False; $\Delta T_H > \Delta T_c$ and $(\dot{m} \Delta T)_H = (\dot{m} \Delta T)_c$ (c_p same)

E) True

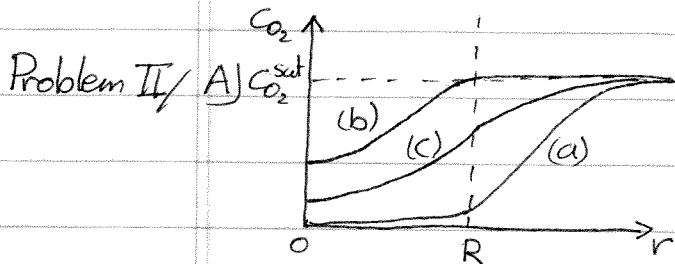
F) True;

G) True; $r^2 N_{Ar} = \text{constant}$

H) True; D_{AB} dominant \Rightarrow no gradients

I) False

J) True; upward facing hot plate has higher h



* for all cases: $\frac{dc}{dr}|_{r=0} = 0$ (symmetry), $c \rightarrow c^{sat}$ for $r \rightarrow \infty$

* (a): very low conc. inside sphere

(b): $c = c^{sat}$ at $r = R$

(c): "in between" (a) and (b)

B) * spray tower; since resistance is mostly in gas phase, mixing of gas phase by droplets will enhance overall mass transfer.

C) * mass transfer is liquid-phase controlled, so k_G does not have real effect on overall coefficient K_L : $\frac{1}{K_L} = \frac{1}{k_L} + \frac{1}{mk_G}$ (k_G large) $\Rightarrow K_L \approx k_L$

Problem III/ * $T_H = 355$ K is constant (sat. steam at 0.51 bar) \Rightarrow condenser behaves like ideal counterflow



from App.

$$A) (\dot{m} c_p \Delta T_c)_c = \dot{m}_{\text{steam}} \cdot h_{fg} \Rightarrow \Delta T_c = \frac{\dot{m}_{\text{steam}} \cdot h_{fg}}{\dot{m}_c \cdot c_{p,\text{cold}}} = \frac{1.5}{15} \cdot \frac{2304 \cdot 10^3 \text{ J/kg}}{4180 \text{ J/kg}\cdot\text{K}} = 55.1 \text{ K}$$

$\Rightarrow T_{C,o} = 335.1 \text{ K} (\Rightarrow T_{\text{cave}} \approx 307 \text{ K})$ $\xrightarrow{\text{guess at 300 K}}$ $\xleftarrow{\text{close enough}}$

B) $F=1$ (T_H constant $\Rightarrow Z=0$)

$$C) UA = \left(\frac{1}{h_i A_i} + \frac{1}{h_o A_o} \right)^{-1} \Rightarrow U = \left(\frac{1}{h_i} + \frac{1}{h_o} \right)^{-1}$$

$A_i = A_o = A$ (thin-walled) \uparrow given

* h_i ? Flow through pipe: $Re_D = \frac{\rho v D}{\mu} = \frac{4 \dot{m}}{\pi D \mu} \quad (\dot{m} = \rho v \frac{\pi D^2}{4}) = \frac{4 \cdot 15/100}{\pi \cdot 0.01 \cdot 700 \cdot 10^{-6}} \xrightarrow{\text{at } T_{cave} \approx 307K} = 2.73 \cdot 10^4$

\Rightarrow Dittus-Boelter: $Nu_D = 0.023 \cdot Re_D^{0.8} \cdot Pr^{0.4} \xrightarrow{\text{water being heated}} = 0.023 \cdot (2.73 \cdot 10^4)^{0.8} \cdot (5.0)^{0.4} = 155$ (turbulent)

$\Rightarrow h_i = \frac{k}{D} Nu_D = \frac{0.62}{0.01} \cdot 155 = 9608 \text{ W/m}^2\text{K}$

$\Rightarrow U = \left(\frac{1}{5000} + \frac{1}{9608} \right)^{-1} = 3289 \text{ W/m}^2\text{K}$

D) $UA \Delta T_{LM} \cdot F = (\dot{m} h_{fg})_{\text{steam}} \Rightarrow A = 100 \cdot \pi D L = \frac{\dot{m} h_{fg}}{U \Delta T_{LM}} \quad \Delta T_{LM} = \frac{75-20}{\ln(75/20)} = 41.6 \text{ K}$

$\Rightarrow L = \frac{1.5 \cdot 2304 \cdot 10^3}{3289 \cdot 41.6 \cdot 100 \cdot \pi \cdot 0.01} = 8.04 \text{ m}$

E) \dot{m} changes for steam, but also $T_{c,o}$ (and thus ΔT_{LM})

\Rightarrow new situation: $\Delta T_c = \frac{1.0 \cdot 2304 \cdot 10^3}{15 \cdot 4180} = 36.7 \text{ K} \Rightarrow T_{c,o} = 316.7 \text{ K} \Rightarrow \Delta T_{LM} = \frac{75-38.3}{\ln(75/38.3)} = 54.6 \text{ K}$

$\Rightarrow U = \frac{\dot{m} h_{fg}}{100 \pi D L \cdot \Delta T_{LM}} = \frac{1.0 \cdot 2304 \cdot 10^3}{100 \cdot \pi \cdot 0.01 \cdot 8.04 \cdot 54.6} = 1671 \text{ W/m}^2\text{K}$

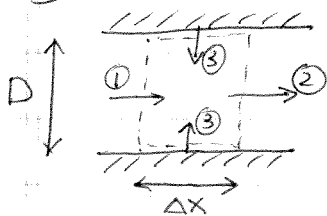
Problem IV/A) k_c ? Flow in pipe $\Rightarrow Re_D = \frac{V_{ave} \cdot D}{\nu \rightarrow 300K \text{ air}} = \frac{0.3 \cdot 0.05}{1.57 \cdot 10^{-5}} = 955$ laminar!

$Sc = \frac{\nu}{D_{AB}} = \frac{1.57 \cdot 10^{-5}}{10^{-5}} = 1.57$

$\Rightarrow Sh_D = \frac{k_c D}{D_{AB}} = 1.86 \cdot \left(\frac{D}{L} \cdot Re \cdot Sc \right)^{1/3} = 1.86 \cdot \left(\frac{0.05}{8} \cdot 955 \cdot 1.57 \right)^{1/3} = 4.32$

$\Rightarrow k_c = 4.32 \cdot \frac{10^{-5}}{0.05} = 8.63 \cdot 10^{-4} \text{ m/s} = 0.863 \text{ mm/s}$

B) Flux $N_A = k_c (C_A^{\text{sat}} - C_A)$ but C_A varies along tube \Rightarrow differential mass balance



$V_{ave} \frac{\pi D^2}{4} C_A|_x + \underbrace{\pi D \Delta x}_{A_{\text{wall}}} N_A = V_{ave} \frac{\pi D^2}{4} C_A|_{x+\Delta x}$

$\Rightarrow V_{ave} \frac{D}{4} \left[\frac{C_A|_{x+\Delta x} - C_A|_x}{\Delta x} \right] = k_c (C_A^{\text{sat}} - C_A)$

$\Rightarrow \frac{dC_A}{dx} = \frac{4k_c}{V_{ave} D} (C_A^{\text{sat}} - C_A) \Rightarrow \frac{dC_A}{C_A^{\text{sat}} - C_A} = \frac{4k_c dx}{V_{ave} D}$

\Rightarrow Integrate from $x=0$ ($C_A=0$) to $x=L$ ($C_{A,\text{out}}$):

$-\ln \left(\frac{C_A^{\text{sat}} - C_{A,\text{out}}}{C_A^{\text{sat}}} \right) = -\ln \left(1 - \frac{C_{A,\text{out}}}{C_A^{\text{sat}}} \right) = \frac{4k_c L}{V_{ave} D}$

$\Rightarrow 1 - \frac{C_{A,\text{out}}}{C_{A,\text{sat}}} = 1 - \frac{P_{A,\text{out}}}{P_{A,\text{sat}}} = \exp \left(-\frac{4k_c L}{V_{ave} D} \right) \Rightarrow P_{A,\text{out}} = P_A^{\text{sat}} \left(1 - \exp \left(-\frac{4k_c L}{V_{ave} D} \right) \right)$

$0.251 \quad = 0.0112 \text{ atm}$

C) All solvent that leaves wall has to get out of the pipe

$M_A \cdot C_{A,\text{out}} \cdot \frac{\pi D^2}{4} \cdot V_{ave} = M_A \frac{P_{A,\text{out}}}{RT} \frac{\pi D^2}{4} V_{ave} = 85 \cdot 10^{-3} \cdot \frac{0.0112 \text{ atm}}{8.206 \cdot 10^{-5} \frac{\text{m}^3 \cdot \text{atm}}{\text{mol} \cdot \text{K}}} \cdot \frac{\pi (0.05)^2 \cdot 0.3}{4} = 2.28 \cdot 10^{-5} \text{ kg/s}$

D) * Required heat flow $q = \dot{m} \cdot h_{fg} = 2.28 \cdot 10^{-5} \text{ kg/s} \cdot 300 \text{ kJ/kg} = 6.83 \cdot 10^{-3} \text{ kJ/s}$

* Air temp change: $q = \dot{m} c_p \Delta T = \rho V_{ave} \frac{\pi D^2}{4} c_p \Delta T$ at 300K

$$\Rightarrow \Delta T = \frac{6.83}{1.177 \cdot 1006 \cdot 0.3 \cdot \frac{\pi}{4} (0.05)^2} = 9.8 \text{ K}$$

* Average temp difference between liquid and bulk air:

$$\Delta T = \frac{q}{h_{ave} \pi D L}$$

h_{ave} from Chilton-Colburn: $h_{ave} = \rho c_p k \left(\frac{Sc}{Pr} \right)^{2/3} = 1.177 \cdot 1006 \cdot 0.863 \cdot 10^{-3} \left(\frac{1.57}{0.708} \right)^{2/3}$

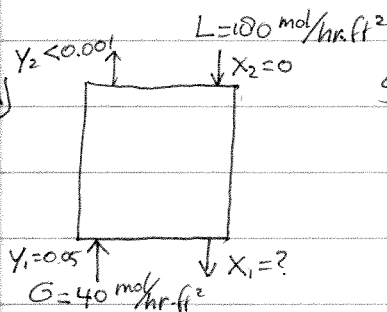
$$= 1.74 \text{ W/m}^2 \cdot \text{K}$$

$$\Rightarrow \Delta T = \frac{6.83}{1.74 \cdot \pi \cdot 0.05 \cdot 6} = 4.2 \text{ K}$$

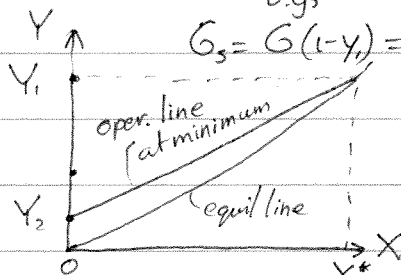
E) * Coldest spot in liquid is near outlet $\sim 286 \text{ K}$

* Average bulk air temp ($\sim 295 \text{ K}$) and film temp ($\sim 293 \text{ K}$) are not really affected by evaporation

Problem IV/A)



Solute free basis: $Y_1 = \frac{0.05}{0.95} = 0.0526$, $Y_2 = \frac{0.001}{0.999} \approx 0.001$, $X_2 = 0$



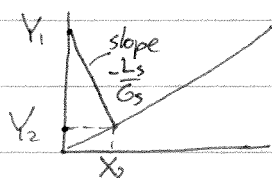
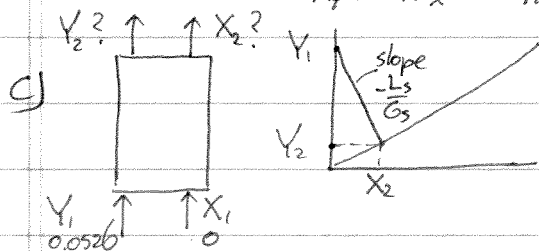
At minimum L_s : $\frac{L_s}{G_s} \Big|_{\min} \Rightarrow X_1 = X_1^* = \frac{X_1^*}{1 - X_1^*}$ with $X_1^* = \frac{Y_1}{2.7} = 0.0185 \Rightarrow X_1^* = 0.0188$

$$\Rightarrow \frac{L_s}{G_s} \Big|_{\min} = \frac{Y_1 - Y_2}{X_1^*} = \frac{0.0526 - 0.001}{0.0188} = 2.76$$

$$\frac{L_s}{G_s} \Big|_{\text{actual}} = \frac{180}{38} = 4.74 > 2.76 \Rightarrow \text{goal can be achieved}$$

B) * $\frac{1/k_y}{1/k_y + 9/k_x}$ with equil. constant $C = \frac{Y}{X} = 2.7$

$$\Rightarrow \frac{1/k_{ya}}{1/k_{ya} + 9/k_{xa}} = \frac{1/15}{1/15 + 2.7/50} = 0.552 \Rightarrow 55.2\% \text{ in gas phase}$$



Operating line: $Y = Y_1 - \frac{L_s}{G_s} X$ Equil. line: $Y = 2.7 \cdot X$

$$\Rightarrow \text{at intersection } (X_2, Y_2): Y_1 - \frac{L_s}{G_s} X_2 = 2.7 X_2 \Rightarrow X_2 = \frac{Y_1}{7.44}$$

$$\Rightarrow X_2 = 0.00707 \Rightarrow Y_2 = 2.7 \cdot X_2 = 0.0191$$

D) (Y_2, X_2) in C achieved in very tall co-flow tower

Not low enough!

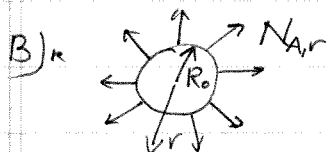
Problem VI/A) * Hirschfelder equation for naphthalene (A) in air (B)

$$M_A = 128 \text{ g/mol}, M_B = 29 \text{ g/mol}, T = 303 \text{ K}, P = 1 \text{ atm}, \sigma_A = 6.2 \text{ \AA}, \sigma_B = 3.617 \text{ \AA}$$

$$\frac{\epsilon_A}{\pi} = 550 \text{ K}, \frac{\epsilon_B}{\pi} = 97 \text{ K} \Rightarrow \sigma_{AB} = \frac{6.2 + 3.617}{2} = 4.91 \text{ \AA}, \epsilon_{AB} = \sqrt{550 \cdot 97} = 231 \text{ K}$$

$$\Rightarrow D_{AB} = \frac{0.001858 (303)^{3/2} (1/128 + 1/29)^{1/2}}{1 \cdot (4.91)^2 \cdot 1.273} = 0.060 \text{ cm}^2/\text{s} \quad \left(\Rightarrow \frac{\pi \epsilon_{AB}}{RT} = 1.312 \Rightarrow \Omega_{AB} = 1.273 \right)$$

$$= 6.6 \cdot 10^{-6} \text{ m}^2/\text{s}$$



* CV: air outside sphere \Rightarrow spherical coordinates

* 1D diffusion ($N_{A,r}$ only), pseudo steady state, no bulk reaction

$$\Rightarrow \text{Gen. diff. equation: } \vec{\nabla} \cdot \vec{N}_A + \frac{\partial C_A}{\partial t} = 0 \Rightarrow \frac{1}{r^2} \left(\frac{d}{dr} (r^2 N_{A,r}) \right) = 0$$

$$\Rightarrow r^2 N_{A,r} = \text{constant}$$

* stagnant medium and $y_A \ll 1$ (max $y_A = 0.0015$)

$$\Rightarrow \text{Fick's rate equation: } N_{A,r} = -C D_{AB} \frac{dy_A}{dr} + y_A (N_{A,r} + N_{B,r})$$

$$\text{C) } \int_{r=R_0}^{\infty} \frac{r^2 N_{A,r}}{r^2} dr = - \int_{y_A^{\text{sat}}}^0 C D_{AB} dy_A \Rightarrow \underbrace{(r^2 N_{A,r})}_{\text{constant}} \left[-\frac{1}{r} \right]_{R_0}^{\infty} = -C D_{AB} [y_A]_{y_A^{\text{sat}}}^0$$

$$\Rightarrow (r^2 N_{A,r}) \frac{1}{R_0} = C D_{AB} y_A^{\text{sat}} \Rightarrow (r^2 N_{A,r}) = \frac{P_A^{\text{sat}}}{RT} D_{AB} R_0$$

$$\text{* total molar flow } W_A = 4\pi r^2 N_{A,r} = 4\pi \frac{P_A^{\text{sat}}}{RT} D_{AB} R_0$$

$$\Rightarrow \text{mass flow } W_A \cdot M_A = 4\pi M_A \frac{P_A^{\text{sat}}}{RT} D_{AB} R_0 = 4\pi \cdot 0.128 \frac{\text{kg/mol}}{\text{kg/mol}} \frac{0.0015 \text{ atm}}{8.206 \cdot 10^{-5} \cdot 303} \cdot 6.6 \cdot 10^{-6} \cdot 0.005$$

$$= 3.21 \cdot 10^{-9} \text{ kg/s}$$

$$= 0.015 \text{ g/hr}$$

D) * With convection around sphere

$$Re_D = \frac{VD}{\nu} = \frac{0.2 \cdot 0.01}{1.57 \cdot 10^{-5}} = 127.4 \quad \left\{ \begin{array}{l} \text{Eqn. 30-9: } Sh_D = 2 + 0.552 \cdot (127.4)^{1/2} \cdot (238)^{1/3} = 10.32 \\ Sc = \frac{\nu}{D_{AB}} = \frac{1.57 \cdot 10^{-5}}{6.6 \cdot 10^{-6}} = 2.38 \end{array} \right. \Rightarrow k_c = \frac{Sh_D \cdot D_{AB}}{D} = \frac{10.32 \cdot 6.6 \cdot 10^{-6}}{0.01} = 0.00681 \text{ m/s}$$

$$\Rightarrow N_A|_{\text{conv}} = k_c \cdot C_A^{\text{sat}}$$

$$\text{* From C): } N_A|_{\text{diff } r=R_0} = C_A^{\text{sat}} \cdot \frac{D_{AB}}{R_0} \Rightarrow \text{enhancement } \frac{N_A|_{\text{conv}}}{N_A|_{\text{diff}}} = \frac{k_c R_0}{D_{AB}} = \frac{0.00681 \cdot 0.005}{6.6 \cdot 10^{-6}} = 5.2$$

$$\text{E) * Mothball contains } \frac{\frac{4}{3}\pi R_0^3 P}{M_A} = \frac{\frac{4}{3}\pi (0.005)^3 \cdot 1140}{0.128} = 4.66 \cdot 10^{-3} \text{ mol naphthalene}$$

$$\text{with half distributed across closet } C_{A,\text{bulk}} = \frac{2.33 \cdot 10^{-3}}{81} = 2.88 \cdot 10^{-5} \text{ mol/m}^3$$

$$\text{* } C_A^{\text{sat}} = \frac{P_A^{\text{sat}}}{RT} = \frac{0.0015}{8.206 \cdot 10^{-5} \cdot 303} = 0.060 \text{ mol/m}^3 \Rightarrow C_{A,\text{bulk}} \ll C_A^{\text{sat}}$$