

MSE 2001B Quiz #5, 9-25-2015

Your name (print) \_\_\_\_\_ Your major \_\_\_\_\_ Score \_\_\_\_\_/10

Signature \_\_\_\_\_

KE

1. Rank the magnitudes of the diffusion coefficients from *the least to the greatest* for the following systems:

- (1) N in Fe at 700 °C
- (2) Cr in Fe at 700 °C
- (3) N in Fe at 900 °C
- (4) Cr in Fe at 900 °C

Note: both Fe and Cr have the BCC crystal structure, and the atomic radii for Fe, Cr, and N are 0.124, 0.125, and 0.065 nm, respectively. (4 pts)

- a) (1), (2), (3), (4)
- ☒ b) (2), (4), (1), (3)
- c) (3), (1), (2), (4)
- d) (2), (4), (3), (1)

2. The *time-dependent* diffusion can be described by (3 pts)

- a) Fick's first law
- ☒ b) Fick's second law

3. One of the practical applications of diffusion is case hardening of steel. If you wish to *increase* the amount of carbon at a given depth below the surface, should you

3.1. Decrease the temperature (assume all other variables constant) (1 pt)

- (a) True
- ☒ (b) False

3.2. Increase the carbon content in the furnace (assume all other variables constant) (1 pt)

- ☒ (a) True
- (b) False

3.3. Increase the time the steel part remains in the carburizing furnace (assume all other variables constant) (1 pt)

- ☒ (a) True
- (b) False