

# 热力学与统计物理-第三次作业

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1.17

We can treat this as binomial distribution: Each molecules only have two states: In the volume or not. The probability of the molecules in the volume is:

$$p = \frac{V}{V_0} \quad (1.1)$$

Then, use Gaussian approximation:

$$P(N) = (2\pi N_0 p q)^{-\frac{1}{2}} \exp\left[-\frac{(N - N_0 p)^2}{2N_0 p q}\right] \quad (1.2)$$

So, we can get the result:

$$\begin{aligned} P(N, N + dN) &= \int_N^{N+dN} (2\pi N_0 p q)^{-\frac{1}{2}} \exp\left[-\frac{(N - N_0 p)^2}{2N_0 p q}\right] dN \\ &= (2\pi N_0 p q)^{-\frac{1}{2}} \exp\left[-\frac{(N - N_0 p)^2}{2N_0 p q}\right] dN \\ &= (2\pi N_0 \frac{V(V_0 - V)}{V_0^2})^{-\frac{1}{2}} \exp\left[-\frac{(V_0 N - N_0 V)^2}{2N_0 V(V_0 - V)}\right] dN \end{aligned} \quad (1.3)$$

1.18

Answer:

This question equal to the off-lattice 3D random walk.

For each step, we have the constant length  $l$ , and a random direction that can be described by  $\theta, \phi$ ,  $\theta$  is the angle between the projection of direction vector on XY plane and the X positive semi-axis, and  $\phi$  is the angle between the direction vector and XY plane. For N steps:

$$\{\theta_1, \theta_2, \dots, \theta_N, \phi_1, \phi_2, \dots, \phi_N\} \quad (2.1)$$

Then, we have:

$$\begin{cases} \overline{r_x^2} = l^2 \times \overline{\sum_{i=1}^N (\cos^2 \phi_i \cos^2 \theta_i)} = l^2 \times N \times \overline{\cos^2 \phi \cos^2 \theta} \\ \overline{r_y^2} = l^2 \times \overline{\sum_{i=1}^N (\cos^2 \phi_i \sin^2 \theta_i)} = l^2 \times N \times \overline{\cos^2 \phi \sin^2 \theta} \\ \overline{r_z^2} = l^2 \times \overline{\sum_{i=1}^N (\sin^2 \phi_i)} = l^2 \times N \times \overline{\sin^2 \phi} \end{cases} \quad (2.2)$$

Using the periodicity of triangle function:

$$\begin{cases} \overline{\cos^2 \theta} = \int_0^{2\pi} \cos^2 \theta d\theta = \frac{1}{2} + \frac{1}{2} \int_0^{2\pi} \cos(2\theta) d\theta = \frac{1}{2} \\ \overline{\sin^2 \theta} = \int_0^{2\pi} \sin^2 \theta d\theta = \frac{1}{2} - \frac{1}{2} \int_0^{2\pi} \cos(2\theta) d\theta = \frac{1}{2} \end{cases} \quad (2.3)$$

Same for  $\phi$ ,  $\phi$  and  $\theta$  are independent variable, so:

$$\begin{cases} \overline{\cos^2 \phi \cos^2 \theta} = \overline{\cos^2 \phi} \times \overline{\cos^2 \theta} \\ \overline{\cos^2 \phi \sin^2 \theta} = \overline{\cos^2 \phi} \times \overline{\sin^2 \theta} \end{cases} \quad (2.4)$$

Substitute (2.4), (2.3) into (2.2):

$$\begin{cases} \overline{r_x^2} = \frac{N}{4} l^2 \\ \overline{r_y^2} = \frac{N}{4} l^2 \\ \overline{r_z^2} = \frac{N}{2} l^2 \end{cases} \quad (2.5)$$

Finally, we get:

$$\overline{r^2} = \overline{r_x^2} + \overline{r_y^2} + \overline{r_z^2} = N \times l^2 \quad (2.6)$$

2.1

Answer:

For this particle, all of its energy is kinetic energy, so we have:

$$P = \sqrt{2mE} \quad (3.1)$$

And the classical phase space is:

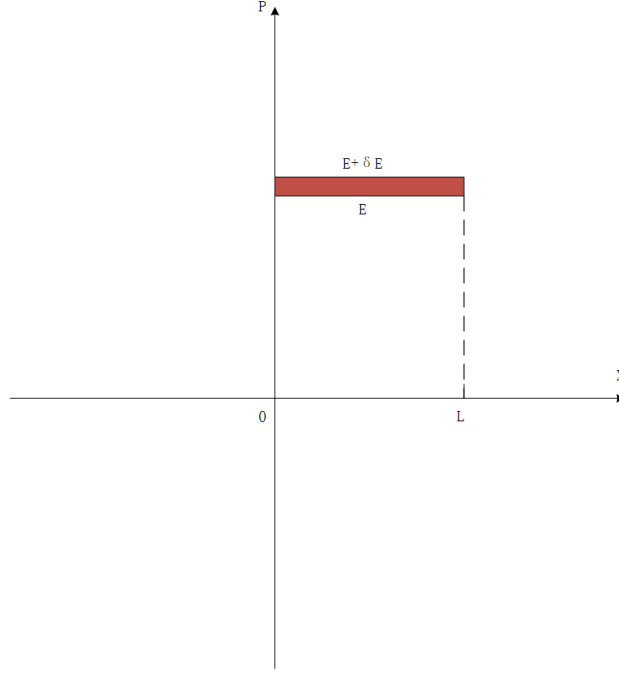


图 1: Classical phase space

2.4

Answer:

(a). The energy of the system is:

$$E = -(n_1 - n_2)\mu H = -(2n_1 - N)\mu H \quad (4.1)$$

So:

$$n_1 = \frac{N}{2} - \frac{E}{2\mu H} \quad (4.2)$$

Since  $\delta E$  is very small compared to  $E$ :

$$\Omega(E) = \int_E^{E+\delta E} \Omega(E') dE' \approx \Omega(E) \delta E = \Omega(n_1) \delta n_1 \quad (4.3)$$

From (4.2):

$$\delta n_1 = \frac{\delta E}{2\mu H} \quad (4.4)$$

Then:

$$\Omega(E) = \frac{N!}{n_1!(N - n_1)!} \frac{\delta E}{2\mu H} = \frac{N!}{(\frac{N}{2} - \frac{E}{2\mu H})!(\frac{N}{2} + \frac{E}{2\mu H})!} \frac{\delta E}{2\mu H} \quad (4.5)$$

(b) From (4.5):

$$\ln \Omega(E) = \ln N! - \ln \left(\frac{N}{2} - \frac{E}{2\mu H}\right)! - \ln \left(\frac{N}{2} + \frac{E}{2\mu H}\right)! + \ln \frac{\delta E}{2\mu H} \quad (4.6)$$

With Stirling's formula:

$$\ln n! \approx n \ln n - n \quad (4.7)$$

(4.6) approximate to:

$$\begin{aligned} \ln \Omega(E) &= N \ln N - \left(\frac{N}{2} - \frac{E}{2\mu H}\right) \ln \left(\frac{N}{2} - \frac{E}{2\mu H}\right) \\ &\quad - \left(\frac{N}{2} + \frac{E}{2\mu H}\right) \ln \left(\frac{N}{2} + \frac{E}{2\mu H}\right) + \ln \frac{\delta E}{2\mu H} \end{aligned} \quad (4.8)$$

(c) For this system, the spin is parallel or antiparallel have an equal probability. So:

$$P(n_1) = \left(\frac{1}{2}\pi N\right)^{-\frac{1}{2}} \exp\left[-\frac{(n_1 - \frac{N}{2})^2}{\frac{N}{2}}\right] \quad (4.9)$$

Consider (4.2):

$$P(E) = \left(\frac{1}{2}\pi N\right)^{-\frac{1}{2}} \exp\left(-\frac{E^2}{2N\mu^2 H^2}\right) \quad (4.10)$$

So, the  $\Omega(E)$  is:

$$\Omega(E) = P(E) \times 2^N = 2^N \left(\frac{1}{2}\pi N\right)^{-\frac{1}{2}} \exp\left(-\frac{E^2}{2N\mu^2 H^2}\right) \quad (4.11)$$

2.7

Answer:

(a). The energy's change is:

$$\Delta E = E(L_x + dL_x) - E(L_x) \quad (5.1)$$

The change of system energy is equal to the negative value of system work to the outside:

$$W = -\Delta E \quad (5.2)$$

The system's work to outside given by:

$$W = \int_{L_x}^{L_x + dL_x} F_x(x') dx' \approx F_x dL_x \quad (5.3)$$

Since  $dL_x$  is a small amount, The approximation in the above formula can be satisfied. Then, from (5.1) to (5.3), we can get:

$$F = -\frac{E(L_x + dL_x) - E(L_x)}{dL_x} = -\frac{\partial E}{\partial L_x} \quad (5.4)$$

Q.E.D (b). The energy of this particle is given by:

$$E = \frac{\hbar^2}{2m} \pi^2 \left( \frac{n_x^2}{L_x^2} + \frac{n_y^2}{L_y^2} + \frac{n_z^2}{L_z^2} \right) \quad (5.5)$$

Then we can calculate the force that particle exert on the wall:

$$\begin{cases} F_x = 2 \frac{\hbar^2}{2m^2} \pi^2 \frac{n_x^2}{L_x^3} \\ F_y = 2 \frac{\hbar^2}{2m^2} \pi^2 \frac{n_y^2}{L_y^3} \\ F_z = 2 \frac{\hbar^2}{2m^2} \pi^2 \frac{n_z^2}{L_z^3} \end{cases} \quad (5.6)$$

Then we can calculate the pressure on three direction:

$$\begin{cases} P_x = 2 \frac{\hbar^2}{2m^2} \pi^2 \frac{n_x^2}{L_x^2 V} \\ P_y = 2 \frac{\hbar^2}{2m^2} \pi^2 \frac{n_y^2}{L_y^2 V} \\ P_z = 2 \frac{\hbar^2}{2m^2} \pi^2 \frac{n_z^2}{L_z^2 V} \end{cases} \quad (5.6)$$

The total pressure is:

$$P = \frac{P_x + P_y + P_z}{3} = \frac{2}{3V} \frac{\hbar^2}{2m^2} \pi^2 \left( \frac{n_x^2}{L_x^2} + \frac{n_y^2}{L_y^2} + \frac{n_z^2}{L_z^2} \right) \quad (5.7)$$

The average is:

$$\bar{P} = \frac{2}{3} \frac{\bar{E}}{V} \quad (5.8)$$

2.10

Answer:

We can treat  $\bar{P}$  as a function of V:

$$\bar{P} = KV^{-\gamma} \quad (6.1)$$

Since the work is done through a quasistatic process, we can divide the change of volume into three dimensions' length change, here we only consider X direction, the other two directions will be the same and easy to prove:

$$F_x = \bar{P}_x S_x \quad (6.2)$$

$$W = \int_{X_s}^{X_f} F_x dx = \int_{X_s}^{X_f} K(S_x X)^{-\gamma} S_x dx = K S_x^{-\gamma+1} \int_{X_s}^{X_f} X^{-\gamma} dx \quad (6.3)$$

Then:

$$W = -K(\gamma - 1)(V_f^{-\gamma+1} - V_s^{-\gamma+1}) \quad (6.4)$$

Replace K with  $\bar{P}_s V_s^\gamma$ :

$$W = -\bar{P}_s V_s^\gamma (\gamma - 1)(V_f^{-\gamma+1} - V_s^{-\gamma+1}) \quad (6.5)$$

2.11.

Answer:

First, we'll calculate the difference of internal energy between state A and B.

$$\Delta E = -W = \int_{V_A}^{V_B} \bar{P} dV = \int_{V_A}^{V_B} \alpha V^{(-5/3)} dV = -3600J \quad (7.1)$$

(a).

$$W = \int_{V_A}^{V_{B'}} \bar{P} dV = 22400J \quad (7.2)$$

With the first law:

$$\Delta E = Q - W = -3600J \quad (7.3)$$

$$Q = 18800J \quad (7.4)$$

(b)

$$W = \int_{V_A}^{V_B} \bar{P} dV = 11500J \quad (7.5)$$

From (7.3):

$$Q = 7950J \quad (7.6)$$

(c)

$$W = \int_{V'_A}^{V_B} \bar{P} dV = 700J \quad (7.7)$$

From (7.3):

$$Q = -2900J \quad (7.8)$$