Inorganic Chemical Nomenclature

Binary Compound Names for Nonmetal Ions

H-	hydride	N ³⁻	nitride	O ²⁻	oxide	F-	fluoride
H⁺	hydrogen	P ³⁻	phosphide	S ²⁻	sulfide	Cl ⁻	chloride
		As ³⁻	arsenide	Se ²⁻	selenide	Br -	bromide
				Te ²⁻	telluride	l ⁻	iodide

Polyatomic Ion Names

NH ₃	ammonia	CIO ₄	perchlorate	C ₂ O ₄ ²⁻	oxalate
NH ₄ ⁺	ammonium	CIO ₃	chlorate	O ₂ ²⁻	peroxide
H ₃ O ⁺	hydronium	CIO ₂ -	chlorite	PO ₄ ³⁻	phosphate
CH ₃ COO ⁻ (C ₂ H ₃ O ₂ ⁻)	acetate	CIO-	hypochlorite	HPO ₄ ²⁻	monohydrogen phosphate
AsO ₄ ³⁻	arsenate	CrO ₄ ²⁻	chromate	H ₂ PO ₄	dihydrogen phosphate
BO ₃ ³⁻	borate	Cr ₂ O ₇ ²⁻	dichromate	PO ₃ ³⁻	phosphite
B ₄ O ₇ ²⁻	tetraborate	CN ⁻	cyanide	SeO ₄ ²⁻	selenate
BrO ₃	bromate	OH ⁻	hydroxide	SiO ₃ ²⁻	silicate
BrO ⁻	hypobromite	10 ₄ -	periodate	SiF ₆ ²⁻	hexafluorosilicate
CO ₃ ²⁻	carbonate	10 ₃ -	iodate	SO ₄ ²⁻	sulfate
HCO ₃	hydrogen carbonate (bicarbonate)	10 ⁻	hypoiodite	HSO ₄	hydrogen sulfate (bisulfate)
		MnO ₄	permanganate	SO ₃ ²⁻	sulfite
		NO ₃	nitrate	HSO ₃ -	hydrogen sulfite (bisulfite)
		NO ₂	nitrite	C ₄ H ₄ O ₆ ²⁻	tartrate
				S ₂ O ₃ ²⁻	thiosulfate

Common Acid Names

HC ₂ H ₃ O ₂ CH ₃ COOH	acetic acid	HNO ₃	nitric acid	
H ₂ CO ₃	carbonic acid	H ₃ PO ₄	phosphoric acid	
HCI	hydrochloric acid	H ₂ SO ₄	sulfuric acid	