

# **IIT - ORGANIC CHEMISTRY**

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## **NURTURE**

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**Corporate Office: NAIVEDHYAM, Plot No. SP-11, Old INOX, Indra Vihar,  
Kota (Raj.) 324005**

DPP # 06

Time : 30 Min.

1. Match the column

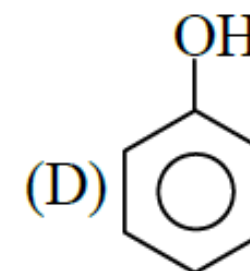
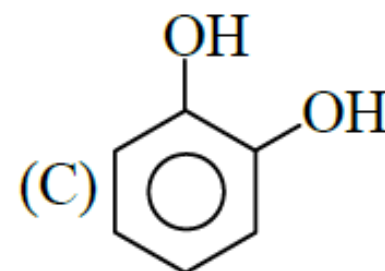
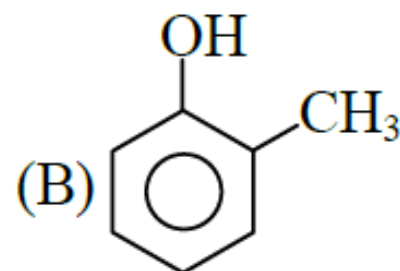
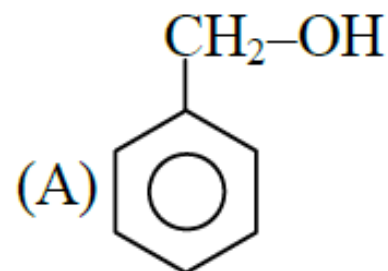
**Column-I**

- (A) Carbocation  $\text{CH}_3^{\oplus}$   
(B) Carbanion  $\text{CH}_3^{\ominus}$   
(C) Carbon free radical  $\dot{\text{C}}\text{H}_3$   
(D) Ammonia  $\ddot{\text{N}}\text{H}_3$

**Column-II**

- (P) Para magnetic  
(Q) Diamagnetic  
(R) Electron deficient central atom  
(S)  $\text{sp}^2$  hybridisation  
(T)  $\text{sp}^3$  hybridisation

2. Homologue of phenol is :



3. Which of the following is the correct order of -I effect.
- (A)  $-\text{NO}_2 > -\text{CN} > -\text{COOH}$  (B)  $-\text{F} > \text{Cl} > -\text{Br} > -\text{I}$
- (C)  $-\overset{\oplus}{\text{N}}\text{H}_3 > -\overset{\oplus}{\text{N}}\text{H}_3\text{Me} > -\overset{\oplus}{\text{N}}\text{H}_3\text{Me}_3$  (D)  $-\text{C}\equiv\text{CH} > -\text{Ph} > -\text{CH}=\text{CH}_2$
4. What is the correct order of +I effect of the following groups.
- |               |                |                |                       |             |             |             |
|---------------|----------------|----------------|-----------------------|-------------|-------------|-------------|
| $-\text{O}^-$ | $-\text{CO}_2$ | $-\text{CR}_3$ | $\text{R}_2\text{CH}$ | $-\text{D}$ | $-\text{T}$ | $-\text{H}$ |
| a             | b              | c              | d                     | e           | f           | g           |
- (A)  $a > b > c > d > f > e > g$  (B)  $a > c > d > b > e > f > g$
- (C)  $a > c > b > d > f > e > g$  (D)  $a > f > b > d > e > c > g$
5. Which is/are first member of given homologous series ?
- (A) Alkadiene  $\Rightarrow \text{CH}_2=\text{C}=\text{CH}_2$  (B) Alkenyne  $\Rightarrow \text{HC}\equiv\text{C}-\text{CH}=\text{CH}_2$
- (C) Cynide  $\Rightarrow \text{CH}_3\text{CH}_2\text{CN}$  (D) Ketone  $\Rightarrow \text{CH}_3-\overset{\text{O}}{\parallel}{\text{C}}-\text{CH}_3$

6. How many of the following are  $-I$  effect (negative inductive effect) showing groups.

- (a)  $-\text{NO}_2$       (b)  $-\text{CO}_2\text{H}$       (c)  $-\text{CH}_3$       (d)  $-\overset{\oplus}{\text{N}}\text{H}_3$       (e)  $-\text{O}^-$   
(f)  $-\text{CN}$       (g)  $-\text{CMe}_3$       (h)  $-\text{Cl}$       (i)  $-\text{H}$

7. How many groups can show  $-I$  effect ?

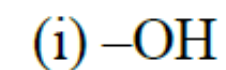
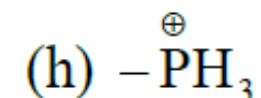
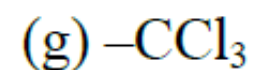
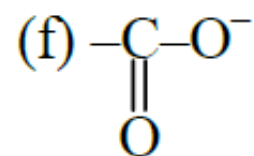
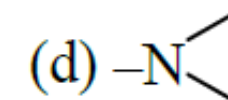
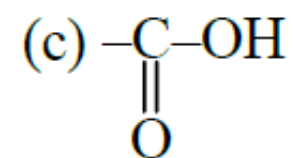
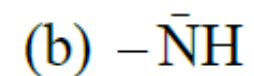
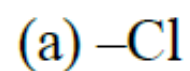
$-\text{CCl}_3$ ,  $-\text{NO}_2$ ,  $-\text{OH}$ ,  $-\text{N}(\text{CH}_3)_2$ ,  $-\text{SO}_3\text{H}$ ,  $-\text{O}^\ominus$ ,  $-\text{CHO}$ ,  $-\text{Cl}$ ,  $-\text{COO}^\ominus$

8. The number of groups showing +I effect = X.

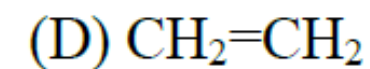
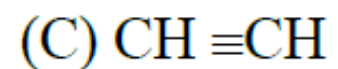
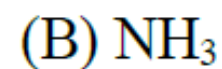
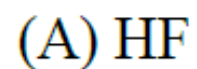
The number of groups showing -I effect = Y.

Find value XY e.g. if X = 3. Y = 4. Answer is 034

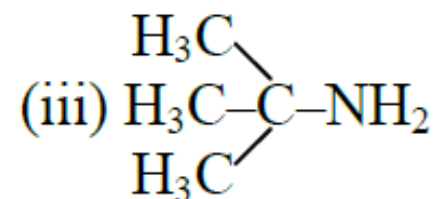
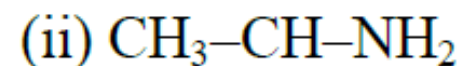
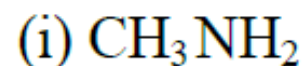
Groups are given as follows :



9. Which of the following is more acidic than  $\text{H}_2\text{O}$  :



10. Which of the following is more basic than  $\text{NH}_3$  in gaseous phase :



(iv) All of above