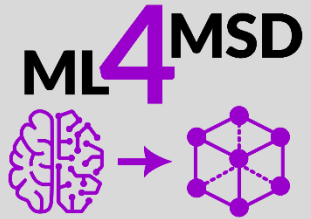


ME 5374-ST



Machine Learning for Materials Science and Discovery

Fall 2025

Asst. Prof. Peter Schindler

Lecture 14 – Atomistic Modeling 2, Density Functional Theory (DFT)

- Quantum Mechanics and Schrödinger Equation
- Hartree-Fock
- Hohenberg-Kohn Theorems
- Kohn-Sham Equation, DFT, and Local Density Approximation (LDA)
- Band Diagrams and Reciprocal Space
- User Choices: Basis Sets, Atomic Potentials, and Exchange-Correlation Functionals
- GPAW and ASE Tutorial

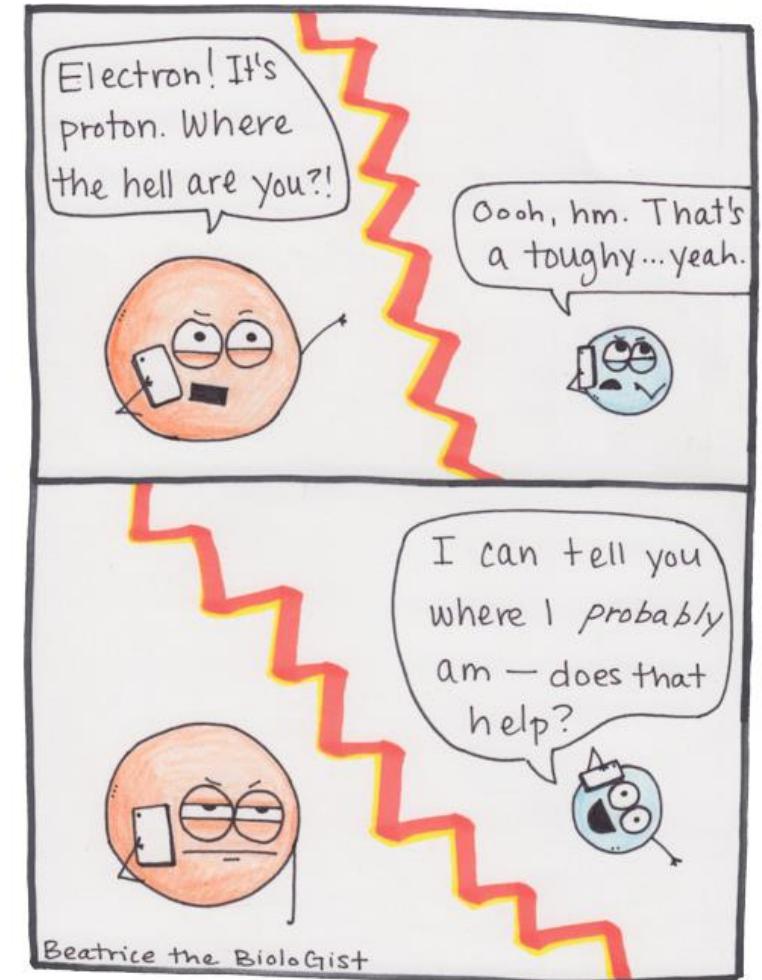
Quantum vs. Classical Mechanics

Classical Mechanics

$$\left. \begin{aligned} \vec{F} &= \dot{\vec{p}} & \vec{p} &= m \cdot \dot{\vec{r}} \\ H &= \frac{\vec{p}^2}{2m} + V \\ \vec{r}(t), \underbrace{\{\vec{r}(t_0), \dot{\vec{r}}(t_0), \ddot{\vec{r}}(t_0), \dots\}}_{\text{initial conditions}} \end{aligned} \right\} \text{Deterministic!}$$

Quantum Mechanics

$$\left. \begin{aligned} \hat{H}\psi(\vec{r}, t) &= i\hbar \frac{\partial}{\partial t} \psi(\vec{r}, t) \\ \vec{p} &\rightarrow \hat{\vec{p}} = -i\hbar \vec{\nabla} \\ \rho(\vec{r}, t) &= |\psi(\vec{r}, t)|^2 \\ \Delta \vec{r} \cdot \Delta \vec{p} &\geq \hbar / 2 \end{aligned} \right\} \text{Probabilistic!}$$



Notation: $\dot{\vec{r}} = d\vec{r}/dt$

Applying Quantum Mechanics to Materials

Schrödinger equation can be solved exactly for Hydrogen (*one* e⁻)

Instead of orbits (Bohr) the solutions represent *electron clouds* (probability distributions) of different shapes → **s,p,d,f orbitals**

$$\hat{H} = -\frac{\hbar^2}{2m} \nabla^2 - \frac{e^2}{4\pi\epsilon r}$$

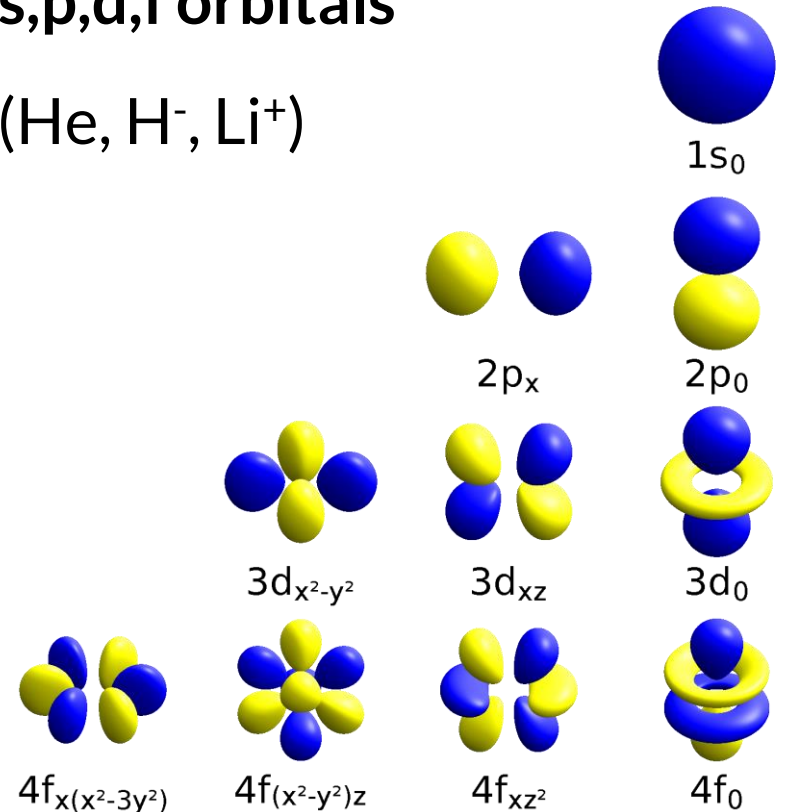
Can also be solved analytically for *two* electrons (He, H⁻, Li⁺)

$$\hat{H} = -\hbar^2 \left[\frac{1}{2\mu} (\nabla_1^2 + \nabla_2^2) + \frac{1}{M} \nabla_1 \cdot \nabla_2 \right] + \frac{e^2}{4\pi\epsilon_0} \left[\frac{1}{|\vec{r}_2 - \vec{r}_1|} - \frac{Z(\vec{r}_1 + \vec{r}_2)}{r_1 r_2} \right]$$

However, *three*-electron system **cannot** be solved analytically

→ *Approximations* necessary

$$\mu = mM / (m + M)$$



Towards Hartree-Fock: Multi-electron Hamiltonian

Multi electron Hamiltonian (N electrons $\{\vec{r}, m_e\}_N$, M atoms $\{\vec{R}, m_n\}_M$)

$$\hat{H} = \underbrace{-\sum_{i=1}^N \frac{\hbar^2 \nabla_{\vec{R}_i}^2}{2m_n}}_{E_{\text{kin, nucl}}} - \underbrace{\sum_{i=1}^N \frac{\hbar^2 \nabla_{\vec{r}_i}^2}{2m_e}}_{E_{\text{kin, el}}} + \underbrace{\sum_{i=1}^N \sum_{j=1}^M \frac{Z_m}{|\vec{r}_i - \vec{R}_j|}}_{\text{nucl - el}} + \underbrace{\sum_{i=1}^N \sum_{j<i}^N \frac{1}{|\vec{r}_i - \vec{r}_j|}}_{\text{el - el}} + \underbrace{\sum_{i=1}^M \sum_{j<i}^M \frac{Z_i Z_j}{|\vec{R}_i - \vec{R}_j|}}_{\text{nucl - nucl}}$$

$$\hat{H}\Phi(\vec{r}_1, \dots, \vec{r}_N, \vec{R}_1, \dots, \vec{R}_M; t) = i\hbar \frac{\partial}{\partial t} \Phi(\vec{r}_1, \dots, \vec{r}_N, \vec{R}_1, \dots, \vec{R}_M; t)$$

Mass of nucleus is at least **1800 times** greater than mass of electron

Like: Mass of



to mass of



Towards Hartree-Fock

Multi electron Hamiltonian (N electrons $\{\vec{r}, m_e\}_N$, M atoms $\{\vec{R}, m_n\}_M$)

$$\hat{H} = \underbrace{-\sum_{i=1}^N \frac{\hbar \nabla_{\vec{R}_i}^2}{2m_n}}_{E_{\text{kin, nucl}}} - \underbrace{\sum_{i=1}^N \frac{\hbar \nabla_{\vec{r}_i}^2}{2m_e}}_{E_{\text{kin, el}}} + \underbrace{\sum_{i=1}^N \sum_{j=1}^M \frac{Z_m}{|\vec{r}_i - \vec{R}_j|}}_{\text{nucl - el}} + \underbrace{\sum_{i=1}^N \sum_{j<i}^N \frac{1}{|\vec{r}_i - \vec{r}_j|}}_{\text{el - el}} + \underbrace{\sum_{i=1}^M \sum_{j<i}^M \frac{Z_i Z_j}{|\vec{R}_i - \vec{R}_j|}}_{\text{nucl - nucl}}$$

$$\hat{H}\Phi(\vec{r}_1, \dots, \vec{r}_N, \vec{R}_1, \dots, \vec{R}_M; t) = i\hbar \frac{\partial}{\partial t} \Phi(\vec{r}_1, \dots, \vec{r}_N, \vec{R}_1, \dots, \vec{R}_M; t)$$

Born-Oppenheimer approximation

$$\Phi(\{\vec{r}\}_N, \{\vec{R}\}_M; t) = \underbrace{\chi(\{\vec{R}\}_M; t)}_{\text{classically}} \cdot \underbrace{\Psi(\{\vec{r}\}_N, \{\vec{R}\}_M)}_{\text{fixed } \{\vec{R}\}_M}$$

$$\left[-\sum_{i=1}^N \frac{\hbar \nabla_i^2}{2m} + \sum_{i=1}^N V_{\text{nuc}}(\vec{r}_i) + \sum_{i=1}^N \sum_{j<i}^N \frac{1}{|\vec{r}_i - \vec{r}_j|} \right] \Psi(\vec{r}_1, \dots, \vec{r}_N) = E \cdot \Psi(\vec{r}_1, \dots, \vec{r}_N)$$

Hartree-Fock

Hartree approximation: Electrons are **uncorrelated**

$$\Psi(\vec{r}_1, \vec{r}_2, \dots, \vec{r}_N) = \psi_1(\vec{r}_1) \cdot \psi_2(\vec{r}_2) \cdot \dots \cdot \psi_N(\vec{r}_N)$$

N single electron equations (Hartree equ.) – **Mean field** approx.

$$\left[-\frac{\hbar^2 \nabla^2}{2m} + V_{\text{nuc}}(\vec{r}) + \sum_{j \neq i}^N \int d\vec{r}' \frac{|\psi_j(\vec{r}')|^2}{|\vec{r} - \vec{r}'|} \right] \psi_i(\vec{r}) = E_i \cdot \psi_i(\vec{r})$$

Missing: Electron correlation and Fermion **exchange interaction**

$$\Psi(\vec{r}_1, \vec{r}_2) = \psi_1(\vec{r}_1) \cdot \psi_2(\vec{r}_2) = \psi_2(\vec{r}_2) \cdot \psi_1(\vec{r}_1) \neq -\Psi(\vec{r}_2, \vec{r}_1)$$

$$\Psi(\vec{r}_1, \vec{r}_2 = \vec{r}_1) = \psi_1(\vec{r}_1) \cdot \psi_2(\vec{r}_1) \neq 0 \quad \text{Violates Pauli exclusion}$$

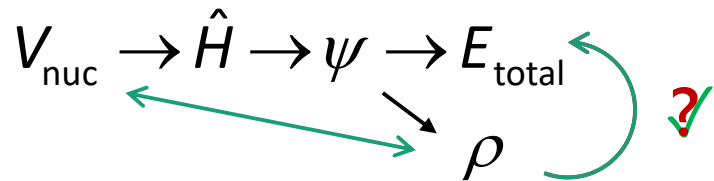
Slater determinant adds exchange **exactly**
(but not correlation)

$$\Psi(\vec{r}_1, \vec{r}_2) = \frac{1}{\sqrt{2}} \begin{vmatrix} \psi_1(\vec{r}_1) & \psi_2(\vec{r}_1) \\ \psi_1(\vec{r}_2) & \psi_2(\vec{r}_2) \end{vmatrix}$$

Performance of Hartree-Fock

- **Binding energies** for materials generally *underestimated* by 10-50%
→ due to *neglected correlation effects*
- Despite that, **geometries** for molecules/solids are pretty ok (few % deviations)
- **Bandgaps** typically off by factor of order unity
- The exact exchange of HF calculations make a good starting point for **perturbative methods** to include correlation effects
- Since HF calculations are usually *less accurate* than DFT and require equal or greater computational effort
→ Materials calculations typically employ DFT

Hohenberg-Kohn Theorems



1. Electron density *uniquely* defines multi-electron system:

$$E_{\text{total}} = E[\rho(\vec{r})]$$

2. $\min[E[\rho(\vec{r})]] = E[\rho_0(\vec{r})]$ where ρ_0 = ground state density

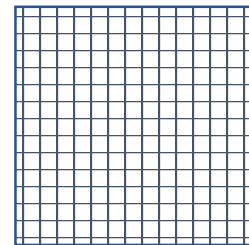
Profound feature: *Calculation of wavefunction not required*

Number of parameters needed for

p^{3N} N -electron wavefunction

p^3 electron density

Np^3 N single-electron wavefunctions



Grid with p points
in each dimension

Overcoming the Exponential Wall

For example, look at **Oxygen** atom (8 electrons)

$\Psi(\vec{r}_1, \vec{r}_2, \vec{r}_3, \vec{r}_4, \vec{r}_5, \vec{r}_6, \vec{r}_7, \vec{r}_8)$ depends on 24 coordinates in 3D

Let's do a rough sampling of the *multi-electron wavefunction*

10 entries / coordinate \longrightarrow 10^{24} entries

1 Byte / entry \longrightarrow 10^{24} Bytes

1 TB = 10^{12} Bytes / HDD \longrightarrow 10^{12} Hard disk drives

0.5 kg / HDD \longrightarrow $5 \cdot 10^8$ tons of HDDs
(500 million tons $\sim 10 \times$ Great Wall)

Equivalent sampling for *KS single-electron wavefunctions*:

8 independent coordinates in 3D $\rightarrow 8 \cdot 10^3$ Bytes = **8 kB**

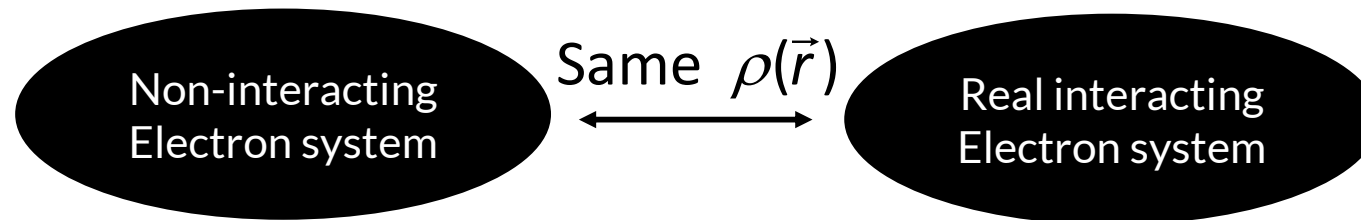
Density Functional Theory (DFT) & Kohn-Sham Equations

But the kinetic energy (as a functional of density)
→ requires *exact multi-electron* wavefunction



Instead

introduce fictitious, *single-electron*, uncorrelated electron states (like in HF)
that give the **same total electron density** as the exact multi-electron state.



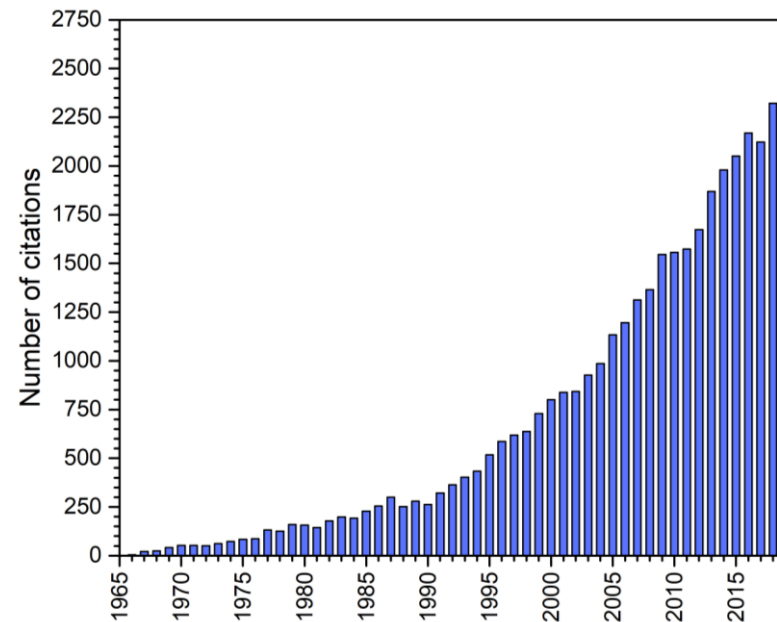
$$\left[-\frac{\hbar^2 \nabla^2}{2m} + V_{\text{nuc}}(\vec{r}) + \sum_j \int d\vec{r}' \frac{\rho(\vec{r}')}{|\vec{r} - \vec{r}'|} + \underbrace{\frac{\delta E_{\text{xc}}[\rho(\vec{r})]}{\delta \rho(\vec{r})}}_{V_{\text{xc}}[\rho(\vec{r})]} \right] \psi_i(\vec{r}) = E_i \cdot \psi_i(\vec{r})$$

LDA: The True Success of DFT

In principle, KS equations are *exact*. However, form of XC in terms of density remains *unknown* → necessary to use **approximations**

Simplest approach for exchange-correlation functional:
Consider a **uniform electron gas** of density ρ

$$E_{\text{XC}}^{\text{LDA}}[\rho(\vec{r})] = \int d\vec{r} \, \rho(\vec{r}) \epsilon_{\text{XC}}[\rho(\vec{r})]$$



“Self-consistent equations including Exchange and Correlation Effects”
W. Kohn and L. J. Sham,
Phys. Rev. 140 (1965)

LDA: Trends and Limitations

- Favors homogeneous density: *Overestimates* binding energies (unlike HF)
- *Bond lengths, bond angles, vibration frequencies agree* with experiment to within a few percent. Bond lengths generally too small (overbound)
- *More accurate* than Hartree-Fock!
- Lack of **long-range correlations**: Unable to describe *London* or *dispersive interactions* (van der Waals)
- **Self-interaction** of electrons problematic:
 - **Localized states** (*d, f*-states) are not described very well
 - **Band gaps** *too small* (factor ~ 2) \rightarrow optical abs. spectra wrong & dielectric constant too large.
- Works *well* for **covalent, ionic bonds** but *not well* for weak bonds (Hydrogen)
- **Wrong asymptotic behavior** of the effective potential: decays exponentially rather than $1/r \rightarrow$ no Rydberg series in atoms/molecules and negatively charged ions unstable
- Cohesive energies, elastic properties, bulk modulus up to $\sim 20\%$ error

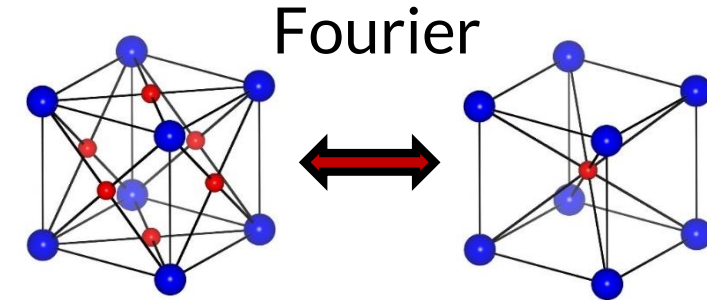
Reciprocal Space and Brillouin Zone

Lattice basis vectors: $\vec{a}, \vec{b}, \vec{c}$

Reciprocal lattice (k -space)
basis vectors are given by

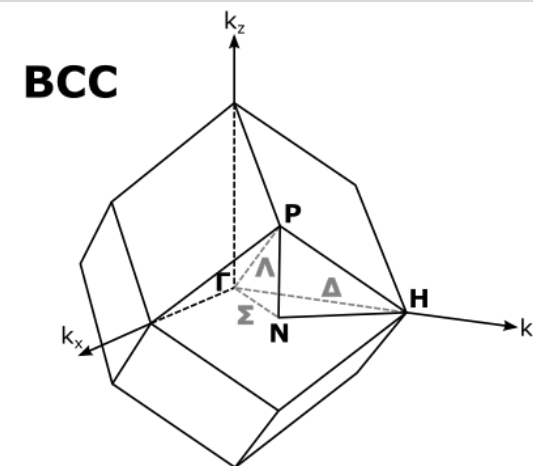
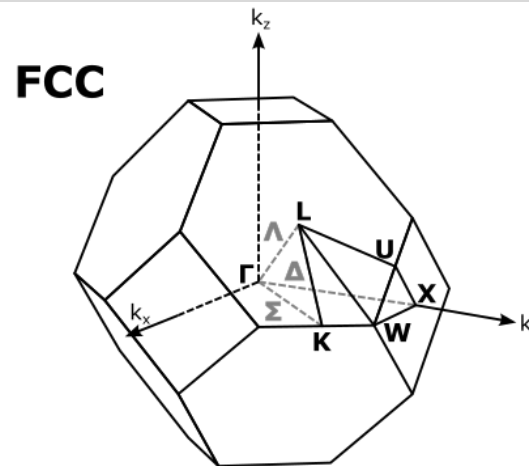
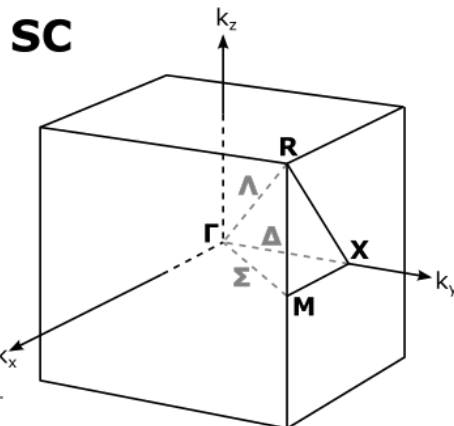
$$\vec{a}^* = \frac{2\pi}{V}(\vec{b} \times \vec{c}) \quad \vec{b}^* = \frac{2\pi}{V}(\vec{a} \times \vec{c}) \quad \vec{c}^* = \frac{2\pi}{V}(\vec{a} \times \vec{b})$$

where $V = \vec{a} \cdot (\vec{b} \times \vec{c})$



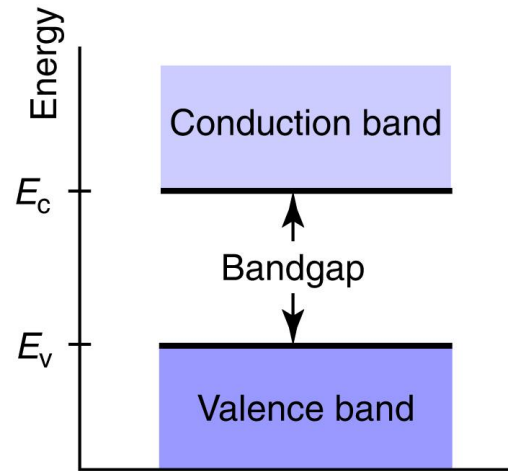
Analogy in signal processing
→ time vs. frequency domain

1st Brillouin Zone = Wigner-Seitz cell in reciprocal space



Band Diagrams

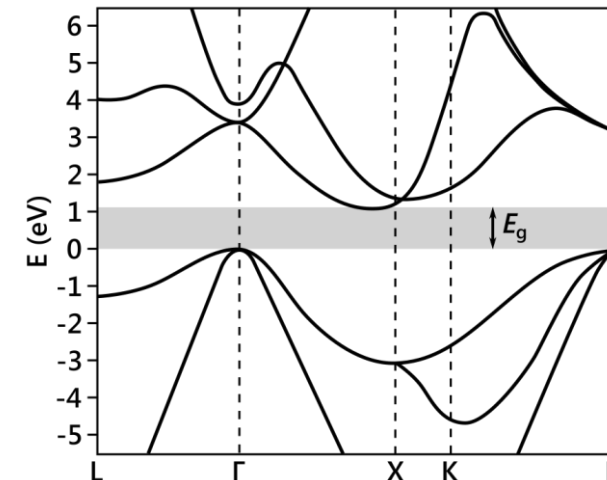
Semiconductor Devices



Space coordinate

Sufficient for
understanding most
electronic devices

DFT / Optoelectronics



k -space coordinate

Detailed band diagram,
essential to understand
opto-electronic devices

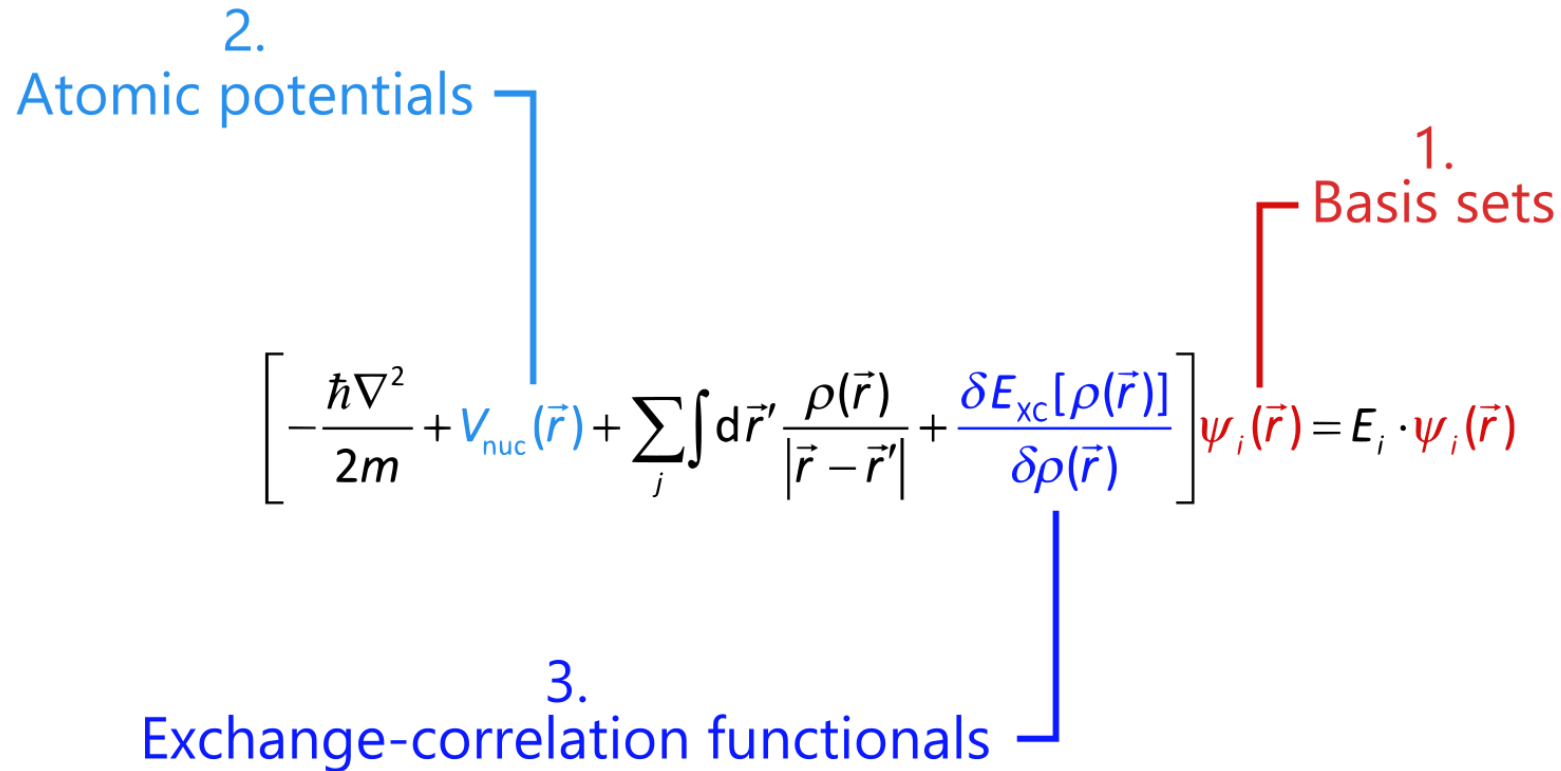
Overview of DFT Knobs to Turn

2. Atomic potentials

1. Basis sets

$$\left[-\frac{\hbar^2 \nabla^2}{2m} + V_{\text{nuc}}(\vec{r}) + \sum_j \int d\vec{r}' \frac{\rho(\vec{r}')}{|\vec{r} - \vec{r}'|} + \frac{\delta E_{\text{xc}}[\rho(\vec{r})]}{\delta \rho(\vec{r})} \right] \psi_i(\vec{r}) = E_i \cdot \psi_i(\vec{r})$$

3. Exchange-correlation functionals



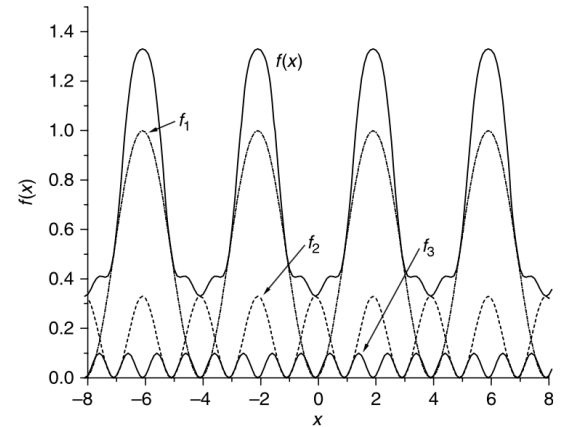
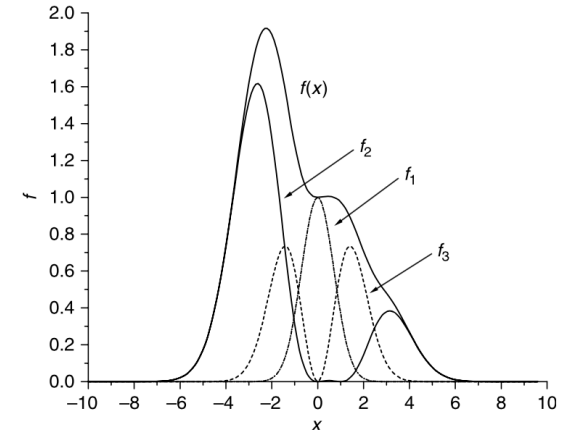
1. Basis Sets

Localized (centered around atoms)

- Resemble atomic orbitals (e.g., Gaussians, Slater)
- Describes *molecules* well/efficiently
→ mainly used by chemists

Extended (plane waves)

- Complete basis set
- Periodic boundary conditions (PBC)
- Describes bulk *crystal* structures well/efficiently
→ mainly used by physicists & material scientists



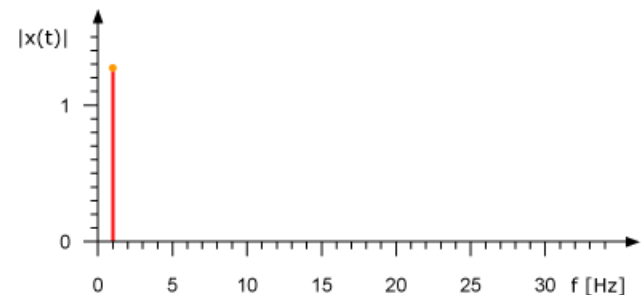
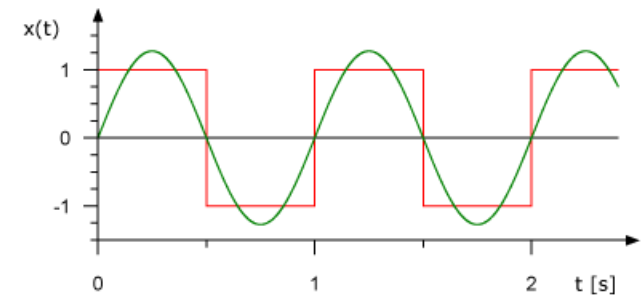
Plane Waves: Pros & Cons

+

- E_{cutoff} only parameter to *systematically improve* basis set
- Same basis set for molecules and solids
- No finite-basis set errors on forces (“*Pulay forces*”)
- Established, highly optimized algorithms (FFTs, efficient charge-density mixing, etc.)

-

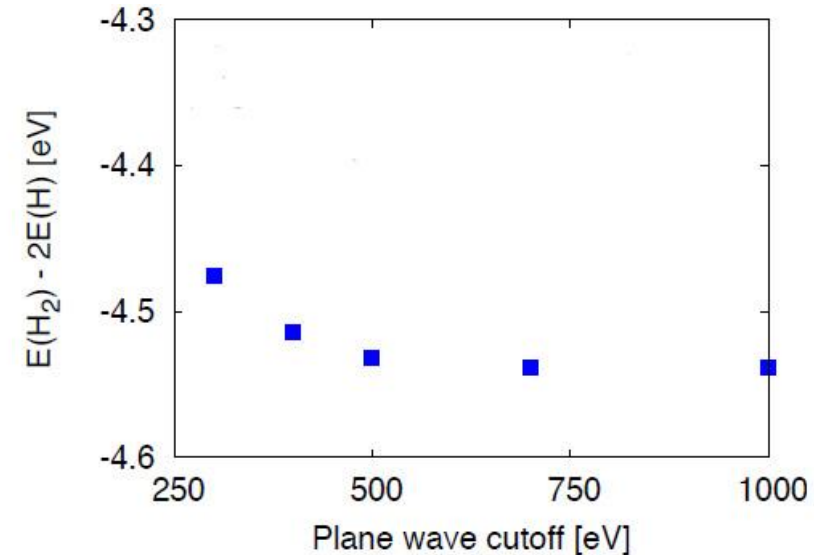
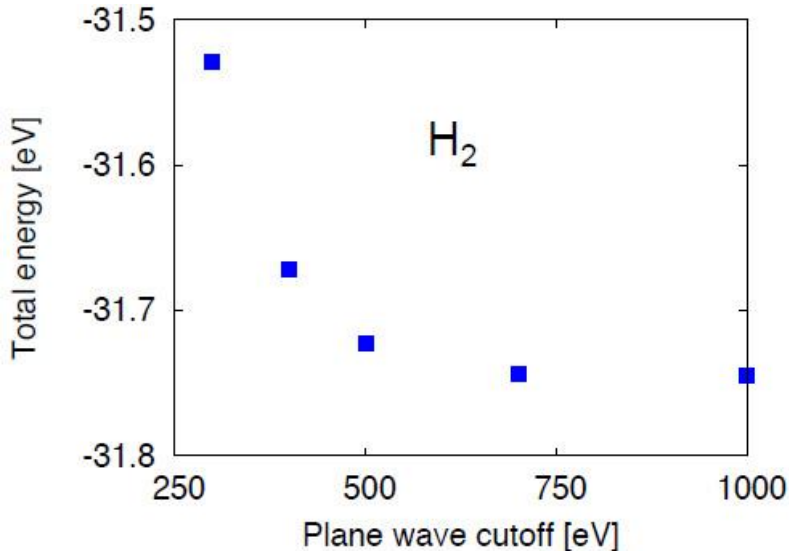
- Pseudopotentials needed (→ not all-electron)
- Large basis sets required (especially for vacuum)



Convergence (w.r.t. E_{cutoff})

Calculation of **energy differences** using the same calculation parameters (XC functional, k -point mesh, PW cutoff, PPs) typically *converge faster*

This is due to *systematic cancellation of errors* associated with lack of total convergence of BZ integration, PW cutoff, etc.

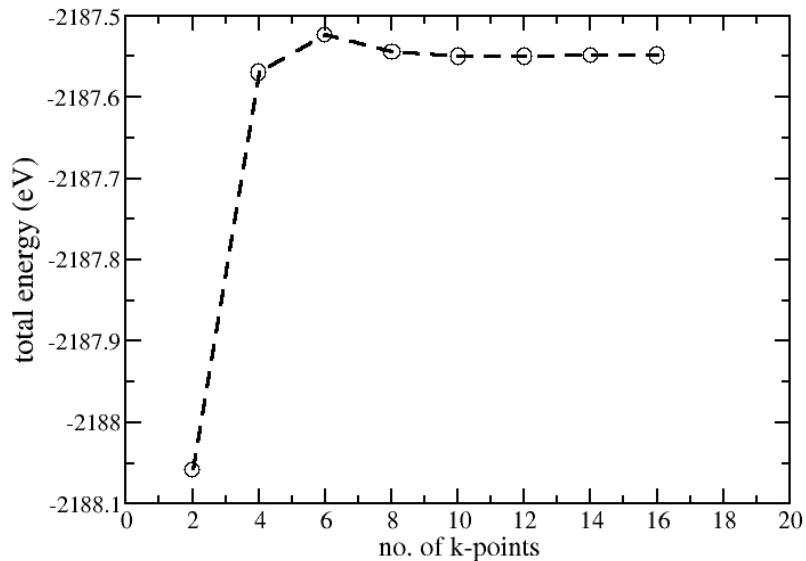


Convergence (w.r.t. k -points)

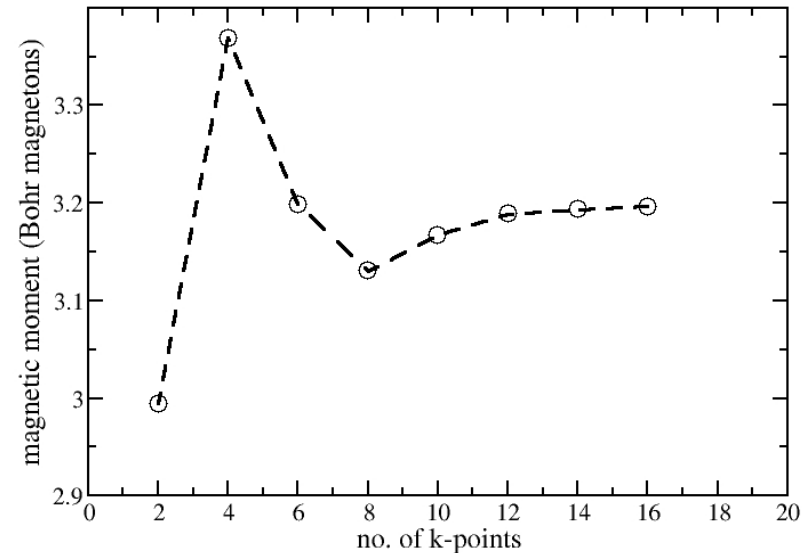
Convergence is not necessarily monotonous

Convergence of magnetic moment or forces may require denser k -point sampling than energy

k-point test for hexagonal Co with 30Ry cutoff and $a=2.49\text{\AA}$



Magnetic moment vs. k -point sampling for hex Co with $a=2.49\text{\AA}$



2. Pseudopotentials

Electrons divided into

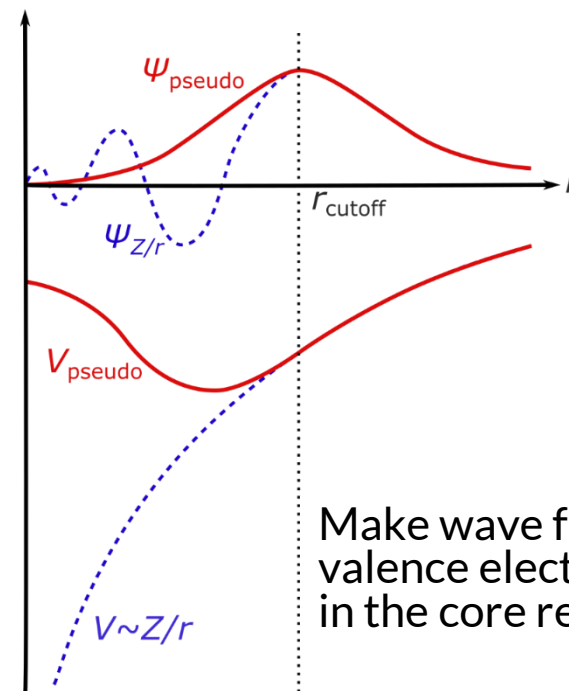
- *Core electrons* – chemically **inactive**; ignore explicit description
- *Valence electrons* – chemically **active**; include screening of nucleus

Valence states also require high energy cutoffs near the core region

>10,000 plane waves required versus 10s of Gaussians per atom

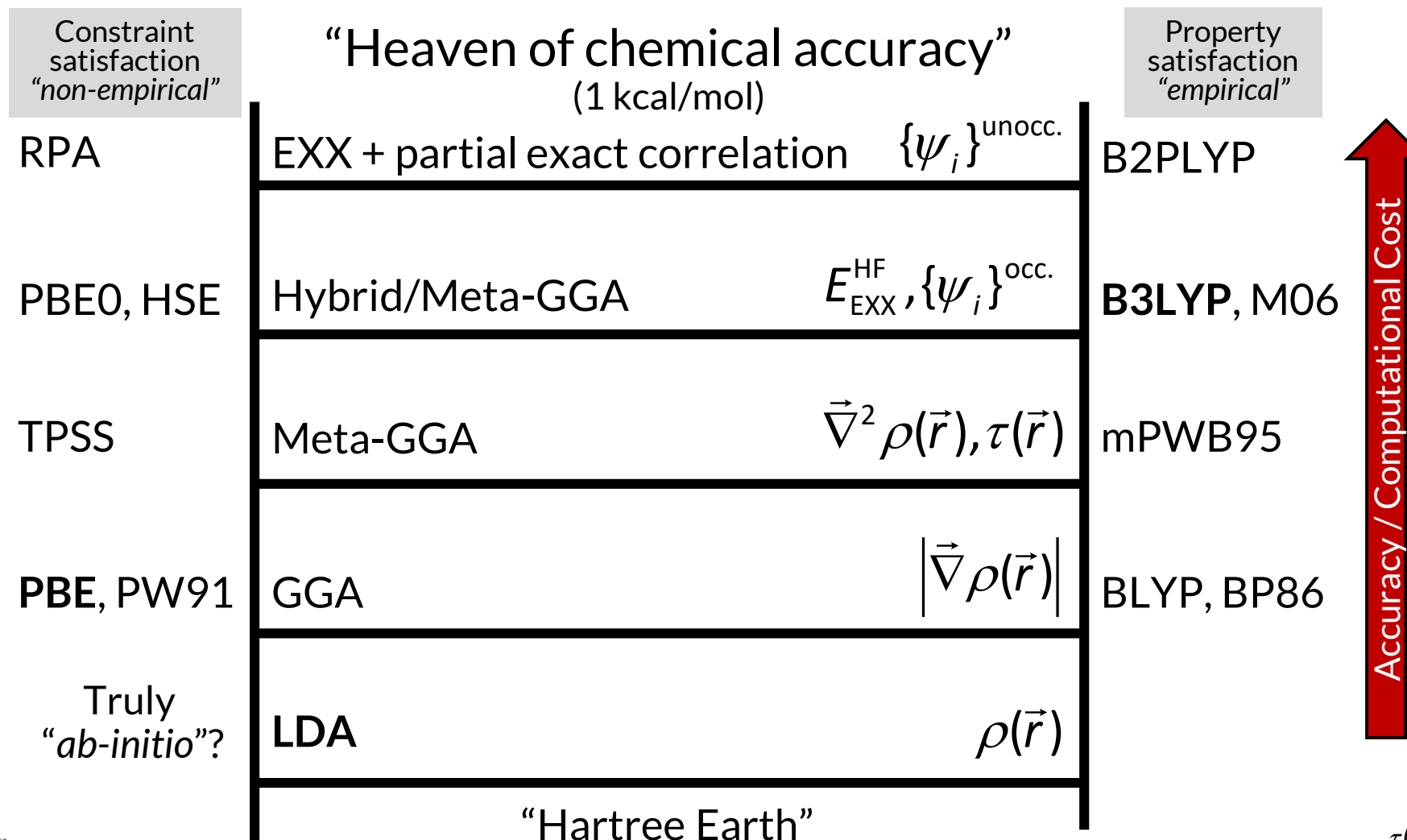
“*Ab initio*” potentials determined from an *all electron calculation* for a single atom.

Valence eigenvalues and wave function properties at boundary **reproduced by pseudopotential**.



Make wave functions of valence electrons **soft** in the core region

3. XC Functionals – Jacob’s Ladder



$$\tau(\vec{r}) = \sum_i^N |\vec{\nabla} \psi_i(\vec{r})|^2$$

Accuracy of Common XC Functionals

Mol.	Exp.	UHF	LSDA	PBE	MGGA	B3LYP	MP2
H ₂	174.5	133.9	180.3	166.7	182.5	—	165.7
Li ₂	39.3	4.8	37.9	31.7	35.9	33.5	25.5
Be ₂	4.8	11.2	20.6	15.6	7.2	—	−1.6
N ₂	364.0	183.3	427.1	387.6	365.2	365.6	368.1
F ₂	62.1	−15.9	124.6	85.1	68.8	57.7	111.6
LiH	92.4	52.6	96.9	85.2	93.1	92.9	86.1
OH	169.6	108.4	197.9	175.0	171.8	172.3	165.7
HF	225.7	154.6	259.1	226.3	221.0	222.1	227.9
H ₂ O	370.0	245.4	424.9	373.2	366.7	368.1	366.5
NH ₃	473.9	318.7	537.5	480.8	476.2	478.4	462.1
CH ₄	668.2	522.7	737.2	669.0	671.1	670.4	661.3
CO	413.2	277.3	476.3	428.4	408.0	408.3	423.9
NO	243.7	84.5	316.2	273.9	252.6	248.0	242.2
Cl ₂	92.4	—	132.1	103.7	94.7	87.8	—
Mean average error:		−93.8	41.1	7.7	1.5	0.8	0.3
RMS error:		109.8	48.4	14.0	4.6	0.7	15.6
Computational cost:		N^4	N^3	N^3 – N^4	N^3 – N^4	N^4	N^5

Limitations of DFT

- Choosing best **XC functional** depends on property/application
- Typically, computational costs *limit unit cells* to (few) 100 atoms
- Does not directly describe electronic *excitations*
- Tends to **underestimate bandgaps** due to self-interaction error and over-delocalized wavefunctions
- Does not reproduce *non-adiabatic* effects (Born-Oppenheimer)
- Problems arise when *accurate* xc-effects are *crucial* like in strongly correlated systems
- Incorrectly describes (or not at all) *relativistic* effects

Ab initio Molecular Dynamics (AIMD)

Hellmann-Feynman theorem

$$\frac{dE_{\lambda}}{d\lambda} = \langle \psi_{\lambda} | \frac{dH_{\lambda}}{d\lambda} | \psi_{\lambda} \rangle$$






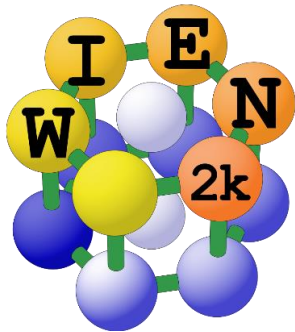






Gives us forces directly from the electron density:

$$F_{x,i} = -\frac{\partial E}{\partial X_i} = -\langle \psi | \frac{dH}{dX_i} | \psi \rangle = Z_i \left(\int d\vec{r} \rho(\vec{r}) \frac{x - X_i}{|\vec{r} - \vec{R}_i|^3} - \sum_{\alpha \neq i}^M Z_{\alpha} \frac{X_{\alpha} - X_i}{|\vec{R}_{\alpha} - \vec{R}_i|^3} \right)$$

These forces do **not** require parametrization like classical molecular dynamics!

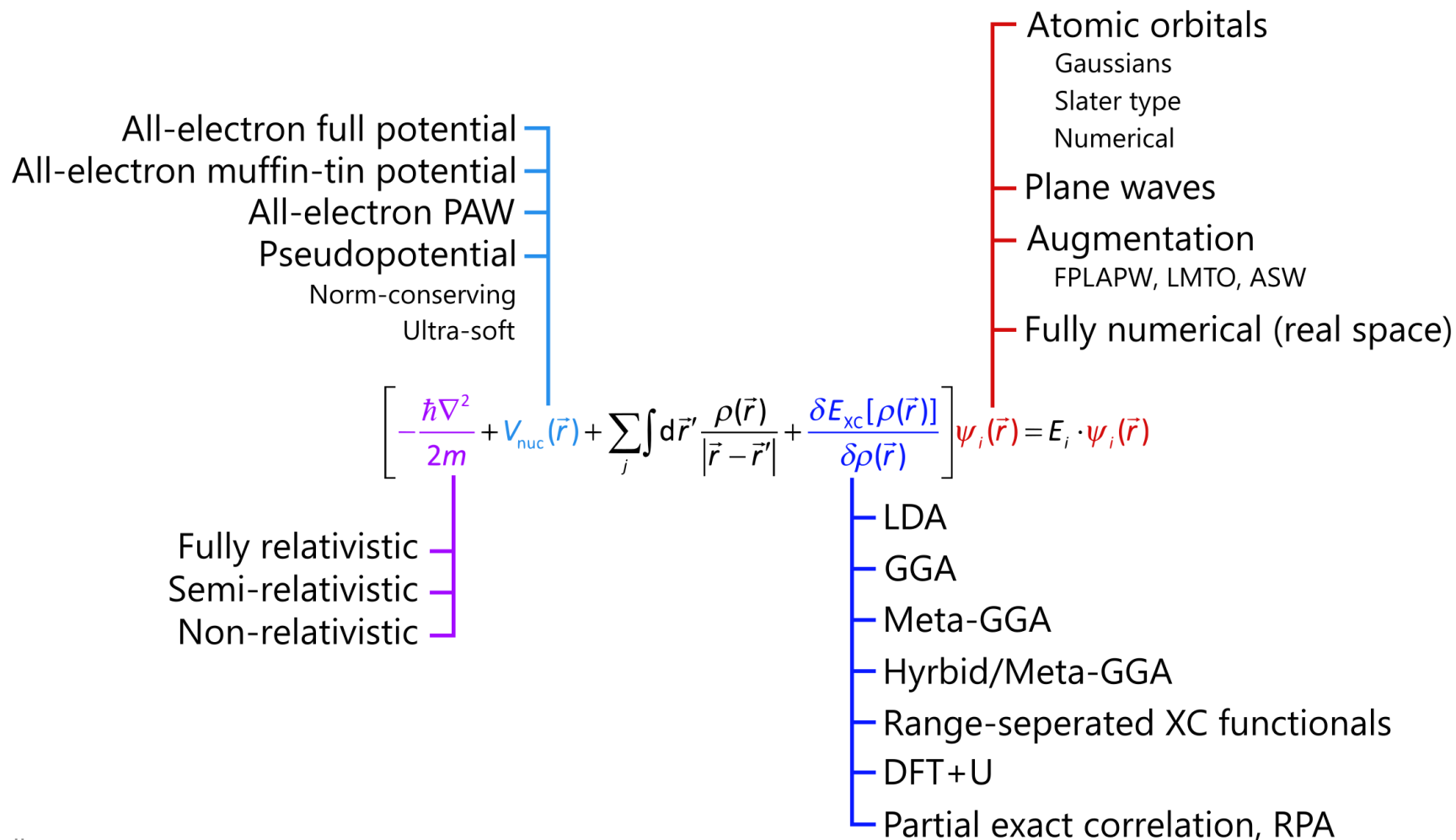
→ “*Ab initio Molecular Dynamics*”

Overview of (some) DFT Codes

Periodic			Molecular
Plane waves	All-electron (augmented)	Local orbitals	
    Dacapo	 	 	  Gaussian  Turbomole 

Remark: Rough classification;
most codes fall into several categories

Overview of DFT Variants (and Zoo of Acronyms)

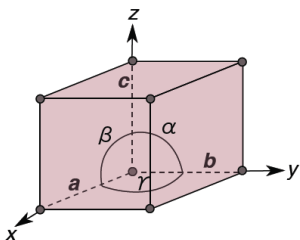


DFT Property Predictions

$$\left[-\frac{\hbar^2 \nabla^2}{2m} + V_{\text{nuc}}(\vec{r}) + \sum_j \int d\vec{r}' \frac{\rho(\vec{r}')}{|\vec{r} - \vec{r}'|} + \frac{\delta E_{\text{xc}}[\rho(\vec{r})]}{\delta \rho(\vec{r})} \right] \psi_i(\vec{r}) = E_i \cdot \psi_i(\vec{r}) \quad \text{“ab-initio”}$$

Structural

Lattice constants
Bond length

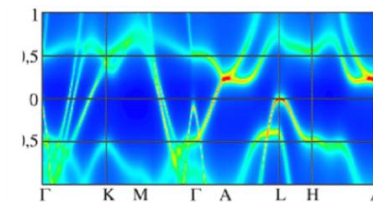


Mechanical

Bulk modulus
Stress tensor σ_{ij}

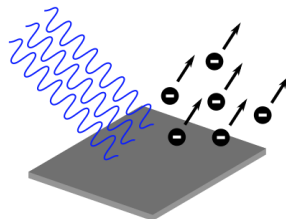
Optical / Electrical

Dielectric constant
Absorption spectra
Density of states
Band structure



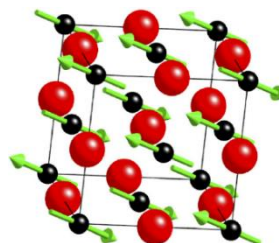
Surface

Work function
Surface/cleavage energy
Adsorption energy



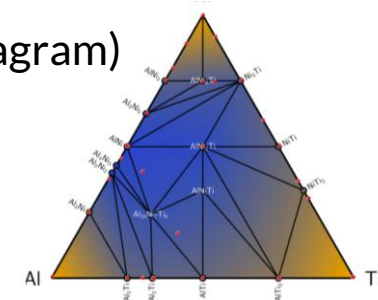
Magnetic

Magnetic ordering
Magnetic moment



Thermodynamic

Vibrational entropy
Phase stability (Hull diagram)



Practical and High-throughput DFT



Writing
DFT input
files by hand



Using
Python

Tutorial: GPAW and ASE

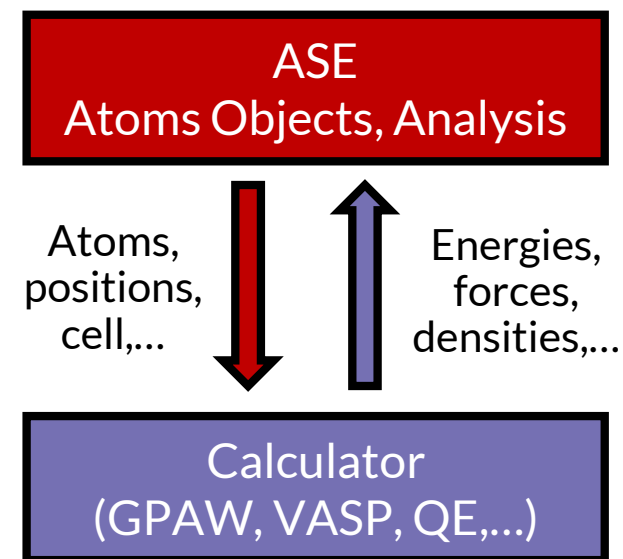
Atomic Simulation Environment (ASE) is a *Python package* for

- Building **atomic structures**
- Structure optimization and molecular dynamics
- **Analysis** and visualization

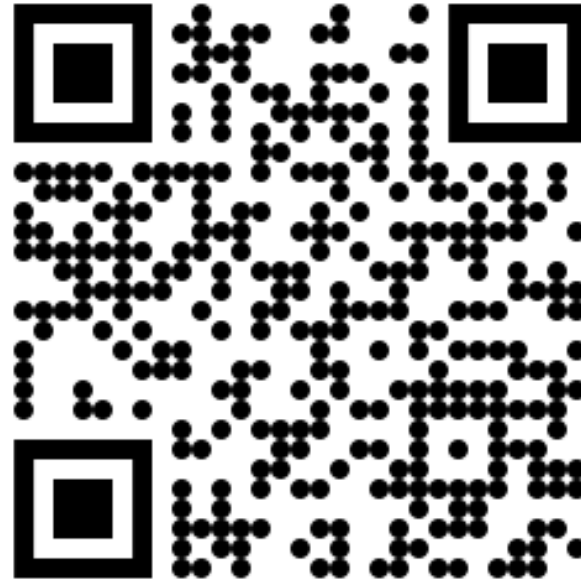
ASE relies on *external calculator* which provides total energies, forces, etc.

- **GPAW**, VASP, QE, Castep, Abinit, etc.
- Input files are *Python scripts*
(simple format, yet powerful!)

Has simple graphical user interface (**ase gui**)



Lecture Feedback



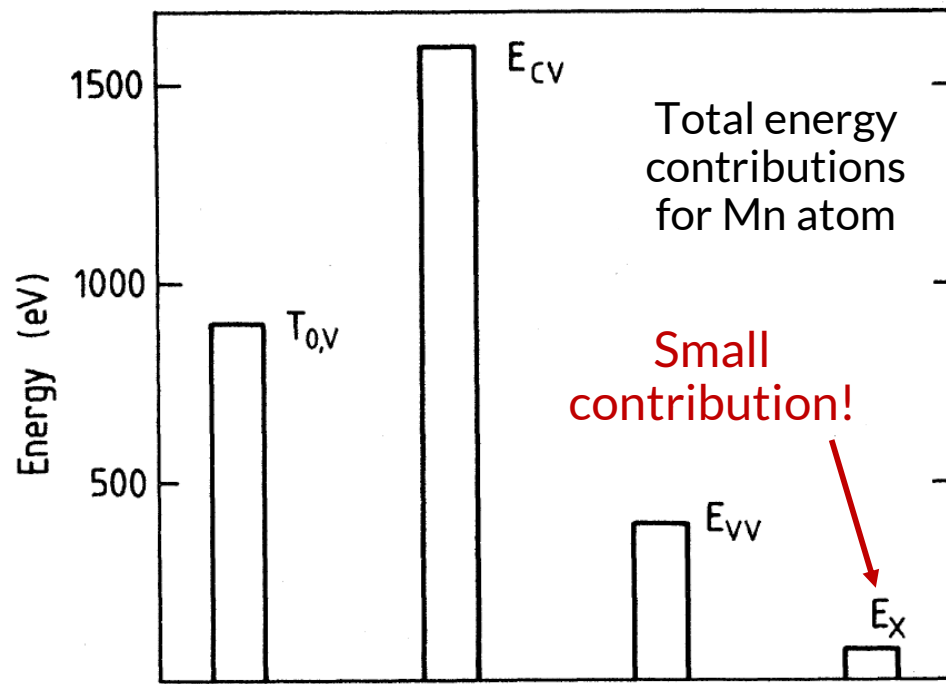
Please, scan the QR code and take a minute to let me know how the lecture was and mention any **feedback/questions**

This form is **anonymous!**

Supplementary Slides

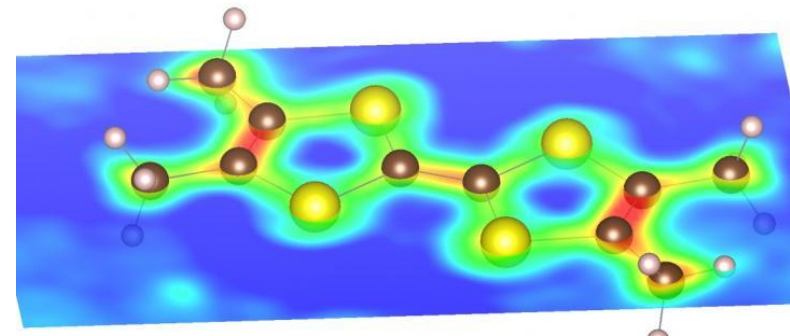
But why Does it Work so well?

Works surprisingly well, because **valence electrons** (which accounts for bonding) often have *slowly varying densities*



But, kinetic part!

$$\left[-\frac{\hbar^2 \nabla^2}{2m} + V_{\text{nuc}}(\vec{r}) + \sum_j \int d\vec{r}' \frac{\rho(\vec{r}')}{|\vec{r} - \vec{r}'|} + \frac{\delta E_{\text{xc}}[\rho(\vec{r})]}{\delta \rho(\vec{r})} \right] \psi_i(\vec{r}) = E_i \cdot \psi_i(\vec{r})$$

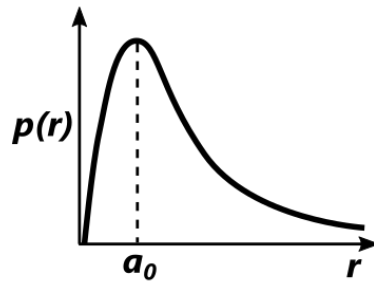


Jones, Gunnasson, Rev. Mod. Phys. 61 (1989)

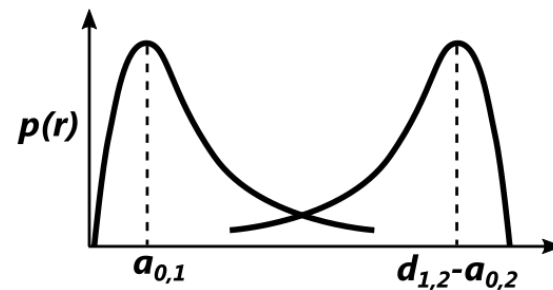
Credit: Shunsuke Kitou

Energy Band Diagrams - Qualitative

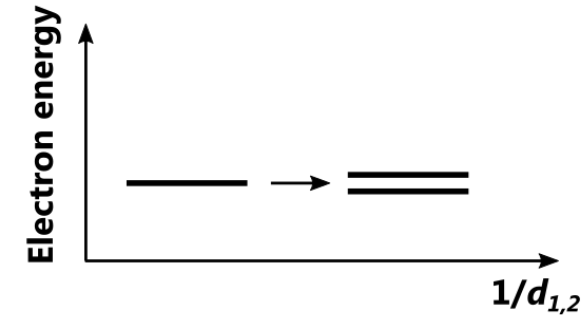
Probability distr. 1-atom



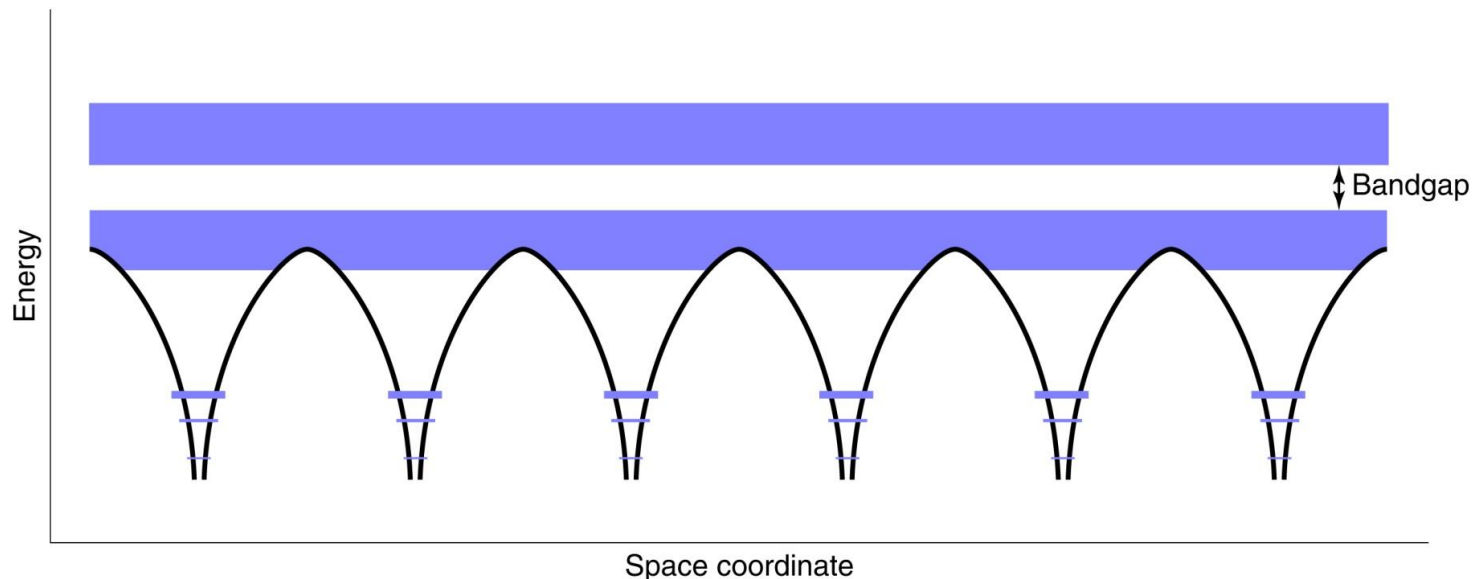
Overlap of 2 atoms



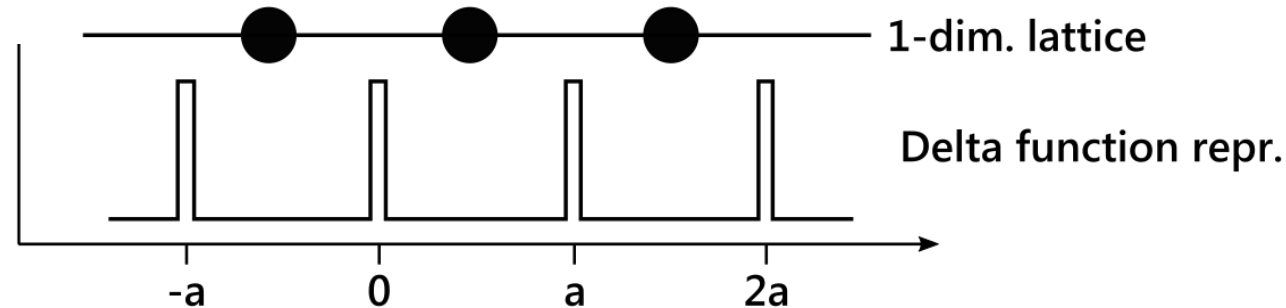
Overlap causes energy level split



Pauli's exclusion principle - One electron (Fermion) per quantum state.



Bloch Waves

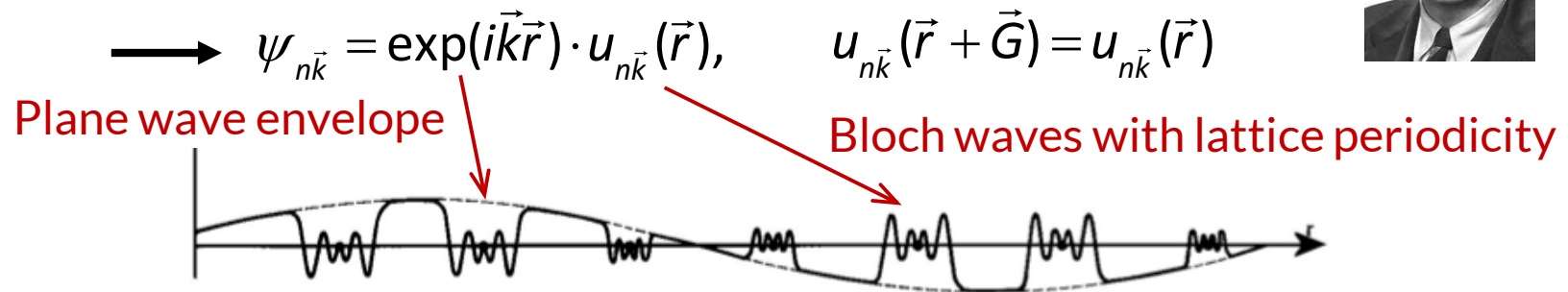


$$-\frac{\hbar^2}{2m_0} \frac{d^2 \psi(x)}{dx^2} + U(x)\psi(x) = E\psi(x)$$

Potential has lattice symmetry $U(x+a) = U(x)$

Bloch theorem (Felix Bloch, 1928):

Periodicity of lattice potential $U(\vec{r} + \vec{G}) = U(\vec{r})$

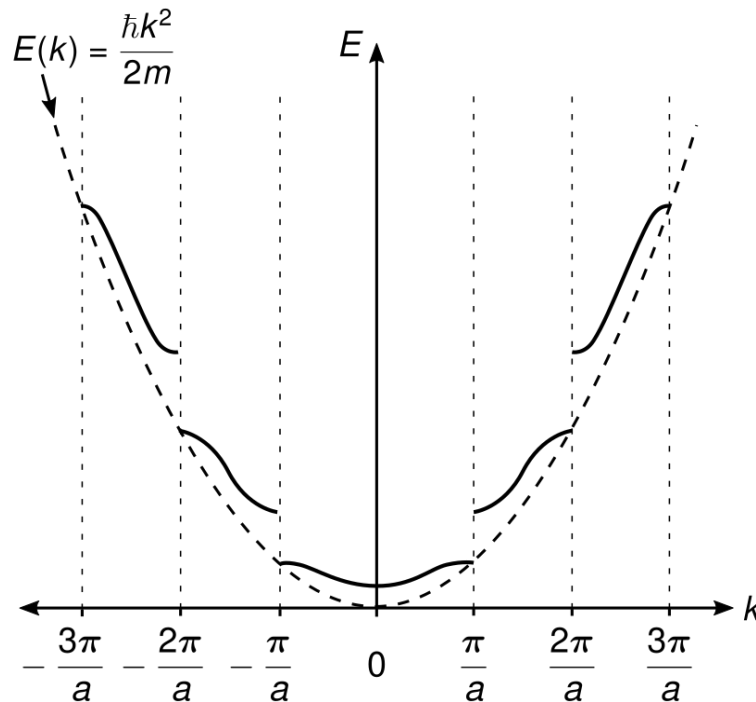


Band Diagrams – Quantitative

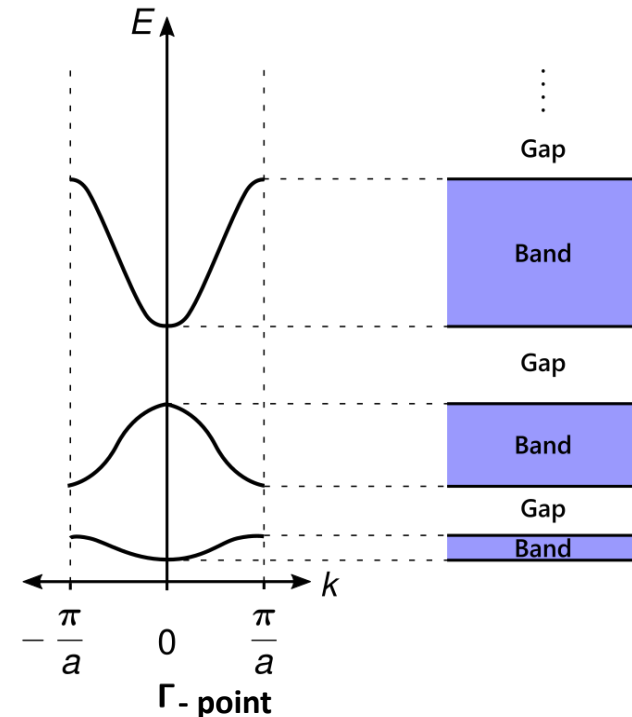
Solving Schrödinger equation gives dispersion relationship: E vs. wave vector k

For free electron:
$$E(k) = \frac{\hbar^2 k^2}{2m}$$

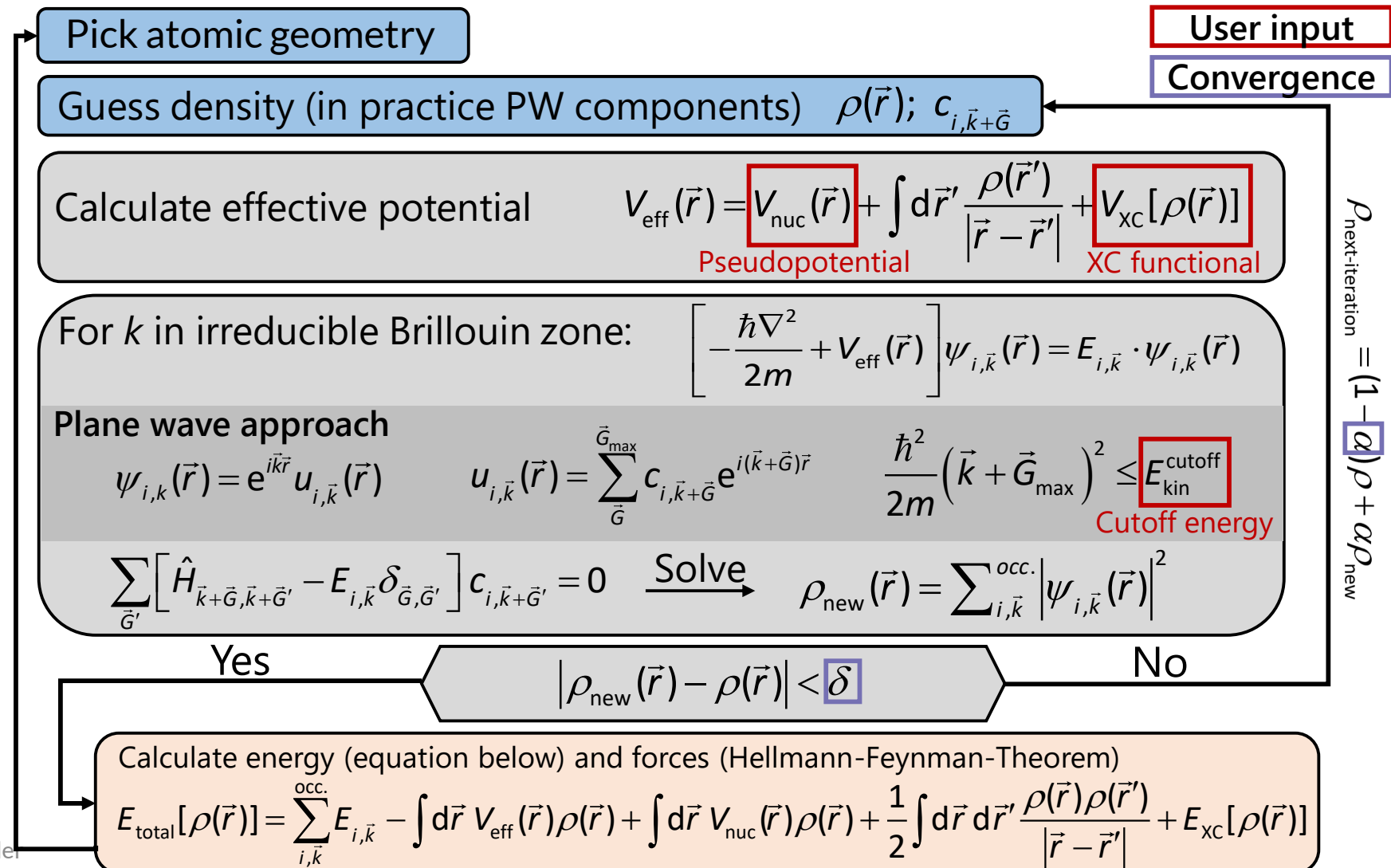
Dispersion relationship $E(k)$ for 1-dim. lattice



Reduced zone scheme
(1st Brillouin zone)



DFT Calculation Procedure – Self-Consistent Field (SCF)



Relativistic Effects

Electrons deep in the core region can exhibit kinetic energies sufficient to produce relativistic effects.

These effects are generally not important for $Z < 14$ (silicon)

Scalar relativistic effects

Can be incorporated into pseudopotentials from relativistic all electron calculation

Spin-orbit interactions

These occur when valence electron spins interact with magnetic fields generated by relativistic core states

These can be incorporated within a spin-density functional theory approach

Types of Pseudopotentials

Pseudopotentials are

- not unique (e.g., optimizable smoothness)
- generally energy- and l -dependent (non-local)

Norm-conserving [Hamann, Schluter, Chiang, 1979]

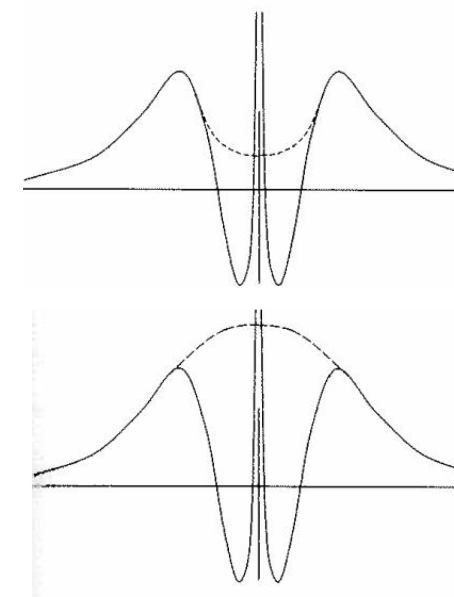
- Generally most *accurate* (transferrable)
but most *expensive* (high # of PWs required)

Ultra-soft [Vanderbilt, 1990]

- Requires *lower cutoffs* for first shell atoms
- No norm conservation – the missing *core charge density* must be specified along with the PP

Projector Augmented Wave (PAW) [Blochl, 1994]

- Includes core states as “frozen-core” for all electron calculations
- Similar to ultrasoft: *Linearized version* of PAW



A Look at a Pseudopotential File

Not important for
Z < 14 (silicon)

l dependency

```
<PP_INFO>
Generated using Vanderbilt code, version 7 3 6
Author: kfg      Generation date: 1 4 15
Automatically converted from original format
1 The Pseudo was generated with a Scalar-Relativistic Calculation
1.600000000000E+00 Local Potential cutoff radius
nl pn l occ Rcut Rcut US E pseu
4S 4 0 2.00 0.000000000000 1.500000000000 -7.27487498300
4P 4 1 6.00 0.000000000000 1.500000000000 -4.59453496500
4D 4 2 10.00 0.000000000000 1.700000000000 -0.83552577500
5S 5 0 0.50 0.000000000000 1.500000000000 -0.56705074500
5P 5 1 0.00 0.000000000000 1.500000000000 -0.24577891300
</PP_INFO>
```

```
<PP_HEADER>
0 Version Number
Ag Element
US Ultrasoft pseudopotential PP-type
T Nonlinear Core Correction
SLA PW PBX PBC PBE Exchange-Correlation functional XC functional used
19.000000000000 Z valence
-295.21248450600 Total energy
0.00000 0.00000 Suggested cutoff for wfc and rho
2 Max angular momentum component
887 Number of points in mesh
5 6 Number of Wavefunctions, Number of Projectors
Wavefunctions nl l occ
4S 0 2.00
4P 1 6.00
4D 2 10.00
5S 0 0.50
5P 1 0.00
</PP_HEADER>
```

Practical Pseudopotential Tips

- Norm conserving PPs with small core radii (requiring high PW cutoffs) are generally the *most accurate* PPs
- Use the **same type** of PP for all atoms in your system
- PPs are created for use with a specific **exchange-correlation functional** – ensure all PPs in your system were created with the *same functional*
- Compute property of interest twice *using different types* of PPs to (roughly) assess variation
- The PW cutoff for a set of atoms will be determined by **hardest PP** of the set

Generate or Download Pseudopotentials:

opium.sourceforge.net/ (Opium PP Generation Project)

th.fhi-berlin.mpg.de/th/fhi98md/fhi98PP/ (fhi98PP PP program)

physics.rutgers.edu/gbrv/ (**Vanderbilt Ultra-soft PPs**)