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Screening of phase-separation CO₂ absorbent using machine learning combined with molecular information
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Abstract (less than 300 words)
<p>Carbon dioxide capture technologies have been focused to overcome global warming. A chemical absorption process using an aqueous amine solvent is one of the most established technologies of CO₂ capture because of the high absorption rate and high reactivity with CO₂. However, this process operation needs a large amount of energy, causing high regeneration costs. Recently, to reduce energy consumption, phase-separation absorbents have been reported. The phase-separation absorbents are mainly composed of mixed solvents of alkanolamine and organic solvents like glycol ether or alcohol. When the absorbent reacts with CO₂, a single liquid phase of the phase-separation absorbent transforms into two liquid phases, CO₂-rich and -lean phases. This absorbent can reduce the temperature difference between the absorption and desorption operations from 80 °C of the conventional chemical absorption process to 40 °C [1]. However, it is difficult to know phase behaviors of mixed-solvent absorbents, resulting in the necessity of screening experiments to find phase-separation absorbents [2, 3].</p> <p>In this work, we developed a screening method of the phase behaviors of the mixed-solvent absorbents using supervised machine learning models: random forest, logistic regression, and support vector machine. There are 61 mixed-solvent absorbents containing alkanolamine/glycol ether or alcohol in a dataset. The dataset was split into a training set and test set, which contains 31 and 30 data of the absorbents, respectively. Extended-connectivity fingerprint [4] or molecular surface charge distributions calculated from a COSMO calculation [5] are used as molecular descriptors in the machine learning models. The machine learning models successfully predicted phase states of the mixed-solvent absorbents before and after CO₂ absorption with accuracies of more than 90 %. Furthermore, we analyzed contributions of explanatory variables to predict the phase states using the learned model.</p> <p>[1] H. Machida et al., Int. J. Greenh. Gas Control. 75 (2018) 1-7 [2] F. Barzagli et al., Int. J. Greenh. Gas Control. 60 (2017) 100-109 [3] W. Zhang et al., Appl. Energy 195 (2017) 316–323 [4] D. Rogers and M. Hahn, J. Chem. Inf. Model. 50(5) (2010) 742-754 [5] A. Klamt and G. Schüürmann, J. Chem. Soc. Perkin trans. 2., 5 (1993) 799-805</p>
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