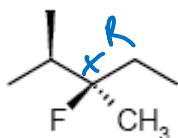


## CHEM 2E03 Tutorial #5

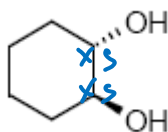
1. a) For each of the following molecules determine if the compound is chiral, achiral, or meso.
- b) Where appropriate, identify all chirality centres in the structure with an asterisk (\*) and determine the configuration (*R* or *S*).
- c) Calculate the maximum number of stereoisomers the molecule can have.

a)



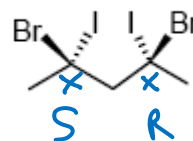
Chiral  $2^1$  Stereoisomers

b)



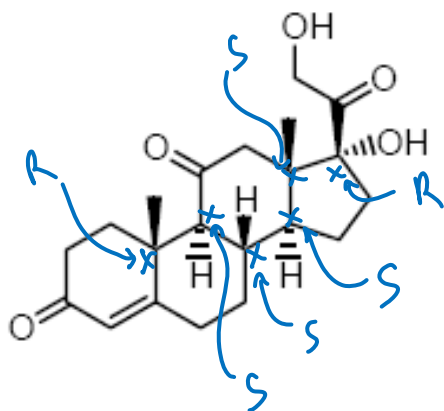
Chiral  $2^2$  Stereoisomers

c)



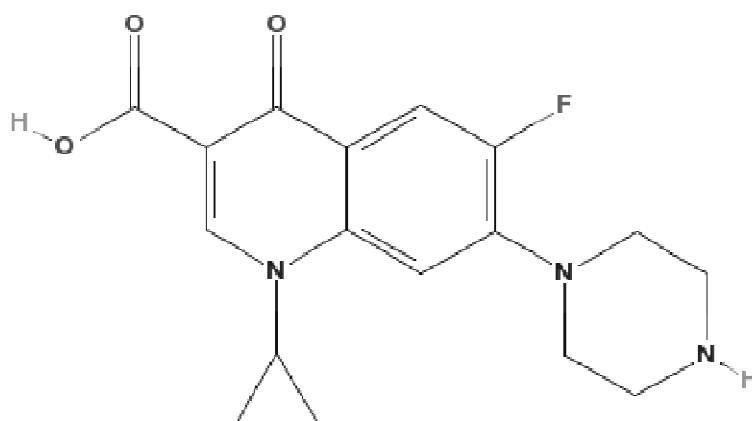
Meso (achiral)  
 $2^2$  Stereoisomers

d) Cortisone



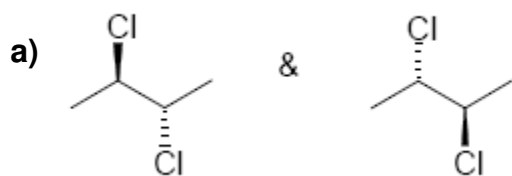
Chiral  $2^6$  Stereoisomers

e) Ciprofloxacin (Anthrax treatment)

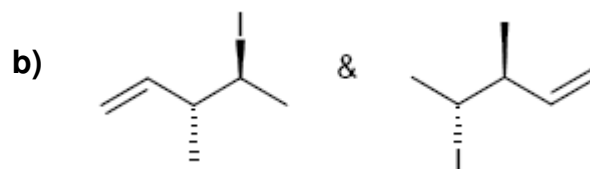


Achiral

2. Identify the relationship between the following pairs of structures by describing them as constitutional isomers, enantiomers, diastereomers, meso compounds, or identical.



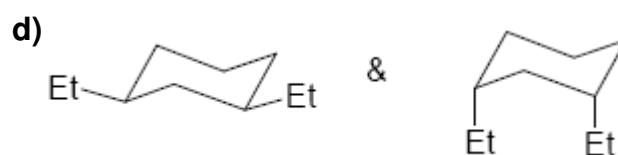
**Meso compounds**



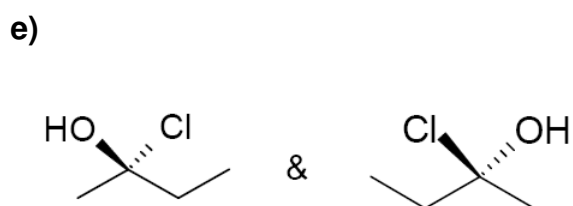
**Enantiomers**



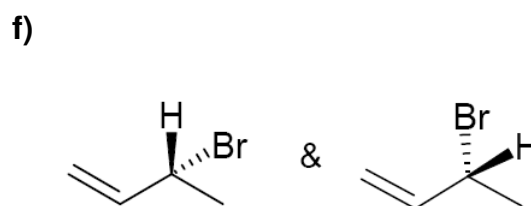
**Identical**



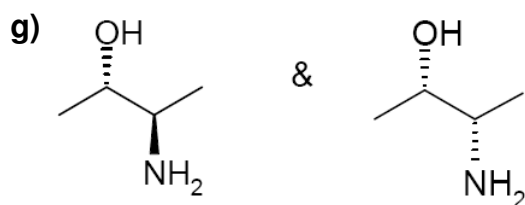
**Meso compounds**



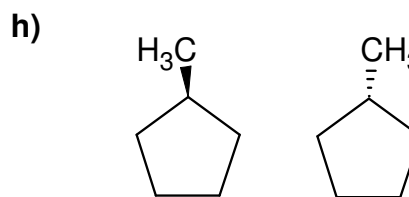
**Identical**



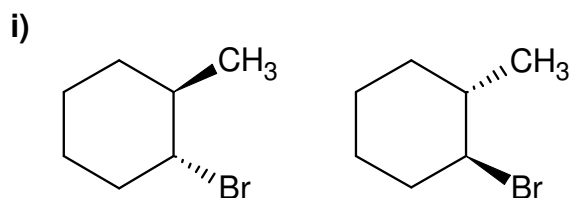
**Identical**



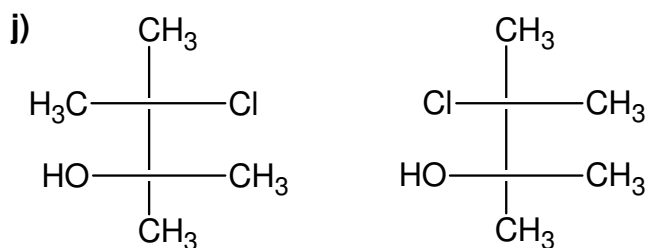
**Diastereomers**



**Identical**

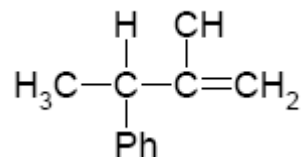


**Enantiomers**



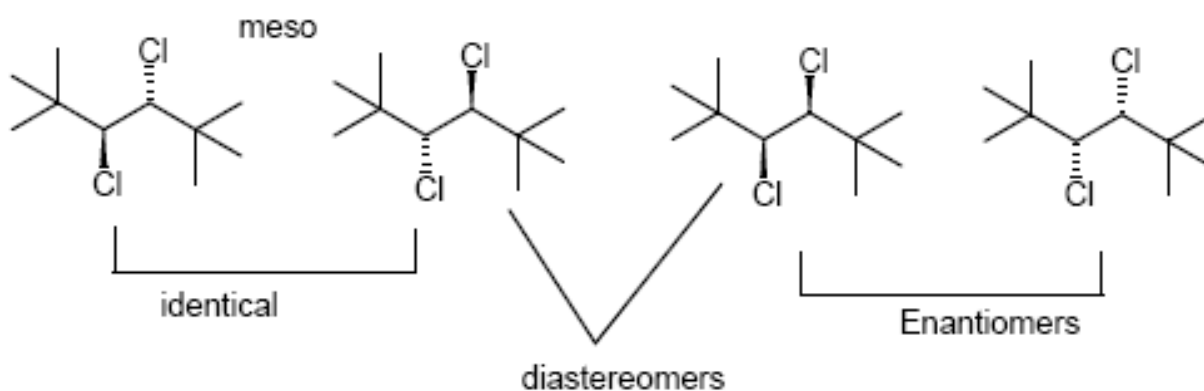
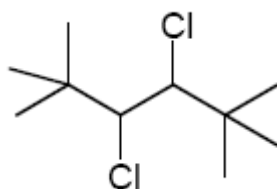
**Identical - there is no chirality centre in this molecule**

3. The specific rotation of the R enantiomer of the following alkene is  $[\alpha]_D^{20} = +76$  degree·mL·g<sup>-1</sup>·dm<sup>-1</sup>, and its molecular mass is 146.2 g/mol. What is the observed rotation of a 0.5 M solution of this compound in a 5 cm sample path?

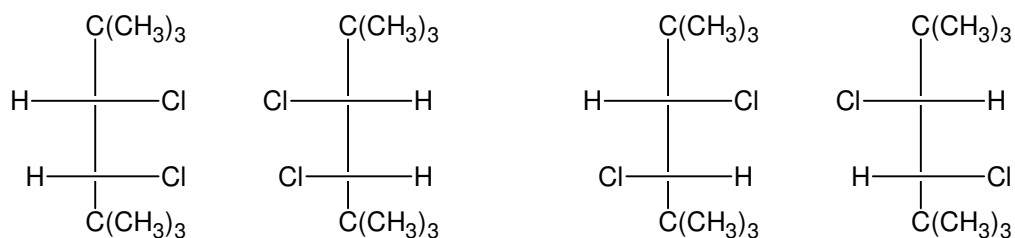


$$\begin{aligned} \alpha &= [\alpha] c l \\ &= (+76 \text{ degree}\cdot\text{mL}\cdot\text{g}^{-1}\cdot\text{dm}^{-1}) (0.5 \text{ mol/L} \times 1\text{L}/1000 \text{ ml} \times 146.2 \text{ g/mol}) (0.5 \text{ dm}) \\ &= +2.78^\circ \end{aligned}$$

4. Draw all the stereoisomers of the following compound. Indicate the relationship between the molecules.



Fisher projections make it easier to 'see' the relationships between these molecules



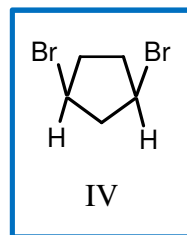
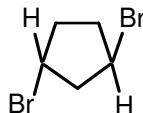
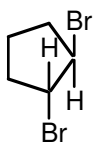
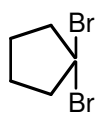
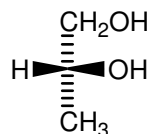
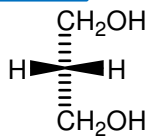
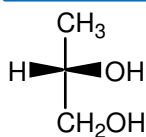
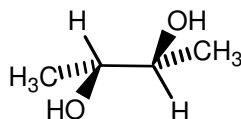
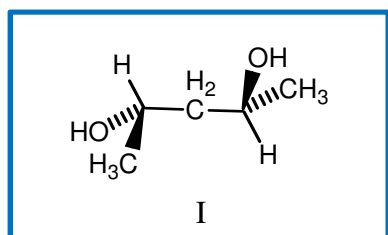
**Meso –plane of symmetry**  
 Rotating the 2<sup>nd</sup> molecule 180°  
 in the plane of the paper produces  
 the 1<sup>st</sup> molecule

**Enantiomers: nonsuperposable  
 mirror images**

5. What is the molecular formula for the alkane of smallest molecular weight which possesses a chiral center?

$C_7H_{16}$  is the smallest alkane that possesses a chiral centre.

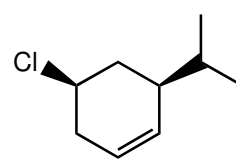
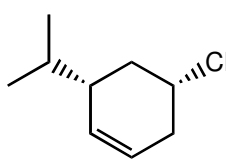
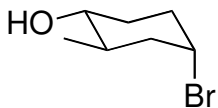
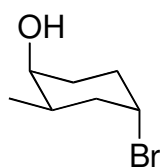
6. Which of the following is (are) a *meso* compounds?



7. What is the relationship between the following sets of compounds?

A) Diastereomers

B) Identical



C) Enantiomers

D) Diastereomers

