

Density functional theory: the basics

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Basic concepts

Wave function is a complex-value entity that describes the quantum state of one or more particles. Example: free electron $\Psi(\mathbf{r}, \mathbf{k}) = e^{i\mathbf{k}\cdot\mathbf{r}}$

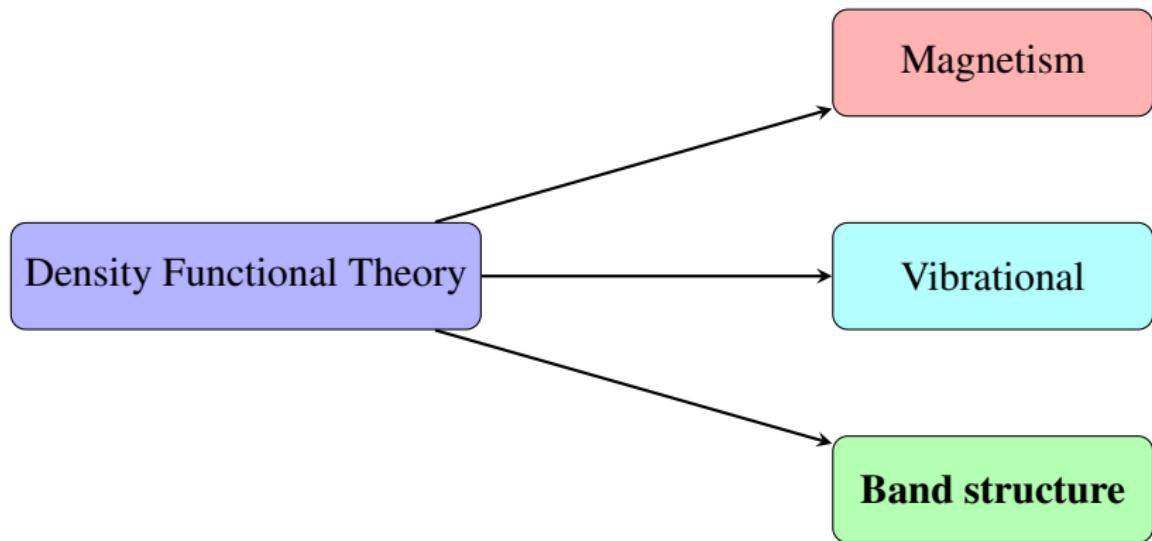
Expected value in quantum mechanics is the act of performing a measurement on a observable. Example: $E = \langle \Psi | \hat{H} | \Psi \rangle$. Explicitly

$$\langle \Psi | \hat{H} | \Psi \rangle = \int \cdots \int \Psi^*(\vec{x}_1, \dots, \vec{x}_N, \vec{R}_1, \dots, \vec{R}_M) \hat{H} \Psi(\vec{x}_1, \dots, \vec{x}_N, \vec{R}_1, \dots, \vec{R}_M) dV \quad (1)$$

Coordinate system In this lecture, \vec{x}_N reffers to the coordinate system of an electron N, whereas \vec{R}_M to a nucleus M

Goal

The goal of the Density Functional Theory is to find the ground state energy E_0 for a given system and its ground state electronic structure, from which one could calculate further material properties such as



The problem

One can start to solve the following problem

$$\hat{H}\Psi(\vec{x}_1, \dots, \vec{x}_N, \vec{R}_1, \dots, \vec{R}_M) = E\Psi(\vec{x}_1, \dots, \vec{x}_N, \vec{R}_1, \dots, \vec{R}_M), \quad (2)$$

where

$$\hat{H} = -\frac{1}{2} \sum_{i=1}^N \nabla_i^2 - \frac{1}{2} \sum_{A=1}^M \frac{1}{M_A} \nabla_A^2 - \sum_{i=1}^N \sum_{A=1}^M \frac{Z_A}{r_{iA}} + \sum_{i=1}^N \sum_{j>i}^N \frac{1}{r_{ij}} + \sum_{A=1}^M \sum_{B>1}^M \frac{Z_A Z_B}{R_{AB}} \quad (3)$$

Too many dimensions! Impossible to solve, but one can attempt to make it simpler. Adiabatic or Born Oppenheimer approximation:

$$\Psi(\vec{x}_1, \dots, \vec{x}_N, \vec{R}_1, \dots, \vec{R}_M) = \Psi_{tot} = \Psi_{ele}(\vec{x}_1, \dots, \vec{x}_N) * \Psi_{nuc}(\vec{R}_1, \dots, \vec{R}_M) \quad (4)$$

Born Oppenheimer approximation

Then, the new Hamiltonian will be

$$\hat{H} = -\frac{1}{2} \sum_{i=1}^N \nabla_i^2 - \sum_{i=1}^N \sum_{A=1}^M \frac{Z_A}{r_{iA}} + \sum_{i=1}^N \sum_{j>i}^N \frac{1}{r_{ij}} = \hat{T} + \hat{V}_{Nc} + \hat{V}_{ee} \quad (5)$$

Example: Water molecule H₂O. 10 electrons. Problem has 30 dimensions!!

Variational principle

The variational principle simply states that any other trial wave function will always give a upper bound energy than the true ground state one

$$\langle \Psi_{trial} | \hat{H} | \Psi_{trial} \rangle = E_{trial} \geq E_0 = \langle \Psi_0 | \hat{H} | \Psi_0 \rangle \quad (6)$$

Therefore, the strategy is to minimize the functional $E[\Psi]$

$$E_0 = \min_{\Psi \rightarrow N} E[\Psi] = \min_{\Psi \rightarrow N} \langle \Psi | \hat{T} + \hat{V}_{nc} + \hat{V}_{ee} | \Psi \rangle \quad (7)$$

Our summary so far

$$\{N, Z_A, R_A\} \Rightarrow \hat{H} \Rightarrow \Psi_0 \Rightarrow E_0 \quad (8)$$

Hartree-Fock

One attempt to minimize the energy functional is to use the Slater determinants

$$\Psi_0 = \Phi_{SD} = \frac{1}{\sqrt{N!}} \det \{\chi_1(\vec{x}_1), \dots, \chi_N(\vec{x}_N)\}, \quad (9)$$

then one has to

$$E_{HF} = \min_{\Phi_{SD} \rightarrow N} E[\Phi_{SD}] \quad (10)$$

$$\langle \Phi_{SD} | \hat{H} | \Phi_{SD} \rangle = \langle i | \hat{h} | i \rangle + \frac{1}{2} \sum_{i=1}^N \sum_{j=1}^N \langle ii | jj \rangle - \langle ij | ji \rangle. \quad (11)$$

Using the variational principle

$$\frac{\delta F}{\delta \chi_j^*} = \langle \Phi_{SD} | \hat{H} | \Phi_{SD} \rangle - \sum_i \epsilon_i \left(\int |\chi_i(\vec{x}_i)|^2 - 1 \right) \quad (12)$$

Hartree-Fock

It will give the Hartree-Fock equation

$$\hat{f}\chi_i = \epsilon_i\chi_i; \quad i = 1, 2, 3, \dots, N \quad (13)$$

with

$$\hat{f} = -\frac{1}{2}\nabla^2 - \sum_A^M \frac{Z_A}{r_{iA}} + V_{HF}(i) \quad (14)$$

and

$$V_{HF}(i) = \sum_j^N \left(\hat{J}_j(\vec{x}_1) - \hat{K}_j(\vec{x}_1) \right) \quad (15)$$

$$\hat{J}_j(\vec{x}_1) = \int |\chi_j(\vec{x}_2)|^2 \frac{1}{r_{12}} d\vec{x}_2 \quad (16)$$

$$\hat{K}_j(\vec{x}_1)\chi_i(\vec{x}_1) = \int \chi_j^*(\vec{x}_2) \frac{1}{r_{12}} \chi_i(\vec{x}_2) \chi_j(\vec{x}_1) d\vec{x}_2 \quad (17)$$

Electronic density

Instead of using the wave function with $3N$ dimensions, one could use the electron density

$$\rho(\vec{r}) = N \int |\Psi(\vec{x}_1, \dots, \vec{x}_N)|^2 d\vec{x}_1 \dots d\vec{x}_N \quad (18)$$

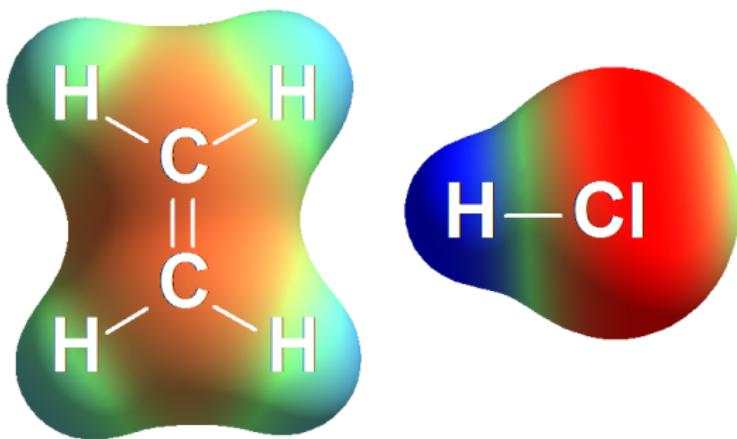


Figure 1: Electronic density

HK Theorems

First HK theorem: The external potential, V_{ext} , where the electrons are immersed is a unique functional of the electronic density $\rho(\vec{r})$

Second HK theorem: The energy functional $E[\rho]$ is minimized by the electronic ground state density $\rho_0(\vec{r})$.

Kohn-Sham approach

Starting from the HK theorems

$$E_0 = \min_{\rho \rightarrow N} \left(F[\rho] + \int \rho(\vec{r}) V_{Ne} d\vec{r} \right) \quad (19)$$

where $F[\rho]$ is an universal functional

$$F[\rho] = T[\rho] + J[\rho] + E_{ncl}[\rho]. \quad (20)$$

In Eq. 20, only $J[\rho]$ is known, while the explicit forms of the other two are unknown. Let us now consider a system with non-interacting electrons.

$$F[\rho] = T_S[\rho] + J[\rho] + E_{xc}[\rho] \quad (21)$$

where $T_S[\rho]$ is the kinetic energy of the non-interacting electrons, such as $T_S \neq T_C$

$$E_{xc}[\rho] = (T[\rho] - T_S[\rho]) + (E_{ee}[\rho] - J[\rho]) = T_C[\rho] + E_{ncl}[\rho] \quad (22)$$

Kohn-Sham approach

Then, one has to minimize

$$\begin{aligned} E[\rho] &= T_S[\rho] + J[\rho] + E_{Ne}[\rho] + E_{xc}[\rho] \\ &= -\frac{1}{2} \sum_i^N \int \phi_i(\vec{r}_1)^* \nabla^2 \phi_i(\vec{r}_1) d\vec{r}_1 \\ &\quad + \frac{1}{2} \sum_i^N \sum_j^N \int \int |\phi_i(\vec{r}_1)|^2 \frac{1}{r_{12}} |\phi_j(\vec{r}_2)|^2 d\vec{r}_1 d\vec{r}_2 \\ &\quad - \sum_i^N \int \sum_A^M \frac{Z_A}{r_{iA}} |\phi_i(\vec{r}_1)|^2 d\vec{r}_1 + E_{ex}[\rho] \end{aligned} \tag{23}$$

Kohn-Sham equation

Finally, if one can use the variational principle to get

$$\left(-\frac{1}{2} \nabla^2 + V_{eff} \right) \phi_i = \epsilon_i \phi_i, \quad (24)$$

with

$$V_{eff}(\vec{r}) = \int \frac{\rho(\vec{r}_2)}{r_{12}} d\vec{r}_2 + V_{xc}(\vec{r}_1) - \sum_A \frac{Z_A}{r_{iA}} \quad (25)$$

Notes about the exchange-correlation potential

The problem is solved correctly if V_{xc} is known! The whole complexity of the problem boils down to that potential.

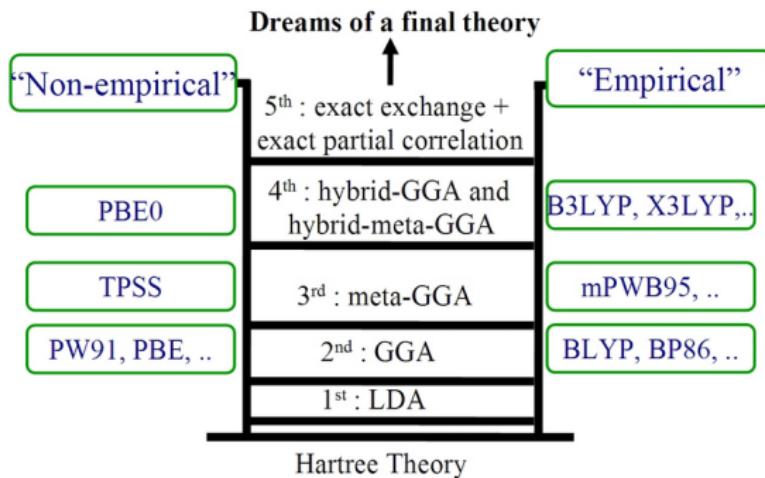


Figure 2: Jacob's ladder

Kohn-Sham scheme

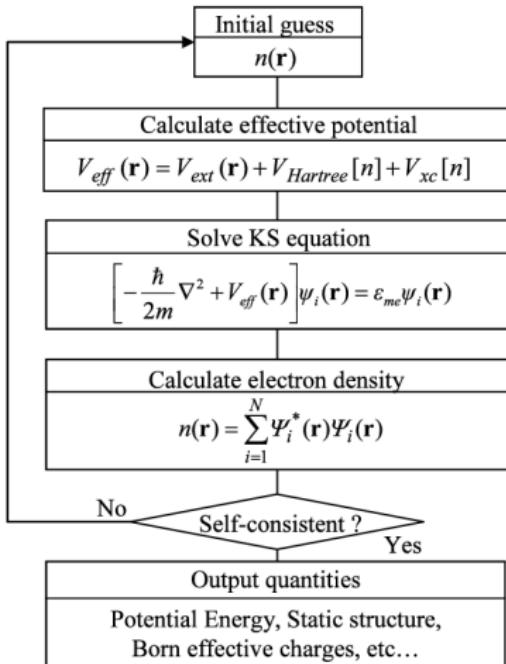


Figure 3: Kohn-Sham scheme