

W9209/9210/9211/9212



DUBLIN INSTITUTE OF TECHNOLOGY

School of Mathematical Sciences

DT9209 MSc Applied Mathematics

DT9210 MSc Applied Mathematics

DT9211 MSc Applied Mathematics

DT9212 MSc Applied Mathematics

WINTER EXAMINATIONS 2017/2018

MATH 9971: CLASSICAL MECHANICS AND THERMODYNAMICS

DR R. IVANOV

DR C HILLS

DR S MITCHELL

18:30 – 20:30 pm, Tuesday, 9 January 2018

Duration: 2 hours

Answer three questions

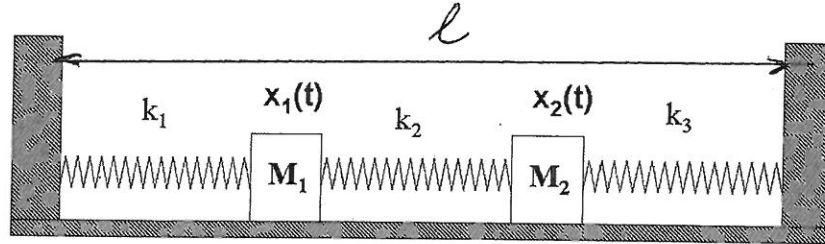
All questions carry equal marks

Approved calculators may be used

Mathematical tables are provided

New Cambridge Statistical Tables are NOT permitted

1. Two point masses M_1 and M_2 are connected with three ideal springs with elasticities k_1 , k_2 and k_3 , as shown to the figure below. The distance l is constant. There is no friction.



- Introducing appropriate generalised coordinates, write down the Lagrangian of the system. (6)
- Write down the Lagrangian equations of motion. (8)
- Find the equilibrium positions of the two masses. (9)
- Find the possible frequencies of the oscillations about the equilibrium found in part c). (10)

[33]

2. A point mass m moves in the field with a scalar potential $V(\mathbf{x})$, where $\mathbf{x} = (x_1, x_2, x_3)$ is the position vector in Cartesian coordinates.

- Find the Hamiltonian and write down the Hamiltonian equations of motion for the point mass in cylindrical coordinates (r, θ, z) where

$$x_1 = r \cos \theta, \quad x_2 = r \sin \theta, \quad x_3 = z.$$

(17)

- Simplify the equations in the case of a central force $V = V(r)$ where r is introduced in part a). Do not solve these equations. Find the conserved momenta. (6)
- Using the conservation of the Hamiltonian and the conserved momenta from part b) (or otherwise) demonstrate that in the case of a potential $V = \alpha r^2$ with α being a positive constant, the trajectory is a closed orbit. (10)

[33]

3. a) Show that for an ideal gas the thermal capacities (per 1 mol) satisfy the relation

$$C_p - C_V = R$$

where $R = 8.3144598 \text{ J}/(\text{mol} \cdot \text{K})$ is the gas constant.

(10)

- b) In a polytropic process the thermal capacity C (per 1 mol) is constant. Show that an ideal gas undergoing a polytropic process satisfies the relation $pV^n = \text{constant}$ with a constant index n . Express n through C , C_p and C_V .

(18)

- c) Show that any adiabatic process for an ideal gas is also polytropic and find n for an adiabatic process.

(5)

[33]

4. a) Two bodies with initial temperatures T_1 and T_2 and constant thermal capacities C_1 and C_2 are put in contact until they reach equilibrium and their temperatures equalise. Find the equilibrium temperature, the entropy change and demonstrate that this process is irreversible. Take for simplicity $C_1 = C_2$.

(13)

- b) Two bodies with initial temperatures T_1 and T_2 and constant thermal capacities C_1 and C_2 are used as a heater and a cooler of a heat engine. Find the maximal work that can be produced by the heat engine. Solve the problem for arbitrary C_1 and C_2 first and then obtain the solution for $C_1 = C_2$ as a particular case.

(20)

[33]

Year: 20/17/2018

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Institiúid Teicneolaíochta Bhaile Átha Cliath



Session: Winter / Summer / Autumn

Question:

1

Page:

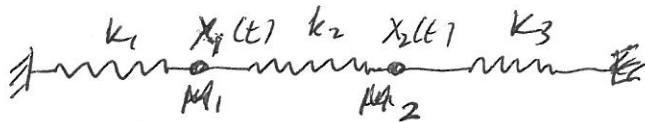
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Sample Examination Solutions

Qn part [e.g. 4 a) iii)]

Mark

a)

For generalised coordinates $x_{1,2}(t)$ like on the figureThe kinetic energy is $T = \frac{m_1 \dot{x}_1^2}{2} + \frac{m_2 \dot{x}_2^2}{2}$

the potential energy is

$$V = \frac{1}{2} k_1 x_1^2 + \frac{1}{2} k_2 (x_2 - x_1)^2 + \frac{1}{2} k_3 (\ell - x_2)^2$$

$$L = T - V = \frac{m_1 \dot{x}_1^2}{2} + \frac{m_2 \dot{x}_2^2}{2} - \frac{k_1 x_1^2}{2} - \frac{k_2 (x_2 - x_1)^2}{2} - \frac{k_3 (\ell - x_2)^2}{2}$$

b)

$$\frac{d}{dt} \left(\frac{\partial L}{\partial \dot{x}_1} \right) - \frac{\partial L}{\partial x_1} = 0$$

$$\frac{d}{dt} (m_1 \dot{x}_1) - (-k_1 x_1 + k_2 (x_2 - x_1)) = 0$$

$$m_1 \ddot{x}_1 + k_1 x_1 - k_2 (x_2 - x_1) = 0$$

$$\boxed{m_1 \ddot{x}_1 + (k_1 + k_2) x_1 - k_2 x_2 = 0} \quad \text{Eq. 1.}$$

Question continued on next page



OR

Total Final Mark Allocation



Year: 20 / 20

Module Code:

(to be submitted with exam coversheet)

Session: Winter / Summer / Autumn (select)

School of Mathematical Sciences
Scoil na nEolaíochtaí MatamaiticiúlaDublin Institute of Technology
Institiúid Teicneolaíochta Bhaile Átha Cliath

Question:

1

Page:

2

Sample Examination Solutions

Qn part (e.g. 4 a) ii)

Mark

$$\frac{d}{dt} \left(\frac{\partial L}{\partial \dot{x}_2} \right) - \frac{\partial L}{\partial x_2} = 0$$

$$\frac{d}{dt} (M_2 \dot{x}_2) - (-k_2(x_2 - x_1) + k_3(l - x_2)) = 0$$

$$M_2 \ddot{x}_2 + k_2(x_2 - x_1) - k_3(l - x_2) = 0$$

$$M_2 \ddot{x}_2 + (k_2 + k_3)x_2 - k_2x_1 = k_3l \quad \text{Eq. 2.}$$

(c)

Assume that $x_k = l_k + q_k$ where l_k is the distance at equilibrium, $l_k = \text{const.}$

From eq. 1 $\Rightarrow (k_1 + k_2)l_1 - k_2l_2 = 0$

From eq. 2 $-k_2l_1 + (k_2 + k_3)l_2 = k_3l$

$$l_1 = \frac{\begin{vmatrix} 0 & -k_2 \\ k_3l & k_2 + k_3 \end{vmatrix}}{\begin{vmatrix} k_1 + k_2 & -k_2 \\ -k_2 & k_2 + k_3 \end{vmatrix}} = \frac{k_2k_3l}{(k_1 + k_2)(k_2 + k_3) - k_2^2} = \frac{k_2k_3l}{k_1k_2 + k_1k_3 + k_2k_3}$$

$$l_2 = \frac{\begin{vmatrix} k_1 + k_2 & 0 \\ -k_2 & k_3l \end{vmatrix}}{\begin{vmatrix} k_1 + k_2 & -k_2 \\ -k_2 & k_2 + k_3 \end{vmatrix}} = \frac{(k_1 + k_2)k_3l}{k_1k_2 + k_1k_3 + k_2k_3}$$

Also $l_2 = l_1 + \frac{k_1k_3l}{k_1k_2 + k_1k_3 + k_2k_3}$

Question continued on next page



OR

Total Final Mark Allocation



Year: 20__ / 20__

Module Code:

Session: Winter / Summer / Autumn

Question: 1

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Page: 3

Sample Examination Solutions

Qn part (e.g. 4a) iii)

Mark

(d) Assuming $q_k = q_{k0} e^{i\omega t}$, thus $\ddot{q}_k = -\omega^2 q_k$

2

From eq(1) and eq. 2 we have

$$\begin{cases} -M_1 \omega^2 q_1 + (k_1 + k_2) q_1 - k_2 q_2 = 0 \\ -M_2 \omega^2 q_2 + (k_2 + k_3) q_2 - k_2 q_1 = 0 \end{cases}$$

2

or

$$\begin{cases} (-M_1 \omega^2 + k_1 + k_2) q_1 - k_2 q_2 = 0 \\ -k_2 q_1 + (k_2 + k_3 - M_2 \omega^2) q_2 = 0 \end{cases}$$

the compatibility condition is

$$\Rightarrow (k_1 + k_2 - M_1 \omega^2)(k_2 + k_3 - M_2 \omega^2) - k_2^2 = 0$$

3

$$M_1 M_2 \omega^4 - [M_1 (k_2 + k_3) + M_2 (k_1 + k_2)] \omega^2 + k_1 k_2 + k_1 k_3 + k_2 k_3 = 0$$

2

$$\omega_{\pm}^2 = \frac{M_1 (k_2 + k_3) + M_2 (k_1 + k_2) \pm \sqrt{[M_1 (k_2 + k_3) + M_2 (k_1 + k_2)]^2 - 4 M_1 M_2 (k_1 k_2 + k_1 k_3 + k_2 k_3)}}{2 M_1 M_2}$$

1

two possible frequencies.

Further examination shows that the discriminant of the above eq. is positive, i.e.

$\omega_{\pm}^2 > 0$ - two possible frequencies

not required

Question continued on next page

OR

Total Final Mark Allocation

33

Year: 20 / 20

School of Mathematical Sciences
Scoil na nEolaíochtaí Matamaiticiúla

Module Code:

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Session: Winter / Summer / Autumn

Question:

2

Page:

4

Sample Examination Solutions

Qn part (e.g. 4a) (iii)

Mark

(a)

The Lagrangian is

$$L = T - V = \frac{m}{2} (\dot{x}_1^2 + \dot{x}_2^2 + \dot{x}_3^2) - V(\vec{x})$$

$$\dot{x}_1 = \dot{r} \cos \theta + r(-\sin \theta) \dot{\theta}$$

$$\dot{x}_2 = \dot{r} \sin \theta + r \cos \theta \dot{\theta}$$

$$\dot{x}_1^2 + \dot{x}_2^2 = \dot{r}^2 + r^2 \dot{\theta}^2$$

$$L = \frac{m}{2} (\dot{r}^2 + r^2 \dot{\theta}^2 + \dot{z}^2) - V(r, \theta, z)$$

$$p_r = \frac{\partial L}{\partial \dot{r}} = m \dot{r} \Rightarrow \dot{r} = \frac{p_r}{m}$$

$$p_\theta = \frac{\partial L}{\partial \dot{\theta}} = m r^2 \dot{\theta} \Rightarrow \dot{\theta} = \frac{p_\theta}{m r^2}$$

$$p_z = m \dot{z} \Rightarrow \dot{z} = \frac{p_z}{m}$$

$$H = p_r \dot{r} + p_\theta \dot{\theta} + p_z \dot{z} - L =$$

$$= p_r \left(\frac{p_r}{m} \right) + p_\theta \left(\frac{p_\theta}{m r^2} \right) + p_z \left(\frac{p_z}{m} \right) -$$

$$- \frac{m}{2} \left[\left(\frac{p_r}{m} \right)^2 + r^2 \left(\frac{p_\theta}{m r^2} \right)^2 + \left(\frac{p_z}{m} \right)^2 \right] + V(r, \theta, z)$$

$$H = \frac{1}{2m} (p_r^2 + \frac{p_\theta^2}{r^2} + p_z^2) + V(r, \theta, z)$$

5

3

3

Question continued on next page



OR

Total Final Mark Allocation



Year: 20__ / 20__

Module Code:

(to be submitted with exam cover sheet)

Session: Winter / Summer / Autumn (delete)

Question:

School of Mathematical Sciences
Scoil na nEolaíochtaí Matamaiticiúla

Dublin Institute of Technology
Institiúid Teicneolaíochta Bhaile Átha Cliath



Page:

5

Sample Examination Solutions

Qn part (e.g. 4a) iii)

Mark

$$\dot{p}_r = -\frac{\partial H}{\partial r} = -\frac{p_\theta^2}{2m} \left(-\frac{2}{r^3}\right) - \frac{\partial V}{\partial r} = \frac{p_\theta^2}{mr^3} - \frac{\partial V}{\partial r}$$

$$\dot{p}_\theta = -\frac{\partial H}{\partial \theta} = -\frac{\partial V}{\partial \theta}$$

$$\dot{p}_z = -\frac{\partial H}{\partial z} = -\frac{\partial V}{\partial z}$$

$$\dot{r} = \frac{\partial H}{\partial p_r} = \frac{p_r}{m}$$

$$\dot{\theta} = \frac{\partial H}{\partial p_\theta} = \frac{p_\theta}{mr^2}$$

$$\dot{z} = \frac{\partial H}{\partial p_z} = \frac{p_z}{m}$$

6

b)

When $V = V(r) \Rightarrow \frac{\partial V}{\partial r} \neq 0$ but $\frac{\partial V}{\partial \theta} = 0, \frac{\partial V}{\partial z} = 0$

$\Rightarrow p_\theta = \text{const}$ (Angular momentum conserv.)

$p_z = \text{const}$ (center of mass motion)

} 2

$$\dot{p}_r = \frac{p_\theta^2}{mr^3} - \frac{dV(r)}{dr}$$

$$\dot{r} = \frac{p_r}{m}, \quad \dot{\theta} = \frac{p_\theta}{mr^2}, \quad \dot{z} = \frac{p_z}{m}$$

} 4

Question continued on next page



OR

Total Final Mark Allocation (check)

Year: 20 ___ / 20 ___

Module Code:

Session: Winter / Summer / Autumn *(delete)*

Question:

School of Mathematical Sciences
Scoil na nEolaíochtaí Matamaiticiúla

Dublin Institute of Technology
Institiúid Teicneolaíochta Bhaile Átha Cliath



Page:

6

Sample Examination Solutions

Qn part (e.g. 4 a) ii)

Mark

c) The trajectory is a line, which represents the intersection of 2 surfaces in the (r, p_r, t) space

(i) $H = \text{const}$, $H(r, p_r) = \text{const}$

2

(ii) $H = H(r, p_r)$

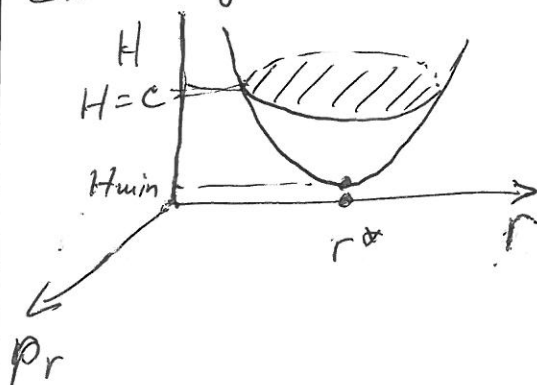
Note that $H(r, p_r) = \frac{1}{2m} \left(p_r^2 + \frac{p_\theta^2}{r^2} \right) + \alpha r^2 + \frac{p_z^2}{2m}$ 2

and $\Rightarrow H(r, p_r)$ has a minimum at

$\frac{\partial H}{\partial r} = \frac{p_\theta^2}{mr^3} - 2\alpha r = 0 \Rightarrow (r^*)^4 = \frac{p_\theta^2}{2\alpha m} = \text{const.}$ 2

$\frac{\partial H}{\partial p_r} = \frac{p_r}{m} = 0 \Leftrightarrow \dot{r} = 0 \Rightarrow r = r^* = \text{const.}$

In every direction $(r, p_r) \rightarrow \infty$, $H \rightarrow \infty$



The section $H = c$ is a closed orbit.

4

Question continued on next page



OR

Total Final Mark Allocation

33

Year: 20 __ / 20 __

School of Mathematical Sciences
Scoil na nEolaíochtaí MatamaiticiúlaModule Code: []Dublin Institute of Technology
Institiúid Teicneolaíochta Bhaile Átha Cliath

Session: Winter / Summer / Autumn

Question: 3Page: 7

Sample Examination Solutions

Qn part (e.g. 4 a) ii)

Mark

(a)

From the first principle

$$\delta Q = du + dw$$

For a process with $p = \text{const} \Rightarrow$

$$C_p dT = C_v dT + p dV ; pV = RT$$

$$\Rightarrow p dV = R dT \quad (p = \text{const})$$

$$\Rightarrow C_p dT = C_v dT + R dT \Rightarrow \boxed{C_p - C_v = R}$$

3

3

1

3

(b)

$$C dT = C_v dT + p dV$$

$$d(pV) = R dT \Rightarrow dT = \frac{1}{R} d(pV)$$

$$\Rightarrow (C - C_v) \frac{1}{R} d(pV) = p dV$$

$$\frac{C - C_v}{R} (p dV + V dp) = p dV$$

$$\frac{C - C_v}{C_p - C_v} p dV - p dV + \frac{C - C_v}{C_p - C_v} V dp = 0$$

$$\frac{C - C_v - C_p + C_v}{C_p - C_v} p dV + \frac{C - C_v}{C_p - C_v} V dp = 0$$

3

3

Question continued on next page



OR

Total Final Mark Allocation



Year: 20__ / 20__

School of Mathematical Sciences
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Session: Winter / Summer / Autumn (select)

Dublin Institute of Technology
Institiúid Teicneolaíochta Bhaile Átha Cliath

Question:

3

Page:

8

Sample Examination Solutions

Qn part (e.g. 4 a) iii)

Mark

	$\frac{C - C_p}{C - C_v} \cdot \frac{dV}{V} + \frac{dp}{p} = 0$	8
	$\Rightarrow \frac{C - C_p}{C - C_v} \ln V + \ln p = \text{const}$	
	$p V^{\frac{C - C_p}{C - C_v}} = \text{const}$	3
	$n = \frac{C - C_p}{C - C_v}$	1
(c)	For an adiabatic process $\delta Q = 0 \Rightarrow C = 0$	4
	$\Rightarrow n = \frac{C_p}{C_v}$	1

Question continued on next page



OR

Total Final Mark Allocation (40%)

33

Year: 20 ___ / 20 ___

School of Mathematical Sciences
Scoil na nEolaíochtaí MatamaiticiúlaModule Code: (to be submitted with exam cover sheet)

Session: Winter / Summer / Autumn (select)

Dublin Institute of Technology
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Sample Examination Solutions

Qn part (e.g. 4 a) iii)

Mark

(a)

let us suppose that $T_1 > T_2$
 then, if T is the equilibrium temperature
 we have

$$C_1(T_1 - T) = C_2(T - T_2)$$

$$C_1 T_1 + C_2 T_2 = (C_1 + C_2) T \Rightarrow T = \frac{C_1 T_1 + C_2 T_2}{C_1 + C_2}$$

$$T = \frac{1}{2}(T_1 + T_2) \quad \text{when } C_1 = C_2$$

$$\Delta S = \Delta S_1 + \Delta S_2 = \int_{T_1}^T \frac{\delta Q_1}{T} + \int_{T_2}^T \frac{\delta Q_2}{T} = \int_{T_1}^T \frac{C_1 dT}{T} + \int_{T_2}^T \frac{C_2 dT}{T}$$

$$= C_1 \ln \frac{T}{T_1} + C_2 \ln \frac{T}{T_2} = \ln \frac{T^{C_1+C_2}}{T_1^{C_1} T_2^{C_2}}$$

when $C_1 = C_2 = C$

$$\Delta S = C \ln \frac{T^2}{T_1 T_2} = C \ln \frac{(T_1 + T_2)^2}{4 T_1 T_2} > 0$$

$$\text{because } (T_1 + T_2)^2 > 4 T_1 T_2 \Leftrightarrow (T_1 - T_2)^2 > 0$$

Since $\Delta S > 0$ the process is irreversible

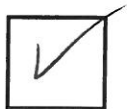
5

4

2

2

Question continued on next page



OR

Total Final Mark Allocation (max)



Year: 20 / 20

Module Code:

(to be submitted with exam cover sheet)

Session: Winter / Summer / Autumn

Question:

4

School of Mathematical Sciences
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Institiúid Teicneolaíochta Bhaile Átha Cliath

Page:

10 of 10

Sample Examination Solutions

Qn part (e.g. 4 a) iii)

Mark

b)

The maximal work is achieved when the process is quasistatic, that is, when the process is reversible. Then $\Delta S = 0$

$$\text{Recall } \Rightarrow \Delta S = \ln \frac{T^{C_1+C_2}}{T_1^{C_1} T_2^{C_2}} = 0 \Rightarrow$$

$$T^{C_1+C_2} = T_1^{C_1} T_2^{C_2} \Rightarrow T = T_1^{\frac{C_1}{C_1+C_2}} T_2^{\frac{C_2}{C_1+C_2}}$$

$$\text{Then } C_1(T_1 - T) = C_2(T - T_2) + W_{\max}$$

$$W_{\max} = C_1 T_1 + C_2 T_2 - (C_1 + C_2) T$$

$$W_{\max} = C_1 T_1 + C_2 T_2 - (C_1 + C_2) T_1^{\frac{C_1}{C_1+C_2}} T_2^{\frac{C_2}{C_1+C_2}}$$

$$\text{When } C_1 = C_2 = C$$

$$W_{\max} = C(T_1 + T_2 - 2\sqrt{T_1 T_2}) = C(\sqrt{T_1} - \sqrt{T_2})^2 > 0$$

when $C_1 \neq C_2$ $W_{\max} > 0$ as well, this can be proven, but not required

Question continued on next page

OR

Total Final Mark Allocation

33