

The liquid-vapour interface of QDO water

Flaviu Cipcigan Andrew Jones Jason Crain

Vlad Sokhan Glenn Martyna

The liquid-vapour interface of QDO water

1. Molecular models
2. The Quantum Drude Oscillator (QDO)
3. Introducing QDO water
4. Liquid-vapour interface of QDO water



ARCHITECTURE: 1950

ARCHITECTURE: NOW

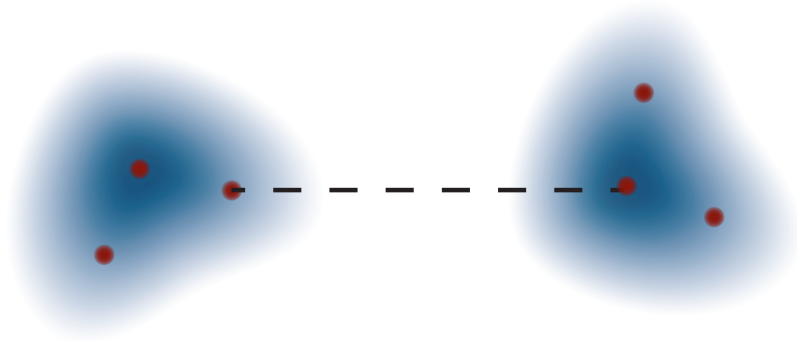


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Challenge

efficiently model intermolecular interactions



Challenge

efficiently model intermolecular interactions

Solution

begin with simple building blocks



Challenge

efficiently model intermolecular interactions

Solution

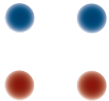
begin with simple building blocks and
assemble them into molecules



Molecular building blocks

Molecular building blocks

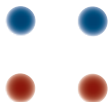
Electrostatics



Point charges

Molecular building blocks

Electrostatics



Point charges

Constraints



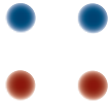
Translational



Rotational

Molecular building blocks

Electrostatics



Point charges

Constraints



Translational



Rotational

Non-Coulomb forces



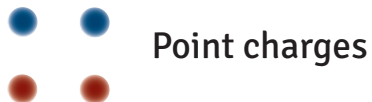
van der Waals



repulsion

Molecular building blocks

Electrostatics



Constraints



Non-Coulomb forces



Response (limited)

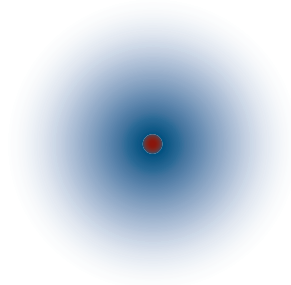


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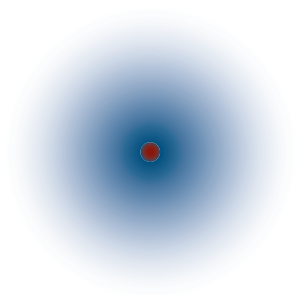
Quantum Drude Oscillator (QDO)

Light negative particle tethered harmonically
to a heavy positive, oppositely charged nucleus



Quantum Drude Oscillator (QDO)

Light negative particle tethered harmonically
to a heavy positive, oppositely charged nucleus



Free parameters

μ reduced mass

ω spring frequency

q charge

Quantum Drude Oscillator: Response

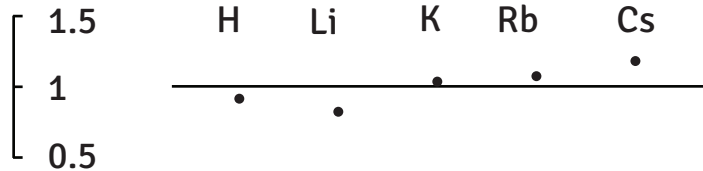
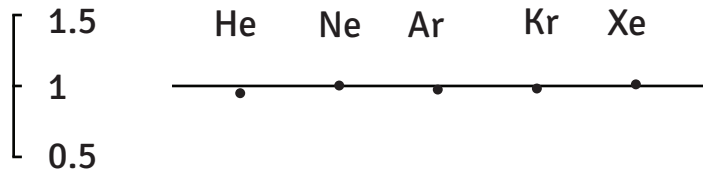
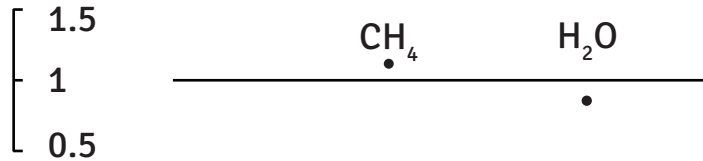
$$\text{Polarisation} \quad \alpha_l = \frac{\left[\frac{q^2}{\mu\omega^2} \right] \left[\frac{(2l-1)!!}{l} \right] \left[\frac{\hbar}{2\mu\omega} \right]^{l-1}}{\text{dipole}}$$

Quantum Drude Oscillator: Response

Polarisation $\alpha_l = \frac{\left[\frac{q^2}{\mu\omega^2} \right] \left[\frac{(2l-1)!!}{l} \right] \left[\frac{\hbar}{2\mu\omega} \right]^{l-1}}{\text{dipole}}$

Dispersion $C_6 = \frac{3}{4}\alpha_1\alpha_1\hbar\omega \quad \bigg| \quad \text{dipole-dipole}$
 $C_8 = 5\alpha_1\alpha_2\hbar\omega \quad \bigg| \quad \text{dipole-quadrupole}$

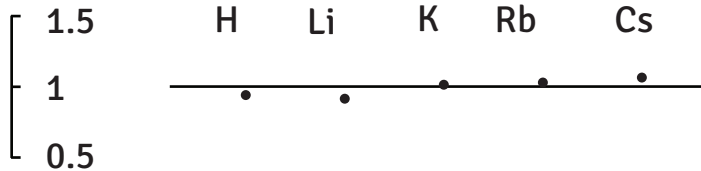
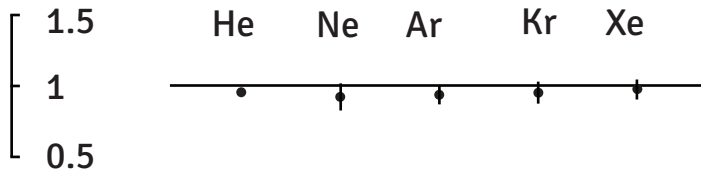
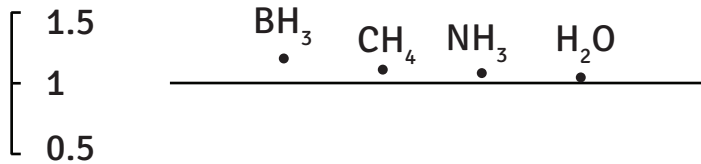
Quantum Drude Oscillator: Invariants



Polarisation

$$\sqrt{\frac{20}{9}} \frac{\alpha_2}{\sqrt{\alpha_1 \alpha_3}} = 1$$

Quantum Drude Oscillator: Invariants

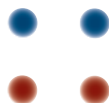


Dispersion

$$\sqrt{\frac{49}{40}} \frac{C_8}{\sqrt{C_6 C_{10}}} = 1$$

Molecular building blocks

Electrostatics



Point charges

Constraints



Translational



Rotational

Non-Coulomb forces



van der Waals



repulsion

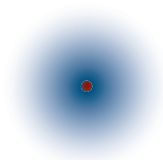
Response



Fluctuating dipole



Charge transfer



Quantum Drude Oscillator

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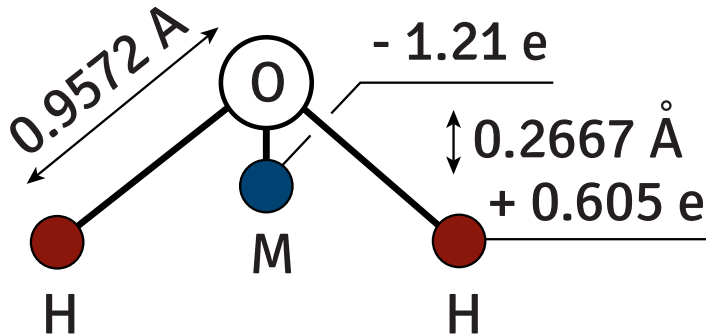
3. Introducing QDO water

Frame QDO Repulsion Damping Sampling

4. Liquid-vapour interface of QDO water

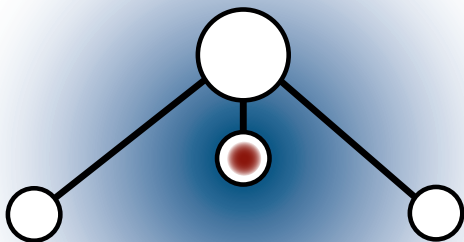
1. Frame

ground state moments



2. QDO

molecular response



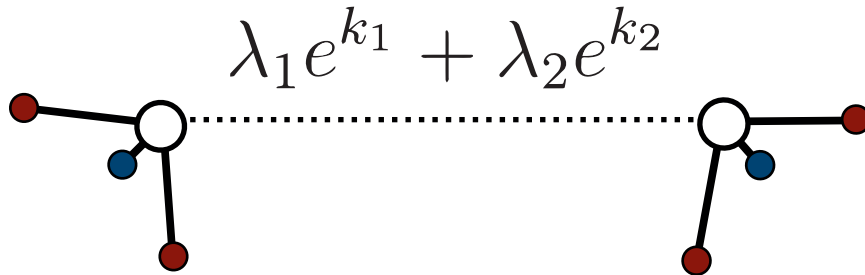
$$\mu = 0.3656 \text{ amu}$$

$$\omega = 0.6287 \omega_h$$

$$q = -1.1973 \text{ e}$$

3. Repulsion


Short range correction



4. Electrostatic Damping

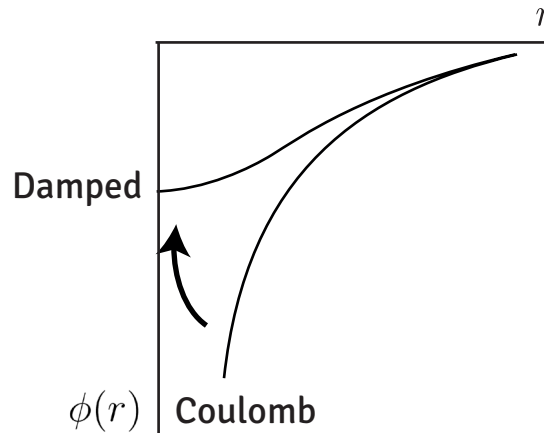
Short range correction

Gaussian charges


$$-\frac{\text{erf}(\gamma r)}{r}$$

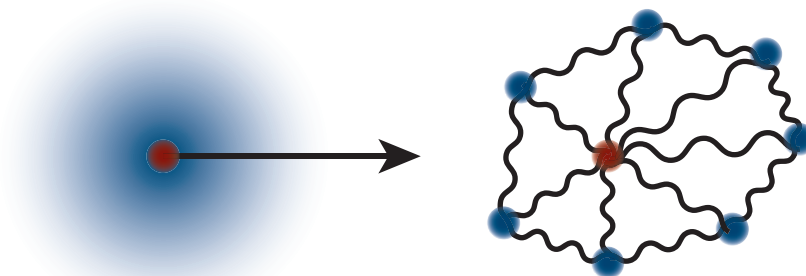
4. Electrostatic Damping

Short range correction



5. Efficient Sampling

Path Integral Molecular Dynamics

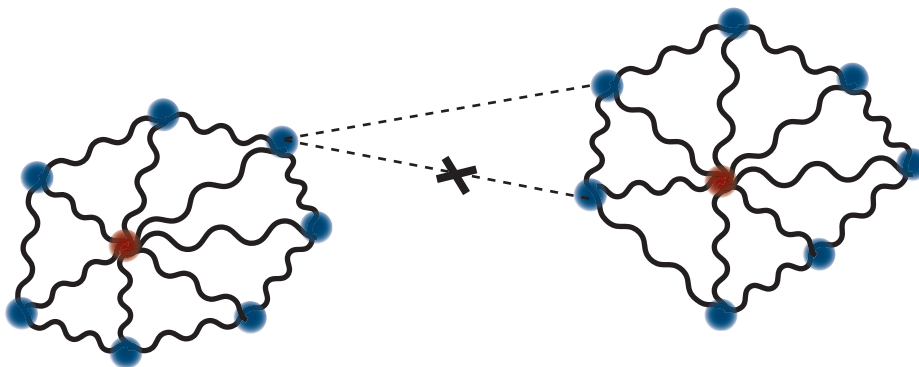


replace electron

with chain of beads

5. Efficient Sampling

Path Integral Molecular Dynamics



no cross interactions between beads

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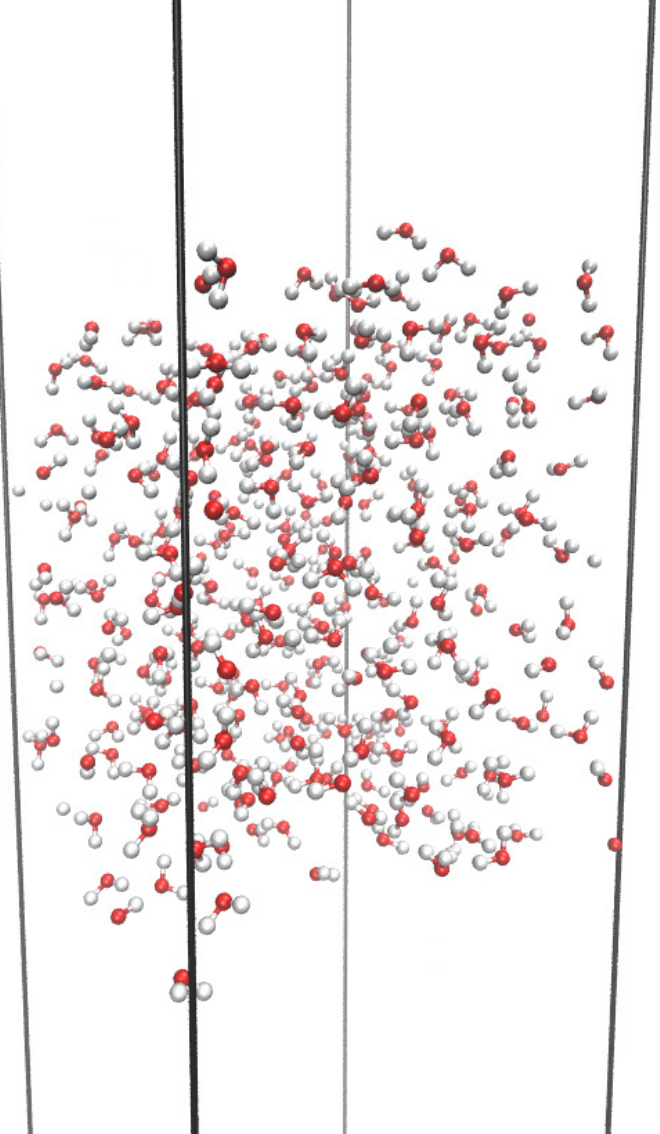
1. Setup

300 QDO-water molecules

Periodic boundaries

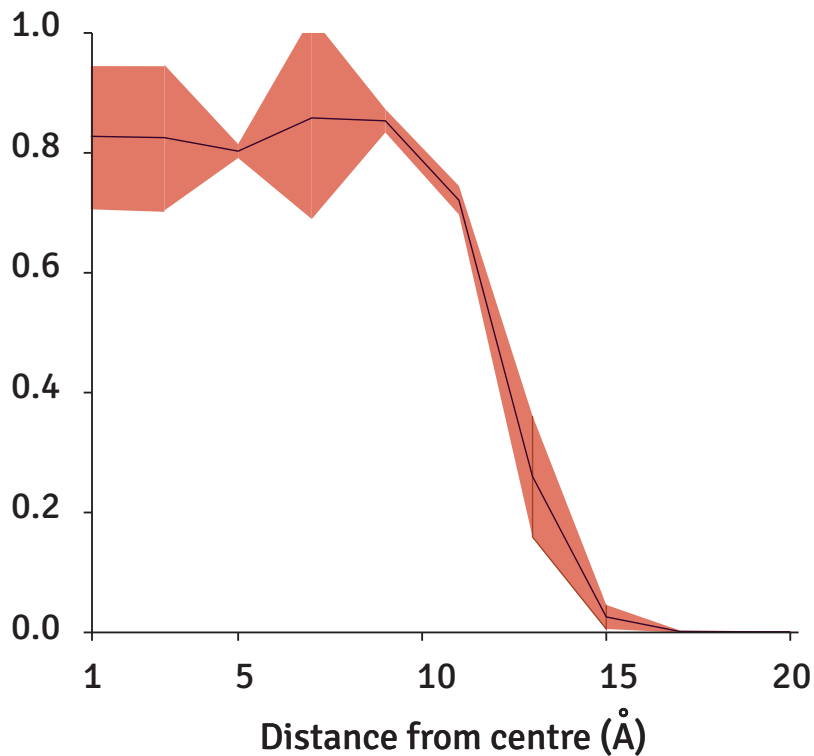
Unit cell

$20.80126 \times 20.80126 \times 80 \text{ \AA}^3$

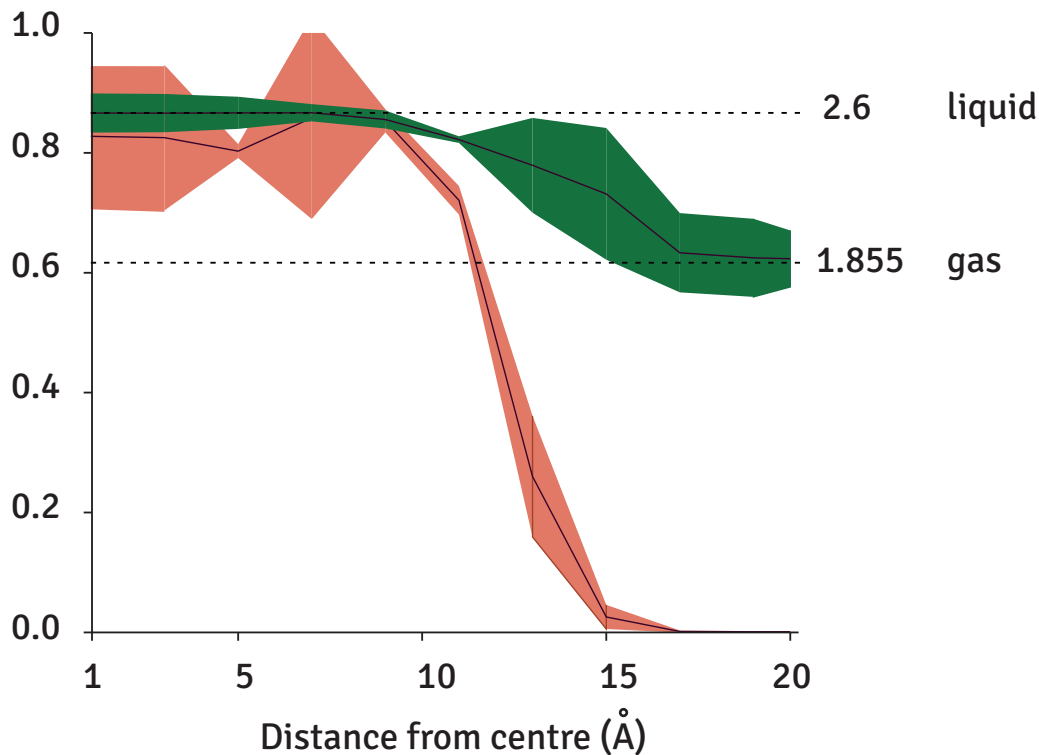


2. Density

(g·cm⁻³)

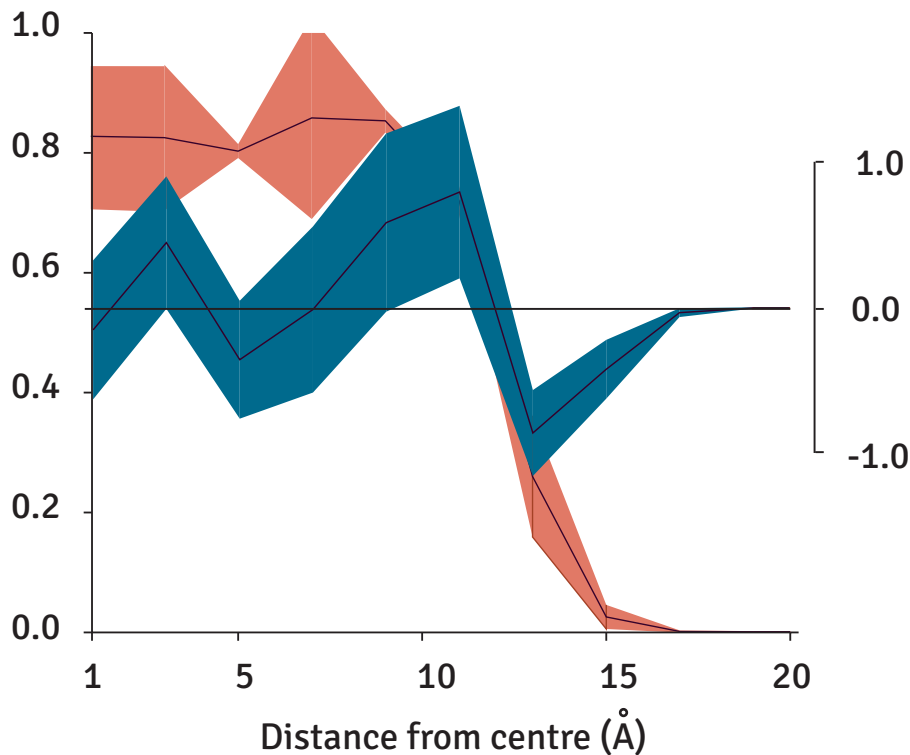


3. Dipole moment (Debye)



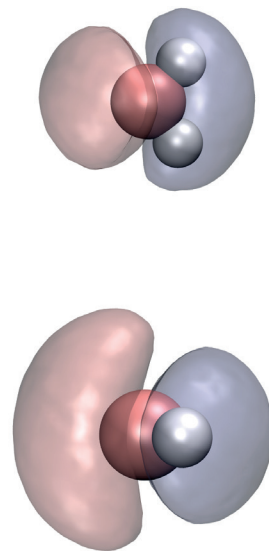
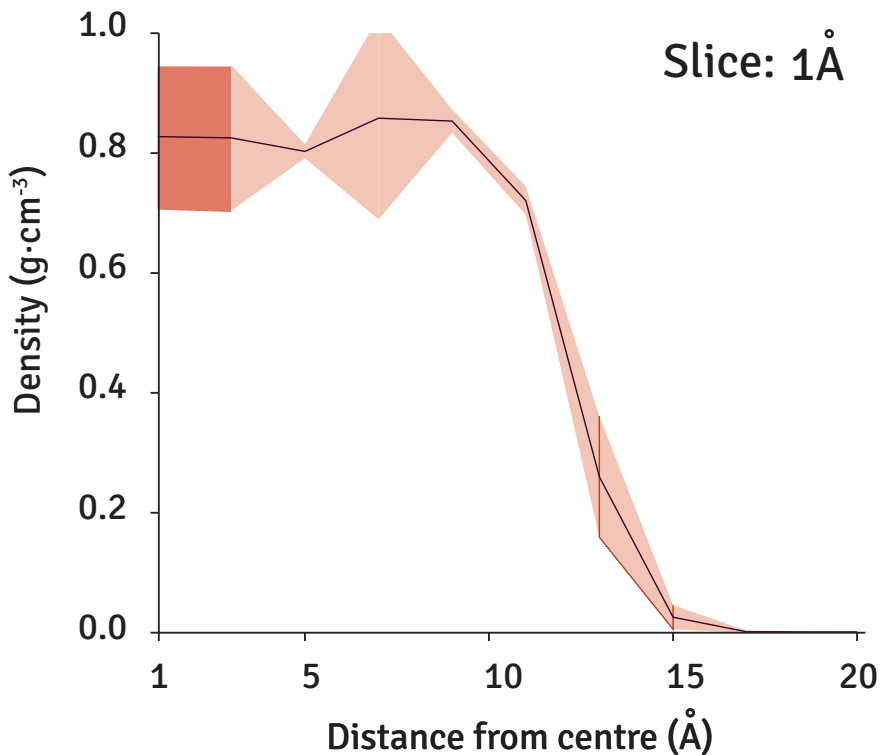
3. Surface charge density ●

($e \cdot \text{\AA}^{-3} \cdot 10^{-3}$)



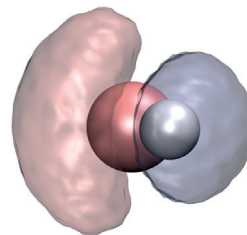
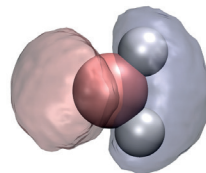
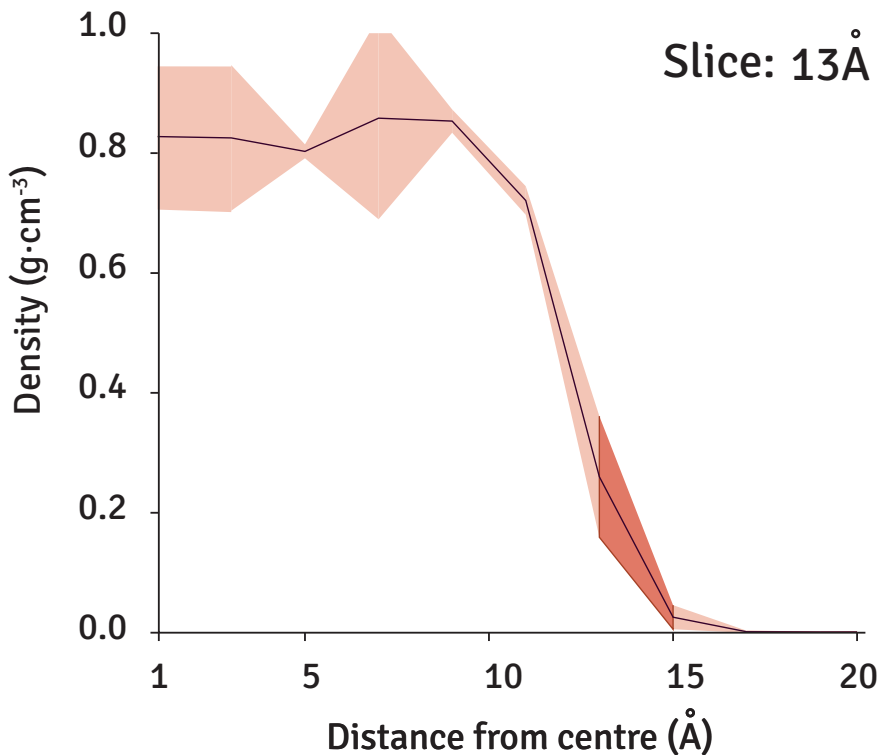
4. Electronic distribution

(minus gas phase ground state charge density)



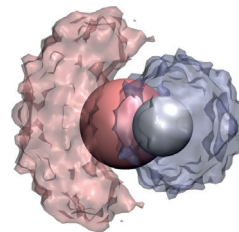
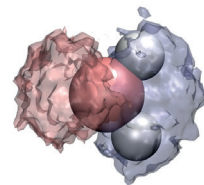
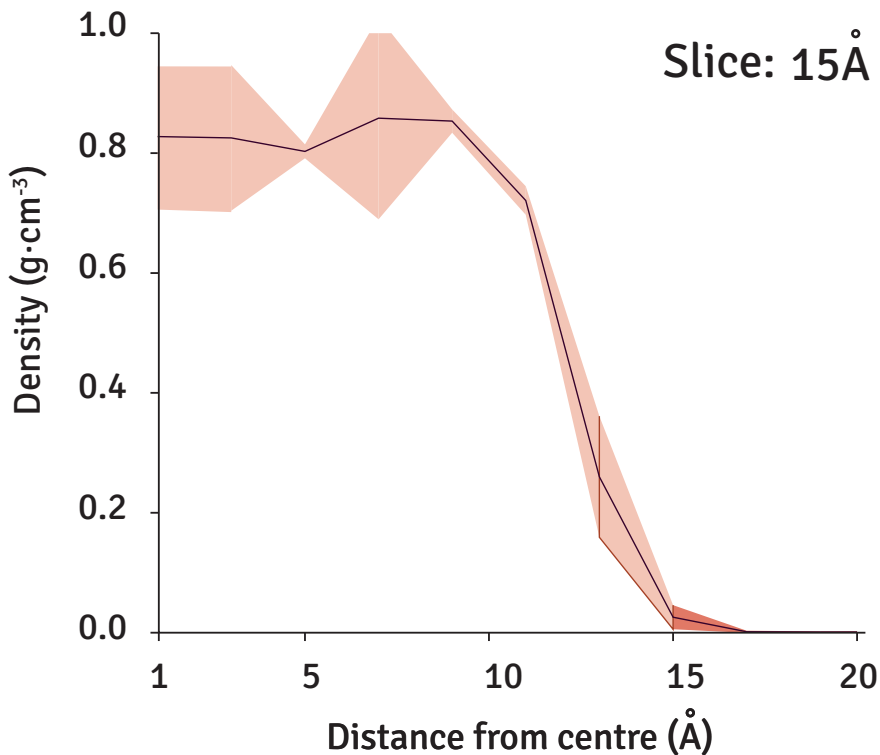
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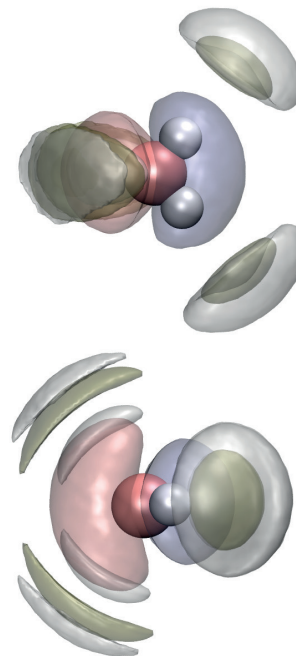
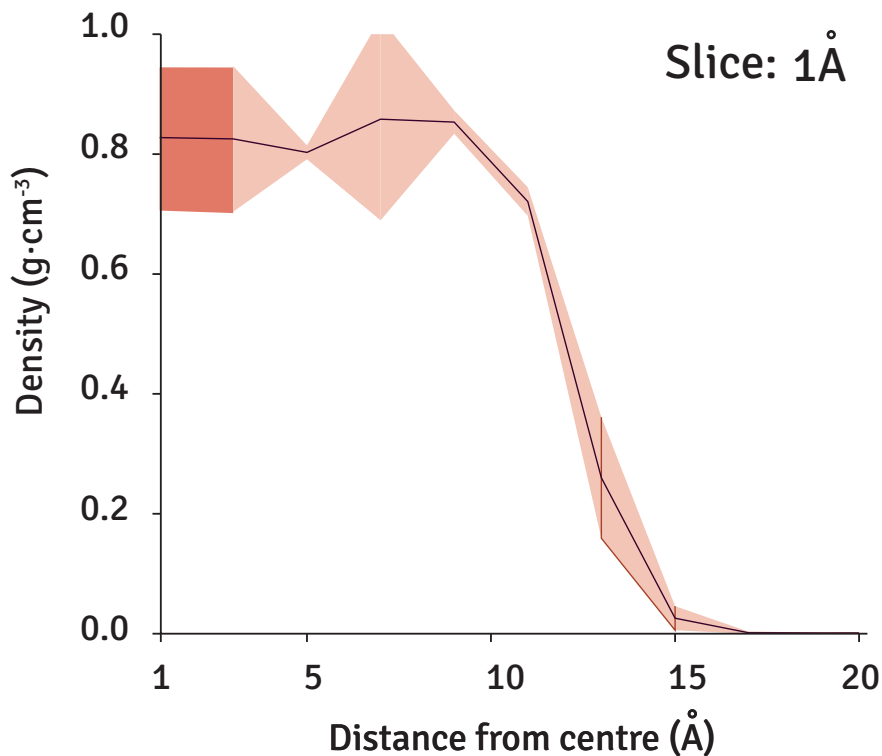


4. Electronic distribution

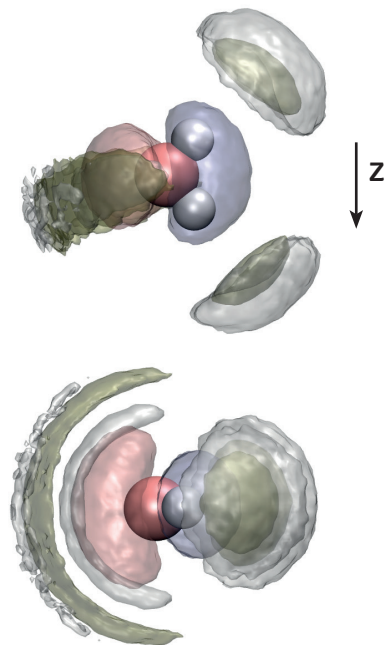
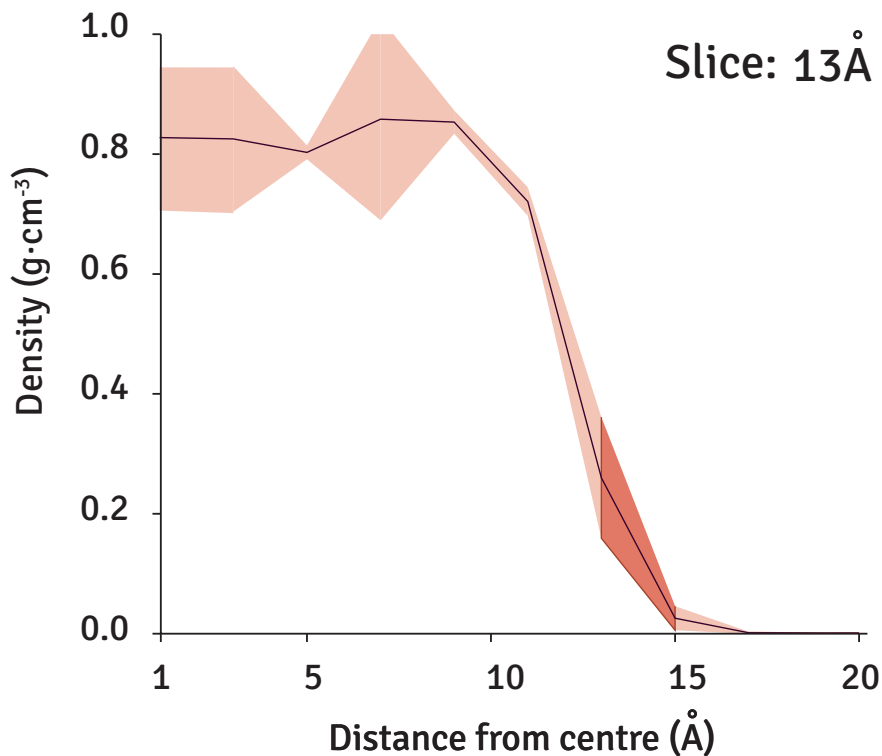
(minus gas phase ground state charge density)



5. Nearest neighbour distribution



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Conclusions

QDOs are an accurate model for long range forces

QDO water has a physical liquid-vapour interface

Next steps

Exploration of the interface's structure and the effects of dispersion and polarisation on physical properties

References

A. Jones, Quantum drude oscillators for accurate many-body intermolecular forces, PhD thesis, The University of Edinburgh

A. Jones, F. Cipcigan, V. Sokhan, J. Crain, G. Martyna, Electronically coarse grained water, PRL (under review)





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Liquid radial distribution function

