The Drude Model - "Free electron" theory of metals Metals: > excellent electrical * thermal ductile (not brittle) high lustre (shiney) - Drude Model -> almost, but-not-completely useless (think v=d of SSP) Basic Idea! Free Electron Gas' & = s > gas of e's' - apply theory of gases to + free é's bonne around Remember PV=nk

delocalized valence e's conduction e's

Note: - Structure of the atom was unknown at the time (1900)

- Model assumes stationary (+) charges to balance the current-currying e's

How dense is our e gas? -> about the same as (ie. # e's/cm3) the metal (atoms/cm3)

n= N = # atoms x # moles x # valence e

n= NA gm Z

Pm -> Avogadro's # 6.022×10

Pm -> mass density

Z -> # valence e

A -> atonic mass

n~ 1022 e / cm3

-> Compare to ideal gas @ STP: 1~10 atom/em3

- Unsurprisingly, this corresponds to a volume per é similar to the Bohr atom.

- Important Assumptions of the Drude Medel
 - 1.) Between collisions:
 - a) Neglect e-e interactions Independent Electron Approx.
 - b) Neglect e-ion interactions Free Electron Approx.
 - c) Newton's laws govern é movement
 - 2) collisions are instantaneous & after the e velocity
 - 3) Average time between collisions is 2:

 Probability that e collides in time dt is a dt
 - 4.) Electrons achieve thermal equilibrium to surroundings only through collisions
 - a) velocity of e after collision is NOT related to it's velocity before the collision
 - b) direction of e after collision is random
 - c) magnitude of velocity related to temperature:

hot: v 1

edd: v t

He will use the Drude Model to 'explain'

- 1) DC electrical conductivity
- 2) Hall Effect & Magnetores: stance
 - 3) AL electrical conductivity
 - 4) Thermal conductivity