

Note: - Structure of the atom was unknown at the time (1900) - Model assumes stationary (+) charges to balance the current-currying e's How dense is our é gas? -> about the same as (ie. It é's/cm³) the metal (atoms/cm³) (ie. # e's/cm3) n= N = # atoms x # moles x # Valence e NA -> Avogadrols # 6.022 x10 n=NA gm Z ?m -> mass density Z -> # valence e A - atomic mass n~ 1022 e / cm3 > Compare to ideal gas @ STP: 1 ~ 10 atom/cm3 - Unsurprisingly this corresponds to a volume per e similar to the Bohr atom.

Important Assumptions of the Drude Medel

- 1.) Between collisions:
  - a) Neglect e-e interactions

Independent Electron Approx.

b) Neglect e-ion interactions

Free Electron Approx.

- c) Newton's laws govern e movement
- 2) collisions are instantaneous & after the
- 3.) Average time between coll-310005 ; 5 2:

Probability that e collides in time dt is a dt

- 4.) Elections achieve thermal equilibrium was surroundings only through collisions
  - a) velocity of e efter collision is NOT related to it's velocity before the collision
  - b) direction of e after collision is random
  - c) magnitude of velocity related to temperature:

hot: v 1

edd: V

We will use the Drude Model to 'explain' 1) DC electrical conductivity 2) Hall Effect & Magnetores: stance 3) AC electrical conductivity 4) Thermal conductivity