

Solution:

According to Bohr Model, the e is revolving around the nucleus which is at rest. However, in reality the nucleus of electron both are notating about a common centre of mass.

Nucleus
$$\omega = b\omega$$

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Take
$$a+b=r$$
,

 $ma = Mb \Rightarrow a = M/mb$
 $a+b=(\frac{M}{m}b+b)$

$$a+b=b\left(\frac{M+m}{m}\right)$$

$$b=\left(\frac{m}{M+m}\right)(a+b)$$

This Coulomb force is responsible for e and nucleus to rotate about the centre of mass.

For
$$e^- = F = M\omega^2 a$$

For nucleus $F = M\omega^2 b = M\omega^2 \left(\frac{m}{m+M}\right) (a+b)$
 $= M\omega^2 (a+b)$
 $= \omega^2 (a+b)$
 $= \omega^2 (a+b)$
 $= \omega^2 (a+b)$

Total Energy
$$E = \frac{1}{2} m \omega^{2} a^{2} + \frac{1}{2} M \omega^{2} b^{2} - \frac{Ze^{2}}{4\pi \epsilon_{0} (a+b)}$$

$$= \frac{1}{2} H r^{2} \omega^{2} - H r \omega^{2}$$

$$E = - H r^{2} \omega^{2}$$