

BB 101: MODULE II  
***PHYSICAL BIOLOGY***

## Review of Lecture 7

- Microtubules and Actin filaments are polymers of tubulin and actin monomers
- Dynamics of Microtubule and Actin Filaments
- Treadmilling of microtubule and actin filament
- A simple model for cytoskeletal filament polymerization, treadmilling and dynamic instability

# Proteins and their structures

- Tubulin and actin monomers are protein molecules
- Proteins can be of different shapes

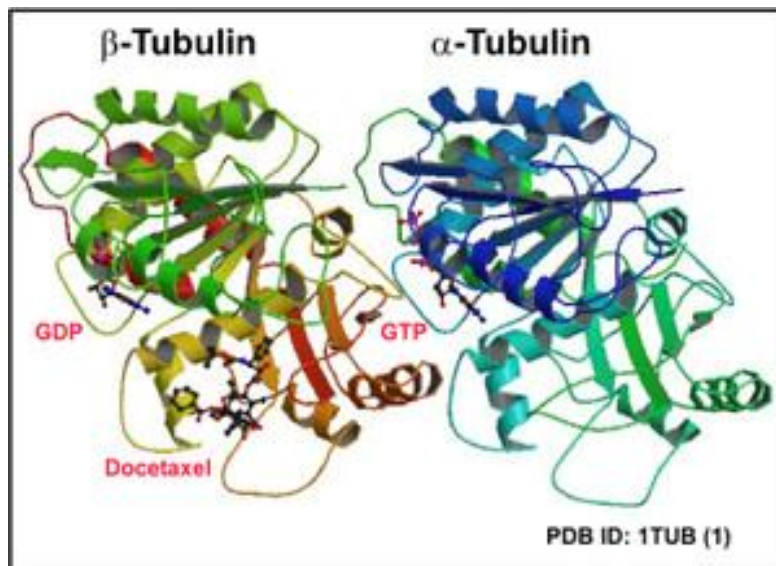


Figure Source: <http://www.mastcell-basophil.net/wiki/wiki-start/microtubules-and-mast-cell-signaling/>

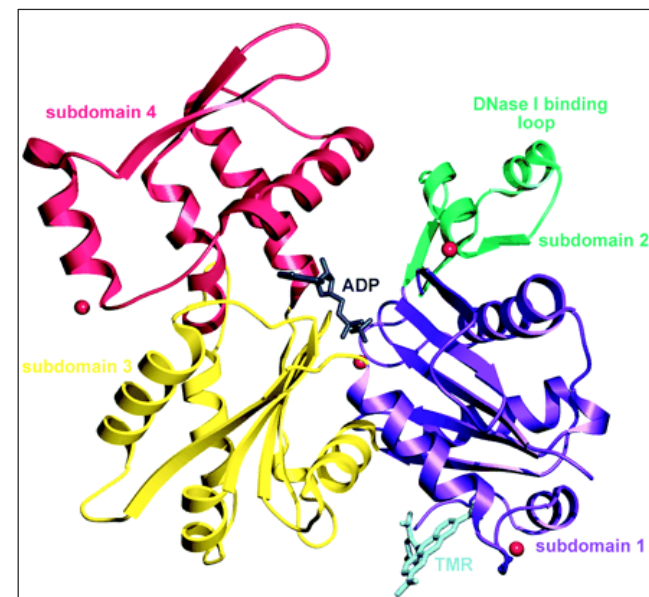


Figure Source: Otterbein et al., Vol. 293, 708-711 (2001)

# Proteins and their structures

- Proteins perform their function by folding into different shapes
- Proteins are sequence of amino acids
- Given a sequence of amino acids can be predict the structure of the protein?

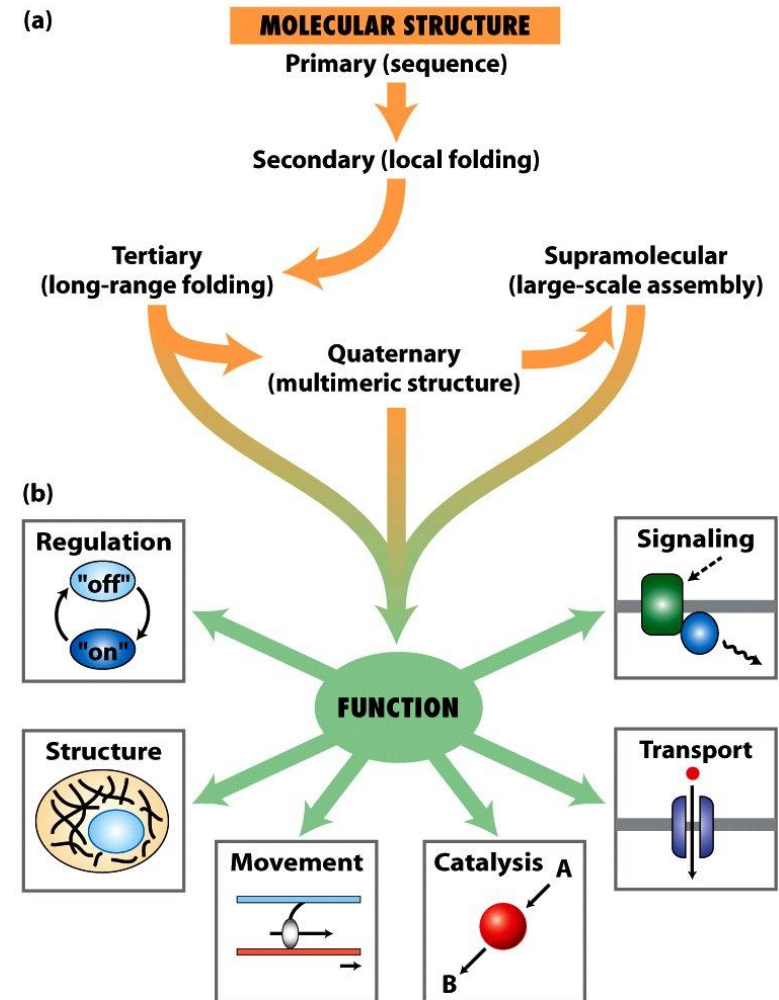


Figure 3-1  
*Molecular Cell Biology, Sixth Edition*  
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# Proteins and their structures

## (a) Primary structure

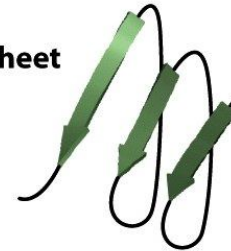
– Ala – Glu – Val – Thr – Asp – Pro – Gly –

## (b) Secondary structure

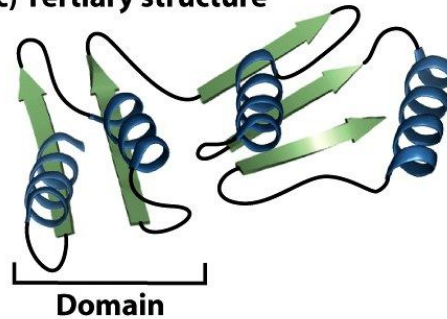
$\alpha$  helix



$\beta$  sheet



## (c) Tertiary structure



## (d) Quaternary structure

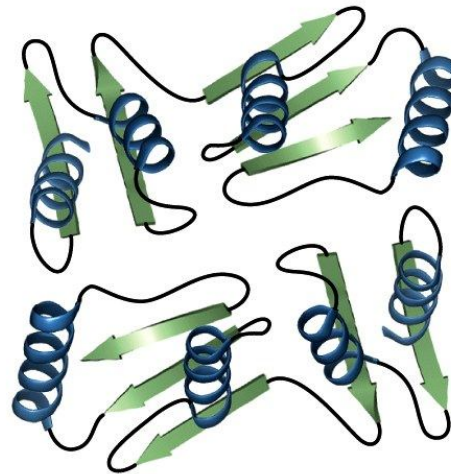


Figure 3-2  
*Molecular Cell Biology, Sixth Edition*  
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# Proteins and their structures

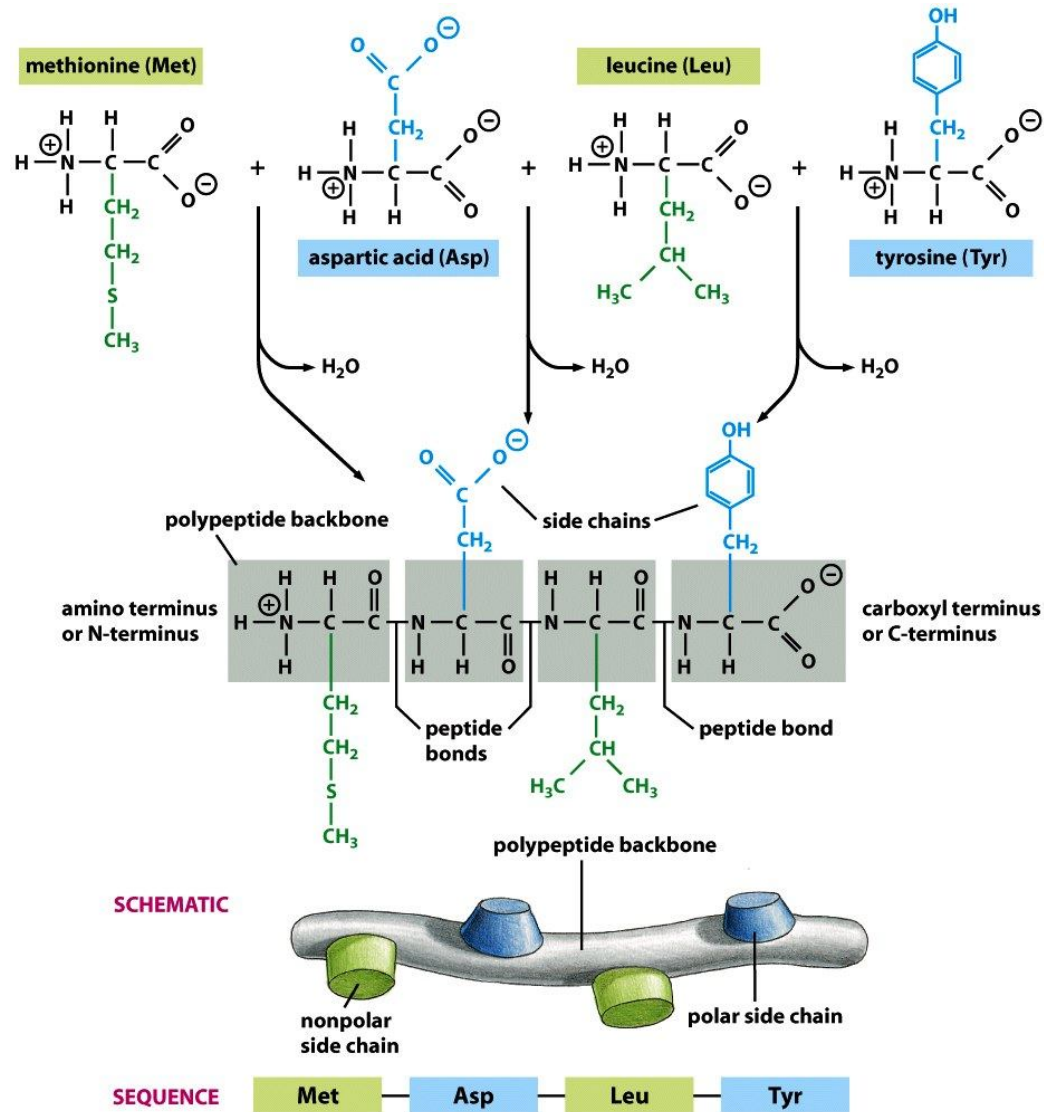


Figure 3-1 *Molecular Biology of the Cell* (© Garland Science 2008)

# Proteins and their structures

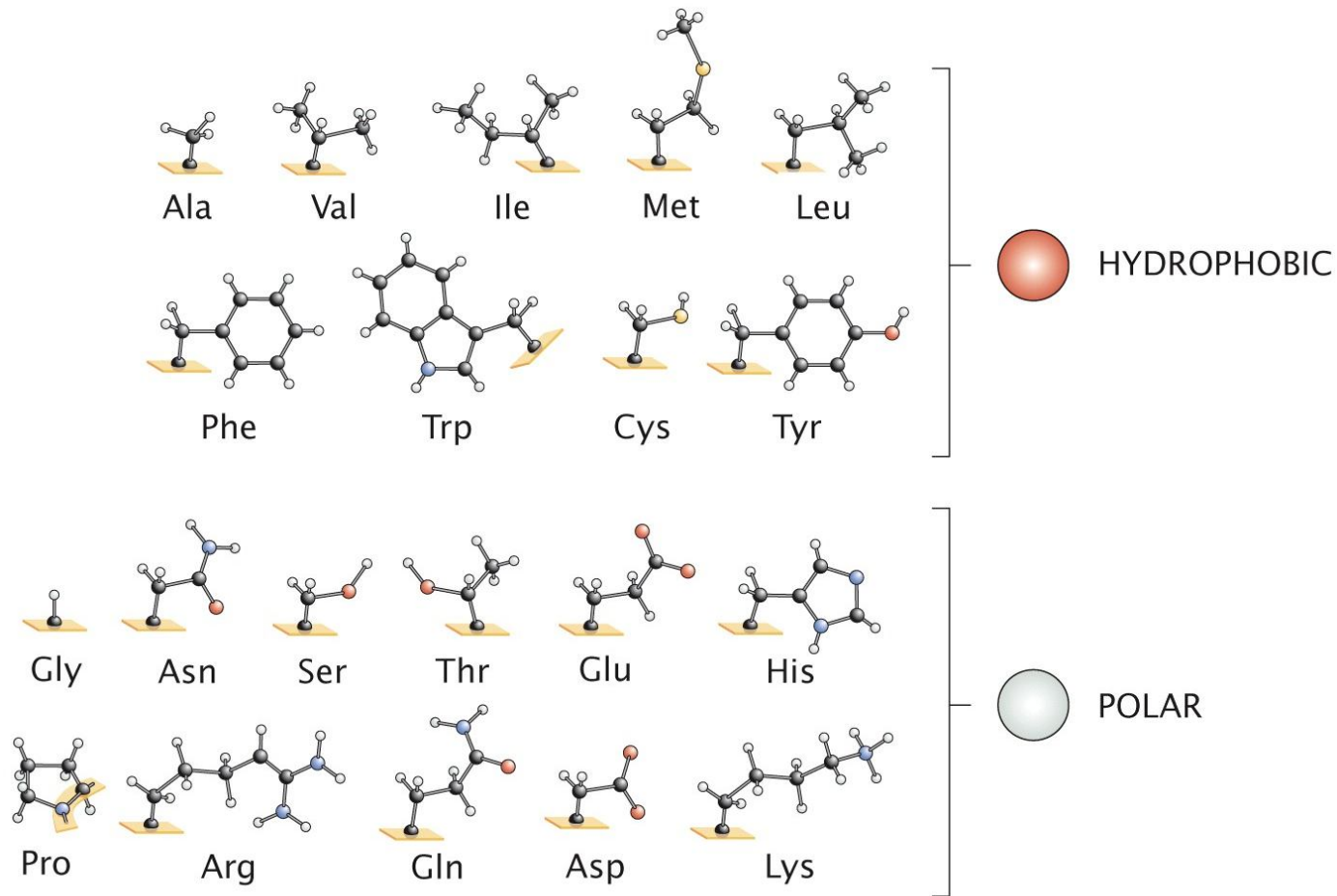


Figure 8.28 Physical Biology of the Cell, 2ed. (© Garland Science 2013)

# Protein Structures and Folding

- Proteins are sequence of amino acids
- **Given a sequence of amino acids can be predict the structure of the protein?**
- Classical mechanics tell us that protein would prefer the structure/conformation which minimizes energy
- However, this is not true when thermal fluctuations are present



# Protein Structures are free energy minimizers

- In presence of thermal fluctuations, a protein folds into the structure which minimizes free energy, out of all the possible ways that a particular chain of amino acids can fold up
- What is free energy?
- Helmholtz free energy
- Gibbs free energy

$$A = U - TS$$

$$G = H - TS$$

$$H = U + pV$$

# Free Energy

- Free energy=Energy- Temperature x Entropy
- Free energy can be minimized by either decreasing energy or increasing entropy
- In order to calculate free energy we have to calculate entropy
- **What is entropy? How to calculate entropy?**

# Entropy



# Micro-state and Macro-state

- Entropy is a measure of the microscopic degeneracy of a macroscopic state (“macro-state”)
- In other words, entropy can be computed by counting the number of possible microscopic arrangements/states (“micro-states”) for a given macroscopic state (“macro-state”)

$$S = k_B \ln W$$

$W$  is the number of possible microscopic arrangements (or “micro-states”)

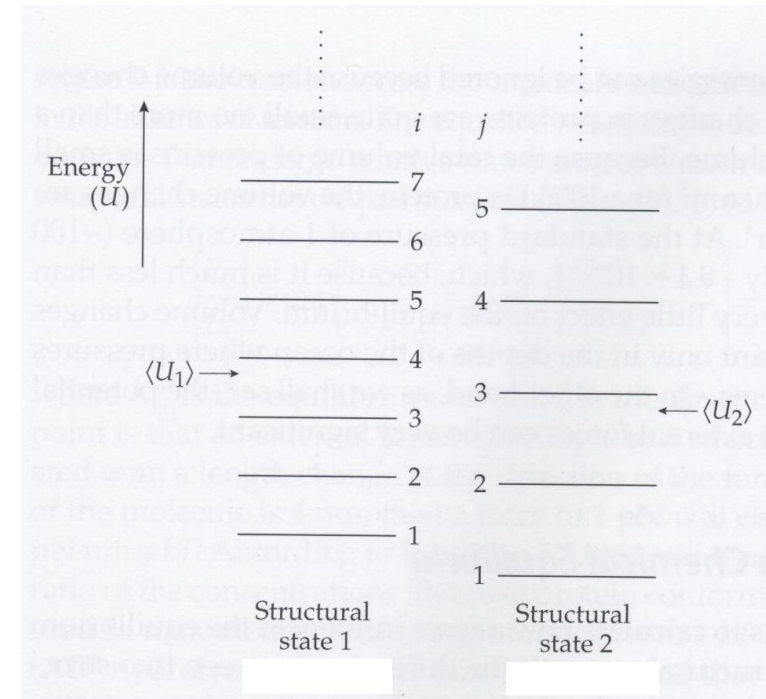
# Micro-state and Macro-state

- Consider a protein that can exist in two **structural states**. For example open (1) and closed (2) ion channel protein.
- However, due to thermal fluctuations, these structural states consists of enormous number of **conformations states**, where a conformational state is a set of positions of all the atoms
- Structural states  $\Rightarrow$  “Macro-states”
- Conformational states  $\Rightarrow$  “Micro-states”

# Micro-state and Macro-state

- We know that probability of finding protein any microstate  $i$  or  $j$  is given by Boltzmann law

$$p_i = \frac{1}{Z} e^{-\frac{U_i}{k_B T}} \quad p_j = \frac{1}{Z} e^{-\frac{U_j}{k_B T}}$$



- What is the probability of finding protein in a given macrostate?
- The probability of finding protein in a given macrostate is given sum of all  $p_i$  or  $p_j$

## Micro-state and Macro-state

- Thus, Probability of finding in a macrostate  $X$  is given by

$$p(X) = \sum_{i \in X} p_i = \sum_{i \in X} \frac{1}{Z} e^{-\frac{U_i}{k_B T}}$$

- Using relation  $G = -k_B T \ln Z$  between partition function and free energy it can be shown that

$$\boxed{p(X) = \frac{1}{Z} e^{-\frac{G(X)}{k_B T}}} \quad G(X) = -k_B T \ln \left( \sum_{i \in X} e^{-\frac{U_i}{k_B T}} \right)$$

- Similar to Boltzmann formula for microstate, energy replaced by free energy of macrostate

# Proof of $G = -k_B T \ln Z$

$$\left. \begin{aligned} G &\equiv \langle U \rangle - TS \\ \langle U \rangle &= \sum_{i=1}^N U_i P_i \\ S &= -k \sum_{i=1}^N p_i \ln p_i \end{aligned} \right\} \Rightarrow G = \sum_{i=1}^N U_i P_i + kT \sum_{i=1}^N p_i \ln p_i = \sum_{i=1}^N P_i \left\{ U_i + kT \ln p_i \right\}$$

$$\left. \begin{aligned} P_i &= \frac{1}{Z} \exp \left[ -\frac{U_i}{kT} \right] \end{aligned} \right\}$$

$$\Rightarrow G = \sum_{i=1}^N \frac{1}{Z} \exp \left[ -\frac{U_i}{kT} \right] \left\{ U_i + kT \ln \left( \frac{1}{Z} \exp \left[ -\frac{U_i}{kT} \right] \right) \right\} = \frac{1}{Z} \sum_{i=1}^N \exp \left[ -\frac{U_i}{kT} \right] \left\{ U_i - kT \ln Z - U_i \right\}$$

$$\left. \begin{aligned} Z &= \sum_{i=1}^N \exp \left[ -\frac{U_i}{kT} \right] \end{aligned} \right\}$$

$$\Rightarrow G = -kT \ln Z \Rightarrow Z = \exp \left( -\frac{G}{kT} \right)$$



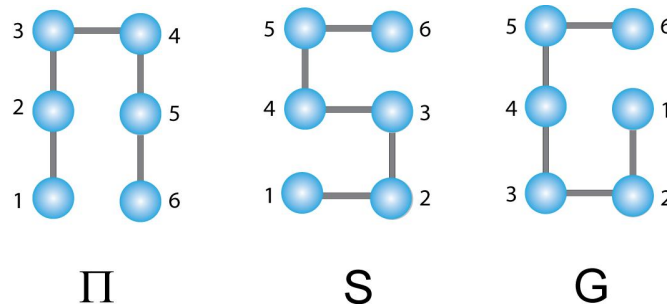
## So far...

- Proteins can exist in many structural states “macrostates”
- Each “macrostate” consists of many “microstates”
- “macrostate” with minimum free energy is preferred
- $G = H - TS$ ,  $G = -k_B T \ln Z$  and  $S = k_B \ln W$
- Probability of finding a microstate  $X$  is given by

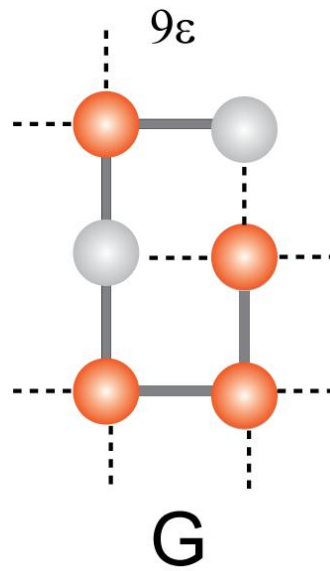
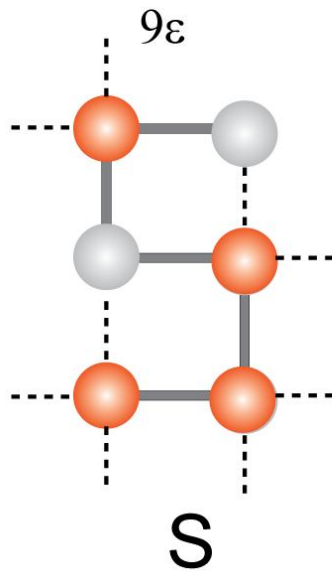
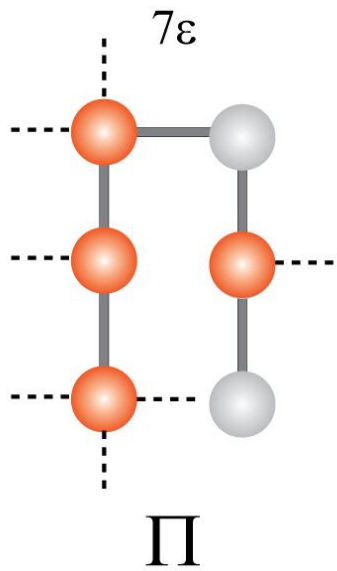
$$P(X) = \frac{1}{Z} e^{-\frac{G(X)}{k_B T}}$$

# HP models of protein folding

- Consider a protein consisting of six amino acids consisting of H and P
- The sequence of the amino acids is HHHPHP
- Suppose that can fold into three possible structures  $\Pi$ , S and G such that energy increases by  $\varepsilon$  for every contact of H with either P or solvent molecule
- What will be the structural state of the protein?



# HP Models of Protein Folding



$$G_{\Pi} = 7\varepsilon - k_B T \ln 1$$

$$G_{others(S \text{ or } G)} = 9\varepsilon - k_B T \ln 2$$

$\Pi$  Structure will be preferred as long as  $G_{\Pi} < G_{others}$

Or,

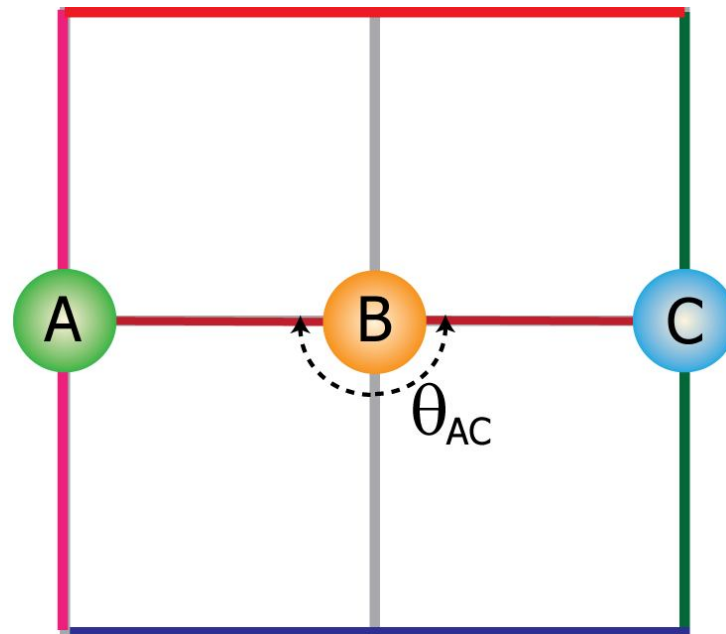
$$7\varepsilon - k_B T \ln 1 < 9\varepsilon - k_B T \ln 2$$

Or,

$$T < \left( \frac{\varepsilon}{k_B} \right) \left( \frac{2}{\ln 2} \right)$$

## Another Toy Models of Protein Folding

- Consider a protein consisting of three amino acids A, B and C connected with bonds of equal length
- This protein is placed on a square grid such that only A and C are free to rotate in a plane such that the bonds are always aligned along the grid lines



## Another Toy Models of Protein Folding

- Bending energy of the protein is given by

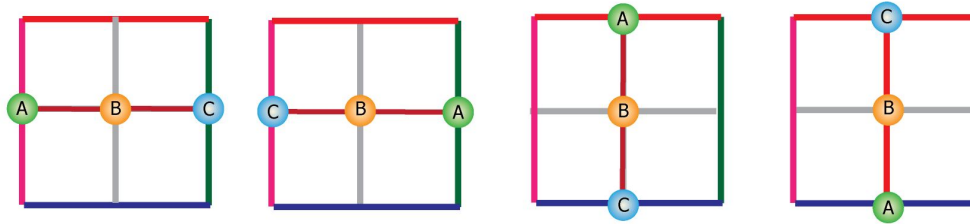
$$\epsilon = A(1 + \cos\theta_{AC})$$

**Whether the structural state of the protein will be bent or straight?**

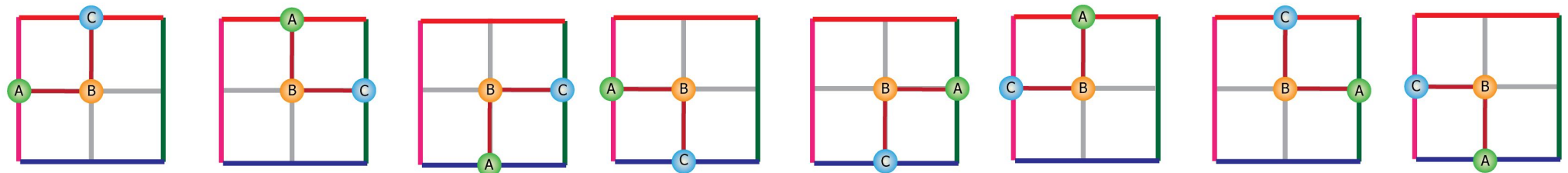
- What is the energy of the straight conformation?
- What is the energy of the bent conformations?
- Classical Mechanics: It should be in straight state

# Number of Straight and Bent conformations

Straight



Bent

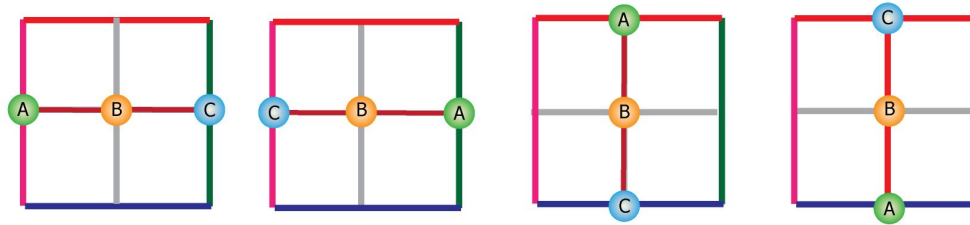


- No. of straight conformations=4
- No. of bent conformations=8

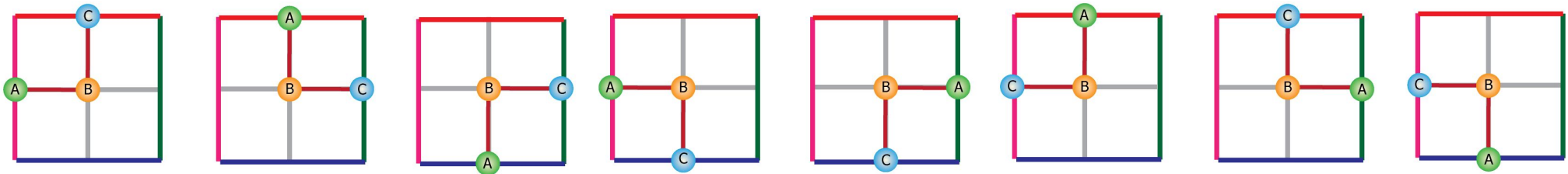
# Calculate Average Bending Energy

$$\epsilon = A(1 + \cos\theta_{AC})$$

Straight



Bent



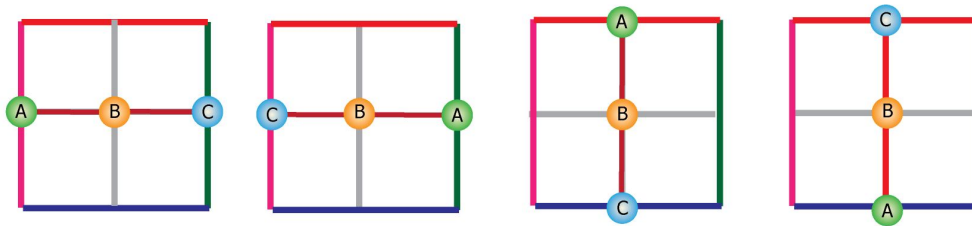
$$\epsilon_{Straight} = 0$$

$$\epsilon_{bent} = A$$

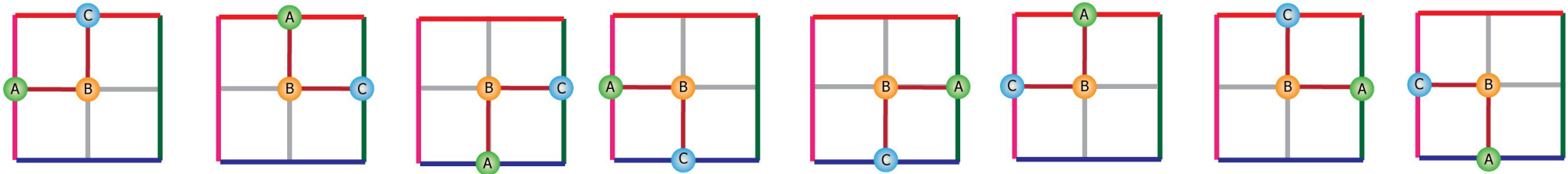
# Calculate Entropy

$$S = k_B \ln W$$

Straight



Bent



$$W_{straight} = 4$$

$$S_{straight} = k_B \ln 4 = 1.4 k_B$$

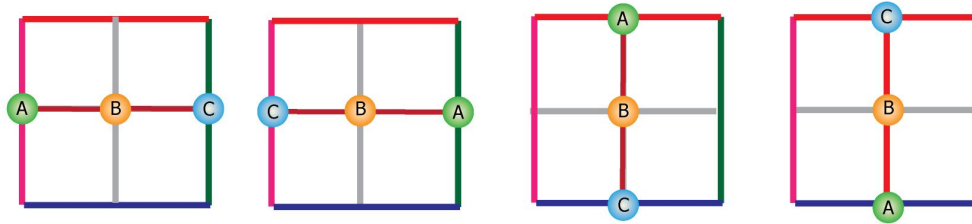
$$W_{bent} = 8$$

$$S_{bent} = k_B \ln 8 = 2.1 k_B$$

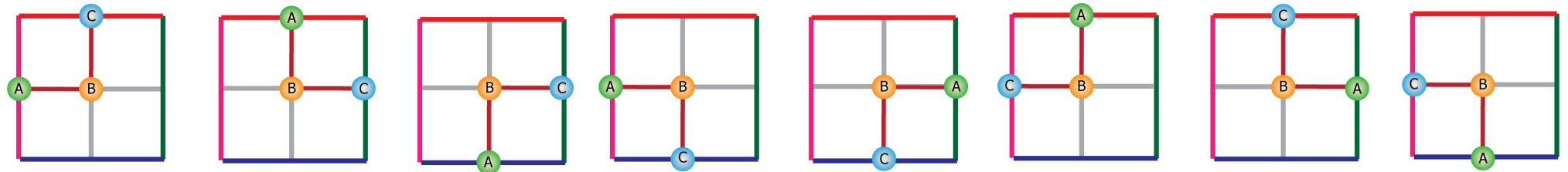


# Calculate Free Energy

Straight



Bent



$$G_{straight} = 0 - 1.4 k_B T$$

$$G_{bent} = A - 2.1 k_B T$$

# Bent or Straight

$$G_{Straight} = 0 - 1.4 k_B T$$

$$G_{bent} = A - 2.1 k_B T$$

- When  $A = 0.1 k_B T$  bent structure is preferred
- However, when  $A = 1.5 k_B T$ , straight structure is preferred

# Protein folding in reality

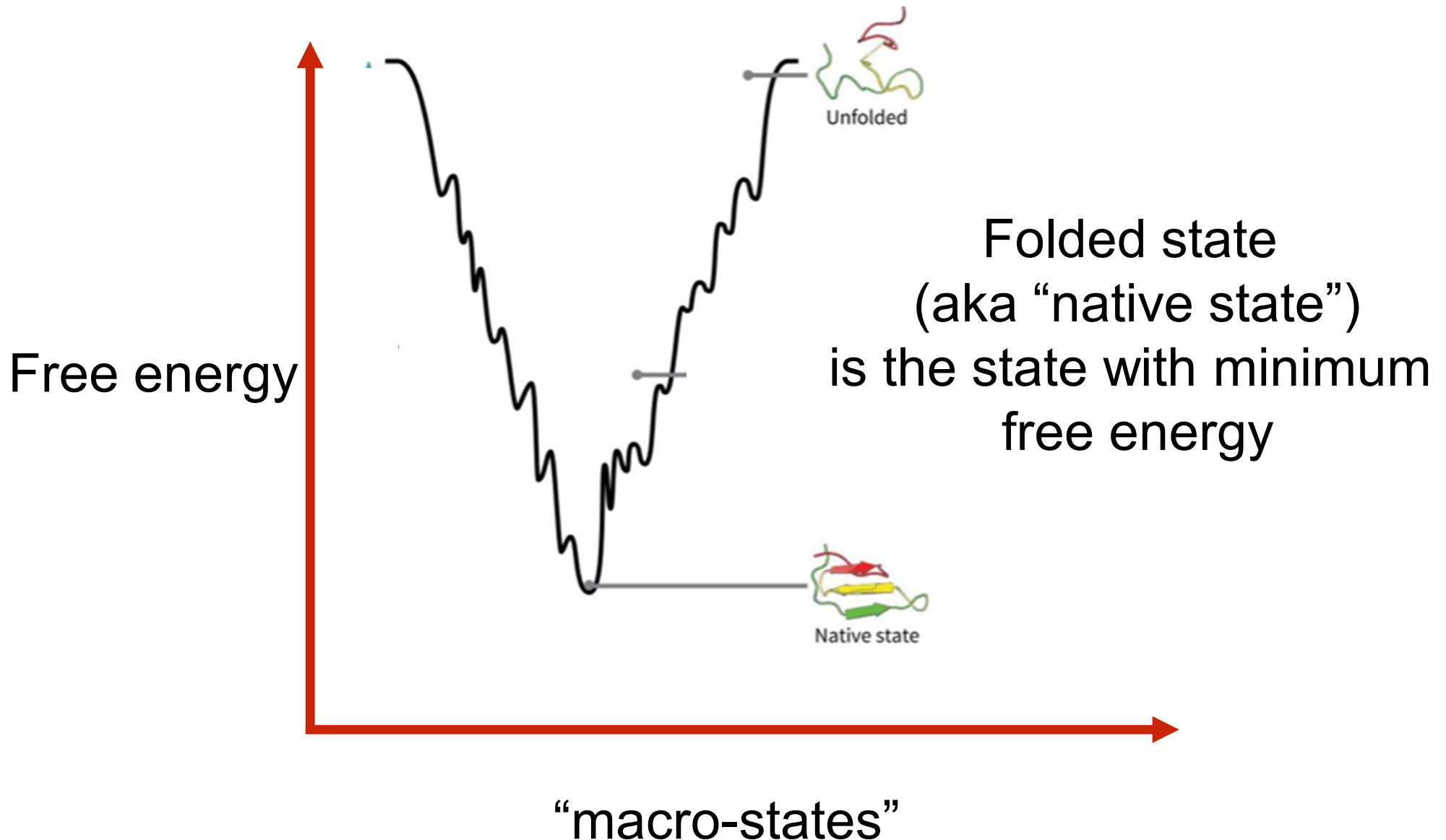
*Here we considered only toy models for protein folding*

*However, in reality*

- Protein monomers have many types of interaction: electrostatic, bending, Van der Waals etc
- Protein monomers interact with water (hydrophobic/hydrophilic)
- Energy/Enthalpy is more complicated than simple bending example we discussed
- One has to worry about entropy of the whole system (protein monomers+water+other ions like  $\text{Na}^+$  and  $\text{Cl}^-$ )

# Protein folding in reality

Typical proteins “see” such a free energy landscape



# Ramachandran Plot

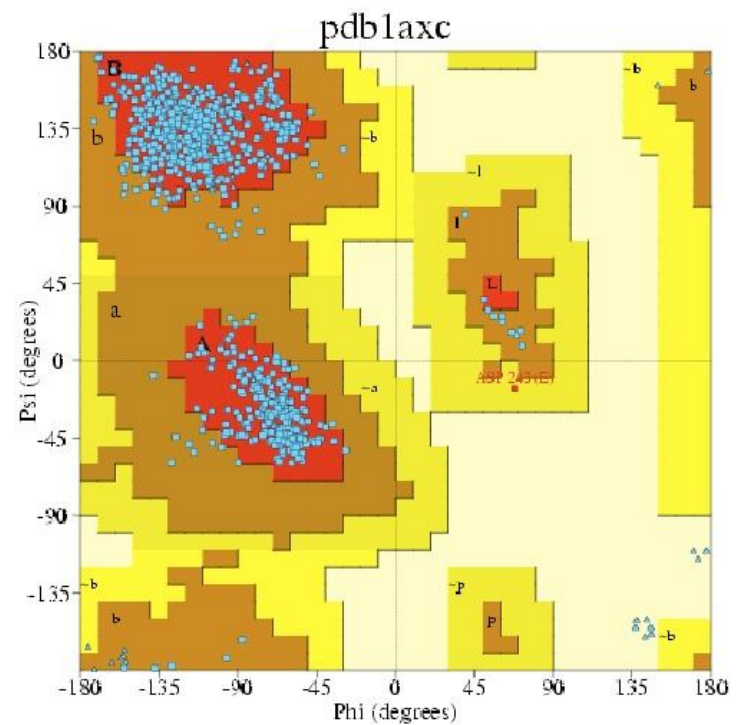
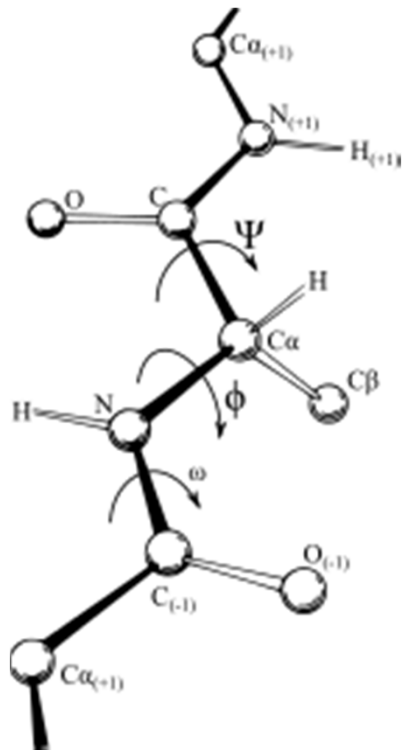
About 50 years ago, G. N. Ramachandran, an Indian Physicist, made a famous discovery on proteins

Ramachandran and his colleagues said that, due to various constraints of arrangements of atoms in 3D, neighboring amino acids (protein monomer) in a protein can't fold into any shape — there are some constraints that their arrangements have to satisfy



# Ramachandran Plot

The set of “allowed” angles can be plotted: This plot is called the “Ramachandran Plot”



# Summary

- Proteins and their structures
- Proteins are free energy minimizers
- Microstate and Macrostate
- Relations  $G = H - TS$  and  $G = -k_B T \ln Z$
- $S = k_B \ln W$
- HP model and a Toy model of protein folding
- Some aspects of real protein folding