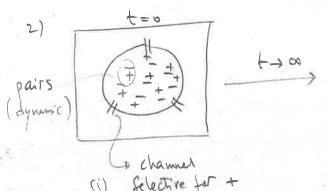
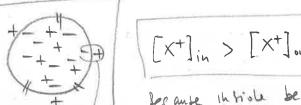


[X]in = [X]out for t-100 because of dillusion

Note: - maroscyic: no change - microscopic: constant change but on average -o steady: 8tate



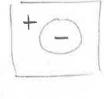
- (i) Elective for +
- (ii) to net charge ( 2+2=0)
- 3) Analogous for Selective chamels: [X-Jin > [X-Jout, Vin > Vout



fecante in trole becomes "affrective" (Vin <0)

on Intace

net dage =D Vin < Vont



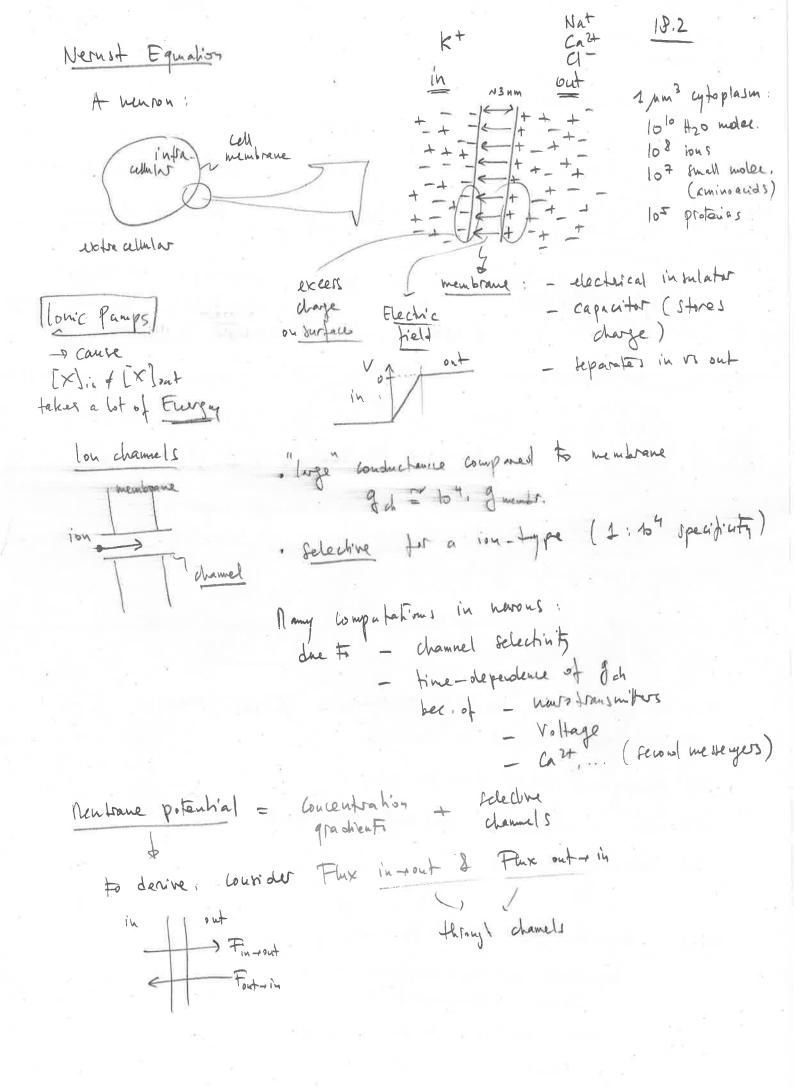
This example has all the ingredients of. the membrane potential in newous

- 1) A physical barrier (in vs out)
- 2) [X] in & [X] out somentration observes in US out
- 3) Selective channels

(non): expression for Vin (Vont =0)

- defendency on concentrations

- defendency on concentrations
- channel properties (# dran., conductance, ...)
- ion proposies: Xt, Xt, ...



Example: - Contribution of Kt memmyful. most charmels open @ rest mostly permeable to K+ - Counder Pleady-State = Equilibrium potential (other ions, e.g. O, not shown) Fout-ric = 8. [K+] out Vont=0 (Louvention)

Vint=0 (Louvention)

Vint=Veg

Vint=Veg Fraction of ions with themal energy > 9V ( Veg <0 = D e 40T <0 ) @ Equilibrium Fout - in = Fin - out [kt]out = e kgT D Veg = kot in [kt]out Solutions for kt in general: Veg - kgt. In [X]out Nemst Equation Holds for chamels beleative to I in type ofermer use GHR-Equation (below) Charge of

From Nemit Eq. - IT => IVIT: more thermal energy => more ions can vois the potential barrier in vs. out - 1217 -> IVI L: Lee, larger Energy bonnier (DE = 29V) - charge in charge bign (2) =0 change in figs (V) - dep. on concentration

Approximate concentrations in hemous

For example 
$$E^{+}$$
 le  $\frac{20}{400}$  = le  $\frac{1}{20}$  = -1. Le  $20 = -3$   $\frac{1}{20} = -3.25 = -35 \text{ mV}$ 
 $4 = 1.38 \cdot 10^{-23} \text{ J/k}$  Bolzmann constant

 $4 = 1.60 \cdot 10^{-19} \text{ G}$ 
 $4 = 1.60 \cdot 10^{-19} \text{ G}$ 
 $4 = 24 - 27 \text{ mV}$ 
 $4 = 200 \text{ J/k}$ 
 $4 = 20$ 

Note: Veg does lot defend on of e.g. # of chamels, Sch ... Bat: - time - to - equilibrain obes teperd on it hand rs [ ] - Voo for external wment obser defend on JL!

( bee next becture)

busious Capacitance of the membrane

$$G_m = c_m \cdot A$$
  $C_m \cong 10 \, n = 10 \, m = 10 \, m$ 

· How nony ions hecoled to get Vm = - 70 mV?

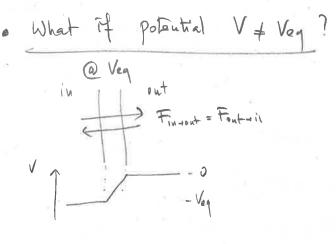
$$Q_{m} = \frac{Q_{m}}{V_{m}} \qquad Q_{m} = \frac{Q_{m} \cdot V_{m}}{1} = \frac{10^{-1} \cdot 70 \cdot 10^{-3}}{1 \cdot 10^{-11}} = \frac{10^{-1} \cdot 70 \cdot 10^{-11}}{1 \cdot 10^{-11}} = \frac{10^{-1} \cdot$$

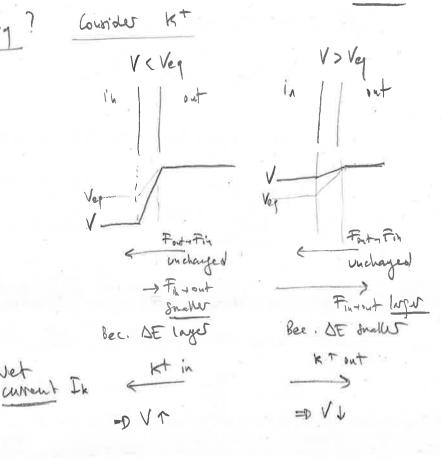
· How many ions in a herrow? Volume of nevon = 10 pm (= radinis = 10-100 pm) = 0 contours to 14 ions

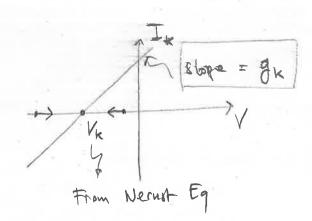
=D ony 1:105 ions wentriht to membane potential @ vert (if any 1 type was involved ... the below) = Dassimption teems ok

But Gin & Gin & T2 } things get work for Smeller Volume & rd ?

ratio of t







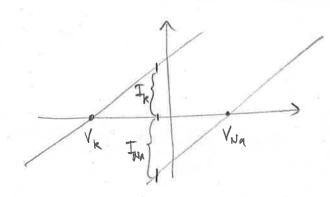
Bottom Line: (i) It mes to pull

V towards Vk

(ii) Pull is stronge it

The is larger

Cutious 2 channel Expres



New Veg given by Ik = - INA

Veg depends on conductances of JNa & JA The CC FNA \$ x >> 8 NA Veg = VNA Veg = Vk

A lot of computations in humans based on this process.

- 1) open a chamel type with Viereual
- 2) V is palled towards Vowersal

Example: huhibitory of Excitatory syrapses: activate inh. Via time

. What if dramets are permeable to more ion types? GHK-Equation: V = 185. ln Pk. [K]out + PNa. [Na]out + Pa. [CI]in
Pk. [K]in + PNa [Na]in + Pal. [a]out

P = perneability

Quertins	Coretion!	S

· Vinh vs Vrest Vrent vs Vk+ Vinh vs. Va-

. The role of purps

Veg + Vohn 4 Veg + Vohn Veg Vohn Veg Vohn Veg Vohn Veg Tong Vohn Veg + Vohn A Veg + Vohn Veg Vohn Veg

literature Dayan & Abbott, Theoretical vernomène (ouline)

Chapters 546