Practical Bioinformatics (Bio334) Metabolic network analysis

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Two kinds of bioinformatic analyses

Analyses of genotypes

Genome sequences
Single nucleotide polymorphisms
Gene families

Analyses of (molecular) phenotypes

Protein structure
Microarray gene expression data
Protein interaction data

Most challenging:

Predicting phenotype from genotype

Important to understand

gene functions
how mutations alter phenotypes
genetic risk factors for disease
how new phenotypes (adaptations, innovations) originate
how natural selection changes organisms

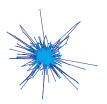
Still impossible for

whole organisms even proteins

Possible for some kinds of biological networks ...

Genome-scale networks comprise hundreds to thousands of genes and gene products

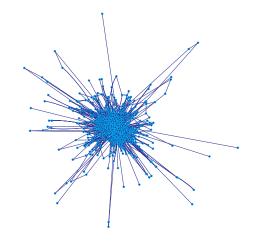
Protein interaction networks
Transcriptional regulation networks
Metabolic networks



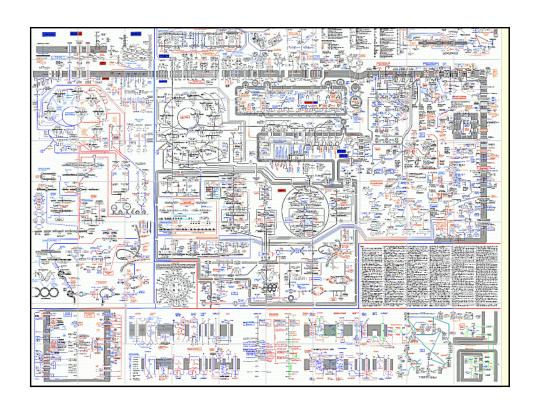
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Mathematical characterization based on qualitative understanding of network topology

Metabolic networks







A metabolic network is a set of chemical reactions whose two main functions are to produce

chemical energy

(for maintenance of cell functions and for biosyntheses)

molecular building blocks for biosyntheses

These reactions are catalyzed by enzymes that are encoded by genes.

Even in bacteria like *E.coli*, several hundred such enzymatic reactions are necessary to fulfill these functions.

You already know small parts of metabolic networks: glycolysis

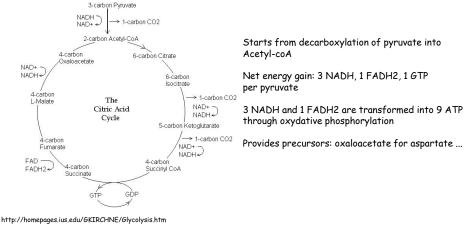
no need to memorize Converts glucose into pyruvate this pathway or other Net gain: 2 ATP, 2 NADH per glucose biochemical pathways Dihydroxyacetone phosphate Provides precursors (just for refresher) phosphate Gyceraldehyde-3-phosphate Gyceraldehyde-3-phosphate Gyceraldehyde-3-phosphoglycerate CADP 3-Phosphoglycerate CADP 3-Phosphoglycerate CADP 3-Phosphoglycerate CADP 3-phosphoglycerate for serine ... Aerobically: leads into citric acid cycle 2-Phosphoglycerate ↓ 2-Phosphoglycerate Anaerobically: creates molecules like ethanol Phosphoenolpyruvate

ADP

Pyruvate Phosphoenolpyruvate

ADP
Pyruvate or lactate to accept electrons from and regenerate NAD from NADH http://homepages.ius.edu/GKIRCHNE/Glycolysis.htm Glucose + 2ADP + 2NAD + 2P; + 2H+-> 2Pyruvate + 2ATP+ 2NADH

You already know small parts of metabolic networks: citric acid cycle



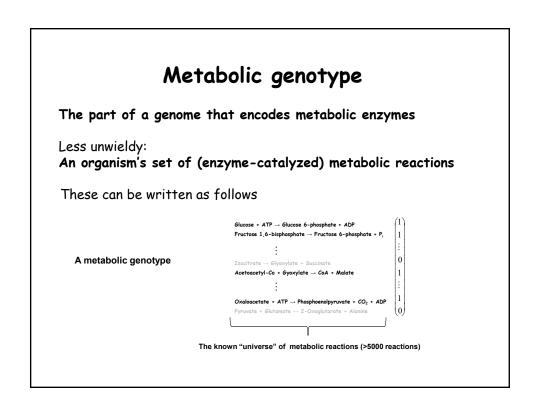
You already know small parts of metabolic networks: central carbon metabolism

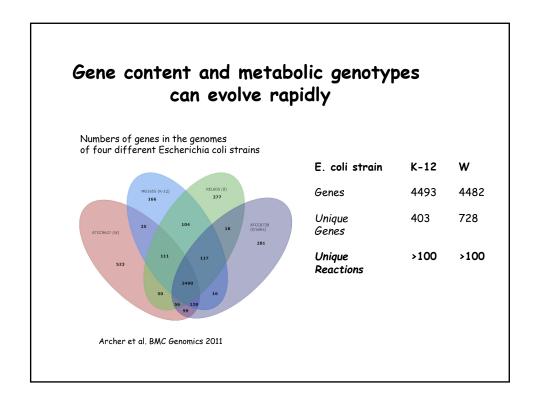
- Glycolysis
- Citric acid cycle
- Pentose phosphate shunt
 - Produces NADPH and precursors for biosyntheses (e.g., ribose-5-phosphate to synthesize RNA and DNA)

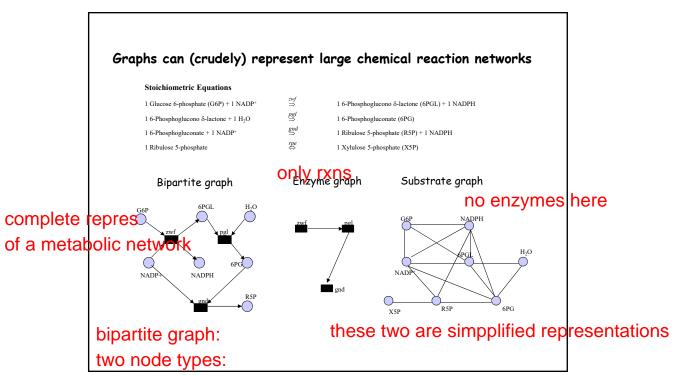
The metabolic network of a whole organism comprises many reactions

Organism	Reactions	Metabolites
E. coli S. cerevisiae A. thaliana	2077 1412 1567	1039 713 1748
B. aphidicola buchnera aphidicola	263	240

buchnera aphidicola lives within aphids so it doesnt need so many reactions since it's an endosymbiont (they only provide aphids with amino acids, and also their genome has been greatly reduced)

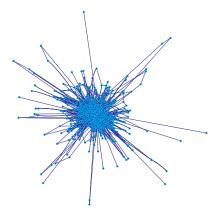






metabolites and enzymes and only they are connected with each other. no connection between metabolites directly.

An enzyme graph representation of the metabolic network of the yeast Saccharomyces cerevisiae





Graph-based analyses of metabolic networks neglect that they have a biological function and phenotype

To understand this phenotype, one needs to understand how matter <u>flows</u> through these networks

Metabolic flux: the rate at which an enzyme converts substrate into product per unit time.

Metabolic phenotype

Most general: All the molecules that a metabolism can synthesize in a given chemical environment

The most important of these molecules are biomass precursors (amino acids, DNA and RNA building blocks etc.).

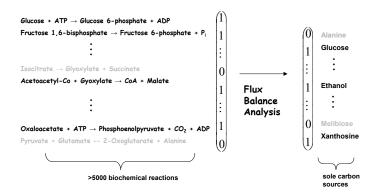
A metabolism is viable if it can synthesize all of them.

 $\mbox{\bf More specific:}$ the spectrum of nutrients on which a metabolism is viable

A carbon utilization phenotype



One can predict metabolic phenotypes from metabolic genotypes



For optimal cell growth, metabolic networks need to produce biochemical precursors in well-balanced amounts.

This requires a specific distribution of metabolic fluxes through enzymatic reactions in the network

Flux balance analysis can predict this distribution, and the maximally possible rate at which biochemical precursors, and thus biomass, can be synthesized by a metabolism

Flux balance analysis needs

- 1. a list of chemical reactions known to be catalyzed by enzymes in a given organism
- Information about nutrients in the chemical environment of a cell and their <u>uptake</u> rate (usually in mol/g dry weight [DW] and hour)

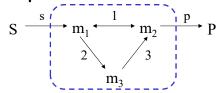
Flux balance analysis computes

1. <u>allowable</u> metabolic fluxes through a metabolic network (fluxes that do not violate the law of mass conservation)

2. within the set of allowable fluxes, those that have desirable properties (e.g., maximal rate of biomass production, maximal biomass yield per unit of carbon source.)

Metabolic flux through central carbon metabolism of *E.coli* growing at a maximally possible rate in a glucose-minimal medium thickness of arrows reflects at which reflects a

A simple chemical reaction network



Two external metabolites

- 1 substrate (nutrient) S
- 1 product P

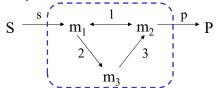
Two transport reactions s,p.

- s determines the uptake rate of S
- p determines the rate at which P is excreted

Three internal metabolites (m_i)

Three internal reactions (1,2,3)

A simple chemical reaction network



Metabolite concentrations m_i change according to the equations

$$\frac{dm_1}{dt} = v_s - v_1 - v_2$$

$$\frac{dm_2}{dt} = v_1 + v_3 - v_p$$

$$\frac{dm_3}{dt} = v_2 - v_3$$

$$\frac{d\vec{m}}{dt} = \mathbf{S}\vec{v}$$

$$\mathbf{S} = \begin{pmatrix} 1 & -1 & -1 & 0 & 0 \\ 0 & 1 & 0 & 1 & -1 \\ 0 & 0 & 1 & -1 & 0 \end{pmatrix}$$

 v_i metabolic flux through reaction i

$$\vec{v} = (v_s, v_1, v_2, v_3, vP)^\mathsf{T}$$

Stoichiometry matrix

Rows: internal metabolites Columns: reactions

An example stoichiometric matrix for bimolecular reactions

Stoichiometric Equations

1 Glucose 6-phosphate (G6P) + 1 NADP+

1 6-Phosphoglucono d-lactone (6PGL) + 1 NADPH

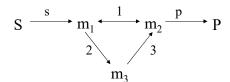
1 6-Phosphoglucono d-lactone + 1 H₂O

1 6-Phosphogluconate + 1 NADP+

1 Ribulose 5-phosphate (R5P) + 1 NADPH

1 Ribulose 5-phosphate (X5P)

Stoichiometric Matrix



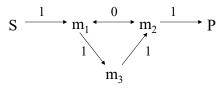
FBA assumes that metabolism is in a steady state where the concentrations of metabolites no longer change

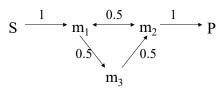
$$\frac{d\overrightarrow{m}}{dt} = 0$$

$$\mathbf{S}\vec{v}=0$$

The solutions of these equations are the allowable metabolic fluxes. They form the so-called <u>null space of S</u>

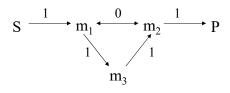
Two allowable flux distributions for our example network



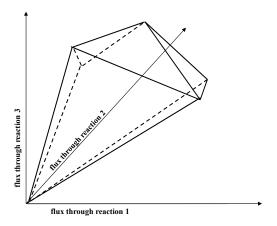


All fluxes of the form (1,x,1-x,1-x,1), $0 \le x \le 1$ are allowable

A flux distribution that maximizes the rate at which reaction 2 (producing m_3) proceeds



The null space of a complex metabolic network forms a high-dimensional "flux cone" (a convex polytope)



Several important properties of a metabolic network can be expressed as a weighted sum of fluxes

$$Z(\vec{v}) = \sum_{i=1}^{m} c_i v_i$$

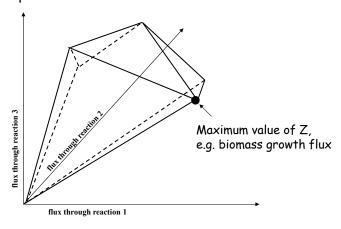
Example:

In the biomass growth reaction,

 v_i is the rate at which essential biochemical precursor i is produced by a metabolic network.

 c_i is a constant that reflects the relative contribution of precursor i to biomass (can be estimated from the biomass composition of a cell.)

FBA uses linear programming to determine regions within the flux cone where some flux or a linear function Z of multiple fluxes is maximal.



Z is also sometimes called an objective function

A biomass growth reaction for E.coli

 $\begin{array}{c} 0.513689 \text{ ala-L} + 0.295792 \text{ arg-L} + 0.241055 \text{ asn-L} + 0.241055 \text{ asp-L} + 0.091580 \text{ cys-L} + \\ 0.263160 \text{ gln-L} + 0.263160 \text{ glu-L} + 0.612638 \text{ gly} + 0.094738 \text{ his-L} + 0.290529 \text{ ile-L} + \\ 0.450531 \text{ leu-L} + 0.343161 \text{ lys-L} + 0.153686 \text{ met-L} + 0.185265 \text{ phe-L} + 0.221055 \text{ pro-L} + \\ 0.215792 \text{ ser-L} + 0.253687 \text{ thr-L} + 0.056843 \text{ trp-L} + 0.137896 \text{ tyr-L} + 0.423162 \text{ val-L} + \\ 0.026166 \text{ datp} + 0.027017 \text{ dctp} + 0.027017 \text{ dgtp} + 0.026166 \text{ dttp} + 0.133508 \text{ ctp} + \\ 0.215096 \text{ gtp} + 0.144104 \text{ utp} + 0.013894 \text{ murein5px4p[p]} + 0.019456 \text{ kdo2lipid4[e]} + \\ 0.063814 \text{ pe160} + 0.075214 \text{ pe161} + 0.177645 \text{ k} + 0.011843 \text{ nh4} + 0.007895 \text{ mg2} + \\ 0.004737 \text{ ca2} + 0.007106 \text{ fe2} + 0.007106 \text{ fe3} + 0.003158 \text{ cu2} + 0.003158 \text{ mn2} + \\ 0.003158 \text{ mobd} + 0.003158 \text{ cobalt2} + 0.003158 \text{ zn2} + 0.004737 \text{ cl} + 0.003948 \text{ so4} + \\ 0.003948 \text{ pi} + 0.000576 \text{ coa} + 0.001831 \text{ nad} + 0.000447 \text{ nadp} + 0.000223 \text{ fad} + 0.000223 \text{ thff} + \\ 0.000223 \text{ mlthf} + 0.000223 \text{ 10fthf} + 0.000223 \text{ thmpp} + 0.000223 \text{ pydx5p} + 0.000223 \text{ pbme} + \\ 0.000223 \text{ sheme} + 0.462000 \text{ h20} \end{array}$

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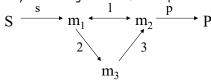
biomass

Limiting nutrient

a nutrient whose uptake rate limits the synthesis of biomass or of another molecule

Lowering its uptake rate would lower the synthesis rate

In our simple example network S is always a limiting nutrient for the production of P.



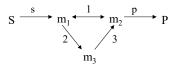
In complex networks and chemically complex environments, not all available nutrients need to be limiting.

(Relative) Yield

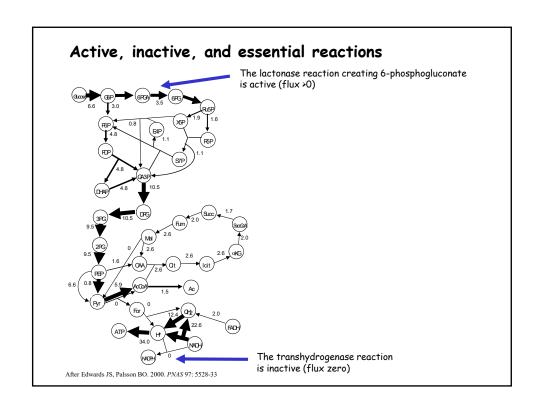
The amount of a metabolic product produced (relative to a unit of nutrient input).

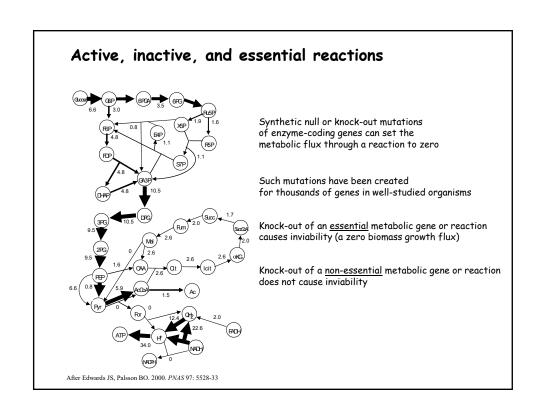
A measure of metabolic efficiency

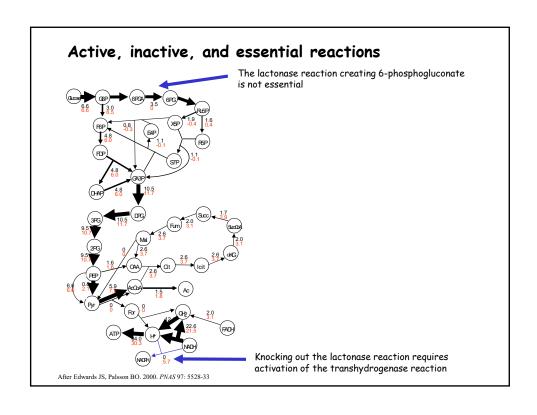
Example 1: In our simple network, if v_s =1 [mmol / g DW /hr] then the yield of product P is also v_p =1 (in steady state).

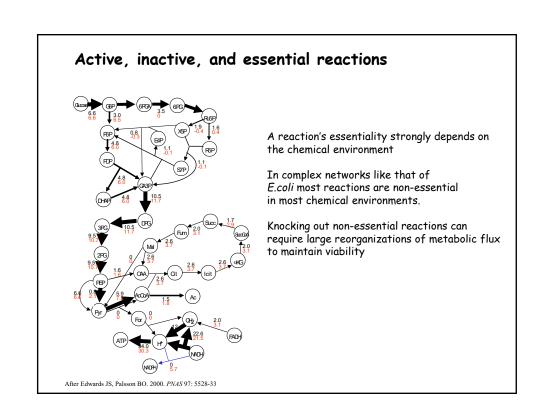


Example 2: ATP yield per mole of glucose in central metabolism Glycolysis: 2 moles ATP oxidative phosphorylation: approximately 30 moles ATP

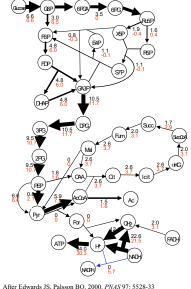








Active, inactive, and essential reactions



all essential reactions are active...

all silent reactions are inessential...

some active reactions are essential, others are inessential...

... in a given environment

Example questions for flux balance analysis

Can a given organism (metabolism) survive in environment X?

How fast could it grow in this environment?

Why are many enzymatic reactions dispensable in any one environment?

Why do some metabolisms have many reactions, while others have few?

Does network function and flux influence network evolution

Is it possible to design "resistance-proof" antimetabolic drugs?

Overview practical exercises

Exercise block 1 ("paper and pencil" analysis of toy networks):

translating reaction diagrams into stoichiometry matrices dependency of synthesis rates on uptake rates essential and active reactions maximal yields

Exercise block 2 (glycolysis and central carbon metabolism):

essential reactions computing reaction numbers limiting nutrients, alternative carbon sources, yields

Exercise block 3 (genome-scale metabolic networks):

expensive and cheap metabolites essential reactions

exam: 1 problem which is similar to one of the problems solved

Overview practical exercises

Exercise block 4 (genome-scale metabolisms):

essential reactions and flux distributions in different environments rich and minimal environments alternative carbon sources

Exercise block 5 (genome-scale metabolisms):

metabolic system analysis and antimetabolic drugs