

建模

估计罐子里的液体温度

一、估计

假设在稳定状态下液体的温度是以每秒 0.1C 的速度加热。但是真实液体温度可能会出现一些波动，用下面的公式来描述系统**状态方程**：

$$\hat{T}_t^- = \hat{T}_{t-1} + v * \Delta t + w$$

- \$w_n\$ 是随机过程噪声，假设我们的模型是很准确的，方差为Q设为0.15
- 测量误差（标准差）为 0.1摄氏度,R=0.1
- 采样时间：每 5 秒进行一次测量
- ^表示估计值
- 表示先验值

预估噪声协方差方程有

$$P_t^- = P_{t-1} + Q$$

二、更新

观测方程泰勒展开

$$z_t = T_t + v$$

- T_t 就是传感器的温度，v是更新误差

卡尔曼增益更新

$$K = P_t^- * (P_t^- + R)^{-1}$$

状态更新（修正）方程

$$\hat{T}_t = \hat{T}_t^- + K * (z_t - \hat{T}_t^-)$$

更新后的后验估计协方差

$$P_t = (I - K)P_t^-$$

在测量点的真实温度是：50.479°C, 51.025°C, 51.5°C, 52.003°C, 52.494°C, 53.002°C, 53.499°C, 54.006°C, 54.498°C, and 54.991°C.

误差0.1内

温度计测量数据：z=

{50.45°C, 50.967°C, 51.6°C, 52.106°C, 52.492°C, 52.819°C, 53.433°C, 54.007°C, 54.523°C, and 54.99°C}

三、迭代

第零次迭代

在第一次迭代之前，需要初始化卡尔曼滤波器，然后预测下一个状态（也就是第一个状态）。

1.初始化

用手摸，盲猜初始化温度是

$$\hat{T}_{(0, 0)} = 10$$

由于没有温度计，这个猜测值是非常不精准的，于是把初始估计标准差 (σ) 设为 ± 100 \$。初始化的估计不确定性(Estimate Uncertainty)是预测误差的方差（自己设置的）

$$P_{0, 0} = \sigma^2 = 10000$$

这里方差非常高。如果我们的初始值更合理，卡尔曼滤波器会收敛更快。

2.估计

$$\hat{T}_{(1, 0)} = 10$$

预估噪声协方差

$$P_{1, 0} = P_{0, 0} + Q = \sigma^2 + Q = 10000.15$$

第一次迭代

1.更新

卡尔曼增益更新

$$K = P_{1, 0} * (P_{1, 0} + R)^{-1} = 0.999999$$

状态更新（修正）

$$\hat{T}_{(1, 1)} = \hat{T}_{(1, 0)} + K * (z_0 - \hat{T}_{(1, 0)}) = 50.45$$

更新后的后验估计协方差

$$P_{1, 1} = (I - K)P_t * P_{1, 0} = 0.01$$

2.估计

$$\hat{T}_{(2, 1)} = \hat{T}_{(1, 1)} = 50.45$$

预估噪声协方差

$$P_{2, 1} = P_{1, 1} + Q = \sigma^2 + Q = 0.16$$

第二次迭代

1.更新

卡尔曼增益更新

$$K = P_{2, 1} * (P_{2, 1} + R)^{-1} = 0.9412$$

状态更新（修正）

$$\hat{T}_{(2, 2)} = \hat{T}_{(2, 1)} + K * (z_1 - \hat{T}_{(2, 1)}) = 50.94$$

...

n	y_n	Current state estimates ($K_n, \hat{x}_{n,n}, p_{n,n}$)	Prediction ($\hat{x}_{n+1,n}, p_{n+1,n}$)
1	50.45°C	$K_1 = \frac{10000.15}{10000.15 + 0.01} = 0.999999$ $\hat{x}_{1,1} = 10 + 0.999999 (50.45 - 10) = 50.45^\circ C$ $p_{1,1} = (1 - 0.999999) 0.10000.15 = 0.01$	$\hat{x}_{2,1} = \hat{x}_{3,3} = 50.45^\circ C$ $p_{2,1} = 0.01 + 0.15 = 0.16$
2	50.967°C	$K_2 = \frac{0.16}{0.16 + 0.01} = 0.9412$ $\hat{x}_{2,2} = 50.45 + 0.9412 (50.967 - 50.45) = 50.94^\circ C$ $p_{2,2} = (1 - 0.9412) 0.16 = 0.0094$	$\hat{x}_{3,2} = \hat{x}_{3,3} = 50.94^\circ C$ $p_{3,2} = 0.0094 + 0.15 = 0.1594$
3	51.6°C	$K_3 = \frac{0.1594}{0.1594 + 0.01} = 0.941$ $\hat{x}_{3,3} = 50.94 + 0.941 (51.6 - 50.94) = 51.56^\circ C$ $p_{3,3} = (1 - 0.941) 0.1594 = 0.0094$	$\hat{x}_{4,3} = \hat{x}_{3,3} = 51.56^\circ C$ $p_{4,3} = 0.0094 + 0.15 = 0.1594$
4	52.106°C	$K_4 = \frac{0.1594}{0.1594 + 0.01} = 0.941$ $\hat{x}_{4,4} = 51.56 + 0.941 (52.106 - 51.56) = 52.07^\circ C$ $p_{4,4} = (1 - 0.941) 0.1594 = 0.0094$	$\hat{x}_{5,4} = \hat{x}_{4,4} = 52.07^\circ C$ $p_{5,4} = 0.0094 + 0.15 = 0.1594$
5	52.492°C	$K_5 = \frac{0.1594}{0.1594 + 0.01} = 0.941$ $\hat{x}_{5,5} = 52.07 + 0.941 (52.492 - 52.07) = 52.47^\circ C$ $p_{5,5} = (1 - 0.941) 0.1594 = 0.0094$	$\hat{x}_{6,5} = \hat{x}_{5,5} = 52.47^\circ C$ $p_{6,5} = 0.0094 + 0.15 = 0.1594$
6	52.819°C	$K_6 = \frac{0.1594}{0.1594 + 0.01} = 0.941$ $\hat{x}_{6,6} = 52.47 + 0.941 (52.819 - 52.47) = 52.8^\circ C$ $p_{6,6} = (1 - 0.941) 0.1594 = 0.0094$	$\hat{x}_{7,6} = \hat{x}_{6,6} = 52.8^\circ C$ $p_{7,6} = 0.0094 + 0.15 = 0.1594$
7	53.433°C	$K_7 = \frac{0.1594}{0.1594 + 0.01} = 0.941$ $\hat{x}_{7,7} = 52.8 + 0.941 (53.433 - 52.8) = 53.4^\circ C$ $p_{7,7} = (1 - 0.941) 0.1594 = 0.0094$	$\hat{x}_{8,7} = \hat{x}_{7,7} = 53.4^\circ C$ $p_{8,7} = 0.0094 + 0.15 = 0.1594$
8	54.007°C	$K_8 = \frac{0.1594}{0.1594 + 0.01} = 0.941$ $\hat{x}_{8,8} = 53.4 + 0.941 (54.007 - 53.4) = 53.97^\circ C$ $p_{8,8} = (1 - 0.941) 0.1594 = 0.0094$	$\hat{x}_{9,8} = \hat{x}_{8,8} = 53.97^\circ C$ $p_{9,8} = 0.0094 + 0.15 = 0.1594$

9	54.523°C	$K_9 = \frac{0.1594}{0.1594 + 0.01} = 0.941$ $\hat{x}_{9,9} = 53.97 + 0.941 (54.523 - 53.97) = 54.49^\circ C$ $p_{9,9} = (1 - 0.941) 0.1594 = 0.0094$	$\hat{x}_{10,9} = \hat{x}_{9,9} = 54.49^\circ C$ $p_{10,9} = 0.0094 + 0.15 = 0.1594$
10	54.99°C	$K_{10} = \frac{0.1594}{0.1594 + 0.01} = 0.941$ $\hat{x}_{10,10} = 54.49 + 0.941 (54.99 - 54.49) = 54.96^\circ C$ $p_{10,10} = (1 - 0.941) 0.1594 = 0.0094$	$\hat{x}_{11,10} = \hat{x}_{10,10} = 54.96^\circ C$ $p_{11,10} = 0.0094 + 0.15 = 0.1594$

下图比较了真实值、测量值及估计值：

