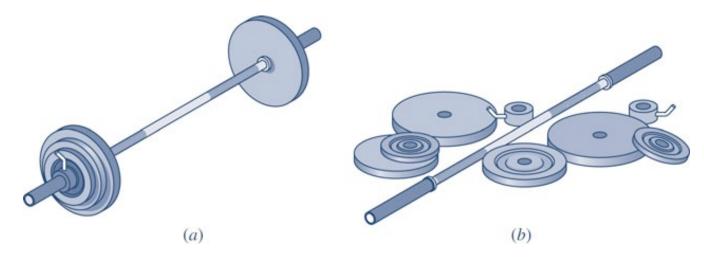
Extensive Property

- ▶ Depends on the size or extent of a system.
- ►Examples: mass, volume, energy., 崎, 沙養
- ► Its value for an overall system is the sum of its values for the parts into which the system is divided.



▶ Its value may vary with time but not position.

Intensive Property

- ► Independent of the size or extent of a system.
- Examples: pressure, temperature.
- ► Its value is not additive as for extensive properties.
- ► May vary from place to place within the system at any moment function of both position and time.

Equilibrium

- ► When a system is isolated, it does not interact with its surroundings; however, its state can change as a consequence of spontaneous events occurring internally as its intensive properties such as temperature and pressure tend toward uniform values. When all such changes cease, the system is at an equilibrium state.
- Equilibrium states and processes from one equilibrium state to another equilibrium state play important roles in thermodynamic analysis.

<u>Units (1 of 2)</u>

- A unit is any specified amount of a quantity by comparison with which any other quantity of the same kind is measured (e.g., meter, kilometers, feet, and miles are all *units of length*).
- Two systems of units:
 - SI (Système International d'Unités)
 - English Engineering units.

Units (2 of 2)

TABLE 1.3 Units for Mass, Length, Time, and Force

	SI		English	
Quantity	Unit	Symbol	Unit	Symbol
mass	kilogram	kg	pound mass	lb
length	meter	m	foot	ft
time	second	s	second	S
force	newton	N	pound force	lbf
	$(= 1 \text{ kg} \cdot \text{m/s}^2)$		$(=32.1740 \text{ lb} \cdot \text{ft/s}^2)$	

In these unit systems, mass, length, and time are base units and force has a unit derived from them using,

$$F = ma \qquad (Eq. 1.1)$$

SI:
$$1 \text{ N} = (1 \text{ kg})(1 \text{ m/s}^2) = 1 \text{ kg·m/s}^2$$
 (Eq. 1.2)

English:

1 lbf =
$$(1 lb)(32.1740 ft/s^2) = 32.1740 lb \cdot ft/s^2$$
 (Eq. 1.5)