Appendix A: Physical Constants and Conversion Factors

PHYSICAL CONSTANTS

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Avogadro's number, N_A=6.023\times 10^{26} molecules/kgmole Boltzmann's constant, k=1.381\times 10^{-23} J/(molecule·K) Electron charge, e=1.602\times 10^{-19} C Electron mass, m_e=9.110\times 10^{-31} kg Faraday's constant, F=96,487 kC/kgmole electrons = 96,487 kJ/(V·kgmole electrons) Gravitational acceleration (standard), g=32.174 ft/s² = 9.807 m/s² Gravitational constant, k_G=6.67\times 10^{-11} m³/(kg·s²) Newton's second law constant, g_c=32.174 lbm·ft/(lbf·s²) = 1.0 kg·m/(N·s²) Planck's constant, \hbar=6.626\times 10^{-34} J·s/molecule Stefan-Boltzmann constant, \sigma=0.1714\times 10^{-8} Btu/(h·ft²·R⁴) = 5.670\times 10^{-8} W/(m²·k⁴) Universal gas constant \Re=1545.35 ft·lbf/(lbmole·R) = 8314.3 J/(kgmole·K) = 8.3143 kJ/(kgmole·K) = 1.9858 Btu/(lbmole·R) = 1.9858 kcal/(kgmole·K) = 1.9858 cal/(gmole·K) = 0.08314 bar·m³/(kgmole·K) = 82.05 L·atm/(kgmole·K) Velocity of light in a vacuum, c=9.836\times 10^8 ft/s = 2.998\times 10^8 m/s
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UNIT DEFINITIONS

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1 coulomb (C) = 1 A·s
                                                                           1 ohm (\Omega) = 1 \text{ V/A}
1 dyne = 1 \text{ g} \cdot \text{cm/s}^2
                                                                           1 pascal (Pa) = 1 \text{ N/m}^2
1 erg = 1 dyne·cm
                                                                           1 poundal = 1 lbm \cdot ft/s^2
1 farad (F) = 1 \text{ C/V}
                                                                           1 siemens (S) = 1 A/V
1 henry (H) = 1 \text{ Wb/A}
                                                                           1 \text{ slug} = 1 \text{ lbf} \cdot \text{s}^2/\text{ft}
1 hertz (Hz) = 1 cycle/s
                                                                           1 tesla (T) = 1 Wb/m^2
1 joule (J) = 1 \text{ N} \cdot \text{m}
                                                                           1 volt (V) = 1 W/A
                                                                           1 watt (W) = 1 J/s
1 lumen = 1 candela · steradian
                                                                           1 weber (Wb) = 1 V·s
1 \text{ lux} = 1 \text{ lumen/m}^2
1 newton (N) = 1 \text{ kg} \cdot \text{m/s}^2
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CONVERSION FACTORS

Length	Energy
$1 \text{ m} = 3.2808 \text{ ft} = 39.37 \text{ in} = 10^2 \text{ cm} = 10^{10} \text{ Å}$	$1 \text{ J} = 1 \text{ N} \cdot \text{m} = 1 \text{ kg} \cdot \text{m}^2/\text{s}^2 = 9.479 \times 10^{-4} \text{ Btu}$
$1 \text{ cm} = 0.0328 \text{ ft} = 0.394 \text{ in} = 10^{-2} \text{ m} = 10^{8} \text{ Å}$	1 kJ = 1000 J = 0.9479 Btu = 238.9 cal
$1 \text{mm} = 10^{-3} \text{m} = 10^{-1} \text{cm}$	1 Btu = 1055.0 J = 1.055 kJ = 778.16 ft·lbf = 252 cal
1 km = 1000 m = 0.6215 miles = 3281 ft	1 cal = 4.186 J = 3.968×10^{-3} Btu
1 in = 2.540 cm = 0.0254 m	1 Cal (in food value) = 1 kcal = 4186 J = 3.968 Btu
1 ft = 12 in = 0.3048 m	1 erg = 1 dyne·cm = 1 g·cm ² /s ² = 10^{-7} J
1 mile = 5280 ft = 1609.36 m = 1.609 km	$1 \text{ eV} = 1.602 \times 10^{-19} \text{J}$

(Continued)

CONVERSION FACTORS

Area

$$1 \text{ m}^2 = 10^4 \text{ cm}^2 = 10.76 \text{ ft}^2 = 1550 \text{ in}^2$$

$$1 \text{ ft}^2 = 144 \text{ in}^2 = 0.0929 \text{ m}^2 = 929.05 \text{ cm}^2$$

$$1 \text{ cm}^2 = 10^{-4} \text{ m}^2 = 1.0764 \times 10^{-3} \text{ ft}^2 = 0.155 \text{ in}^2$$

$$1 \text{ in}^2 = 6.944 \times 10^{-3} \text{ ft}^2 = 6.4516 \times 10^{-4} \text{ m}^2 = 6.4516 \text{ cm}^2$$

Volume

$$1 \text{ m}^3 = 35.313 \text{ ft}^3 = 6.1023 \times 10^4 \text{ in}^3 = 1000 \text{ L} = 264.171 \text{ gal}$$

$$1 L = 10^{-3} \text{m}^3 = 0.0353 \, \text{ft}^3 = 61.03 \, \text{in}^3 = 0.2642 \, \text{gal}$$

1 gal =
$$231 \text{ in}^3 = 0.13368 \text{ ft}^3 = 3.785 \times 10^{-3} \text{ m}^3$$

$$1 \text{ ft}^3 = 1728 \text{ in}^3 = 28.3168 \, L = 0.02832 \, \text{m}^3 = 7.4805 \, \text{gal}$$

$$1 \text{ in}^3 = 16.387 \text{ cm}^3 = 1.6387 \times 10^{-5} \text{ m}^3 = 4.329 \times 10^{-3} \text{ gal}$$

Mass

$$1 \text{ kg} = 1000 \text{ g} = 2.2046 \text{ lbm} = 0.0685 \text{ slug}$$

1 lbm =
$$453.6$$
 g = 0.4536 kg = 3.108×10^{-2} slug

$$1 \text{ slug} = 32.174 \text{ lbm} = 1.459 \times 10^4 \text{ g} = 14.594 \text{ kg}$$

Force

$$1 \text{ N} = 10^5 \text{ dyne} = 1 \text{ kg} \cdot \text{m/s}^2 = 0.225 \text{ lbf}$$

$$1 \text{ lbf} = 4.448 \text{ N} = 32.174 \text{ poundals}$$

1 poundal =
$$0.138 \, \text{N} = 3.108 \times 10^{-2} \, \text{lbf}$$

Power

(Continued)

$$1 \text{ W} = 1 \text{ J/s} = 1 \text{ kg} \cdot \text{m}^2/\text{s}^3 = 3.412 \text{ Btu/h} = 1.3405 \times 10^{-3} \text{ hp}$$

$$1 \text{ kW} = 1000 \text{ W} = 3412 \text{ Btu/h} = 737.3 \text{ ft·lbf/s} = 1.3405 \text{ hp}$$

1 Btu/h =
$$0.293 \text{ W} = 0.2161 \text{ ft} \cdot \text{lbf/s} = 3.9293 \times 10^{-4} \text{ hp}$$

$$1 \text{ hp} = 550 \text{ ft \cdot lbf/s} = 33000 \text{ ft \cdot lbf/min} = 2545 \text{ Btu/h} = 746 \text{ W}$$

Pressure

$$1 \text{ Pa} = 1 \text{ N/m}^2 = 1 \text{ kg/(m} \cdot \text{s}^2) = 1.4504 \times 10^{-4} \text{ lbf/in}^2$$

$$1 \, \text{lbf/in}^2 = 6894.76 \, \text{Pa} = 0.068 \, \text{atm} = 2.036 \, \text{in Hg}$$

$$1 \text{ atm} = 14.696 \, \text{lbf/in}^2 = 1.01325 \times 10^5 \, \text{Pa}$$

$$= 101.325 \, \text{kPa} = 760 \, \text{mm} \, \text{Hg}$$

$$1 \text{ bar} = 10^5 \text{ Pa} = 0.987 \text{ atm} = 14.504 \text{ lbf/in}^2$$

$$1 \text{ dyne/cm}^2 = 0.1 \text{ Pa} = 10^{-6} \text{ bar} = 145.04 \times 10^{-7} \text{ lbf/in}^2$$

1 in Hg =
$$3376.8 \, \text{Pa} = 0.491 \, \text{lbf/in}^2$$

1 in
$$H_2O = 248.8 \, \text{Pa} = 0.0361 \, \text{lbf/in}^2$$

MISCELLANEOUS UNIT CONVERSIONS

Specific Heat Units

$1 \text{ Btu/(lbm} \cdot {}^{\circ}\text{F}) = 1 \text{ Btu/(lbm} \cdot \text{R})$

$$1 \text{ kJ/(kg} \cdot \text{K}) = 0.23884 \text{ Btu/(lbm} \cdot \text{R}) = 185.8 \text{ ft} \cdot \text{lbf/(lbm} \cdot \text{R})$$

1 Btu/(lbm·R) = 778.16 ft·lbf/(lbm·R) =
$$4.186 \text{ kJ/(kg·K)}$$

Energy Density Units

$$1 \text{ kJ/kg} = 1000 \text{ m}^2/\text{s}^2 = 0.4299 \text{ Btu/lbm}$$

1 Btu/lbm =
$$2.326 \, \text{kJ/kg} = 2326 \, \text{m}^2/\text{s}^2$$

Energy Flux

$$1 \text{ W/m}^2 = 0.317 \text{ Btu/(h·ft}^2)$$

$$1 \text{ Btu/(h·ft}^2) = 3.154 \text{ W/m}^2$$

Heat Transfer Coefficient

$$1 \text{ W/(m}^2 \cdot \text{K}) = 0.1761 \text{ Btu/(h} \cdot \text{ft}^2 \cdot \text{R})$$

$$1 Btu/(h \cdot ft^2 \cdot R) = 5.679 W/(m^2 \cdot K)$$

Thermal Conductivity

$$1\,W/(m\!\cdot\!K) = 0.5778\,Btu/(h\!\cdot\!ft\!\cdot\!R)$$

$$1 \; Btu/(h \cdot ft \cdot R) \; = \; 1.731 \; W/(m \cdot K)$$

Temperature

$$T(^{\circ}F) = \frac{9}{5}T(^{\circ}C) + 32 = T(R) - 459.67$$

$$T(^{\circ}C) = \frac{5}{9}[T(^{\circ}F) - 32] = T(K) - 273.15$$

$$T(R) = \frac{9}{5} T(K) = (1.8)T(K) = T(^{\circ}F) + 459.67$$

$$T(K) = \frac{5}{9}T(R) = T(R)/1.8 = T(^{\circ}C) + 273.15$$

Density

$$1 \text{ lbm/ft}^3 = 16.0187 \text{ kg/m}^3$$

$$1 \text{ kg/m}^3 = 0.062427 \text{ lbm/ft}^3 = 10^{-3} \text{ g/cm}^3$$

$$1 \text{ g/cm}^3 = 1 \text{ kg/L} = 62.4 \text{ lbm/ft}^3 = 10^3 \text{ kg/m}^3$$

Viscosity

$$1 \text{ Pa} \cdot \text{s} = 1 \text{ N} \cdot \text{s/m}^2 = 1 \text{ kg/(m} \cdot \text{s}) = 10 \text{ poise}$$

1 poise = 1 dyne·s/cm² = 1 g/(cm·s) =
$$0.1 \text{ Pa·s}$$

1 poise =
$$2.09 \times 10^{-3} lbf \cdot s/ft^2 = 6.72 \times 10^{-2} lbm/(ft \cdot s)$$

1 centipoise =
$$0.01$$
 poise = 10^{-3} Pa·s

$$1 \text{ lbf} \cdot \text{s/ft}^2 = 1 \text{ slug/(ft} \cdot \text{s}) = 47.9 \text{ Pa} \cdot \text{s} = 479 \text{ poise}$$

1 stoke =
$$1 \text{ cm}^2/\text{s} = 10^{-4} \text{ m}^2/\text{s} = 1.076 \times 10^{-3} \text{ ft}^2/\text{s}$$

1 centistoke =
$$0.01$$
 stoke = 10^{-6} m²/s = 1.076×10^{-5} ft²/s

$$1 \text{ m}^2/\text{s} = 10^4 \text{ stoke} = 10^6 \text{ centistoke} = 10.76 \text{ ft}^2/\text{s}$$