

Yonsei University Graduate Class

Energy Materials: Design, Discovery and Data

Solar Energy Conversion

Prof. Aron Walsh

Department of Materials
Imperial College London

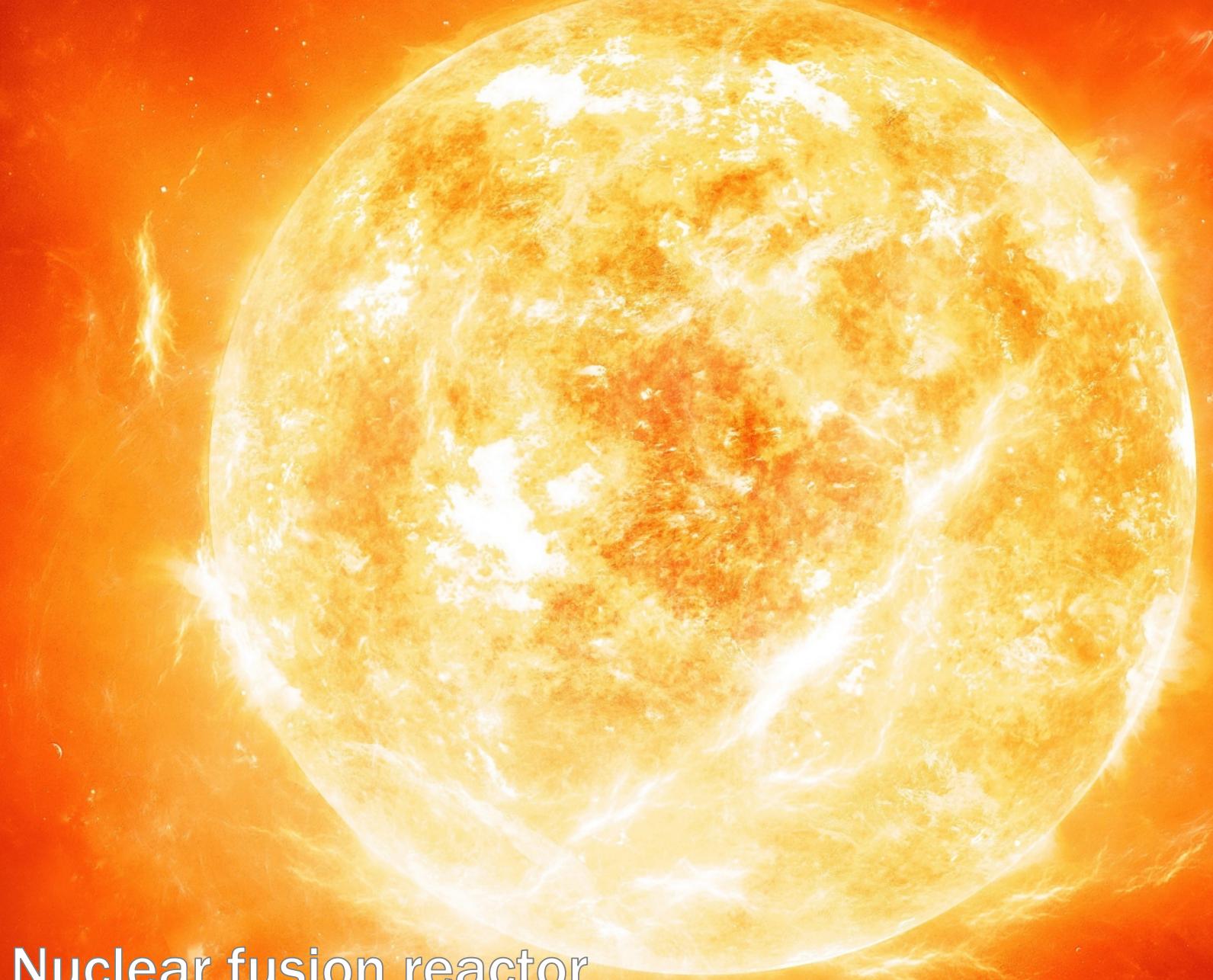


<https://wmd-group.github.io>



@lonepair

385 YW (Yotta = 10^{24}): Luminosity of the Sun



Nuclear fusion reactor

385 YW (Yotta = 10^{24}): Luminosity of the Sun

174 PW (Peta = 10^{15}): Power reaching Earth



Blackbody emission at ~ 5800 K

385 YW (Yotta = 10^{24}): Luminosity of the Sun

174 PW (Peta = 10^{15}): Power reaching Earth

15 TW (Tera = 10^{12}): Power usage on Earth



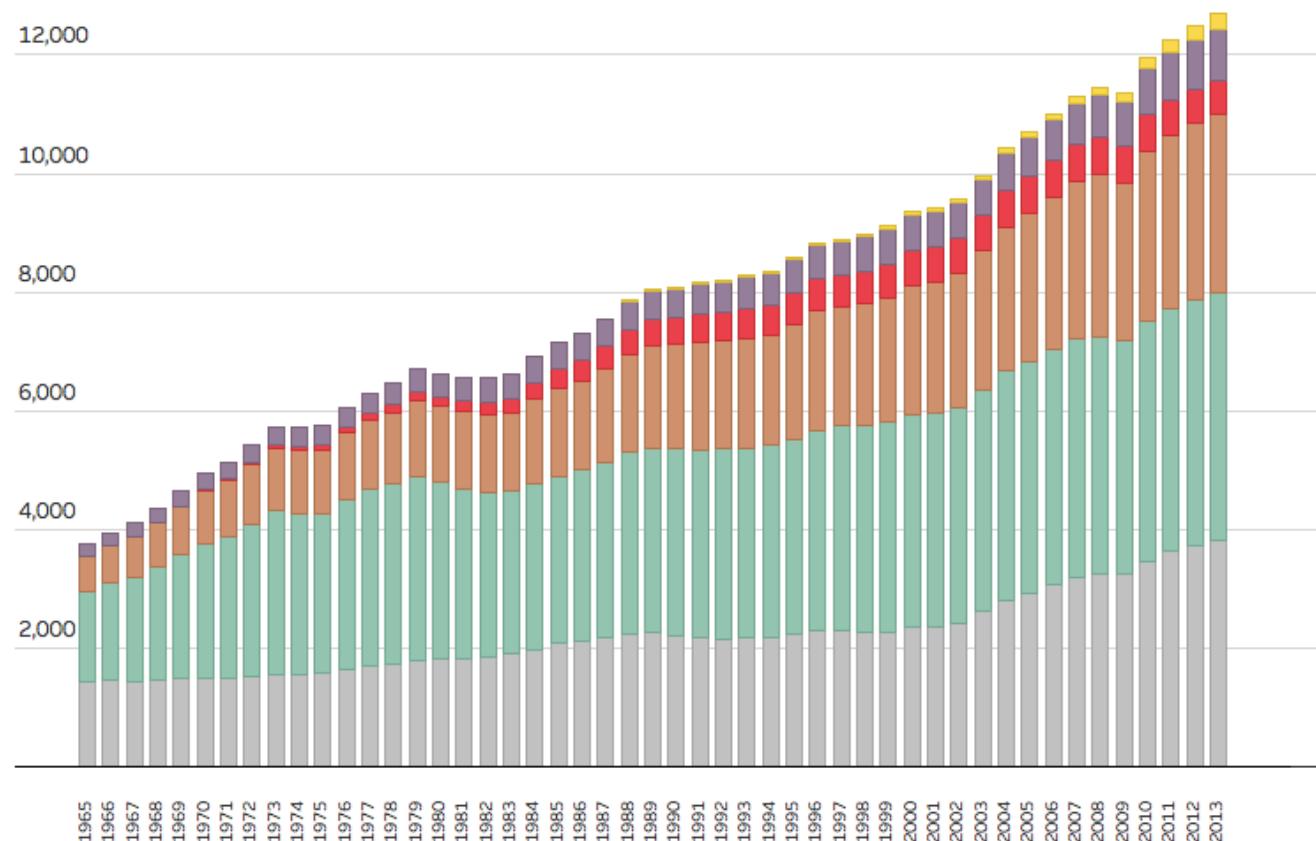
20% electricity consumption on lighting

Global Trend – Total Energy

Global energy use by source

In millions of tons of oil equivalent

Coal Oil Natural gas Nuclear Hydroelectricity Other renewables



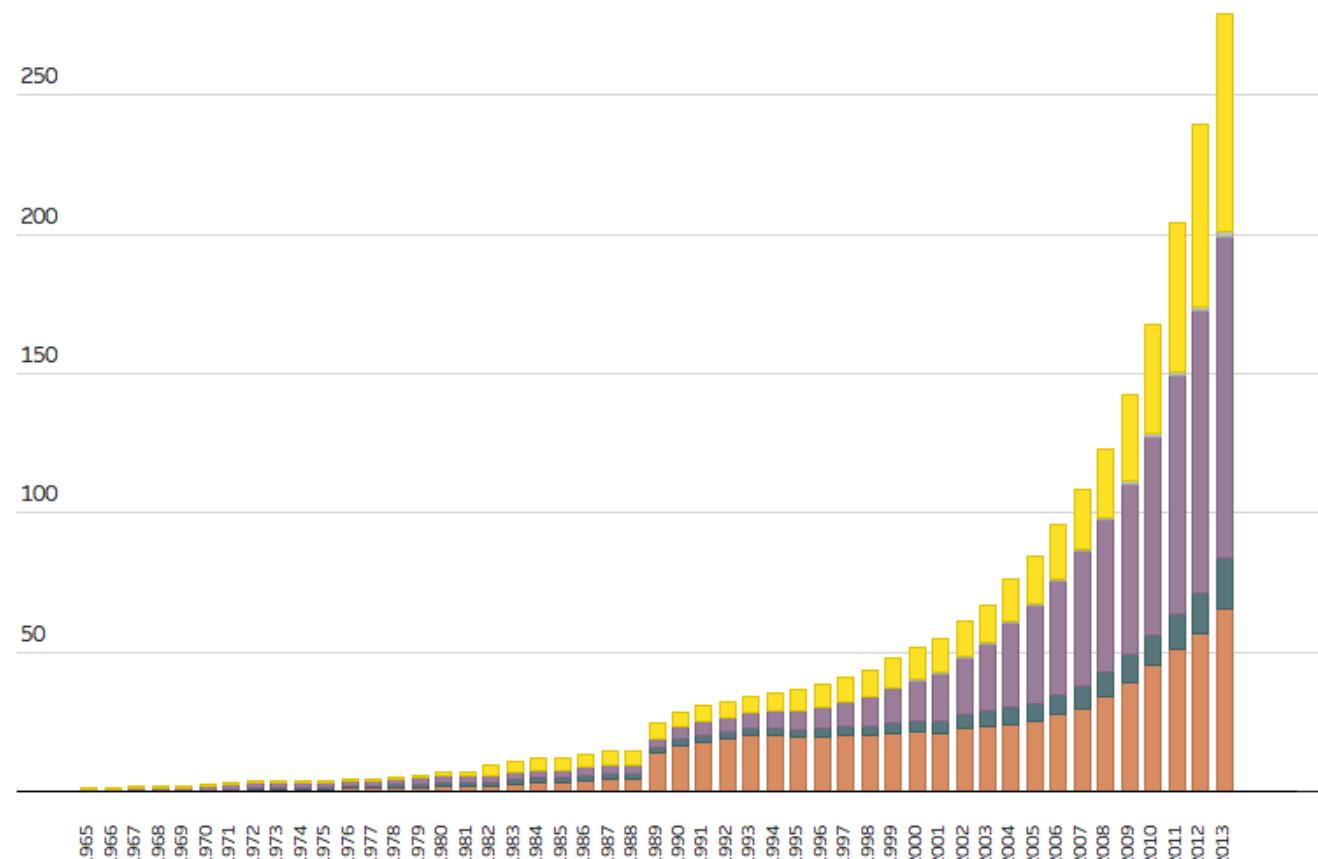
Source: BP Statistical Review of Energy 2014

Global Trend – Renewable Energy

Global renewable energy consumption

Excluding hydropower. In million tons of oil equivalent.

North America South America Europe Middle East Africa Asia Pacific



Source: BP Statistical Review of World Energy 2014.

Solar Energy Future



Solar Energy Future

Battersea Coal Plant (1953, UK)

503 MW

Latest Developments



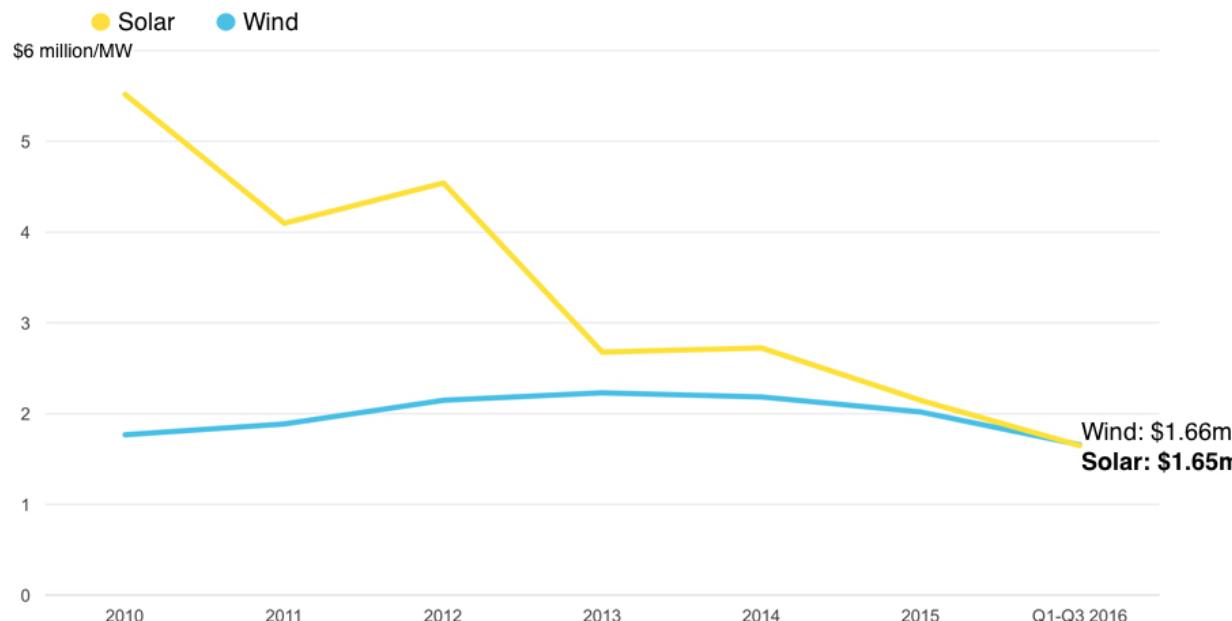
<http://earthobservatory.nasa.gov/IOTD/view.php?id=89668>

Latest Developments

“Renewables are robustly entering the era of undercutting fossil fuel prices”

Solar Surprise: Prices Fall Below Wind

A turning point for renewables in lower-income countries



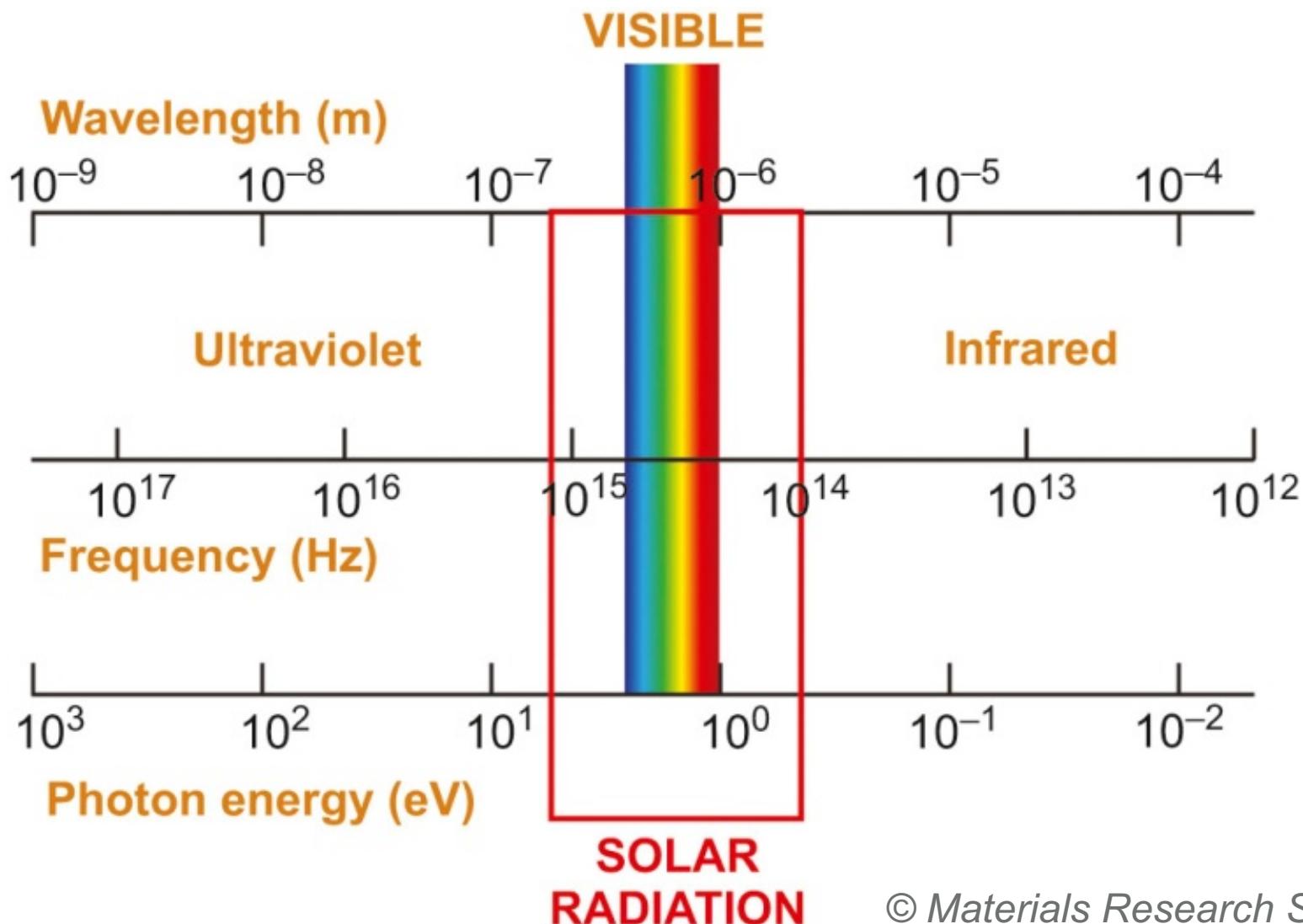
Class Question

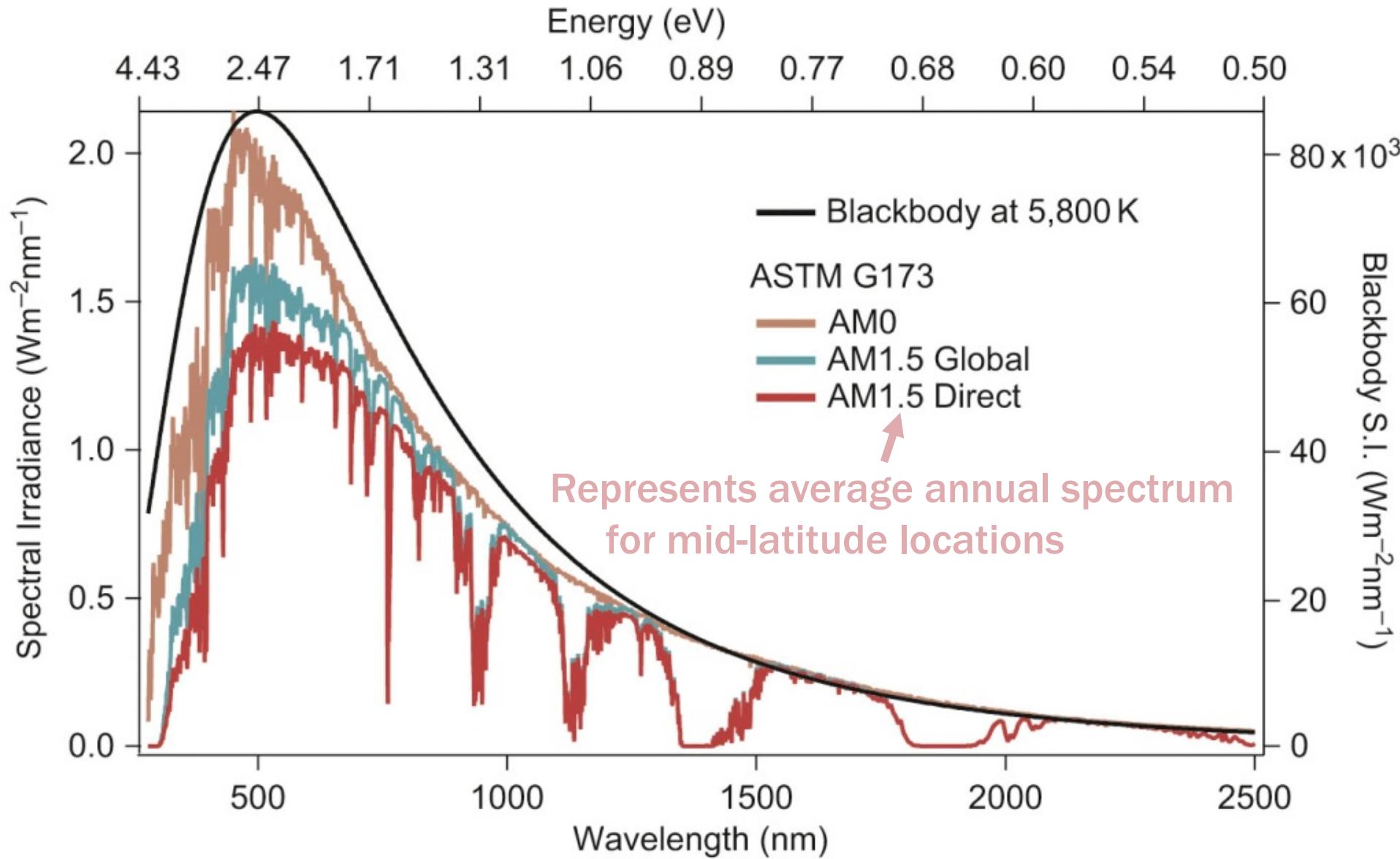
How is South Korea's
electricity generated?

Talk Outline: Solar Energy

- 1. Solar Energy Fundamentals**
- 2. Current Photovoltaic Technologies**
- 3. Case Study: Hybrid Perovskites**

Solar Spectrum





Data: <http://rredc.nrel.gov/solar/spectra/am1.5>

Class Question

How to convert sunlight
to electricity?

Solar Energy Timeline

Archimedes Death Ray (212 BC)



Solar Energy Timeline

LA LUMIÈRE

392466

SES CAUSES ET SES EFFETS

PAR


M. EDMOND BECQUEREL

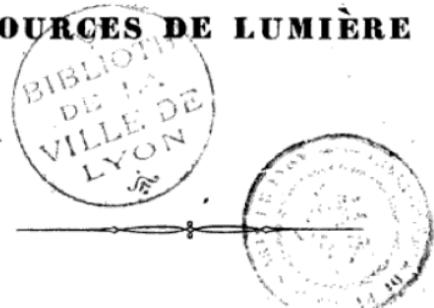
DE L'ACADEMIE DES SCIENCES

DE L'INSTITUT DE FRANCE

PROFESSEUR AU CONSERVATOIRE IMPÉRIAL DES ARTS ET MÉTIERS, ETC., ETC.

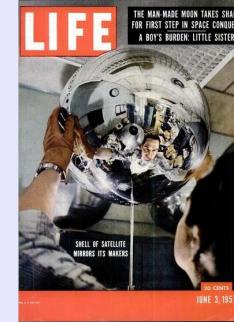
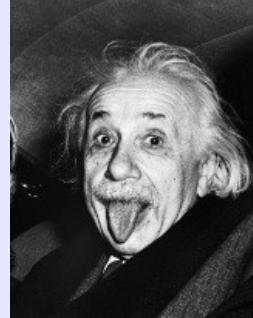
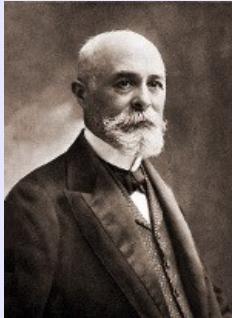
TOME PREMIER

SOURCES DE LUMIÈRE



Expériences de décembre 1862.	MATIÈRES placées dans la flamme du chalumeau à gaz oxygène et hydrogène.	PHOTOMÉTRÉ.			
		Angle observé.	Logarithme de I.	Intensité lumineuse I.	TEMPÉRATURE calculée.
	Chaux vive.....	1°18'	3,2884330	"	1566,9
	Magnésie	1.19	3,2766000	"	1564,6
	Platine.....	2.22	2,7682170	"	1463,2
	Palladium.....	4.12	2,2705248	"	1364,0
Fer.....	Effets lumineux va- riables; éclats par instants.	5.44	2,0008810	"	"
	Cuivre... Effets lumineux très- variables d'intensité.	25.32	0,7309726	"	"

Solar Energy Timeline



1839:
Edmund Bequerel
discovers
photovoltaic effect
in electrolytic cell

1873:
Willoughby Smith
observes
photoconductivity
in Selenium

1905:
Einstein
publishes paper
on photoelectric
effect (not
photovoltaic
effect!)

1954:
First PV cell made
at Bell Labs, USA.
Used to power
telephone
repeaters – 4%
efficient.
Daryl Chapin, Calvin
Fuller, Gerald Pearson.

1958:
First PV cell in
space. Vanguard
Satellite. Max
power 1W!
Followed Explorer VI
and VII satellites
(1959)

1960:
Hoffman
Electronics
achieves **14%** PV
cells!

Solar Energy Timeline



1977:
U.S. DOE launches
Solar Energy
Research Institute.
Now NREL.
Total PV
manufacturing
capacity exceeds 500
kW

1982:
First MW
capacity solar
plant goes
online.
Hesperia,
California

1994:
GaAs solar cells
> 30%
NREL



1992:
CdTe solar cell
15.9% efficient
Univ. South
Florida.

2000:
First Solar starts
production

2008:
Peak in Si cost as
production of SC
grade Si slows.
**Thin film gets
excited**



REVIEW ARTICLE

PUBLISHED ONLINE: XX JULY 2014 | DOI: 10.1038/NPHOTON.2014.134

nature
photronics

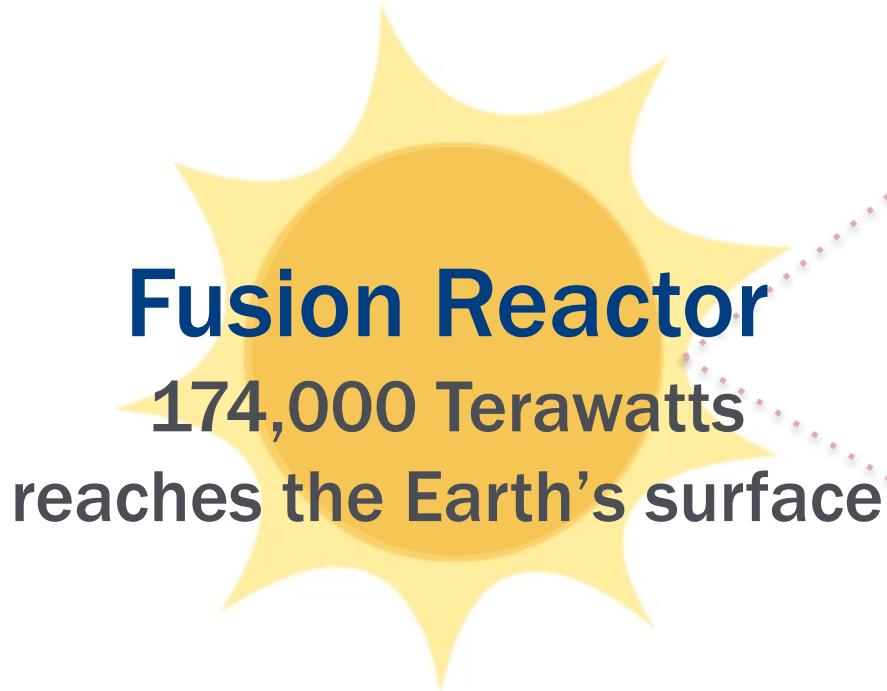
The emergence of perovskite solar cells

Martin A. Green^{1*}, Anita Ho-Baillie¹ and Henry J. Snaith²

EPSRC

Engineering and Physical Sciences
Research Council

Solar Electricity and Fuel



Fusion Reactor
174,000 Terawatts
reaches the Earth's surface

Electricity
Solar Cells

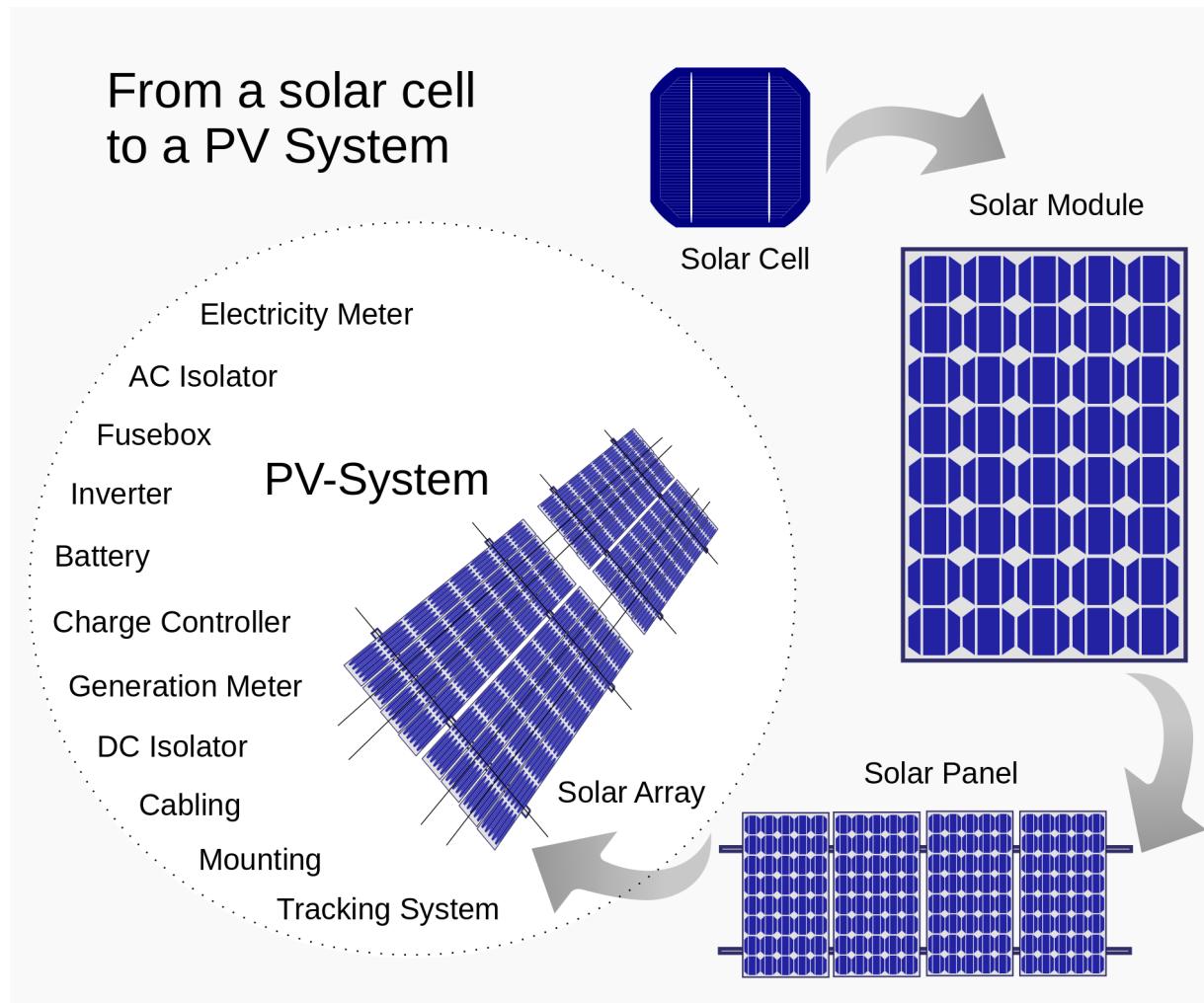
High efficiency (20 – 50%)

Chemical Energy
Solar Fuels

Low efficiency (< 5%)

Physics (electron–hole separation) is easier than
chemistry (oxidation/reduction reactions)

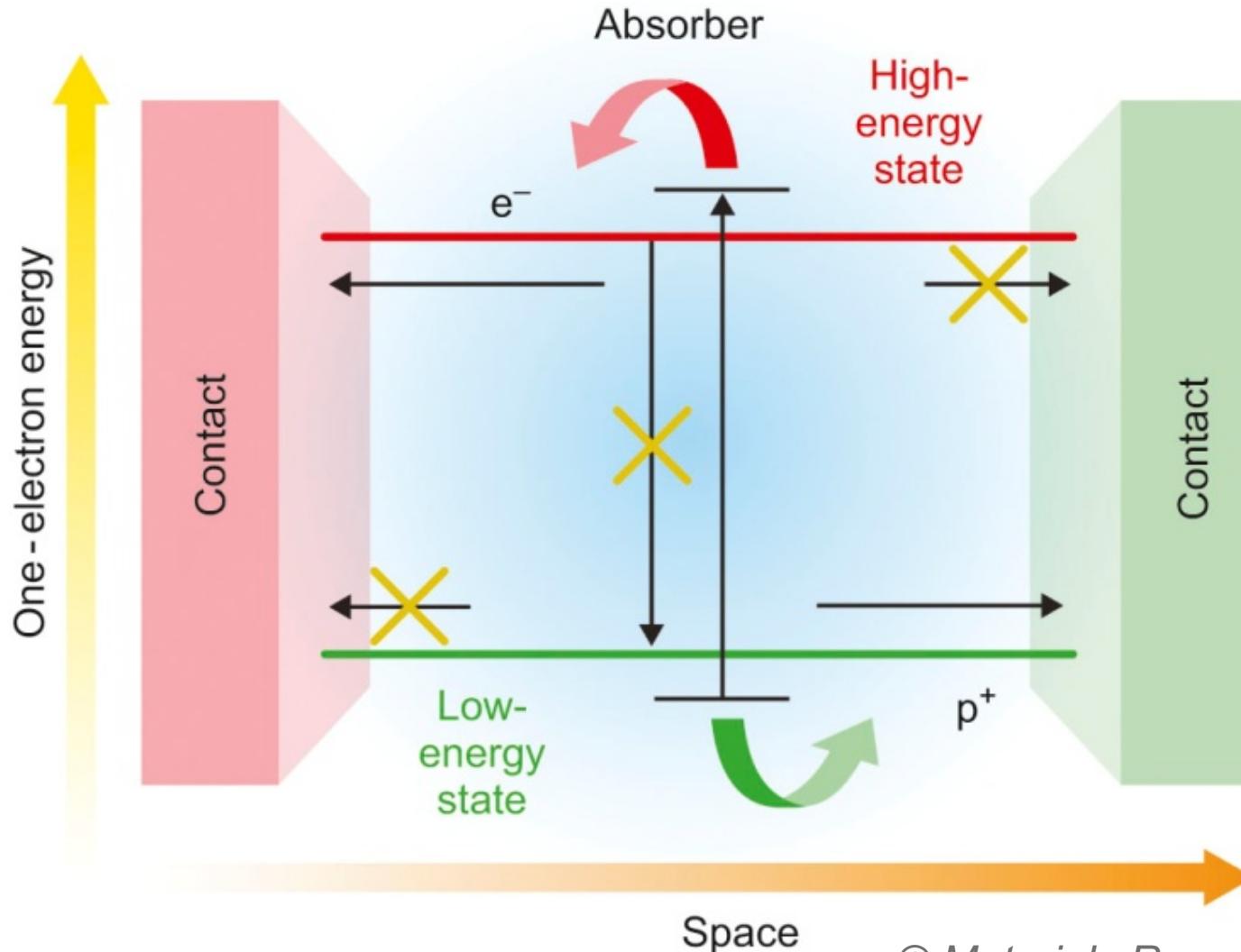
Our Focus: Solar Cells



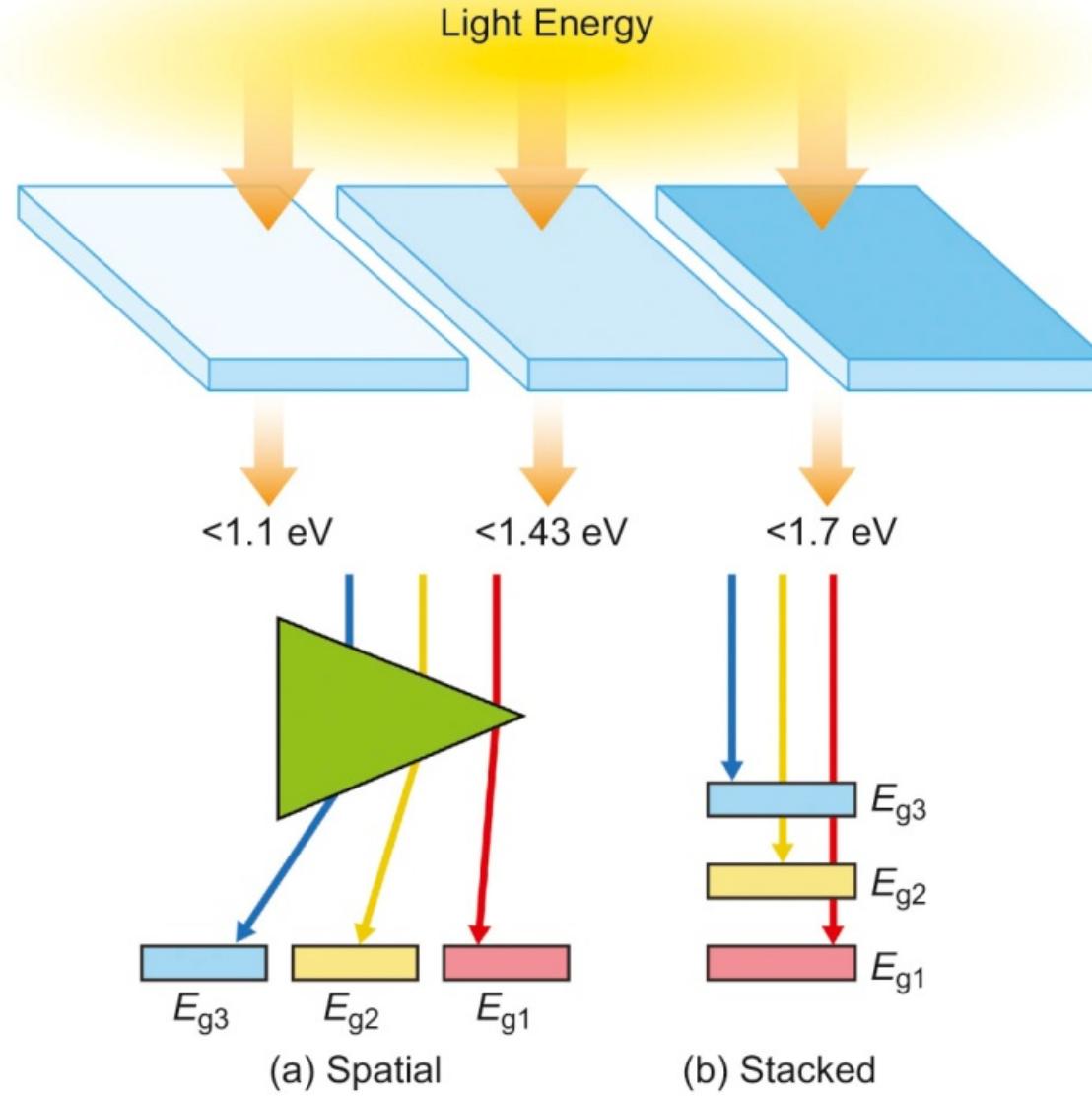
Our Focus: Semiconductors

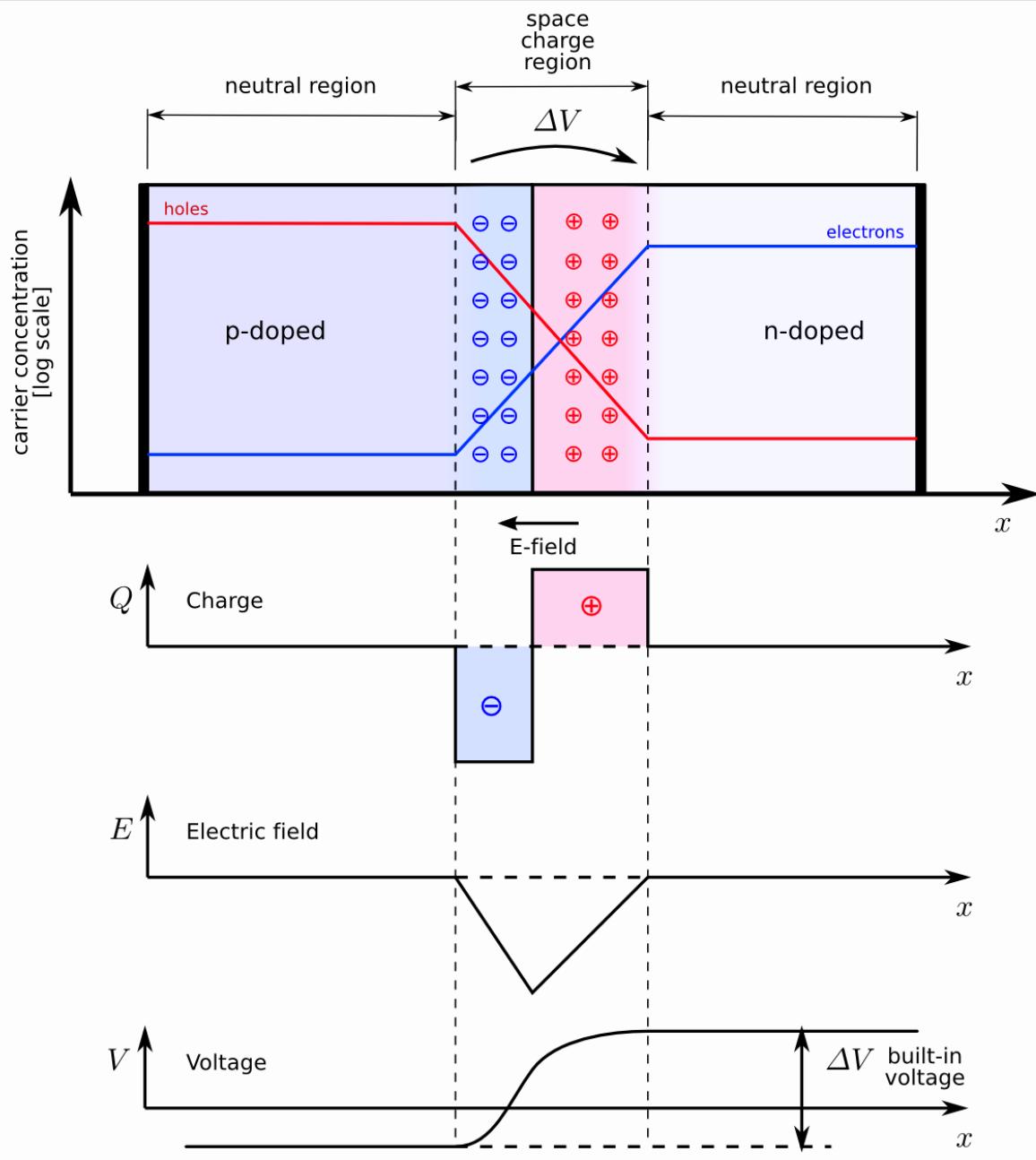
- **Optical Materials** – light collection and trapping, spectral up / down conversion
- **Electronic Materials** – primary light absorption and light-to-electricity conversion. Electrical contacts
- **Thermal Materials** – heat conversion and collection, e.g. thermoelectrics, thermionic emission, thermophotovoltaics

Single Junction Solar Cell



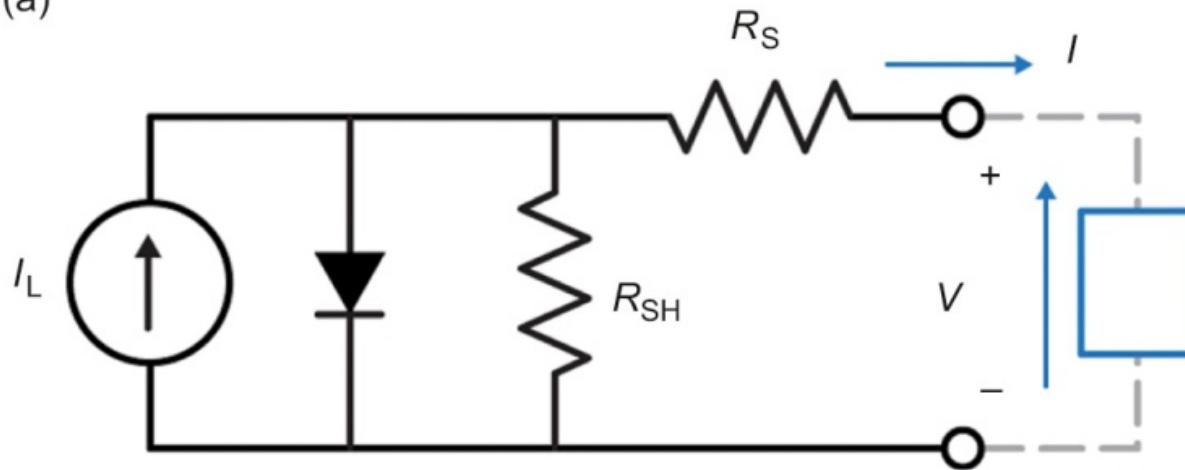
Multi Junction Solar Cell



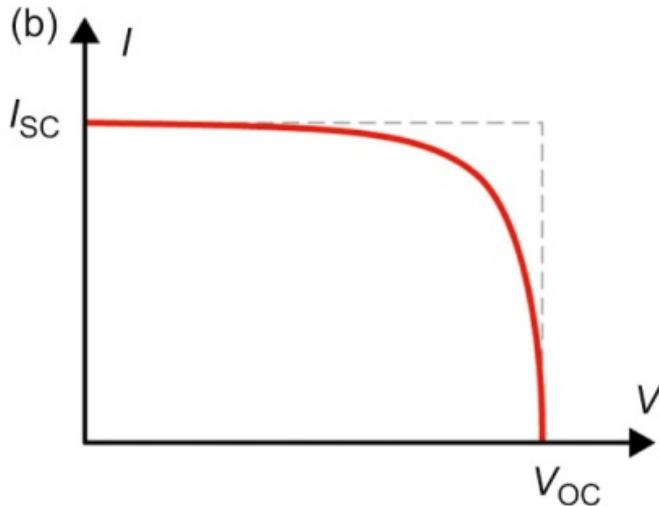


Equivalent Electric Circuit

(a)



(b)



Series Resistance (R_S) – limitation of real materials: resistance at interfaces + bulk
<http://www.pveducation.org/pvcdrom/series-resistance>

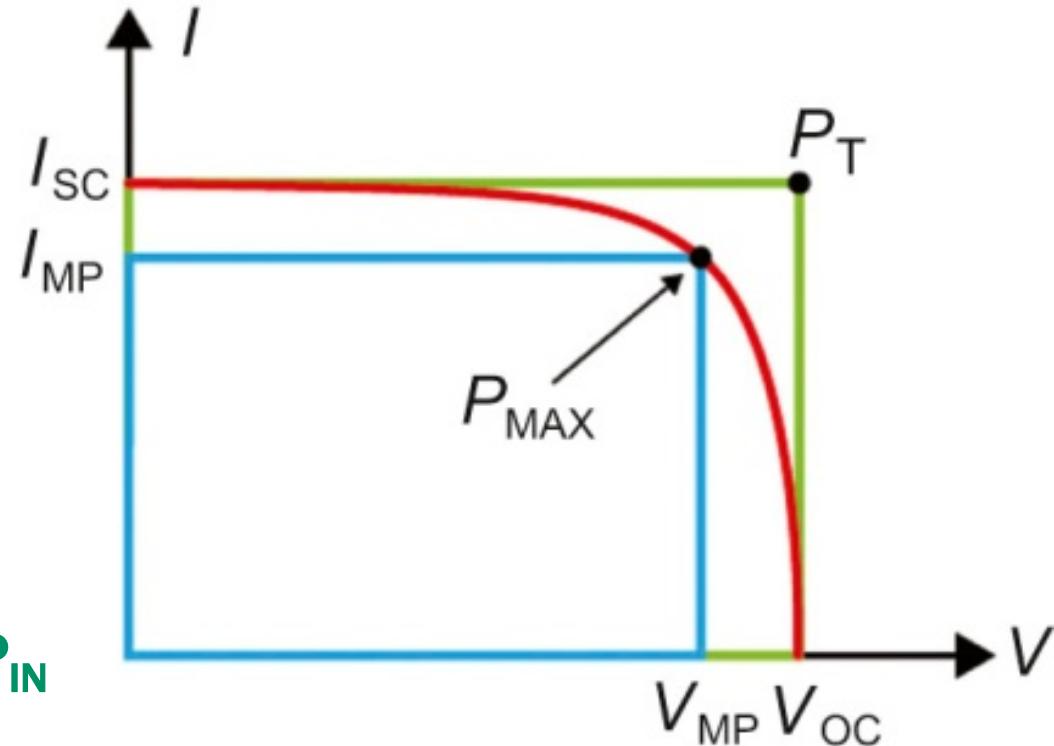
Shunt Resistance (R_{SH}) – limitation of device fabrication: short circuit pathways
<http://www.pveducation.org/pvcdrom/shunt-resistance>

Fill Factor and Photovoltaic Efficiency

$$FF = \frac{P_{MAX}}{P_T} =$$

Fill Factor

$$\frac{I_{MP} \cdot V_{MP}}{I_{SC} \cdot V_{OC}}$$



Device Efficiency: $\eta = P_{MAX} / P_{IN}$

External Quantum Efficiency: $EQE(\lambda)$

Efficiency of photons to e-h pairs

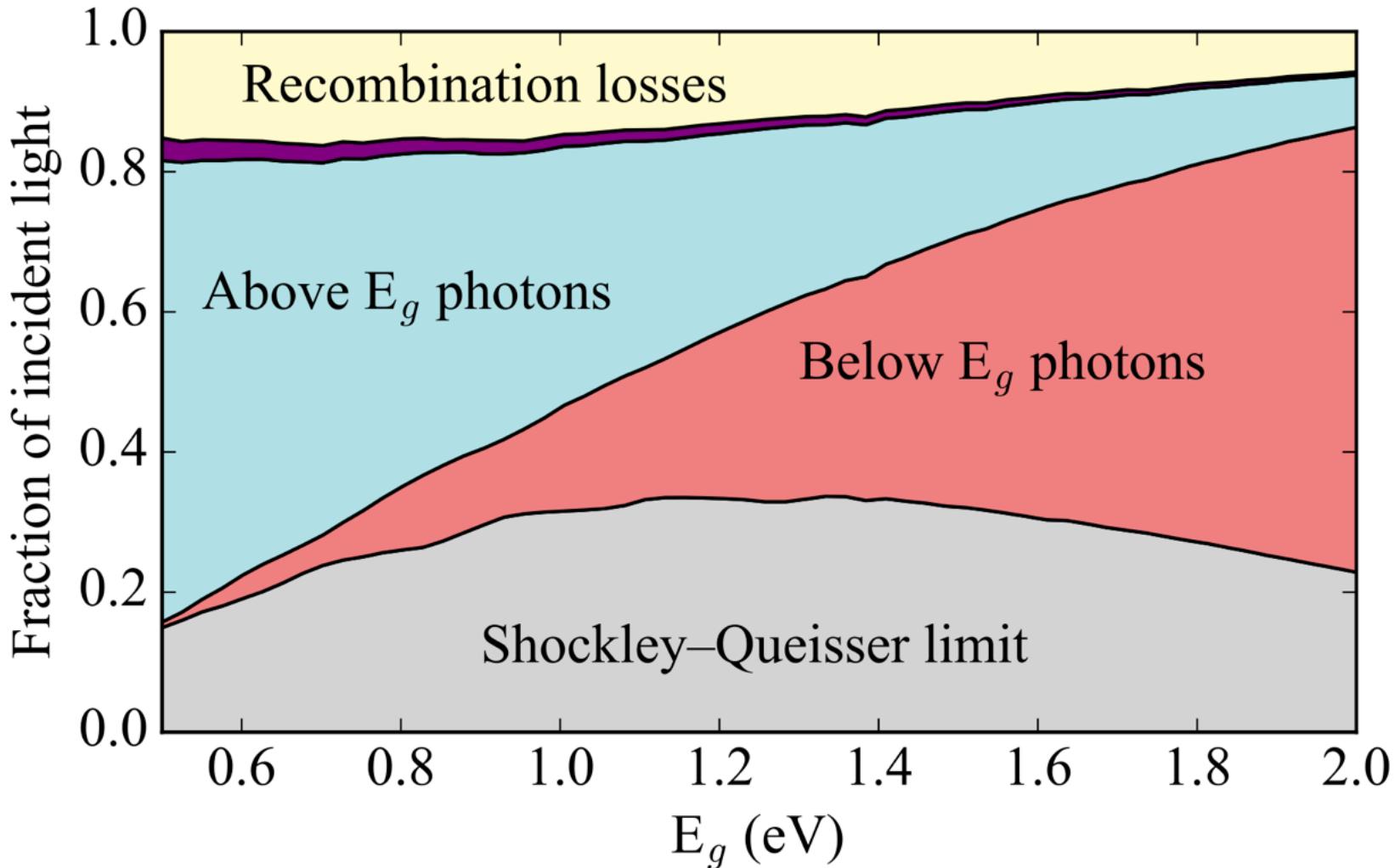
Internal Quantum Efficiency: $IQE(\lambda)$

Exclude reflectance and transmission

Class Question

What is the maximum efficiency that can be achieved?

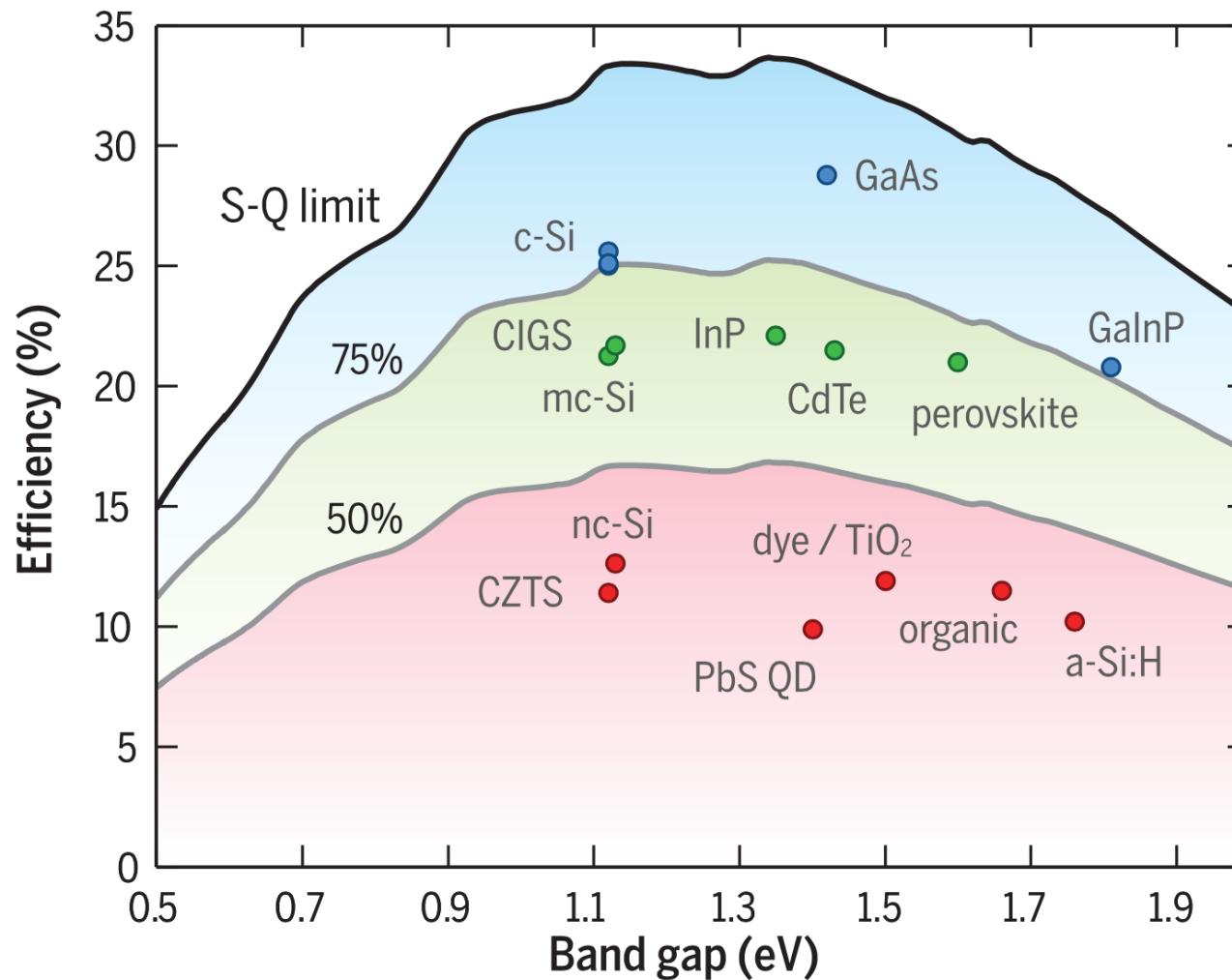
pyExample in Jupyter Notebook



Talk Outline: Solar Energy

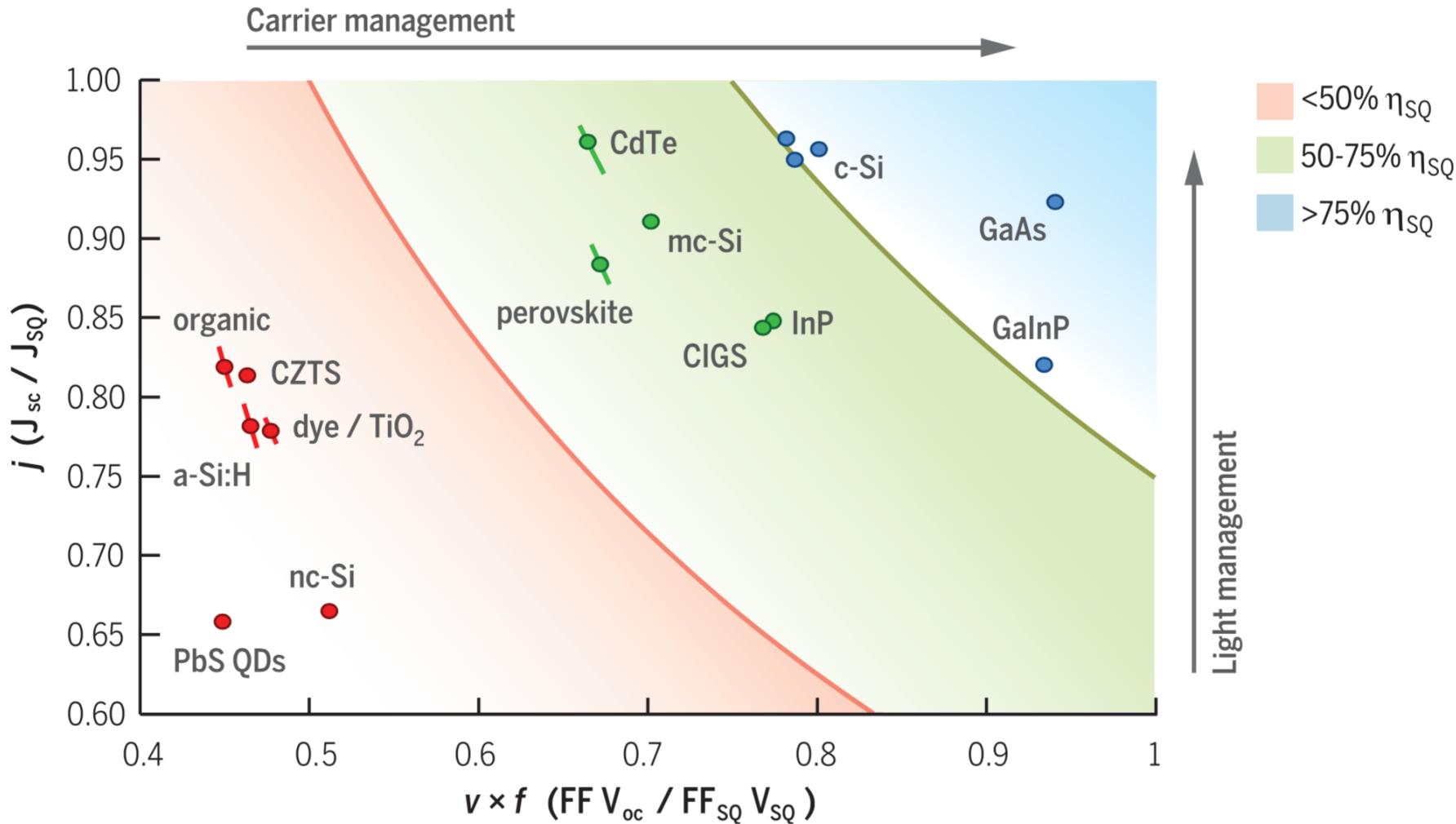
- 1. Solar Energy Fundamentals**
- 2. Current Photovoltaic Technologies**
- 3. Case Study: Hybrid Perovskites**

Many Materials and Technologies

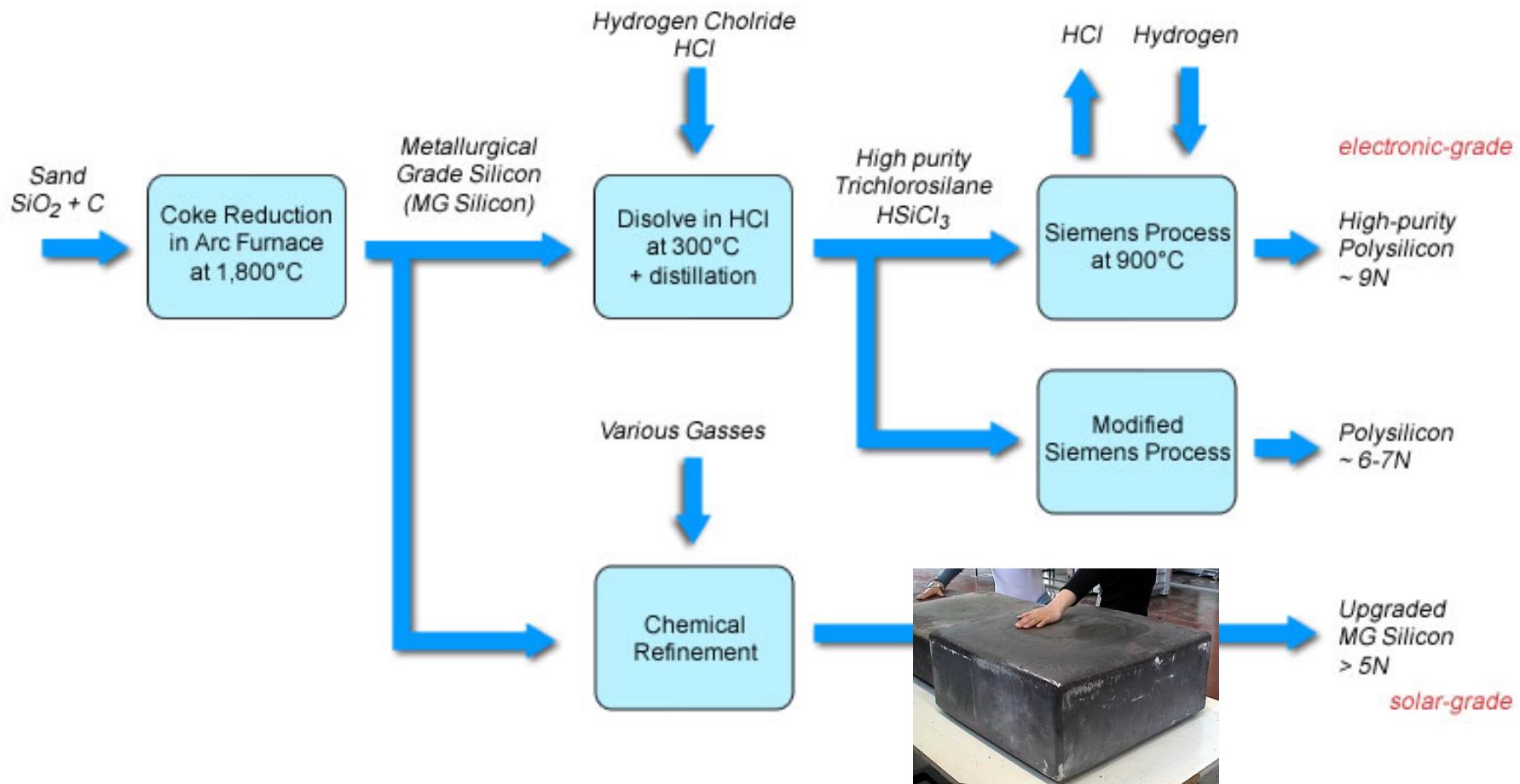


A. Polman et al, Science 352, 307 (2016)

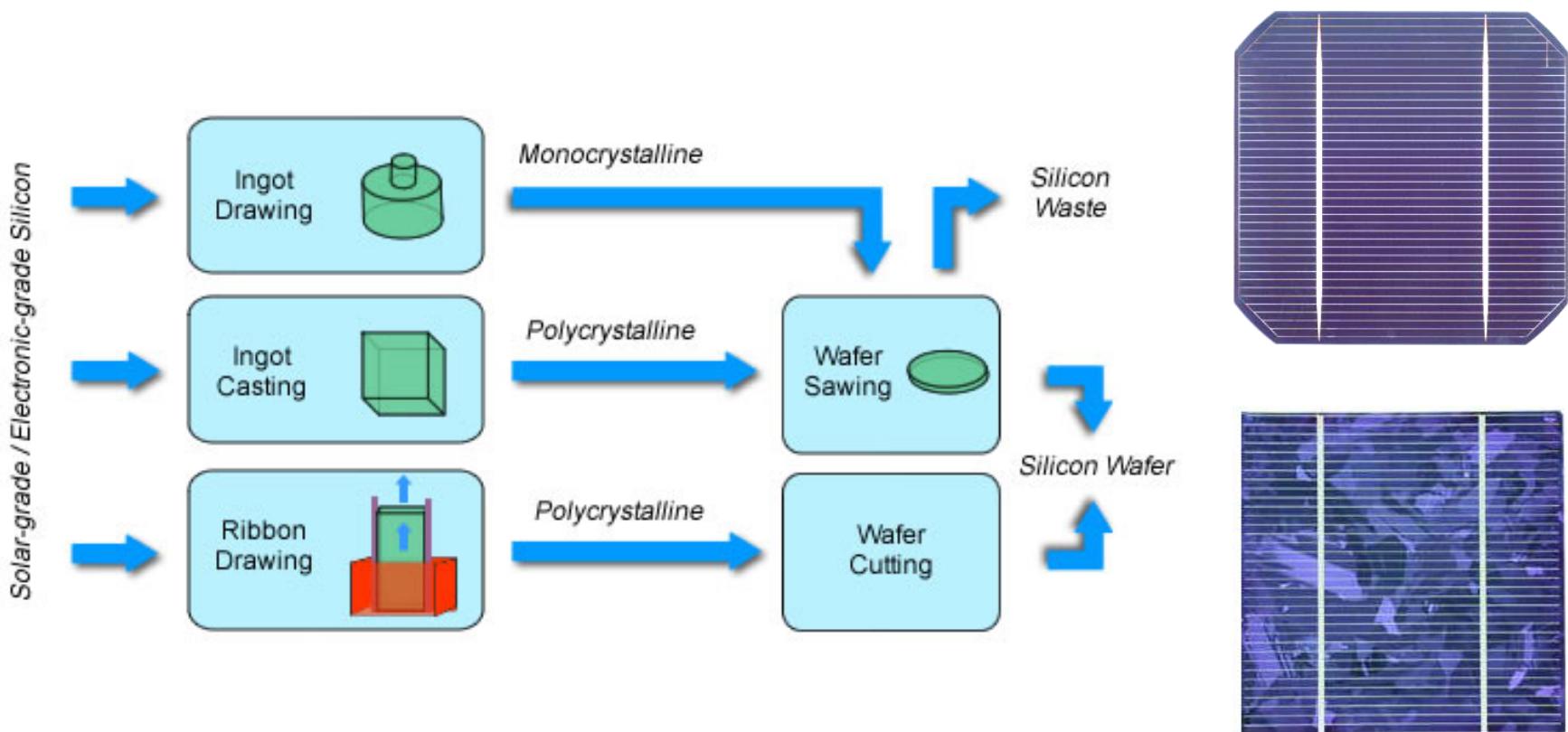
Efficiency Bottlenecks



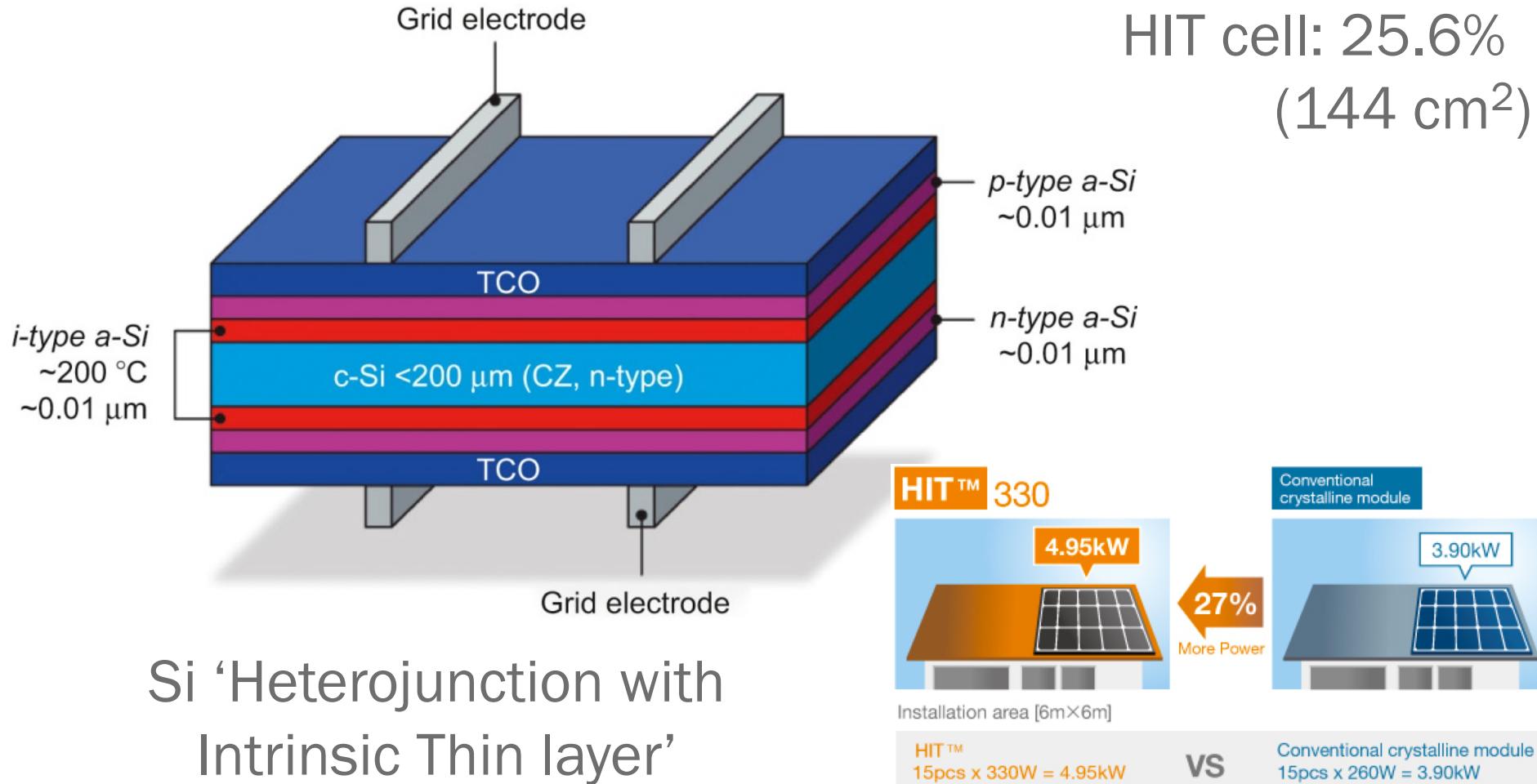
Silicon: Energy Intensive Processing



Silicon: Energy Intensive Processing



Silicon: Robust Technology



Si ‘Heterojunction with
Intrinsic Thin layer’

Silicon: Robust Technology

HIT Double
Photovoltaic Module



Front surface



Back surface



Patio Awning: 1.4kW
Solar Living Design Lakewood, CO

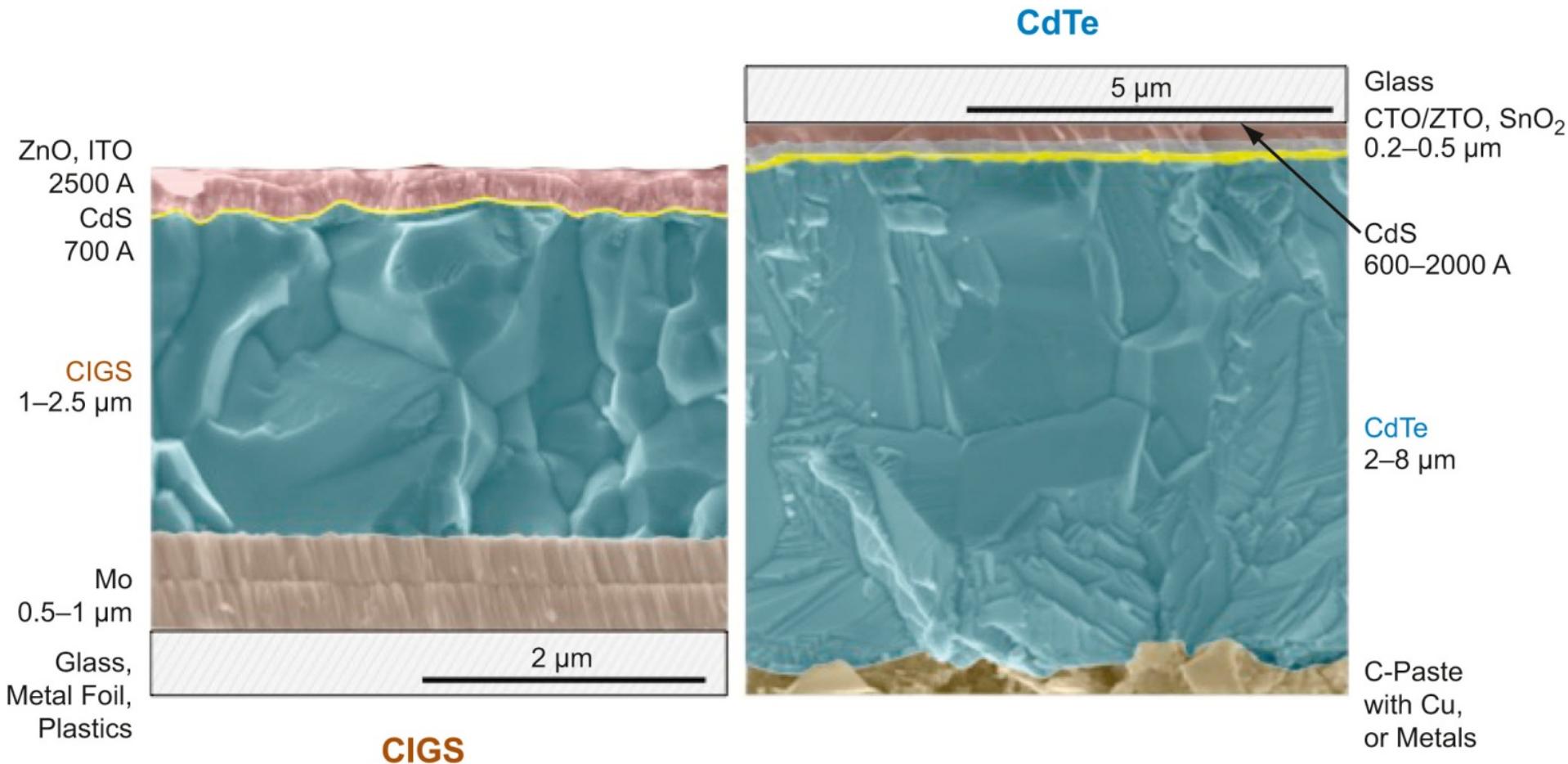


Canopy: 11.9kW
Acme International Services, Inc. Cayman Islands.



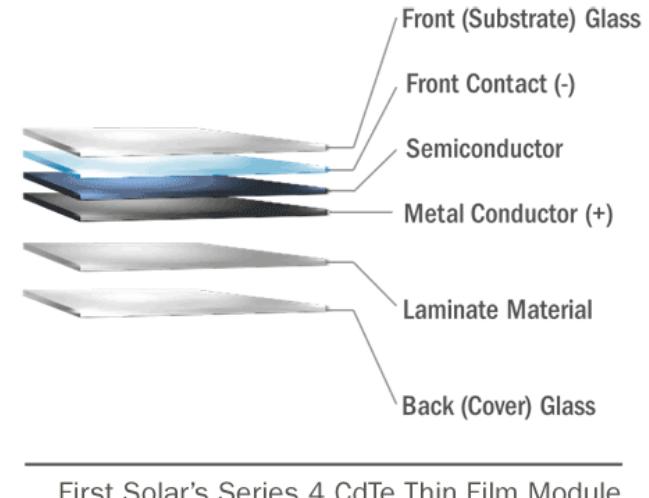
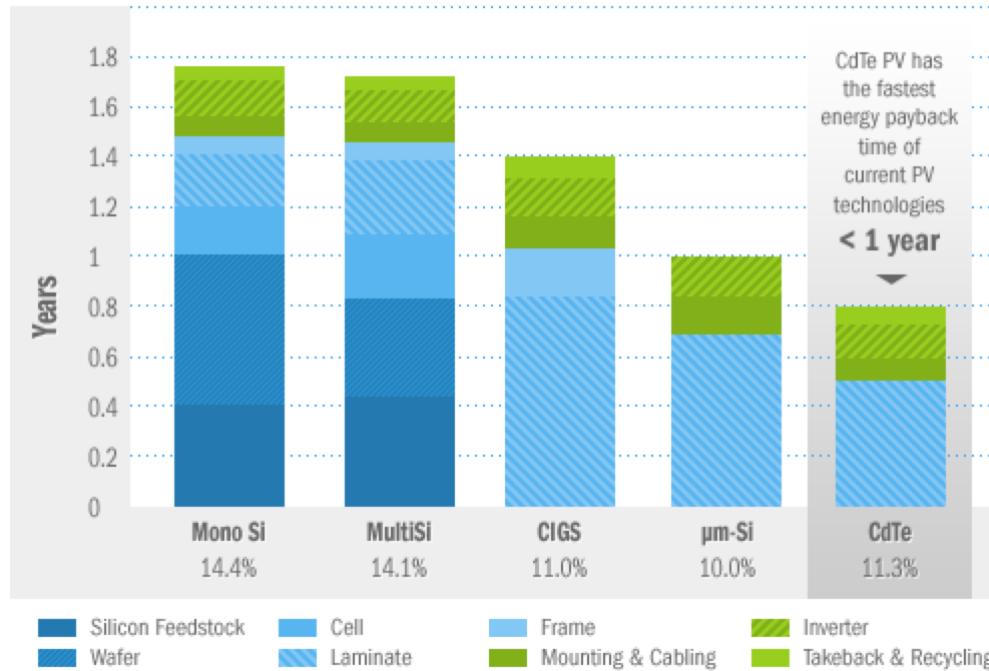
Canopy: 10.4kW
New American Home, Las Vegas, NV.

Thin Films: Indirect to Direct Band Gap



Thin Films: Rapid Energy Payback

$$\text{EPBT} = E_{\text{input}} / (E_{\text{output}} / \text{yr})$$



First Solar's Series 4 CdTe Thin Film Module



Thin Films: Many New Materials

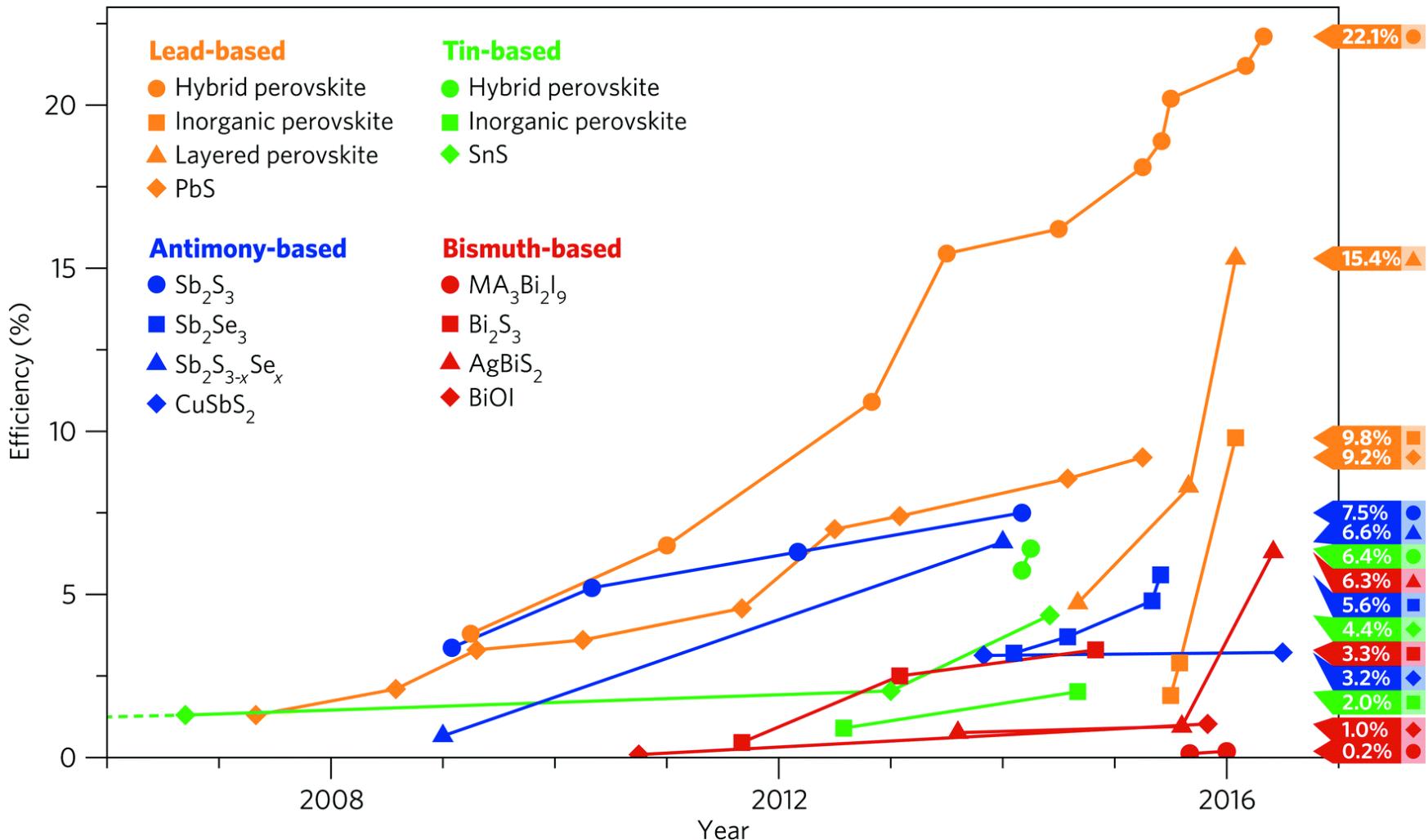
Discovering and optimising new candidate materials is an active research field

Cu_2O	SnS	SnZrS_2	CdTe	Cu_3N	ZnSnP_2
BiFeO_3	CuInS_2	CuSbS_2	WSe_3	CuTaN_2	Cu_3PS_4
CuTaS_4	$\text{Cu}_2\text{ZnSnS}_4$	CuBiS_2	Sb_2Se_3	LiZnN	FeSi_2
Fe_2SiS_4	Cu_2S	Zn_3N_2	AgSbS_2	FeSe ₂	FeS ₂
CsPbI_3	$\text{CH}_3\text{NH}_3\text{PbI}_3$	CsSnI_3	CuTaN_2	Bi_2S_3	Sb_2S_3

Talk Outline: Solar Energy

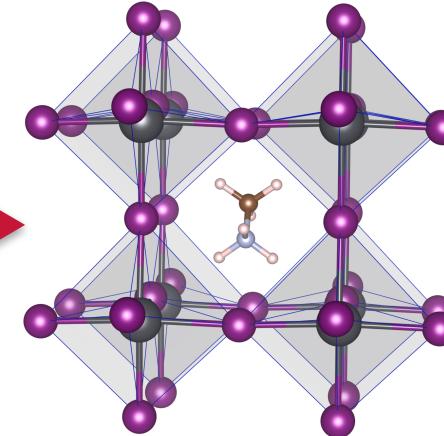
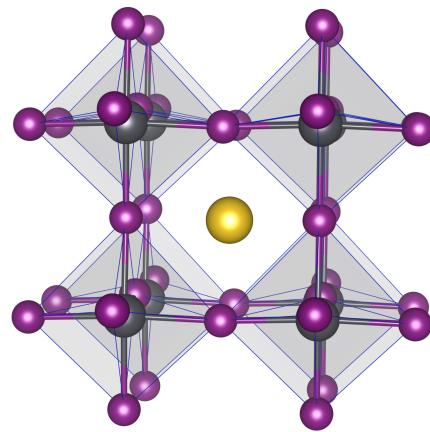
- 1. Solar Energy Fundamentals**
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Rise of Perovskite Solar Cells



Hybrid Organic-Inorganic Perovskites

Inorganic
 CsPbI_3



Hybrid
 $\text{CH}_3\text{NH}_3\text{PbI}_3$
or MAPI

Brief History

- (1958) – Photoconductivity in CsPbI_3 (Møller)
- (1978) – Synthesis of $\text{CH}_3\text{NH}_3\text{PbI}_3$ (Weber)
- (1994) – Metallic transition in $\text{CH}_3\text{NH}_3\text{SnI}_3$ (Mitzi)
- (2009) – Perovskite dye cell (Miyasaka)
- (2013) – Planar thin-film solar cell (Snaith)

Why Hybrid Perovskites?

Essentials for Solar Cells

- Strong optical absorption ($E_g \sim 1.6$ eV)
- Light electron and hole masses (*conductive*)
- Easy to synthesise (*cheap and scalable*)

Advanced Features

- Large dielectric constants: carrier separation (weak excitons) and transport (low scattering)
- Slow $e\text{-}h$ recombination: low losses, large V_{oc}
 - Relativistic effects – spin-orbit coupling
 - Ferroelectricity – dynamic fluctuations

Dielectric Response

Organic-Inorganic Dielectrics

$$\epsilon_0 = \epsilon_\infty + \epsilon_{ionic} + \epsilon_{rotation} + \epsilon_{other}$$

Standard Inorganic Dielectric

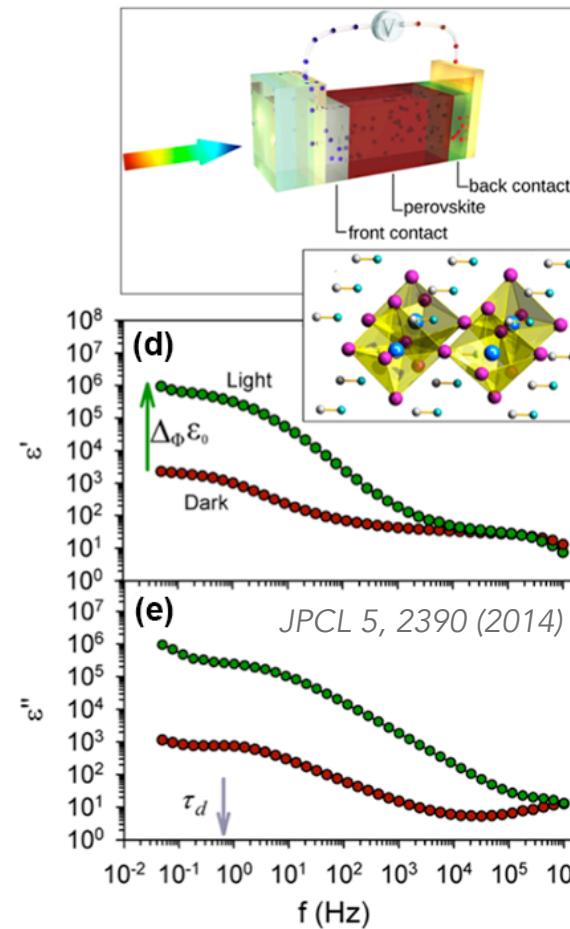
**Microstructure
Conductivity
Contacts**

CdTe	CuInSe ₂	Cu ₂ ZnSnS ₄
$\epsilon_0 \sim 10$	$\epsilon_0 \sim 13$	$\epsilon_0 \sim 13$
$\epsilon_\infty \sim 7$	$\epsilon_\infty \sim 8$	$\epsilon_\infty \sim 7$
$\epsilon_{ionic} \sim 3$	$\epsilon_{ionic} \sim 5$	$\epsilon_{ionic} \sim 6$

$$\begin{aligned} \text{CH}_3\text{NH}_3\text{PbI}_3 \\ \epsilon_0 \sim 34 \\ \epsilon_\infty \sim 5 \\ \epsilon_{ionic} \sim 20 \\ \epsilon_{rotation} \sim 9(300\text{K}) \end{aligned}$$

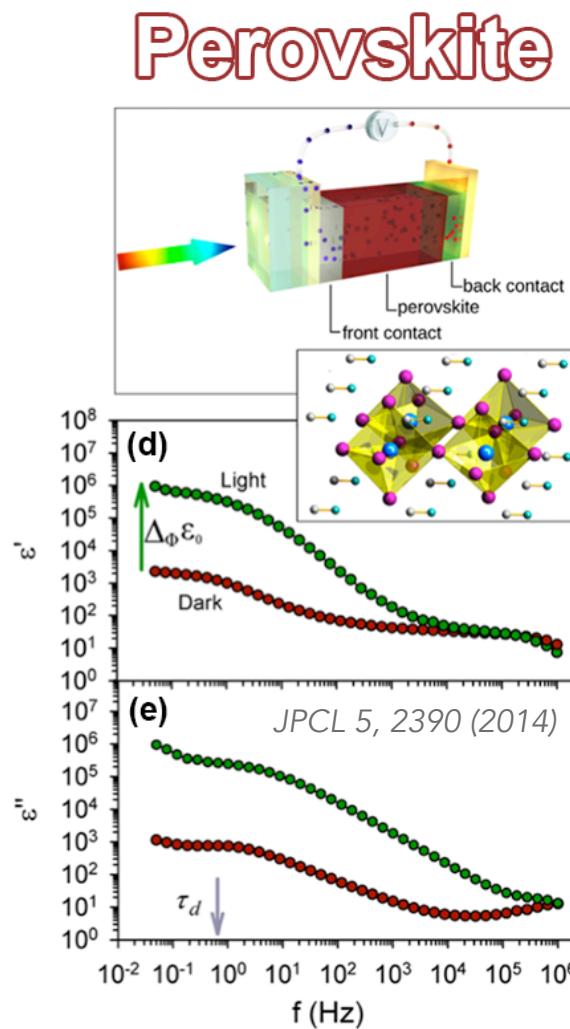
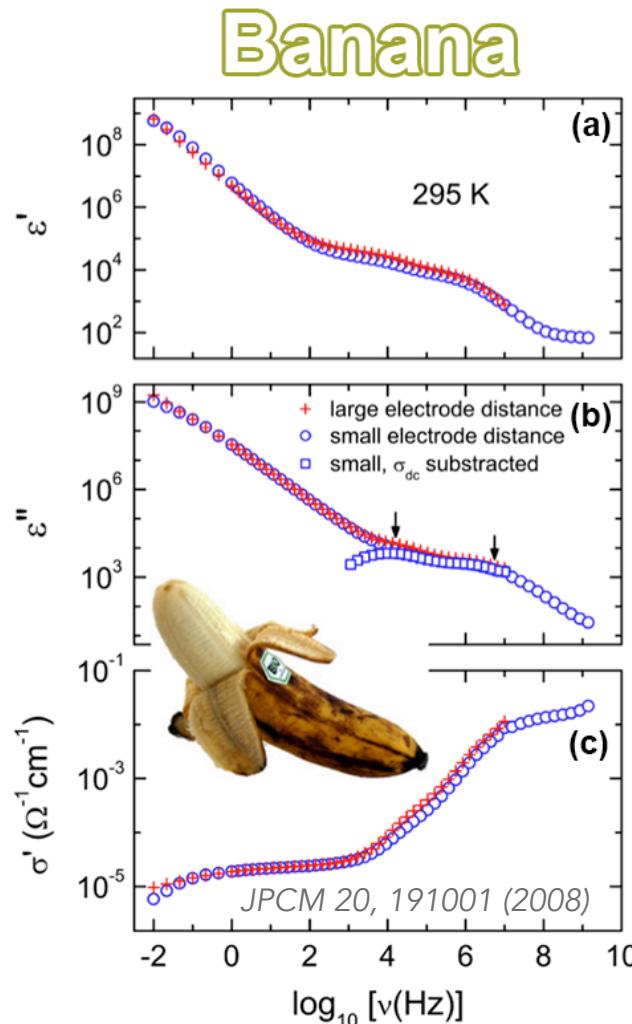
“Giant Dielectric Constant”

Perovskite

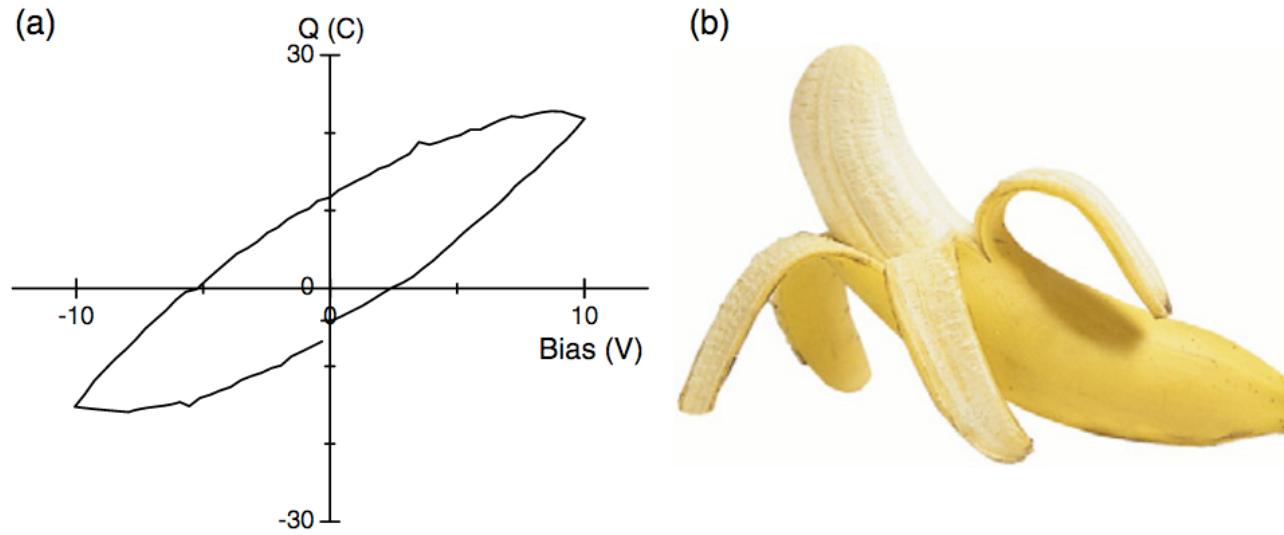


Frost and Walsh, Acc. Chem. Res. 49, 528 (2016)

“Giant Dielectric Constant”



Bananas are Lossy Dielectrics



Conclusions

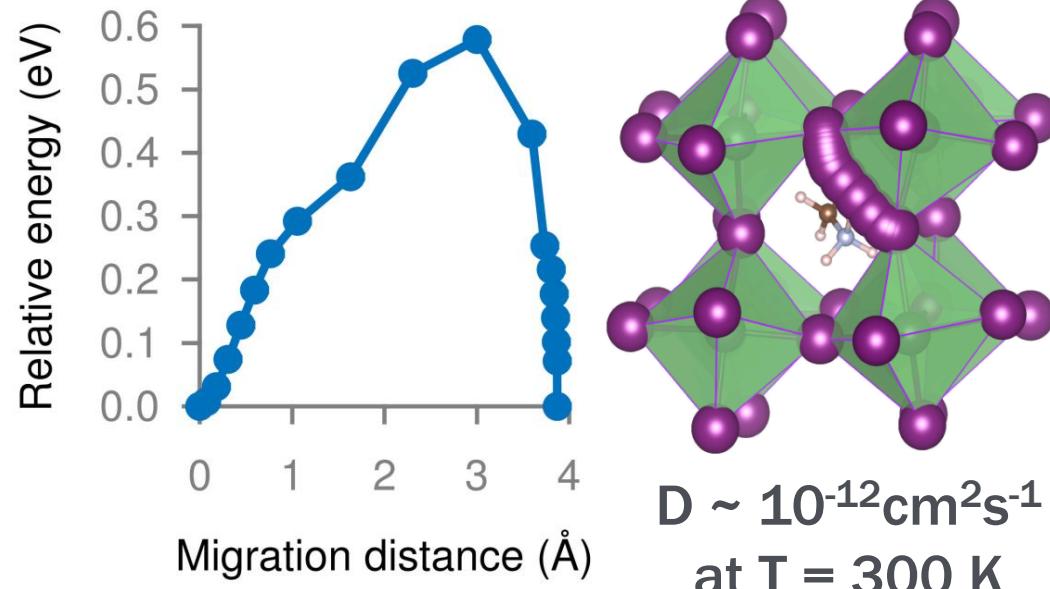
If your ‘hysteresis loops’ look like figure 1(a), please do not publish them. Publish data that are saturated and have a region in Q versus V that is concave. Bananas are not ferroelectric, and it is easy to be misled by closed $Q(V)$ loops.

Mixed Ionic–Electronic Conductors

Reservoir of charged point defects (site vacancies)
in thermodynamic equilibrium: V_{MA}^- , V_{Pb}^{2-} , V_I^+

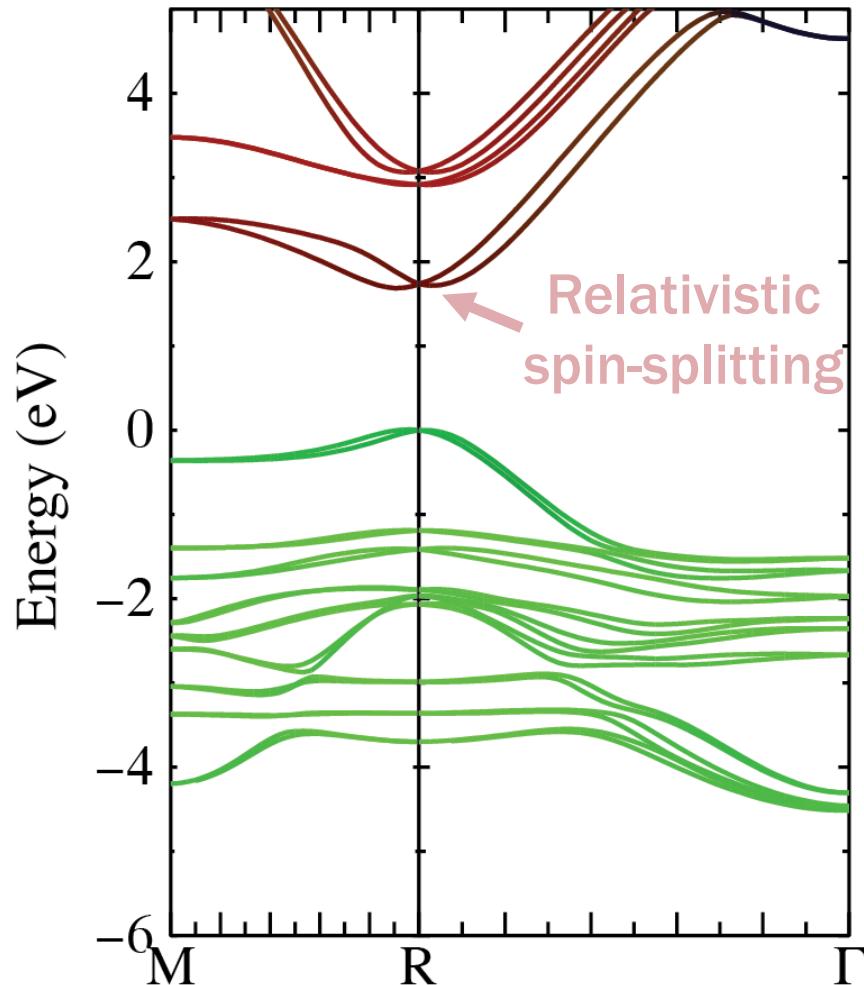
Angewandte Chemie 54, 1791 (2015)

Vacancy	E_a (eV)
CH_3NH_3^+	0.8
Pb^{2+}	2.3
I^-	0.6



C. Eames et al, Nature Comm. 6, 8497 (2015)

Electronic Band Structure (MAPI)



Conduction
Band

Degeneracy removed
by Δ_{CF} and Δ_{SOC}

Pb 6p⁰

$$E_g^{QSGW} = 2.7 \text{ eV} \rightarrow 1.7 \text{ eV (SOC)}$$

Valence
Band

I 5p⁶
Pb 6s²

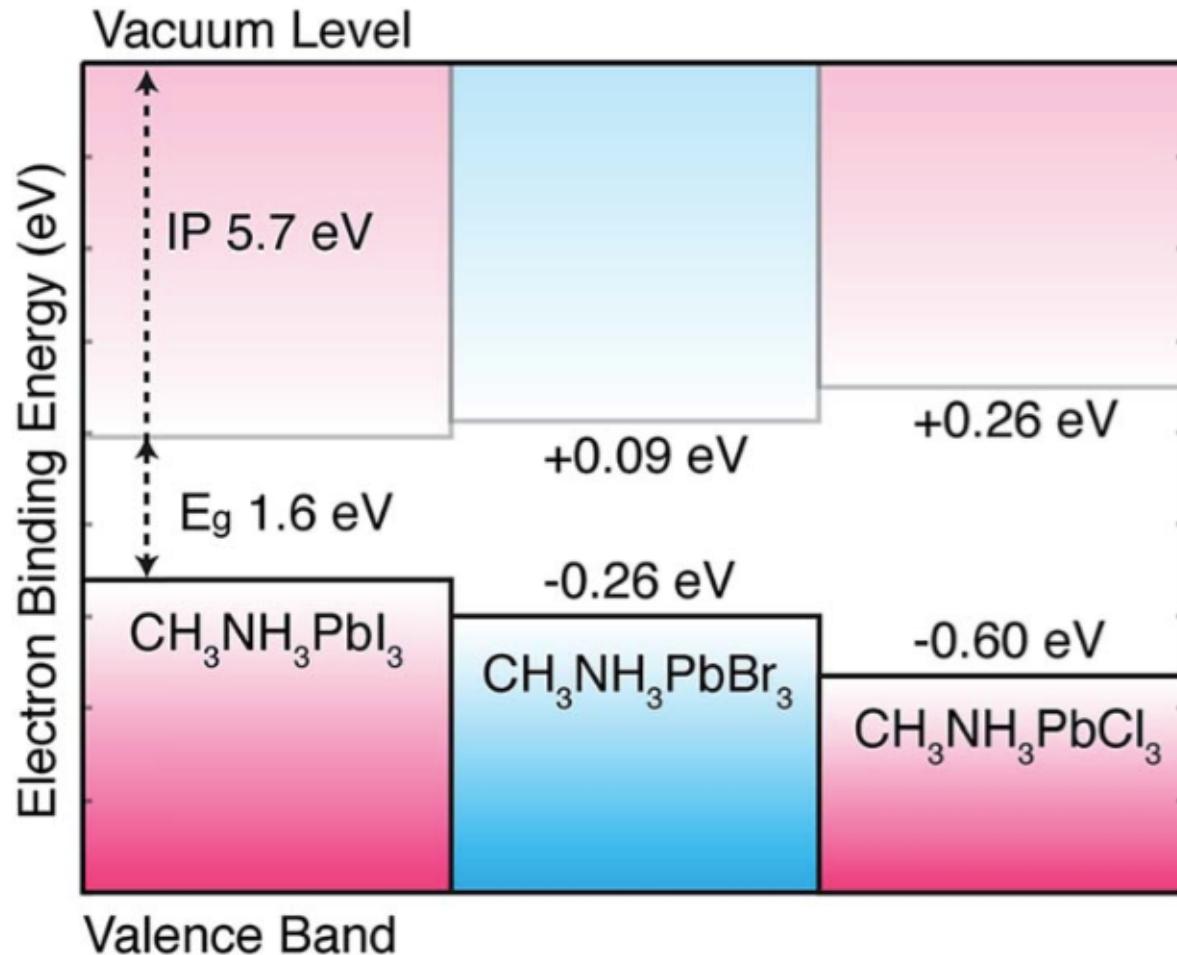
Band Gap Engineering: Principles

A site: Usually electronically inactive, but can change structure (volume and tilting pattern)

B site: Forms lower conduction band (Pb is key for electron affinity and transport)

X site: Forms upper valence band (I is key for ionisation potential and hole transport)

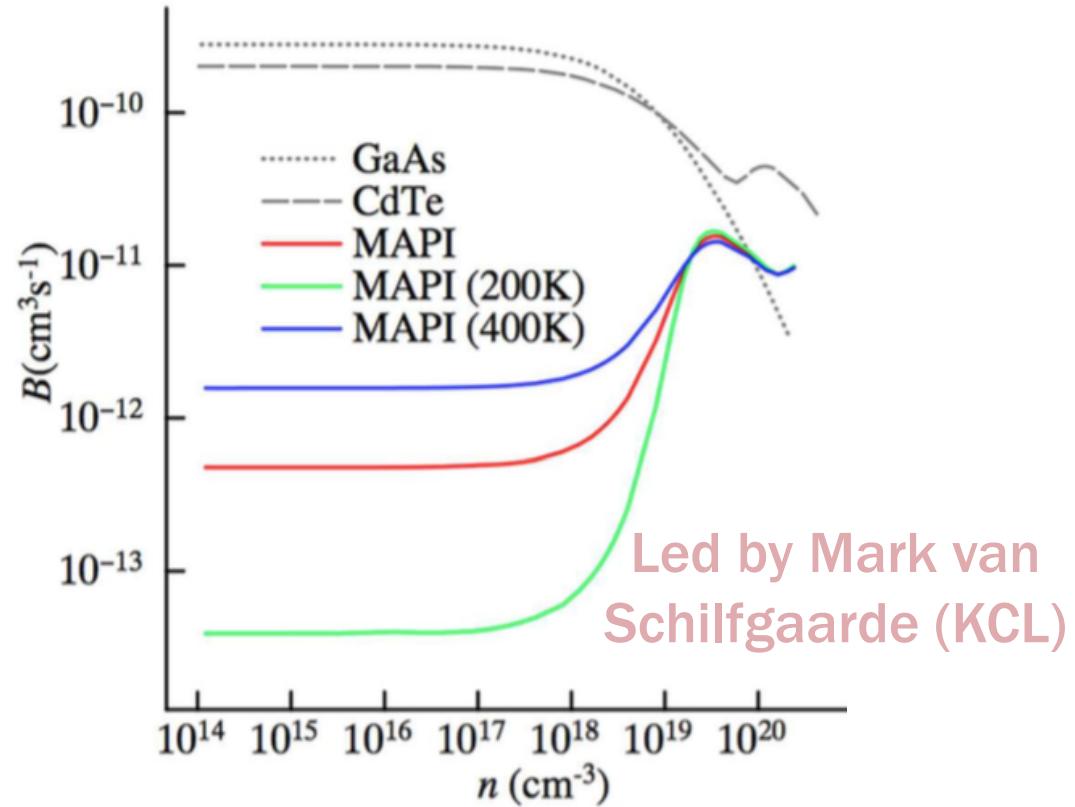
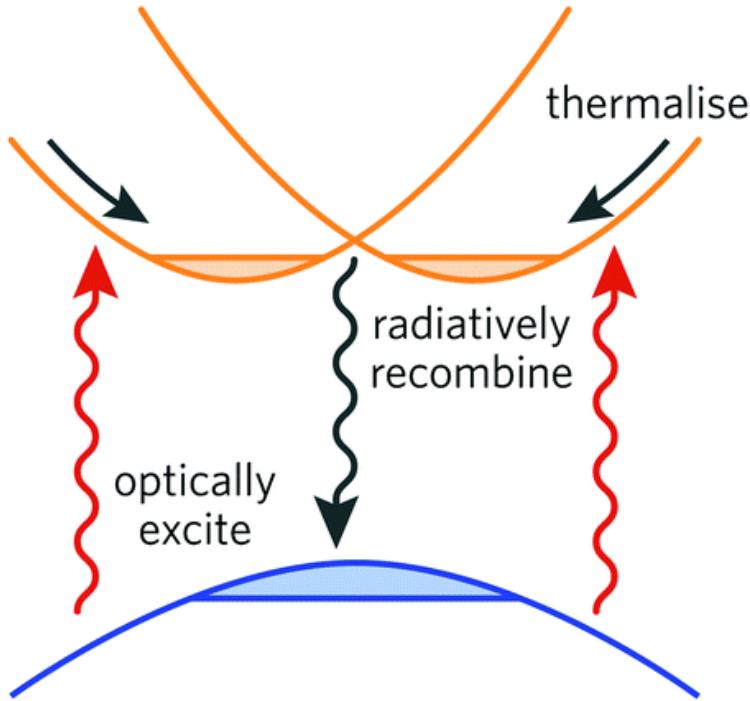
Band Gap Engineering: Halides



Band offsets from
heterojunction core
level alignments:
HSE06+SOC

Rashba and Radiative Recombination

Rashba splitting of conduction band reduces radiative recombination at low fluence



Support for Weakly Indirect Gap

Validation from experiment is growing

Indirect to direct band gap transition under pressure

Led by Bruno Ehrler (AMOLF); Energy. Environ. Science (2017);
DOI: [10.1039/c6ee03474h](https://doi.org/10.1039/c6ee03474h)

Indirect to direct band gap transition with fluence

Led by Sam Stranks (Cambridge); Nature Materials (2016);
DOI: [10.1038/nmat4765](https://doi.org/10.1038/nmat4765)

Giant Rashba splitting in MAPbBr_3 with ARPES

Led by Thomas Fauster (Erlangen); Physical Review Letters (2016); DOI: [10.1103/PhysRevLett.117.126401](https://doi.org/10.1103/PhysRevLett.117.126401)

Perovskites: Challenges & Opportunities

Local Structure – correlation lengths

Dynamic Disorder – connection to optoelectronics

Ionic Conductivity – how to limit

Electrical Conductivity – control p-type and n-type

Chemical Stability – breakdown with O₂ and H₂O

Surfaces & Interfaces – poor understanding

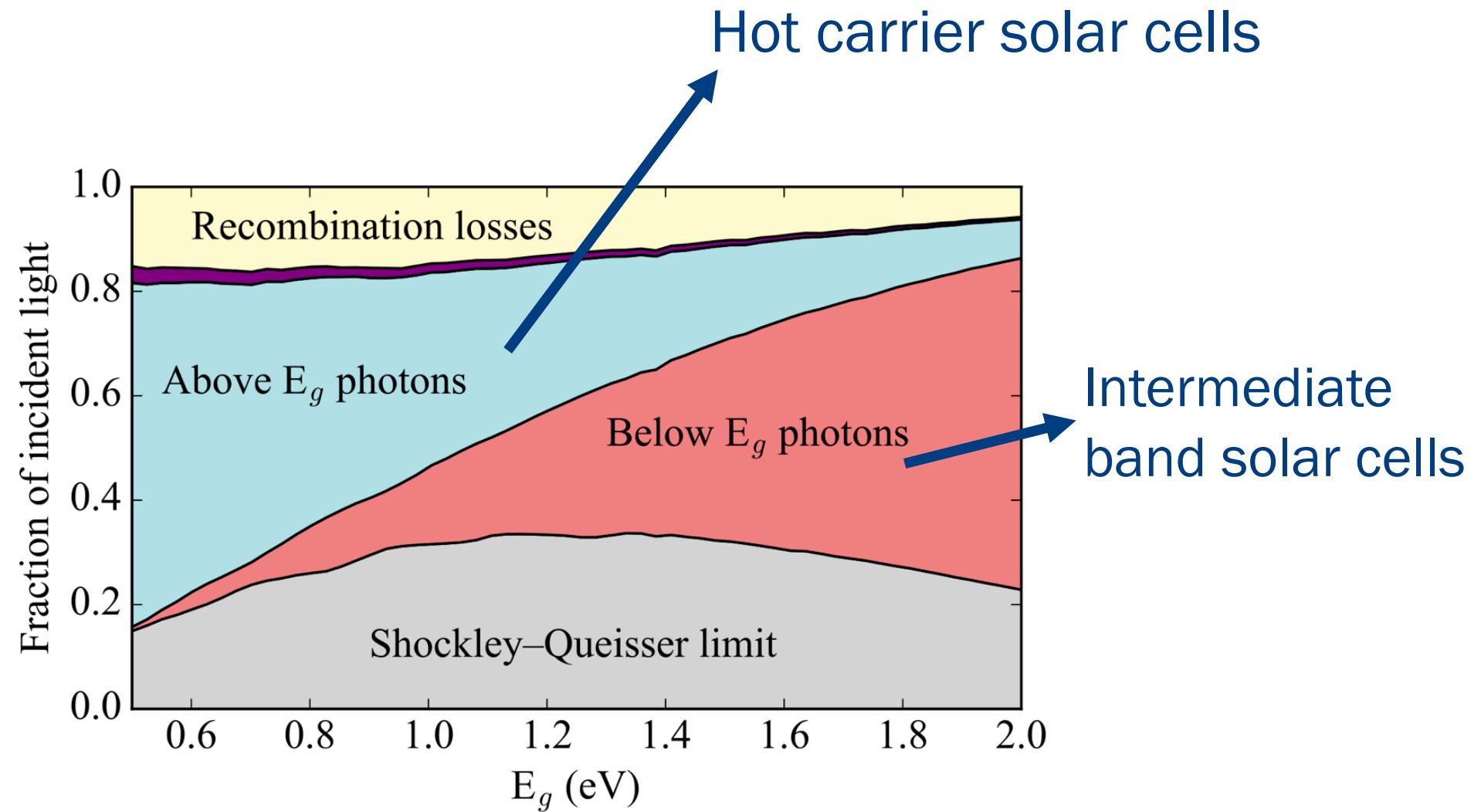
Alloys – thermodynamics and photo-stability

Hysteresis – how to eliminate

Beyond 3D – 2D and 1D hybrid perovskites

Beyond Pb – alternative non-toxic metals

Beyond S-Q Limit ($> 30\%$ Efficiency)



Summary – Key Points

- Motivation and basic science of solar energy harvesting
- Fundamental limits for light to electricity conversion
- Effective materials for photovoltaics are semiconductors that absorb sunlight and transport charges
- Halide perovskites have the potential to displace existing technologies