Effective Core Potentials70 Years of Development –

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Overview

Fundamentals of electronic structure calculations

The idea of an effective core potential (ECP)

Various types of effective core potentials

Recent extensions of the ECP idea

Summary

Fundamentals

Fundamentals of electronic structure calculations

Consider stationary (i. e. timeindependent) states of a system (atom, molecule, ..., crystal) with n electrons and K space-fixed nuclei.

Solve the Schrödinger equation:

$$\mathcal{H}\Psi_k = E_k \Psi_k$$

for E_k and Ψ_k , where k = (0), 1, 2, ... (under some boundary conditions \rightarrow eigenvalue problem).

Zero-, one-, and two-electron parts of the Hamilton operator (in atomic units):

$$\mathcal{H} = \mathcal{H}_0 + \mathcal{H}_1 + \mathcal{H}_2$$

$$\mathcal{H}_0 = \sum_{\lambda < \mu}^K \frac{Z_\lambda Z_\mu}{R_{\lambda \mu}} \quad - \text{ nuclear repulsion } (Z_\lambda \text{ nuclear charge,} \\ R_{\lambda \mu} \text{ internuclear distance})$$

$$\mathcal{H}_1 = \sum_{i=1}^n h(i) \quad - \text{ one-electron part } (h \text{ is a one-electron operator})$$

$$\mathcal{H}_2 = \sum_{i < j}^n g(i,j) \quad - \text{ two-electron part } (g \text{ is a two-electron operator})$$

State functions Ψ_k can be expanded in antisymmetrized products Φ_j of single particle functions $\psi_i(\mathbf{x})$ (spin orbitals or spinors), e. g.

$$\Psi_k = \sum_j C_{jk} \Phi_j$$

$$\Phi_j = \frac{1}{\sqrt{n!}} \left| \psi_{j_1} \psi_{j_2} \dots \psi_{j_{n-1}} \psi_{j_n} \right|$$

 Φ_i — Slater determinants

This expression automatically fulfills the Pauli principle for fermions (e. g. electrons)

$$\Phi_j(\ldots,\mathbf{x}_a,\ldots,\mathbf{x}_b,\ldots) = -\Phi_j(\ldots,\mathbf{x}_b,\ldots,\mathbf{x}_a,\ldots)$$

while for bosons (e. g. α -particle, ⁴He)

$$\Phi_j(\ldots,\mathbf{x}_a,\ldots,\mathbf{x}_b,\ldots) = +\Phi_j(\ldots,\mathbf{x}_b,\ldots,\mathbf{x}_a,\ldots)$$

A simple example: n = 2

$$\Phi(\mathbf{x}_{1}, \mathbf{x}_{2}) = \frac{1}{\sqrt{2}} |\psi_{k} \psi_{l}| = \frac{1}{\sqrt{2}} \begin{vmatrix} \psi_{k}(\mathbf{x}_{1}) & \psi_{k}(\mathbf{x}_{2}) \\ \psi_{l}(\mathbf{x}_{1}) & \psi_{l}(\mathbf{x}_{2}) \end{vmatrix}$$
$$= \frac{1}{\sqrt{2}} (\psi_{k}(\mathbf{x}_{1}) \psi_{l}(\mathbf{x}_{2}) - \psi_{l}(\mathbf{x}_{1}) \psi_{k}(\mathbf{x}_{2}))$$
$$= -\Phi(\mathbf{x}_{2}, \mathbf{x}_{1})$$

 \rightarrow electrons are indistinguishable ('each electron uses every single particle function')

For the ground state of the helium atom, He $1s^2$ ¹S, choose:

$$\psi_k(\mathbf{x}) = \psi_{1s}(\mathbf{r})\alpha \qquad \qquad \psi_l(\mathbf{x}) = \psi_{1s}(\mathbf{r})\beta$$

(or vice versa), and determine $\psi_{1s}(\mathbf{r})$.

Determination of the 'best' set of single electron functions $\{\psi_j\}$ from the variational principle, applied to the energy expectation value

$$E[\{\psi_j\}] = <\Psi_k|\mathcal{H}|\Psi_k> \qquad \text{with} \qquad <\Psi_k|\Psi_k> = 1$$

$$E_k \leq \min_{\{\psi_j\}} E[\{\psi_j\}]$$

under the orthogonality and normalization conditions $<\psi_i|\psi_j>=\delta_{ij}$

→ Euler-Lagrange equations

$$\frac{\delta E}{\delta \psi_j} = 0$$

Their precise form depends on the underlying 'physical picture' (e. g. non-relativistic / relativistic treatment) and the form of the state function Ψ_k involved.

Canonical Fock equations for the spatial part $\psi_j(\mathbf{r})$ of single electron functions (closed-shell case, non-relativistic treatment):

$$\frac{\delta E}{\delta \psi_j} = 0 \quad \Rightarrow \quad (\mathcal{F} - \varepsilon_j)\psi_j = 0$$

$$\mathcal{F} = h + \sum_i (2\mathcal{J}_i - \mathcal{K}_i)$$

$$\mathcal{J}_i \psi_j(\mathbf{r}) = \psi_j(\mathbf{r}) \int d\tau' \psi_i^*(\mathbf{r}') g \psi_i(\mathbf{r}')$$

$$\mathcal{K}_i \psi_j(\mathbf{r}) = \psi_i(\mathbf{r}) \int d\tau' \psi_i^*(\mathbf{r}') g \psi_j(\mathbf{r}')$$

with one- and two-electron operators (in atomic units)

$$h = \frac{1}{2}\mathbf{p}^2 + \sum_{\lambda} V_{\mathsf{nuc}}^{\lambda}(\mathbf{r}_{\lambda}) \qquad g = \frac{1}{|\mathbf{r} - \mathbf{r}'|}$$

 $\begin{array}{l} \mathbf{p} = -\mathrm{i}\nabla & \text{---} \text{ operator of linear momentum} \\ V_{\mathrm{nuc}}^{\lambda}(\mathbf{r}_{\lambda}) & \text{----} \text{ nuclear potential (for point-like nuclei: } -Z_{\lambda}/r_{\lambda}) \\ \varepsilon_{j} & \text{----} \text{ orbital energy} \\ \psi_{j} & \text{----} \text{ orbital (spatial part of the spin orbital)} \end{array}$

Relativistic treatment:

One- and two-electron operators

$$h = c\alpha \cdot \mathbf{p} + c^2(\beta - \mathbf{1}_4) + \sum_{\lambda} V_{\mathsf{nuc}}^{\lambda}(\mathbf{r}_{\lambda})$$
 $g = \frac{1}{|\mathbf{r} - \mathbf{r}'|} + \text{corrections}$

 4×4 Dirac matrices $\alpha = (\alpha_x, \alpha_y, \alpha_z)$ and β

Inclusion of corrections in $g \longrightarrow Gaunt$ or Breit operators for electron interaction

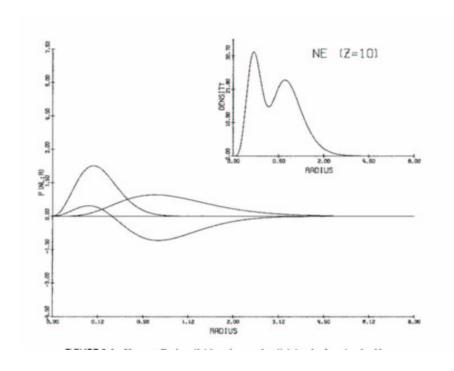
4-component spinor ('single electron function')

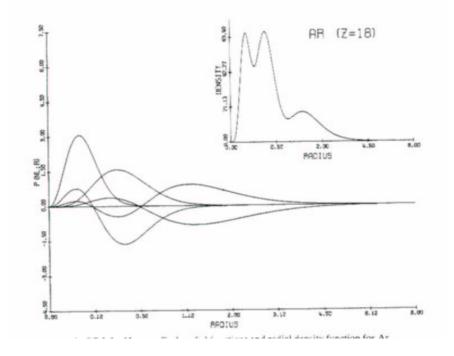
$$\psi_j(\mathbf{x}) = \begin{pmatrix} \psi_{1,j}(\mathbf{r}) \\ \psi_{2,j}(\mathbf{r}) \\ \psi_{3,j}(\mathbf{r}) \\ \psi_{4,j}(\mathbf{r}) \end{pmatrix}$$

Electronic structure of atoms

Spatial part of atomic spin orbitals can be written as

$$\psi_{nlm}(\mathbf{r}) = r^{-1}P_{nl}(r)Y_{lm}(\hat{\mathbf{r}})$$

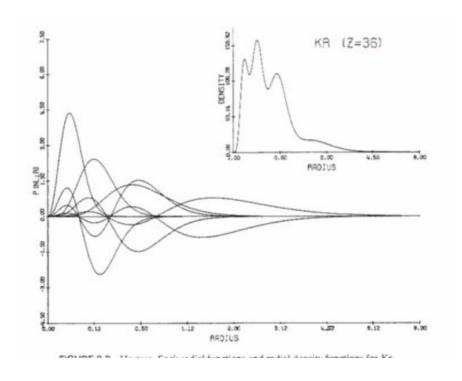


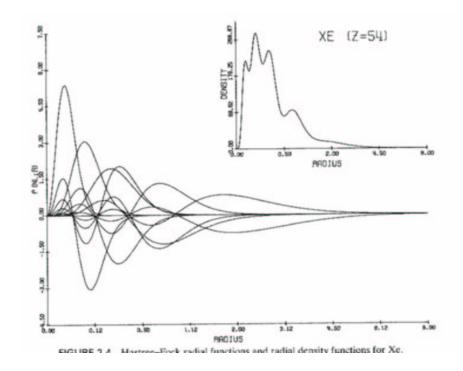


Ne [2He] $2s^2 2p^6 {}^1S$

Ar [$_{10}$ Ne] $3s^2 3p^6 1S$

From: Charlotte Froese Fischer: *The Hartree-Fock method for atoms*. Wiley, New York, 1977

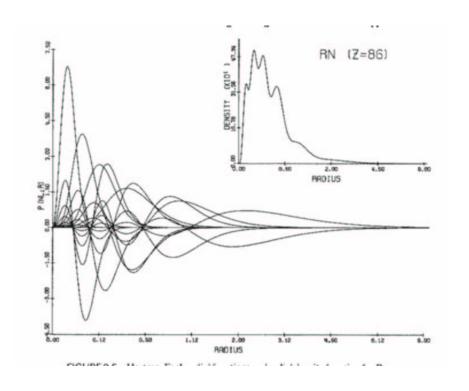




 $\mathrm{Kr}~[_{18}\mathrm{Ar}]~3d^{10}~4s^2~4p^6~^1S$

 ${\rm Xe} \ [{\rm _{36}Kr}] \ 4d^{10} \ 5s^2 \ 5p^6 \ ^1S$

From: Charlotte Froese Fischer: *The Hartree-Fock method for atoms*. Wiley, New York, 1977



Rn [$_{54}$ Xe] $4f^{14}$ $5d^{10}$ $6s^2$ $6p^6$ 1S

From: Charlotte Froese Fischer: *The Hartree-Fock method for atoms*. Wiley, New York, 1977

Chemical experience: Elements from the same group of the periodic table behave similarly in chemical reactions \rightarrow similar 'valence part'

Is a reduction of our equations to a 'valence system' with only

$$n_v = n - \sum_{\lambda}^{K} (Z_{\lambda} - Q_{\lambda})$$

electrons possible (Q_{λ} core charges)? Yes, but not rigorously! For our examples (noble gas atoms): ns^2 np^6 , $n_v = 8$

Can one study the influence of the core on the valence part in atoms, and then transfer this knowledge to the study of molecules?

Yes, with very good accuracy for valence part properties!

Important decision: Which part becomes the 'core', and what remains for the 'valence system'? (Iterative improvement - starting from the chemist's 'valence system' - is possible)

Valence Hamilton operator (in atomic units):

$$\mathcal{H}_v = V_{cc} + \sum_{i=1}^{n_v} h_v(i) + \sum_{i < j}^{n_v} \frac{1}{r_{ij}}$$

$$V_{cc} = \sum\limits_{\lambda<\mu}^K \frac{Q_\lambda Q_\mu}{R_{\lambda\mu}}$$
 — core-core and core-nuclear repulsion

$$h_v(i) = -\frac{1}{2}\nabla_i^2 + U_{cv}(i)$$
 — effective one-electron operator

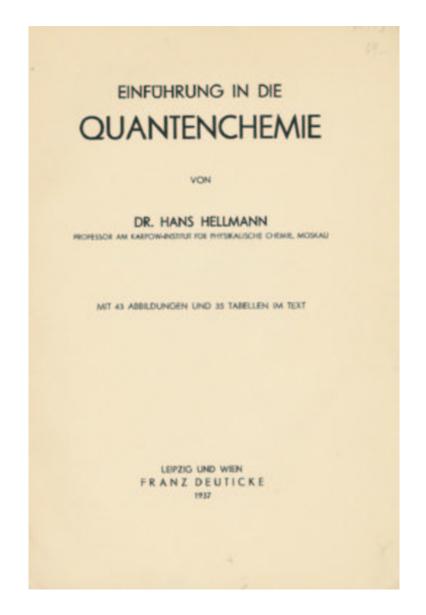
effective core potential (ECP):

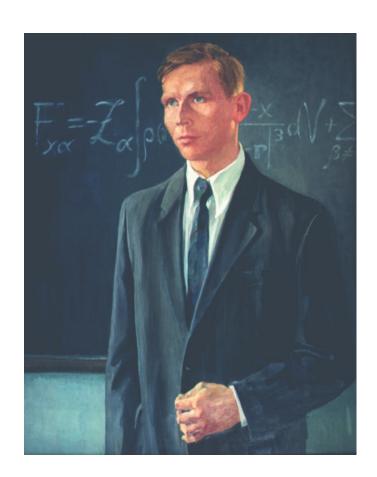
$$U_{cv}(i) = \sum_{\lambda}^{K} \left\{ -\frac{Q_{\lambda}}{r_{\lambda i}} + \Delta U_{cv}^{\lambda}(\mathbf{r}_{\lambda i}) \right\}$$

The idea of an effective core potential

- Г. Гельман / Н. Hellmann (10.06.1934), О комбинированном приближенном расчете проблемы многих электронов / Über ein kombiniertes Störungsverfahren im Vielelektronenproblem, Докл. Акад. Наук СССР / С. R. (Dokl.) Acad. Sci. URSS 4 (1934) 442–444 (in Russian), 444–446 (in German).
- \rightarrow "Zusatzpotential" $\partial T/\partial \varrho$, derived from Thomas-Fermi theory
- H. Hellmann (26.11.1934), A New Approximation Method in the Problem of Many Electrons, J. Chem. Phys. 3 (1935) 61.
- H. Hellmann (17.12.1934; 03.12.1935), Ein kombiniertes Näherungsverfahren zur Energieberechnung im Vielelektronenproblem, *Acta Physicochim. URSS* $\bf 1$ (1934/1935) 913–940; Ein kombiniertes Näherungsverfahren zur Energieberechnung im Vielelektronenproblem. II, *Acta Physicochim. URSS* $\bf 4$ (1936) 225–244.
- \rightarrow dependence on angular momentum (l, not j = l + s)
- H. Hellmann, W. Kassatotschkin (15.03.1936), Metallic Binding According to the Combined Approximation Procedure, *J. Chem. Phys.* **4** (1936) 324–325.
- H. Hellmann, W. Kassatotschkin (16.05.1936), Die metallische Bindung nach dem kombinierten Näherungsverfahren, *Acta Physicochim. URSS* **5** (1936) 23–44.
- \rightarrow analytic expression with parameters adjusted to experimental atomic excitation energies, application to alkali metals
- Г. Гельман (23.10.1936), Квантовая Химия, ОНТИ, Москва, 1937 г.

Hans Hellmann (März 1937), Einführung in die Quantenchemie, Deuticke, Leipzig, 1937.

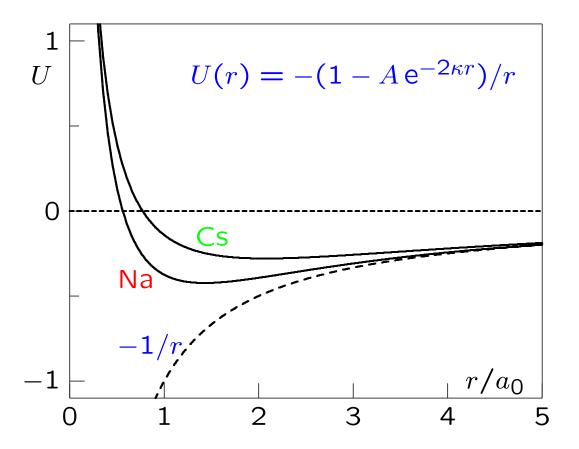




Hans Hellmann (1903–1938)

$$U = V_{\text{core}} + \frac{\partial T}{\partial \varrho} \longrightarrow U(r) = -\frac{1}{r} + \frac{A}{r} e^{-2\kappa r}$$

Portrait of Hans Hellmann painted by Tatyana Livshits, Moscow, according to a photo taken in 1933



Effective core potential U(r), in atomic units, for sodium (Na: A=1.826; $\kappa=0.536$) and cesium (Cs: A=1.672; $\kappa=0.333$), used in this form by Hellmann and Kassatochkin in a study of metallic binding in alkali metals [H. Hellmann, W. Kassatotschkin: Die metallische Bindung nach dem kombinierten Näherungsverfahren. Acta Physicochim. U. R. S. S. 5 (1936) 23–44; parameters adjusted to experimental term energies].

Independent parallel development

Paul Gombás (04.03.1935), Über die metallische Bindung, *Z. Phys.* **94** (1935) 473–488.

ightarrow Thomas-Fermi theory applied to the atomic core, using a representation of the core density given by W. Lenz and H. Jensen

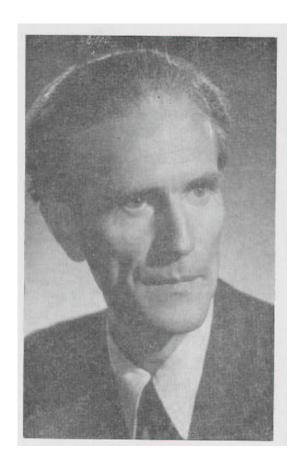
Paul Gombás, Über eine vereinfachte Formulierung der Besetzungsvorschrift der Quantenzustände von Atomen und deren Anwendung zur Bestimmung der Atomterme, *Z. Phys.* **118** (1941/1942) 164–180.

Paul Gombás, Zur Berechnung von Atomtermen, Z. Phys. 119 (1942) 318–324.

Paul Gombás, Pseudopotentiale, Springer, Wien, 1967.

- *** Dependence on angular momentum explicitly included (i. e. $U \rightarrow U_l$)
- *** Development of l-dependent potentials for exchange and correlation





Pál Gombás (1909–1971)

Photograph taken from obituary written by R. Gáspár, Acta Phys. Acad. Sci. Hung. 30 (1971) 109

Early applications of the Hellmann potential

- H. Preuß, Untersuchungen zum kombinierten Näherungsverfahren. Z. Naturforsch. A 10 (1955) 365–373
- K. Ladányi, Zur Theorie der Edelmetalle, *Acta Phys. Acad. Sci. Hung.* **5** (1955) 361–380
- L. Szasz, G. McGinn, Energy-Term Calculations with Hellmann-Type Pseudopotential, J. Chem. Phys. **42** (1965) 2363–2370

Pseudopotential Theory of Atoms and Molecules

Levente Szasz Fordham University L. Szasz, Pseudopotential Theory of Atoms and Molecules, Wiley, New York, 1985

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Search for a theoretical basis

$$\left(-\frac{1}{2}\nabla^2 + U^{(\mathsf{G})\mathsf{PK}}\right)\varphi_k^v = \varepsilon_k \varphi_k^v$$

A complicated operator ($U^{(G)PK}$) appears, which requires full knowledge of the solutions of the all-electron case.

Orbital transformation:

$$\varphi_k^v = \psi_k^v + \sum_{j \in c} c_{jk} \psi_j^c$$

valence orbital (nodes) $\psi_k^v o$ pseudo-valence orbital (nodeless) $arphi_k^v$

- (a) I. Fényes, Csillagászati Lapok (Budapest) 6 (1943) 49 (based on Hartree theory);
- (b) P. Szépfalusy, *Acta Phys. Acad. Sci. Hung.* **5** (1955) 325 & **6** (1956) 273 (based on Hartree-Fock theory);
- (c) E. Antončík, Czech. J. Phys. 7 (1957) 118 & 9 (1959) 291;
- (d) J. C. Phillips, L. Kleinman, Phys. Rev. 116 (1959) 287 ('pseudopotential');
- (e) A. U. Hazi, S. A. Rice, J. Chem. Phys. 45 (1966) 3004;
 J. D. Weeks, A. Hazi, S. A. Rice, Adv. Chem. Phys. 16 (1969) 283



Local operators (multiplicative operator):

$$U(\mathbf{r})$$
 or $U(r)$

'acts' in the same way onto every function determine one function U(r) (as Hellmann did)

Semilocal operators:

$$U^{\lambda} = \sum_{l} U^{\lambda}_{l}(r) \mathcal{P}^{\lambda}_{l}$$
 or $U^{\lambda} = \sum_{j} U^{\lambda}_{j}(r) \mathcal{P}^{\lambda}_{j}$

Projector on angular momentum subspace on center λ , e. g.:

$$\mathcal{P}_l^{\lambda} = \sum_{m=-l}^l |\lambda lm> <\lambda lm| \qquad \left(ext{or:} \quad \mathcal{P}_j^{\lambda} = \sum_{m=-j}^j |\lambda jm> <\lambda jm|
ight)$$

'acts' in the same way for given angular momentum determine one function $U_l(r)$ for every angular momentum l

Nonlocal operators:

$$U^{\lambda} = \sum_{ij} |\chi_i^{\lambda} > C_{ij} < \chi_j^{\lambda}|$$

'acts' differently on different functions choose / determine a suitable set of functions (a basis) $\{\chi_i^{\lambda}(\mathbf{r})\}$

Important ways to determine U_l^{λ}

(a) Inversion of the Schrödinger equation for the (nodeless) pseudoorbital, subsequent fitting to some useful analytic expression (\rightarrow shape-consistent)

$$\left(-\frac{1}{2}\nabla^2 + U_l^{\lambda}\right)\varphi_l = \varepsilon\varphi_l \quad \Rightarrow \quad U_l^{\lambda}(\mathbf{r}) = \varepsilon + \frac{1}{2}\frac{\varphi_l''}{\varphi_l} \approx \sum_k c_k f_k(\alpha_k, \mathbf{r})$$

(b) Direct determination or optimization (e. g. via least squares) of the parameters in an analytic expression for U_l^{λ} with respect to some reference data

$$U_l^{\lambda}(\mathbf{r}) = \sum_k c_k f_k(\alpha_k, \mathbf{r}) \quad \Rightarrow \quad S = \sum_i w_i \left(\langle \mathcal{O}^{\mathsf{ECP}} \rangle_i - \langle \mathcal{O}^{\mathsf{ref}} \rangle_i \right)^2 = \min$$

$$\Rightarrow \quad \{c_k, \alpha_k\}$$

- $<\mathcal{O}^{\mathsf{ECP}}>_i$ value of property $\mathcal O$ evaluated with ECP for state i
- $<\mathcal{O}^{\mathsf{ECP}}>_i$ reference value of that property for state i

Shape-consistent pseudopotentials:

$$U^{\lambda}(\mathbf{r}) = U_{L}^{\lambda}(r) + \sum_{l=0}^{L-1} \left(U_{l}^{\lambda}(r) - U_{L}^{\lambda}(r) \right) \mathcal{P}_{l}^{\lambda}$$

$$U_{L}^{\lambda}(r) = \sum_{k} A_{k} r^{n_{k}} \exp\left(-\alpha_{k} r^{2}\right)$$

$$U_{l}^{\lambda}(r) - U_{L}^{\lambda}(r) = \sum_{k} B_{k} r^{m_{k}} \exp\left(-\beta_{k} r^{2}\right)$$

keep the resulting pseudo-valence orbital φ^v identical with the valence orbital ψ^v beyond some matching radius: $\varphi^v(r) \equiv \psi^v(r)$ for $r \geq R_c$ determine parameters through inversion of a Schrödinger-type equation

- (a) W. A. Goddard III, Phys. Rev. 174 (1968) 659
- (b) Ph. Durand, J.-C. Barthelat, Theor. Chim. Acta 38 (1975) 283
- (c) L. R. Kahn, P. Baybutt, D. G. Truhlar, J. Chem. Phys. 65 (1976) 3826
- (d) P. A. Christiansen, Y. S. Lee, K. S. Pitzer, J. Chem. Phys. 71 (1979) 4445
- (e) Hay / Wadt Christiansen / Ermler / Pitzer Stevens / Basch / Krauss / Jasien / Cundari

available e. g. from URL http://www.clarkson.edu/~pac/reps.html

Generalized Relativistic Effective Core Potentials (GRECP):

$$U = U_{n_{v}LJ}(r) + \sum_{l=0}^{L} \sum_{j=|l-1/2|}^{j=l+1/2} \left[U_{n_{v}lj}(r) - U_{n_{v}LJ}(r) \right] \mathcal{P}_{lj}$$

$$+ \sum_{n_{c}} \sum_{l=0}^{L} \sum_{j=|l-1/2|}^{j=l+1/2} \left\{ \left[U_{n_{c}lj}(r) - U_{n_{v}lj}(r) \right] \mathcal{P}_{n_{c}lj} \right.$$

$$+ \mathcal{P}_{n_{c}lj} \left[U_{n_{c}lj}(r) - U_{n_{v}lj}(r) \right] \right\}$$

$$- \sum_{n_{c}} \sum_{l=0}^{L} \sum_{j=|l-1/2|}^{j=l+1/2} \mathcal{P}_{n_{c}lj} \left[\frac{U_{n_{c}lj}(r) + U_{n'_{c}lj}(r)}{2} - U_{n_{v}lj}(r) \right] \mathcal{P}_{n'_{c}lj}$$

two new terms added to the semi-local form already known requires a technique to invert the Schrödinger equation for valence orbitals with nodes the various radial parts, which appear above, are fitted to analytic expressions uses projection operators onto outer core pseudospinors

- (a) A. V. Titov, A. O. Mitrushenkov, I. I. Tupitsyn, Chem. Phys. Lett. 185 (1991) 330;
- (b) N. S. Mosyagin, A. V. Titov, A. V. Tulub, Phys. Rev. A 50 (1994) 2239;
- (c) A. V. Titov, N. S. Mosyagin, Int. J. Quantum Chem. 71 (1999) 359

available from URL http://qchem.pnpi.spb.ru/publication/recp.html for Ba, Yb, Hg-Rn

Energy-consistent pseudopotentials:

$$U^{\lambda}(\mathbf{r}) = U_{L}^{\lambda}(r) + \sum_{l=0}^{L-1} \left(U_{l}^{\lambda}(r) - U_{L}^{\lambda}(r) \right) \mathcal{P}_{l}^{\lambda}$$

$$U_{L}^{\lambda}(r) = \sum_{k} A_{k} r^{n_{k}} \exp\left(-\alpha_{k} r^{2}\right)$$

$$U_{l}^{\lambda}(r) - U_{L}^{\lambda}(r) = \sum_{k} B_{k} r^{m_{k}} \exp\left(-\beta_{k} r^{2}\right)$$

Determine parameters through optimization w. r. t. observable quantities (e. g. energies, as Hellmann did — minimize least squares sum)

Reference data may come from many different sources (e. g. experiment, or various theoretical levels — make an exclusive choice)

Dolg / Schwerdtfeger / Wedig / Stoll / Preuß

available from URL http://www.theochem.uni-stuttgart.de/pseudopotentials

Error analysis for semilocal ECPs

The modern ECP approach relies heavily on error compensation. Some important errors involved are

- (a) frozen core approximation (implicit or explicit assumption of a 'frozen core')
- (b) core valence separation (core and valence part are considered as independent subsystems)
- (c) introduction of effective one-electron operators, replacing in part oneand two-electron operators from the all-electron approach
- (d) (pseudo) orbital transformation (removal of nodal structure of valence orbitals [not for model potentials])

For further details see

- (a) W. H. E. Schwarz, *Theor. Chim. Acta* **11** (1968) 307, 377, *Acta Phys. Acad. Sci. Hung.* **27** (1969) 391
- (b) T. C. Chang, P. Habitz, B. Pittel, W. H. E. Schwarz, Theor. Chim. Acta 34 (1974) 263
- (c) B. Pittel, W. H. E. Schwarz, Chem. Phys. Lett. 46 (1977) 121
- (d) T. C. Chang, P. Habitz, W. H. E. Schwarz, *Theor. Chim. Acta* 44 (1977) 61

ab initio model potentials (AIMP — non-local ECPs)

$$-\sum_{\lambda}^{K} \frac{Z_{\lambda}}{r_{\lambda i}} + \sum_{j \in c} \left\{ 2\mathcal{J}_{j}(i) - \mathcal{K}_{j}(i) \right\} = \sum_{\lambda}^{K} \left\{ -\frac{Q_{\lambda}}{r_{\lambda i}} + \Delta U_{cv}^{\lambda}(\mathbf{r}_{\lambda i}) \right\}$$

approximate $\Delta U_{cv}^{\lambda}(\mathbf{r}_{\lambda i})$ by a sum of semilocal and nonlocal terms with adjustable parameters (for every core λ)

$$-\frac{Z_{\lambda}}{r_{\lambda i}} + 2\sum_{c \in \lambda} \mathcal{J}_{c}(i) - \sum_{c \in \lambda} \mathcal{K}_{c}(i) = -\frac{Q_{\lambda}}{r_{\lambda i}} + \sum_{k} c_{k}^{\lambda} r_{\lambda i}^{n_{k}^{\lambda}} \exp\left(-\alpha_{k}^{\lambda} r_{\lambda i}^{2}\right) + \sum_{p,q} |\chi_{p}^{\lambda} > A_{pq}^{\lambda} < \chi_{q}^{\lambda}| + \sum_{c \in \lambda} |\psi_{c}^{\lambda} > B_{c}^{\lambda} < \psi_{c}^{\lambda}|$$

no orbital transformation required — nodal structure of valence orbitals retained S. Huzinaga, A. A. Cantu, *J. Chem. Phys.* **55** (1971) 5543; V. Bonifacić, S. Huzinaga, *J. Chem. Phys.* **60** (1974) 2779; Y. Sakai, S. Huzinaga, *J. Chem. Phys.* **76** (1982) 2537; M. Klobukowski, *J. Comput. Chem.* **4** (1983) 350; S. Huzinaga, L. Seijo, Z. Barandiarán, M. Klobukowski, *J. Chem. Phys.* **86** (1987) 2132; Z. Barandiarán, L. Seijo, *J. Chem. Phys.* **101** (1994) 4049; C. M. Marian, U. Wahlgren, *Chem. Phys. Lett.* **251** (1996) 357; L. Seijo, Z. Barandiarán, E. Harguindey, *J. Chem. Phys.* **114** (2001) 118

available from URL http://www.uam.es/quimica/aimp and http://www.thch.uni-bonn.de/tc/TCB.download.html

Additional refinements

Correction for point-like core-core interaction

Core polarization potentials

Spin-orbit interaction potentials

Correction for point-like core-core interaction

If necessary (e. g. when large-core ECPs are used), one can replace

$$V_{cc} = \sum_{\lambda < \mu}^{K} \frac{Q_{\lambda} Q_{\mu}}{R_{\lambda \mu}} \quad \to \quad V_{cc} = \sum_{\lambda < \mu}^{K} \left\{ \frac{Q_{\lambda} Q_{\mu}}{R_{\lambda \mu}} + \Delta V_{cc}^{\lambda \mu} (\mathbf{R}_{\lambda \mu}) \right\}$$

and parameterize the correction terms, e. g.

$$\Delta V_{cc}^{\lambda\mu}(\mathbf{R}_{\lambda\mu}) = B_{\lambda\mu} \exp\left(-b_{\lambda\mu}R_{\lambda\mu}\right)$$

(for each pair of centers (λ, μ)).

Core polarization potentials

To account for (i) core polarization and (ii) (part of) core-valence correlation, one may add:

$$V_{cpp} = -\frac{1}{2} \sum_{\lambda} \alpha_{\lambda} \mathbf{f}_{\lambda}^{2}$$

 α_{λ} — electric dipole polarizability of core λ

 f_{λ} — electric field at core λ , due to all other cores or nuclei, and all valence electrons (has to include a short-range cutoff)

⁽a) W. Müller, J. Flesch, W. Meyer, J. Chem. Phys. 80 (1984) 3297;

⁽b) P. Fuentealba, Thesis, Univ. Stuttgart, 1984

Spin-orbit interaction potentials

$$V_{SO}^{\lambda}(i) = \sum_{l=1}^{l_{\text{max}}} \frac{2}{2l+1} \Delta U_l^{\lambda}(r_{\lambda i}) \mathcal{P}_l^{\lambda} \mathbf{l}_i \cdot \mathbf{s}_i \mathcal{P}_l^{\lambda}$$
$$\Delta U_l^{\lambda}(r) = U_{l,j=l+1/2}^{\lambda}(r) - U_{l,j=l-1/2}^{\lambda}(r)$$

- (a) P. Hafner, W. H. E. Schwarz, Chem. Phys. Lett. 65 (1979) 537;
- (b) P. A. Christiansen, Y. S. Lee, K. S. Pitzer, J. Chem. Phys. 71 (1979) 4445;
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- M. Dolg, in: J. Grotendorst (ed.), Modern Methods and Algorithms of Quantum Chemistry (NIC Series, Vol. 1), John von Neumann Institute for Computing, Jülich, 2000, pp. 479–508 [2nd ed. (NIC Series, Vol. 3), pp. 507–540]
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For earlier reviews see references given in the reviews above.

Pseudopotentials in solid state science

IN THE THEORY OF METALS

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W. A. Harrison, Pseudopotentials in the Theory of Metals, Benjamin, New York, 1966

based on the work by:

J. C. Phillips, L. Kleinman, *Phys. Rev.* 116 (1959) 287; M. H. Cohen, V. Heine, *Phys. Rev.* 122 (1961) 1821; V. Heine, I. V. Abarenkov, *Philos. Mag.* 9 (1964) 451 & 12 (1965) 529

W. A. BENJAMIN, INC., Rending, Massachusett

Non-local pseudopotentials, designed to be used in plane-wave DFT calculations for periodic systems (crystals) – general form

$$U^{\lambda} = U^{\lambda}_{\text{loc}} + \sum_{ij} |\chi^{\lambda}_i > B^{\lambda}_{ij} < \chi^{\lambda}_j| \quad \text{or} \quad U^{\lambda} = \sum_{ij} |\chi^{\lambda}_i > B^{\lambda}_{ij} < \chi^{\lambda}_j|$$

The 'smoother' (or 'softer') the pseudopotential, the better (faster) the convergence of the plane wave expansions. The 'lack' of deep potential wells, due to introduction of the pseudopotential, was already recognized by Hellmann (see the German ed. of his book, Fig. 4, p. 39)

Norm-conserving pseudopotentials:

- (a) D. R. Hamann, M. Schlüter, C. Chiang, Phys. Rev. Lett. 43 (1979) 1494
- (b) L. Kleinman, D. M. Bylander, *Phys. Rev. Lett.* **48** (1982) 1425
- (c) G. B. Bachelet, D. R. Hamann, M. Schlüter, *Phys. Rev. B* **26** (1982) 4199
- (d) D. Vanderbilt, Phys. Rev. B 32 (1985) 8412
- (e) N. Troullier, J. L. Martins, Solid State Commun. 74 (1990) 613
- (f) X. Gonze, R. Stumpf, M. Scheffler, *Phys. Rev. B* 44 (1991) 8503;
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- (g) S. Goedecker, M. Teter, J. Hutter, *Phys. Rev. B* **54** (1996) 1703;
 C. Hartwigsen, S. Goedecker, J. Hutter, *Phys. Rev. B* **58** (1998) 3641

(Ultra)soft pseudopotentials:

D. Vanderbilt, *Phys. Rev. B* **41** (1990) 7892

see also PWScf, URL http://www.pwscf.org, for solid state DFT calcs. using plane wave basis, S. Baroni et al., Trieste, Italy

Recent extensions of the ECP idea

Effective group (or fragment) potentials (EGP / EFP)

Instead of 'replacing' only an atomic core by an effective one-electron operator (ECP), a fragment of the molecular system (an inactive group of atoms, or an inactive part of the electronic structure) is 'replaced' by an effective potential: the effective fragment (or group) potential

Some references (incomplete):

- (a) K. Ohta, Y. Yoshioka, K. Morokuma, K. Kitaura, *Chem. Phys. Lett.* **101** (1983) 12 (effective fragment potential)
- (b) M. v. Arnim, S. D. Peyerimhoff, Theor. Chim. Acta 87 (1993) 41;
 M. v. Arnim, S. D. Peyerimhoff, Chem. Phys. Lett. 210 (1993) 488
 (effective potential for spectator groups)
- (c) J. A. M. Romero, J. F. Sanz, J. Chem. Phys. 99 (1993) 1255 (group model potential)
- (d) D. Andrae, Thesis, Univ. Stuttgart, 1994 (energy-consistent approach)
- (e) F. Alary, R. Poteau, J.-L. Heully, J.-C. Barthelat, J.-P. Daudey, *Theor. Chem. Acc.* **104** (2000) 174 (effective group potentials)

Embedding potentials

Instead of 'replacing' only an atomic core by an effective one-electron operator (ECP), the complete surrounding environment of a system (in a crystal) is 'replaced' by an effective potential: the embedding potential

→ see presentation given by L. Seijo

Summary

Effective core potentials (ECPs)

- a tool to study systems involving heavy atoms
- only a 'valence part' is treated explicitly (\neq the chemist's 'valence system')
- not (too) much loss of accuracy in valence properties
- most important relativistic effects (scalar relativistic effects) included in formally non-relativistic approaches
- not uniquely defined many varieties exist