

The all-seeing eye of Auger electron spectroscopy: a study on aqueous KCl

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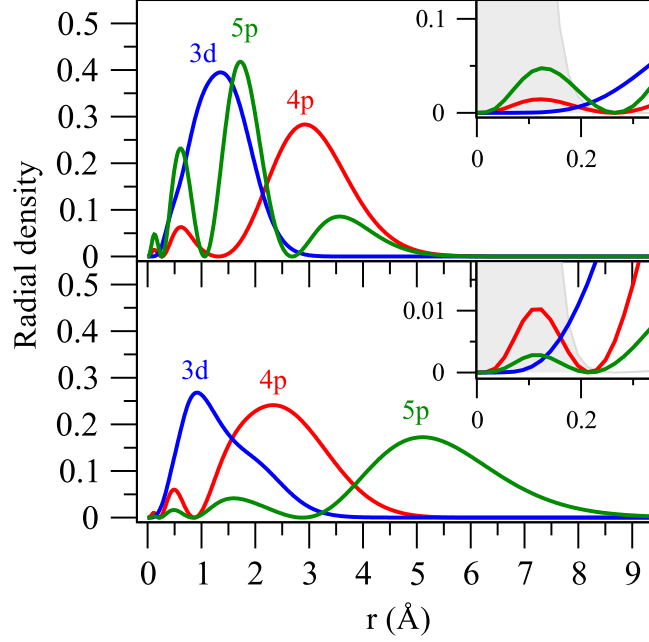


Figure 1: Radial density distributions of the singly-occupied natural orbital occupied by the excited electron corresponding to the $1s \rightarrow 4p$, $1s \rightarrow 3d$ and $1s \rightarrow 5p$ core excitations in K^+ (lower panel) and Cl^- (upper panel). The insets show the region of distances relevant for the overlap with the $1s$ core orbital whose radial density is shown as a grey shaded area.

References

- (1) Föhlisch, A.; Feulner, P.; Hennies, F.; Fink, A.; Menzel, D.; Sanchez-Portal, D.; Echenique, P. M.; Wurth, W. *Nature* *436*, 373.

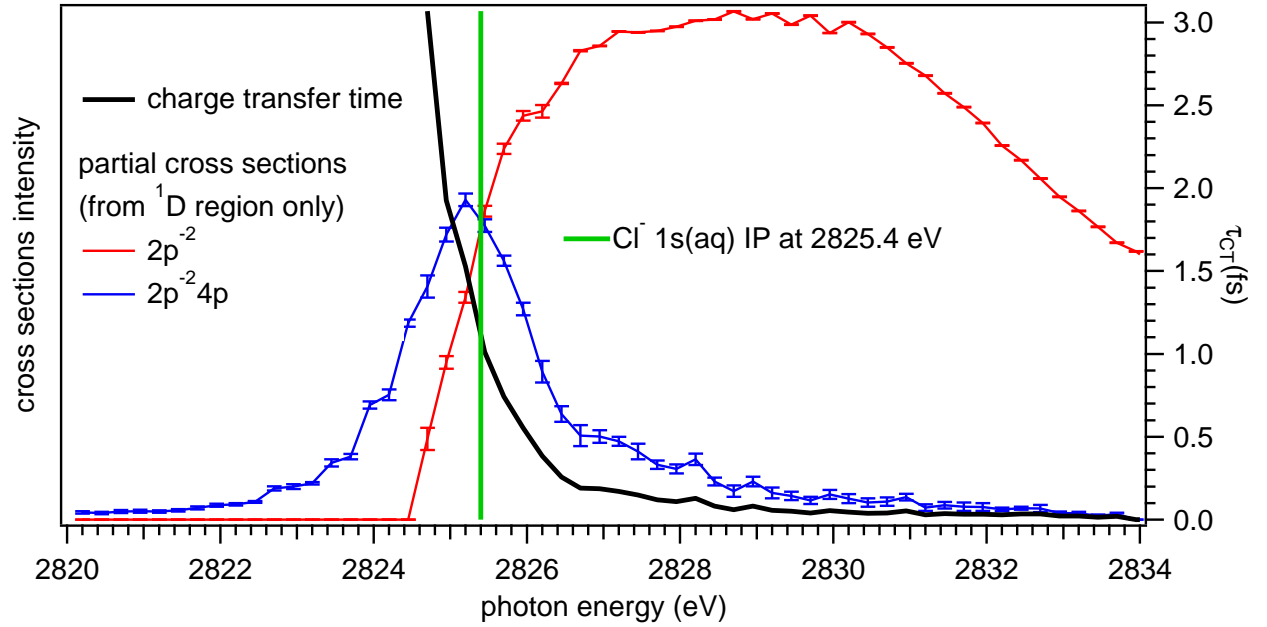


Figure 2: Partial cross sections and charge transfer time extracted from Fig. ???. The blue and red curves are obtained by integrating the area of the $2p^{-2}$ and $2p^{-2}4p$ final states (1D state region only) at each photon energy step. From these curves we determine the charge transfer time τ_{CT} according to the formula $\tau_{CT} = \tau l/d$, with τ being the Cl 1s core-hole lifetime and l/d being the intensity ratio of the localized ($2p^{-2}4p$) and delocalized ($2p^{-2}$) states at a given excitation energy.¹ The green line defines the $Cl_{aq}^{-}(1s)$ ionization potential.