

The electronic structure of aqueous KCl revealed by X-ray absorption and Auger electron spectroscopy

The all-seeing eye of Auger electron spectroscopy: a study on aqueous KCl

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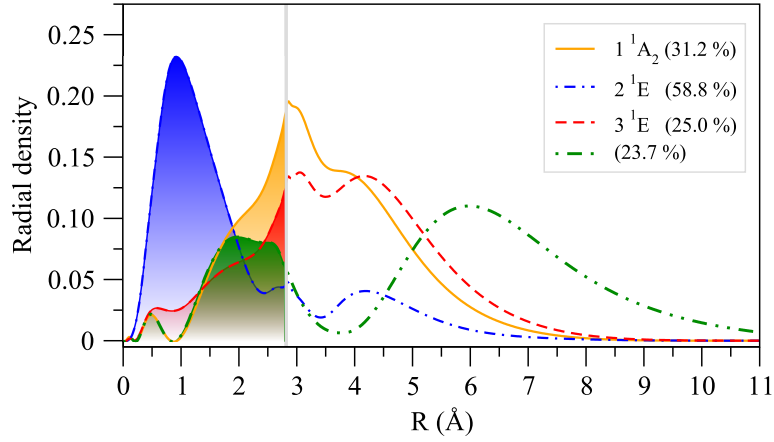


Figure 1: Radial density distributions of the singly-occupied natural orbitals occupied by the excited electron in the core excited states 3609.11 eV (1^1A_2), 3609.41 eV (2^1E) and 3609.64 eV (3^1E) of $K^+(H_2O)_6$. The grey line at 2.816 Å represents the equilibrium K-O distance.

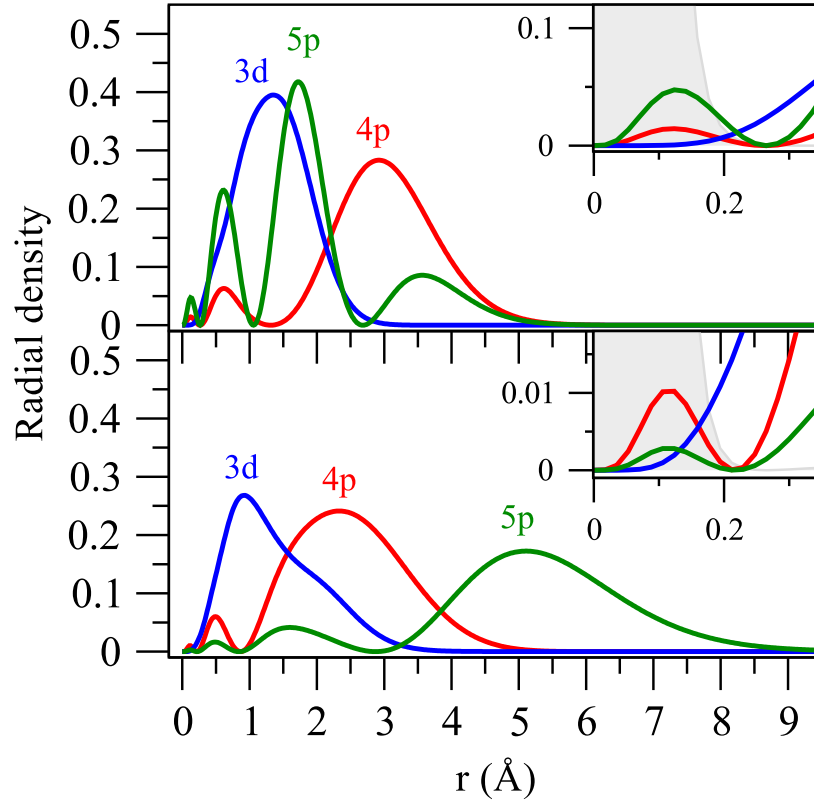


Figure 2: Radial density distributions of the singly-occupied natural orbital occupied by the excited electron corresponding to the $1s \rightarrow 4p$, $1s \rightarrow 3d$ and $1s \rightarrow 5p$ core excitations in K^+ (lower panel) and Cl^- (upper panel). The insets show the region of distances relevant for the overlap with the $1s$ core orbital whose radial density is shown as a grey shaded area.