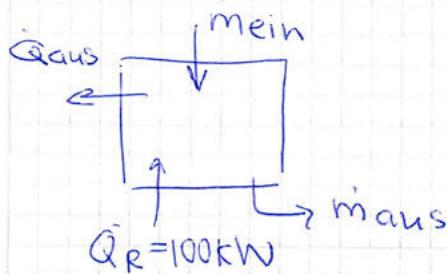


# Aufgabe 1

a) ges  $\dot{Q}_{\text{aus}}$



$$\dot{Q} = m(\dot{h}_{\text{ein}} - \dot{h}_{\text{aus}}) + \dot{Q}$$

$\dot{h}_{\text{ein}} (70^\circ\text{C})$

$$\dot{h}_{\text{ein}} - \dot{h}_{\text{aus}} = c_{\text{if}}(T_{\text{ein}} - T_{\text{aus}}) + v_{\text{it}}(P_{\text{ein}} - P_{\text{aus}})$$

$$\dot{h}_{\text{ein}} - \dot{h}_{\text{aus}} = c_{\text{if}}(288.15\text{K} - 298.15\text{K})$$

$$\text{b)} \bar{T} = \frac{\int_s^a T dS}{S_a - S_e} = h$$

$$\text{b)} \bar{T} = \frac{\int_s^a T dS}{S_a - S_e} = \frac{h_a - h_e}{S_a - S_e} \quad \text{da reversibler stationärer Prozess}$$

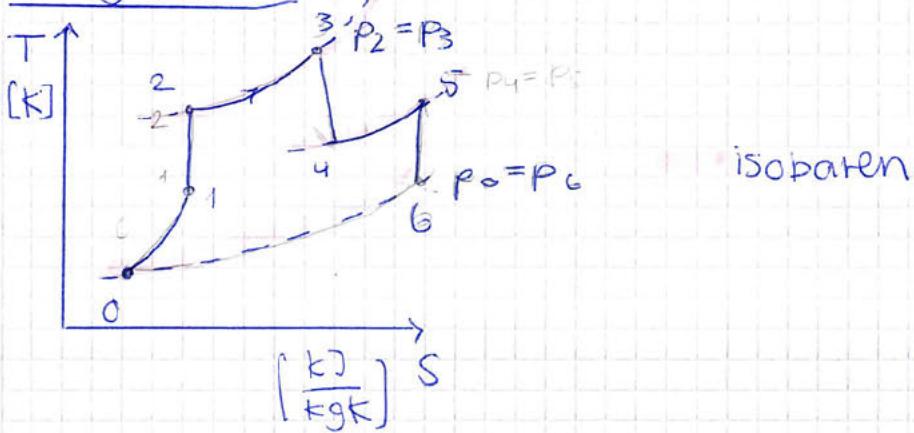
$$\dot{h}_{\text{ein}} - \dot{h}_{\text{aus}} = c_{\text{if}}(T_{\text{ein}} - T_{\text{aus}}) + v_{\text{it}}(P_{\text{ein}} - P_{\text{aus}})$$

$$\text{c)} \dot{Q} = m_{\text{ein}}(S_{\text{ein}} - S_{\text{aus}}) + \sum \frac{Q_j}{T_j} + S_{\text{erz}}$$

$$\text{d)} \frac{dE}{dt} = \dot{m}_w(\dot{h}_{\text{ein}} - \dot{h}_{\text{aus}}) + \sum Q - W$$

$$\text{e)} \Delta S_{12} = n_1(S_1 - m_2 S_2) - m_1 S_1 = \sum \frac{Q_j}{T_j} + S_{\text{erz}}$$

## Aufgabe 2

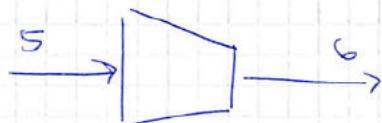


isobaren

b)  $m_6 = p_6 A_6 W_6$

$$\frac{T_6}{T_5} = \left( \frac{p_6}{p_5} \right)^{\frac{n-1}{n}}$$

$$T_6 = 328.075 \text{ K}$$



$$p = \frac{1}{V_6}$$

$$p_6 V_6 = m_6 R T_6$$

$$R = c_p - \frac{c_p}{n} = 0.287 \frac{\text{kJ}}{\text{kg K}}$$

c)  $\Delta_{\text{ex,str}} = m_{\text{ges}} (h_6 - h_0 - T_0 (s_6 - s_0))$

$$h_6 - h_0 = c_p (T_6 - T_0) = 97.4311 \frac{\text{kJ}}{\text{kg}}$$

$$T_6 = 340 \text{ K} \quad T_0 = -30^\circ\text{C} = 243.15 \text{ K}$$

$$s_6 - s_0 = c_p \ln \left( \frac{T_6}{T_0} \right) - R \ln \left( \frac{p_6}{p_0} \right)$$

$$= 0.3373 \frac{\text{kJ}}{\text{kg K}}$$

$$\Delta_{\text{ex,str}} =$$

gehört zusammen

b)  $\left( \frac{T_6}{T_5} \right) = \left( \frac{V_6}{V_5} \right)^{n-1}$

$$\sqrt[n-1]{\frac{T_6}{T_5}} V_5 = V_6 =$$

$$p_6 V_6 = R T_6$$

$$V_6 = \frac{R T_6}{p_6} = 4.93 \frac{\text{m}^3}{\text{kg}}$$

$$m_6 = \frac{1}{V_6} A_6 W_6$$

$$\frac{m_6 V_6}{A_6} = W_6$$

$$a) O = \text{mges} \Delta \text{exstr}_{6,0} + \Sigma - W - \bar{E}_{\text{verl}}$$

$$W_{ob} =$$

$$\bar{E}_{\text{verl}} = \text{mges} \Delta \text{exstr}_{6,0} - W$$

### Aufgabe 3

a) ges:  $p_{g1}$   $m_g$

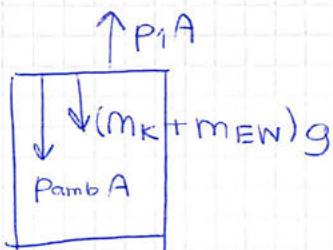
$$\frac{KJ}{kgK} \quad \frac{kmol}{kg}$$

$$T_{g1} = 500^\circ C = 773.15 K$$

$$V_g = 3.14 L = 0.00314 m^3$$

$$R_g = \frac{R}{M} = 0.1663 \frac{KJ}{kgK}$$

$$p_{g1} V_{g1} = m_g R_g T_1$$



$$h = 5.70 m$$

$$A = 5^2 \pi = 0.00785 m^2$$

$$p_{amb} A + (m_k + m_{EW}) g = p_1 A$$

$$p_{amb} + \frac{g}{A} (m_k + m_{EW}) = p_1 = 1.4 \text{ bar} = p_{g1}$$

$$\frac{p_{g1} V_{g1}}{R_g T_1} = 0.003142 \frac{KJ}{kgK} = m_g$$

b) ges:  $T_{g2}$ ,  $p_{g2}$

$p_{g2} = p_{g1} = 1.4 \text{ bar}$ , da sich das Gewicht des EW nicht verändert.

$$p_{g2} V_{g2} = m_2 R T_2$$

$$\frac{p_{g2} V_{g2}}{m_2 R} = \frac{1.4 \text{ bar} \cdot 3.14 L}{3.42 g \cdot 0.1663 \frac{KJ}{kgK}} = 7$$

$$c) \Delta E = Q_{12} - W_{12} \quad W_{12} = \int p_{12} dV = p_{12} (V_2 - V_1)$$

Systemgrenze um das Gas.

$$\Delta E_{12} = m(u_2 - u_1)$$

$$u_2 - u_1 = c_v (T_2 - T_1) = 0.633 (0.003 - 500) = -316.498 \frac{KJ}{kg}$$

$$\kappa = \frac{c_p}{c_v} = \frac{R + c_v}{c_v} = 1.263$$

$$W_{12}^{rev} = \frac{R(T_2 - T_1)}{1 - \kappa} = -112.822 \frac{KJ}{kg}$$

$$n = \kappa$$

$$W_{12}^{rev} = m \omega_{12}^{rev} = -0.386 \frac{KJ}{kg}$$

$$\Delta E_{12} = m(u_2 - u_1) = Q_{12} - W_{12}$$

$$u_2 - u_1 = -316.498 \frac{\text{kJ}}{\text{kg}}$$

$$W_{12} = -0.386 \text{ kJ}$$

$$m(u_2 - u_1) + W_{12} = Q_{12} = -1468 \text{ kJ}$$

$$m = 3.42 \text{ g}$$

a)  $T_{g2} = T_{2\text{EW}} = 0.003^\circ\text{C}$

$$\frac{u_2 - u_{\text{fest}}}{u_{\text{flüssig}} - u_{\text{fest}}} = x$$

$u_{\text{fest}}$

TAB1

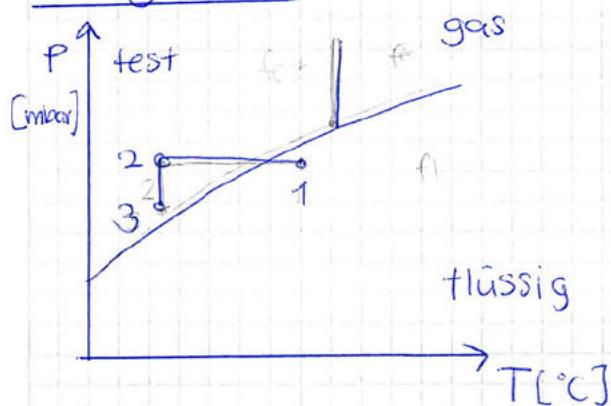
$$u_{\text{fest}}(0.003^\circ\text{C}) \stackrel{!}{=} -333.442 \frac{\text{kJ}}{\text{kg}}$$

$$u_{\text{flüssig}}(0.003^\circ\text{C}) = -0.033 \frac{\text{kJ}}{\text{kg}}$$

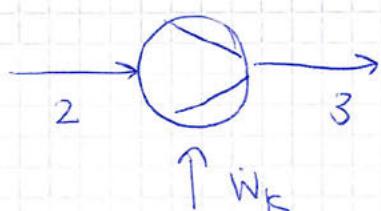
$u_2$

$87$

#### Aufgabe 4



b)



$$0 = \dot{m} (h_2 - h_3) - \dot{W}_k \quad \frac{\dot{W}_k}{h_2 - h_3} = \dot{m}$$

$$\underline{h_2 = h_3} \approx h_2 \quad h_3(8\text{bar}) =$$

$$s_2 = s_{\text{B}}$$

$$s_3 \quad \text{TAB 12}$$

$$s_3 \text{ superheated} \quad s_3(8\text{bar}) =$$

$$c) \quad x = \frac{h_1 - h_f}{h_g - h_f}$$

$$d) \quad \varepsilon_k = \frac{|Q_{zul}|}{|\dot{W}_k|} = \frac{|Q_{zul}|}{|Q_{ab} - Q_{zul}|}$$

e) Die Temperatur würde sich immer weiter senken, bis sie irgendwann stagniert, da keine Wärme mehr entzogen werden kann.

