

$$1a) \dot{Q}_{\text{aus}} = \dot{Q}_R - \dot{m}_{\text{ein}} (h_{\text{aus}} - h_{\text{ein}})$$

$$= \dot{Q}_R - \dot{m}_{\text{ein}} = 100 \text{ kW} - \frac{42.18 \text{ kW}}{0.3 \text{ kg/s}} = 57.82 \text{ kW}$$

$$h_{\text{ein}} = 1267.0 \frac{\text{kJ}}{\text{kg}}$$

$$h_{\text{aus}} = 1107.6 \frac{\text{kJ}}{\text{kg}}$$

b) \bar{T}

$$0 = \dot{m} (s_e - s_a) + \frac{\dot{Q}}{\bar{T}}$$

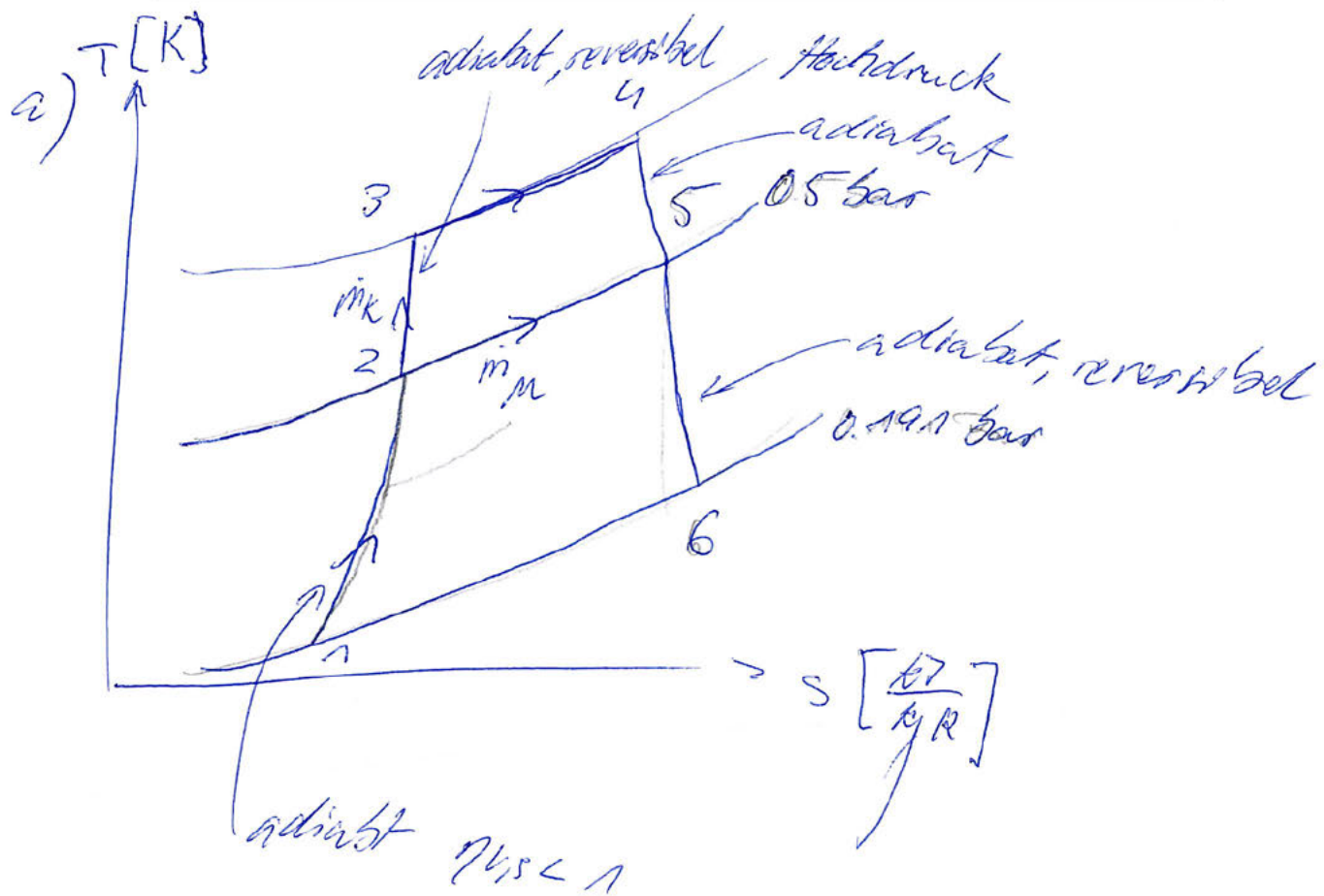
$$\frac{\dot{Q}}{\bar{T}} = \dot{m} (s_a - s_e)$$

$$\bar{T} = \frac{\dot{Q}}{\dot{m} (s_a - s_e)}$$

$$\bar{T} = \frac{\int_{s_e}^{s_a} T ds}{s_a - s_e} = \frac{T_2 - T_1}{\ln\left(\frac{T_2}{T_1}\right)}$$

$$= 293.12 \text{ K}$$

$$c) s_{\text{erz}} = \frac{\dot{Q}}{\bar{T}} = \frac{65 \text{ kW}}{295 \text{ K}} = 0.22 \frac{\text{W}}{\text{K}}$$



b)

$$T_6 = T_5 \left(\frac{P_6}{P_1} \right)^{\frac{1.4-1}{1.4}} = 328.075 \text{ K}$$

~~$\Delta KE =$~~

$$0 = \Delta m_i \left(h_i + \frac{w_i^2}{2} \right)$$

$$w = \sqrt{2(h_6 - h_5)}$$

$$KE = W = \int_5^6 p dV$$

$$\frac{w^2}{2} = \frac{1}{\gamma - 1} (T_2 - T_1)$$

$$\frac{w^2}{2} = c_p \cdot \Delta T = 92.95 \frac{\text{kJ}}{\text{kg}}$$

$$w^2 = 18490 \frac{\text{kJ}}{\text{kg}} \quad w_6 = 136 \frac{\text{m}}{\text{s}}$$

c)

$$\Delta e_{x, str} = e_{x, str, 6} - e_{x, str, 0}$$

$$\begin{aligned} \bar{r}_{luft} &= \frac{1.874 \frac{\text{J}}{\text{mol K}}}{28.97 \frac{\text{g}}{\text{mol}}} \\ &= 257.02 \end{aligned}$$

$$\Delta e_{x, str} = h - h_0 - T_0 (s - s_0) + ke$$

$$\Delta e_{x, str, 6} = h_6 - h_0 - T_0 (s_6 - s_0) + ke_6 - ke_0$$

$$= c_p \cdot \Delta T_{6,0} - T_0 c_p \ln\left(\frac{T_6}{T_0}\right) - T_0 R \ln\left(\frac{P_6}{P_0}\right) \quad \text{if } P_6 = P_0$$

~~$$= 1006 \cdot (340 - 273.15) - (273.15) \left(1006 \ln\left(\frac{340}{273.15}\right) + 287.02 \ln\left(\frac{0.191}{0.5}\right) \right)$$~~

~~$$= 67.126 \frac{\text{kJ}}{\text{kg}}$$~~

$$= c_p \left(\Delta T_{6,0} - T_0 \ln\left(\frac{T_6}{T_0}\right) \right) + ke$$

$$= 1006 \frac{\text{kJ}}{\text{kg K}} \left(340 \text{ K} - (273.15 - 30) \text{ K} - (273.15 - 30) \text{ K} \ln\left(\frac{340}{273.15 - 30}\right) \right)$$

$$= 15.42 \frac{\text{kJ}}{\text{kg}} + \frac{\left(510 \frac{\text{m}}{\text{s}} \right)^2}{2} = 115.47 \frac{\text{kJ}}{\text{kg}}$$

d)

$$e_{x, verl} = e_{x, str, 06} - e_{x, 9b} - ke = 100 \frac{\text{kJ}}{\text{kg}} + 969.582 \frac{\text{kJ}}{\text{kg}} - 130.05 \frac{\text{kJ}}{\text{kg}}$$

$$e_{x, 9b} = \left(1 - \frac{T_0}{T} \right) Q = 969.582 \frac{\text{kJ}}{\text{kg}} = 969.58 \frac{\text{kJ}}{\text{kg}}$$

$$ke = 130.05 \frac{\text{kJ}}{\text{kg}}$$

3a)

 $P_{g,1}$ m_g

$$m_g = \frac{pV}{RT}$$

$$p = \bar{p} = 166.3 \text{ Pa}$$

$$P_{g,1} = \frac{F_G}{A} + p_{\text{atmosph}} = 140094 \text{ Pa} \approx 14 \text{ bar}$$

$$\bar{F} = (m_R + m_{EW}) \cdot g$$

$$A = (0.05 \text{ m})^2 \pi = 0.00785 \text{ m}^2$$

$$m_g = \frac{140094 \text{ Pa} \cdot 3.14 \cdot 10^{-3} \text{ m}^3}{\frac{8314 \text{ J/molK}}{50 \text{ g/mol}} \cdot (273.15 + 500) \text{ K}} = 3.422 \text{ g}$$

b)

$$P_{g,2} = P_{g,1}, \text{ da } P_{g,2} = \frac{F_G}{A} + p_{\text{atmosph}}$$

$$T_{G,2} = T_{EW,1}, \text{ da thermodynamisches Ggw entsteht}$$

~~$dU_{EIS} = dQ$~~ und nicht nicht alles an Eis geschmolzen ist.

$$c) \text{ ~~$dU_{EIS} = dQ$~~ } \quad m \cdot dU = Q_m$$

$$= m_g \int_{T_n}^{T_2} c_p(T) dT = Q_m$$

$$= 3.422 \text{ g} \cdot 0.633 \frac{\text{kJ}}{\text{kg}} (500^\circ\text{C} - 0.003^\circ\text{C})$$

$$= 1082.998 \text{ kJ}$$

$$3d) \quad x_{EIS} = \frac{\phi - \phi_f}{\phi_g - \phi_f}$$

$$= \frac{u_2 - u_f}{u_g - u_f} = \frac{-185.090 \frac{\text{kJ}}{\text{kg}} + 333.458 \frac{\text{kJ}}{\text{kg}}}{\dots}$$

Zustand 1
0.06 kg Eis
0.09 kg Wasser

$$x_{EIS} = \frac{u_2 - u_{\text{flüssig}}}{u_{\text{fest}} - u_{\text{flüssig}}} = \frac{-185.090 + 0.045}{-333.458 + 0.045} = 0.555$$

$$u_1 = -333.458 \frac{\text{kJ}}{\text{kg}} \cdot 0.06 \text{ kg} - 0.045 \frac{\text{kJ}}{\text{kg}} \cdot 0.09 \text{ kg} = -20.009 \text{ kJ}$$

$$Q_m = \Delta u = u_2 - u_1$$

$$u_2 = Q_m + u_1 = 15007 - 200097 = -18509 \text{ kJ}$$

$$u_2 = \frac{u_2}{m} = -185.090 \frac{\text{kJ}}{\text{kg}}$$

4c) ges.: x_n

$$A_m: h_g = h_g(8 \text{ bar}) = 264. \pi \frac{\text{kJ}}{\text{kg}}$$

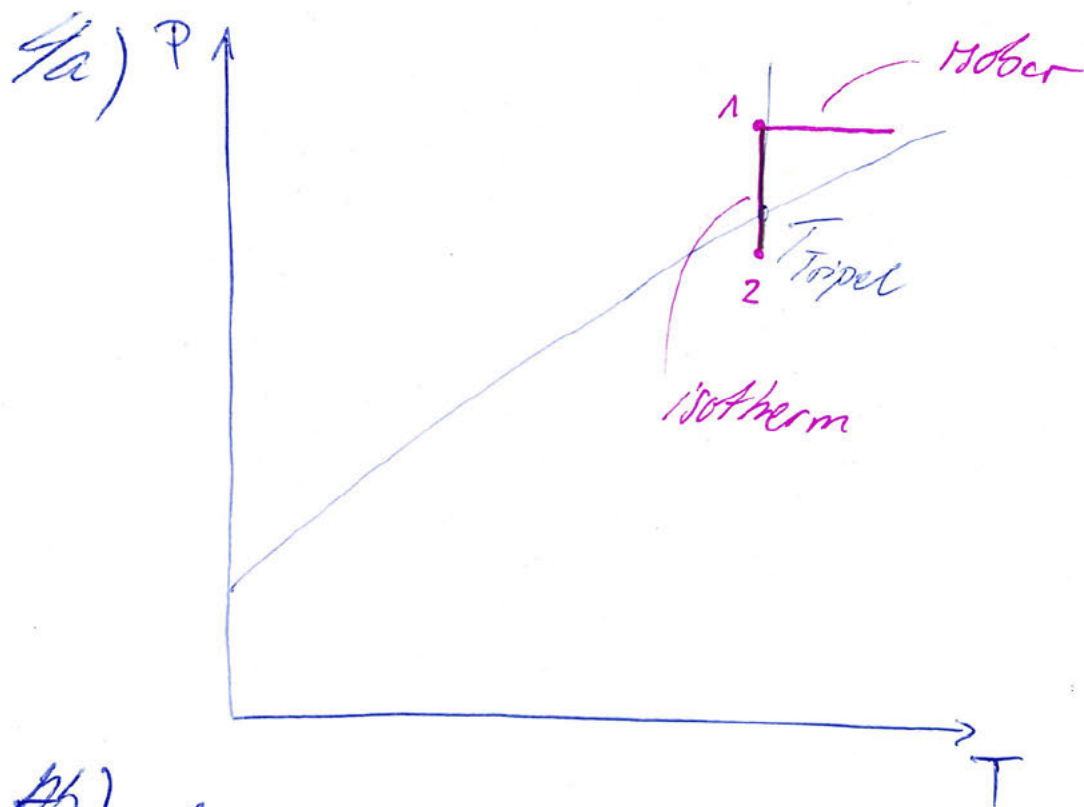
$$\dot{m} = \frac{\dot{Q}_{Kf}}{h_g}$$

$$x_n = \frac{s_g - s_{n,g}}{s_{n,g} - s_{n,f}}$$

$$4d) \epsilon_K = \frac{|\dot{Q}_{zu}|}{|\dot{W}_{el}|} = \frac{|\dot{Q}_K|}{|\dot{W}_K|}$$

$$\dot{Q}_K = \dot{m}(h_2 - h_1)$$

e) Die Temperatur im Innenraum würde weiterhin sinken, da der Innenraum des Gefrier-trockners nach außen hin adiabatisch ist.



4b) ges.: \dot{m}_{R134a}

$$\dot{W}_K = \dot{m} (h_2 - h_3)$$

$$h_3 = h_A(\text{Isobar}) = 93.42 \frac{\text{kJ}}{\text{kg}}$$

$$h_2 = h_g(-22^\circ\text{C}) = 235.31 \frac{\text{kJ}}{\text{kg}}$$

$$s_2 = s_3 = 0.9056 \frac{\text{kJ}}{\text{kgK}}$$

$$h_4 = h_g(\text{Isobar}) = 269.15 \frac{\text{kJ}}{\text{kg}}$$

$$\frac{\dot{W}_K}{h_2 - h_3} = \dot{m} = \frac{28 \text{ W}}{235.31 - 93.42 \frac{\text{kJ}}{\text{kg}}} = 0.172 \frac{\text{kg}}{\text{s}}$$