Afg.

E-Bil. Ein-Aus (stat., isotherm, isochor)

Q = mein (hein - hous) + Qalus + QR hein - hf (70°C) = 292.98 kg haus = hr (100°C)= 479.04 kg

 $Q_{aus} \approx -62.782 \text{ kW}$   $= \frac{h_{aus}^{kr} - h_{ein}^{kr}}{S_{aus}^{kr} - S_{ein}^{kr}} \approx \frac{1_{aus} - 1_{ein}}{2 \ln \left(\frac{1_{aus}}{T_{ein}}\right)} \approx 293.12 \text{ K}$   $h_{aus} - h_{ein} = c \left(\frac{1_{aus} - 1_{ein}}{T_{ein}}\right) + v \left(\frac{1_{aus}}{T_{aus}}\right) = 0$ Solus Sein = clan The

c) 5-3:1. Reaklorwound (stationary kein Massenstrom) 10 Sens - - gave = 01 272.74 1

d) E-3:lanz 1-2 (offen, adiabal da 9712+ garan = 09, isochar)

= U = U2 (my + amz) = u, my = amz heinz

-> am = - 12 my - 4, my = 3756.84 kg

u= uf (70°C) 22 292.95 kg

4= 4 (100°C) + x0 (4 (100°C) - 4 (100°C)) = 429.3778 kg

hein= hf (20°C) = 83.96 kg

e) = 52 1000 m1 (st (20c) + xp (sg)

-0 f = my 5, - mz 52 = my 5, - 52 (my + 4 mz) = - 1387.6 5= 5+ x (5-5) 27.33774 6k

52 = 5 (70°C) = 0.9549 Kgk

=> 05 nz = mzz - m, 5, = 1387.6 K



- Afg. 2 T[k] > 5 [k]
  - 6) P5 5 = 2 15 => 5 = 25 = 3.476 kg mil 7 = n cv - cv = cv (n-1) = 402. 4 Fak 16 = (Ps) == == (Ps) == 2328.07 k mil p6 = p. 1. HS: all = tonst => uz- uz + kez - kez = 0 => cr (T\_6-15) + 2 (w2-w52) =0 => w= N w3 - Zc, (T, -T,) = 507. 244 m/s
  - c) mit w, = 5 10 mg = ex,sh = ho-ho-10 (s, -s) + 1/2 (w2 - w2) = nc, (T, -T) - Top ln (T) - T21n (P) 7+ 1/2 (w2 - w2) ≈725.67
  - d) Ex-Bil. Trie brenk (stationain) The expert = -4 ex, shr + (7 - \frac{10}{7a}) Abo quo = 8 43. 9 73 kg

Afg. 3 2= 7 = 761. 289 5k

A = 4 D2 = 0.00785 m2

-> B

T= 0.01°C

Pg= (mew + mk) 9A + pand = 7.4 boar

Pg~ g= mg ? Ig1 => mg = 13 Ig1 ≈ 3.422g

6) Tg = 273.16k, die Tempenatur des Eiswassers (Tripelpunkt von Wasser)

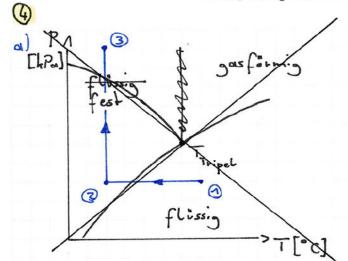
Pgz = pg, = 7.4 bar, der Ampient Pressure und das Gewicht von Eiswosser + Kolben bleiben unversindent.

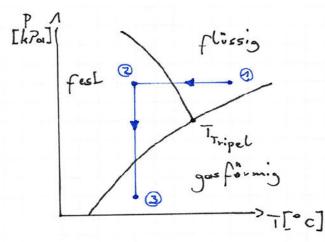
c) E-Bilanz Gas (geschlossen, isoban) = U = mg (vz- vn) = Qnz - Wnz => Q12 = mg cr (T2-T1) + mg [ pg1 dv = mg cv (12-T1) = pg, (V2-mg/12) Vz = ma / 12 = 0. mm L

=> Qnz = -7.507 kg

d) E-Bil Wasser (geschlossen, isobar, isochor) a Une = Qnz = 0 = 1 = 1 = 1 = - 184.97.06 kg Un = U + x ( U Fest - U Firssing) - 200,0406 Fig x = U - V Firsting - 1 0.5548 = 0.555

Afg. 4





$$h_{z} = h_{3} \left(-16^{\circ} c\right)^{\frac{4}{2}} = 232.74 \frac{10}{10}$$

$$h_{3} = h\left(8 \log_{10} s = s_{3}\right)^{\frac{4}{2}} = 232.74 \frac{10}{10}$$

$$h_{3} = h\left(8 \log_{10} s = s_{3}\right)^{\frac{4}{2}} = 273.61 - 264.75 = 0.9066 + 264.75 \approx 277.31 \frac{10}{10}$$

$$s_{z} = s_{3} = s_{3} \left(-16^{\circ} c\right)^{\frac{4}{2}} = 0.9298 \frac{10}{10}$$

$$s_{z} = s_{3} = s_{3} \left(-16^{\circ} c\right)^{\frac{4}{2}} = 0.9298 \frac{10}{10}$$

e) 
$$h_1 = h_1 = h_1 (8601) = 93.42 = 6$$

$$x_1 = \frac{h_1 - h_1}{h_2 - h_1} \approx 0.3076$$
bei  $T_1 = -76^{\circ}C$ 

d) 
$$\mathcal{E}_{k} = \frac{|\dot{q}_{su}|^{2}}{|\dot{w}_{k}|} = \frac{\dot{q}_{k}}{|\dot{w}_{k}|}$$

$$E - 3; L. \gamma - 2 \qquad (slat, isobar)$$

$$\Rightarrow \dot{q}_{k} = \dot{m}_{k \gamma 3 k_{-}} (h_{2} - h_{\gamma}) \approx 120.36 \text{ W}$$

$$\dot{\omega}_{E_{k}} = 4.299$$

e) Die Temperatur würde weiter Sinken bis Tr = T; und dann Konshant bleiben bei der gleichen Temperatur wie im Verdampfer (Tr)