

# Aufgabe 1

(a)  $\dot{Q}_{\text{aus}}$  über wand

Energie Bilanz um Kessel

$$\dot{Q} = \dot{m} [h_c - h_a] + \sum \dot{Q}_j - \sum \dot{E}_{\text{kin}}^0$$

$$h_c = h_f + x \cdot (h_g - h_f) \Rightarrow h_f + x \cdot h_{gf}$$

$$h_g = h_g[100^\circ\text{C}] \quad h_f = h_f[100^\circ\text{C}]$$

$$= 419.04 + 0.005 \cdot 2257.0 = 430.325$$

$$\Rightarrow -\dot{Q} = 0.3 \frac{\text{kg}}{\text{s}} \cdot 3(430.325 - 292.98) = \underline{\underline{-41.264 \text{ kW} = \dot{Q}_{\text{aus}}}}$$

$$\text{b)} \overline{h_f} = \frac{\int_e^a T ds}{s_a - s_e} = \frac{T \cdot (s_a - s_e)}{s_a - s_e}$$

$$s_a = 0.99549 \frac{\text{kJ/kg} \cdot \text{K}}{s_e} = 7.7553 \frac{\text{kJ/kg} \cdot \text{K}}{\text{K}}$$

(b)

$$\text{c)} \dot{S}_{\text{erz}} \Rightarrow \dot{Q} = \dot{m} [s_c - s_a] + \sum \dot{S}_j + \dot{S}_{\text{erz}} \quad \text{Temperaturanhebung}$$

$$\dot{S}_{\text{erz}} = \dot{m} [s_c - s_a] + \frac{\dot{Q}_{\text{R}}} {T} - \frac{\dot{Q}_{\text{aus}}}{T} = \dot{m} [s_c - s_a] + \frac{\dot{Q}_{\text{R}} - \dot{Q}_{\text{aus}}}{T}$$

$$\dot{S}_{\text{erz}} = \cancel{0.99549} 0.3 \cdot \cancel{0.99549} [7.7553 - 0.99549] + \frac{100 \text{ kW} - 41.264 \text{ kW}}{298.15 \text{ K}}$$

$$\dot{S}_{\text{erz}} = \underline{\underline{2.225 \frac{\text{kJ/kg} \cdot \text{K}}{}}}$$

$$\text{d)} \text{ halboffenes System!} \quad \Delta E = m_2 u_2 - m_1 u_1 = \Delta m h_i + \sum Q_j - \sum E_{\text{kin}}^0$$

$$E = U + K_e + P_e^0 \rightarrow \Delta E = \Delta U$$

$$\Delta U = \Delta m h_i + \sum Q_j = \Delta m h + \sum Q_j$$

$$\Delta m = \frac{\Delta U + Q_{R,12}}{\Delta h} = 1 = \Delta u + \frac{Q_{R,12}}{\Delta h \Delta m}$$

$$\Delta U = \Delta m \cdot \Delta u$$

$$\Delta u = \frac{Q_{R,12}}{\Delta h \Delta m}$$

$$1 - \Delta u = \frac{Q_{R,12}}{\Delta h \Delta m} \Rightarrow (1 - \Delta u) \Delta m = \frac{Q_{R,12}}{\Delta h}$$

$$\Delta u: u_f[20^\circ\text{C}] = 83.95 \frac{\text{kJ}}{\text{kg}}$$

$$u[70^\circ\text{C}] = 292.95 \frac{\text{kJ}}{\text{kg}}$$

$$\Delta h: h[20^\circ\text{C}] = 83.96 \frac{\text{kJ}}{\text{kg}}$$

$$h[70^\circ\text{C}] = 292.98 \frac{\text{kJ}}{\text{kg}}$$

$$\Delta m = \frac{Q_{R,12}}{\cancel{\Delta h}} / ((\Delta h)(1 - \Delta u))$$

$$\Delta m = \frac{Q_{R2}}{\eta h(1-\alpha u)} = \frac{3.5 \cdot 10^6 \text{ kJ}}{209.35(1-20\%)} \approx \underline{803.76 \text{ kg}}$$

Δ

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$\Delta m$  aus Aufgabenstellung!

$$c) \Delta S_{12} = m_2 s_2 - m_1 s_1 \quad \cancel{(m_2 - m_1)(s_2 - s_1)} = m_2 s_2 - m_1 s_1 + \cancel{m_1}$$

$$\Rightarrow \Delta m \quad \cancel{m_2(s_2 - s_1)} \Rightarrow \cancel{m_2(s_2 - s_1)} = m_1(s_1 - s_2) =$$

$$\Rightarrow s_2 = 0.9549$$

$$s_1 = s_f + 0.005 \cdot (s_g - s_f) = 1.3069 + 0.005 \cdot (7.3549 - 1.3069) \\ = 1.33714$$

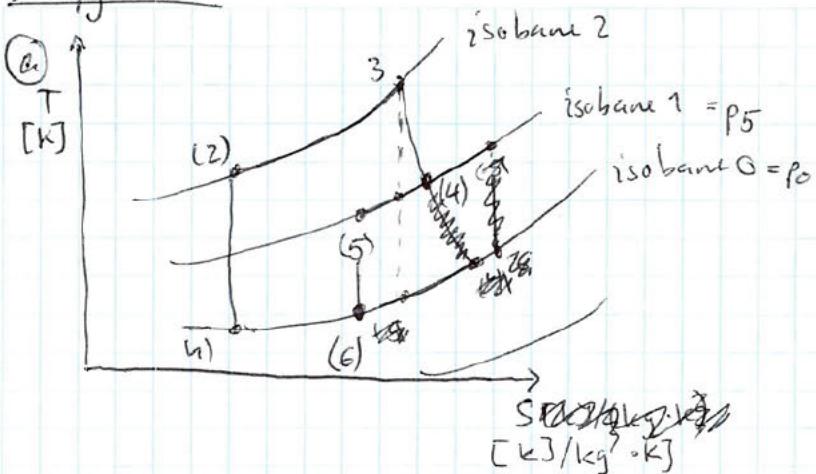
$$\Rightarrow m_1 = 5755 \text{ kg} \quad m_2 = 5755 + 3600 = 9355 \text{ kg}$$

$$\Delta S_{12} = \cancel{5755} \quad 9355 \cdot 0.9549 - 5755 \cdot 1.33714 =$$

$$= 8433.0895 - 7695.26407 = \underline{1237.85 \text{ kJ/kg} \cdot \text{K}}$$

## Aufgabe 2

### Aufgabe 2



(b)  $\dot{m}_G, T_6$  Da idealer Gas mit polytropem Tverlust / fests

$$\frac{T_6}{T_5} = \left(\frac{p_G}{p_5}\right)^{\frac{n-1}{n}} \quad n=k=1.4 \quad p_G = p_0 = 0.191 \text{ Bar} \quad T_5 = 431.9 \text{ K}$$

$$T_6 = 431.9 \text{ K} \cdot \left(\frac{0.191}{0.5}\right)^{\frac{0.4}{1.4}} = \underline{\underline{328.075 \text{ K}}}$$

$$p_5 = 0.5 \text{ Bar}$$

$$\begin{aligned} \rho_A &= \dot{m} \\ \rho_v &= \dot{m} \quad \rightarrow \dot{m}_6 = \dot{m}_5 \quad \left(\frac{v_1}{v_2}\right)^{n-1} = \left(\frac{p_2}{p_1}\right)^{\frac{n-1}{n}} \Rightarrow v_6 = v_5 \cdot \left(\frac{p_6}{p_5}\right)^{\frac{1}{n}} \\ v_5 &= \frac{RT_5}{p_5} \quad R = \frac{\bar{R}}{M} \quad \cancel{\frac{0.314}{1.006}} = c_p^{\text{ig}} - c_v^{\text{ig}} \quad k = \frac{c_p^{\text{ig}}}{c_v^{\text{ig}}} \rightarrow \cancel{R} c_v^{\text{ig}} = \frac{c_p^{\text{ig}}}{k} \\ &\quad R = 1.006 - 0.718 = 0.287 \frac{\text{kJ}}{\text{kg} \cdot \text{K}} \quad = 0.718 \frac{\text{kJ}}{\text{kg} \cdot \text{K}} \end{aligned}$$

$$(c) \Delta e_{x,\text{str}} = e_{x,\text{str},6} - e_{x,\text{str},0}$$

$$\begin{aligned} \dot{m} e_{x,\text{str}} &= \dot{m} \left[ h_6 - h_0 + T_6 (s_6 - s_0) + k \epsilon + \cancel{f(e)} \right] \Rightarrow e_{x,\text{str}} = h_6 - h_0 - T_0 (s_6 - s_0) + \frac{k \epsilon}{\dot{m}} \frac{\omega^2}{2} \\ h_6 - h_0 &= \int_{T_0}^{T_6} c_p^{\text{ig}} dT = c_p^{\text{ig}} (T_6 - T_0) = 1.006 \cdot (328.075 \text{ K} - 243.15 \text{ K}) = 85.43 \frac{\text{kJ}}{\text{kg}} \\ s_6 - s_0 &= \int_{T_0}^{T_6} \frac{c_p^{\text{ig}}}{T} dT - R \ln \left( \frac{p_6}{p_0} \right) = c_p^{\text{ig}} \ln \left( \frac{T_6}{T_0} \right) - R \cdot \ln \left( \frac{p_6}{p_0} \right) = \\ &= 1.006 \ln \left( \frac{328.075 \text{ K}}{243.15 \text{ K}} \right) - 0.287 \cdot \ln(1) = 1.006 \cdot \ln \left( \frac{328.075 \text{ K}}{243.15 \text{ K}} \right) - 0 \\ \Rightarrow T_0 \cdot \Delta s &= 243.15 \cdot 0.301 = 73.27 \frac{\text{kJ}}{\text{kg} \cdot \text{K}} \quad = 0.301 \frac{\text{kJ}}{\text{kg} \cdot \text{K}} \end{aligned}$$

$\omega_6$  von angegebenen stellung, die nicht gefunden vorher!

$$\Delta ke = \frac{\omega^2}{2} = \rho e \bar{E}_{x,str} = 85.43 \frac{kg}{kg} - 73.27 \frac{kg}{kg} + \frac{510 \frac{m^2}{s^2} kg}{2} = 12.16 \frac{kg}{kg} + \frac{130.050 \frac{m^2}{s^2} kg}{2}$$

$$\Delta w = \omega_6 - \omega_2 = 510 - 266 = 300 \frac{m}{s}$$

~~$$\textcircled{1} \quad \Delta w = \frac{\Delta w^2}{2} = 48050 \frac{kg \cdot m^2}{s^2} \Rightarrow ke = 48.050 \frac{kg}{s^2}$$~~

$$e_{x,str} = 85.43 - 73.27 + 48.050 = 60.21 \frac{kg}{kg^2}$$

①  $c_{x,verl}$  stationärer  $\dot{E}_x$  Prozess  $\rightarrow$  Exergie Bilanz um Triebwerk

$$0 = \dot{m} \cdot \dot{e}_{x,str} + \underbrace{\varepsilon \left(1 - \frac{T_0}{T}\right) \dot{Q}_j}_{\dot{E}_{x,Q,j}} - \dot{E}_{x,n} - \dot{E}_{x,verl}$$

$$\begin{aligned} \dot{E}_{x,Q} &= \left(1 - \frac{T_0}{T}\right) \dot{Q} \\ &= \left(1 - \frac{248.15 K}{1289 K}\right) \cdot 1195 = 769.582 \frac{kg}{kg} \cdot \dot{m} \Rightarrow \dot{e}_{x,Q} = \frac{969.582 \frac{kg}{kg}}{kg} \end{aligned}$$

$$\Rightarrow \dot{e}_{x,verl} = \dot{e}_{x,str} + \dot{e}_{x,Q}$$

$$\dot{e}_{x,verl} = - \dot{e}_{x,str} + \dot{e}_{x,Q} = 969.582 \frac{kg}{kg} - 60.21 \frac{kg}{kg} = \underline{\underline{909.37 \frac{kg}{kg}}}$$

### Aufgabe 3

(a)  $P_{g,1}$  mg

$$P_{g,1} = P_{\text{amb}} + P_{\text{gewicht}} = P_{\text{amb}} + P_{g,2} = D$$

$$D = 0.1 \text{ m}$$

$$P_{g,2} = 1 \text{ Bar} + 0.1002 \text{ Bar} = \underline{\underline{1.1002 \text{ Bar}}} = \underline{\underline{1.1 \text{ Bar}}}$$

$$\text{für Masse: } pV = m \cdot R \cdot T \rightarrow m = \frac{pV}{RT}$$

$$\begin{aligned} V_{g,1} &= 3.14 \cdot 10^{-3} \text{ m}^3 \\ T_{g,1} &= 773.15 \text{ K} \end{aligned}$$

$$R = \frac{8.314}{M} = \frac{8.314}{0.166} = 0.166 \frac{\text{kJ}}{\text{kgK}}$$

$$\begin{aligned} P_{\text{gewicht}} &= \frac{m_k \cdot g}{A} + \frac{m_{\text{EW}} \cdot g}{A} \\ &= 32 \text{ kg} \cdot 9.81 \frac{\text{m}}{\text{s}^2} + 0.1 \text{ kg} \cdot 9.81 \frac{\text{m}}{\text{s}^2} \end{aligned}$$

$$= 1002.36 \text{ Pa}$$

$$\begin{aligned} m &= \frac{11002.36 \text{ Pa} \cdot 3.14 \cdot 10^{-3} \text{ m}^3}{773.15 \text{ K} \cdot 0.166} = 2.692 \text{ kg} \\ &= \underline{\underline{2.692 \text{ kg}}} \end{aligned}$$

$\Delta$  (b)  $x_{E,2} > 0$   $T_{g,2}, P_{g,2}$   $T_{g,2}$  ist gleich wie  $\cancel{T_{1,g}}$ , da sich das EW noch im "Nuss-fest" gebiet ~~noch~~ ist.

(c)

c)  $\alpha_{12}$  zwischen 1-2 Energie Bilanz:  $\frac{dE}{dt} = \{Q_j - \sum_{i,v,n} \dot{E}_{i,v,n}^{\circ}\}$  (geschlossenes System)

$$\frac{d}{dt}(U + K_e + \cancel{q_e}) = \Delta Q \quad (\text{Tz aus Aufgabenstellung})$$

$$\begin{aligned} \Delta U_{12} &= \Delta Q_{12} \Rightarrow \Delta u_{1,2} = \int_{T_1}^{T_2} c_v^{\circ} dT = c_v^{\circ} (T_2 - T_1) = \cancel{c_v^{\circ}} q \\ \Delta u_{1,2} &= 0.633 \cdot (-0.003) = \cancel{-0.001.9} \frac{\text{kJ}}{\text{kg}} \end{aligned}$$

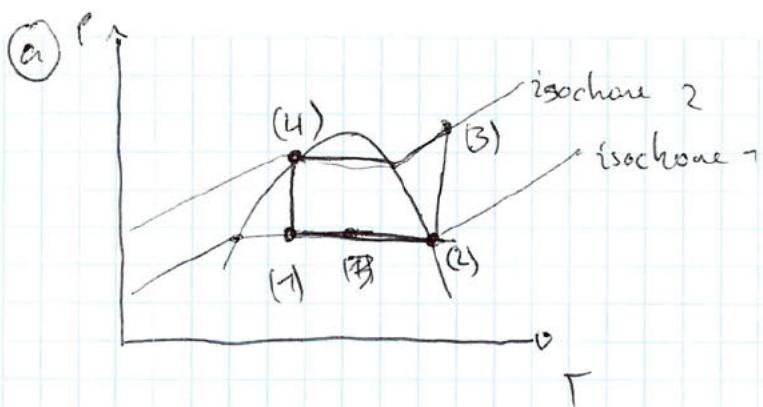
$$\Delta q_{12} = 0.0019 \frac{\text{kJ}}{\text{kg}} \rightarrow m \cdot \Delta q_{12} = 1 \text{ kg} \cancel{- \frac{1900}{1900}} = 1900 \cancel{\frac{1900}{1900}} \frac{\text{kJ}}{\text{kg}}$$

$\Rightarrow 1900 \text{ kJ/Kugel}$  führt zu einem Temperaturunterschied von  $-190^\circ$

①  $x_{EIS,2}$

Δ

## Aufgabe 4



(b)  $\dot{m}_{v=134}$  Energie Bilanz um ~~Drossel~~ Kompressor Verdichter

$$0 = \dot{m} (h_c - h_a) - \dot{w}_k \rightarrow \dot{m} = \frac{\dot{w}_k}{h_c - h_a}$$

$\Delta h = 243.78$  nach drossel

(c)  $x_1$  nach drossel  $x = \frac{\phi - \phi_f}{\phi_g - \phi_f}$  mit  $\phi = v, u, h, s$

$$\Rightarrow h = 243.78 \text{ da drossel isenthalp per definition}$$

⑦  $E_k = \frac{|Q_{ab}|}{|W_t|} = \frac{6}{28} = \underline{\underline{0.214}}$

a

⑧ Das Wasser würde in die kritische Phase übergehen!