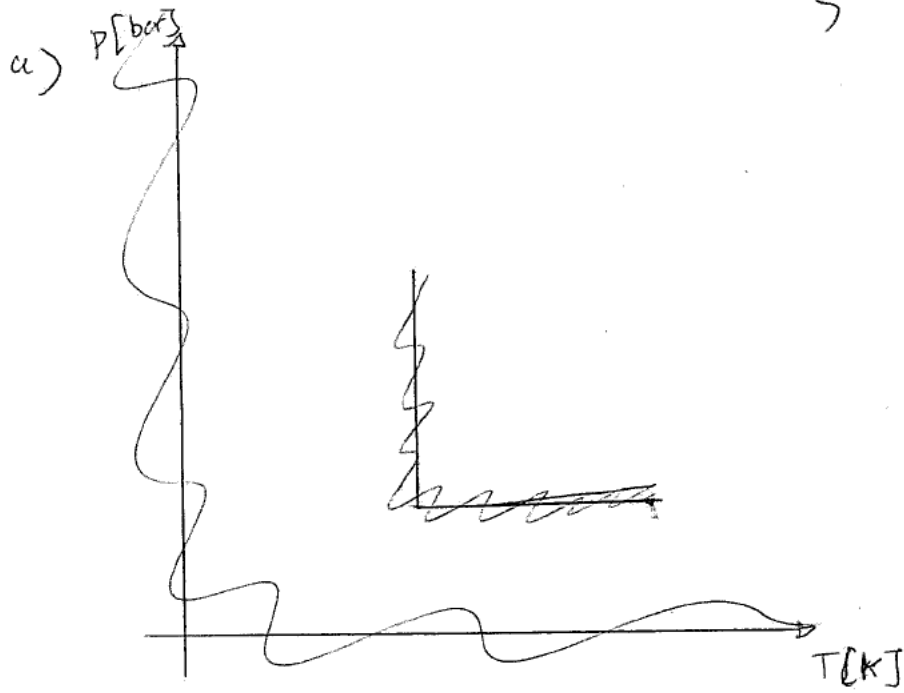


Aufgabe 4

5

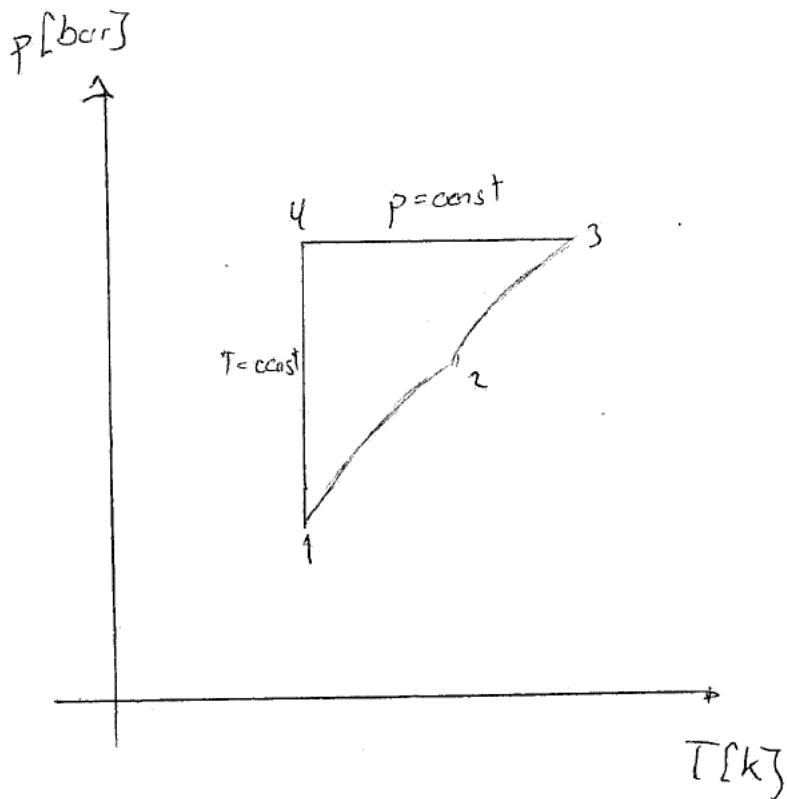


1-2: ~~p~~ T p

2-3: $S_2 = S_3$ p

3-4: isobar T $p = \text{const.}$

4-1: isenthalpe $h_4 = h_1$ p $T = \text{const.}$



Aufgabe 4

b)

$$\dot{m}(h_2 - h_3) + \dot{Q} - \dot{W} = 0$$

$$\dot{m} = \frac{\dot{W}}{h_2 - h_3}$$

$$s_2 = s_3$$

~~$$h_3 = h_2 - \frac{\dot{W}}{\dot{m}}$$~~

$$s_3 = s_2 = 0.9066$$

c)

d)

$$e) \quad \epsilon_k = \frac{|\dot{Q}_{zu}|}{|\dot{W}|} = \frac{|\dot{Q}_k|}{\dot{W}_k}$$

$$\dot{Q}_k = -\dot{m}h$$

a) Aufgabe 3

4

$$pV = mRT \quad (m_{EW} + m_{eis} = 0,16 \text{ kg})$$

$$\Sigma F = 0$$

$$p_{g,1} \cdot A - 0,1 \cdot g - 32 \cdot g - p_{amb} \cdot A = 0$$

$$p_{g,1} = \frac{(0,1 + 32) \cdot g}{A} + p_{amb}$$

$$A = \left(\frac{D}{2}\right)^2 \cdot \pi = \left(\frac{10 \cdot 10^{-3}}{2}\right)^2 \cdot \pi = 0,0078 \text{ m}^2$$

$$= \frac{(0,1 + 32) \cdot 9,81}{0,0078} + 1 \cdot 10^5 = 1,4 \text{ bar}$$

$$m_{g,1} = \frac{p_{g,1} \cdot V_{g,1}}{R \cdot T_{g,1}}$$

$$R = \frac{R}{M} = \frac{8314}{50} = 166,28$$

$$= \frac{1,4 \cdot 10^5 \cdot 3,14 \cdot 10^{-3}}{166,28 \cdot 773} = 0,0025 \text{ kg} = 2,5 \text{ g}$$

$$= 0,0034 \text{ kg} = 3,4 \text{ g}$$

b) $h_{gas} = h_{EW}$

○ Aufgabe 3

Energiebilanz

$$Q - W = \Delta U + \Delta KE^{PO} + \Delta PE^{PO}$$

$$Q_{12} = W_{12} + \Delta U_{12}$$

$$\Delta U_{12} = U_2 - U_1 = m_g (u_2 - u_1) = m_g (c_v (T_{g,2} - T_{g,1}))$$

$$= 3.6 \cdot 10^{-3} \cdot 0.633 (0.003 - 500)$$

$$= -1.14 \text{ kJ}$$

b) $T_{g,2} = T_{EW,2} = 0.003 \text{ } ^\circ\text{C}$

$$\begin{aligned} Q_{12} &= \Delta U = m \cdot \Delta u = (m_{EW} + m_{eis}) (x_{eis} \cdot u_{Fest} + (1 - x_{eis}) \cdot u_{flüssig}) \\ 1500 \text{ J} &= 0.16 \text{ kg} (x_{eis} (u_{Fest} - u_{flüssig}) + u_{Fest}) \\ x_{eis} &= \frac{\frac{1500 \text{ J}}{0.16} - u_{Fest}}{u_{Fest} - u_{flüssig}} = \frac{\frac{15 \text{ kJ}}{0.16} + 333.4}{-333.4 + 0.033} = \end{aligned}$$

$$Q_{12} = \Delta U = u_2 - u_1$$

$$\frac{u_2}{m_{tot}} = x_{eis,2} \cdot u_{Fest} + (1 - x_{eis,2}) \cdot u_{flüssig}$$

$$\frac{u_1}{m_{tot}} = x_{eis,1} \cdot u_{Fest} + (1 - x_{eis,1}) \cdot u_{flüssig}$$

more on another page

c) Aufgabe 2

3

$$-\Delta_{ex, str} = h_0 - h_6 - T_0 (S_0 - S_6) + \Delta h_k + \cancel{A_{pe}^{p0}}$$

$$h_6(340K) = 340.42$$

$$S_6^0(340K) = 1.8279$$

$$h_0(243K) = \frac{250.05 - 240.02}{250 - 240} (243 - 240) + 240.02$$

$$= 243.03 \frac{kJ}{kg}$$

$$S_0^0(243K) = \frac{1.51917 - 1.47824}{250 - 240} (243 - 240) + 1.47824$$

$$= 1.490519 \frac{kJ}{kgK}$$

$$-\Delta_{ex, str} = h_0 - h_6 - T_0 \left(S_0^0 - S_6^0 - R \ln \left(\frac{p_0}{p_6} \right) \right) + \frac{w_0^2}{2} - \frac{w_6^2}{2}$$

$$= 243.03 - 340.42 - 243 \left(1.490519 - 1.8279 - 287 \ln \left(\frac{0.191}{0.5} \right) \right) + \frac{200^2}{2} - \frac{510^2}{2}$$

$$= -95 \frac{kJ}{kg}$$

$$\Delta_{ex, str} = 95 \frac{kJ}{kg}$$

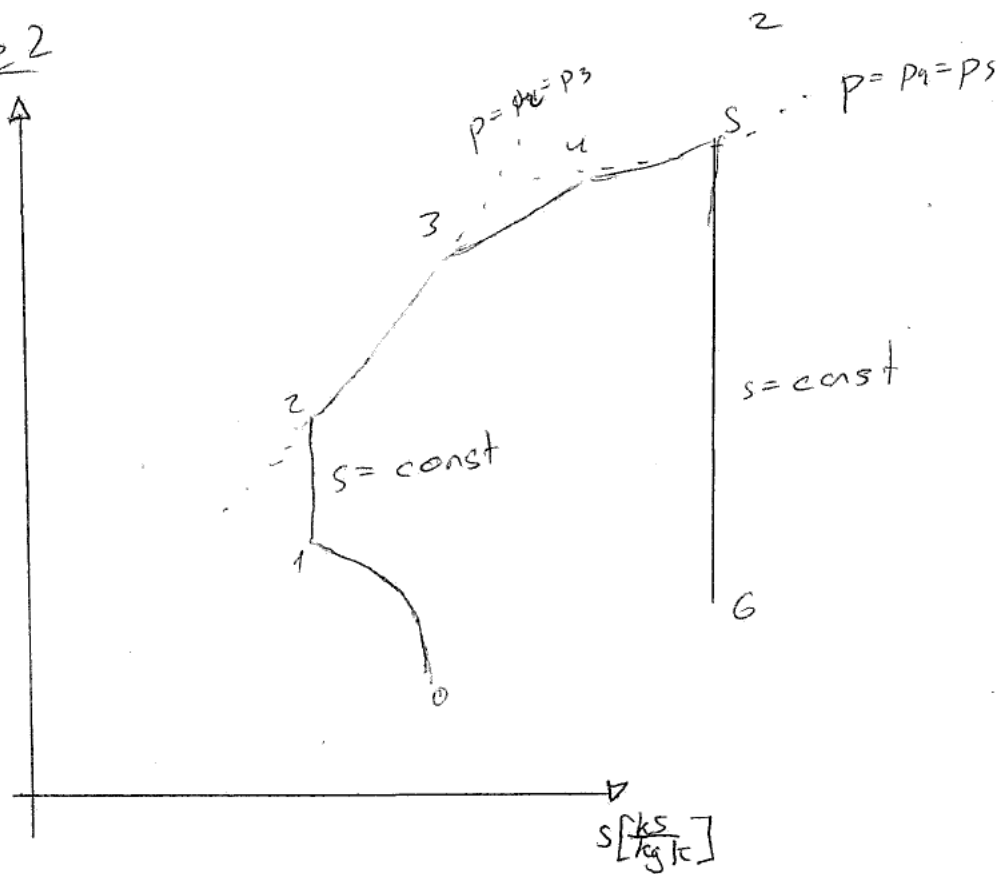
d)

$$\dot{E}_{x, verl} = T_0 \cdot \dot{S}_{erz}$$

$$\dot{S}_{erz} =$$

Aufgabe 2

a) $T[K]$



0-1: ~~s~~ s T

1-2: isentrope $s_1 = s_2$

2-3: isobar T

3-4: s

4-5: $p_4 = p_5$

5-6: ~~s~~ $s_5 = s_6$

Aufgabe 2

b) Energie bilanz

$$0 = \dot{m} \left(h_s - h_0 + \frac{w_s^2 - w_0^2}{2} + g(z_s - z_0) \right) + \dot{Q} - \dot{W}$$

isentrope $s_s = s_0$

$$s_s^0(431.9 \text{ K}) = \frac{2.0887 - 2.06533}{440 - 430} (431.9 - 430) + 2.06533$$

$$= 2.06977 \quad (\text{Wert von A-22})$$

$$s_0 - s_s = s_0^0 - s_s^0 - R \ln \left(\frac{p_0}{p_s} \right) = 0 \quad R = \frac{8314}{28.57} = 287$$

$$s_0^0 = s_s^0 + R \ln \left(\frac{p_0}{p_s} \right)$$

$$p_0 = p_0 = 0.181 \text{ bar}$$

$$p_s = 0.5 \text{ bar}$$

$$= 2.06977 + 287 \cdot \ln \left(\frac{0.181}{0.5} \right) \cdot 10^3$$

$$= 1.7935 \frac{\text{kJ}}{\text{kg} \cdot \text{K}}$$

$$s^0(T = 325 \text{ K}) = 1.78249$$

$$s^0(T = 330 \text{ K}) = 1.79783$$

~~T = 340 K~~

$$T_0 = \frac{330 - 325}{1.79783 - 1.78249} (1.7935 - 1.78249) + 325 \quad (\text{Wert von A-22})$$

$$= 328.5 \text{ K}$$

$$0 = \dot{m} \left(h_s - h_0 + \frac{w_s^2}{2} - \frac{w_0^2}{2} \right)$$

$$w_0^2 = 2 \left(h_s - h_0 + \frac{w_s^2}{2} \right)$$

$$h_s = \frac{441.61 - 431.43}{440 - 430} (431.9 - 430) + 431.43 \quad (\text{Wert von A-22})$$

$$= 433.36 \frac{\text{kJ}}{\text{kg}}$$

$$h_0 = \frac{330.34 - 325.31}{330 - 325} (328.5 - 325) + 325.31 \quad (\text{Wert von A-22})$$

$$= 328.83 \frac{\text{kJ}}{\text{kg}}$$

$$w_0 = \sqrt{2 \left(433.36 - 328.83 \right)} = 507 \frac{\text{m}}{\text{s}}$$

Aufgabe 1

1

a) Energiebilanz:

$$0 = \dot{m}(h_e - h_a) + \dot{Q}_R - \dot{Q}_{\text{aus}}$$

$$\dot{Q}_{\text{aus}} = \dot{m}(h_{\text{ein}} - h_{\text{aus}}) + \dot{Q}_R$$

$$h_{\text{ein}} = 292.98 \frac{\text{kJ}}{\text{kg}}$$

$$h_{\text{aus}} = 419.04 \frac{\text{kJ}}{\text{kg}}$$

$$\begin{aligned}\dot{Q}_{\text{aus}} &= 0.3 (292.98 - 419.04) + 100 \\ &= 62.182 \text{ kW}\end{aligned}$$

b)

$$\dot{Q}_{\text{aus}} = 65 \text{ kW}$$

$$\bar{T}_{\text{KF}} = \frac{\int_{\text{ein}}^{\text{aus}} T ds}{s_{\text{aus}} - s_{\text{ein}}}$$

Aufgabe 1

c) $\bar{T}_{kf} = 295 \text{ K}$

$$0 = \dot{m}(s_e - s_a) + \sum \frac{\dot{Q}_j}{\bar{T}} + \dot{S}_{erz}$$

$$\dot{S}_{erz} = -\dot{m}(s_e - s_a) - \sum \frac{\dot{Q}_j}{\bar{T}_j}$$

d) $\cancel{\Delta m} \Delta m [h] + Q - \cancel{W} = \Delta U + \cancel{\Delta KE} + \cancel{\Delta PE}$

e) $\Delta S = m_2 s_2 - m_1 s_1$

$$m_2 = m_1 + \Delta m_{12}$$