Q5.1

5 pts a) radiative transfer $I(x) = I_0 e^{-n\nabla x}$

 X_m is the dist. over which $I_0 \rightarrow I_0/e$ $I(X_m) = I_0 e^{-nOX_m} = I_0/e = I_0 e^{-1}$

3 pts b) $T = X/x_m$ $\Rightarrow X = Tx_m$ $I(t) = I_0 e^{-n\sigma x_m t} = I_0 e^{-(1) \cdot t} = I_0 e^{-t}$

6 pts c) slab of Thickness I cm absorbs 15% of The light. ie $I(1cm) = 0.85I_0$.

> Since $I(t) = I_0 e^{-T}$, This means $e^{-T} = 0.85$ ln(0.85) = -T

T = 0.16

And $x_m = x/t$, so $x_m = \frac{1 \text{ cm}}{-\ln(0.85)} = 6.15 \text{ cm}$

Now how thick would the slab be to absorb 99% of the light?

That occurs when $e^{-T} = 0.01$

T = ln(100) = 4,61

and since $x = T \times_m$. That Thickness required is x = (4.61)(6.15 cm) = 28.3 cm.

$$\langle E \rangle = \int_{0}^{\infty} EF(E)dE = \int_{0}^{\infty} E \cdot \frac{2}{\pi} kT \left(\frac{E}{kT}\right)^{2} exp\left(\frac{E}{kT}\right) dE$$

 $Sub X = \frac{E}{kT}$ $dX = \frac{1}{kT} dE$

(E) = 2 ITH I LIX X 2 e-X KTOLX

 $= 2 L T \int_{0}^{\infty} x^{3/2} e^{-X} dx = 2 L T 3 \sqrt{4} = \frac{3}{2} L T$

The only tricky part is beeping track of all the kT's in the substitution.

6 pts

(E) = 3 KT A particle whose energy is 4 times the mean has energy $4 \times \frac{3}{2} kT = 6 kT$.

Suppose that GhT is enough to ionize on H atom, i.e. $6kT = 13.6eV \times 1.602 \times 10^{19} J = 2.18 \times 10^{-16} J$

 $T = 13.6 \text{ eV} \times 1.602 \times 10^{-19} \text{ J}$ $= V \qquad 6 \times 1.38 \times 10^{-23} \text{ J/k}$

= 2.6×104K

and if this is a proton its KE (\$mv2) -> speed v given by

 $\frac{1}{3}mV^2 = E$ $V = \left(\frac{2E}{m}\right)^{1/2} = \left(\frac{2 \times 2.18 \times 10^{-18} \text{J}}{1.67 \times 10^{-27} \text{lg}}\right) = 5.1 \times 10^{4} \text{ m/s}$

C) Similar computation but now instead of requiring 13.6 eV, we need 1855 to go n=1 -> n=2

5 ptsc) Confinued Bohr model energies are 13,6 eV (12) So DE for n=1-> n=2 is DE=13.6eV (1-4) by analogy with previous part, the necessary temp is $T = (2.6 \times 10^4 \text{ K}) \times 3/4 = 2.0 \times 10^4 \text{ K}$ speed will be 51 km/s (3/4) 2 = 44 km/s Both of these temperatives are much higher than the Surface of the Sun, which is only about 6000K. It atoms on the Sun's surface probably exist in a variety of ionization and excitation states but most of them will probably be neutral atoms in the n=1 state. (More into in later chapters.)

25.3 4 pts a) Check blackbody flux per unit area for a temp of 310 K. Flux from a blackbody surface $F = \sqrt{58}T^4$ for 310 K This is $5.67 \times 10^{-8} \frac{W}{m^2 \cdot K^4} \times (310 \, \text{K})^4 = 523 \, \text{W/m}^2, \text{ok}$ 523 W = 523 J x 24×3600s x 2,0 m2 = 9.0×107 J 4 pts Assuming a surface area of maybe 2,0 m²?
This value is hard to guess but 1 m² seems too little food calones: 9.0×107 Jx / Cal = 2.1×104 Cal

5 pts

$$I_{\nu} = \frac{2h\nu^{3}/c^{2}}{e^{h\nu/kT}-1} = \frac{2h\nu^{3}/c^{2}}{e^{5.86}}$$

Black Body Flux is $F = \int_0^\infty dx \int_{\mathbb{R}^2} Ir \cos\theta dx$

solid angle de = sinodody

$$= \int_{0}^{\infty} I_{\nu} d\nu \cdot 2\pi \cdot \frac{1}{2} = \pi \int_{0}^{\infty} \frac{2h\nu^{3}/c^{2}}{e^{h\nu/h\tau}-1} d\nu$$

Sub x = hV/kT. $dx = (\frac{h}{kT})dv$

$$\chi^{3} = \frac{h^{3}v^{3}}{(kT)^{3}}$$
 so $\frac{2hv^{3}}{c^{2}} = \frac{h^{3}v^{3}}{(kT)^{3}}, \frac{(kT)^{3}}{h^{2}}, \frac{2}{c^{2}} = \frac{2\chi^{3}(kT)^{3}}{c^{2}}$

 $F = \pi \int_{0}^{\infty} \frac{2x^{3}(kT)^{3}}{C^{2}h^{2}} \frac{1}{e^{X}-1} \left(\frac{kT}{h}\right) dx$

$$= \frac{2\pi (kT)^{3}(kT)}{C^{2}} \int_{h^{2}}^{\infty} (hT) \int_{0}^{\infty} \frac{x^{3}}{e^{x}} dx \approx \frac{2\pi (kT)^{4}}{C^{2}} \frac{\pi^{4}}{15}$$

 $\frac{2}{15}\frac{2\pi^{5}k^{4}}{c^{2}h^{3}}$, $T^{4} \approx v_{SB}$. T^{4}