

11. Light & Colour

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Agenda

- Physics of Light.
- Interactions between light and the environment
- Integrating radiance
- Types of light
- Color theory
- Color spaces





What's the Difference?

- Light is a physical phenomenon
 - Photons of varying wavelength
 - A spectral distribution
- Colour is a perceptual phenomenon
 - How the human eye interprets it
 - Based on the rods and cones in the eye
- What you learned in school is simplified

Photons

- Light is a particle – a photon
 - But also a (sine) wave with a frequency
- Energy of a photon is related to frequency:
 - $E = hf = \frac{hc}{\lambda}$
 - $h \approx 6.626 \times 10^{-34} \frac{kg\ m^2}{s}$ (Planck's constant)
 - $f = \frac{c}{\lambda}$ is the frequency
 - $c \approx 2.99 \times 10^8 \frac{m}{s}$ is speed of light in a vacuum

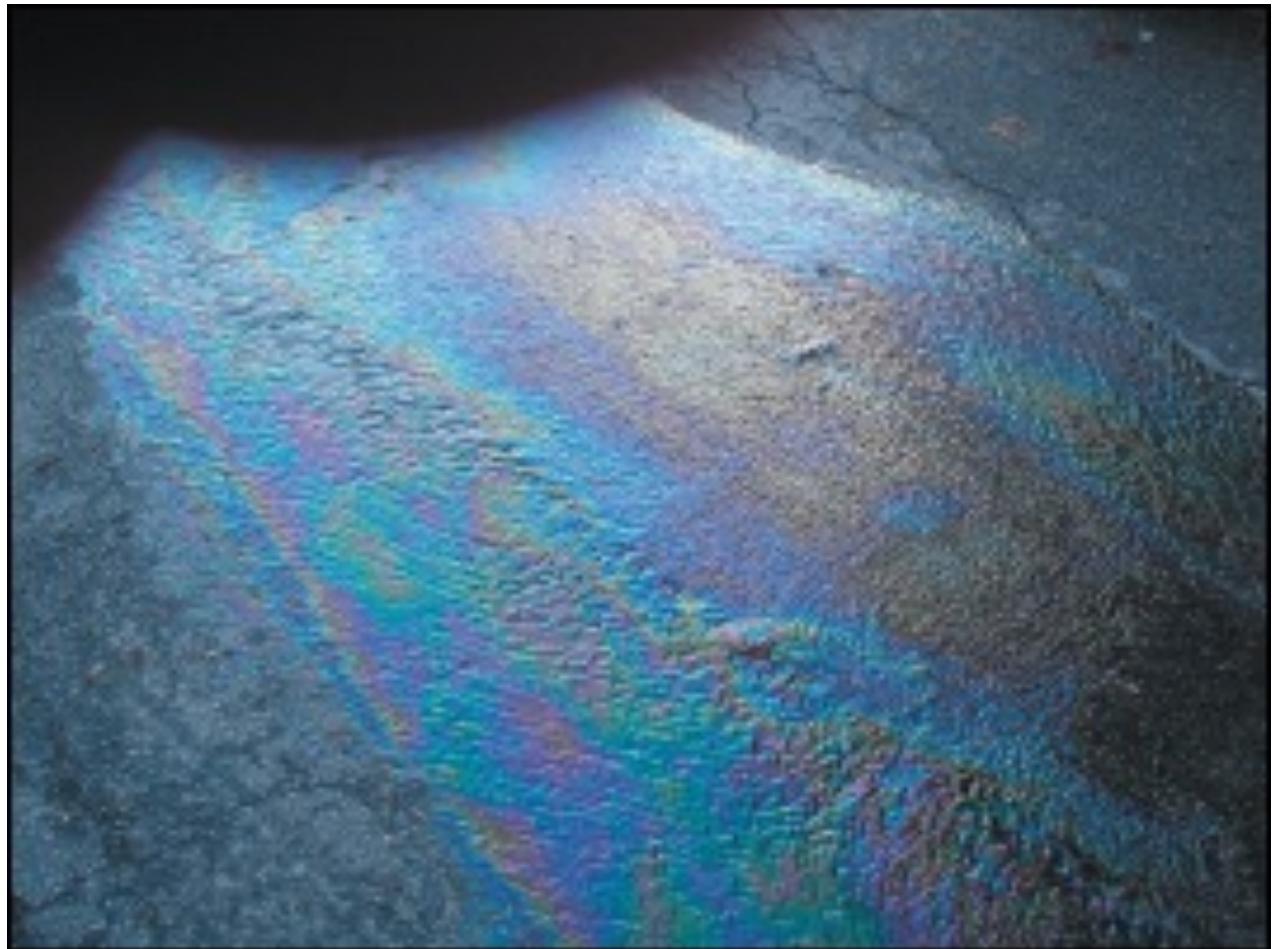


Units for Light

- Energy (J) depends on wavelength
- Power (W) is energy/unit time
- Irradiance (W/m^2) is power/unit area
 - Also known as radiosity
- Radiance ($W/m^2 sr$) is irradiance/unit angle
 - Conserved along a ray
 - Defers inverse-square law to surface

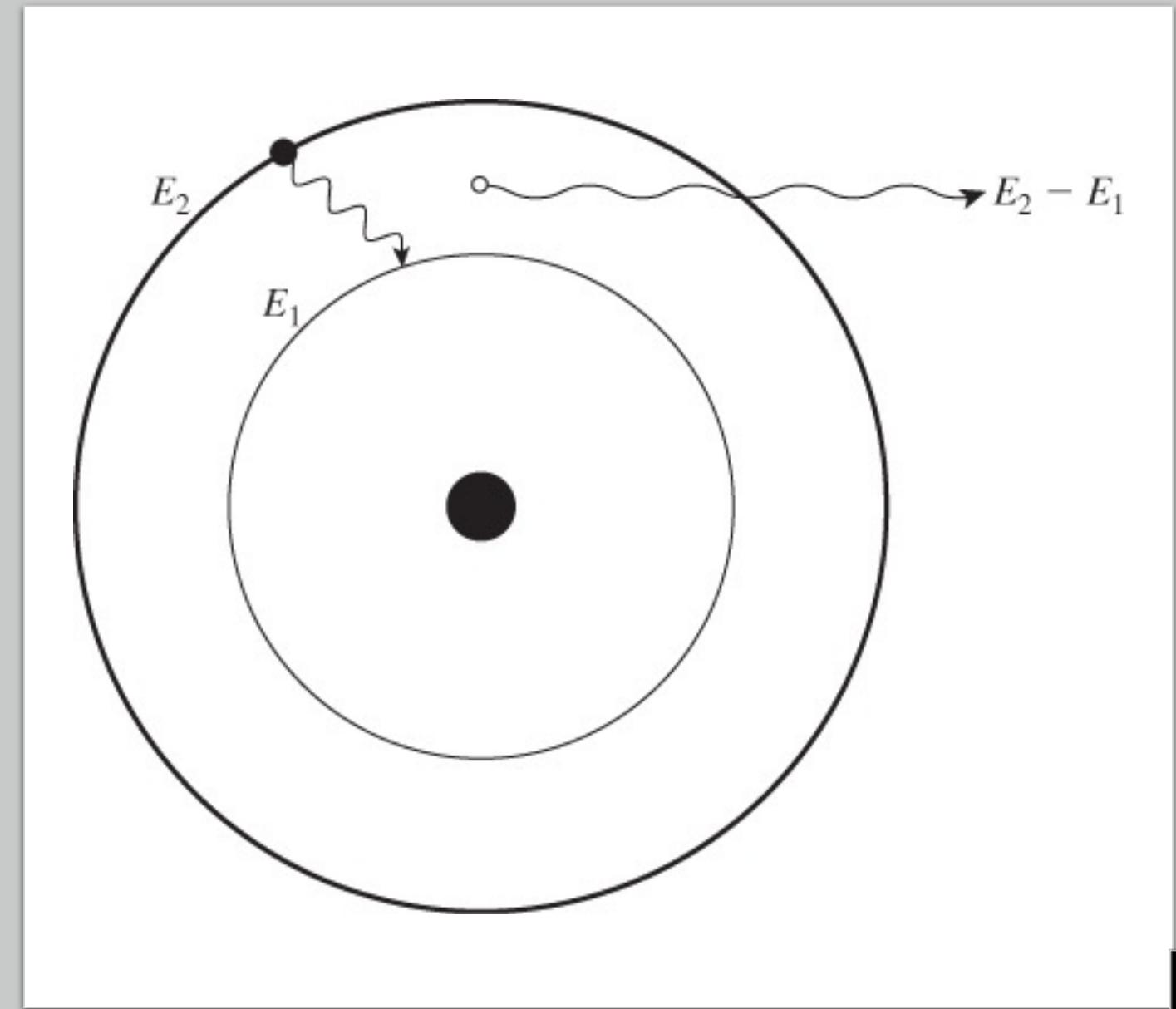
Why This Matters

- Thin films generate rainbows of colour
- Many other phenomena also affected



Atomic Orbitals

- Electrons orbit atoms
- Fixed radii called orbitals
 - Related to electron energy
- Electrons capture photons
 - Jump to higher orbitals
- Or they release photons
 - Drop to lower orbitals



Photon Frequencies

- Single atom – very strict orbitals
 - Photons have specific frequencies
- Blocks of material have roaming electrons
 - So they have bands of frequencies
 - But photons still restricted to these bands



Electron Effects

- Electrons may be tightly or loosely bound
 - Tight: photon released at same frequency
 - Loose: photon released at lower frequency
 - Spare energy absorbed as heat
 - Reflective material \Leftrightarrow conductors
 - Transparent material \Leftrightarrow insulators





Fluorescence

- Energy absorbed, transformed to heat
- Electron releases lower-energy photon
- Lower energy => lower frequency
- Lower frequency => different colour
- UV light can drop to visible light
 - Fluorescence

Time Delays

- Photon re-radiation is not immediate
- There is a time lag
- Small lag + re-emission of identical photon
 - Means the effective speed of light is slower
 - Different materials => different speeds
 - Known as virtual transition
- Large lag + energy loss
 - Phosphorescence

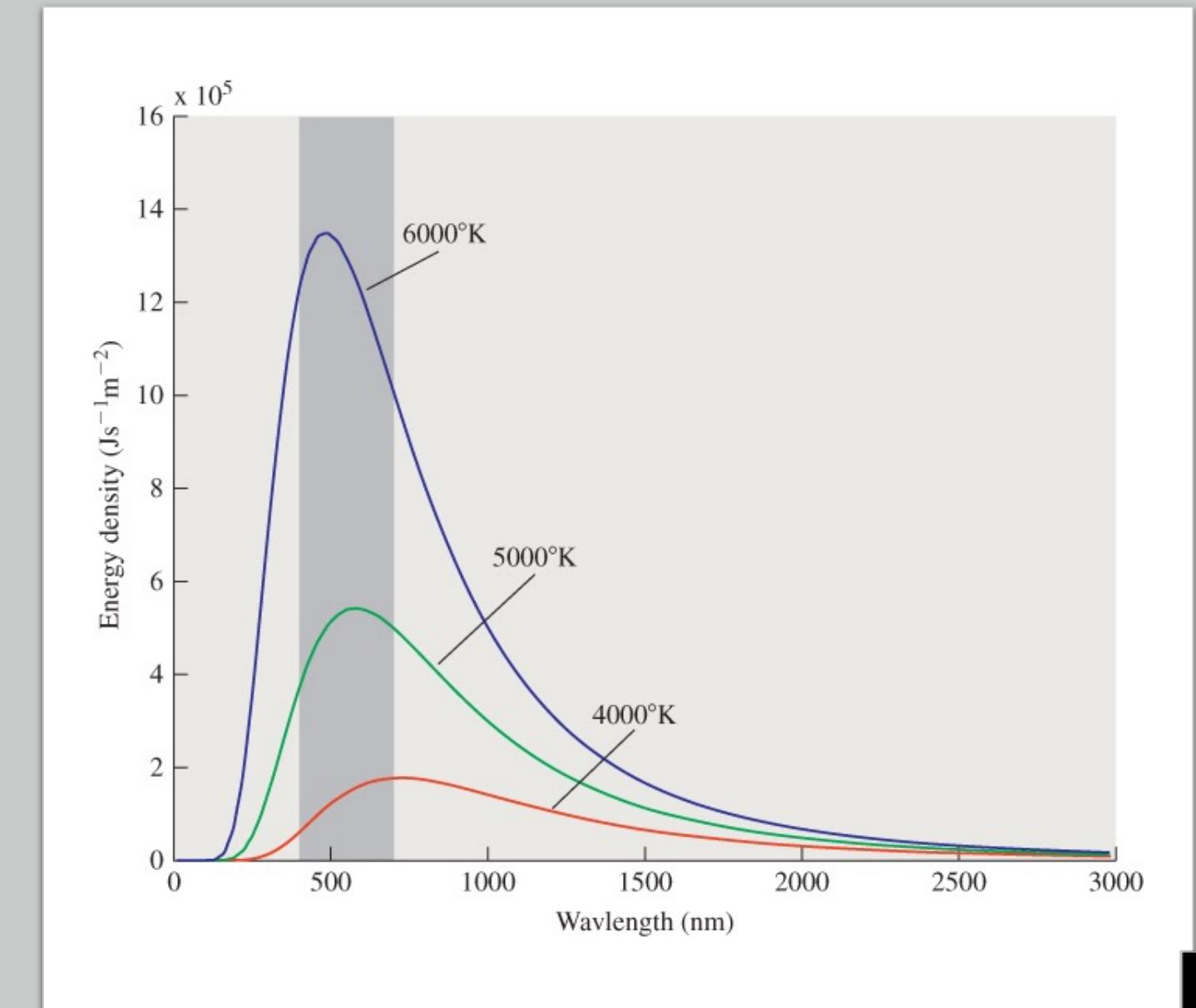
Carbon

- Carbon has loosely bound electrons
 - Can absorb photons of many frequencies
 - Store energy in vibrations (heat)
 - Dark materials warm up
- Reverse this to generate photons
 - Heat up soot so atoms vibrate
 - They start kicking out photons



Black-body Radiation

- Heat up a black body
- Radiates *lots* of photons
- Radiation depends on temperature
- Stefan-Boltzmann law
 - $Power = \sigma T^4$
 - $\sigma = 5.67 \times 10^{-8} \frac{W}{m^2 K^4}$



Wave Nature of Light

- Photons *also* behave like waves
 - wave length – distance between peaks
 - wave velocity – speed of peaks



Planar Waves

- Generalisation of sine waves to space
- Light wave has an associated electric field
- So, a wave along x-direction is:

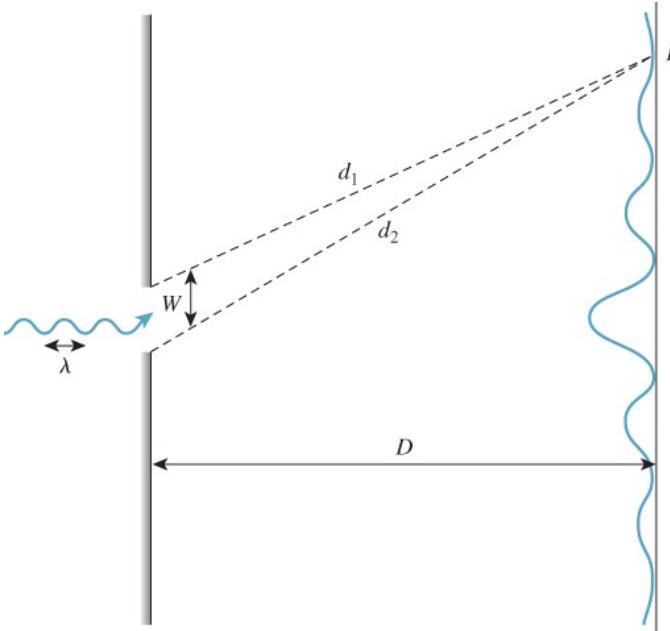
$$E_y(x, 0, 0, t) = A_y \sin\left(2\pi \frac{x}{\lambda} - 2\pi \frac{c}{\lambda} t + \Delta_y\right)$$

$$E_z(x, 0, 0, t) = A_z \sin\left(2\pi \frac{x}{\lambda} - 2\pi \frac{c}{\lambda} t + \Delta_z\right)$$

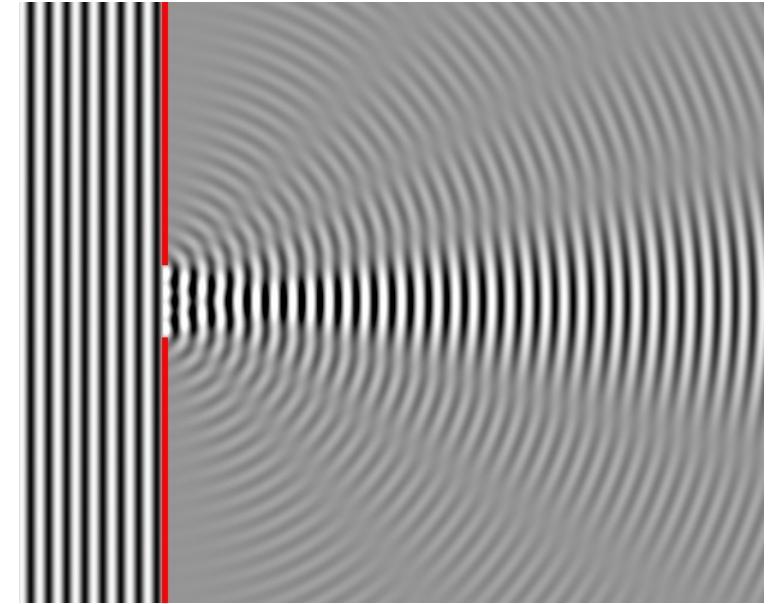
- All others are just rotations of this



Diffraction



At *one* distance D



At *all* distances

- Waves passing through a slit *diffract*
 - depending on width of slit



Polarisation

- Change the origin in x or t
 - $\Delta y, \Delta z$ will change
 - But their difference will not
 - This is the basis of polarisation

$$E_x(x, 0, 0, t) = 0$$

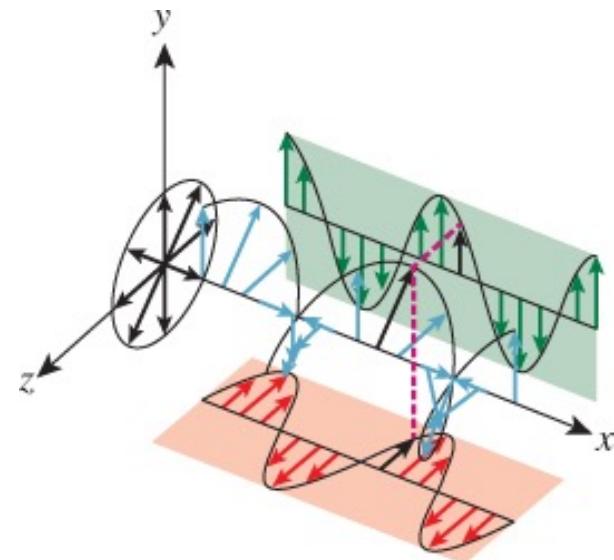
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$$E_z(x, 0, 0, t) = A_z \sin \left(2\pi \frac{x}{\lambda} - 2\pi \frac{c}{\lambda} t + \Delta_z \right)$$



Circular Polarisation

- y, z are out of phase
- Represents a helical light wave
- Projection sideways gives sine, cosine
- Projection forward gives a circle



Let $t = 0$

$$A_y = A_z$$

$$\Delta_y = 0$$

$$\Delta_z = \pi/2$$

Then

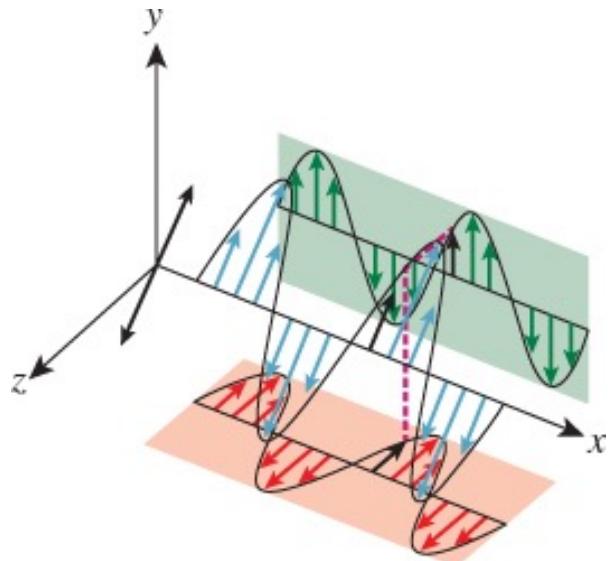
$$E_x(x, 0, 0, t) = 0$$

$$E_y(x, 0, 0, t) = A_y \sin\left(2\pi \frac{x}{\lambda}\right)$$

$$E_z(x, 0, 0, t) = A_z \cos\left(2\pi \frac{x}{\lambda}\right)$$

Linear Polarisation

- y, z are in phase
- Light wave is planar in nature
- Projects to a straight line
 - And can have any axis $(0 \ A_y \ A_z)^T$



Let $t = 0$

$$A_y = A_z$$

$$\Delta_y = 0$$

$$\Delta_z = 0$$

Then

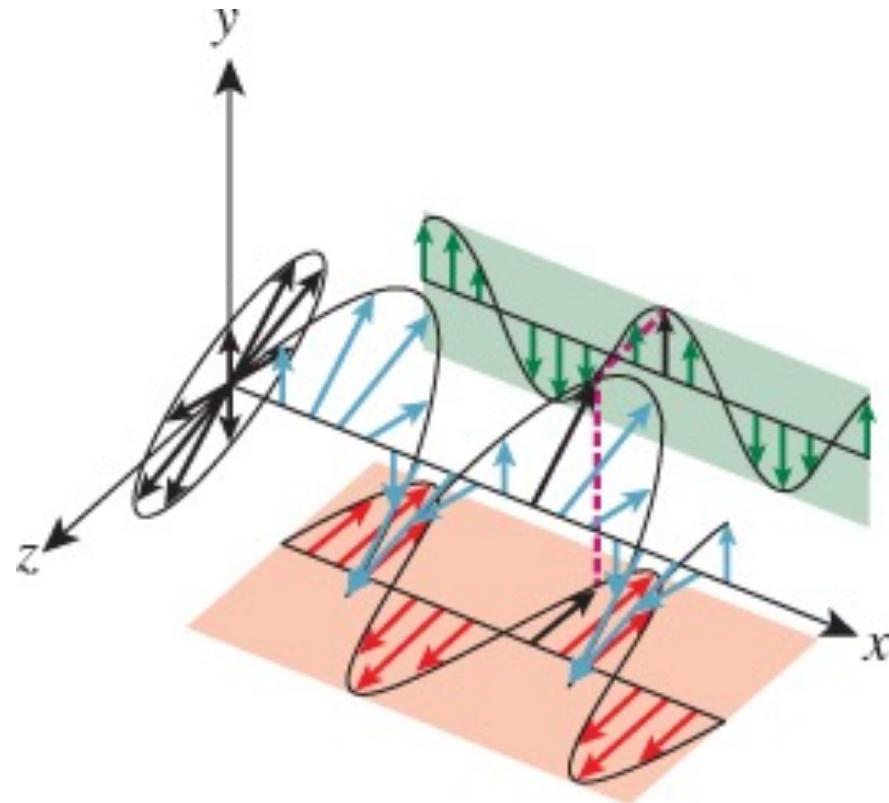
$$E_x(x, 0, 0, t) = 0$$

$$E_y(x, 0, 0, t) = A_y \sin\left(2\pi \frac{x}{\lambda}\right)$$

$$E_z(x, 0, 0, t) = A_z \sin\left(2\pi \frac{x}{\lambda}\right)$$

Elliptical Polarisation

- Any other case
 - a combination of circular and linear





Polarising Material

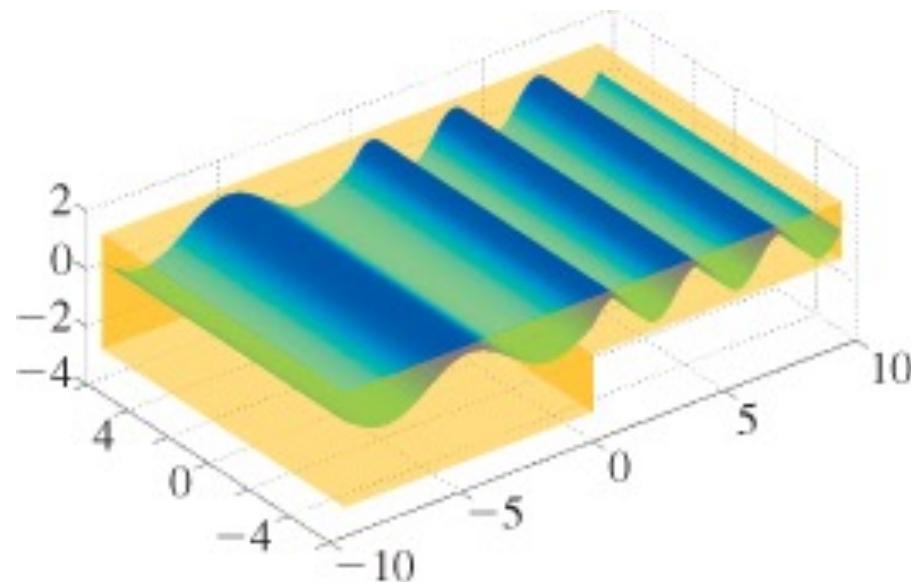
- Some material is polarising
 - Transparent to one polarisation
 - Opaque to opposite polarisation
- Commonly used for driving glasses
- Also used in LCD screens
- And in monitor glass
- More in a few minutes ...



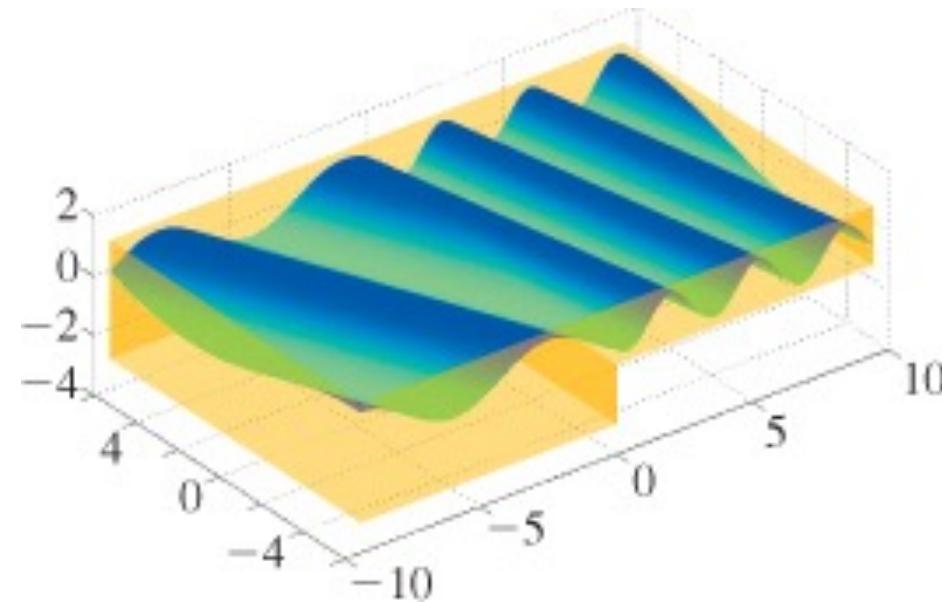
Refractive Index

- Recall that light speed depends on material
- The refractive index is the ratio with c
 - Vacuum: $s = 1.0000$
 - Air: $s = 1.0003$
 - Water: $s = 1.33$
 - Diamond: $s = 2.42$
- Actually depends on wavelength

Wave Retardation



Uniform



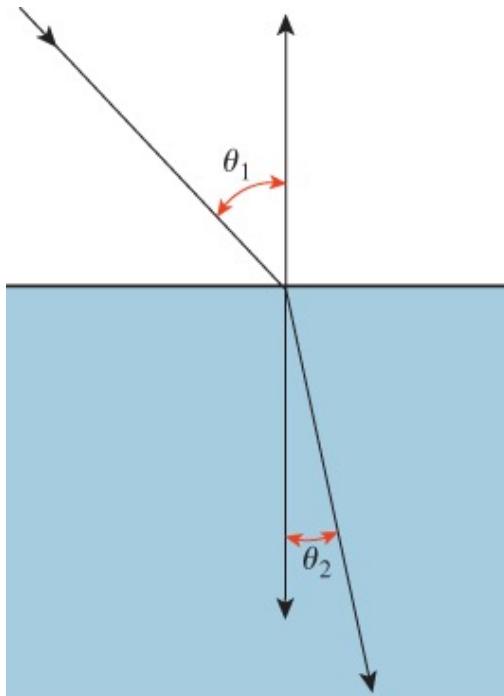
Slowed on One Side

- As waves slow down, they bunch up
- Frequency is constant, wavelength changes
- Slowing one side changes the direction



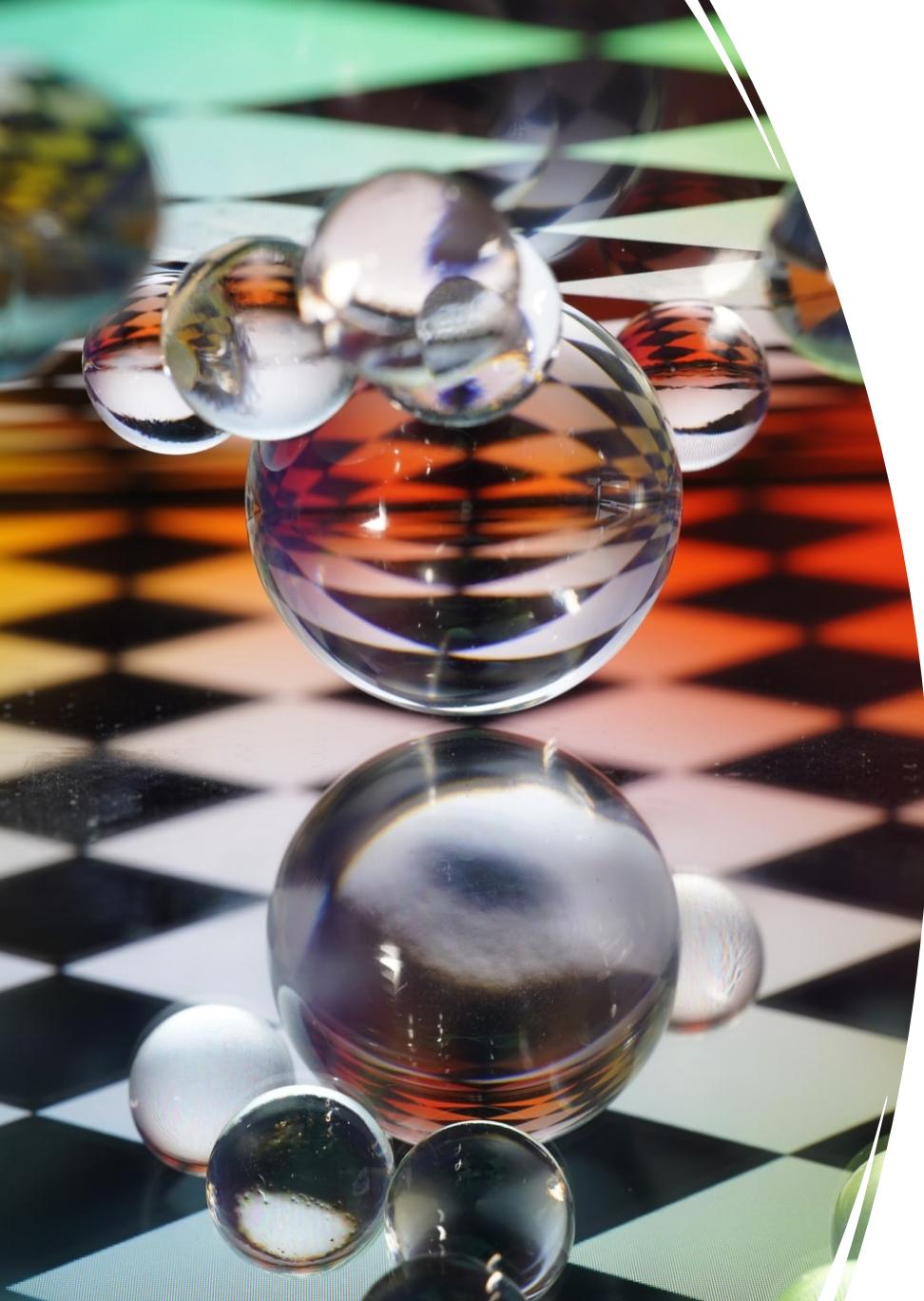
Snell's Law

- Waves retard based on angle
- So this rotates the wave
 - Based on indices of refraction



$$\frac{n_2}{n_1} = \frac{\sin \theta_1}{\sin \theta_2}$$



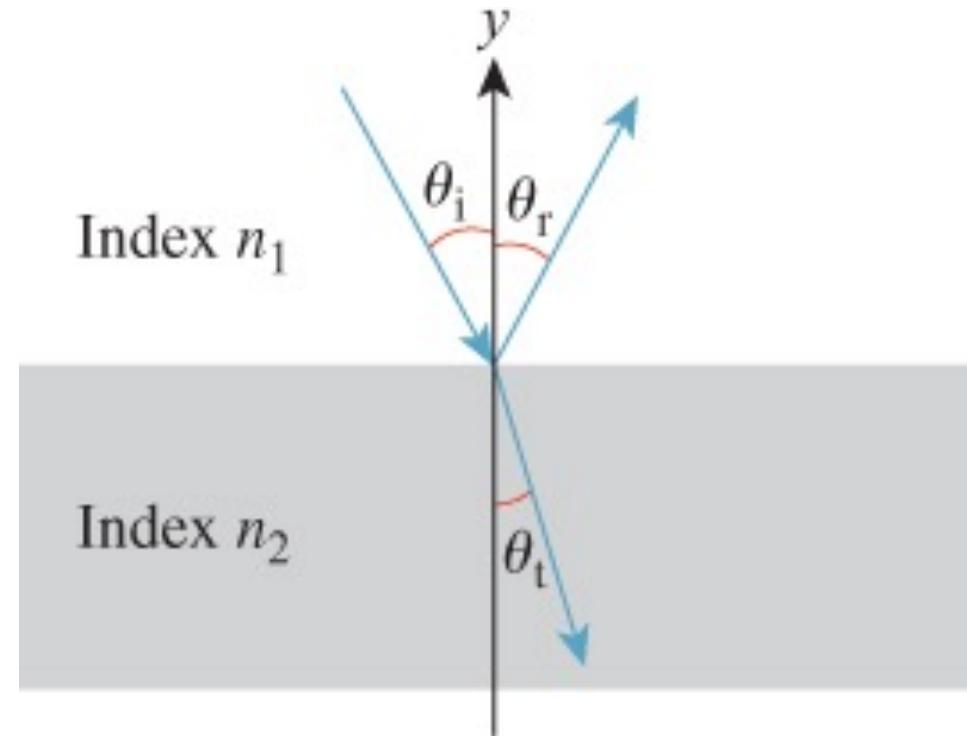


Reflection & Refraction

- Not all light *can* refract
- So energy must go somewhere
 - Reflection
 - Refraction
 - Absorption
- Governed by Fresnel equations

Polarised Reflection

- Light reflected gets polarised
 - p-polarisation:
 - parallel to surface
 - perpendicular to page
 - s-polarisation:
 - perpendicular to light
 - in plane of page

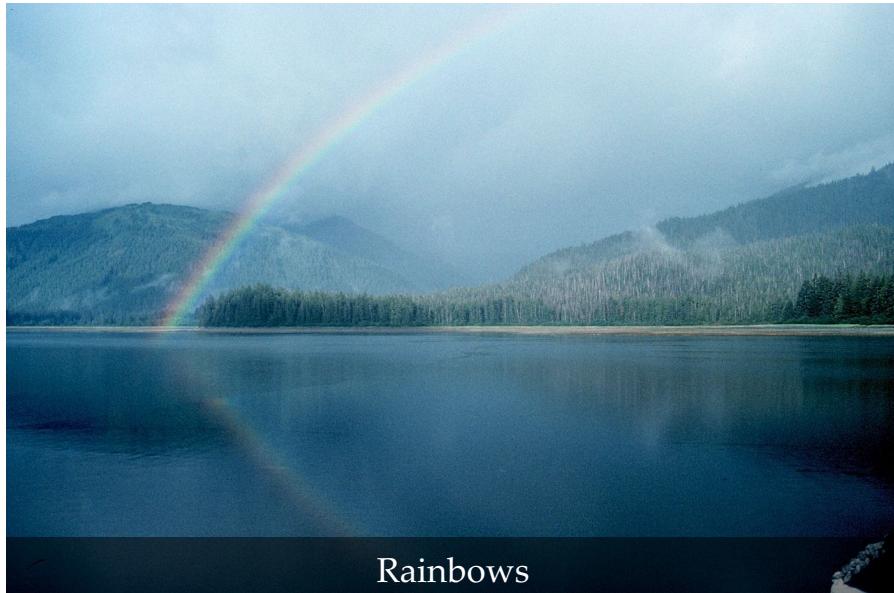


Total Internal Reflection

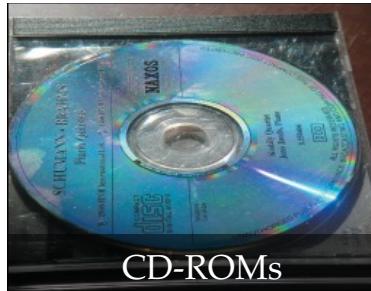
- If angle is too steep, light *cannot* transmit
- All light reflects internally
- Basis of lenses, &c.



Refractive Effects



Rainbows



CD-ROMs



Sundogs



Caustics





Conductors

- Insulators tend to refract light
- Conductors tend to absorb then release
- Rate of release affects reflected light
 - Can be expressed as complex number
 - Real part = index of refraction
 - Imaginary part = coefficient of extinction κ

Simpler Approach

- Treat refraction and extinction separately
- Fresnel reflectance then becomes

$$R_s = \frac{(n_2^2 + \kappa^2) \cos^2 \theta_i - 2n_2 \cos \theta_i + 1}{(n_2^2 + \kappa^2) \cos^2 \theta_i + 2n_2 \cos \theta_i + 1} \text{ and} \quad (26.14)$$

$$R_p = \frac{(n_2^2 + \kappa^2) - 2n_2 \cos \theta_i + \cos^2 \theta_i}{(n_2^2 + \kappa^2) + 2n_2 \cos \theta_i + \cos^2 \theta_i}, \quad (26.15)$$

- But this doesn't cover everything
 - Phenomena such as birefringence (refraction that depends on polarization)



Without Polarisation

- Graphics usually assumes no polarisation
- Makes computations much easier
- Simplifies Fresnel equations to:

$$L(P, \omega_r) = R_F L(P, -\omega_i) \text{ and} \quad (26.18)$$

$$L(P, \omega_t) = (1 - R_F) \frac{n_2^2}{n_1^2} L(P, -\omega_i). \quad (26.19)$$

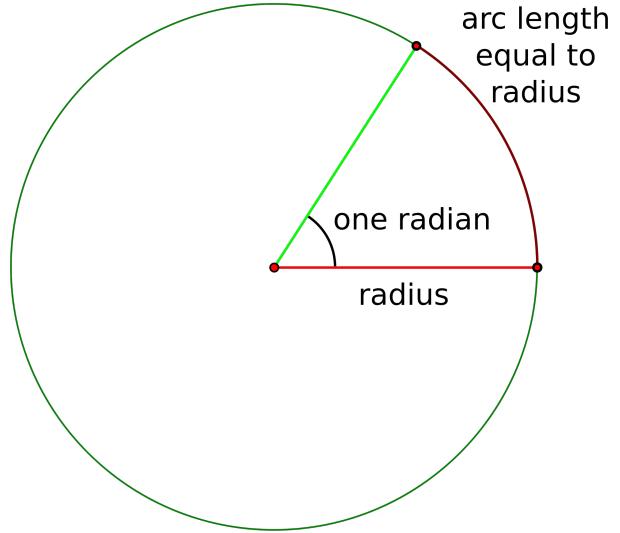


Moving to Calculus

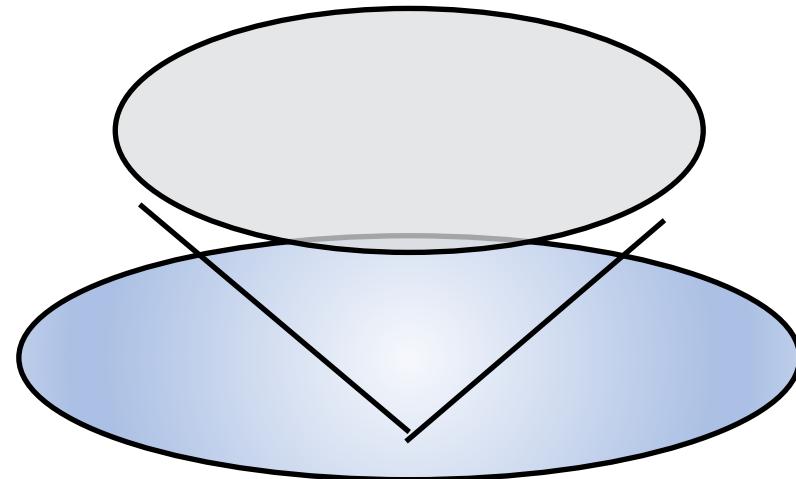
- All of this is per photon
- But there are many photons
 - At least 10^{23} in practice
- So we assume a continuous function
 - called a density function
- Provided that the integral comes out right

Solid Angles

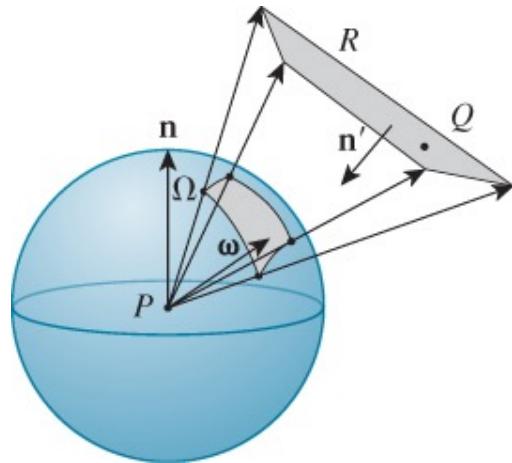
- A radian subtends a unit arc



- A steradian subtends a unit area



Change of Variables



$$A = \int_{\omega_i \in \Omega} g(\omega_i) \omega_i \cdot \mathbf{n} d\omega \quad (26.30)$$

becomes

$$A = \int_{Q \in R} g(N(Q)) N(Q) \cdot \mathbf{n} |JN(Q)| dQ, \quad (26.31)$$

becomes

$$A = \int_{s=0}^w \int_{t=0}^h g(\omega(s, t)) \frac{|\omega(s, t) \cdot \mathbf{n}| |\omega(s, t) \cdot \mathbf{n}'|}{\|Q(s, t) - P\|^2} dt ds. \quad (26.37)$$

- We often have rectangular patches
- We want to integrate in steradians
- This involves a change of variables



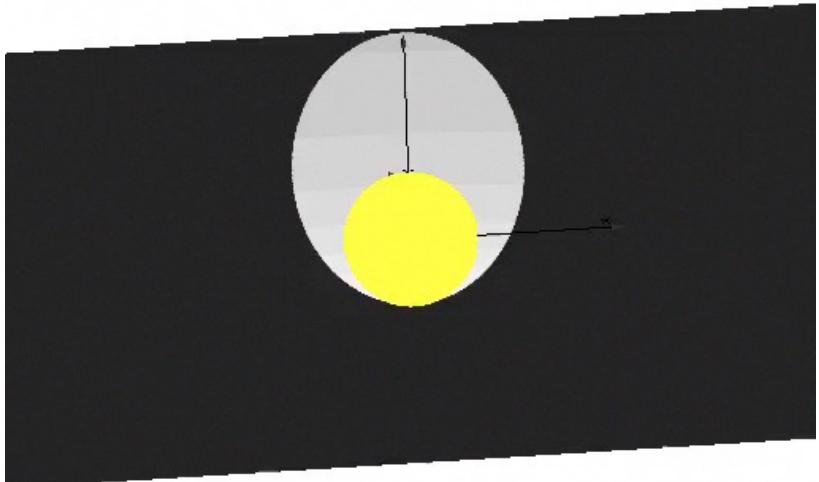
Integrating Radiance

- Energy is an integral of radiance
 - Over wavelengths from λ_0 to λ_1
 - Over all incoming directions ω
 - Over a given pixel/patch $[x_0, x_1] \times [y_0, y_1]$
 - And over a time interval $[t_0, t_1]$

$$energy \approx \int_{t_0}^{t_1} \int_{x_0}^{x_1} \int_{y_0}^{y_1} \int_{\omega \in \Omega} \int_{\lambda_0}^{\lambda_1} L(t, (x, y, 0), -\omega, \lambda) d\lambda d\omega dy dx dt. \quad (26.39)$$



But . . .



$$energy \approx \int_{t_0}^{t_1} \int_{x_0}^{x_1} \int_{y_0}^{y_1} \int_{\omega \in \Omega} \int_{\lambda_0}^{\lambda_1} L(t, (x, y, 0), -\omega, \lambda) d\lambda d\omega dy dx dt. \quad (26.39)$$

becomes

$$energy = \int_{t_0}^{t_1} \int_{x_0}^{x_1} \int_{y_0}^{y_1} \int_{\omega \in \Omega} \int_{\lambda_0}^{\lambda_1} L(t, (x, y, 0), -\omega, \lambda) \omega \cdot \mathbf{e}_3 d\lambda d\omega dy dx dt. \quad (26.40)$$

becomes

$$energy = \int_{t_0}^{t_1} \int_{\lambda_0}^{\lambda_1} \int_{P \in R} \int_{\omega \in \Omega} L(t, P, -\omega, \lambda) |\omega \cdot \mathbf{n}| d\omega dP d\lambda dt, \quad (26.41)$$

- The amount of light depends on the angle ω
 - Dependence is as with diffuse Phong
 - On the dot product with the angle

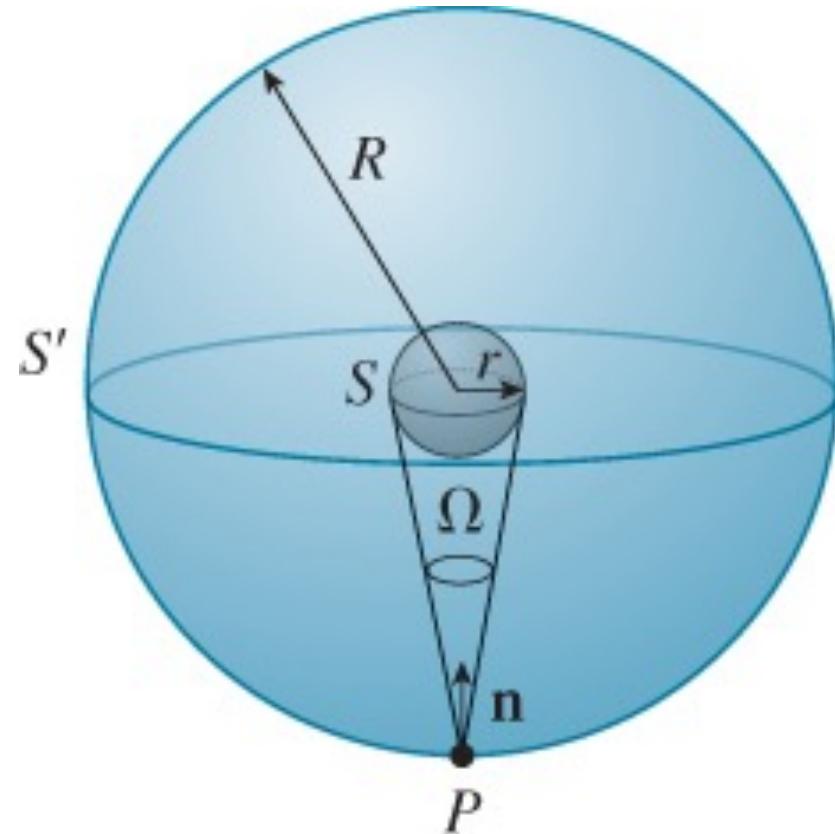


A Lambertian Emitter

- As P gets further from sphere S
 - Angle $\omega \cdot \vec{n} \rightarrow 1$ and

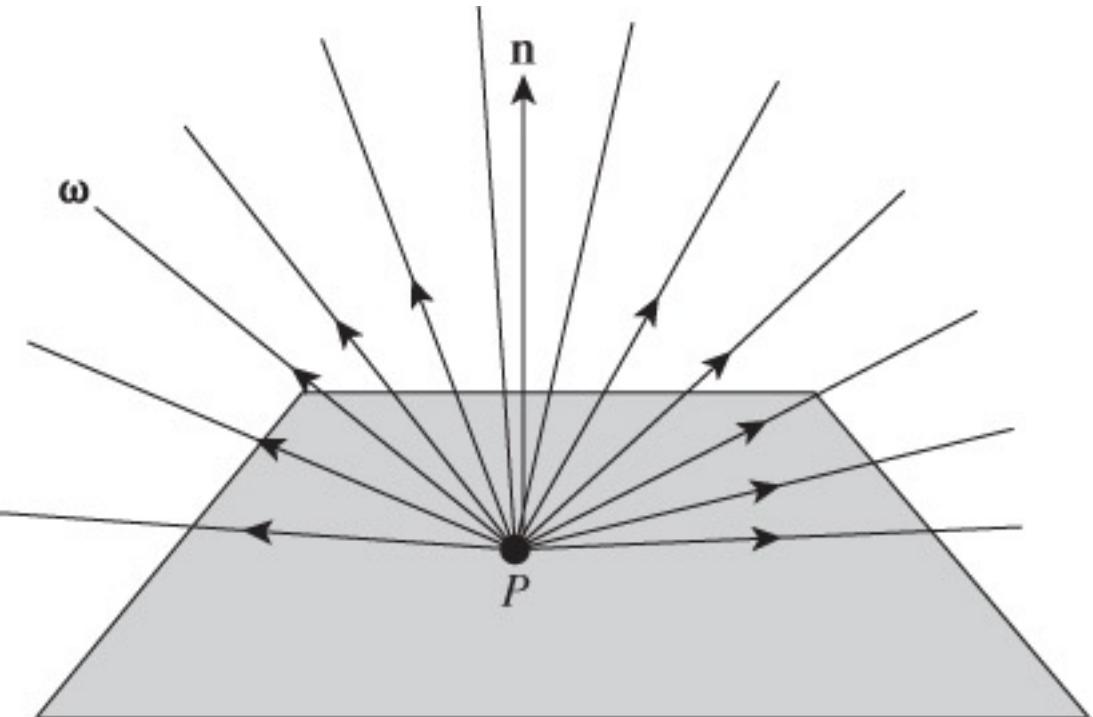
$$D = L \int 1 d\omega = L m(\Omega)$$

- Substitute $m(\Omega) = \frac{\pi r^2}{R^2}$ and
get $L = \frac{\Phi}{4\pi(\pi r^2)}$



Irradiance

- Similarly, we can simplify the irradiance
- By substituting assumptions



$$E(t, P, \mathbf{n}, \lambda) = \int_{\{\omega : \omega_i \cdot \mathbf{n} \geq 0\}} L(t, P, -\omega_i, \lambda) \omega_i \cdot \mathbf{n} d\omega_i. \quad (26.64)$$

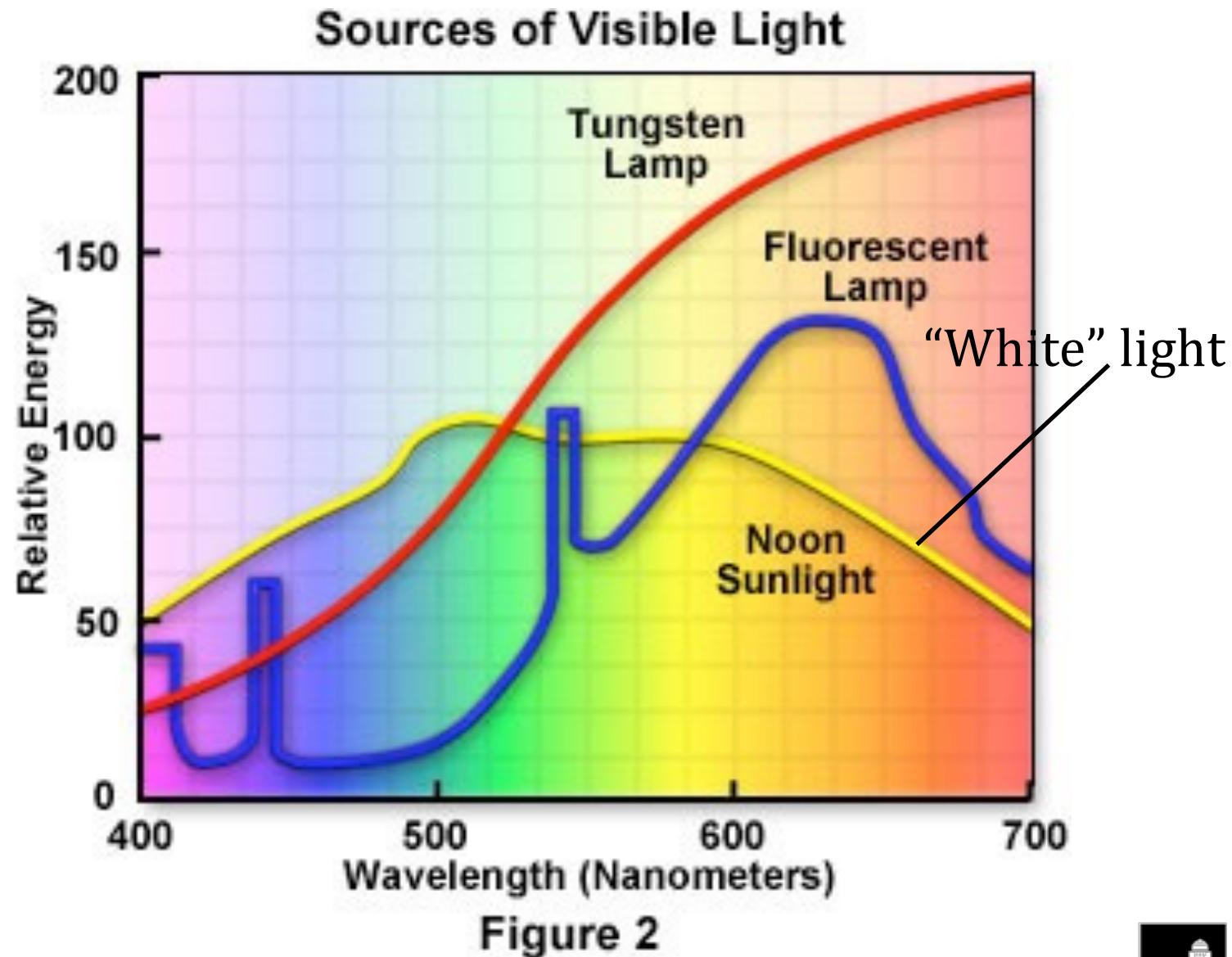
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Types of Light Source

- Sunlight (aka white light)
 - Black-body radiation at c. 6000°C
- Incandescent Bulb (Tungsten)
 - Black-body radiation at c. 2500°C
- Fluorescent Lamp (Mercury)
 - Stimulated emission with phosphorescence
- LED
 - Direct stimulated emission

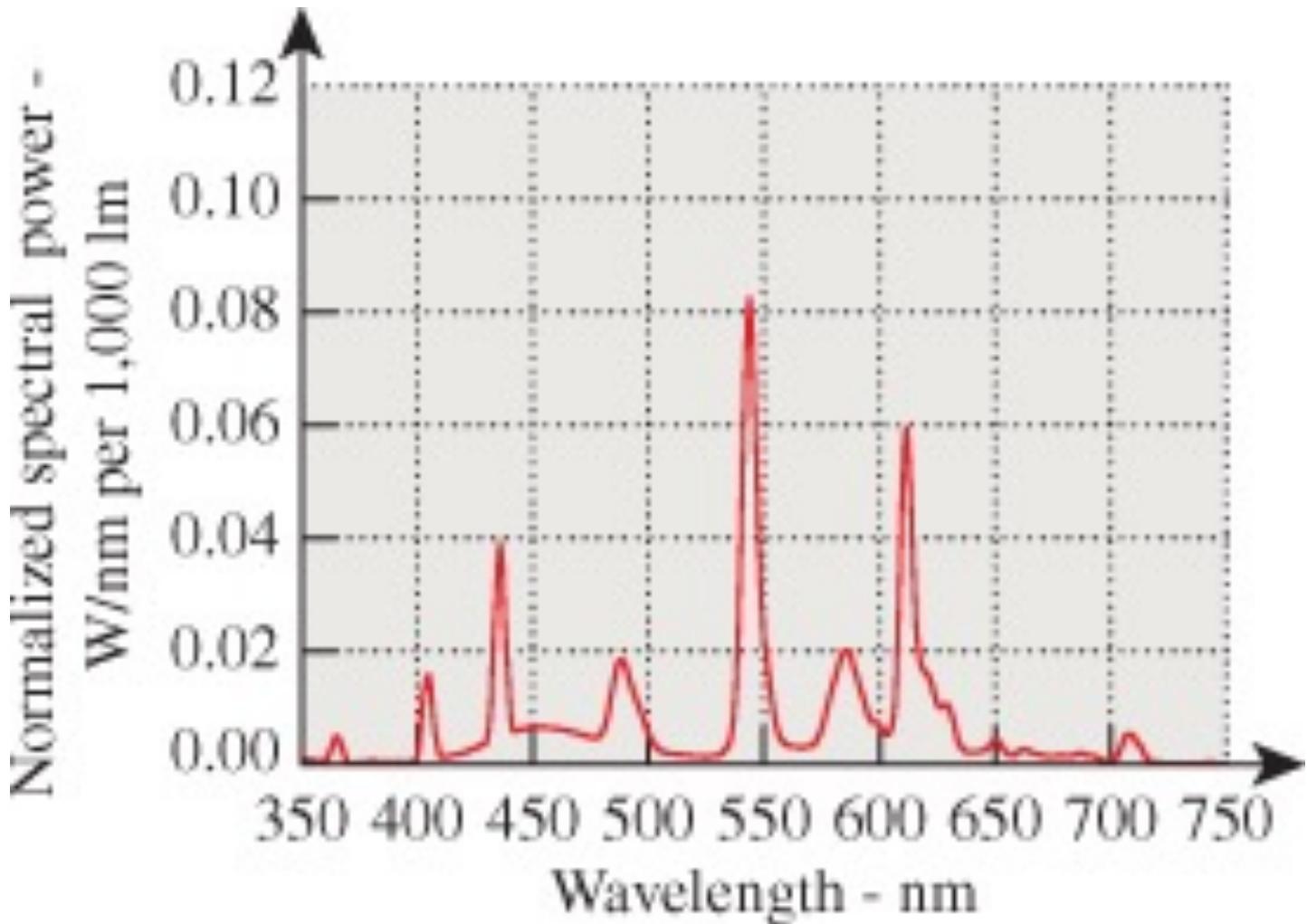


Spectral Distribution



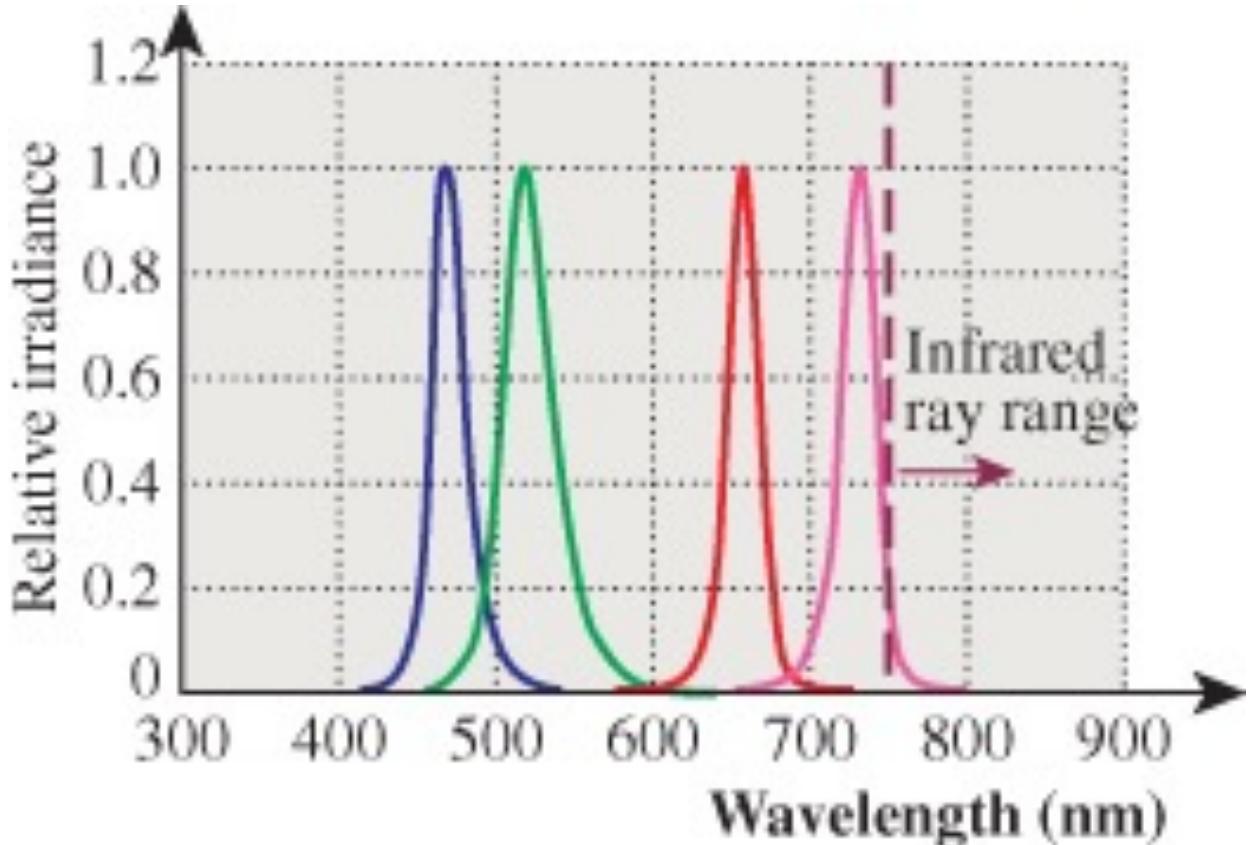
A Fluorescent Lamp

- Note the spikes (impulses)
- Phosphors chosen so that *result* looks white
 - i.e. perceptually close to sunlight



LED Lights

- LEDs are light-emitting diodes
- Close to *mono-spectral*
 - i.e. output only one wavelength

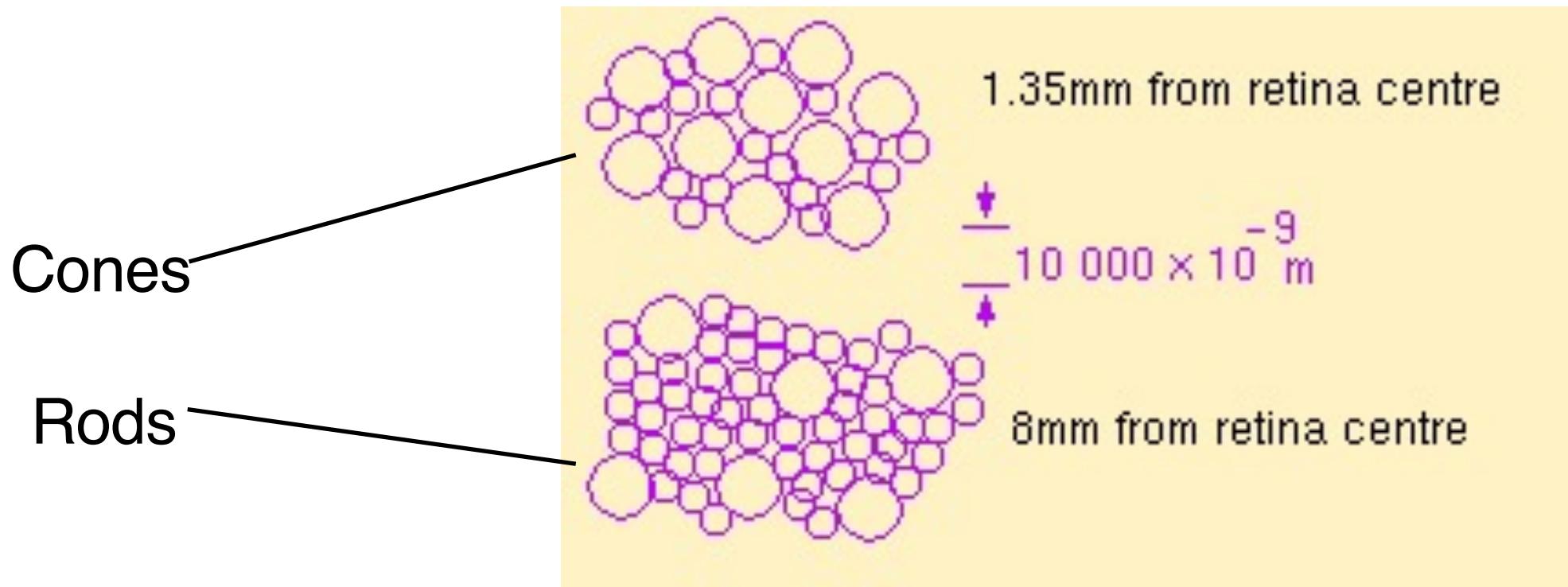


— Blue (470 nm)	— Red (660 nm)
— Green (525 nm)	— FarRed (735 nm)

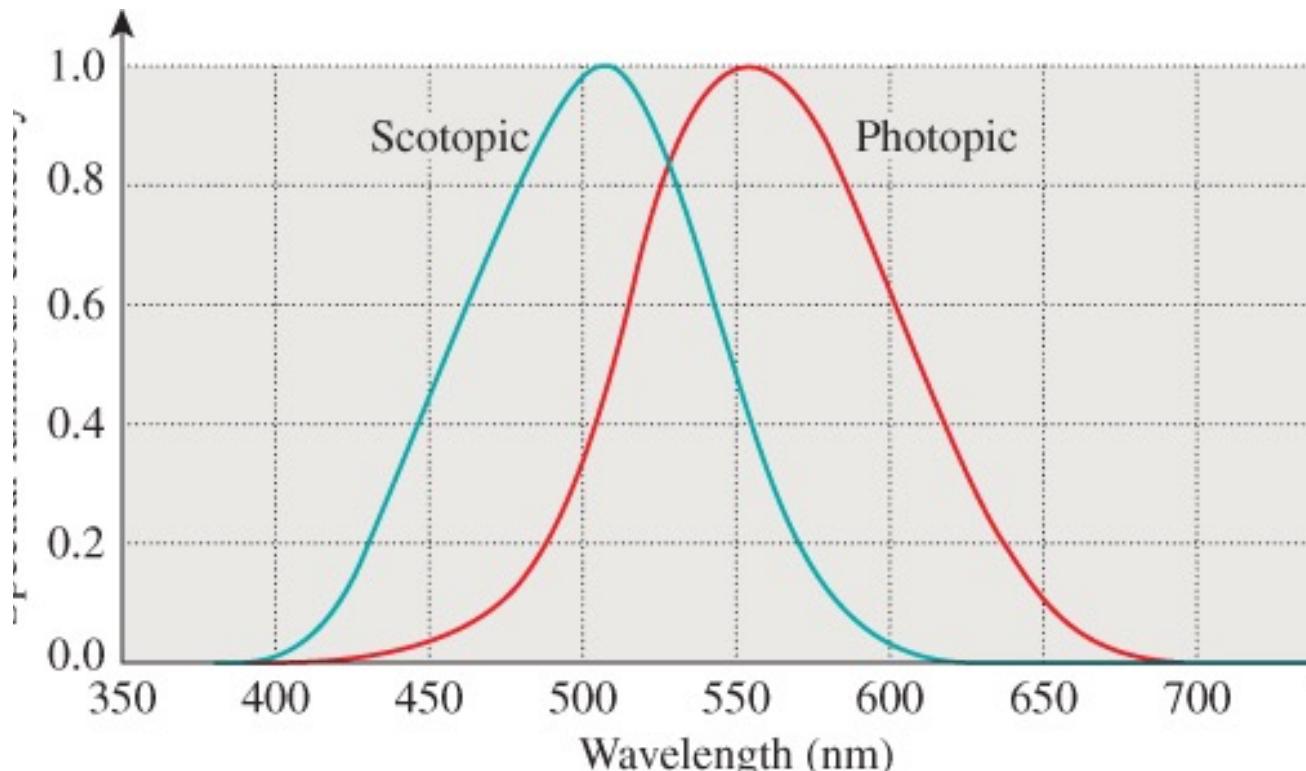


The Retina

- A patchwork of light-sensitive cells
 - Rods: low light conditions (B/W)
 - Cones: ordinary conditions (colour)

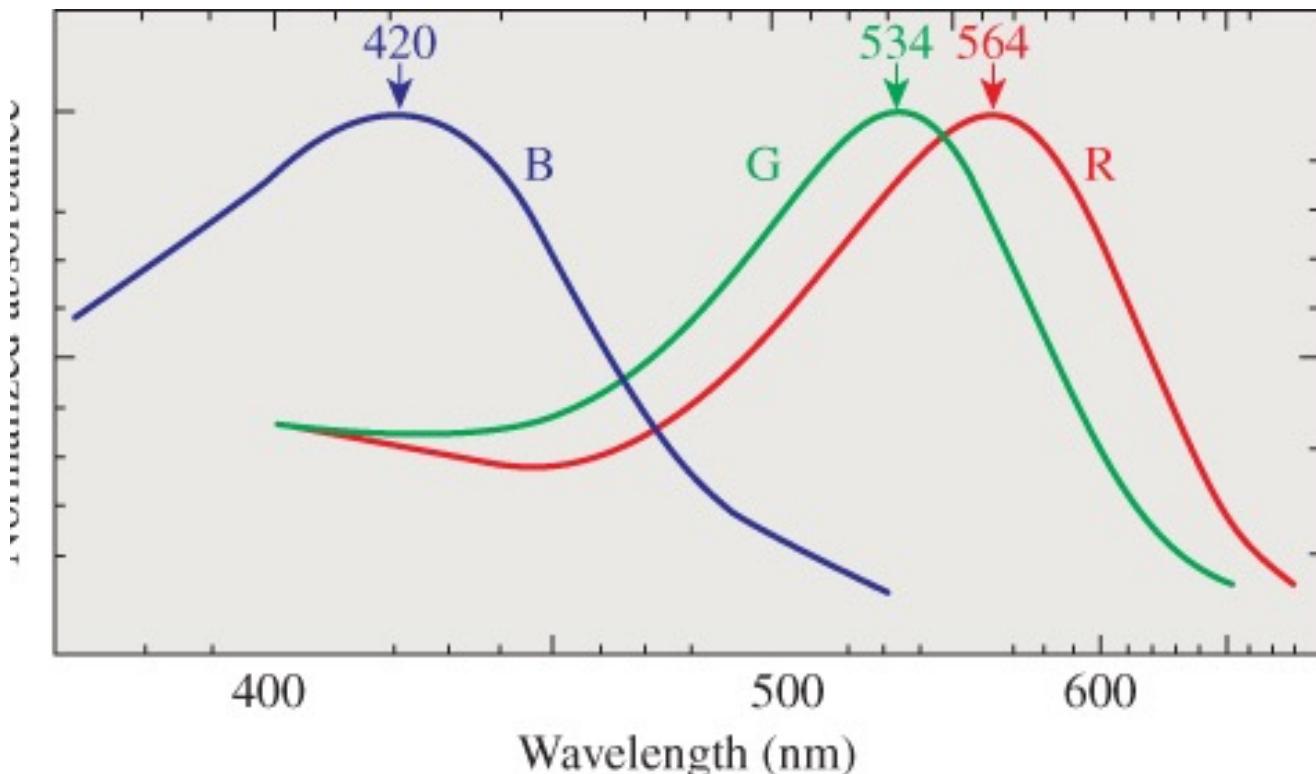


Photopic & Scotopic Vision



- Photopic vision uses the cones
- Scotopic (night) vision uses the rods
- And their efficiency is different
 - Scotopic vision has trouble with red

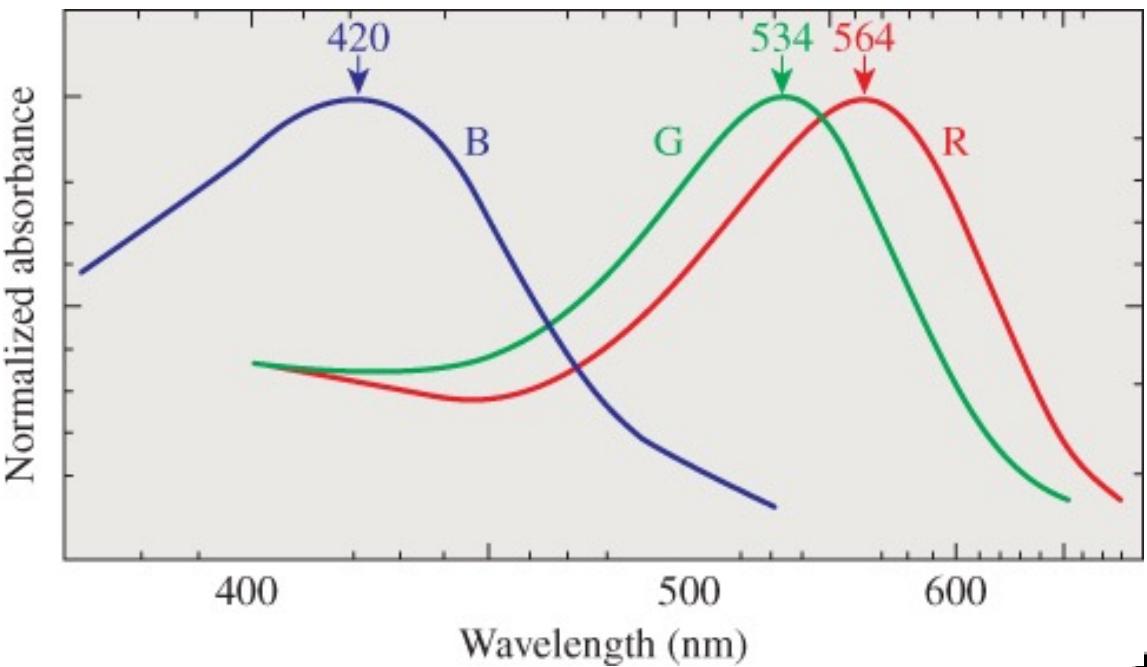
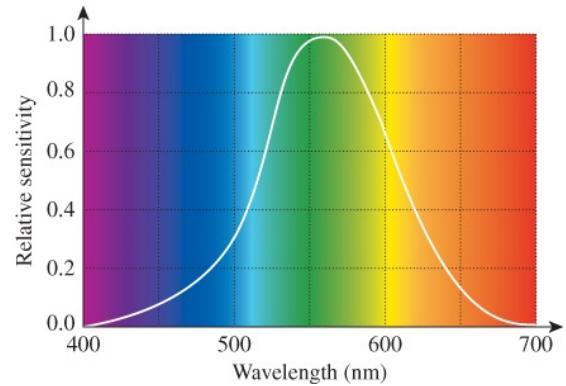
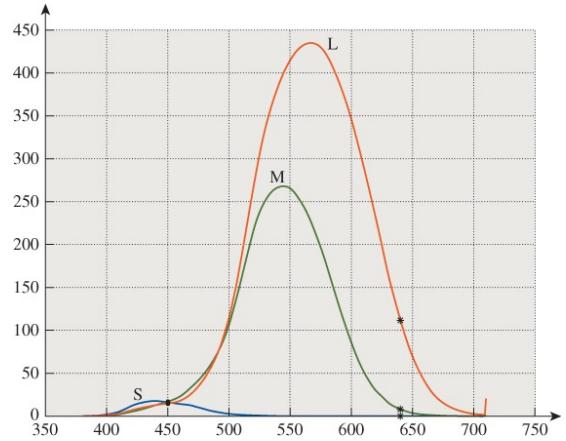
Cones & Colour



- Three types of cones
- Respond to different wavelengths
 - Integrate over a *range* of wavelengths

Luminous Efficiency

- The eye works best with greenish light
- Other wavelengths fall off in response
- So blue cones are overall weak

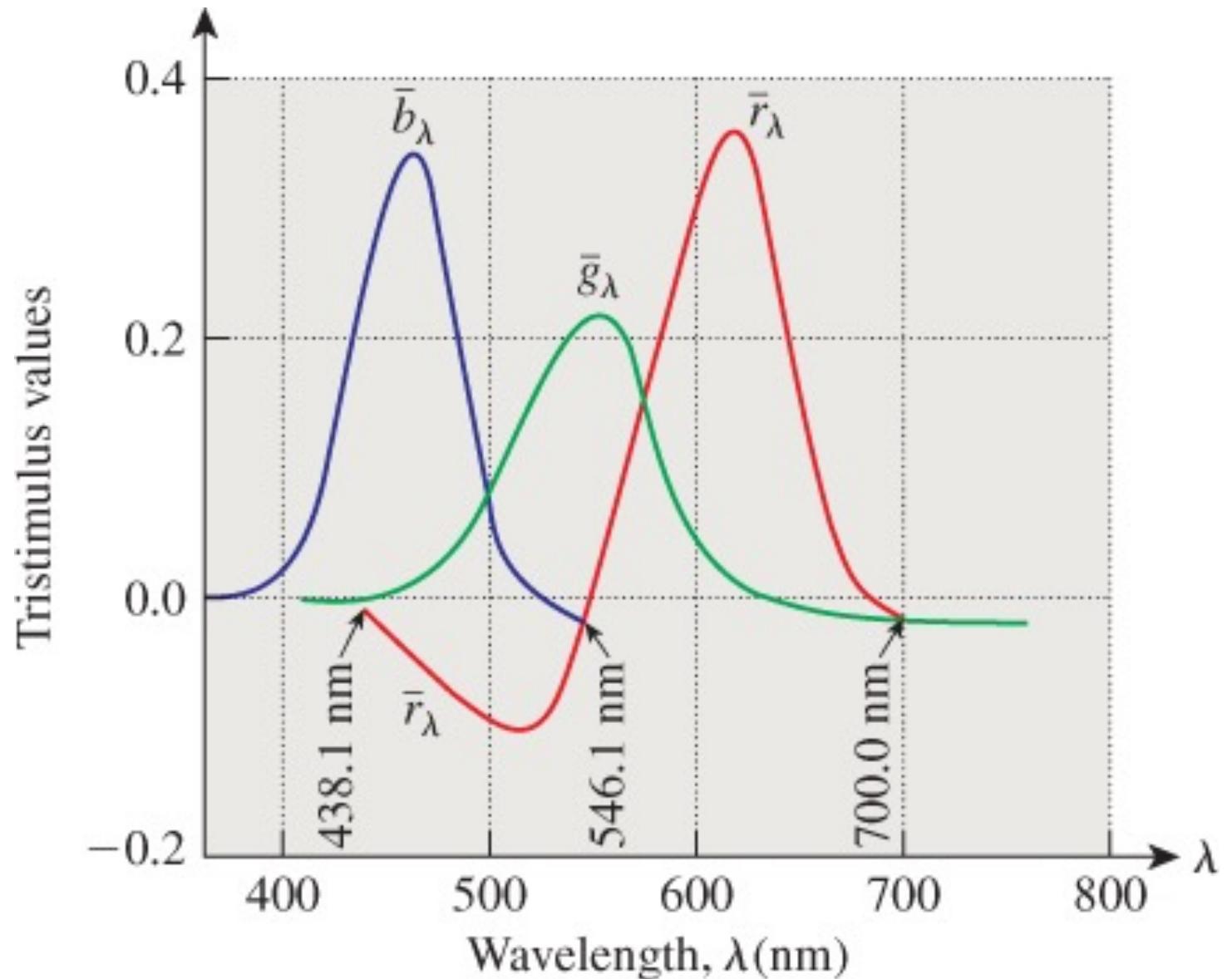


Luminance

- Perceived brightness
- Integrate total energy over all wavelengths
 - $\int I(\lambda)e(\lambda)d\lambda$
- Notice we've now included human response
 - So this is based on perceptual studies
 - Averages over a population

Tri-stimulus Theory

- Mix R, G & B to get any colour desired
- Human can't tell the difference
- But you need *negative* red light to do it



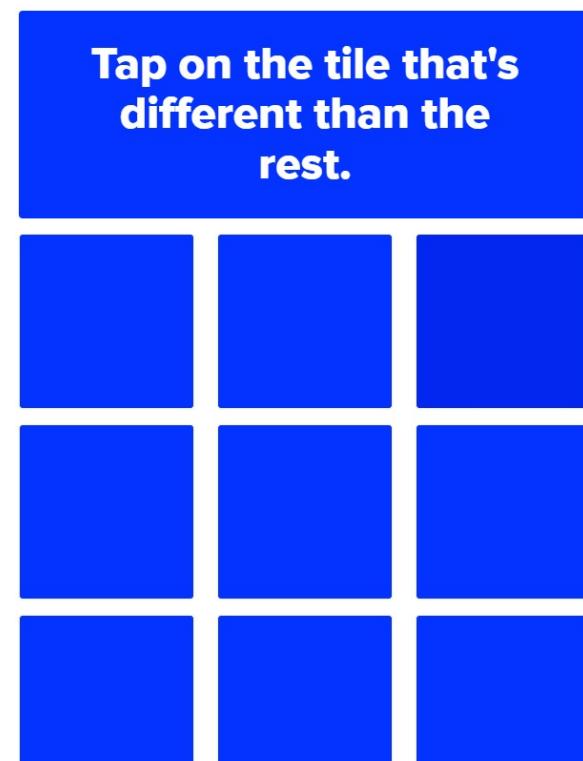
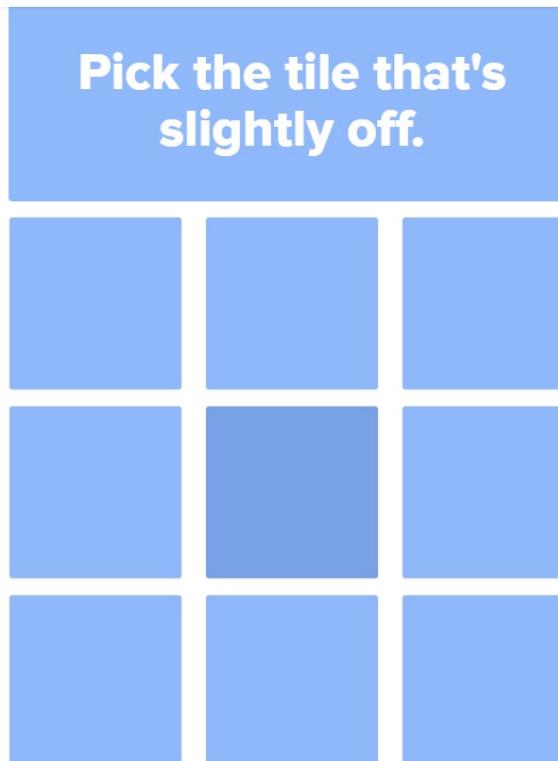
Adaptation

- The eyes use slow chemical reactions
- But they adapt to the light level
- So their response is (mostly) logarithmic
- But too much light washes an image out
 - Even in the areas where it's dark



Just Noticeable Difference

- Humans can just spot a 1% difference
 - at least in the range that displays use
- So we need about 100 logarithmic levels
- And 256 values is enough most of the time
- Hence our 8-bit RGB values
- But they're not uniform
 - So we use gamma correction



Radiance to Pixel Value

- OpenGL/PPM use a simple model:
 - 0-255 / 0.0 – 1.0 in each colour channel
- Human vision isn't like this
 - We see a wide range of intensity
 - Closer to 0.0 – 1,000,000
 - But it's logarithmic in nature
 - And we have to adjust for it

Gamma encoding

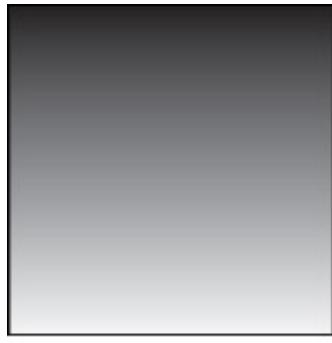
- Assume we have
 - radiance r
 - min & max *possible* r
 - $\Delta r = (\text{max} - \text{min}) / 1.0$
- Then we compute

$$r' = [(r * \Delta r)^{1/\gamma}] * 255.0$$

- With $\gamma = 2.2$ (to match the human eye)



Gamma Examples



Linear brightness



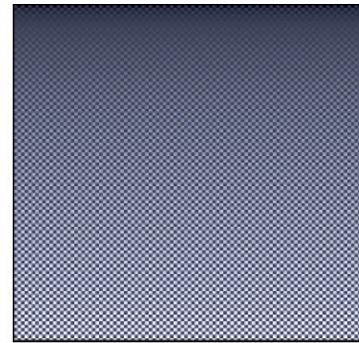
Linear radiance

$$\gamma = 2.0$$



(a)

$$\gamma = 1.0$$



(b)

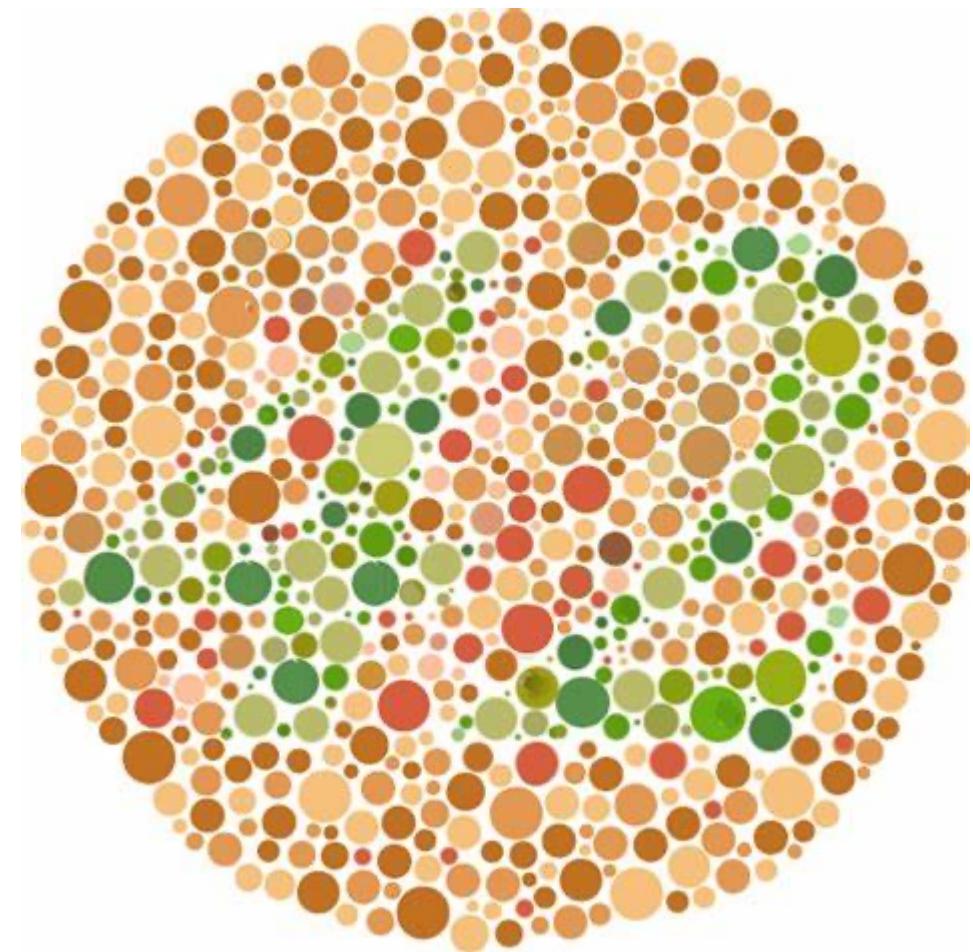
$$\gamma = 2.0$$

- (subject to monitor gamma, &c.)
- Compute in float
- Use gamma to convert to integers

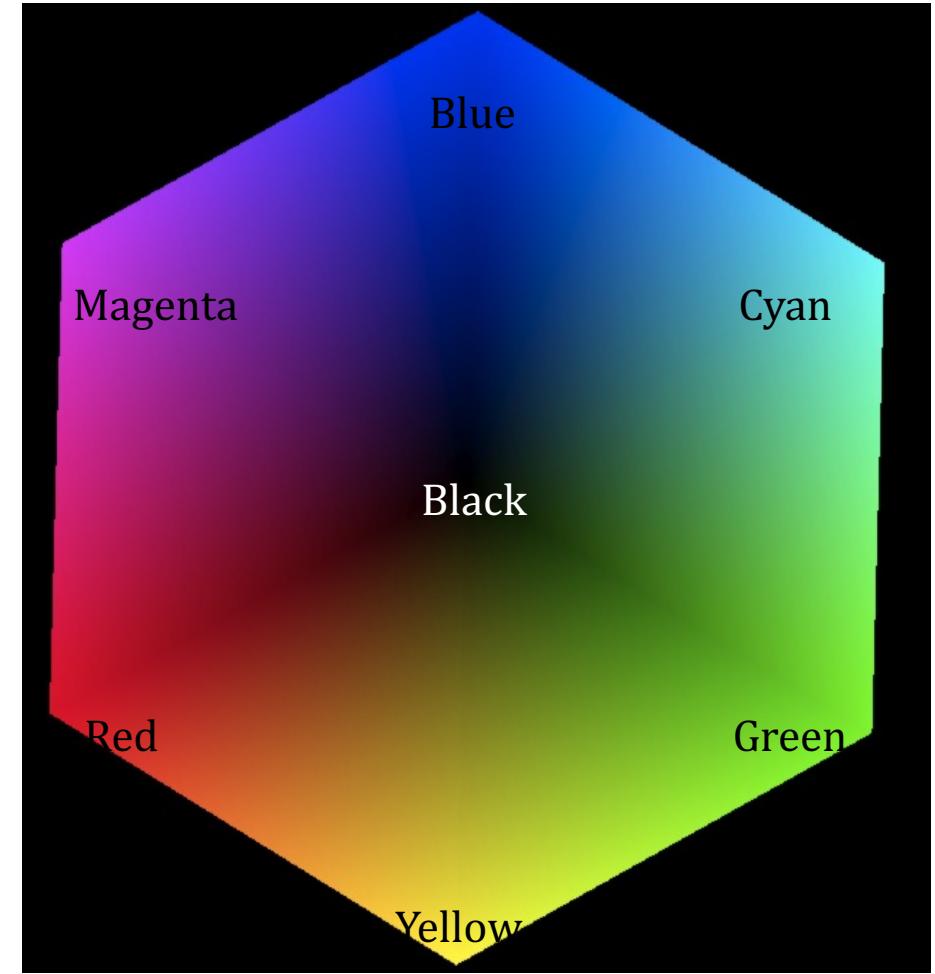


Colour Blindness

- About 8-10% of men, 1% of women
- Most common is red-green
 - only one cone for both red & green
- Yellow-blue also occurs
- And some people only have one type of cone
- Remember to design for colour blindness
 - NEVER rely only on colour

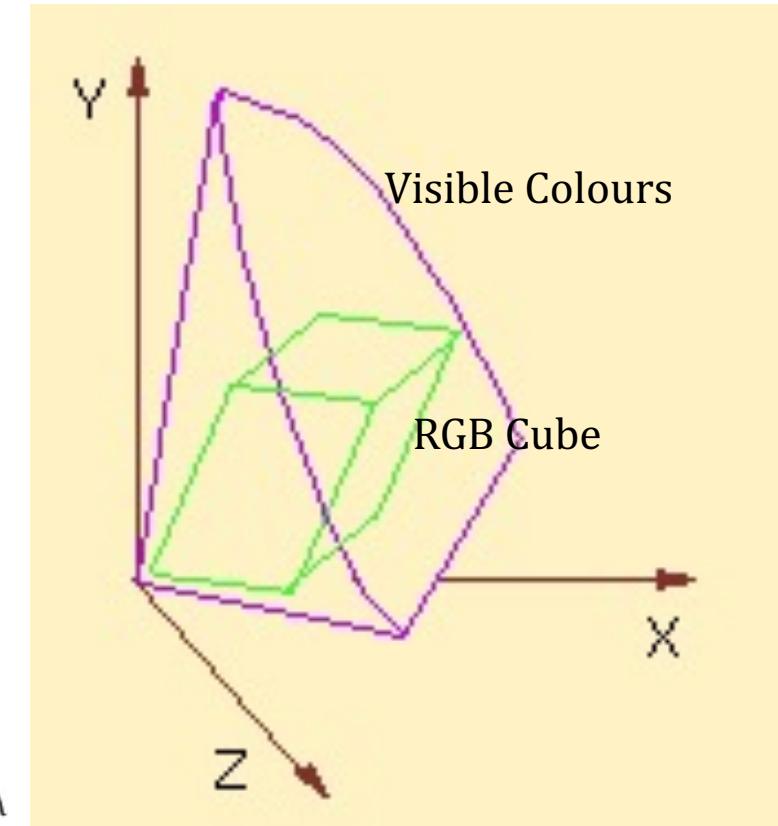
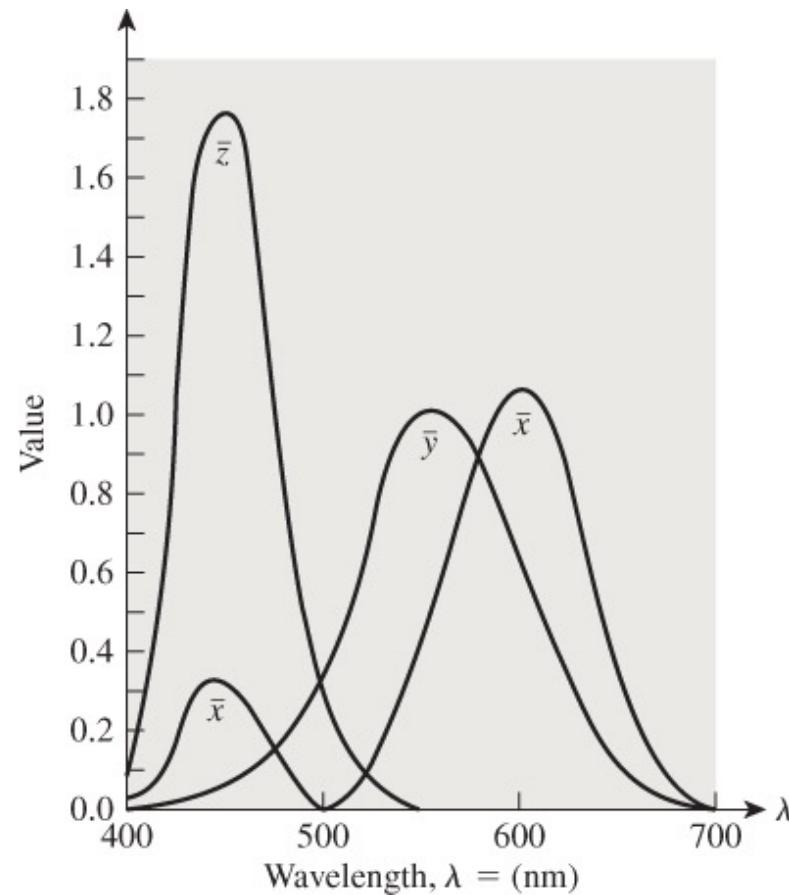


RGB Colour Cube



XYZ Colour Space

- By the CIE (Commission Internationale d'Eclairage)
- Every visible colour uses positive coords



RGB to XYZ

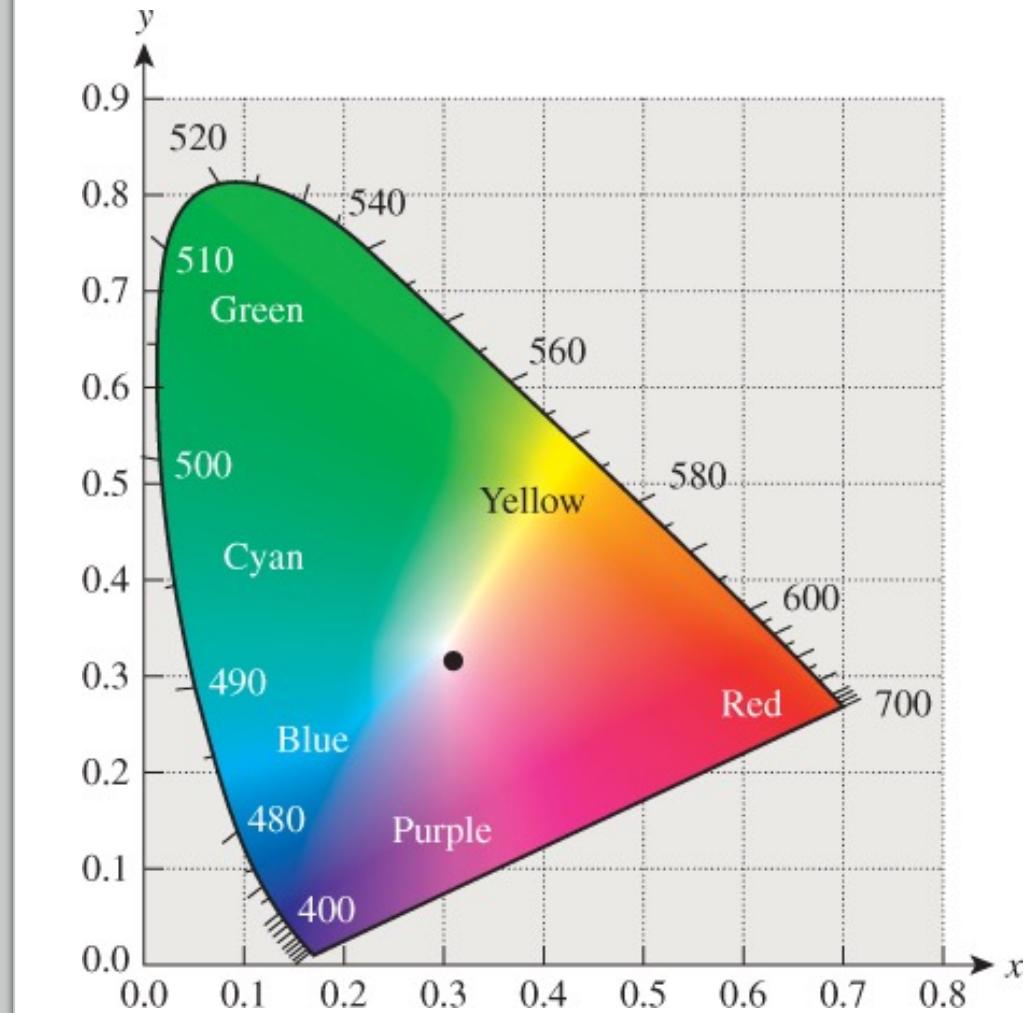
- Just a matrix transformation
- Converts between different colour systems

$$\begin{bmatrix} X \\ Y \\ Z \end{bmatrix} = \begin{bmatrix} 2.36460 & -0.51515 & 0.00520 \\ -0.89653 & 1.42640 & -0.01441 \\ -0.46807 & 0.08875 & 1.00921 \end{bmatrix} \begin{bmatrix} R \\ G \\ B \end{bmatrix}$$



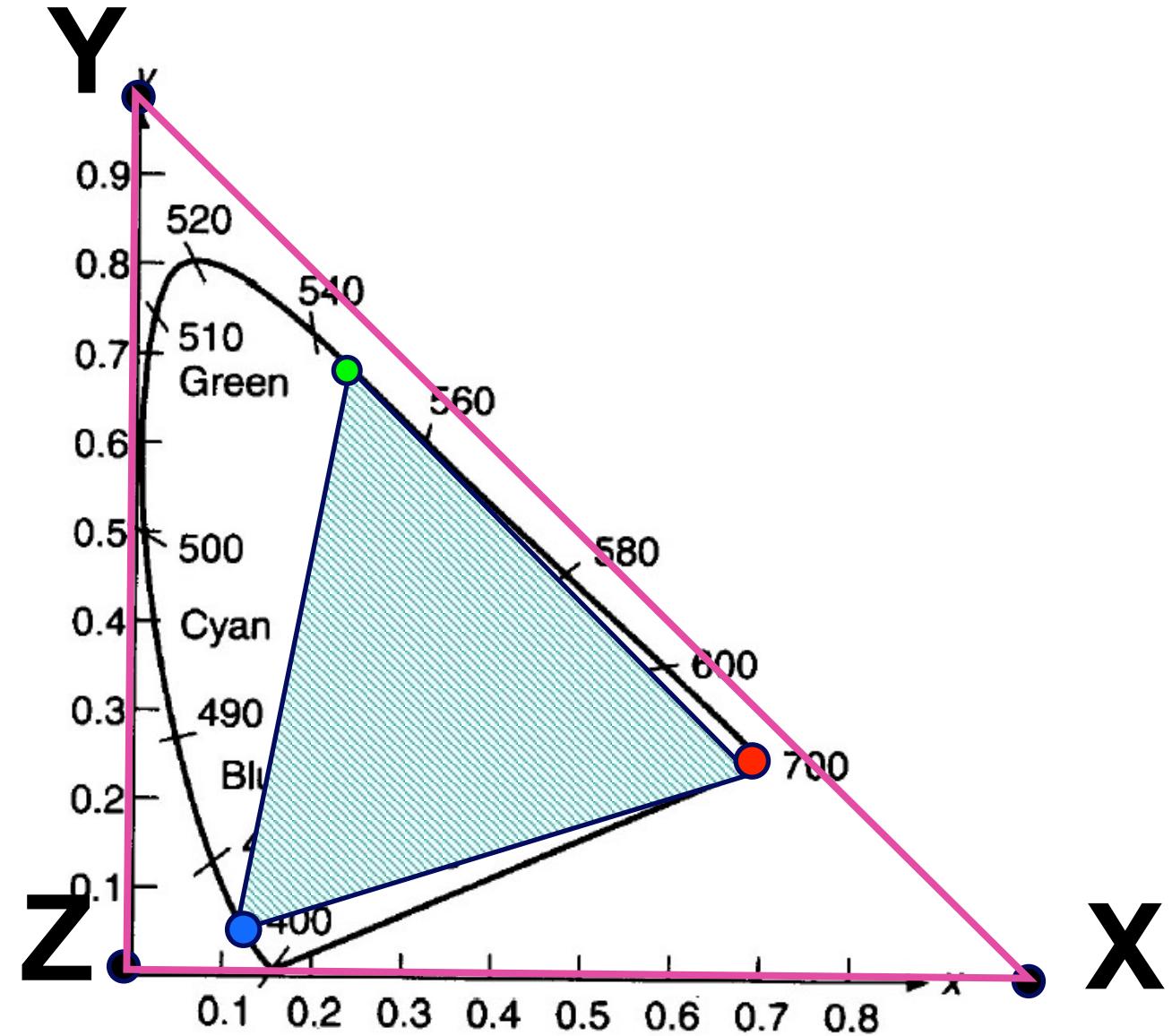
CIE Chromaticity Diagram

- Project colour space to plane $X + Y + Z = 1$
- Pure colours appear along the edge



sRGB Gamut

- Set of colours possible with standard RGB (sRGB)
- Convex hull of positions on the diagram
- Each device has a gamut
- Used to calibrate colour perception
- Still a major problem



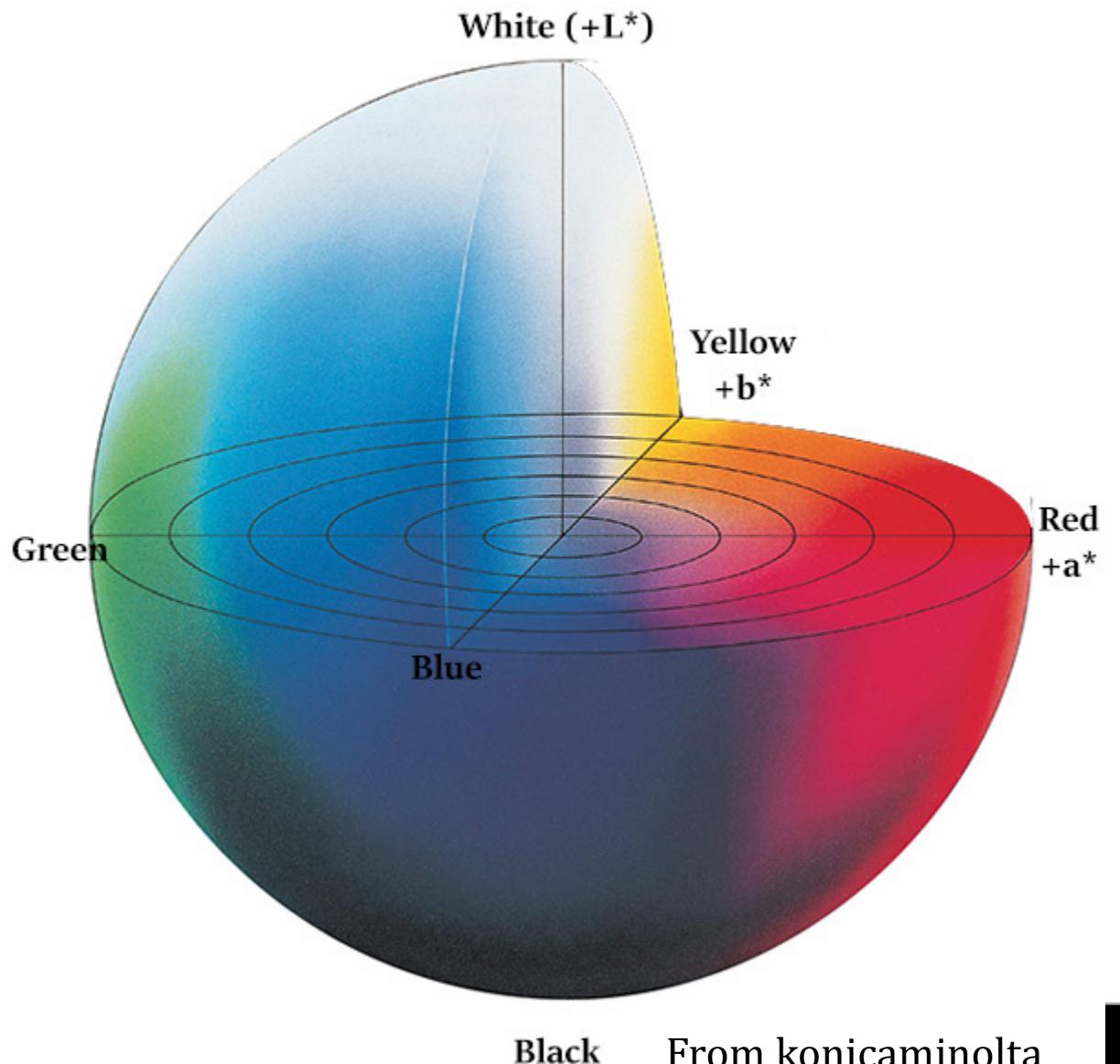
YUV Colour Space

- Early TV was a monochrome signal
 - Measured total intensity Y of light
 - Axis from black to white
- Colour TV is backward compatible
 - Y component is still transmitted
 - Components U,V were added



CIELAB Color space

- Perceptually uniform space:
 - Separates light and chromacity (similarly to YUV)
 - But in this case not focused on transmission, but perception.
- A change of X in one of the axis should match how much change is perceived.
- Computing in 8 bits will break things. Use floats

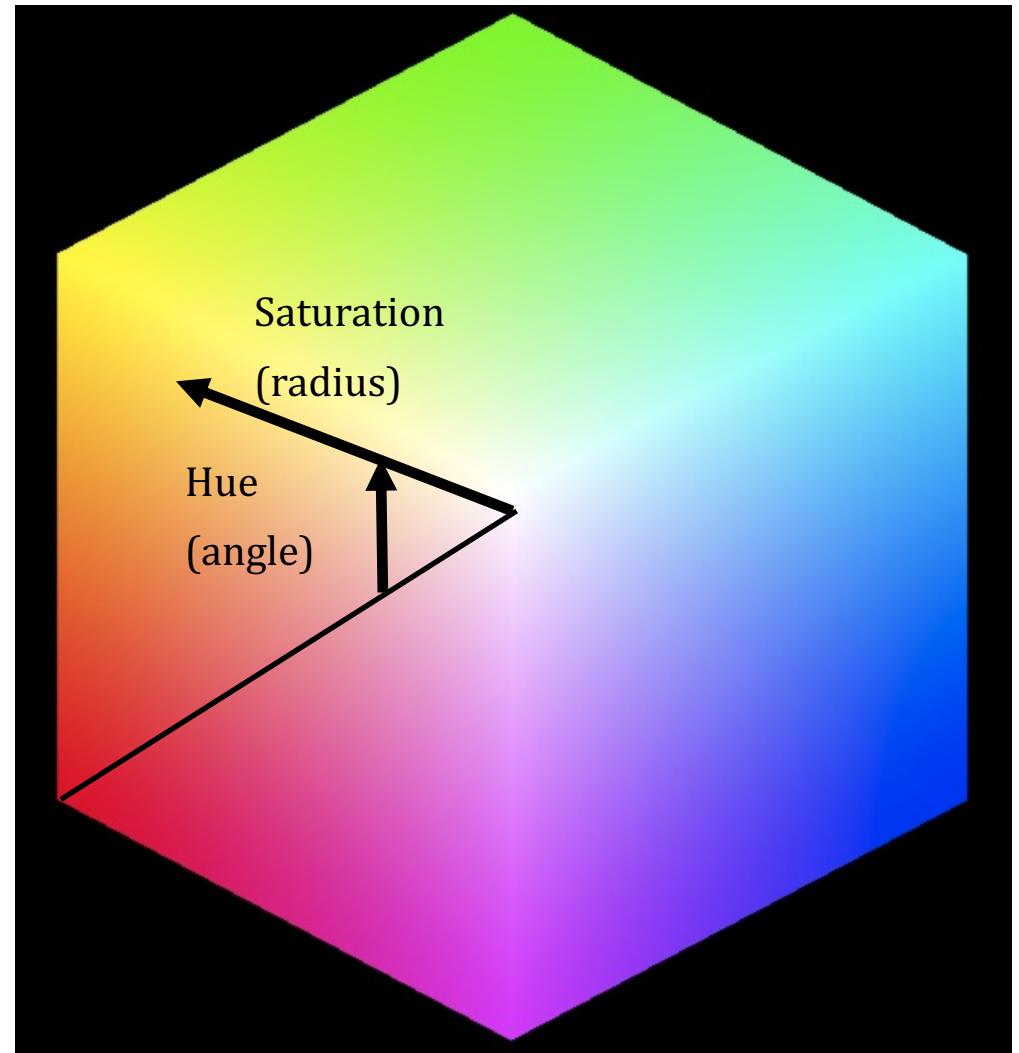


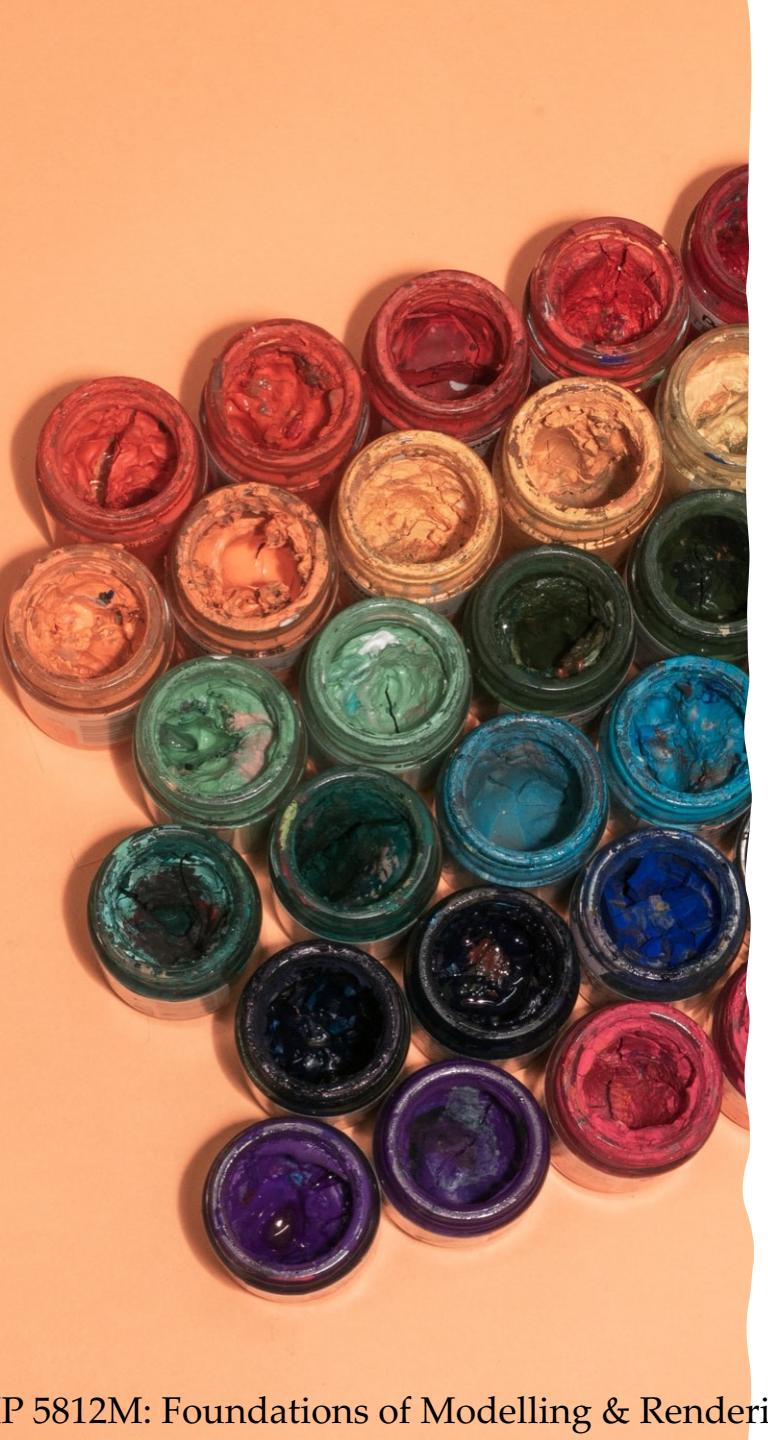
From konicaminolta



HSV Colour Space

- The *colour wheel*
 - H - hue
 - S - saturation
 - V - value
 - (or B - brightness)





Choosing Colours

- How do we get “yellow”?
 - start with the rainbow: R O Y G B I V
 - mix R & G to get Y
- Or look at the colour wheel
- Or just play with RGB until we get it right

Additive Colour

- These models assume light is added
 - Colour is the sum of components
 - Suitable for combining light sources
- But this isn't the only way of doing it
 - Go back to the physics again

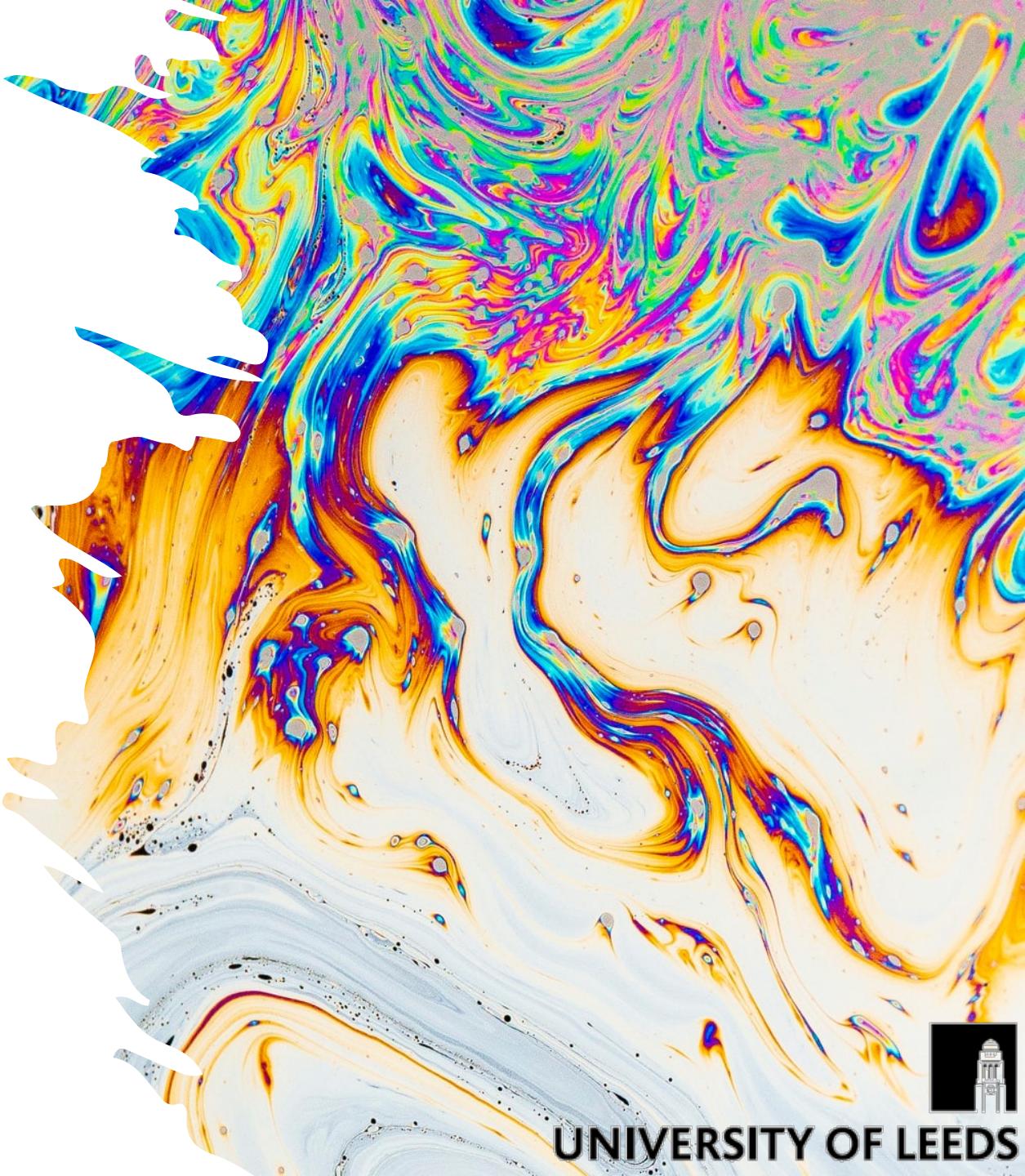


Colour Absorption

- Objects are coloured by pigments
 - that absorb certain colours of light
- E.g. chlorophyll in leaves
 - absorbs almost all red / blue light
 - but reflects green light
 - so output light is green

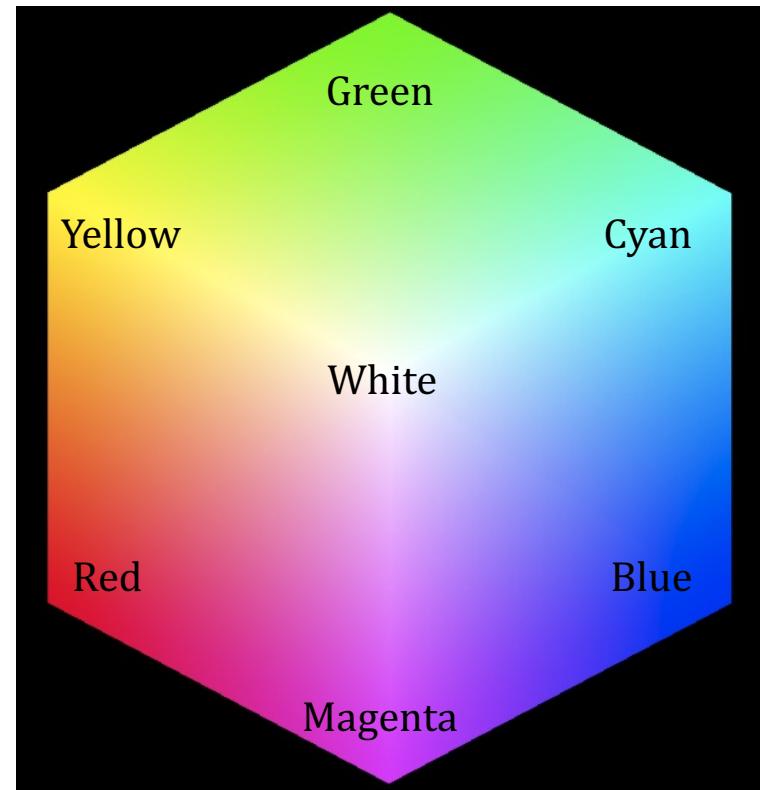
Subtractive Colour

- For pigments, we subtract colour
 - Blue + Yellow = Green
 - blue reflects some green as well
 - so does yellow
 - green is only colour reflected by both
 - We learned this colour model in school



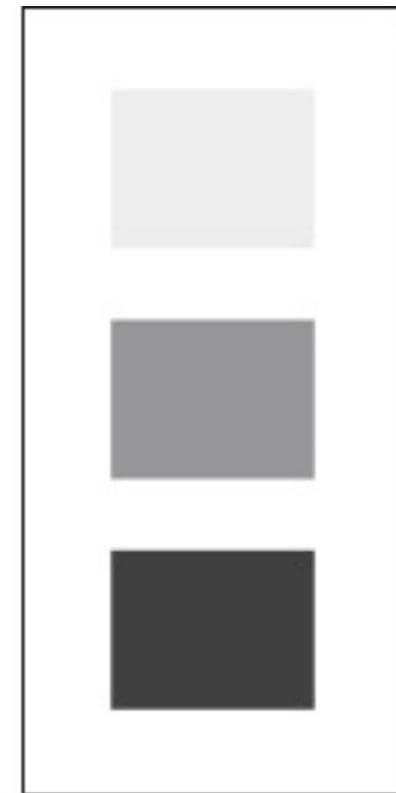
CMYK model

- Printers use CMYK
- Cyan removes red
- Magenta removes green
- Yellow removes blue
- Black removes everything
- Question: Can you run out of color ink just printing black and white?



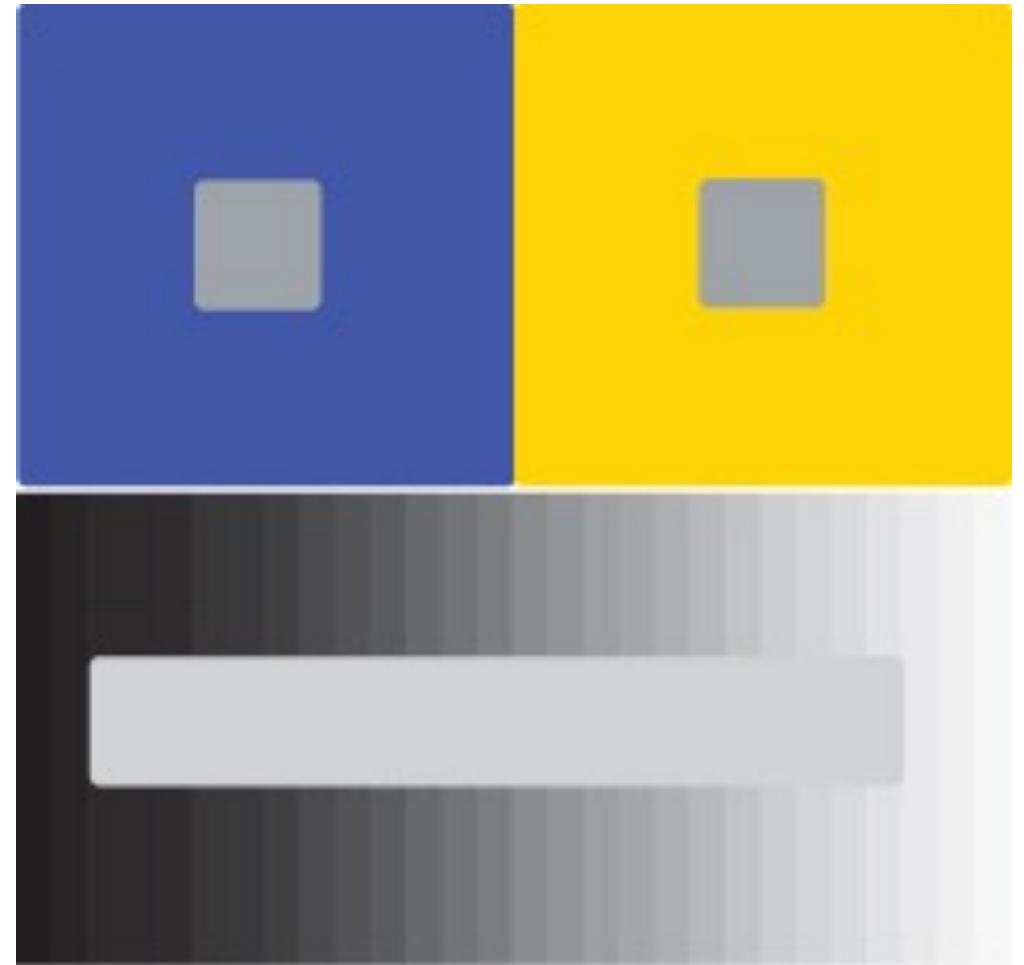
Tone Perception

- Cones don't act separately
- Retina combines signals from cones
- Perception is affected by context
- Which set of squares is lighter?



Simultaneous Contrast

- Tone perception also occurs in colour
- Which greys are the same here?



Colour Interpolation

- Interpolating nearby colours works well
- Interpolating distant colours usually fails
- So films are starting to add wavelengths
 - Known as spectral rendering
- Will we end up with ROYGBIV devices?
 - Time will tell
- In the meantime, assume RGB

Images by

- S3: Braxton Apana
- S4 S10: Sigmund
- S21: Alexandra Tran
- S22: Robbie Dawn
- S25: FLY:D
- S26: ThomasCarlPion
- S29: Etienne Girardet
- S49: Victor Freitas
- S62: Sarah Arista
- S65: Tanalee Youngblood
- S66: Daniel Olah

S50: buzzfeed.com

