

1:
$$\dot{m}_1 = C_1 \dot{p}_1$$

$$\frac{P_s - P_d}{R_1} - \frac{P_1 - P_2}{R_2} = C_1 \dot{p}_1$$

$$C_1 \dot{p}_1 - \left(\frac{1}{R_1} + \frac{1}{R_2}\right) p_1 = -\left(\frac{P_s}{R_1} + \frac{P_2}{R_2}\right)$$

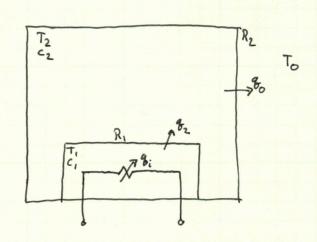
2:
$$\dot{m}_2 = C_2 \dot{p}_2$$

$$\frac{P_1 - P_2}{R_2} = \frac{C_2}{P_2}$$

$$C_{1} \dot{P}_{1} + \left(\frac{1}{R_{1}} + \frac{1}{R_{2}}\right) P_{1} = \frac{P_{5}}{R_{1}} + \frac{P_{2}}{R_{2}}$$

$$C_{2} \dot{P}_{2} + \frac{1}{R_{2}} P_{2} = \frac{P_{1}}{R_{2}}$$

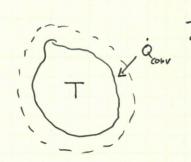
3)



$$C_{1}\overset{\circ}{T}_{1} = g_{2} - \frac{1}{R_{1}} \left(T_{2} - T_{2} \right) \longrightarrow C_{1}\overset{\circ}{T}_{1} + \frac{1}{R_{1}} T_{1} = g_{1} + \frac{1}{R_{1}} T_{2}$$

$$C_{2}\overset{\circ}{T}_{2} = \frac{1}{R_{1}} \left(T_{1} - T_{2} \right) - \frac{1}{R_{2}} \left(T_{2} - T_{0} \right) \longrightarrow C_{2}\overset{\circ}{T}_{2} + \left(\frac{1}{R_{1}} + \frac{1}{R_{2}} \right) T_{2} = \frac{1}{R_{1}} T_{1} + \frac{1}{R_{2}} T_{0}$$

If Cz were a really small number the increase in temperature to be really large to balance and keep equality.



To P, +, cp, A, h

$$\frac{\rho + c_p ?}{h A} \mathring{T} = \mathbb{I}_0$$

$$\dot{E} = -\frac{1}{R} \left[T_{S} - T_{A} \right]$$

$$Pc_{P} + T_{S} = -\frac{1}{R} [T_{S} - T_{A}]$$
; $R = \frac{1}{hA}$
 $Pc_{P} + T_{S} = -\frac{1}{hA} [T_{S} - T_{A}]$

$$h = \frac{\rho c_p + T_s}{A \left[T_A - T_s\right]}, \quad \forall = \frac{4}{3} \pi \left(\frac{D}{2}\right)^2, \quad A = 4\pi \left(\frac{D}{2}\right)^2$$

$$h = \frac{1 p D^{2} c_{0} T_{s}}{6 [T_{A} - T_{s}]} | p = 7920 \frac{k_{s}}{n^{3}}$$

$$D = 25E^{-3} m$$

$$S_{p} = 500 \frac{7}{k_{s}} c$$

$$T_{s} = 76 c$$

$$T_{A} = 22 c$$

$$T_{s} = -0.05 c/s$$

h = 0.38+ 15.27 200